# PHYSICAL PROPERTIES OF CHEMICAL COMPOUNDS-II

A systematic tabular presentation of accurate data on the physical properties of 476 organic straight-chain compounds compiled by R. R. Dreisbach of the Dow Chemical Co. These comprehensive and basic data were determined for specially prepared high purity compounds. In addition to the precisely measured properties, the author has calculated new values for many constants based upon his new experimental values.



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### Introduction

This is a continuation of Mr. Dreisbach's compilation of physical properties of organic compounds. Data on 511 cyclic compounds were published in 1955 as Number 15 of the Advances in Chemistry Series under the title of "Physical Properties of Chemical Compounds." The present volume includes 476 acyclic compounds.

As in the earlier volume, this compilation contains many data not hitherto published. It also includes parameters which can be used for interpolating and extrapolating the determined data for practically all of the compounds listed.

> H. S. NUTTING, Director Central Research Index The Dow Chemical Co.

# Physical Properties of Chemical Compounds

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## Definition of the Symbols and Parameters Used, with the Methods of Calculating the Parameters

Mol. % Pur.: Mole % purity by weight.

F.P.: Freezing point, °C.

F.P. 100%: Freezing point curve extrapolated to 100% purity.

B.P. 760 mm., 100 mm., etc.: Boiling points at these pressures, °C.

P<sub>25</sub>: Pressures at 25°C., in mm.

 $P_{\epsilon}$ : Pressure corresponding to temperature  $t_{\epsilon}$  in mm.

d<sup>20</sup>, etc.: Density at 20°C., etc., g./ml.

a. b: Constants of Law of Rectilinear Diameters,  $d_v + d_L = a + bt$ .  $d_v = density of the vapor, g./ml.; d_L = density of the liquid, g./ml.$ 

n<sub>D</sub><sup>20</sup>, etc.: Refractive index for the sodium line at 20°C., etc.

C: Constant of the Eykman equation,  $(n_D^2 - 1)/(n_D + 0.4) \times 1/d = C$ 

MR (obsd.): Molal refraction (obsd.) =  $(n_D^2 - 1)/(n_D^2 + 2) \times M/d = MR$  at 20 °C. (M = mol. wt.)

MR (calcd.): Molal refraction calculated from atomic refractive indices. See page 8.

- $(n_D d/2)$ : Refractivity intercept equals refractive index minus one half the density, both at the same temperature, 20°C.
- D: Dielectric constant determined at a frequency of 10<sup>5</sup> (cycles/sec.) and at 25°C. unless otherwise noted. When reported as data of The Dow Chemical Co., error about ±0.005. Where Reference 5 is noted it was obtained by squaring the refractive index at 20°C.
- A, B, C: Constants of the Antoine vapor pressure equation for the liquid state, giving P (pressure) in mm. and t (temperature) in °C. This is in the range between the temperatures as indicated. These temperatures in general are the boiling point at 30 mm. to a  $T_R$  of 0.75 to 0.80. See method of obtaining A, B, C on page 6.

Antoine equation:  $\log P = A - B/(t + C)$ .

 $A^*$ ,  $B^*$ , K, c,  $t_k$ ,  $t_x$ : Constants of the saturated vapor density equation

$$\log d_v(g./ml.) = A^* - B^*/(t+C)$$
 to the temperature  $t_k$ .

$$\log d_v(g./ml.) = A^* - B^*/(t+C) + K/(1.1 T_c - 273.2 - t) + c$$

from temperature  $t_k$  to a reduced temperature,  $T_R$ , of 0.92

 $t_k$  = Temperature at which it is necessary to change from the simple vapor density equation to the corrected vapor density equation in the higher ranges, °C.

$$t_k = t_x + K/c$$
 and  $t_x = (1.1 T_c - 273.2)$ °C.

 $A^*$  and  $B^*$  where the latent heat at the atmospheric boiling point is available.

$$V_q - V_L = (31381.7 \times \Delta Hv \times dt/dp)/T$$

Where the latent heat is not available use

$$M(\Delta Hv)/T_B = 21.0$$
 and from this  $\Delta Hv = (T_B \times 21.0)/M$ 

The value 21.0 (or any other value as 21.4 say) is obtained from the nearest related compound which has a latent heat available. Then proceed as in case where latent heat is available for Vg value at B.P.

Since  $d_v = 1/Vg$ 

$$\log d_{v760} = A^* - B^*/(t_B + C)$$
 at 760 mm.

$$\log d_{v30} = A^* - B^*/(t_{30} + C)$$
 at 30 mm.

Solve for  $A^*$ ,  $B^*$ , since t and  $d_v$  at 760 mm. and 30 mm. and C are known.

- A', B', C': Constants of the Antoine vapor pressure equation below 30 mm. pressure, covering the temperature range as indicated. See method of obtaining the constants on page 6.
- $A'^*$ ,  $B'^*$ : Constants of the vapor density equation below 30 mm. These two values are obtained by using the boiling point at 30 mm. and the pressure at 25°C. (obtained from the values A', B', C') and assuming that at 25°C. the relationship PV/RT = 1. Then we have Vg at 25°C. = RT/MP = 62,  $361 \times (25 + 273.2)/MP$ .

Then dv = 1/Vg. Inserting these values of vapor density we then solve the two equations for the values of  $A'^*$  and  $B'^*$  as in the case of  $A^*$  and  $B^*$ .

- Ac, Bc, Cc: Constants of the Antoine vapor pressure equation for the liquid state from  $T_R = 0.75$  (or a higher  $T_R$  as indicated) to the critical temperature. See method of obtaining the constants on page 7.
- Cryoscopic Constants, A°, B°: Cryoscopic constants for calculating mole % purity. See J. Research Natl. Bur. Standards, 35 (1945); RP 1676.

t, °C.: Temperature at which a mole of the vapor occupies 22.414 liters and the vapor is in equilibrium with the liquid, in °C.

$$te = \frac{B^*}{(A^* - \log dv_e)} - C$$

dt/dp: Rate of change of boiling point with pressure, given by equation  $dt/dp = B/(2.3026 \times P \times (A - \log P)^2)$  °C./mm. Also  $dt/dp = (t + C)^2/2.3 PB$ 

 $\Delta Hm$ : Latent heat of fusion in cal./g.

 $\Delta Hv$ : Latent heat of vaporization at the temperature designated, cal./g.

 $t_r(d, e)$ : The latent heat of vaporization at temperature  $t_e$  as given by the equation  $\Delta H_v = d - et$ , indicates the accuracy of this equation at temperature  $t_e$ .

 $\Delta H_v/T_e$ : Molal latent heat of vaporization at  $t_e$  divided by  $T_e$ . (Equal to the mola entropy of vaporization at  $t_e$ .)

- d, e; d', e': These are parameters of the latent heat of vaporization equation.  $\Delta Hv(\text{cal./} g.) = d et$ . This is valid between the temperatures indicated. It has been found that the latent heat between the boiling point at 30 mm, and the boiling point at 760 mm, is almost a linear function of the temperature. As seen in most cases, this equation holds almost to temperature  $t_e$ . Above and below this the latent heat is not linear with temperature except for short intervals.
- $d_c$ : Critical density, g./ml.
- $v_e$ : Critical volume, ml./g.
- t<sub>c</sub>: Critical temperature, °C. See also page 7.
- $P_c$  mm.: Critical pressure in mm. Where this was not obtained from the literature it is calculated as follows (The Thomson method, private communication from George W. Thomson): The critical temperature is inserted in the Antoine equation, using the A, B, and C values to calculate the critical pressure.

This value is too low. This is then multiplied by 1.07 and is assumed to be the critical pressure. In the great majority of cases, this will agree with determined values to within  $\pm 3\%$ . For high boiling compounds this value must be decreased, since in most cases there is somewhat irregular drift with increasing temperature, so this should be continually lowered as the boiling point becomes increasingly higher.

PV/RT: Compressibility at the temperature designated.

$$z = PV/RT$$

where P = pressure in mm., V = volume in ml./mole, and R = 62361.

ΔHc: Heat of combustion, kcal./mole, gas at constant pressure, 298.16 °K. or 25 °C.

ΔHf: Heat of formation, kcal./mole, liquid at 298.16°K. or 25°C.

ΔFf: Free energy of formation, kcal./mole, liquid at 298.16°K. or 25°C.

η: Kinematic viscosity in centistokes, at temperature designated. The kinematic viscosity is given by the equation

$$\log \eta = A^v + B^v/T$$

between the temperatures indicated to an accuracy of 1% or better.

B.P. °C., 30 mm.; dt/dp;  $\Delta Hv$ ; PV/RT: These values at 30 mm. are calculated from the Antoine equation using A, B, and C. It has been found that at 30 mm. in almost all Cox chart families the ratio PV/RT is negligibly different from one. This, then, has been taken as one point (the other point being the B.P. at 760 mm.) from which to calculate  $A^*$  and  $B^*$ , always assuming the compressibility as 1.0000 at 30 mm.

- c<sub>p</sub>: Specific heat at constant pressure at temperature designated, cal./g. °K.
- c<sub>v</sub>: Specific heat at constant volume at temperature designated, cal./g. °K.
- f,g, h, f', g', h': Parameters of the heat capacity equation for the liquid for the temperature ranges designated, °K.  $c_p(\text{liquid}) = f + gT + hT^2$
- m, n, o, m', n', o': Parameters of the heat capacity equation for the vapor for the temperature ranges designated, °K.  $c_{v}(\text{vapor}) = m + nT + oT^{2}$
- γ: Surface tension in dynes/cm., at temperature designated.
- [P]: Parachor at the temperature designated:  $M(\gamma)^{1/4}/(d_L - d_v) = [P]$
- [P] Sugd.: Parachor from atomic and structural values as given by Sugden. See table. The parachor value for oxygen as hydroxyl (alcohols) in these tables is taken as 15. Sugden gives the values of 20 for oxygen and 30 for oxygen in esters, which does not seem to work for alcohols and phenols.
- Exp. L.l.; Exp. L.u.: Explosion limits lower and upper range, % by wt.

Dispersion: Specific dispersion,  $10^4(n_F - n_c)/d$ , ml./g. at 25°C.

 $n_F$ ,  $n_c$  = refractive index for F and C lines.

d = density, g./ml.

Flash and Fire Points, °C.: Cleveland open cup (ASTM D 92-46) if not otherwise designated. Closed cup (ASTM D 56-36) will be designated as such.

M Spec.: Mass Spectrograph.

Ultra V.: Ultraviolet.

X-Ray Dif.: X-Ray Diffraction.

Infrared: Infrared Spectrograph.

Solubility at 25°C., in solvents as designated.

Explanation of the methods used for calculating the various parameters in the foregoing:

A, B, C: The A, B, and C constants, except where given by the API reports, are calculated by means of the Thomson method [Chem. Revs. 38, 1-39 (1946)] using the determined boiling points at three different pressures. The three formulas for this are as follows:

$$(y_3-y_2)/(y_2-y_1)\cdot(t_2-t_1)/(t_3-t_2)=1-(t_3-t_1)/(t_3+C)$$
  $B=(y_3-y_1)/(t_3-t_1)\cdot(t_1+C)(t_3+C)$  and  $A=y_1+B/(t_1+C)$ 

where  $y_1$ ,  $y_2$ , and  $y_3$  are equal to  $\log P_1$ ,  $\log P_2$ , and  $\log P_3$  at temperatures  $t_1$ ,  $t_2$ , and  $t_3$ . Unless the data for the three points are *very* accurate, the C value can be considerably in error. As a check on this method an empirical formula developed by Thomson (private communication from George W. Thomson) will give a much better value of C if the data are much in error. This formula is  $C = 239 - 0.19t_B$ . The A and B values can then be readily determined from the two points given, since they are much less critical.

A', B', C' (for pressures below 30 mm.): Applicable when molar heats of vaporization are available at 25°C. and the Antoine equation can be used to obtain the boiling point at 30 mm. Let A, B, C be the constants of the usual Antoine equation valid above 30 mm. and let A', B', C' be the constants of the Antoine equation sought for below 30 mm. These two equations are taken to give the same value of the pressure-temperature slope at 30 mm.

$$\log 30 = A - B/(t_1 + \hat{C}) = A' - B'/(t_1 + C')$$

$$B/(t_1 + C)^2 = B'/(t_1 + C')^2$$

Since PV/RT may be assumed to be 1.0000 at  $t_1$ , the temperature corresponding to 30 mm., and is also 1.0000 at 25 °C., the molar heat of vaporization at 25 °C.,  $M\Delta Hv_2$  is given by

 $M\Delta H v_2 = 2.3026~RB'~[(t_2 + 273.2)/(t_2 + C')]^2$  where  $t_2 = 25$  °C. To solve for A', B', C' let  $g_2 = M\Delta H v_2/2.3026~R(t_2 + 273.2)^2 = M\Delta H v_2/406883$  if  $t_2 = 25$  °C.

Since  $t_1$ ,  $t_2$ , and all values on left-hand side of equations above are known, then B' and C' are readily obtained as follows:

$$[B'/(t_2 + C')^2][(t_1 + C')^2B'] = g_2(t_1 + C')^2/B' = \text{say, } h^2$$
  
Then  $C' = (t_1 - ht_2)/(h - 1)$  and  $B' = g_2(t_2 + C')^2$   
Also  $B' = B[(t_1 + C')/(t_1 + C)]^2$   
 $A' = \log 30 + B'/(t_1 + C')$  since  $P_1 = 30$  mm.

These formulas were developed with the aid of George Thomson.

When heats of vaporization at 25°C. are not known:

In this case the C' value is estimated and A' and B' are calculated from known data. It was noticed that C' has a value approximately 18 higher than C when latent heats at 25°C. are known. By adding this increment to C we have C', then B' from the relation for the first case

$$B' = B[(t_{30} + C')/(t_{30} + C)]^2$$
 and then A' as in first case.

In the case of the alkenes and alkynes the A', B', C' and  $A'^*$ ,  $B'^*$  were not calculated by the above method, since the data for these compounds is much less reliable than in the case of the alkanes.

Ac, Bc, Cc: This method was developed by George Thomson [Chem. Revs. 38, No. 1, 23 (1946)] and is similar to the one for obtaining A', B', C'. It is assumed that parameters A, B, C of the Antoine equation are good to a  $T_R$  0.75 or a higher reduced temperature, and this temperature corresponds to the 25°C. in the case of A', B', C', and the critical point corresponds to the 30-mm. point.

B/
$$(t_1 + C)^2 \times (t_c - t_1)/(y_c - y_1) = 1 + (t_c - t_1)/(t_1 + Cc)$$
  
and  $Bc = (y_c - y_1)/(t_c - t_1) \times (t_1 + Cc)(t_c + Cc)(t_c + C_c)$ ; Ac = B/ $t_c$  + C<sub>c</sub> + y<sub>c</sub> where  $t_1$  °C. =  $T_R$  0.75,  $t_c$  °C. = critical temperature  $y_1 = \log P$  at  $t_1$ ,  $y_c = \log P_c$ 

The first equation is used to evaluate Cc, the second, Bc, and the third, Ac.

Association: The association in the vapor phase of organic acids seems to vary inversely as the temperature for some acids, at least for part of the range. In part of the range, and also apparently for some acids over the whole range, the association is fairly constant. The association is given in these sheets by the formula  $M_x = p - rt$ . For instance, for acetic acid this formula would be  $M_x = 2.225 - 0.004085 t$  from 0° to  $100^{\circ}$ C. From  $100^{\circ}$ C. to a  $T_R$  of 0.92,  $M_x = 1.85$ . That is to say, the vapor density as calculated by the  $A^*$ ,  $B^*$  formula would have to be multiplied by this correction factor to take care of the association. Further, if the reciprocal of the density is used as calculated to give vapor volume, it would be necessary to divide by 1.85 to get the actual vapor volume.

t<sub>c</sub>: Where the critical temperature has not been determined, it is calculated by Watson's equation:

 $T_e/T_c = 0.283 \ (M/d_s)^{0.18}$ 

where  $d_s = \text{liquid density, g./ml.}$  at the boiling point, and M = molecular weight. This is used for all hydrocarbons and halohydrocarbons.

f, g, h, m, n, o, etc.: For a short temperature range the equation  $C_p = f + gT + hT^2$  reproduces almost exactly determined data. The parameters were set up on the IBM machines using eight determined values where that many or more were available.

The IBM machines were used to set up the Antoine constants from determined data. A preliminary C value was obtained from the equation C=239.  $-0.19t_B$ . A and B were then obtained and new C values either side of the first C used and new A and B values found. In each case above, the boiling points at the experimental pressures were calculated and compared with the determined boiling points.

Actually the value of C was generally obtained from C = 239.  $-0.19t_B$ , since the determined values must be *very very* accurate to give better values of C.

#### **Cox Chart Families**

1. Alkanes

Haloalkanes
 Alkenes

4. Haloalkenes

5. Diolefins

6. Alkynes

### Atomic Refractive Indices Used for Computing Molecular Refractive Index

All values are for the sodium line.

All V	alues are for th	e sodium line.	
Carbon singly bound and alone Carbon singly bound Carbon double bond Carbon triple bond Carbon conjugated Hydrogen Oxygen—hydroxyl Oxygen—ethereal Oxygen—ketonic Oxygen—as ester Sulfur—as SH Sulfur—as RSR Sulfur—as RCNS	2.592 2.418 1.733 2.398 1.27 1.100 1.525 1.643 2.211 1.64 7.69 7.97	NO as nitrites NO as nitrosoamine NO <sub>2</sub> as alkyl nitrite NO <sub>2</sub> as alkyl nitrate NO <sub>2</sub> as nitroparaffin NO <sub>2</sub> as nitro aromatic NO <sub>2</sub> as nitramine Fluorine Chlorine Bromine Iodine	5.91 5.37 7.44 7.59 6.72 7.30 7.51 0.95* 5.967 8.865 13.900
	7.91		
Sulfur—as RSSR Nitrogen	8.11		
As aliphatic primary amine	2.45		
As aromatic primary amine	3.21		
As aliphatic secondary amine	2.65		
As aromatic secondary amine	3.59		
As aliphatic tertiary amine	3.00		
As aromatic tertiary amine	4.36		
As hydroxylamine	2.48		
As hydrazine	2.47		
As aliphatic cyanide	3.05		
As aromatic cyanide	3.79		
As aliphatic oxime	3.93		
As primary amide	2.65		
As secondary amide	<b>2</b> . $27$		
As tertiary amide	2.71		

<sup>\*</sup> This value for one fluorine atom attached to carbon. The value 1.1 is to be used for each fluorine atom in polyfluorides.

#### Atomic and Structural Constants for Calculation of Parachor

	Sugden		Sugden
CH <sub>2</sub>	39.0	Br	68.0
C	. 4.8	[	91.0
H		Single bond	
0		Double bond	
O (Alcohol)		Triple bond	46.6
O <sub>2</sub> (Ester)		B-membered ring	16.7
N		I-membered ring	11.6
N (nitrile)		5-membered ring	8.5
S		6-membered ring	6.1
F		-membered ring	
Cl		$\mathbf{M}$ iphatic alcohol $oxed{1}$ subtract	6.0

								No. 1	
NAME	Methane					ST	RUCTURAL	FORMUL	A
							CVI		
Mole % Pur.		ecul		Molecular Weight 16.04	42		CH <sub>4</sub>		
	•	Ref.			Ref.				Ref.
F.P. ℃	-182.48	2	dt/dP			f	to		
F.P. 100%		_	°C/mm -140°C	0.00537	5	g	<u>°K</u>		
B. P. °C 760 mm	-161.49	2	BP	0.0160	2	h			-
100 30	-181.45	2	t <sub>e</sub>	0.3567	5	f' g'	to *K		
10	-190. -195.51 <sup>≠</sup>	2	30 mm	0.2141	2	h'			
1	-207.0	5	ΔHm cal/g ΔHv cal/g	14.025	-	m	300 to	0.1259	4
Pressure mm -140°C	3289 7	5	-140°C	115.67	3	n	600 <b>⊸</b> K	0.0015	4
t <sub>e</sub>	272.1	5	30 mm BP	133.34 121.87	5 2			0.0666	-
Density			t_	125.30	5	m'	700 to	0.1408 0.0013	4
g/ml-160°C	0.4222 0.4075	<b>3</b>	te (d, e)	126.36	5	0'	1000 1	-0.0 <sub>6</sub> 34	4
d <sup>t</sup> -150 4 -140	0.3916	3	ΔHv/T <sub>e</sub>	19.99	5	Sur	face tension	- 0	-
a -160°C	0.4242	3	d -190 to e -160 °C	56.90 0.4023	5		es/cm180℃		2
Bef. Index	-0.00126	4	d' to			8	-170 -160	15.8 13.7#.	2
n <sub>D</sub> 20°C			e' °C	0.1/2	-	Par	achor [P]	236 · M.	
25 30			d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.162 6.17	2 2		,20°C		
"C"		$\vdash$	1 -	-82.5	2		30 <b>4</b> 0		
MR (Obs.)		-	P <sub>c</sub> mm	34808.	2		Sugd.	73.2	5
MR (Calc.)	6.818	5	PV/RT -140°C	0.9208	4	Exp	. L.1.%/wt. u.		
(nD-d/2) Dielectric			30 mm	1.0000	5	Dis	persion		
A -180 to	6,61184	2	BP t <sub>e</sub>	0.9628 0.9726	4 5		sh Point C		
B ∟°C_	389.93	2	tc	0.290	2		e Point		
C	266.00	2	ΔHc kcal/m ΔHf	191.76	2		Spec. raV.		
A*L165 to B*L130 °C	0.74077 362.78	4	ΔFf				lay Dif. ared	Yes	2
K	7.79	4	Viscosity				bility in +		Ē
t <sub>k</sub> [-130 to	-0.15125	4	centistokes 7 -180 °C	0.418	2	Ac	etone	!	
t <sub>x</sub>  -100 °C	-63.0	5	-175	0.364	2		rbon tet. nzene		
A'   to B'   °C			-170 -165	0.325 0.295	2 2	Et	her		
č. – – –			BV -180 to	101.76	4		Heptane hanol		ĺ
A!* to			AV  -160 ℃	₹,52937	4		ter		
B'* °C			(B <sup>V</sup> )  to				ter in		-
Ac to Bc t <sub>c</sub> °C			(A <sup>V</sup> )  °C	0.50000					
Cc			c <sub>p</sub> liq.300 °K 400	0.53310	2 2				
Cryos. A° consts. B°	0.0138 0.0057	2 2	c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	-172.65	4	c <sub>v</sub> vap.						
# solid		نا	# at saturation	on pressure	لـــــا	+ ~-	ams/100 gra	me ecles-	<u> </u>
	ES: 1-Dow	2-A		Calc. from de	t. da				
SOURCE:		AP					,		
PURIFICAT	ION:	AP							
	E REFERE								
			-						

NAME	Ethane					STRUCTURAL FORM	ULA
						сн,сн,	
Mole % Pur.	Ref. Mo	lecul rmul	ar C <sub>2</sub> H <sub>6</sub>	Molecular Weight 30.06	68	0303	
	<del> </del>	Ref.			Ref		Ref
F.P. °C	-183.27 <sup>≠</sup>	2	dt/dP	Γ		f to	
F.P. 100%			*C/mm		۱. ا	g   <u>*K</u>	
B. P. °C			25°C BP	0.00213 0.0244	5 2	h .	- 1
760 mm 10 <b>0</b>	-88.63 -119.33	2 2	t <sub>e</sub>	0.0351	5	f' to	
30	-132.74	4	30 mm	0.3350	5	g'	į
1 <b>0</b> 1	-142.88 -159.51	5	ΔHm cal/g	22.728	2	h'	
Pressure		<u> </u>	ΔHv cal/g			m to	
mm 25°C	29290.	5	25°C	76.86	5 5	n   *K	
t <sub>e</sub>	480.9	5	30 mm BP	129.71 116.97	2	<u> </u>	
Density			l t_	119.45	5	m' to co	
g/ml-60°C	0.509 0.363	3	t <sub>e</sub> (a, e)	119.33	5	o'   '	
d <sub>4</sub> 30	0.30	3	AHv/Te	20.36	5	Sunface tension	
a	0.364	4	d   -132 to	91.37 0.2888	5 5	Surface tension dynes/cm110°C 19.5	57 2
ь	-0.0363	4	_d'to	0.2000		-100 17.9	
Ref. Index			e' °C			-90 16.3	31 2
<sup>n</sup> D 20°C			d g/ml	0.203	2	Parachor [P] 20°C	
30	<u> </u>		d g/ml vc ml/g tc °C	4.92 32.27	2 2	30	
"C"			P <sub>c</sub> mm	36632.	2	40 Sugd. 112. 2	.   5
MR (Obs.)		ł	PV/RT		$\vdash$	Exp. L.1.%/wt.	-
MR (Calc.) (nD-d/2)	İ		25°C	1.0000	5	u.	
Dielectric		<b>-</b>	30 mm BP	1.0000 0.9622	5 5	Dispersion	
A  -132 to	6,80266	2	te	0.9735	5	Flash Point °C Fire Point	
B 1-44 °C	656.40	2	t <sub>c</sub>	0.288	2	M Spec.	
C	256.00	2	ΔHc kcal/m	341.26	2	Ultra V.	
A*  -132 to B*   -90 °C	0.98156	5	ΔFf	1		X-Ray Dif. Infrared	- 1
K	1		Viscosity			Solubility in +	
t			centistokes 7 -120 °C	0.420	2	Acetone	
tk to	1	1	7 -120 ℃ -110	0.438 0.387	2	Carbon tet.	ĺ
A' to			-100	0.348	2	Benzene Ether	1
₿', ∟ _ <u>°</u> ⊆			-90 B <sup>V</sup>   <sub>-130</sub> to	0.314	2	n-Heptane	
A¹* to			A   -80 °C	135.71 2.75615	4 4	Ethanol Water	l
B'* °C	1		(B <sup>V</sup> ) to	-		Water in	
Ac   -44 to	7.67290	5	(A <sup>V</sup> ) °C				
Bc t °C	1096.9	5	c <sub>p</sub> liq. °K	<u> </u>	$\vdash$		
Cc — -	320.54	2	=				
Cryos. A° consts. B°	0.04256 0.0095	2	c <sub>p</sub> vap.300°K 400	0.41225 0.52148	2 2		1
t <sub>e</sub> °C	-96.7	5	c <sub>w</sub> vap.	",,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	-		
$T_{\mathbf{R}} = 0.79$			≠ at saturation	pressure	ш	grams/100 grams sol	
	ES: 1-Dow				dat	a 5-Calc. by formula	A CIII
SOURCE:		AI			,		
PURIFICAT	ION:	Al					
	RE REFERE					· · · · · · · · · · · · · · · · · · ·	
			- Louing				

							No. 3	
NAME	Propane					STRUCTURAL F	ORMULA	1
Mole % Pur.	Ref. Mol	ecul		Molecular Veight 44.09	4	СН <sub>3</sub> -СН <sub>2</sub> -(	CH <sub>3</sub>	
		Ref.			Ref.		F	Ref.
F.P. °C F.P. 100%	-187.69	2	dt/dP °C/mm			f to g °K		
B. P. °C 760 mm 100 30 10	-42.07 -79.63 -96.07 -108.5	2 2 4 4	25°C BP t <sub>e</sub> 30 mm	0.00561 0.0298 0.0355 0.4109	4 2 5 4	h   to g'  °K		
Pressure mm 25°C	-129. 7095.	5 4	ΔHm cal/g ΔHv cal/g 25°C	19.10 81.76	2	m 300 to 600 K	0.0014	4
Density	608.5	4	30 mm BP te	114.70 101.76 102.98	5 2 5	m' 700 to	0.0960	4
g/ml 20°C dt 25 4 30	0.5005 <sup>‡</sup> 0.4928 <sup>‡</sup> 0.4861	2 2 4	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	102.94	5	n' 1000 <b>°K</b>	0.0012 -0.0 <sub>6</sub> 40	4
a b	0.5375 -0.0 <sub>3</sub> 99	4	d -100 to e -45 °C d' to	91.68 0.2396	5 5		19.2 17.85	2 2
Ref. Index n <sub>D</sub> 20°C 25 30	1, 2898	5	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	0. 220 4. 358	2 2	-50 Parachor [P] 20°C 30	16.49	2
"C"	0.766	5	tc°C P <sub>c</sub> mm	96.8 31928.	2 2	40 Suad 1	51 2	5
MR (Obs.) MR (Calc.) (nD-d/2)	16.054 1.04	5 5	PV/RT 25°C 30 mm	0.8461 1.0000	4 5	Sugd. 1 Exp. L.1.%/wt. u. Dispersion	51.2	
Dielectric  A -130 to B _ 5°C	1.66 6.82973 813.200	2 2	BP t <sub>e</sub> t <sub>c</sub>	0.9612 0.9669 0.278	4 5 2	Flash Point C Fire Point		
A* -100 to B* -40 °C K	248.00 1.05579 756.21	2 5 5	ΔHc kcal/m ΔHf ΔFf	488.53 -28.643	2 2	M. Spec. Ultra V. X-Ray Dif. Infrared	Yes	2
c t <sub>k</sub> -40 to t <sub>x</sub> 65 °C	17.62 -0.13012 134.0	4 4 5	Viscosity centistokes 7 -80°C -70	0.524 0.470	2 2	Solubility in + Acetone Carbon tet. Benzene		
A'   to B'   _ °C C' to			-60 -50 <b>B<sup>V</sup></b>   -85 to A <sup>V</sup>   -30 °C	0.425 0.387 190.47 2.73347	2 2 4 4	Ether n-Heptane Ethanol Water		
A'* to B'* °C Ac  5 to	7. 33829	4	(B <sup>V</sup> )  -140 to	184.09 Z.76598	4	Water in Viscosity		
Bc tc °C Cc Cryos, A°	1090.0 287.8 0.05802	4 4	c <sub>p</sub> liq. °K	0,40051	2	centistokes η -130°C -90	1.126	2 2 2
consts. B°	0,0073 -46.98	2 5	c <sub>p</sub> vap300°K 400 c <sub>v</sub> vap.	0.51118	2	-40	0.355	۷.
$T_{\mathbf{R}} = 0.75$			# at saturation			grams/100 gram		t
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by form	nula	
SOURCE:			PI					
PURIFICAT LITERATUE	ION: RE REFEREI		S:					

								No. 4	
NAME	n-Butane					STRUCTU	RAL 1	FORMULA	
						CH	CH CH		
Mole	Ref. Mo	10001		Molecular		CH <sub>3</sub>	сн <sub>2</sub> сн	2013	
% Pur.	Fo	rmul		Weight 58.12	0				
		Ref.			Ref.				Ref.
F.P. ℃	-138.350	2	dt/dP			f	to		
F.P. 1009	<b>'</b>		*C/mm 25*C	0,01768	4	8   _	_ •K		
B. P. *C 760 mm	-0.50	2	BP	0.03465	2	h +			
100	-44.17	2	t <sub>e</sub>	0.03597	l I	f' g'	to °K		
30 10	-63.30 -77.76	4	30 mm	0.4778	4	h' -	<b>^</b> `-		
1	-101.5	5	ΔHm cal/g	19.167	2		00 to	0.0351	4
Pressure mm 25°C	1,022		ΔHv cal/g 25°C	86, 63	2	n   _6	00 °K	0.0013	4
mm 25-C	1823. 724.6	5	30 mm	105.08	5	0		-0.0 <sub>6</sub> 40	4
Density	1 1	$\vdash$	BP t <sub>e</sub>	92.09 92.35	2		00 to	0.0974	4
g/ml 20°C	0.5788 0.5730	2	(4,0)	92.35	5	n'   110	00 •K	0.0012 -0.0 <sub>6</sub> 40	4
d <sup>t</sup> 25 4 30	0.5671	2	ΔHv/T <sub>e</sub>	19.77	5			6.0	<u> </u>
	0.6039	4	d   -60 to	91.99 0,2069	5	Surface tendentes dynes/cm.		18.43	2
ь	-0.0399	4	<b> -å,-¦ − %-%</b>		4	*	-20	17.22	2
Ref. Index	· 1 3326 <sup>‡</sup>	2	e' 30 °C	0.2228	4	Parachor	-10 [p]	16.02	2
25	1.3292	2	d g/ml	0. 228 4. 387	2 2	Parachor	20°C		
30	1.3252	5	v <sub>c</sub> ml/g t <sub>c</sub> °C	152.01	2		30 40		
"C"	0.7730	4	P <sub>c</sub> mm	28477.	2			190.2	4
MR (Obs.) MR (Calc.	1 20 772	2 5	PV/RT	<del> </del>		Exp. L.1.	6/wt.		
(nD-d/2)	1.0432	2	25°C 30 mm	0.9286	4 5	u. Dispersion			
Dielectric	1.776	5	BP	0.9582	4	Flash Poir		-60.	3
A -60 to		2	t <sub>e</sub> t <sub>c</sub>	0.9595 0.274	5 2	Fire Point		-00.	
B <u>45</u> °C	2 945.9 240.0	2 2	ΔHc kcal/m	635, 05	2	M Spec.			
A*   -60 to	1.11497	5	ΔHf	-35.29	2	Ultra V. X-Ray Dif.			
B* 30 °C	881.81	5	ΔFf	<del></del>	-	Infrared		Yes	2
С	ļ	ļ	Viscosity centistokes			Solubility i	n +		
tk to		1	7 -40 °C	0.491	2	Acetone Carbon te	t.		
t <sub>x</sub> ; °C		<u> </u>	-30 -20	0.446 0.407	2	Benzene			
B' •0			-10	0.375	2	Ether n-Heptane	,	!	
C'		<u> </u>	B <sup>V</sup>   -40 to A <sup>V</sup>   10 °C	234.16	4	Ethanol			
A'* to B'* *C				Z. 68698		Water Water in			
Acl 45 to		4	ll 4. v.	227.42	4	Viscosity			
Bc t °C	1299.	4	(A')  -40 °C c <sub>p</sub> liq. °K	₹.71750	4	centistoke	90°C	0.01	,
Cc	289.1	4	<b>41</b> -		_		90°C 50	0.91 0.545	2
Cryos. A° consts. B°		2 2	c <sub>p</sub> vap.300°K 400	0. 40261 0. 50929	2		0	0.350	2
t <sub>e</sub> °C	-1.75	5	c <sub>v</sub> vap.	1					
	_1	•	$T_R = 0.75 T_c$	<u> </u>	L	+ grams/10	0 grar	ns solvent	<u> </u>
			PI 3-Lit. 4-0	Calc. from det	. da	ta 5-Calc.	by for	mula	
SOURCE:		AP							
PURIFICA'	TION:	AP	ľ						
LITERATU	RE REFERE	NCES	: 3 NFPA 325						

								No. 5	
NAME	2-Methylp	ropar	ne			ST	RUCTURAL	FORMUL	A
	Isobutane						сн <sub>3</sub> сн-	CH <sub>2</sub>	
Mole % Pur.	Ref. Mo	ecul:		Molecular Weight 58,12	0		сн <sub>3</sub>	,	
	<b>.</b>	Ref.			Ref.				Ref.
F.P. °C	-159.600	2	dt/dP			f	to		
F.P. 100% B.P. °C		$\vdash$	°C/mm 25°C	0.0132	5	g	' <u>°K</u>		
760 mm	-11.730	2	BP	0.0337 0.0361	2 5	h f'			_
100 30	-54.07 -72.52	2 4	t <sub>e</sub> 30 mm	0.4600	4	g'	to K		
10	-86.4	4	ΔHm cal/g	<del> </del>	2	h'			
1	-109.2	5	ΔHr cal/g	18.668	-	m	300 to	-0.0058	4
Pressure mm 25°C	2611.	4	25°C	78.63	2	n o	l_6 <u>0</u> 0_ <b>•</b> K	0.0015	4
t <sub>e</sub>	697.	5	30 mm BP	99. 79 87. 56	5 2			-0.0 <sub>6</sub> 58	*
Density	#		t_	87.99	5	m' n'	700 to	0.0920 0.0013	4
g/ml 20°C	0.5572 <sup>≠</sup> 0.5510 <sup>≠</sup>	2 2	le (a, e)	88.00	5	0'	1000 _1	-0.0641	4
d <sub>4</sub> 30	0.5450	4	ΔHv/T <sub>e</sub>	19.72	5	Sur	face tension		
a	0.5846	4 4	d -75 to e -10 °C	85.20 0.2011	5	dyn	es/cm40°C	17.68	2
b	-0.0398	*	d' -10 to	84.71	4	8	-30 -20	16.48 15.28	2 2
Ref. Index n <sub>D</sub> 20°C	1.3169	5	e'   25 °C	0.2431	4	Par	achor [P]	15.20	-
45			d g/ml	0.221 4.525	2 2		20°C		
30 "C"	0.7/75	┝╤┤	vc ml/g tc °C	134.98	2		30 40		
MR (Obs.)	0.7675	5	P <sub>c</sub> mm	27360.	2			190, 2	5
MR (Calc.)	20.772	5	PV/RT	0.0003	,	Exp	. L.1.%/wt.		
(nD-d/2)			25°C 30 mm	0.9083 1.0000	5	Dis	u. persion		
Dielectric	1.734	5	BP	0.9648	4 5		sh Point C	-87.	5
A -75 to B _30 °C	6.74808 882.80	2 2	t e t	0.9668 0.283	2	Fir	e Point		
c '	240.00	2	∆Hc kcal/m	633.05	2		Spec. ra V.		
A* -75 to	1.04221	5	ΔHf ΔFf	-37.87	2	X-F	lay Dif.		
B* _ 0 °C	820.37	5	Viscosity				ared	Yes	2
°			centistokes				ibility in <sup>†</sup> etone		
t <sub>k</sub> to c			η -50 °C -40	0.619 0.549	2 2	Ca	rbon tet.		
A'   to			-30	0.491	2		nzene he <i>r</i>		
B'°C_			-20 B <sub>v</sub> -90 to	0,443	2	n-	Heptane	•	
A'* to		$\vdash$	B' -90 to A' -55 °C	265.10 2.6035	4		hanol iter		
B'* °C			(B <sup>V</sup> )  -55 to	273.73	4	Wa	ter in		
Acl 30 to	7.42067	4	(A <sup>V</sup> )  0 °C	2.5653	4	Vis	cosity		
Bc t <sub>c</sub> °C	1288.1 296.7	4 4	c <sub>p</sub> liq. °K			1	tistokes -80°C	0.946	2
Cryos. A°	0.04234	2	_	0.40003	2	7	-60	0.703	2
consts. B°	0.0057	2	c <sub>p</sub> vap 300°K 400	0.51222	2		-10	0.403	2
t <sub>e</sub> °C	-13.90	5	c <sub>v</sub> vap.	L					
≠ at satura	tion pressure		$T_R = 0.75 T_C$				ams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A		Calc. from de	t. da	<b>ta</b> 5	-Calc. by for	mula	
SOURCE:			PI						
PURIFICAT			PI						
LITERATUI	RE REFERE	NCES	<b>5:</b>						

	·						No. 6	
NAME	n-Pentane	:				STRUCTURAL	FORMULA	L
Γ								
			<del></del>		$\dashv$	CH <sub>3</sub> (CH <sub>2</sub> )	3 <sup>CH</sup> 3	
Mole		lecul		Molecular	,			
% Pur. 99	.86   2   Fo	rmul	-5-12	Weight 72.14	=			
		Ref.	ļ	·	Ref			Ref.
F.P. C F.P. 100%	-129.721	2	dt/dP *C/mm		1 1	f to		
B. P. *C	<del>`</del>	-	25°C	0: 05257	4	g <u>*K</u>	1	1
760 mm	36.074	2	BP	0.03856 0.03631	2 5	h	<del> </del>	├
100 30	-12.59 -33.93	2	<b>t</b> e		4	f' to		[
10	-50.1	4	30 mm	0.5334	2	h'	1	
1	-76.63	5	ΔHm cal/g	27.805	۲	m   300 to	0,0310	4
Pressure			∆Hv cal/g 25°C	87.54	2	n 600 °K	0.0014	4
mm 25°C	512.5 825.9	4 5	30 mm	98.5 <b>3</b>	5	° ;	-0.0642	4
Density	+	۰	BP	85.38 84.92	2 5	m'   700 to	0.0976	4
g/ml 20°C		2	te (d, e)	84.92	5	n' 11000 •K		
dt 25 4 30	0.62139 0.61649	2	AHV/Te	19.65	5	8.	-0.0641	*
30	0.64604	4	d   -35 to	92.16	5	Surface tension dynes/cm, 20°C	16.00	2
ь	-0.03904	4	<u>-å,-</u> ,-, <u>-</u> ⁴0_ <b>:</b> €		5	dynes/cm. 20°C	14.95	2
Ref. Index			d'   to			40	13.8*	2
<sup>n</sup> D 20°C		2	d g/m1	0, 232	2	Parachor [P]		
30	1.35472 1.35194	2	IV ml/g	4.311 196.62	2	20°C 30	231.0 231.0	4
"C"	0.7664	4	, C		2	40	230.5	4
MR (Obs.)	+	2	P <sub>c</sub> mm	25316.	2		229. 2	5
MR (Calc.		5	PV/RT 25°C	0,9662	5	Exp. L.1.%/wt.	3.6 18.6	3
(nD-d/2)	1.04436	2	30 mm	1.0000	5	Dispersion	98.0	2
Dielectric	1.843	5	BP t <sub>e</sub>	0.9547 0.9523	4 5	Flash Point °C	-40.0	3
A -35 to B 80 °C		2 2	tc	0.268	2	Fire Point	ļ	
C	232.0	2	ΔHc kcal/m	782.04	2	M Spec. Ultra V.		
A* -35 to		5	ΔHf ΔFf	-41.36 -2,25	2 2	X-Ray Dif.		
B* ∟60 °C	995.37	5 4	Viscosity	-2.23	۲	Infrared	727.	1
c	-0.16075	4	centistokes	}		Solubility in + Acetone		1
t <sub>k</sub>   60 to t <sub>x</sub>   160 °C	185. 243.6	4 5	7 0 °C	0.432	2 2	Carbon tet.	<b>80</b>	
A' to		-	10 20	0.401 0.375	2	Benzene	∞	l
B' _ 'C			30	0.351	2	Ether n-Heptane	<b>∞</b>	1
<u> </u>	1		B <sup>V</sup>   -10 to A <sup>V</sup>   40 °C	251.11	4	Ethanol	oc	1
A'* to B'* *C				2,71711	4	Water Water in		
Acl 80 to	<del></del>	5	1	261.39	4	Viscosity	1	
Bc tc	1411.3	5	(A <sup>V</sup> )  -10 °C	₹. 67863	4	centistokes		
Co	279.1	5	c <sub>p</sub> liq. °K			ካ -60°C	0.803 0.514	2 2
Cryos. A.	0.04906	2	c <sub>p</sub> vap.300°K	0.40016	2	-10	0.469	2
consts, B°	0.0042	2	c <sub>v</sub> vap.	0.5 <b>0</b> 633	2			
te °C	38, 54	5	•	L	لـــا	L	L	L
TR = 0.7			at saturation	pressure		grams/100 gra	ms solvent	<u> </u>
SOURCE:	J=3: 1-DO₩		PI 3-Lat, 4-0	aic. Irom det	. dai	ta 5-Calc. by for	mula	
PURIFICAT	PION:		<del></del>					
			PI : 3 NFPA 325	<del></del>				
TILERAIU	ne refekel	NUES	: 3 NFPA 325	)				

Isopenta  Ref. 2	Molecul			-	ST	RUCTURAL	FORMUL	A
99 Ref. 2	Molecul							
1			Molecular			СН <sub>3</sub> СНСН СН <sub>3</sub>	I <sub>2</sub> CH <sub>3</sub>	
-159, 900	Formula		Weight 72.140	-	<del></del>			la .
1 -137.700	Ref.		<del>                                     </del>	Ref.			1	Re
<b>†</b>	<u> </u>	dt/dP °C/mm			f g	to *K		
		25°C	0.0412	4	h			
27.852		BP t	0.03815	2 5		*0		$\vdash$
-41.10		11 -	1	1 1	g'	! <u>*K</u>		l
-56.9	4		<del>                                     </del>		h'			
-82.8	-   5		1	┢┋┪	m	300 to	0.0043	
688.1	4	25°C	81.47	2		_6 <u>0</u> 0_•K		4
793.8	5						-0.0631	-
1		t_	80.73	5	m'	700 to	0.1052	
		te (d, e)	80.72	5	o'	1 (1000 -K		
		ΔHv/T <sub>e</sub>	19.27	5	<u> </u>		3.3637	H
0,640	050 4	d -40 to	86.45	5			15.00	2
-0.039	2 4		0.1968		8,	30	13.93#	2
		e' °C					12.73	5
		d g/ml	0.234	2	Par		220.0	4
		V mi/g			ł	30	229.9	4
0.766	51 4	1 _	1	1 1		40 Sugard	220 2	5
			23004.		E		227.2	1-
		25°C	0.9478	5	Exp	u.		
		30 mm	1.0000		Dis	persion	98.6	2
<del></del>		t		5			-60.	5
		t <sub>c</sub>	0.269	2				⊬
233.097		ΔHc kcal/m	780.12	2				
		ΔFf			X-I	Ray Dif.		
. 355.73	"	Viscosity	1				Yes	2
.		centistokes						
		1 (			Ca	rbon tet.	∞ ∞	
<del> </del>		10	0.396	2			00	
.	ŀ	20		$\overline{}$			oc .	
ļ		B' -10 to					oc	
	į	<u> </u>	-					
7 256	32 5	11			Vic	cosity		T
1325.2	5	<del></del>	2.55042	+*-		tistokes		
275.8	-	P			η	-50°C	0.790	2
		c <sub>p</sub> vap.300°K	0.39559	2		-20 30	0.535	2 2
+		<del>- 4</del> 00	0.50578	2				1
		ll	D Dress	لــــــــــــــــــــــــــــــــــــــ	L	/100	<u> </u>	<u> </u>
				- د. ه				ıt
, EG; 1-DC			Caic, irom de	t. da	ua 5	-Caic. by for	muia	
'ION:								
			- E14					
ne Kefe	RENCE	o: 3 NBS Circ	2. 514					
	-20.14 -41.10 -56.9 -82.8  688.1 793.8  0.619 0.604 -0.035  1.355 1.347 0.766 25.299 1.041 1.843 6.789 1020.012 233.095 1.152 955.73	-20.14	-20.14	Color	te 0.0369 5 -20.14 2 -41.10 4 -56.9 4 -82.8 5  688.1 4 793.8 5  0.61967 2 0.61462 2 0.60951 4  0.64050 4 -0.0392 4  1.35373 2 1.35088 2 1.34762 4 0.7661 4 25.292 2 0.25.290 5 1.04390 2 1.843  3 6.78967 2 1.843  3 6.78967 2 1.15247 5 955.73 5  1.15247 5 955.73 5  1.25632 5 1325.2	Ton   Continue   Con	1.   1.   1.   1.   1.   1.   1.   1.	Column

								No. 8	
NAME	2, 2-Dime	thylp	ropane		l	STR	UCTURAL 1	FORMULA	
	Neopentan	e					ÇH <sub>3</sub>		
<del></del>	TT		T				CH <sub>3</sub> C CI	H <sub>3</sub>	
Mole % Pur.	Ref. Mo	lecul rmul	ar C <sub>5</sub> H <sub>12</sub>	Molecular Weight 72.14	6		ĊH <sub>3</sub>	1	
<del>- 7 - 11 - 1</del>	1 1 10	Ref.		Weight	Ref	<del></del>			Ref.
F. P. *C	-16.550	2	dt/dP			f	to		
F.P. 100%			°C/mm			g	<u>*K</u>		
B. P. *C			25°C BP	0.0244 0.03652	4 4	h			
760 mm 100	9.503 -36.32	2 2	t <sub>e</sub>	0.0370	5	f'	to		
30	-56.27	4	30 mm	0.4973	4	g'	K_		
10 1	-71.3 -95.9	5	ΔHm cal/g	10.786	2	h'		2 21 52	
Pressure	1		ΔHv cal/g 25°C	72.15	١, ١	m n	300 to 600 °K	-0.0159 0.0016	4
mm 25°C	1285.2 750.8	4 5	30 mm	72.15 86.88	2 5	•		-0.0 <sub>6</sub> 62	4
Density	<b>+</b>	-	BP	75.37 75.43	<b>2</b> 5	m'	700 to	0.1199	4
g/ml 20°C	0.5910	2	te (d, e)	75.43	5	n' o'	1 <u>1000</u> •K_	0.0012	4
dt 25 4 30	0.5910 <sub>4</sub> 0.5851 <sup>‡</sup> 0.5792	2	AHv/Te	19.27	5			-0.0642	*
a	0,61601	4	d   -60 to	77.03	5		s/cm. 20°C	1 <b>2.0</b> 5	5
Ъ	-0.00102	4	<u>ૄ</u> યું¦ ⊸ દુઃ	0.1750	5	3,	30	10.98	5
Ref. Index	₹	2	e' •C	<u> </u>			40	9.94	5
25	1.339	2	d g/ml vc ml/g	0.238 4.200	2 2	Para	chor [P]		
30	1.335	4	tc C	160.60	2		30 40		
"C"	0.7779	4	P <sub>c</sub> mm	2399 <b>3.</b>	2			229.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT		_	Exp.	L. 1. %/wt.		
(nD-d/2)	25.290 <sub>4</sub> 1.046 <sup>‡</sup>	2	25°C 30 mm	0.9 <b>32</b> 1 1.0 <b>0</b> 00	5	Dian	u. ersion	98.≠	2
Dielectric	1,801	5	BP	0.9554 0.9557	4 5		h Point °C	-75.	5
A -60 to B _55 °C		2 2	te t <sub>c</sub>	0.269	2	Fire	Point		Ĺ
c C	237.0	2	AHc kcal/m	777.37	2	M Sp Ultra		Yes	2
A* -60 to		5	ΔHf ΔFf (gas)	-44.98 -3.66	2 2	X-R	y Dif.		
B* ∟ <sup>20</sup> °C	886. 37	5	Viscosity	3.00	H	Infra		Yes	2
c	_{		centistokes	0 (2)			oility in +	<b>so</b>	
t <sub>k</sub> to			7 -10 °C	0.626 0.576	2		bon tet.	oc	
A'   to			0 5	0.534 0.498	2 2	Eth	zene er	<b>80</b>	
B', ∟ _ º	<u>-</u>		BV   10 to	484.9	4		leptane anol	90 90	
A'* to			A'   5 °C	3.95415	4	Wat	er		
B'* °C			(B <sup>V</sup> ) to			Wat	er in		<u> </u>
Ac   55 to		5 5	(A <sup>V</sup> )  °C						
Bc Ltc_°C	280.8	5	c <sub>p</sub> liq. ∘K						
Cryos. A	0.00595	2	cp vap.300°K	0.40487	2				
consts. B°	+	_	c <sub>v</sub> vap.	0.52047	2				
t <sub>e</sub> °C	9. 167	5			L	L		l	<u> </u>
T <sub>R</sub> = 0.7		2 AT	# at saturation	pressure		gra	ms/100 gran	ns solvent	<u> </u>
SOURCE:	I-DOW		PI 3-Lit. 4-C	aic. Irom det	. da	ta 5-	Calc. by for	mula	
PURIFICAT	TION:		PI	<del></del>					
	RE REFERE								
			•						
L									

								No. 9	
NAME	n-He	xa ne				ST	RUCTURAL	. FORMUL	A
		,					CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub>	СН	
Mole	Ref.	Molecula		Molecular			3, 3, 2, 4	3	
% Pur. 99.	99   2	Formula	6.14	Weight 86.17		ı .			Ref
F, P, *C	-95, 348	Ref.	14/17		Ref.			7	Kei.
F.P. 100%			dt/dP °C/mm		١	f g	to °1		
B. P. °C	/0.74		25°C BP	0.1524 0.04191	2	h			
760 mm 100	68.740 15.81	2 2	t	0.0363	5	ſ'	t	5	
30	-7.44	4 4	30 mm	0.5815	4	g'	•1	K	
10 1	-25,1 -54.0	5	∆Hm cal/g	36.137	2	h'			ļ
Pressure			ΔHv cal/g 25°C	87.50	2	m n	300 to   600 °1		
mm 25°C	151.3 914.5	4 5	30 mm	93.37	5	•	l 1	-0.0643	
t <sub>e</sub> Density	714.3		BP	80.03 79.01	2 5	m'	700 to		
g/ml 20°C	0.659	37 2	t <sub>e</sub> (d, e)	78.98	5	n'	1000 •1		
dt 25 d4 30	0,654		ΔHv/T <sub>e</sub>	19.57	5	0'		-0.0640	1
A 30	0,677		d -10 to	92.06	5		face tension		2
b	-0.038		e 75 °C to	0.1751	5	الا الا	es/cm. 20°C 30	17.38	2
Ref. Index		10/ 2	• · · · · · · · · · · · · · · · · · · ·			ļ	40	16.36	2
<sup>n</sup> D 20°C	1.374		d <sub>c</sub> g/ml	0.234	2	Par	achor [P] 20°C	271.0	4
30	1.369	38 4	l v ml/g	4.271 234.7	2		30	270.9	4
"C"	0.760	7 4	t <sub>c</sub> *C	22739.	2		40 Suga	270.9 1. 268.2	5
MR (Obs.)	29.907		PV/RT	<del> </del>	<del>                                     </del>	Ext	L. 1. %/wt		31
MR (Calc. (nD-d/2)	29.908 1.045		25°C	0.9847	<b>5</b>	1 -	u.	18.0	31
Dielectric	1.890	) 3	30 mm BP	1.0000 0.9503	4	J	persion	98.0	2
A -10 to	6.877	76 2	te	0.9447 0.264	5 2		sh Point °C e Point	-26.0	3'
B (110 °C	1171.53 224.366	2 2	tc ΔHc kcal/m	928, 93	2	М.	Spec.	Yes	1
A* -10 to	1,257		ΔHf	-47.52	2		ra V. Ray Dif.		
B*[100 °C		5	ΔFf	-1.03	2		ared	60.	1
K	-0.131		Viscosity centistokes		l		ubility in +		
tk 100 to	117.	4 5	ິ <b>ກ</b> 30 °C	0.4389	2		etone rbon tet.	90	
195 °C	285.6	-   3	· 40 50	0.4085 0.3817	2 2	Be	nzene	•	
B' C			60	0.3577	2		her Heptane	<b>80</b>	
C'			B <sup>V</sup>   15 to A <sup>V</sup>   60 °C	297.4 2.66156	4	Et	hanol	•	
A'* to B'* °C	ļ		(BV)1-30 to	309.4	4		ater ater in		
Ac 110 to	7,319	38 4	(A <sup>V</sup> )  15 °C	2.61885	4	Visc	osity	T	
Bc tc C	1483.1	4	c <sub>p</sub> liq. °K	1 2.51535	١ <del>٠</del>	11 .	tistokes -20°C	0 401	,
Ce— —	265.9	4			1.	א	20	0.691 0.4741	2
Cryos, A° consts, B°	0.049		c <sub>p</sub> vap.300K 400	0.39885 0.50446	2 2		70	0.336	2
t <sub>e</sub> °C	74.76	5	c vap.	0.30440	-				
$T_{R} = 0.75$			1		_	" + g	rams/100 gr	ams solver	ıt.
REFEREN		2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		API							
PURIFICA'	TION:	API							
LITERATU	RE REFE	RENCES	5: 3 NBS 514;	3' NFPA 325					

								No. 10	
NAME	2-Methylp	entar	ne			STR	UCTURAL 1	FORMULA	
	Isohexane						сн <sub>3</sub> снсн <sub>2</sub> с	н.сн.	
Mole	2.6			Malandan			CH <sub>3</sub>	23	
		lecul rmul		Molecular Weight 86, l	72		3		
		Ref.	l .		Ref.				Ref.
F.P. *C	-153.670	2	dt/dP			f	to		
F.P. 100	6		°C/mm	0 1142		g	<u>*</u> K		
B. P. °C 760 mm	(0.37)		25°C BP	0.1143 0.04141	2	h			
100	60.271 8.06	2 2	-t	0.0366	5	f'	to		
30 10	-14.82	4	30 mm	0.5717	4	g'	K_		
1	-32.1 -60.55	5	∆Hm cal/g	17.407	2	h'	1 200 4	0.0150	-
Pressure			ΔHv cal/g	03.03	١, ١	m n	300 to	0.0150 0.0014	4
mm 25°C		4	25°C 30 mm	82.83 89.76	2 5	0		-0.0650	4
t <sub>e</sub> Density	891.5	5	BP	77.09	2	m'	700 to	0.1533	4
g/ml 20°	0.65315	2	te te (d, e)	76. 25 76. 23	5	n'	11000 <b>•</b> K	<b>0</b> .0011	4
dt 25	0.6485€	2	ΔHv/T <sub>e</sub>	19.41	5	0'		-0.0635	4
	0.64386		d   -15 to	87.26	5		ace tension		
a b	0.67177 -0.0 <sub>3</sub> 890	4	_e _i _65•C	0.1688	5	dyne	s/cm. 20°C 30	17.38 16.37	2 2
Ref. Index			d' to				40	15.36	2
n <sub>D</sub> 20°		2	d <sub>c</sub> g/ml	0.235	2	Para	chor [P]		
25 30	1.36873	2 4	l v mi/g	4.259	2	1	20°C 30	269.7 269.7	4
"C"	0,7613	4	t <sub>c</sub> ·C	224.9	2		40	269.6	4
MR (Obs.	<del></del>	2	P <sub>c</sub> mm	22762.	2	<b></b>		268.2	5
MR (Calc.	29.908	5	PV/RT 25°C	0.9793	5	Exp.	L. 1. %/wt. u.		
(nD-d/2)	1.04488	2	30 mm	1.000 <b>0</b>	5	Disp	ersion	98.6	2
Dielectric		5	BP t <sub>e</sub>	0.9512 0.9463	4 5		h Point °C	-14.	5
A -15 t B 100 °		2 2	tc	0.269	2	ļ	Point		
c	226.572	2	ΔHc kcal/m	927.23	2	M Sp Ultra			
A*  -15 to	1.22708	5	ΔHf ΔFf	-48.82 -1.97	2	X-R	ay Dif.		
B* 175 °C	<u>C</u> 1063.37	5	Viscosity		<del>-</del> -	Infra		Yes	2
°			centistokes				oility in +	<b></b>	
t <sub>k</sub>   t			7 0 ℃	0.5525 0.4746	3	Car	bon tet.	œ	
A' t	<del>,  </del>	-	30	0.4349	3	Ber Eth	zene	00 00	
B' *	<u>s</u>		B <sup>V</sup>   -10 to	<del> </del>	-	n-H	leptane	<b>x</b>	
C' t	_	-	B' -10 to A'   40 °C	286.99 T.69184	4	Eth. Wat	anol	oc	
A'* t			(BV) to	1	ا		er in		
Ac   100 to	7.31635	4	(A <sup>V</sup> ) °C	1					
Bc tc_	C 1468.9 271.5	4	c <sub>p</sub> liq. °K	<b>†</b>	<del>                                     </del>	1			
Cryos. A		2	(I -	0.40:07	١, ١				
consts. B		2	c <sub>p</sub> vap.300°K 400	0.40187 0.51061	2				
t <sub>e</sub> °C	65.38	5	c <sub>v</sub> vap. 388.56°K	0.366	3	<u> </u>			
$T_R = 0$						T gra	ms/100 gran	ns solvent	t
	CES: 1-Dow			Calc. from det	da:	ta 5-	Calc. by for	mula	
SOURCE:	TION		PI PI						
PURIFICA									
LITERAT	KE REFERE	NCES	3 ASTM 109	•					

No. 11 3-Methylpentane NAME STRUCTURAL FORMULA CH3CH2CH CH2CH3 ĊH<sub>3</sub> Ref. Molecular Molecular Mole  $C_{6}H_{14}$ % Pur. 99.8 Weight 86.172 3 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.1262 B.P. °C h ВP 0.04182 2 760 mm 63, 282 2 t<sub>e</sub> 0.0366 5 ſ١ to 100 10.53 2 g' <u>°К</u> 30 -12.61 4 30 mm 0.5781 4 10 -30.1 4 h' ∆Hm cal/g -58.87 5 300 to 0.0804 4 m AHv cal/g Pressure n \_6<u>0</u>0\_°K 0.0014 4 25°C 83.96 mm 25°C 189.8 4 o 0.0643 4 30 mm 90.29 5 te 902.1 5 RP 77.88 2 0.0648 m' 1 700 to Density 4 76.99 5 te (d, e) n' 0.0013 1000 °K g/ml 20°C 4 0.66431 2 76.97 5 ٥' -0.0<sub>6</sub>46 4 25 0.65976  $\mathbf{d_{4}^{t}}$ 2 AHv/Te 5 19.40 30 0.65519 4 Surface tension -15 to 88.23 5 4 0.68259 dynes/cm. 20°C 18.12 2 <u>70</u> 0.1636 5 ь -0.0<sub>3</sub>88 4 ă۳ 17.08 to 40 16.03 2 e' Ref. Index 20°C <sup>n</sup>D 1.37652 2 [P] Parachor d<sub>c</sub> g/ml 0.235 2 1.37386 2 25 20°C 267.9 4 vc ml/g tc °C 4.259 2 4 30 1.37106 30 4 267.8 t<sub>c</sub> 231.2 2 40 267.6 4 "C" 0.7582 4 P<sub>c</sub> mm 2 23431. Sugd 268.2 5 MR (Obs.) 29.802 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 29.908 5 25°C 0.9822 5 (nD-d/2)1.04436 2 30 mm 5 1.0000 Dispersion 97.1 2 Dielectric 1.895 5 BP 0.9531 4 Flash Point °C -35. 5 A -15 to te tc 0.9480 5 6.84887 2 Fire Point 0.273 2 B (105 °C 1152.368 2 M. Spec. C 227, 129 2 AHc kcal/m 927.87 2 Ultra V. ΔHf -48.28 2 A\* -15 to 1.22845 5 X-Ray Dif. ΔFf Z 1078.38 -1.34 B\*| 80 °C Infrared 2 Yes ĸ Viscosity Solubility in centistokes Acetone œ to 0 °C  $t_k$ 0.5774 31 Carbon tet. °C œ  $t_{\mathbf{x}}$ 15 0.5068 31 Benzene œ 25 31 0.4653 A'ı to Ether œ °C 40 0.4156B' n-Heptane œ B<sup>V</sup> A ċ١ -10 to 305.47 4 Ethanol 00 50 °C **Z.** 64336 4 Water A1\* to B'\* Water in °C (B<sup>V</sup>) to Acl 105 to (A<sup>V</sup>) 7.36387 4 °C Bc tc °C 1520.0 cp liq. ۰ĸ Cc 276.46 4 c<sub>p</sub> vap.300°K Cryos. Aº 0.39885 2 consts. Bo 400 0.50446 2 te °C c, vap. 68.84 5  $T_{\mathbf{R}} = 0.75 \, \mathbf{T}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 ACS 74, 1951 (1952); 3' Timmermans

							No. 12	
NAME	2, 2-Dimetl	ylbu	tane		STRUCTURAL :	FORMULA		
	Neohexane				CH <sub>3</sub>			
						сн <sub>3</sub> с сн <sub>2</sub>	CH <sub>3</sub>	
Mole % Pur. 99.		lecul rmul	ar C <sub>6</sub> H <sub>14</sub>	Molecular Weight 86.17	2	CH <sub>3</sub>		
		Ref.			Ref.			Ref.
F.P. *C	-99.870	2	dt/dP			f to		
F.P. 100%			*C/mm 25*C	0.0814		g LK_		
B. P. *C 760 mm	49.741	2	BP	0.0814 0.04117	2	h		L_
100	-2.0	2	t <sub>e</sub>	0.0372	5	f' to		l
30 10	-24.49 -41.5	4	30 mm	0.5619	4	g'   ' <u>*</u> K_	1	
ĩ	-69.3	5	ΔHm cal/g	1.607	2	1 222	0.0046	4
Pressure	T		ΔHv cal/g 25°C	76, 79	2	m 300 to	0.0014	4
mm 25°C	319.1 864.9	4 5	30 mm	84.62	5	•	-0.0 <sub>6</sub> 47	4
Density	+	-	BP +	72.96 72.32	2 5	m'   700 to	-0.0210	
g/ml 20°C		2	te te (d, e)	72.32	5	n' 11000 °K	0.0016	
dt 25	0.64446 0.63972	2	AHV/T	19.06	5		-0.0 <sub>6</sub> 64	
	0,66820	4	d   -25 to		5	Surface tension dynes/cm, 20°C	16.30	2
ь	-0.03889	4	<b>├</b> ╬,┤ <i>─</i> ⁰─ <b>;</b> ;		5	) y 30	15.31	2
Ref. Index			e' 'C			40	14.33	2
<sup>n</sup> D 20°C	1.36876	2 2	d <sub>c</sub> g/ml	0.240	2	Parachor [P] 20°C	267.2	4
30	1.36312	4	t <sub>c</sub> *C	4.166	2 2	30	267.2	4
"C"	0.7608	4	P <sub>c</sub> mm	23309.	2	40 Sugd.	267.1 268.2	4 5
MR (Obs.)		2	PV/RT	23307.	<u> </u>	Exp. L. 1. %/wt.	200.2	<u> </u>
MR (Calc. (nD-d/2)	29.908 1.04418	5 2	25°C	0.9754	5	u.		
Dielectric	1.873	5	30 mm BP	1.0000 0.9544	5	Dispersion	99.8	2
A -25 to	<del></del>	2	te	0.9505	5	Flash Point °C Fire Point	-47.	5
B L_95_• <u>°</u>		2	t <sub>c</sub>	0.274	2	M Spec.		
A*   -25 to	1,14940	5	ΔHc kcal/m ΔHf	924.53 -51.00	2 2	Ultra V.		
B* _ 65 °C		5	ΔFí	-2.90	2	X-Ray Dif. Infrared	Yes	2
K	1		Viscosity centistokes			Solubility in +		_
to to			7 0 °C	0.7144	3	Acetone Carbon tet.	60	
×			20	0.5777	3	Bensene	<b>80</b>	ĺ
A' to			25 30	0.5446 0.5158	3	Ether n-Heptane	€0	İ
c,			B <sup>v</sup>   -10 to	39 <u>0</u> .6	4	Ethanol	<b>80</b>	
A'* to			AV   40 °C	7.42425	4	Water Water in	ļ	
B'* °C	<del></del>	_	(B <sup>V</sup> ) to	}			t	$\vdash$
Bc t °C	1494.7	4	(A <sup>V</sup> )  °C		-		1	
Ce	286.2	4	c <sub>p</sub> liq. ∘K				]	
Cryos. A° consts. B°	0.002321 0.000	2 2	c <sub>p</sub> vap 300 °K 400	0.39560 0.50712	2 2			
t <sub>e</sub> °C	53.84	5	c <sub>v</sub> vap.				l	
$T_R = 0.7$						† grams/100 grai		
	ES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by for	mula	
SOURCE:	TON.	AP	<del></del>					
PURIFICAT		AP				<del></del>		
-IIERAIU	RE REFEREI	NC ES	3 Timmerm	ans				

No. 13 2, 3-Dimethylbutane NAME STRUCTURAL FORMULA сн3сн сн сн3 ċн,ċн, Molecular Ref. Molecular C6H14 % Pur. 99.99 Formula Weight 86.172 Ref Ref. Ref. °C -128.538 2 dt/dP f to F.P. 100% °C/mm g °K 25°C 0.1059 B.P. °C h ВP 0.04173 2 760 mm 57.988 2 0.0370 5 ſ١ ŧ, 100 5.45 2 to g¹ -17.53 <u> °K</u> 30 4 30 mm 0.5738 4 10 - 34. 9 4 h' 2.251 ∆Hm cal/g 2 -63.37 5 300 to -0,0058 m AHv cal/g Pressure 600 °K 0.0015 n 4 25°C 80.77 mm 25°C 234.6 o -0.0<sub>6</sub>54 4 30 mm 87.57 5 887.5 5 t<sub>e</sub> BP 75.65 2 m' 700 to 0.0686 4 Density 74.87 te (d, e) 5 5 n' g/ml 20°C 1000 •K 0.0013 -0.0<sub>6</sub>46 0.66164 44 2 74.86 ٥' 0.65702 2  $d_4^t$ 25 AHv/T 19.19 5 30 0.65237 4 Surface tension d -20 to 84.80 0.68026 5 4 dynes/cm. 20°C 17.37 2 2 °C 0.1578 \_65\_ ь -0.03885 4 30 ăח 16.37 to 40 15.37 2 Ref. Index e' <sup>n</sup>D 20°C 1.37495 2 Parachor [P] d<sub>c</sub> g/ml 0.241 2 25 1.37231 2 20°C 266.3 4 vc ml/g t\_°C 4.154 30 1.36939 4 30 t<sub>c</sub> 266.2 227.1 2 40 "C" 266.1 4 0.7583 5 P<sub>c</sub> mm 23552. 2 Sugd. 5 268. **2** MR (Obs.) 29.810 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 29.908 25°C 0.9800 (nD-d/2)2 u. 1.04413 30 mm 1.0000 5 Dispersion 98.3 2 Dielectric 1.890 5 BP 0.9535 4 Flash Point C -37. 5 0.9489 A -20 to 5 6.80983 2 Fire Point 0.270 2 1700 °C 1127.187 2 M. Spec. С ∆Hc kcal/m 228.900 2 926.40 2 Ultra V. A# -20 to ΔHf -49.48 1.19420 5 X-Ray Dif. ΔFf -1.69 B\* 75 °C 1053, 55 Infrared Yes 2 Viscosity Solubility in c centistokes Acetone to °C œ Carbon tet. t<sub>x</sub> °C œ Benzene œ A' to Ether B' °C B<sub>v</sub> | n-Heptane œ C' to Ethanol œ ۰c A'\* Water to Water in B'\* (BV) °C to Ac | 100 to 7.30917 (A<sup>V</sup>)| °C Bc\_tc\_°C 1481.8 c<sub>p</sub> liq. ۰ĸ 277.4 4 Cryos. A° consts. B° 0.00467 2 cp vap.300°K 0.39177 0.50248 2 c, vap. te °C 62.99 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

									No. 14	
NAME	n-I	Ieptar	ne			$\dashv$	STF	RUCTURAL	FORMULA	1
		_						CH <sub>3</sub> (CH <sub>2</sub>	) <sub>5</sub> CH <sub>2</sub>	
Mole % Pur. 99	. 94 Re:		olecul ormul		Molecular Weight 100 1	98		, <u>.</u>	•	
7			Ref.			Ref				Ref
F.P. °C	-90.	610	2	dt/dP	1		ſ	300 to	0,0373	5
F.P. 1007				°C/mm		١. ١	g	600°K		
B. P. *C				25°C BP	0.4376 0.04479	5	h	:	-0.0639	5
760 mm 100	98. 41.		2 2	t	0.0365	5	f'	to		
30	16.	80	4	30 mm	0.6237	4	g'		4	ļ
10 1	-2. -33.		2	ΔHm cal/g	33.474	2	h'	<u> </u>		L
Pressure	+		+*	ΔHv cal/g	1		m	300 to		
mm 25°C	45.	81	4	25°C	87.18	2	n	-600 •K	0.0014 -0.0 <sub>6</sub> 45	
t <sub>e</sub>	988.		5	30 mm BP	89.14 75.60	2		<u> </u>	<del></del>	┝-
Density				t_	73.96	5	m' n'	700 to	0.1062	
g/ml 20°0		68376 67951		t (d, e)	74.08	5	01	1 1 1 0 0 0	-0.0640	
d <sub>4</sub> 25		67525		ΔHv/T <sub>e</sub>	19. <b>4</b> 6	5	ļ	<u> </u>		-
	0.	70075	4	d   15 to		4		face tension es/cm. 20°C	19.29	5
b		0384	4	<u>-a</u> ,-¦ -¹º5- 🕃		4	3,	30	18.34	5
Ref. Index				e'   *C		] ]		40	17.41	5
<sup>n</sup> D 20°C		38764 38511		d <sub>c</sub> g/ml	0, 235	2	Par	achor [P]		١
30		<b>382</b> 50		ll v_ mi/g	4. 252	2		20°C 30	<b>'</b> }	ļ
"C"	0.	7572	4	16 °C	267. 01	2		40		l
MR (Obs.			2	P <sub>c</sub> mm	20528.	2			. 307.2	5
MR (Calc.			5	PV/RT 25°C	1.0000	5	Exp	. L.1.%/wt.	3.4	3'
(nD-d/2)	1.	04576	2	30 mm	1.0000	5	Dist	u. persion	18.0 97.7	3'
Dielectric	1.	924	3	BP	0.9401	4		h Point °C	-3.89	31
A   15 t		90240		t <sub>e</sub>	0.9326 0.260	5 2		Point	-3.07	-
B 1730.6	1268. 216.		2 2	t <sub>c</sub> AHc kcal/m	1075,85	2	M S	pec.		
A* 15 to		33143	+	AHf	-53.63	2		a V.	}	l
B* 155 °C	1194.		5	ΔFf	0.27	2		ay Dif. ared	593.	1
K '	24.		5	Viscosity	i		<b></b>	bility in +	+	t
t <sub>k</sub>   155 to		14584	4	centistokes 7 30 °C	0.5586	2	Ac	etone	<b>60</b>	
tx 220 °C			5	7 30 ℃ 50	0.4752	2		rbon tet.	<b>so</b>	
A'   to			$\dagger$	70	0.4118	2	Eth	nzene ner	<b>80</b>	1
B' '	2			B <sup>V</sup>   20 to	0.3624	2	n-F	Heptane	∞	1
	+		+	B' 20 to A' 100 °C	344. 88 7. 60962	4 4		nanol ter	<b>*</b>	İ
A'* to					-	1 1		ter in		L
Ac   130 to		3270	5	(A <sup>V</sup> ) <sub>1 20 °C</sub>	391.3 T 44704	4				
Bc tc	1581.	7	5		0.39780	5			1	1
Cc	257.		5	400	0.39780	5				
Cryos. Acconsts, B		05 <b>0</b> 65 0 <b>033</b>	2 2	c <sub>p</sub> vap.300°K	0.39781	2				1
t <sub>e</sub> °C	107.		5	c <sub>v</sub> vap.	0.50320	2				
$T_R = 0.7$			لــــــــــــــــــــــــــــــــــــــ	L	<u> </u>	ш	+ ~=-	ams/100 gra	ma aclus	<del>-</del>
		Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	det				<u> </u>
SOURCE:			AF					0_10.		
PURIFICA	TION:		AI							
		FERF		: 3 NBS Circ.	514: 31 NEE	ρ <sub>Δ</sub> 32	5			
	1.5.	·	., CE3	. J NES CIFC.	214; 2. NFF	A 36	,			

No. 15 2-Methylhexane NAME STRUCTURAL FORMULA CH3CHCH2(CH2)2CH3 Molecular C7H16 сн3 Mole Ref. Molecular % Pur. Weight 100.198 Formula Ref. Ref. F.P. °C F.P. 10 -118,276 2 dt/dP f to 100% °C/mm °<u>K</u> g 25°C 0.3189 4 B. P. \*C h BP 0.04431 2 760 mm 90.052 2 t<sub>e</sub> 0.0366 5 ſ١ 100 34.093 to g' <u>°K</u> 30 9.52 4 30 mm 0.6145 4 -9.09 10 4 h! ∆Hm cal/g 21.907 2 1 -39.7 5 300 to 0.0272 AHv cal/g Pressure n 600 °K 0.0014 25°C 83.02 2 mm 25°C 65.88 -0.0<sub>6</sub>45 o 4 30 mm 85.98 5 t<sub>e</sub> 971.1 5 ВP 73.14 2 m 700 to 0.1062 4 Density 71.85 5 te (d, e) 1000 °K 0.0012 4 g/ml 20°C 0.67859 5 2 71.79 ٥' -0.0640 4 25 0.67439  $\mathbf{d_{4}^{t}}$ AHv/T 19.37 5 30 0.67022 4 Surface tension 10 to 87,50 5 0.69538 a dynes/cm, 20°C 19.29 2 100 °C 0.1595 5 å⊢ -0.03825 Ъ 4 18.32 30 2 to 40 2 Ref. Index 17.36 e' <sup>n</sup>D 20°C 1.38485 [P] Parachor dc g/ml 0.234 2 25 1.38227 2 20°C 309.6 4 vc ml/g t °C 2 4.272 30 1.37979 4 30 309.5 4 t<sub>c</sub> 257.9 2 40 309.4 4 "C" 0.7576 4 P<sub>c</sub> mm 20672. 2 Sugd 307.2 5 MR (Obs.) 34.591 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.526 25°C 0.9897 5 (nD-d/2) 1.04556 2 30 mm 1.0000 Dispersion 98.5 2 Dielectric 1.919 3 BP 0.9467 4 Flash Point C -14. 5 0.9389 5 A 10 to B 125 °C 6.87318 1236.026 t<sub>e</sub> 2 Fire Point 2 0.267 M. Spec. С 219.545 2 AHc kcal/m 1074.14 2 Ultra V. 2 1.30078 ΔHf A\* 10 to -54.93 5 X-Ray Dif. ΔFf -0.68 2 B\*[110 °C 1161.12 Infrared Yes 2 ĸ Viscosity Solubility in centistokes Acetone to 20 °C 0.5570 31 Carbon tet. °C Benzene œ ٦ī to Ether an B١ ۰c n-Heptane œ B<sup>V</sup> A C' to Ethanol œ °C Water A'\* to (BV) Water in B'\* ۰c to Ac | 125 to 7.31001 (A<sup>V</sup>)| °C Bc tc °C 1555.4 c<sub>p</sub> liq 19. 38°C 0.5570 31 Cc 261.5 4 Cryos. A° consts. B° 0.04605 2 vap.300°K 0.39781 2 0.0036 2 0.50320 2 400 c<sub>v</sub> vap. te °C 98.53  $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES: 3 NBS 514; 3' Timmermans

								<b>No.</b> 16	
NAME	3-Meth	ylhexane	•			STR	UCTURAL 1	FORMULA	1
						,	יש כש כשוכי	ם / כם	
	1 1		Т		$\dashv$		H <sub>3</sub> CH <sub>2</sub> CH(C	<sup>1</sup> 2'2 <sup>C</sup> 13	
Mole	Ref.	Molecul		Molecular			CH <sub>3</sub>		
% Pur.		Formul		Weight 100.1	_				_
		Ref.		·	Ref		<del></del>		Ref
F.P. °C			dt/dP			f	to		
F.P. 100%			*C/mm 25*C	0. 3391	4	g	K		
B. P. °C 760 mm	91.85	0 2	BP	0. <b>044</b> 59	2	h	ļ <del></del>		L_
100	35.56	1 2	t <sub>e</sub>	0.0369	5	f'	to	1	
3 <b>0</b> 10	10.85	4	30 mm	0,6179	4	g'		ł	
10	-7.9 - <b>38</b> .6	2 5	ΔHm cal/g	_		h'	<u> </u>		<u> </u>
Pressure		-	ΔHv cal/g			m n	300 to	0.0272	4
mm 25°C	61.59	4	25°C 30 mm	83.68 86.33	2	"		0.0014 -0.0 <sub>6</sub> 45	
te	976.	5	BP	73.43	2			<u> </u>	-
Density			t.	72, 11	5	m'	700 to     1000 K	0.1062	4
g/ml 20°C	0.68 0.68		t (d, e)	72.04	5	0'	! ' <u>-</u> , -, -, -, -, -, -, -, -, -, -, -, -, -,	-0.0640	
dt 25 4 30	0.67		AHv/T	19.33	5	-		-	├
8	0.70		d   10 to		5		s/cm. 20°C	19.79	2
Ъ	-0.03		급,ᆜᅩᅆᇎ		5	3,	30	18.81	2
Ref. Index			• •				40	17.85	2
n <sub>D</sub> 20°C	1.38		d g/ml v ml/g	0.240	2	Para	chor [P]		١.
30	1.38			4.172	2		20°C 30	307.7 307.6	4
"C"	0.75		-	262.4	2		40	307.5	4
MR (Obs.)	34, 46		P <sub>c</sub> mm	21356.	2			307.2	5
MR (Calc.)	34.52		PV/RT 25°C	0, 9915	[ ] ا	Exp.	L.1.%/wt.	l	1
(nD-d/2)	1.04		30 mm	1,0000	5 5	Dian	u. ersion	97.4	2
Dielectric	20° 1.93	3	BP	0.9468	4		h Point °C	-14.	5
A 10 to	6.86		t <sub>e</sub>	0.9389 0.267	5 2		Point	**.	
B [_130 °C	1240.19 219.22		t <sub>c</sub> ΔHc kcal/m	1074.78	2	M S	ec.		
A*  10 to	1, 29		AHC KCal/m	-54.35	2	Ultr	a V.		
B* 110 °C		5	ΔFf	-0.39	2	Infra	ay Dif. red	Yes	2
ĸ	1		Viscosity				bility in +	<del> </del>	┢
£	ł		centistokes 7 °C			Ace	tone	œ	1
ty C	1		7 ·c		(		bon tet.	œ	1
A'   to		-	l	1		Eth	nzene er	00   00	
B' 'C	ł			<del> </del>	$\vdash \vdash$	n-H	leptane	<b>∞</b>	
C'	<b> </b>		B <sup>V</sup> to			Eth Wat	anol	• • • • • • • • • • • • • • • • • • •	
A'* to B!* °C	1			-			ter in		
Ac   130 to	7, 39	633 4		1					
Bc t tc C	1635.1	4							
Cc	270.8	4	c <sub>p</sub> liq. °K		[				
Cryos, A° consts, B°			c <sub>p</sub> vap.300°K	0.39781	2				
t <sub>e</sub> °C	100,58	5	c <sub>v</sub> vap.	0.50320	2				
$T_{R} = 0.75$			L	1		+ gra	ms/100 gran	ns solven	<u>.          </u> t
REFERENC		w 2-AI	PI 3-Lit. 4-(	Calc, from det	da1		Calc. by for		
SOURCE:		AI		·		<u> </u>			
PURIFICAT	ION:	AI	PI						
LITERATUE	E REFE		3: 3 NBS 514						

No. 17 3-Ethylpentane STRUCTURAL FORMULA NAME Triethylmethane CH3CH2CHCH2CH3 Molecular C7H16  $C_2H_5$ Mole Molecular Weight 100.198 % Pur. Ref. Ref F. P. ℃ -118.604 2 dt/dP f to F.P. 100% °C/mm 25°C °K g 0.3584 B. P. ℃ h BP 0.04482 2 760 mm 93, 475 2 0.0367 5 ſ١ t<sub>e</sub> to 100 36.864 2 g' °<u>K</u> 30 12.00 4 30 mm 0.6218 10 -6.8 4 h! ∆Hm cal/g 22,775 2 1 -37.8 5 m 300 to 0.0272 ∆Hv cal/g Pressure n 1 600 °K 0.0014 25°C 84,02 2 mm 25°C 58.05 o -0.0645 4 4 30 mm 86.47 983.0 5 t<sub>e</sub> ВP 73,83 2 m' 700 to 0.1062 4 Density 72.81 5 te te (d, e) 'n 0.0012 1000 °K g/ml 20°C 0.69816 5 2 72.43 o' -0.0640 4  $d_4^t$ 25 0.69395 2 AHv/Te 19.42 5 30 0.68982 4 Surface tension 10 to 88.34 5 0.71499 dynes/cm. 20°C 20.44 2 100 °C 0.1552 5 ь -0.03829 4 30 19.45 2 ٦ī to 18.47 40 2 Ref. Index e'  $^{n}D$ 20°C 1.39339 2 [P] Parachor d<sub>c</sub> g/ml 0.241 2 25 1.39084 2 305.3 20°C 4 vc ml/g t\_°C 4.152 2 30 1.38840 4 30 305.2 4 t<sub>c</sub> 267.6 2 40 305.2 4 "C" 0.7519 4 P<sub>c</sub> mm 21736. 2 Sugd. 307. 2 5 MR (Obs.) 34.283 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.526 5 25°C 0.9917 5 (nD-d/2)1.04431 2 30 mm 1.0000 5 Dispersion 95.7 2 Dielectric 1.939 3 BP 0.9485 4 Flash Point C -14.0 5 0.9404 5 A 10 to 6.87564 2 Fire Point 0.268 2 B 1130 °C 1251.827 2 M. Spec. C AHc kcal/m 2 219.887 2 1075.40 Ultra V. ΔHf -53.77 2 A\* 10 to 1.29505 5 X-Ray Dif. ΔFf 2 1.04 B\* 115 °C 1174.98 2 Infrared Yes ĸ Viscosity Solubility in centistokes Acetone to œ  $\mathbf{t_{k}^{t}}$ Carbon tet. œ °C Benzene 00 A'I to Ether œ В' °C n-Heptane B<sub>v</sub> | œ C' to Ethanol œ °C Water A'\* to Water in B'\* °C (B<sup>V</sup>)| to Acl 130 to 7.29098 4 (AV) °C Bc tc °C 1561.6 c<sub>p</sub> liq. °K Cc 261.0 4 cp vap.300°K Cryos. A° 0.04807 2 0.39781 2 consts. B° 0.0039 0.50320 400 c<sub>v</sub> vap. te °C 102.49  $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES: 3 NBS Circ. 514

								<b>No.</b> 18	
NAME	2, 2-Di	nethylpe	ntane		$\exists$	STRUCT	JRAL CH <sub>3</sub>	FORMULA	
L							CH <sub>2</sub> C	н,сн,	
Mole % Pur. 99	.99 Ref.	Molecul Formul		Molecular Weight 100.1	98		CH <sub>3</sub>	2 3	
		Ref.	<del></del>		Ref				Ref.
F. P. *C	-123, 811		dt/dP	Τ		f	•-		
F.P. 100%			*C/mm			g	to °K		
B. P. *C	1		25°C	0.2138	4	h			
760 mm	79.197		BP	0.04394 0.0372	5	f' + T	to		
100 30	23.858	2 4	t <sub>e</sub> 30 mm	0.6047	4	g'   1	°K		l
10	-18.6	4			2	h'			
1	-48,68	5	ΔHm cal/g	13.892	۲	m I	300 to	0,0272	4
Pressure	1,05 22		ΔHv cal/g 25°C	77.36	2	n	600°K	0.0014	4
mm 25°C	105.23 942.7	4 5	30 mm	81.39	5	•		-0.0645	4
Density	+		BP	69.55	2	m' I	700 to	0.1062	4
g/ml 20°C	0.673	85 2	te te (d, e)	68.44 68.45	5 5	n'   1_1	000 <b>°K</b>	0.0012	4
at 25	0.669	53 2	ΔHv/T	1	5	°'		-0.0640	4
4 30	0.665		d   0 to	19.06		Surface te	ension	<u> </u>	
a	0.691		90.00		5 5	dynes/cm	. 20°C	18.02	2
ь	-0.038	42 4	d' to	1		1	30 40	17.08	2 2
Ref. Index		15 2	e' j •c	ļ	$\sqcup$	Bana aban		10.14	-
<sup>n</sup> D 20°C	1.379		d <sub>c</sub> g/ml	0.248	2	Parachor		306.6	4
30	1.376	94 4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.032 247.7	2 2		30	306.5	4
"C"	0.758	0 4		21584.	2		40	306.4	4
MR (Obs.)	34.617	2	P <sub>c</sub> mm PV/RT	21304.	-	<u> </u>		307. 2	5
MR (Calc.			25°C	0.9878	5	Exp. L.1. u.			
(nD-d/2)	1.045		30 mm	1.0000	5	Dispersio		99.3	2
Dielectric	1.912	3	BP	0.9486	4 5	Flash Poi	nt °C	-25.	5
A 0 to			te t <sub>c</sub>	0.9418 0.268	2	Fire Poin			]
B LIFE.	223.303 223.303		ΔHc kcal/m	1071.45	2	M Spec.			
A*  0 to	$\overline{}$	$\overline{}$	ΔHf	-57.05	2	Ultra V.	•		
B* 95 ℃		5	ΔFf	-1.15	2	X-Ray Dif	١.	Yes	2
к — —			Viscosity		1	Solubility	in +	<b>†</b>	<del>                                     </del>
ξ — — to	-1		centistokes 7 20 °C	0.5773	31	Acetone		<b>∞</b>	İ
t <sub>k</sub> to			7 20 °C	0.5775		Carbon t	et.	<b>∞</b>	
A' to		-		1		Benzene Ether		ο ο	
B' °	2		B <sup>V</sup> l to	<del></del>	$\vdash$	n-Heptan	e	œ	
	+	$\dashv$	B <sup>V</sup> to C			Ethanol Water		<b>∞</b>	
A'* to B'* °C			<u> </u>	-		Water in			İ
Ac   115 to		52 4	1						
Bc t °C		2 4		<del>                                     </del>					
Cc'	290.5	4	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°	0.031 0.003		c <sub>p</sub> vap.300°K 400	0.39781 0.50320	2 2				
te °C	86.57	5	c <sub>v</sub> vap.						
$T_R = 0.7$	5 T <sub>C</sub>					+ grams/1	00 oras	ms solven	<u></u>
REFERENC		w 2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat				
SOURCE:		AF							
PURIFICAT	TION:	AF	<del></del>						
			: 3 NBS 514;	31 Ind Fra	Char	20 220 4	19441		
			. 3 NBG 314,	J ma. Dag.	on <b>o</b> n	<u>30,</u> 330 (	, , , , ,		

								No. 19	
NAME	2, 3-Dime	thylp	entane			ST	RUCTURAL	FORMUL	A
							сн сн сн (	-н сн	
Mole % Pur.	Ref. Mo	lecul mula	ar C <sub>7</sub> H <sub>16</sub>	Molecular Weight 100.19	8		ch <sub>3</sub> ch ch c	20113	
		Ref.			Ref.				Ref.
F. P. ℃			dt/dP			f	to		
F.P. 1009	6		°C/mm 25°C	0.3103	4	g	' <u>*K</u>		
B. P. ℃ 760 mm	89.784	2	<b>B</b> P	0.04482	2	h			_
100	33.237	2	t <sub>e</sub>	0.0369	5	f' g'	to *K		
30 10	8.43	4	30 mm	0.6200	4-	h'			
1	-41.19	5	ΔHm cal/g	+		m	300 to	0.0272	4
Pressure mm 25°C	40.07		∆Hv cal/g 25°C	81,68	2	n	_6 <u>0</u> 0_•K	0.0014	4
t <sub>e</sub>	68.87	5	30 mm	84.58	5	°		-0.0 <sub>6</sub> 45	4
Density			BP te	72.48 71.23	2 5	m'	700 to	0.1062	4
g/ml 20°0		2	t <sub>e</sub> (d, e)	71.19	5	n'	[1000 •K	0.0012 -0.0 <sub>6</sub> 40	4
d <sub>4</sub> 25 30	0.69091 0.68673	2 4	ΔHv/T <sub>e</sub>	19.20	5			. 6	
a	0.71175	4	d 5 to		5		ace tension s/cm. 20°C	19.96	2
Ъ	-0.03819	4	_e100 °C		5	8	30	18.98	2
Ref. Index		١, ١	e' °C	:		<u> </u>	40	18.02	2
<sup>n</sup> D 20°0	1.39196	2 2	d <sub>c</sub> g/ml	0.247	2	Par	achor [P] 20°C	304.8	4
30	1.38696	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.042 264.6	2 2		30	304.7	4
"C"	0.7527	4	P <sub>c</sub> mm	22192.	2		40 Sugd.	304.6 307.2	4 5
MR (Obs. MR (Calc.		2 5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04442	2	25°C 30 mm	0.9903 1.0000	5	_	u.	04.3	2
Dielectric	1.939	3	BP	0.9501	4		ersion sh Point C	96.2 -16.	5
A 5 to		2	t <sub>e</sub>	0.9424 0.268	5 2		Point	-10.	,
B (130 °C	221.823	2 2	t <sub>c</sub> AHc kcal/m	1073.12	2	М.	Spec.		
A*  5 to		5	ΔHf	-55.81	2		a V. ay Dif.		
B*[110 °C		5	ΔFf	-1.27	2		ared	Yes	2
K			Viscosity centistokes				bility in +		
t <sub>k</sub> \ \ \ to		1	<b>າ</b> 15 °C	0.6233	31		etone rbon tet.	<b>8</b> 0	
x			30	0.6000	3'	Be	nzene	<b>∞</b>	
A'  to B'  °C			ļ	<b>_</b>	1		ner Heptane	∞ ∞	
C'			B <sup>V</sup> to A <sup>V</sup> C			Etl	nanol	øo	
A'* to B'* °C			⊩. <u>-</u> v. <del>-</del>	-			ter ter in		
Acl 130 to	+	4	(B') to						
Bc tc °C	1652.5	4	·		$\vdash$				
Ce	276.5	4	P						
Cryos, Accounts, B			c vap.300°K	0.39781 0.50320	2 2				
t <sub>e</sub> °C	98.49	5	c <sub>v</sub> vap.						
$T_{\mathbf{R}} = 0.7$			μ		لــــــــــــــــــــــــــــــــــــــ	+ gr	ams/100 gra	ms solven	t
	CES: 1-Dow	2-A	PI 3-Lit. 4-	-Calc. from de	t. da				
SOURCE:		A	PI						
PURIFICA	TION:	A	Ρĭ						
LITERATU	JRE REFERE	NCE	S: 3 NBS 514;	3' Timmern	nans				
L									

								No. 20	
NAME	2,4-Di	methylpe	entane			STR	UCTURAL :	FORMULA	
							сн <sub>3</sub> сн сн <sub>2</sub>	сн сн.	
	1, (			M-1			CH <sub>3</sub>	CH.	
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 100.1	98		3	3	
		Ref.			Ref				Ref
F, P. *C	-119.24	2 2	dt/dP			ſ	l to		
F. P. 100%			°C/mm	0.3353	١. ١	g	<u>*</u> K		ŀ
B. P. *C 760 mm	90.50		25°C BP	0.2252 0.04376	2	h	L		
100 mm	80.50 25.35		t <sub>e</sub>	0.0369	5	f'.	to		
30 10	1.22	4	30 mm	0.6031	4	g'		1	
1	-47.0	5	AHm cal/g	16.318	2	h'	1 300 to	0.0272	4
Pressure			ΔHv cal/g 25°C	78, 44	2	m	600 °K	0.0014	
mm 25°C	98.39! 946.4	5 4 5	30 mm	82.54	5	٥		-0.0 <sub>6</sub> 45	4
Density	740.4		BP	70.36 69.26	2 5	m'	700 to	0.1062	4
g/ml 20°C	0.67		t <sub>e</sub> (d, e)	69.21	5	n'	11000 •K	0.0012	4
dt 25 4 30	0.66		ΔHv/T	19.21	5	٥'		-0.0 <sub>6</sub> 40	4
- JU	0.69		d   0 to	82.73	5		ace tension	18.15	١,
Ъ	-0.03		_d 90 ℃	0.1537	5	gyne	ss/cm. 20°C 30	18.15	2 2
Ref. Index			• 0				40	16.23	2
<sup>n</sup> D 20°C	1.38		d <sub>c</sub> g/ml	0,239	2	Para	achor [P] 20°C	207 /	١.
30	1.37		C mr/g	4, 192 247, 1	2 2		30	307.6 307.5	4
"C"	0.75	30 4	, -	20824.	2		40 Sund	307.5	4 5
MR (Obs.)	34.619		P <sub>c</sub> mm	20024.	1-1	Fyn	L. 1. %/wt.	307.2	-
MR (Calc.) (nD-d/2)	34.526		25°C	0.9865	5	Exp.	u.		
Dielectric	1.914	$\overline{}$	30 mm BP	1.000 <b>0</b> 0.9487	5 4		ersion	98.6	2
A 0 to	+		te	0.9418	5		h Point °C Point	-24.	5
B 1 115 °C	1192.04	1 2	t <sub>c</sub>	0.270	2	M S			}
C	221.634		ΔHc kcal/m ΔHf	1072.44 -56.17	2 2	Ultr	a V.		
A* 0 to B* 100 °C	1.260 1117.86	088   5	ΔFf	-0.49	2	X-R Infra	ay Dif.		1
K			Viscosity				bility in +	<del> </del>	_
t <sub>k</sub>	-		rentistokes 15 °C	0.5538	3,	Ace	etone	œ	
t <sub>x</sub> C			30	0.5351	31	_	rbon tet. nzene	00   00	Į
A' to				1		Eth	er		
B', ∟ _ <u>•</u> C	1	1	B <sup>V</sup>   to	<u> </u>	$\vdash$		leptane anol	80 80	
A¹* to	<del> </del>		AV C	Ì		Wa	ter	~	
B'* *C	<u> </u>	-	(B <sup>V</sup> ) to			Wa	ter in	<del> </del>	-
Ac   115 to	7.332	247   4	(A <sup>V</sup> )  °C					[	
Bc t <sub>c</sub> C	270.5	4	c <sub>p</sub> liq. ∘K					1	
Cryos. A°	0.034		cp vap300°K	0.39781	2			1	
consts. B°	0.003		400	0.50320	2				
t <sub>e</sub> °C	87.98	5	c <sub>v</sub> vap.	1		L		L	
$T_R = 0.75$							ms/100 grai		t
REFERENC	ES: 1-De			Calc. from det	t. dat	ta 5-	Calc, by for	mula	
SOURCE:	ION	AF	<del></del>						
PURIFICAT		AF							
LITERATUI	KE REFE	RENCES	3 NBS 514;	3' Timmerm	ans				

								No. 21	
NAME	3, 3-Dime	thylp	entane		ST	RUCTURAL	FORMUL	A	
							CH CH C	CII CII	
Mole % Pur. 99		lecul mula		Molecular Weight 100,19	28		сн <sub>3</sub> сн <sub>2</sub> с сн <sub>3</sub>		
, , , , , , , , , , , , , , , , , , ,		Ref.		T Tooling	Ref.				Ref.
F. P. °C	-134.46	2	dt/dP			f	to		
F.P. 100%			°C/mm			gl	<u>*K</u>		
B. P. ℃			25°C <b>B</b> P	0.2674 0.04509	2	h			
760 mm 100	86.064 29.241	2 2	te	0.0374	5	f'	to		
30	4.36	4	30 mm	0.6215	4	g'	'° <u>K</u>		
10	-14.4 -45.3	2	AHm cal/g	16.857	2	h'			
Pressure	-13.3		AHv cal/g			m	300 to	0.0272	4
mm 25°C	82.84	4	25°C	78. 76	2 4	n o	_600 <b>.</b> K	0.0014 -0.0 <sub>6</sub> 45	4
t <sub>e</sub>	966.5	5	30 mm BP	81.94 70.71	2		1 700		_
Density	0 (0007		t <sub>e</sub> ,	69.56	5	m'	700 to	0.1062 0.0012	4
g/ml 20°C	0.69327	2 2	t <sub>e</sub> (d, e)	69.55	5	0'	[1000 TK	-0.0640	4
d <sup>t</sup> 25 4 30	0.68488	4	∆Hv/T <sub>e</sub>	18.95	5		·		
a	0.71004	4	d 0 to	82.54	4		face tension es/cm. 20°C	19.59	2
Ъ	-0.03820	4	e 100 °C to	0.1375	4	8	30	18.62	2
Ref. Index			e' °C				40	17.67	2
<sup>n</sup> D 20°C	1.39092	2	d <sub>c</sub> g/ml	0.239	3'	Par	achor [P] 20°C	304.2	4
30	1. 38595	4	II V_mi/g	4.183 263.0	3' 2		30	304.2	4
"C"	0.7527	4	-	22800.	2		40	304.1	4
MR (Obs.)	34, 332	2	P <sub>c</sub> mm	22800.	<u> </u>			307.2	5
MR (Calc.		5	PV/RT 25°C	0.9898	5	Exp	. L.1.%/wt. u.		
(nD-d/2)	1.04428	2	30 mm	1.0000	5	Dis	persion	97.1	2
Dielectric	1.937	3	<b>B</b> P	0.9520 0.9447	4 5	Fla	sh Point °C	-18.	5
A 5 to B   130 °C	6, 82667 1228, 663	2	te t	0.7111		Fire	e Point		
c 1230_0	225. 316	2	ΔHc kcal/m	1072.57	2		Spec.		
A* 5 to	1.24360	5	ΔHf ΔFf	-56.07	2		a V. lay Dif.		
B* 105 °C	1149. 27	5	<del></del>	-0.69	2	Infr	ared	Yes	2
c			Viscosity centistokes				ıbility in +		
t <sub>k</sub> – to			<b>η °</b> C				etone rbon tet.	∞ ∞	
t <sub>x</sub>						Ве	nzene	∞ ∞	
A'  to B'  °C					$\perp$		her Heptane	<b>∞</b> 0	
_ c' ' =	-	1	B <sup>V</sup> to C				hanol	<b>∞</b>	
A!* to			<del> </del>	-			ter		
B'* °C	ļ		( <b>B</b> <sup>V</sup> )  to			Wa	ter in		
Ac 130 to Bc t °C	7.55060 1795.1	4	(A <sup>V</sup> )  °C		$\sqcup$				
Bc_tc_°C	299.3	4	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°	0.04418 0.0040	2 2	c_vap.300°K	0.39781	2				
t <sub>e</sub> °C	94.53	5	c <sub>v</sub> vap.	0.50320	2				
$T_R = 0.7$	5 T <sub>C</sub>	L	<u> </u>	1	اــــــا	+ gr	ams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da		-Calc. by for		
SOURCE:		AP							
PURIFICA?	TION:	AP							
			5: 3 NBS 514;	3' ASTM 109	)				
			9 HB0 314;	5 AGIWITO	•				

T								No. 22	
NAME	2, 2, 3	-Trime	thylbut	ane			STRUCTURAL CH <sub>3</sub>	FORMULA	A.
							сн <sub>3</sub> с сн	CH	
Mole	Por	14-1	1		Melecules		Ċн <sub>3</sub> ċн		
% Pur.	Ref.	Molec Form		С <sub>7</sub> Н <sub>16</sub>	Molecular Weight 100.	198	CH <sub>3</sub> CF	<b>'</b> 3	
<del></del>			ef			Ref			Ref
F.P. °C	-24.91			/dP	T		f   t	. T	
F.P. 100%				C/mm	ł	i i	g to		ł
B. P. *C	+	<del></del>		5°C	0.2227	4	h .	-	1
760 mm	80.88		2 U -	P	0.04484 0.0377	5	f' to		+-
100 30	24.46		2 t	e 0 mm	0,6152	4	g'   '		1
10	-18.8		4		5, 393	2	h'	7	1
1	-49.3		· —	im cal/g	5, 393	۲	m   300 to	-0,0081	4
Pressure	1,65		1 2	Iv cal/g 5°C	76.42	2	n 600 º1	κ 0.0 <b>0</b> 15	4
mm 25°C	102.36		3	0 mm	80.10	5	0	-0.0654	4
Density	+ 131.3	-+	P	P	69.04	2	m'   700 to	0,1121	4
g/ml 20°C	0.69	011 2	2	(d, e)	67.98 67.97	5	n'   11000°1		
t 25	0.68	588 2	2 .	Hv/T <sub>e</sub>	18.82	5	0'	-0.0642	4
	0.68		- 1	1 0 to		5	Surface tension		Π
a b	0.70 -0.0 <sub>3</sub>		ئے ا	_i90 •		5	dynes/cm. 20°0		2
Ref. Index		1025	—∥ a'		5		30 40	17.77	2 2
n <sub>D</sub> 20°C		944 2	2 e'	<u>i                                     </u>	<del></del>	<del>  _  </del>	Parachor [P]	+	F
45	1.38	692 2	² ∥ de	g/ml ml/g	0.254 3.932	2 2	200	302.4	4
30	1.38		<b>1</b> 1 년	*C	258.3	2	30	302.1	4
"C"	0.75	35 4	P.	mm	22610.	2	40 Sug	301.9 d. 307.2	4 5
MR (Obs.) MR (Calc.			DV	/RT	<del> </del>	<del> </del>	Exp. L.1.%/wt		۱Ť
(nD-d/2)	34.52		,   2	5°C	0.9888	5	u.	1	l
Dielectric		<del></del>	——   3°	0 mm P	1.000 <b>0</b> 0.9516	5 4	Dispersion	98.3	2
A 0 to	+		٠,		0.9447	5	Flash Point °C Fire Point	-22.	5
B 1125 °C			- 11 +		0.269	2		<del></del>	+-
С	226.05	0 2		c kcal/m	1071.78	2	M Spec. Ultra V.	Yes	2
A* 0 to					-56.63 -0.17	2 2	X-Ray Dif.	1	
B* ∟100 °C	1122.41	. ] 5	, I—	scosity	<del> </del>	<del> </del>	Infrared	Yes	2
·	_	- 1		ntistokes			Solubility in	i	ĺ
tk to		1	7	•c	-		Acetone Carbon tet.	e0 e0	
t							Bensene	<b>80</b>	1
A'   to		j			<u> </u>		Ether n-Heptane	e0 e0	1
<u>c,                                    </u>			В <sup>V</sup>	l to			Ethanol	- w	
A1* to			- Āv		_		Water Water in		
B'* *(			(B)	.1		1 1	water in	-	$\vdash$
Ac   125 to	7.40	100   4		<u>′)      °c</u>	:				
Bc Ltc_*C	288.7	4	Cp	liq. °K					}
Cryos, A	0.00	441 2	— II -	vap.300°K	0, 39781	2			
consts. B	0.00		<u>.</u>	400	0.50320	2			
t <sub>e</sub> °C	88.73	5	C <sub>v</sub>	vap.					1
$T_R = 0.7$	5 T <sub>C</sub>				<u> </u>	ئـــــــــــــــــــــــــــــــــــــ	grams/100 gr	ams solven	t t
REFEREN	CES: 1-D	low 2-	API 3	-Lit. 4-	Calc. from de	t. dai	ta 5-Calc, by fo		
OURCE:			API						
PURIFICA	TION:		API				·		
LITERATU									
LITERATU									
LITERATU									
LITERATU									
LITERATU									

								No. 23	
NAME	n-Octane					ST	RUCTURAL	FORMUL	A
							CH (CH )	CH	
Mole	Ref. Mo	lecul		Molecular			сн <sub>3</sub> (сн <sub>2</sub> )6	3	
% Pur. 99.	96 2 For	mul		Weight 114.2	224				
	_	Ref.			Ref.			,	Ref.
F.P. °C F.P. 100%	-56.795	2	dt/dP *C/mm			£	to		
B. P. *C	<del>                                     </del>	-	25°C	1.2698	5	g h	<b>! °К</b> !		
760 mm	125.665	2	BP t <sub>e</sub>	0.04738 0.0364	5	- <u>"</u> ,-	to		
100 30	65.704 39.28	2	30 mm	0.6613	4	g'	*K		
10 1	19.2 -14.9	2 4	ΔHm cal/g	43.397	2	h'			
Pressure	<b>†</b>		AHv cal/g	94 90		m	300 to	0.0265 0.0014	4
mm 25°C	14.036 1064.5	4 5	25°C 30 mm	86.80 85.63	5	٥		-0.0646	
Density	1001.5	ب	BP t	71.91 70.06	2 5	m'		0.1079	4
g/ml 20°C		2	te te (d, e)	70.05	5	n' o'	1000 °K	0.0012 -0.0640	4
dt 25 4 30	0.69849 0.69445	2 4	ΔHv/T <sub>e</sub>	19.45	5		<u> </u>	-0.06-20	-
8	0.71862	4	d 20 to e   140 °C	90.50 0.1479	4		face tension es/cm. 20°C	21.76	2
Bef. Index	-0.03802	4	d' to	1		8	30 <b>40</b>	20.77 19.80	2 2
n <sub>D</sub> 20°C		2	e'   °C	0,235	2	Par	achor [P]		<b> </b>
25 30	1.39505	2 4	d g/ml vc ml/g t °C	4.255	2		`20°C 30	351.2 351.2	4
"C"	0.7547	4	1	296.2	2		40	351.2	4
MR (Obs.)	39.192	2	P <sub>c</sub> mm	18726.	2	F	Sugd.	346.2	5 32
MR (Calc.) (nD-d/2)	39.144 1.04617	5 2	25°C	1.0000	5	Exp	u. L.1.%/wt.	11.3	32
Dielectric	1.948	3	30 mm BP	1.0000 0.9411	5		persion	98.0	2 3 <sup>2</sup>
A 40 to	6.92377	2	t <sub>e</sub>	0.9299 0.256	5 2		sh Point °C e Point	15.56	3*
B (155 ℃ C	1355.126 209.517	2	tc AHc kcal/m	1222,77	2		Spec.	Yes	1
A* 40 to	1.38276	4	ΔHf ΔFf	-59.74 1.58	2 2		ra V. Ray Dif.		
B*[150 °C	1278.24	4 5	Viscosity	1.56	-	<b></b>	ared	Yes	1
\$ C150+5	0.14684	4	centistokes	0.5353			ubility in <sup>†</sup> :etone	<b></b>	
t <sub>k</sub> 150 to t <sub>x</sub> 250 °C	190. 353.	4 5	η 60 °C	0.5353 0.4602	2 2	Ca	rbon tet.	<b>∞</b>	
A'   15 to	7.47176	4	100 120	0.4022 0.3560	2 2	Et	her	<b>80</b>	
B'  _ 40 °C	1641.52 234.5	4	BV 20 to	392, 25	4		Heptane hanol	<b>00</b>	
A'* 15 to	1.87375	4	A   75 °C	2.55168	4	w.	ater		
B'* 40 °C	1541.29	4	(B <sup>V</sup> )  75	387.11	4	<b></b>	eter in	<del> </del>	-
Acl 155 to Bc t <sub>c</sub> °C	7.30712 1648.0	4	(A <sup>V</sup> )  130	2.56694	4		ntistokes		
Cc — —	246.9	4	c <sub>p</sub> liq. 20 °C.	0.52	31		30 <b>°C</b> 70	0.7005 0.4950	2
Cryos, A° consts, B°	0.05329 0.0031	2	c vap 300°K	0.39703 0.50217	2				
t <sub>e</sub> °C	138, 25	5	c <sub>w</sub> vap.					1	
$T_{R} = 0.757$	<sup>r</sup> c					† g	rams/100 gra	ms solven	t
	ES: 1-Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP							
PURIFICAT		AP							
LILEKATU.	RE REFERE	NCES	3 NBS Circ	. 514; 3' Tim	nmer	mans	; 3 <sup>z</sup> NFPA 3.	25	

							No. 24	
NAME	2-Methylho	eptan	e		_	STRUCTURAL	FORMULA	<b>L</b>
Mole % Pur.		lecul		Molecular Weight 114.2	24	сн <sub>3</sub> сн (сн	<sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	
		Ref.	<del></del>		Ref			Ref
F, P. *C F. P. 100%	-109.040	2	dt/dP *C/mm			f to		
B.P. °C 760 mm 100 30	117.647 58.297 32.15	2 2 4	25°C BP t <sub>e</sub> 30 mm	0.903 0.04691 0.0364 0.6542	4 2 5 4	h f' to g'K_		
10 1 Pressure	12.3 -21.7	5	ΔHm cal/g ΔHv cal/g			m   300 to	0.0265 0.0014	
mm 25°C	20.637 1052.	5 5	25°C 30 mm BP	83.02 82.65 70.30	2 5 2	0	-0.0 <sub>6</sub> 46	4
Density g/ml 20°C dt 25 d4 30	0.69792 0.69392 0.68991	2 2 4	te te (d, e) AHv/Te	68.65 68.56 19.46	5 5 5	m'   700 to n'   1000 °K	0.1079 0.0012 -0.0 <sub>6</sub> 40	4 4
a b	0.71390 -0.0 <sub>3</sub> 794	4 4	d   25 to e   130 °C d'   to	87.30 0.1445	5	Surface tension dynes/cm. 20°C 30 40	20.60 19.68 18.77	2 2 2
Ref. Index n <sub>D</sub> 20°C 25 30	1.39494 1.39257 1.39020	2 2 4	e' i °C  d g/ml vc ml/g tc °C	0.234 4.273 288.	2 2 2	Parachor [P] 20°C 30	348.7 348.8	4 4
"C" MR (Obs.)	0.7551 39.231	2	P <sub>c</sub> mm	18848.	2	40 Sugd.	348.8 346.2	5
MR (Calc. (nD-d/2) Dielectric		5 2 5	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9487	5 5 4	Exp. L.1.%/wt. u. Dispersion Flash Point °C	98.5	2
A 35 to B <u>150°C</u> C		2 2 2	t <sub>e</sub> t <sub>c</sub> AHc kcal/m	0.9386 0.263	5 2 2	Fire Point M Spec,		
A*  35 to B*  140 °C K	1.36848	5	ΔHf ΔFf Viscosity	-60.98 0.92	2	Ultra V. X-Ray Dif. Infrared Solubility in +	Yes	2
tk to			centistokes 7°C			Acetone Carbon tet. Benzene	80 80	
B' 35 °C	1729.19 247.4	5 5 5	B <sup>V</sup>   to			Ether n-Heptane Ethanol Water	88 88	
B'* 35 °C  Ac   150 to  Bc   t °C	7.24113	5 4 4	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C			Water in		
Cryos. A	0.0458	2	c <sub>p</sub> liq. °K c <sub>p</sub> vap.300°K	0.39703	2			
t <sub>e</sub> °C	129.66	5	c <sub>v</sub> vap.	0.50217	2			
$T_R = 0.7$						+ grams/100 gran		
	ES: 1-Dow			alc. from det	da:	ta 5-Calc, by for	mula	
SOURCE:		AI						
PURIFICAT LITERATU	TION: RE REFERE	AI NCES						

No. 25 3-Methylheptane NAME STRUCTURAL FORMULA  $CH_3CH_2CH$   $(CH_2)_3CH_3$ CH<sub>2</sub> Mole Ref. Molecular Molecular  $C_{8}H_{18}$ Weight 114, 224 % Pur. 99.99 2 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -120.500 2 dt/dP f to °C/mm g °K 25°C 0.9478 5 B. P. °C h BP 0.04712 2 760 mm 118.925 2 0.0361 5 ſ١ to 100 59.358 2 g' <u>°K</u> 30 33.15 4 30 mm 0.6556 4 10 13.3 2 h! ∆Hm cal/g 23,813 2 5 -21.0 0.0265 300 to m ∆Hv cal/g Pressure \_6<u>0</u>0\_**°K** 0.0014 n 4 25°C 83.35. 2 mm 25°C 19.582 4 o ·0.0<sub>6</sub>46 4 30 mm 83.01 te 1070.6 5 BP 71.30 2 m' 700 to 0.1079 4 Density 5 t<sub>e</sub> t<sub>e</sub> (d, e) 69.60 n' 1000 °K 0.0012 g/ml 20°C 0.70582 2 69.56 5 o' -0.0640 4 d<sub>4</sub> 25 0.70175 2 ΔHv/T<sub>e</sub> 5 19.64 30 0.69767 4 Surface tension Т 25 87.54 5 0.72208 to 4 dynes/cm. 20°C 21.17 2 e di 130 °C 0.1366 5 ь -0.03809 4 30 20.24 2 to 19.32 40 2 Ref. Index e¹ <sup>n</sup>D 20°C 1.39848 [P] Parachor dc g/ml 0.239 2 25 1.39610 2 20°C 347.2 4 vc ml/g t\_°C 4.185 2 30 1.39374 4 30 347.3 4 292. 2 t<sub>c</sub> 40 347.4 4 "C" 0.7531 4  $P_c$  mm 19456. 2 Sugd 346.2 5 MR (Obs.) 39.100 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 39.144 25°C 1.0000 5 (nD-d/2)u. 2 1.04557 30 mm 1,0000 5 97.5 Dispersion 2 Dielectric 1.956 5 BP 0.9601 4 Flash Point C 5 6. 0.9504 30 to 5 6.89944 2 Fire Point 0.264 2 В 1\_150 °C 1331.530 2 M. Spec. С AHc kcal/m 2 1221.76 212.414 2 Ultra V. ΔHf -60.34 2 A\* 30 to 1.33027 5 X-Ray Dif. ΔFf 2 1,12 B\* 140 °C 1246.6 2 Infrared Yes ĸ Viscosity Solubility in centistokes Acetone to œ Carbon tet. °C 00 t<sub>x</sub>\_ Benzene 00 0 to 7,77698 Ether œ В' 30 °C 1797.39 5 n-Heptane B<sub>v</sub> | C' 252.16 5 to Ethanol °C Water 10 to 2.15798 5 Water in B'\* 30 °C 1684.52 (B<sup>V</sup>)| to Ac | 150 to (A<sup>V</sup>) 7.30319 5 °C Bc tc °C 1640.9 c<sub>p</sub> liq. ۰ĸ Cc 252.4 5 c<sub>p</sub> vap.300°K Cryos. A° 0.0587 2 0.39703 2 consts. B° 0.50217 400 c vap. t<sub>e</sub> ℃ 5 131.67  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

<u></u>		***					No. 26	
NAME	4-Meth	ylheptan	e			STRUCTURAL	FORMULA	<b>L</b>
						CH ICH / CHI	сн ) сн	
	$\neg \neg \neg$				$\neg \vdash$	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>2</sub> СН( СН <sub>3</sub>		
Mole % Pur. 99.	99 Ref.	Molecul Formul		Molecular Weight 114.2	224	03		
<b>70 1 41.</b> //	· // 1 - 1	Ref		Worgat 1111	Ref			Ref
F. P. *C	-120.95		dt/dP	T		f   to	T	-
F.P. 100%			*C/mm	ł	Н	g L		
B. P. *C			25°C BP	0.9068 0.04695	5 2	h ,		
760 mm 100	117.70 58.34		t <sub>e</sub>	0.0361	5	f' to		
30	32.23	4	30 mm	0.6534	4	g'   'K	-	ł
10 1	12.4	2 4	ΔHm cal/g	22.675	2	h'		L.
Pressure	<del> </del>		ΔHv cal/g			m   300 to n   600 °K	0.0265	
mm 25°C	20.55		25°C 30 mm	83. 01 82. 79	5	0	-0.0646	
t <sub>e</sub>	1063.6	5	BP	70.91	2	m'   700 to	0,1079	├
Density g/ml 20°C	0.70	463 2	te te (d, e)	69. 23 69. 18	5	n' 1000 K	0.0012	4
<sub>d</sub> t 25	0.70	055 2	ΔHv/T	19.61	5	0'	-0.0 <sub>6</sub> 40	4
	0.69		d   30 to		5	Surface tension		Γ
a b	0.72		<u>e l 130 °C</u>	0.1390	5	dynes/cm. 20°C	21.00	2 2
Ref. Index	<del></del>		d'   to			40	19.15	2
n <sub>D</sub> 20°C	1.39		d <sub>c</sub> g/ml	0, 240	2	Parachor [P]		
25 30	1.39		I V mi/g	4.167	2	20°C 30	347.1 347.2	4
"C"	0.75		_ ~	290.	2	40	347.3	4
MR (Obs.)	+		P <sub>c</sub> mm	19456.	2	Sugd	346.2	5
MR (Calc.	39.14	4 5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	1.04		30 mm	1.0000	5	Dispersion	97.5	2
Dielectric	1.95		BP t <sub>e</sub>	0.9575 0.9477	4 5	Flash Point °C	6.0	5
A 25 to B   150 °C			tc	0.264	2	Fire Point	ļ	<b>├</b> ─-
с ——	212.56	8 2	ΔHc kcal/m	1221.89	2	M Spec. Ultra V.	l	ł
A*   25 to			ΔHf ΔFf	-60.17 1.86	2 2	X-Ray Dif.		
B* 140 °C	1244.16		Viscosity			Infrared Solubility in +	Yes	2
£ .—=	_		centistokes	İ		Solubility in + Acetone	<b>∞</b>	İ
t <sub>k</sub> to			7 ℃			Carbon tet.	· ·	]
A' 0 to	7.89	70 5				Benzene Ether	&C &C	
B' L 35 °C	1860.28 257.54		B <sup>V</sup> to	-	$\vdash$	n-Heptane	· ·	
A'* 10 to			A <sup>V</sup>   C			Ethanol Water	<b>∞</b>	
B'* 35 °C			(B <sup>V</sup> ) to	1		Water in	L	ļ
Ac   150 to	7.31		(A <sup>V</sup> ) °C			ì		
Bc tc_°C	1645.2 253.5	4 4	c <sub>p</sub> liq. °K	<b>†</b>				)
Cryos. A°	0.05		l -	0.20703	,			
consts. B°	0.03		c <sub>p</sub> vap.300°K 400	0.39703 0.50217	2			l
$t_e$ °C $T_R = 0.7$	130.15	5	c <sub>v</sub> vap.			L	1	<u> </u>
REFEREN		ow 2-AF	PI 3-Lit, 4-0	Tale from de		grams/100 gra		<u> </u>
SOURCE:		AF		Calc. from de	QM	ta 5-Calc, by for	muia	
PURIFICAT	TION:	AF				<del></del>		
LITERATU							<del></del>	
			•					

No. 27 3-Ethylhexane NAME STRUCTURAL FORMULA CH3CH2CH (CH2)2CH3 Ċ<sub>2</sub>H<sub>5</sub> Mole Molecular Molecular  $C_{8}H_{18}$ Weight 114.224 % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g °Κ 0.9283 B. P. °C h ВP 0.04719 2 760 mm 118.534 2 0.0359 5 f¹ te tο 100 58,902 2 °K g' 4 0.6559 4 30 32.68 30 mm 10 12.8 2 h١ AHm cal/g -20.47 5 300 to 0.0265 m AHv cal/g Pressure n 1 600 °K 0.0014 25°C 83.19 mm 25°C 20.03 o -0.0646 4 5 30 mm 82.70 1080.9 5 t<sub>e</sub> ВP 71,70 2 m 700 to 0.1079 Density 5 70.52 t<sub>e</sub> (d, e) 0.0012 n' 1000 °K g/ml 20°C 5 0.71358 2 70.01 ٥' 4 -0.0640 dt 4 25 0.70948 5 AHv/Te 19.88 30 0.70537 Surface tension Т d 30 86, 88 5 to 0.72996 4 dynes/cm. 20°C 21.51 0.1281 ᇷᅴ 1<u>30 °C</u> 5 ь -0.0381 4 20.58 30 2 40 19.64 2 Ref. Index e¹ 20°C 1.40162  $^{n}D$ [P] Parachor d<sub>c</sub> g/ml 0.245 2 25 1.39919 2 20°C 345.2 ,ς ... 4 ml/g 4.080 2 30 1.39644 5 30 345.1 4 <sup>t</sup>c 294. 5 40 344.9 4 "C" 0.7498 5 P<sub>c</sub> mm 19914. 5 Sugd. 346.2 5 MR (Obs.) 38.944 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 39.144 25°C 1.0000 (nD-d/2)1.04483 2 u. 30 mm 1.0000 5 Dispersion 96.4 2 0.9687 Dielectric 1.964 5 BP 4 Flash Point C -4. 5 0.9596 5 A 25 to 6.89098 2 Fire Point 0.264 2 B (150 °C 1327.884 2 M. Spec. С 212,595 2 2 ∆Hc kcal/m 1222, 19 Ultra V. ΔHf -59.88 A\* 25 to 1.30712 5 X-Ray Dif. ΔFf 1.80 B\*[ 140 °C 1239.30 Infra red Yes 2 ĸ Viscosity Solubility in centistokes Acetone to t<sub>k</sub> Carbon tet. °C 00 Benzene œ ۸' 0 to 7.30017 Ether œ B' 35 °C 1529.6 5  $\overline{B_{\mathbf{v}}^{\mathbf{v}}}$ n-Heptane œ C' 230. 5 Ethanol œ °C Water 1.72722 A!\* 10 to 5 Water in B'\* 35 °C 5 (B<sup>V</sup>)| 1437.6 to Acl 150 to 7.29602 5 (AV) °C Bc tc °C 1639.8 5 c<sub>p</sub> liq. °K Cc' 253,2 5 c<sub>p</sub> vap.300K Cryos. A 0.39703 2 consts. B° 400 0.50217 2 c vap. t<sub>e</sub> °C 131,67 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

				· · · · · · · · · · · · · · · · · · ·					No. 28	
NAME	2,2-D	imeth	ylhe	xane			STR	UCTURAL	FORMULA	Ł
								сн <sub>3</sub> с - (сн	a) aCHa	
Mole	Ref.	Mole	1		Molecular			CH <sub>3</sub>	2/303	
% Pur. 99		For			Weight 114.2	24				
		I	Ref.			Ref.				Ref.
F.P. °C	-121, 18	3	2	dt/dP			f	300 to	0.0445	5
F.P. 100%	<u>'                                    </u>		_	*C/mm 25*C	0.5772	4	g	600 <u>°K</u>	l	)
B. P. °C 760 mm	106, 84		2	BP	0.04650		h	<b>+</b>	-0.0 <sub>6</sub> 37	5
100	48.22	22	2	te	0.0366	5	f'	to *K	ļ	!
30 10	22.54 3.1	١ ١	4 2	30 mm	0.6417	4	g' h'	'	1	1
1	-28.8		4	ΔHm cal/g		$\sqcup$	m	1 300 to	0, 0265	4
Pressure		. T		∆Hv cal/g 25°C	78.02	2	n'	_600°K	0.0014	4
mm 25°C	34.05 1033.1	, [	4 5	30 mm	79.03	5	°	ì	-0.0646	4
Density	+	$\dashv$	-	BP t	67.7	2 5	m'	700 to	0.1079	
g/ml 20°C			2	te te (d, e)	66. 19	5	n' o'	1 1000 °K	0.0012 -0.0 <sub>6</sub> 40	
dt 25 4 30	0.69	3695	2 4	AHv/Te	19.33	5	L	1	-0.0640	<u> </u>
	0, 71		4	d   20 to		4		ace tension s/cm. 20°C	19.60	2
Ъ	-0.03		4	_e,_  120_ °C		4	3,	30	18.69	2
Ref. Index				e' •C			<u> </u>	40	17.80	2
<sup>n</sup> D 20°C	1.39		2 2	d <sub>c</sub> g/ml	0.245	2	Par	achor [P] 20°C	}	j
30	1.38		4	vc ml/g tc °C	4.088	2 2	1	30		
"C"	0.75	52	4	P <sub>c</sub> mm	19456.	2		40 Sugd.	346.2	5
MR (Obs.)			2	PV/RT		-	Exp	. L.1.%/wt.	310.2	<del>-</del>
MR (Calc. (nD-d/2)	39.14		5 2	25°C	0.99147	5	_	u.		
Dielectric			5	30 mm BP	1.0000	5		ersion	99.7	2
A 20 to	6.83	715	2	t <sub>e</sub>	0.9490	5 2		sh Point °C Point	-3.	5
B 1740.0			2 2	t <sub>c</sub> ΔHc kcal/m	0.264	2	M S	pec.		$\vdash$
A*  20 to	215.07		5	ΔHC KCMI/HI	1218.88	2	Ultr		l	1
B* 130 °C			5	ΔFf	0.72	2		ay Dif. ared	Yes	2
K		1		Viscosity centistokes	1		Solu	bility in +		
the Tto		1		7 °c				etone rbon tet.	ø0 ø0	l
<del>v</del> i			_	]			Be	nzene	œ	
A'   to							Eth	ier Jeptane	<b>00</b>	
C'				B <sup>V</sup> to			Eth	anol	00 00	
A'* to B'* *C				A <sup>V</sup> I — °C	.]		Wa Wa	ter ter in		
Ac   140 to		434	5	(B <sup>V</sup> )  to					<del>                                     </del>	T
Bc t <sub>c</sub>		134	5	(A <sup>V</sup> )  °C			1		1	
Ce	256.0		5	c <sub>p</sub> liq. 300°K 400	0.39891 0.50224	5				
Cryos. A° consts. B°	0.03	54	2	c <sub>p</sub> vap.300°K 400	0.39703 0.50217	2				
t <sub>e</sub> °C	118.07		5	c <sub>w</sub> vap.						
$T_{R} = 0.7$							+ gra	ams/100 gran	ms solven	t
REFEREN	CES: 1-D	ow 2			Calc. from det	t. dat	ta 5-	Calc. by for	mula	
SOURCE:			AF	) I						
PURIFICA'			AF							
LITERATU	RE REFI	EREN	CES	:						

39 No. 29 2, 3-Dimethylhexane STRUCTURAL FORMULA NAME CH3CH CH (CH2)2CH3 ĊH<sub>3</sub>ĊH<sub>3</sub> Mole Ref. Molecular Molecular Weight 114.224 C8H18 % Pur Formula Ref. Ref. Ref. F. P. °C dt/dP f 300 to 0.0445 5 F.P. 100% °C/mm \_6<u>0</u>0\_**°K** 0.0013 5 g 25°C 0.8131 4 B. P. °C 5 h -0.0637 ВP 0.04724 2 115.607 760 mm 2 0.0363 5 ſ١ 100 55.965 2 g' °<u>K</u> 30 29.77 4 30 mm 0.6548 4 10 9.9 2 h' ∆Hm cal/g -25.45 4 1 300 to 0.0265 m AHv cal/g Pressure | 600 °K n 0.0014 25°C 81.17 2 mm 25°C 23, 44 4 0 -0.0646 4 30 mm 81.29 4 te 1066.2 5 BP 70.20 2 0.1079 m' 700 to 4 Density 68.59 5 te (d, e) n' g/ml 20°C 1000 •K 0.0012 0.71214 2 68.57 5 ۰, 4 -0.0640 ď4 25 0.70809 2 ΔHv/Te 5 19,52 30 0.70403 4 Surface tension 30 85.14 d 4 to 0.72832 4 dynes/cm. 20°C 20.99 2 130 0.1293 4 ь -0.03804 4 30 20.07 2 d٦ to 40 19.16 2 Ref. Index e' °C 20°C 1.40113 [P] nD d g/ml vc ml/g t °C Parachor 0.248 2 1.39880 25 2 Z0°C 4.036 2 30 1.39625 4 30 t<sub>c</sub> 293. 2 40 "C" 0.7509 4 P<sub>c</sub> mm 20216. 2 Sugd. 346. 2 5 MR (Obs.) 38.981 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 39. 144 5 25°C 1.0000 (nD-d/2)5 u. 1.04506 2 30 mm 1,0000 5 Dispersion 97.0 2 Dielectric 1.963 5 BP 0.9639 4 Flash Point C 0.9546 6.87004 A | 25 to 2 Fire Point 0.264 2 2 B (150 °C 1315.503 M. Spec. c 2 214.157 AHc kcal/m 1221,45 2 Ultra V ΔHf -60.40 A\* 25 to 5 2 1.29661 X-Ray Dif. ΔFf B\* 140 °C 1228.9 2.17 Infrared Yes 2 ĸ Viscosity Solubility in centistokes Acetone œ to t<sub>k</sub> | t<sub>x</sub> | Carbon tet. œ °C Benzene œ 0 to 8.51379 Ether œ В' 30 °C 2239.64 4 n-Heptane œ B<sub>v</sub> C 288.51 4 to Ethanol 00 °C A'\* 10 to B'\* 30 °C 2.84821 Water 4 Water in (B<sup>V</sup>) 2097.36 4 to Acl 150 to 7.27817 5 (AV) °C 1630.7 Bc Ltc\_ °C c<sub>p</sub> liq. 300°K 0.39891 5 255.6 5 400 0.50224 Cryos. A° cp vap.300°K 0.39703 2 consts. B° 400 0.50217 2 c, vap. t<sub>e</sub> °C 128.23 5 TR  $= 0.75 T_{\rm c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 5-Calc. by formula 4-Calc. from det. data API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

> In PHYSICAL PROPERTIES OF CHEMICAL COMPOUNDS-II: Advances in Chemistry; American Chemical Society: Washington, DC, 1961.

								No. 30	
NAME	2,4-Dime	thyll	hexane			STR	UCTURAL 1	FORMULA	1
l I							сн3 сн	-	
<b> </b>						CF	13CH CH2CH	CH <sub>2</sub> CH <sub>3</sub>	
Mole	Ref. Mo	lecul	ar C <sub>8</sub> H <sub>18</sub>	Molecular					
% Pur.	Fo	_		Weight 114.2	_				
<b></b>		Ref.		<del></del>	Ref.	<u> </u>	. 1		Ref.
F.P. *C F.P. 1007		-	dt/dP	1		f	300 to	0.0445	
B. P. *C	<u> </u>	├	*C/mm 25*C	0.6388	4	g	_ 600 •K	0.0013	1
760 mm	109.429	2	BP	0.04664	2	h	<del>                                     </del>	-0.0 <sub>6</sub> 37	5
100	50.585	2	t <sub>e</sub>	0.0365	5	f' g'	to L*K_		İ
30 10	24.77 5.2	2	30 mm	0.6451	4	h'			
1	-26.9	4	ΔHm cal/g	<b></b>	-	m	300 to	0.0265	4
Pressure			ΔHv cal/g 25°C	79.02	2	n	_600°K	0.0014	4
mm 25°C	30.35 1042.0	5	30 mm	79.82	4	0		-0.0 <sub>6</sub> 46	4
Density	1042.0	3	BP	68.5	2 5	m'	700 to	0.1079	4
g/ml 20°0	0.70036	2	te te (d, e)	66.99	5	n'	1 <u>1</u> 000 •K	0.0012	
dt 25	0.69620	2	ΔHv/T <sub>e</sub>	19.41	5	٥'	 	-0.0 <sub>6</sub> 40	4
	0.69203	4	d   25 to		4		ace tension		_ ا
a b	0.71698 -0.0 <sub>3</sub> 824	4	_e _  120 °C	0.1337	4	dyne	s/cm. 20°C 30	20.05 19.13	2 2
Ref. Index	<del></del>	Ť	d' l to				40	18.22	2
n <sub>D</sub> 20°0	1.39534	2	<del> </del>	0, 245	2	Para	chor [P]		
25 30	1.39291	2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	4.080	2		20°C 30		ŀ
"C"		4	t <sub>c</sub> °C	282.	2	1	40		ŀ
MR (Obs.)	0,7533	-	P <sub>c</sub> mm	19608.	2		Sugd.	346.2	5
MR (Calc.		2 5	PV/RT	0.000	_	Exp.	L.1.%/wt.		
(nD-d/2)	1.04516	2	25°C 30 mm	0.9907 1.0000	5	Dian	u. ersion	97.8	2
Dielectric	1.947	5	BP	0.9591	4		h Point °C	-2.	5
A 15 t		2	t <sub>e</sub>	0.9501 0.264	5 2		Point		
B (_14 <u>5</u> °9	214.790	2 2	t <sub>c</sub>	1220, 15	2	M SI	ec.		
A* 15 to	+	5	ΔHf	-61.47	2	Ultra	a V. ay Dif.		
B* 130 °C		5	ΔFf	0.89	2	Infra		Yes	2
K c	İ		Viscosity centistokes			Solu	bility in +		
t <sub>k</sub>			7 °C				tone bon tet.	00	
t <sub>x</sub> i °C	3		<b>!</b> `				izene	ο ο	
A' to				ĺ		Eth		oc	
c;	-		B <sup>V</sup>   to				leptane anol	00 00	
A1* to	5	T	A <sup>V</sup>   °C			Wat	ter		
B'* °(			(B <sup>V</sup> ) to	1		Wat	ter in		-
Ac   145 to	7. 29625	5	(A <sup>V</sup> ) °C	1					
Bc Lc_°C	258.9	5	cp liq. 300°K	0.39891	5				
Cryos, A	<del></del>		400 c <sub>p</sub> vap.300°K	0.50224 0.39703	5 2				
consts. B			400	0.50217	2				
t <sub>e</sub> °C	121.02	5	c <sub>w</sub> vap.						
$T_R = 0.7$	'5 T <sub>c</sub>					+ gra	ms/100 gran	ns solven	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	da				
SOURCE:		P	API						
PURIFICA	TION:	A	PI						
LITERATU	RE REFERE	NCES	<b>3:</b>						

								No. 31	
NAME	2,5-Dime	thylh	nexane			STRUCTURAL FORMULA			
							СН3	сн3	
Mole % Pur. 99		lecul mul		Molecular Weight 114,22	24	c	:н <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub>		
		Ref.			Ref.				Ref.
F. P. °C	-91.200	2	dt/dP			f	300 to	0.0445	5
F. P. 1009	<u> </u>	├-	*C/mm 25*C	0.6359	4	g	'_600_•K	0.0013	5
B. P. °C 760 mm	109.103	2	BP	0.04646	2	h	<del> </del>	-0.0 <sub>6</sub> 37	5
100 30	50.468	2	t <sub>e</sub>	0.0364	5 4	f' g'	to *K		
10	24.74 5.3	2	30 mm	0,6432	<del> </del>	h'			
1	-26.76	4	ΔHm cal/g	<b>+</b>	$\vdash$	m	300 to	0.0265	4
Pressure mm 25°C	30, 41	4	ΔHv cal/g 25°C	79.21	2	n o	_6 <u>0</u> 0_•K	0.0014	4
t <sub>e</sub>	1040.3	5	30 mm BP	80.03 68.60	2			-0.0646	*
Density			t.	67.08	5	m'	700 to	0.1079 0.0012	4
g/ml 20°0	0.69354	2	le (a, e)	67.04	5	٥'	<u>                                   </u>	-0.0640	4
d <sub>4</sub> 25	0.68513	4	ΔHv/T <sub>e</sub>	19.46	5	Sur	face tension		<del>                                     </del>
a b	0.71032	4	d   25 to e   120 °C	83.39 0.1355	4 4		es/cm. 20°C	19.73	2
Ref. Index	-0.03831	•	d' to			,	30 <b>4</b> 0	18.82 17.92	2
n <sub>D</sub> 20°0	1.39246	2		0,239	2	Par	achor [P]		
25	1.39004	2	d g/ml vc ml/g t °C	4.185	2		20°C 30	346.8 347.6	4
"C"	0,7552	4	`	279.	2		40		
MR (Obs.	39.260	2	P <sub>c</sub> mm	19000.	2	<u> </u>		346.2	5
MR (Calc. (nD-d/2)		5 2	25°C	0.9904	5	Exp	u. L.1.%/wt.		
Dielectric	1.04569	5	30 mm BP	1.0000 0.9585	5 4		persion	99.0	2
A 20 to		2	ll t	0.9495	5		sh Point C e Point	1.	5
B (140 °C		2 2	t <sub>c</sub>	0.264	2		Spec.		
A*  20 to		5	ΔHc kcal/m ΔHf	1219.37 -62.26	2 2	Ult	ra V.		
B*[130 °C		5	ΔFf	0.59	2		Ray Dif. ared	Yes	2
K c			Viscosity centistokes				ability in +		
t <sub>k</sub> to			<b>່າ</b> 15 °C	0.7340	3		etone rbon tet.	ος ος	
t <sub>x</sub> °C		-	30	0,6305	3	Be	nzene	∞	
B'			B <sub>v</sub> to				her Heptane	οο οο	
C'		-	A to				hanol ater	oc	
A'* to B'* °C			(B <sup>V</sup> )  - to	-			ter in		
Ac  140 to		5	(A <sup>V</sup> )  °C						
Bc tc °C	1570.2 251.5	5 5	c <sub>p</sub> liq. 300°K	0.39891	5	1			
Cryos, A	<del></del>	2	H - 400	0.50224 0.39703	5 2				
consts. B		Ĺ	400	0.50217	2				
t <sub>e</sub> °C	120.58	5	c <sub>v</sub> vap.			<u>L</u>		l	
$T_R = 0.7$		<u> </u>	DI 2 II.				ams/100 gra		t
	CES: 1-Dow	2-A A	PI 3-Lit. 4- PI	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE: PURIFICA	TION.		PI						
			S: 3 Timmerm	nans			····		
1									
!									

							No. 32	
NAME	3, 3-Din	nethylhe	xane			STRUCTURAL	FORMULA	<u>.</u>
<u> </u>						CH <sub>3</sub>		
	$\top \top \top$					сн <sub>3</sub> сн <sub>2</sub> с - (с	CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	
Mole		Molecul		Molecular		с́н <sub>3</sub>		
% Pur.		Formula	0 10	Weight 114.2	-			D
		Ref		<u> </u>	Ref		r	Ref
F.P. °C F.P. 100%	-126.10	2	dt/dP *C/mm	]		f to		
B, P. *C			25°C	0.6825	4	g <u> </u>	1	
760 mm	111.969	2	BP	0.04741	5	H	<del> </del>	├
100 30	52.16 25.93	2 4	t <sub>e</sub> 30 mm	0.65557	4	f' to		
10	6. 1	2		0.05557	H	h' i	1	i
1	-27.	5	ΔHm cal/g	<del> </del>	-	m   300to	0.0265	4
Pressure mm 25°C	28.611	5	ΔHv cal/g 25°C	79.20	4	n _600°K	0.0014	
t <sub>e</sub>	1053.406	5	30 mm	79.15 68.5	5 2	<u> </u>	-0.0646	4
Density			BP t_	66.89	5	m'   700 to	0.1079	4
g/ml 20°C	0.710		te te (d,e) AHv/T	66.99	5	n' 1000°K	0.0012 -0.0 <sub>6</sub> 40	
dt 25 4 30	0.705 0.701		ΔHv/T <sub>e</sub>	19.26	5	<u> </u>	6.5	Ļ
a	0-726		d   25 to	82.36	4	Surface tension dynes/cm. 20°C	20, 63	2
Ъ	-0.038	0 4	125 °C	0.1238 80.61	5	<b>3</b> 0	19.72	2
Ref. Index	1 400		e' 25 °C		5	40	18.81	2
n <sub>D</sub> 20°C	1.400 1.397		d g/ml vc ml/g	0.254	2	Parachor [P] 20°C	342.9	4
30	1.395		t <sub>c</sub> *C	3.940 277.9	2 5	30	343.0	4
"C"	0.751	3 4	P <sub>c</sub> mm	17388.	5	40 Sugd	343.0 346.2	4 5
MR (Obs.)	39.009	2	PV/RT		$\vdash$	Exp. L.1.%/wt.	340.2	۲Ť
MR (Calc.) (nD-d/2)	39.144 1.045	09   5	25°C	1.0000	4	u.	]	
Dielectric			30 mm BP	1.0000 0.96198	5 4	Dispersion	97.3	2
A   25 to	6,851	21 2	t <sub>e</sub>	0.95282	5 2	Flash Point °C Fire Point		l
B <u>140°C</u>	1307.882 217.439	2 2	t <sub>c</sub>	0.264	2	M Spec.	<del> </del>	$\vdash$
A*  25 to	1, 282		ΔHc kcal/m ΔHf	1219.97 -61.58	2	Ultra V.	ļ	Ì
B* 135 °C	1221.27	5	ΔFf	1,23	2	X-Ray Dif. Infrared		ļ
K ———			Viscosity	}	1 1	Solubility in +	<b>T</b>	<u> </u>
t <sub>k</sub>   to			centistokes 7 °C			Acetone		
*x			•			Carbon tet. Benzene		
A'   0 to B' <u>35</u> °C	7.239 1503.4	0 5				Ether		
c, C 33 3	235.	5	B <sup>V</sup>   to			n-Heptane Ethanol		ĺ
A** 10 to	1.647		A <sup>V</sup> C			Water Water in	ļ	i
B'* 35 °C	1404.8	5	(B <sup>V</sup> ) to	1		water in		<del> </del>
Ac   140 to	7.261° 1617.9	74 5	(A <sup>V</sup> )  °C		igsquare	Į.		
Cc	257.6	5	c <sub>p</sub> liq. °K	1		ŀ	j	
Cryos. A° consts. B°	0.04	2	c <sub>p</sub> vap.300°K 400	0.39703 0.50217	2			
te °C	124.168	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.75$						grams/100 gra	ms solveni	
REFERENC	ES: 1-Dov	w 2-AF	PI 3-Lit. 4-0	alc. from de	t. dat			
SOURCE:		AF	PI					
PURIFICATI		AF						
LITERATUR	E REFE	RENCES	:					

		,							No. 33	
NAME	3, 4-D	imeth	hylhe	xane			ST	RUCTURAL	FORMUL	A
1 1						1		сн <sub>3</sub> сн	13	
							CI	н <sub>з</sub> сн <sub>2</sub> сн сн	CH2CH3	
Mole % Pur.	Ref.		ecula mula		Molecula <i>r</i> Weight 114.22	4		, -		
// · · · · ·			Ref		T T	Ref.	<del></del>			Ref.
F.P. °C	1		Ver.	14/15	<del> </del>	Ker.	-	T	T	$\vdash$
F.P. 1009	-			dt/dP °C/mm			f g	300 to _600_°K	0.0445 0.0013	
B. P. °C				25°C	0.8713	4	h		-0.0637	1
760 mm	117.72		2	BP t <sub>e</sub>	0.04752 0.03643	2 5	f'	to		
1 <b>0</b> 0 30	57.69 31.31		2 4	30 mm	0.6596	4	g'	! <b>°K</b>		
10	11.3		2	ΔHm cal/g		$\vdash$ $\dashv$	h'			
1	-22.33		4	ΔHv cal/g		$\vdash$	m	300 to	0.0265	4
Pressure mm 25°C	21.64		4	25°C	81.65	2	n	_6 <u>0</u> 0_•K	0.0014	4
t <sub>e</sub>	1066. 3		5	30 mm BP	81.52	5			-0.0 <sub>6</sub> 46	*
Density				t <sub>e</sub>	70.20 68.72	2 5	m'	700 to	0.1079	4
g/ml 20°C			2 2	te (d, e)	68.55	5	n'	1000 •K	0.0012 -0.0 <sub>6</sub> 40	4
d <sub>4</sub> 25 30	0.71		4	ΔHv/T <sub>e</sub>	20.08	5			11.6	
a	0.73		4	d 30 to	85.63	5		face tension es/cm. 20°C	21.64	2
b	-0.03		4	_e   _130 °C to	0.1310	5	8,	30	20.71	2
Ref. Index				e' C				40	19.79	2
<sup>n</sup> D 20°C	1.40		2 2	d <sub>c</sub> g/ml	0.253	2	Par	achor [P] 20°C	242 2	4
30	1.39		4	I V_ mi/g	3.957	2	1	30	343.3 343.2	4
"C"	0.74	86	4	1 _	298. 20824.	2 2		40 50 m²	342.6	4 5
MR (Obs.)	38.84	5	2	P <sub>c</sub> mm PV/RT	20024.	۲			346, 2	-
MR (Calc. (nD-d/2)			5 2	25°C	1.0002	5	Exp	u. L.1.%/wt.		
Dielectric	1.04	$\rightarrow$		30 mm	1.0000	5	Dis	persion	96.6	2
A 30 to			5	BP t <sub>e</sub>	0.9593 0.9496	4 5		sh Point °C	4.	5
B 155°C			2 2	t c	0.264	2		e Point		
c	214.86		2	ΔHc kcal/m	1221.68	2		Spec. ra V.		
A* 30 to			5	ΔHf ΔFf	-60.23 2.03	2 2	X-F	lay Dif.		
B*[140°C	1244. 36		5	Viscosity				ared	Yes	2
·	_	1	ļ	centistokes				ability in Tetone	<b>o</b> o	
t <sub>k</sub> to		ĺ		η °C				rbon tet.	- oo	
A'   0 to		803	5					nzene her	<b>00</b>	
B' 30 °C	1533.24		5			-		Heptane	∞ ∞	
C'	233.		5	B <sup>V</sup> to A °C				hanol	∞	
A'* 10 to B'* 30 °C			5					iter iter in		
Ac  155 to			5	v						
Bc tc °C	1684.6		5		0. 20001	-				
Cc	261.3		5	c <sub>p</sub> liq. 300°K 400	0.39891 0.50224	5 5				
Cryos, A <sup>c</sup> consts, B <sup>c</sup>		ļ	l	c vap.300°K	0.39703	2				
t <sub>e</sub> °C	+		_	c <sub>v</sub> vap.	0.50217	2				
$T_R = 0.7$	130.42 5 T	-	5	L <b>*</b>	L	لــــا	L			<u> </u>
		)OW	2 . A	PI 3-Lit. 4-	Calc from 3-	+ 4-		ams/100 gra		t
	CES: 1-L	,0 <b>W</b>	AP		Calc. from de	i. da	<b>на</b> Э	-Caic. by for	111018	
SOURCE:	TION:		AP						<del></del>	
PURIFICA		ED ES								
LITERATU	RE REF	EKEN	NCES	<b>:</b>						

				_						No. 34	
NAME		3-Ethy	1-2-	meth	ylpentane			STR	UCTURAL	FORMULA	
									CH3C2H5		
	-	$\top$			T			(	сн,сн сн		
Mole		Ref.	Мо	lecul		Molecular	.		•	- ,	
% Pur. 99	. 99	5 2	Fo	rmul	-8-18	Weight 114.2	_				-
	_			Ref.		<del></del>	Ref		<del> </del>		Ref.
F.P. *C F.P. 1009		114.96	<u> </u>	2	dt/dP *C/mm			f	300 to	0.0445	
B. P. *C	+			H	25°C	0.8023	4	g	600•K	0.0013	l
760 mm		115.65	0	2	BP	0.04748	2	h	<del>                                     </del>	-0.0637	5
100 30		55.71		2	t <sub>e</sub>	0.0366	5	f' g'	to		l
10	ı	29.40 9.5		4 2	30 mm	0.6577	4	h'	i	1	
1		-26.1		4	ΔHm cal/g	ļ	$\vdash$	<u></u>	1 300 to	0, 0265	4
Pressure	T				ΔHv cal/g 25°C	80,60	2	n n	_600 °K_	0.0014	
mm 25°C		23.92 063.7		4 5	30 mm	80.74	5	۰	i	-0.0 <sub>6</sub> 46	4
Density	+	003.7		-	BP	69.70	2 5	m'	700 to	0.1079	4
g/ml 20°0	:	0.71	932	2	te te (d, e) AHv/T	68.11	5	n'	1 <u>1</u> 000 • K	0.0012	
dt 25 4 30		0.71		2	ΔHv/T	19.38	5	0'	İ	-0.0 <sub>6</sub> 40	4
	+	0.71		-	d   25 to	<del> </del>	4		ace tension		
a b	-	0.73 -0.0 <sub>3</sub>		4 4	<u>•   130 °C</u>	0.1280	4	dyne	s/cm, 20°C 30	21.52 20.58	2
Ref. Index					d' i to				40	19.65	2
n <sub>D</sub> 20°0		1.40		2	<b></b> _	0,254	2	Para	chor [P]		
25 30		1.40		2 4	d g/ml v ml/g	3.940	2		20°C 30		
"C"	+	0,74		4	t <sub>c</sub> ·C	295.	2		40		
MR (Obs.	+	38.83		-	P <sub>c</sub> mm	20824.	2		Sugd.	346.2	5
MR (Calc.		39.14		2 5	PV/RT	1 0000		Exp.	L.1.%/wt.		
(nD-d/2)	$\perp$	1.04	435	2	25°C 30 mm	1.0000 1.0000	5 5	Dist	u. ersion	96.1	2
Dielectric		1.97	1	5	BP	0.9618	4		h Point °C	3.	5
A 25 t		6.86		2	t <sub>e</sub>	0.9524 0.264	5 2		Point		
B 1_150 °		215.30		2 2	t <sub>c</sub>	1222.11	2	M S	pec.		
A*   25 to	_	1, 29		5	ΔHf	-59.69	2	Ultr.	a V. ay Dif.		
B* 140°		231.73		5	ΔFf	3.03	2	Infra		Yes	2
K C					Viscosity centistokes			Solu	bility in +		
the Te					7 °C	ľ			etone	<b>60</b>	
t <sub>x</sub>					•		1 1		rbon tet. nzene	80 80	
A'   0 to		8.55		4				Eth		<b>60</b>	[
C 30 -		278, 23 292, 31		4	B <sup>V</sup>   to				Heptane anol	80 80	l
A'* 10 to	_	2.88		4	A <sup>V</sup> I °C			Wa	ter		
B'* 30 °	C 2	132.76		4	(B <sup>V</sup> ) to	1		Wa	ter in		-
Ac   150 to	2   .	7.40	735	5	(A <sup>V</sup> )  °C	<u> </u>					
Bc Ltc_	- 1	749.0 271.2		5	c <sub>p</sub> liq. 300°K	0.39891	5				l
Cryos, A'	•	0.05	44	2	c <sub>p</sub> vap.300°K	0.50224					
t <sub>e</sub> °C	+	128 25		5	c <sub>v</sub> vap.	0.50217	2				
$T_R = 0.7$		128, 25		لــًـا	L	L	$\sqcup \sqcup$	L	10.00	<u> </u>	L
REFEREN				2 4 -	OT 2 124 4 4	2-1- 6			ms/100 gran		<u> </u>
SOURCE:	Ų EX	s. 1-D	<b>∨₩</b>	Z-AF		Calc. from det	. dat	a 5-	Calc. by for	mula	
		N.		AF	<del></del>						
PURIFICA			- n	AI							
21121411	, KL	KEFI		1CES	•						

TABLE C. ALKANES

No. 35

45

NAME	3-	Ethyl-	3-met	thylpentane			ST	RUCTURAL	FORMUL		
				· · · · · · · · · · · · · · · · · · ·			CH <sub>3</sub>				
Mole		-( )/-	.1 1		Mala - ula m		(	сн <sub>3</sub> сн <sub>2</sub> с	CH <sub>2</sub> CH <sub>3</sub>		
% Pur. 99	. 995	ef. Mo 2 Fo	rmul		Molecular Weight 114.2	24		C <sub>2</sub> H <sub>5</sub>			
			Ref.			Ref.				Ref.	
F. P. °C		. 870	2	dt/dP			f	300 to	0.0445		
B. P. °C	•		+-	°C/mm 25°C	0.8459	4	g	'_6 <u>0</u> 0_° <u>K</u>	ł	1 1	
760 mm		. 259	2	BP	0.04844 0.0370	2 5	h f'	to	-0.0637	-	
100 30		. 103 . 25	2 4	t <sub>e</sub> 30 mm	0.6713	4	g'	<u>• K</u>			
10 1	-25	. 9	2 4	ΔHm cal/g		$\vdash$	h'				
Pressure	+	· <i>'</i> —	<del>                                     </del>	ΔHv cal/g			m n	300 to	0.0265 0.0014		
mm 25°C		. 01	5	25°C 30 mm	79.49 79.55	2 4	0	<u></u>	-0.0646	4	
Density	1072	. ′	5	BP	69.30	2 5	m'	700 to	0.1079	4	
g/m1 20°		. 72742	2	t <sub>e</sub> (d, e)	67.75 67.76	5	n' o'	[10 <u>0</u> 0 <b>∘</b> K		4	
d <sub>4</sub> 25		. 72354 . 71965	2 4	AHv/T <sub>e</sub>	19.12	5			-0.0 <sub>6</sub> 40	ļ-	
a		. 74293	4	d 30 to	83.08	5		face tension es/cm. 20°C	21.99	2	
Ъ	_	. 0377	4	d' 130 °C to	0.1165	5	8	30 40	21.07 20.15	2 2	
Ref. Inde	- 1	. 40775	2	e'   °C			Par	achor [P]	20.15	-	
D 25	1	. 40549 . 40316	2	d g/ml vc ml/g	0.263 3.808	2 2		20°C			
"C"	+	. 7466	4	tc °C	305.	2		30 <b>4</b> 0			
MR (Obs.		.717	2	P <sub>c</sub> mm	21964.	2			346.2	5	
MR (Calc. (nD-d/2)		. 144 . 04404	5 2	PV/RT 25°C	1.0000	5	Exp	u. L.1.%/wt.			
Dielectric		. 982	5	30 mm BP	1.0000 0.9625	5 4		persion	95.8	2	
A 30 to	-	. 86731	2	t_	0.9528	5		sh Point C e Point	6.	5	
B 1760 .c		. 209 . 684	2 2	tc ΔHc kcal/m	0.265	2	M.	Spec.			
A* 30 to	<del></del>	. 28717	5	ΔHf	-60.46	2		ra V. Ray Dif.			
B*[140 °C	1257	. 01	5	ΔFf Viscosity	2.69	2		ared	Yes	2	
c	_			centistokes				ability in +	<b>∞</b>		
t <sub>k</sub> to				າ °⊂			Ca	rbon tet.	oc		
A'   0 to		. 17016	5					nzene her	ος ος		
B'   30 °C		. 18 . 096	5	B <sup>v</sup> to A <sup>v</sup> °C				Heptane hanol	σο σο		
A'* 10 to	2	.51548	5				Wa	ter	~		
B'* 30 °C	+		5	(B <sup>V</sup> )  to			- W.	ter in		┼	
Ac 160 to	1864	. 49210 . <b>1</b>	5	(A <sup>V</sup> )  °C	0.30001	F					
Ce	286	. 7	5	c <sub>p</sub> liq. 300°K 400	0.39891 0.50224	5					
Cryos. A' consts. B'		. 0392	2	c <sub>p</sub> vap.300°K 400	0.39703 0.50217	2					
t <sub>e</sub> °C	131	. 44	5	c <sub>v</sub> vap.				· · · · · · · · · · · · · · · · · · ·			
$T_R = 0$ .								ams/100 gra		it	
REFEREN	CES:	-Dow		PI 3-Lit. 4-	Calc, from de	t. da	ta 5	-Calc, by for	mula		
SOURCE: PURIFICA	TION·			.PI							
LITERATI		EFERE									

								No. 36	
NAME	2, 2, 3-Tri	meth	vlpentane			STRI	JCTURAL I		`
NAME -			, - p		$\dashv$	01111	сн <sub>3</sub> сн <sub>3</sub>	. 01(1110 -1	•
							:н <sub>3</sub> с - сн	сн.сн.	
Mole	Ref. Mo	lecul		Molecular		`	- •	20113	
% Pur. 99.		rmul		Weight 114.22	24		CH <sub>3</sub>		
		Ref.			Ref				Ref.
F.P. *C	-112,27	2	dt/dP			f	300to	0.0445	5
F.P. 100%	1-11-11-1	-	*C/mm	}	] ]	g !	600°K	0.0013	
B, P. °C	<u> </u>	$\vdash$	25°C	0.6199	4	h !		-0.0637	
760 mm	109.841	2	BP	0.04755 0.0372	5	<u>f'</u> +	to	-	
100 30	49.922 23.69	2	t <sub>e</sub> 30 mm	0,6553	4	g'	•K		]
10	3.9	2		0.0333	+	h'			
11	-28.7	4	ΔHm cal/g		$\vdash$	m l	300 to	0.0265	4
Pressure	22.00	١.,	ΔHv cal/g 25°C	77.24	2	n ;	_600°K_	0.0014	4
mm 25°C	32.06 1043.0	5	30 mm	78.00	4	°i		-0.0646	4
Density		Ť	BP	67.30 65.84	2 5	m'	700 to	0.1079	4
g/ml 20°C	0.71602	2	te te (d, e)	65.83	5	n'	1 <u>1</u> 0 <u>0</u> 0°K	0.0012	4
dt 25 4 30	0.71207	2	ΔHv/T	19.05	5	0'		-0.0 <sub>6</sub> 40	4
	0.70811	4	d   25 to	80.94	4		ce tension		
a b	0.73180 -0.0 <sub>3</sub> 781	4	e   120 °C		4	dyne	30°C 30°C	20.67 19.77	5
Ref. Index	0.03.02	H	G' 6	ĺ		1	40	18.87	5
n <sub>D</sub> 20°C	1.40295	2		0.2(1		Para	chor [P]		
- 25	1.40066	2	d g/ml v ml/g	0.261 3.826	2 2		20°C		
30	1.39811	4	tc °C	294.	Ž		<b>30</b> 40		İ
"C"	0.7500	4	P <sub>c</sub> mm	21432.	2		Sugd.	346.2	5
MR (Obs.) MR (Calc.)	38.925	5	PV/RT		Н	Exp.	L.1.%/wt.		
(nD-d/2)	39.144 1.04494	2	25°C	0.9925	5	_	u.		
Dielectric	1.968	5	30 mm BP	1.0000 0.9585	5 4		rsion	97.2	2
A 20 to	<del></del>	2	te	0.9493	5		Point °C		
B 1150 ℃	1294.875	2	t <sub>c</sub>	0.265	2	M Sp			├
С	218.420	2	ΔHc kcal/m ΔHf	1219.98 -61.44	2 2	Ultra			ı
A* 20 to	1.26422	5	ΔFf	2, 22	2		y Dif.		
B* ∟130 °C	1209. 78	5	Viscosity		$\vdash$	Infra		Yes	2
c	_		centistokes	ļ		Solub Ace	ility in +		
tk   10° C	i i		η ∘c				on tet.	<b>&amp;</b>	ĺ
A' to	<del> </del>	-		İ		Ben		œ	
B'						Ethe n-He	er eptane	80	
C'			B <sup>V</sup>   to	ļ		Etha	nol	<b>∞</b>	l
A'* to	1		AV C	1		Wate	er in		
B'* °C	<del> </del>		(B <sup>V</sup> ) to	1					
Ac   150 to Bc   t <sub>c</sub> °C	7.29189	5	(A <sup>V</sup> )  °C						
Cc Cc	267.9	5	c <sub>p</sub> liq. 300°K 400	0.39891 0.50224	5 5				
Cryos. A°	0.0401	2	c <sub>p</sub> vap.300°K	0.39703	2				
consts. B°			400	0.50217	2				
t <sub>e</sub> °C	121.69	5	c <sub>v</sub> vap.						
$T_{R} = 0.75$	Tc					† gran	ms/100 gran	ns solven	t
	ES: 1-Dow	2-AF		alc, from det	dat	ta 5-0	Calc. by form	nula	
SOURCE:		AF	PI						
PURIFICAT		AI							
LITERATU	RE REFERE	NCES	3:						

TABLE C. ALKANES

No. 37

47

								No. 37	
NAME	2, 2, 4-Tri	neth	ylpentane			ST	RUCTURAL	FORMUL	A
	Isooctane							CH <sub>3</sub>	
-			T		$\neg$		CH3C CH2	CH CH <sub>3</sub>	
Mole % Pur.		ecula mula	ar C <sub>8</sub> H <sub>18</sub>	Molecular Veight 114.22	4		Ċн <sub>3</sub>		
70 I UI.	1 1 1 1 1 1 1	Ref.		vergnt 111.55	Ref.	ī			Ref.
F. P. °C	-107.380	2	dt/dP		1		1		
F.P. 100%	101,300	Ť	°C/mm			f g	to K		
B. P. °C			25°C	0.4226	4	h			
760 mm 100	99.238	2	BP t <sub>e</sub>	0.04651 0.03732	2 5	f'	to		
30	40.667 15.05	2 4	30 mm	0,6398	4	g'	° <u>K</u>		
10	-4.3	2	ΔHm cal/g	0,00,0		h'			
1	-36.1	4	ΔHv cal/g	<del></del>		m	300 to	0.0265	4
Pressure mm 25°C	49.34	4	25°C	73.50	2	n o	_6 <u>0</u> 0_• <u>K</u>	0.0014 -0.0 <sub>6</sub> 46	4 4
t <sub>e</sub>	1009.4	5	30 mm BP	75.31 64.87	4 2				-
Density			t,	63.60	5	m'	700 to	0.1079 0.0012	4
g/ml 20°C	0.69192 0.68777	2 2	t <sub>e</sub> (a, e)	63.59	5	٠, اه	1000 K	-0.0640	4
d <sub>4</sub> 30	0.68361	4	ΔHv/T <sub>e</sub>	18.98	5		f		
a	0.70850	4	d 15 to e 110 °C	77.17	5		face tension es/cm. 20°C	18.77	2
b	-0.03817	4	-å-  → 10 °C to	0.1240	5	8	30	17.88	2
Ref. Index	1.39145	2	e' °C				40	16.99	2
<sup>n</sup> D 20°C	1.38901	2	d <sub>c</sub> g/ml	0.237	2 2	Par	achor [P] 20°C		
30	1.38646	5	v <sub>c</sub> ml/g t <sub>c</sub> °C	4,220 271,15	2		30		
"C"	0.7552	4	P <sub>c</sub> mm	19380.	2		40 Sugd.	346. 2	5
MR (Obs.)	39. 262	2	PV/RT			Exp	. L.1.%/wt.		-
MR (Calc.)   (nD-d/2)	39.144 1.04549	5 2	25°C	0.9910	4	-	u.		
Dielectric	1.936	5	30 mm BP	1.0000 0.9562	5		persion	100.5	2
A 15 to	6,81189	2	t t	0.9478	5		sh Point C e Point	-12.	3
B (_135 ℃ C		2		0, 275	2	<u> </u>	Spec.	Yes	1
A* 15 to	220, 735	5	ΔHc kcal/m ΔHf	1219.01 -61.97	2	Ult	ra V.		-
B* 120 °C	1.26590 1175.54	5	ΔFf	1.65	2		Ray Dif. ared	1772.	ı
к			Viscosity			<b> </b>	ability in +		┢┈
t <sub>k</sub> to			centistokes 7 20 °C	0.7259	1	Ac	etone	∞	
t <sub>x</sub> °C			40	0.5958	1		rbon tet.	00 00	
A' to			60 80	0.4999 0.4270	1 1	Et	her	•	
B'°C_			B 30 to	400.09	4		Heptane hanol	∞ ∞	
A¹* to			B 1 30 to A 1 90 °C	Z. 49766	4	Wa	ater		
B'* °C			(B <sup>V</sup> )  to			W <sub>4</sub>	ter in		-
Acl 135 to Bc t °C	7.27905 1612.2	5	(A <sup>V</sup> )  °C						
Bc tc C	267.7	5	c <sub>p</sub> liq. °K						
Cryos. A°	0.04031	2	c <sub>p</sub> vap.300°K	0.39703	2				
consts. B°	0.0043	2	- 4±00	0.50217	2				
t <sub>e</sub> °C	109.60	5	c <sub>v</sub> vap.			<u> </u>		l	
$T_{\mathbf{R}} = 0.75$	Tc					† gı	ams/100 gra	ms solven	t
	ES: 1-Dow		PI 3-Lit. 4-0	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			.PI						
PURIFICAT			.PI						
LITERATU	RE REFERE	NCES	5: 3 NFPA 325	5					
L									

										<b>No.</b> 38	
NAME	2	3 3.	Tri	nethi	ylpentane		$\neg \tau$	STE	UCTURAL I		
NAME		, , , ,						311		ORMODA	•
									CH3CH3		
Mole		B-6		11		Molecular	- 1	,	сн <sub>3</sub> сн с	H <sub>2</sub> CH <sub>3</sub>	
% Pur. 99	. 87	Ref.		lecul rmul		Molecular Weight 114.2	24		CH <sub>3</sub>		
				Ref.	<del></del>		Ref	Γ			Ref.
F. P. *C	1.1	00.70		2	46/45	1		<del></del>	1 2004	0.0445	<del>                                     </del>
F.P. 100%		00.10		۲	dt/dP *C/mm		i i	f g	300 to	0.0445	5
B. P. °C	+			t	25°C	0.7357	4	h		-0.0637	5
760 mm		14.76		2	BP	0.04833 0.0373	5	<u>r</u>	l to	0.065.	<del>  -</del>
100 30		53.81 27.10		2 4	t <sub>e</sub> 30 mm	0.6676	4	g	l ¹ °K		
10	1	6.9		2		0.8678	+	h'	 I		
1	-:	33.15		4	ΔHm cal/g	<u> </u>	$\vdash$	m	300 to	0.0265	4
Pressure	1.				ΔHv cal/g 25°C	77.87	2	n	_600°K	0.0014	4
mm 25°C		27.00 60.3		5	30 mm	78,34	4	٥	i	-0.0 <sub>6</sub> 46	4
Density	+			Ť	BP	68.10	5	m'	700 to	0,1079	4
g/ml 20°C	:	0.72	619	2	te te (d, e)	66.62	5	n'	1000°K	0.0012	4
dt 25	ı	0.72		2	ΔHv/T	18.99	5	٥'	i	-0.0 <sub>6</sub> 40	4
	+	0.71		4	d   25 to	81,50	5		ace tension		
a b	1.	0.74 -0.03		4	e   125 °C	0.1168	5	dyne	s/cm. 20°C	21.56 20.65	2 2
Ref. Index	+			+	a' to			•	40	19.75	2
n <sub>D</sub> 20°C		1.40		2		<del></del>		Par	chor [P]		
25 30		1.40		2 4	d g/ml vc ml/g t °C	0.264 3.791	2 2		20°C		
"C"	+	1.40		-	tc °C	303.	2	ł	30 40		
	+-	0.74		4	P <sub>c</sub> mm	22040.	2	İ		346. 2	5
MR (Obs.) MR (Calc.		38.76 39.14		2 5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	Ί .	1.04		2	25°C 30 mm	1.0002 1.0000	5 5	n, .	u.		١.
Dielectric		1.98	1	5	BP	0.9607	4		ersion h Point °C	96.1	2
A 25 to		6.84		2	t <sub>e</sub> t <sub>c</sub>	0.9512 0.266	5 2		Point		
B 1_160 °C		28.04 20.37		2 2	ΔHc kcal/m	1220.86	2	M S	pec.		
A*   25 to	_	1.27		5	ΔHf	-60.63	2	Ultr Y-R	a V. ay Dif.		
B* 140 °C	12:	39.27		5	ΔFf	2.54	2	Infra		Yes	2
K C	1			1	Viscosity centistokes			Solu	bility in +		
t <sub>k</sub> to					7 °c	ļ			tone	<b>60</b>	
'X '								Be	nzene	ec ec	
A'   to							1 1	Eth		<b>60</b>	
c,	-				B <sup>v</sup>   to				leptane anol	ec ec	
A'* to					A <sup>V</sup>   °C			Wa			
B'* °C				$oxed{oxed}$	(B <sup>V</sup> ) to	1		⊢ Wa	ter in		├
Ac   160 to	177	7.38	917	5	(A <sup>V</sup> )  °C			Ì			
Cc _ c		79.5		5	c <sub>p</sub> liq. 300°K	0.39891	-5	l			ļ
Cryos, A°	1	0.00	62	2	400	0.50224 0.39703	5 2				
consts. B°	-				400	0.50217	2	1			
t <sub>e</sub> °C T <sub>R</sub> = 0.7		27.46		5	c <sub>v</sub> vap.	L		<u> </u>			<u> </u>
REFEREN		1 0		2 4 -	OT 2 124 4 4	7-1- 6			ms/100 gran		<u> </u>
SOURCE:	, EG:	1-10	- W	2-AF		Calc. from det	. dat	ia 5-	Calc. by form	nula	
	CION				· · · · · · · · · · · · · · · · · · ·						
PURIFICA:			- P - P	AF							
LILERAIU	r.e.	ALF1	in Ei	NUES	•						

Ref.

-109.210

113.467

27.06

7. 1

-25.65

27.01

1051.

53.404

NAME

Mole % Pur. 99.89

F.P. °C F.P. 100%

B. P. °C 760 mm

100

30

10

1

Density

Pressure

mm 25°C

te

2, 3, 4-Trimethylpentane

Molecular

Ref.

٠2

2

4

4

4

5

Formula

 $C_8H_{18}$ 

°C/mm

30 mm

∆Hm cal/g

AHv cal/g

30 mm

25°C

BP

0.7192

0.6585

79.52 79.39

68.37

66.45

dt/dP

25°C

BР

49 No. 39 STRUCTURAL FORMULA сн3сн3сн3 сн,сн сн сн сн, Molecular Weight 114,224 Ref. Ref. f to °Κ g 4 h 0.04761 0.0370 2 ſ١ to °<u>K</u> g' 4 h' 300 to 0.0265 m 0.0014 | 600 °K n 5 5 2 5 5 0 -0,-0646 4 700 to m' 0.1079 4 0.0012 n' |

Density g/ml 20°C	0. 71906	2	t <sub>e</sub> (d, e)	66. <b>4</b> 5 66. 82	5	n 1000 K	0.0012	4
dt 25	0.71503 0.71099	2 4	AHV/Te	19.04	5		-0.0640	4
a b Ref. Index	0.73517 -0.0 <sub>3</sub> 80	4	d 20 to e 120 °C d' to e' °C	82.84 0.1276	5 5	Surface tension dynes/cm. 20°C 30 40	21.14 20.22 19.31	2 2 2
n <sub>D</sub> 20°C 25 30	1.40422 1.40198 1.39931	2 2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.256 3.913 295.	2 2 2	Parachor [P] 20°C 30 40		
"C"	0, 7491	4		20976.	2		346.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	38.868 39.144 1.04469	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5	Exp. L.1.%/wt. u. Dispersion	96.9	2
Dielectric	1.972	5	BP	0.9566	4	Flash Point C	2.	5
	6.85396 1315.084	2	te t <sub>c</sub>	0.9470 0.265	5 2	Fire Point		
A* 25 to B* 135 °C	217.526 1.29238 1230.23	2 5 5	ΔHc kcal/m ΔHf ΔFf	1220.61 -60.98 2.54	2 2 2	M. Spec. Ultra V. X-Ray Dif. Infrared	Yes	2
K			Viscosity centistokes n °C			Solubility in + Acetone Carbon tet. Benzene Ether	80 80 80	_
A'* to B'* °C			$ \begin{array}{c c} B^{\mathbf{v}} & \text{to} \\ A^{\mathbf{v}} & & {}^{\circ}C \\ \hline (B^{\mathbf{v}}) & & \text{to} \end{array} $			n-Heptane Ethanol Water Water in	80	
Acl 150 to	7.33497 1695.6 267.7	5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K					
Cryos, A° consts. B°	0.04147 0.0035	2 2	c <sub>p</sub> vap.300°K 400 c <sub>p</sub> vap.	0.39703 0.50217	2 2		,	
t <sub>e</sub> °C	125.62	5	-VP.			L		<u> </u>
REFERENCE		2-A	PI 3-Lit. 4-0	Calc, from de	+ da	grams/100 gra		<u>t                                     </u>
SOURCE:			PI	,, ue		3-0aic. by 101		
PURIFICATION	ON:	A	PI					
LITERATUR		NCES	3:					
						***************************************		

						<b>No.</b> 40
NAME	2, 2, 3, 3-	Tetra	methylbutane			STRUCTURAL FORMULA
					$\neg$	CH <sub>3</sub> CH <sub>3</sub>
	<del></del>				$\dashv$	сн <sub>3</sub> с - с - сн <sub>3</sub>
Mole	Ref. Mo	lecul	ar c u	Molecular		ċн¸ċн¸
% Pur.		rmul	C <sub>8</sub> H <sub>18</sub>	Weight 114.2	24	33
		Ref.			Ref	Re
F.P. *C	+100.69	2	dt/dP			f to
F.P. 100%	<u> </u>		*C/mm 25*C	0.8149	4	g <u>°K</u> _
B. P. °C 760 mm	106.47	2	BP	0.0476	3	h
100	54.70₹	3	t <sub>e</sub>	0.0373	5	f' to
30 10	31.33 <sup>≠</sup> 13.09 <sup>≠</sup>	2	30 mm	0.6604	4	∬ g'
1	-18.03≠	3	∆Hm cal/g			h'
Pressure		+	ΔHv cal/g	#		m   300 to   -0.0414 4 n   600 °K   0.0017 4
mm 25°C	20.854	3	25°C 30 mm	89.36# 75.46	2 4	-0.0675 4
t <sub>e</sub>	1035.	5	BP	66.2	2	<b>⊩</b> ————————————————————————————————————
Density g/ml 23°C	0.8242	31	t <sub>e</sub>	65.02	5	m'   700 to   0.1205 4 n'   1000 °K   0.0013 4
dt 25	0.8215	31	• (4,6)	64.92	1 1	0'0.0646 4
<sup>4</sup> 4 30	0.8188	3'	ΔHv/T <sub>e</sub>	19.80	5	Surface tension
	0.8352*	4	d   20 to e   120 °C	78.03 0.1113	5	dynes/cm. 20°C   21.14   2
ь	-0.0354	4	_a'to			30 20.22 2 40 19.31 2
Ref. Index		31	•' i •C	<del> </del>		Parachor [P]
D 25	1.10/3		d g/ml v ml/g	0.239 4.189	2 2	20°C
30			t <sub>c</sub> *C	270.8	2	30
"C"			P <sub>c</sub> mm	18620.	3	40 Sugd. 346. 2 5
MR (Obs.)		ا ـ ا	PV/RT	<del></del>		Exp. L.1.%/wt.
MR (Calc. (nD-d/2)	39.144	5	25°C	1.0000	5	u.
Dielectric	<u> </u>		30 mm BP	1.0000 0.9599	5 4	Dispersion
A 101 to	6, 87665	2	te	0.9515	5	Flash Point °C Fire Point
B 1160 °C	1327.8	2	t <sub>c</sub>	0.262	4	M Spec.
<u> </u>	226.0	2	ΔHc kcal/m ΔHf	1218.59 -64.23	2 2	Ultra V.
A* 101 to B* 135 °C		5	ΔFf	3.13	2	X-Ray Dif.
K 1 2 2	- 1250.0		Viscosity			Infrared Solubility in +
tto	-{		centistokes			Solubility in + Acetone
tk   to			∥າ °⊂			Carbon tet.
A'   -20 to	7.92864	3				Benzene Ether
B' [101 °C	1709.428	3	B <sup>V</sup> l to			n-Heptane
C'	233.634	3	B <sup>V</sup> to			Ethanol Water
A'* -20 to B'* 101 °C		5		1		Water in
Ac   160 to		5	, v.			
Bcit C	3213.7	5	<u>-</u>	0.5893#	2	4   1
Cc	438.9	5	c <sub>p</sub> liq. 20°C	0.5893"	3	
Cryos. A° consts. B°	0.00613	2	c <sub>p</sub> vap.300°K 400	0.39703 0.50217	2 2	
t <sub>e</sub> °C	117.83	5	c <sub>w</sub> vap.			
solid state	#104.9°C	# he	at of sublimation	on $T_R = 0.79$	5 T <sub>C</sub>	+ grams/100 grams solvent
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-Calc, by formula
SOURCE:			PI, Lit.			
PURIFICAT	TION:	ΑI	PI, Lit.			
LITERATU	RE REFERE	NCES	: 3 JACS 74,	883 (1952) D.	W.	Scott et al; 3' JACS 71,
3447 (1949)	Wm. F. Seye	er et	al			_

								No. 41	
NAME _	n-Nonane					ST	RUCTURAL	FORMUL.	A
							CH <sub>3</sub> (CH <sub>2</sub> ).	CH.	
Mole % Pur. 99. 9	97   Ref. Mo. 1	ecul mula		Molecular Veight 128.25	50			, 5113	
	1	Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-53.519	2	dt/dP °C/mm 25°C	2 ((0	۰	f g	300 to _600_°K	0.0496 0.0013	5 5
B.P. °C 760 mm 100 30	150.798 87.899 60.16	2 2 4	BP t <sub>e</sub> 30 mm	3.660 0.04967 0.0365 0.6943	5 2 5 4	f' g'	to *K	-0.0 <sub>6</sub> 35	5
10 1	39.0 3.6	2	ΔHm cal/g	28.83	3 <b>2</b>	h'			
Pressure mm 25°C t <sub>e</sub>	4.351 1127.4	5	AHv cal/g 25°C 30 mm BP	86.54 82.67 68.44	2 4 4	m n o	600 •K	0.0294 0.0015 -0.0 <sub>6</sub> 52	4 4
Density g/ml 20°C dt 25 4 30	0.71763 0.71381 0.70999	2 2 4	te te (d, e) AHv/Te	66. 42 66. 00 19. 38	5 5	m' n' o'	700 to	0.1222 0.0013 -0.0 <sub>6</sub> 45	4 4
a b Ref. Index	0. 73290 -0. 0 <sub>3</sub> 762	4 4	d 60 to e 165 °C d' 10 to e' 60 °C	92.11 0.1570 89.29 0.1102	4 4 4		face tension es/cm, 20°C 30 40	22.92 21.95 21.00	2 2 2
<sup>n</sup> D 20°C 25 30	1.40542 1.40311 1.39984	2 2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.244 4.094 322.	2 2 2	Par	20°C		
"C"	0.7509	4	P <sub>c</sub> mm	17279.	2	ľ	40 Sugd.	385.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	43.842 43.762 1.04660	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5	1	L.1.%/wt. u. persion	3.2 11.8 97.2	3' 3' 2
A 60 to	6.93513	2	BP t e t c	0.9380 0.9482 0.250	5 2		sh Point C e Point	31.0	31
B   185 °C C	1428.811 201.619 1.43093	2 2	ΔHc kcal/m ΔHf	1369.70	2	Ult	Spec.	Yes	1
B*  180 °C K	1351.7 24.	4 5	ΔFf Viscosity			Infr	Ray Dif. ared ubility in	Yes	2
t <sub>k</sub> 180 to t <sub>x</sub> 270 °C	-0.15644 227. 381.	5 5 5	7 80 °C 100 120	0.5502 0.4743 0.4151	2 2 2	A c	cetone rbon tet.	80 80	1 1 1
A'   20 to B'   60 °C	7.30801 1630.71 219.5	5 5 5	140 BV 30 to	0, 3675 437, 7	2	n-	her Heptane hanol	8 8 8	1 1 1
A'* 20 to B'* 60 °C	1.76177 1536.61	5 5	A <sup>V</sup>   90 °C   (B <sup>V</sup> )  90 to	2.50135 427.1	4		ater ater in		
Acl 185 to Bc t <sub>c</sub> °C Cc —	7,43583 1842,3 254.0	5 5 5	(A <sup>V</sup> ) 150 °C c <sub>p</sub> liq. 300°K 400	2.53159 0.39979 0.50256	4 5 5		cosity tistokes 40°C	0.7921 0.6515	2 2
Cryos. A° consts. B°	0.03856	3 <sup>2</sup>	c vap.300K 400	0.39649 0.50136	2		60 110 150	0.4430 0.3473	2 2
t <sub>e</sub> °C T <sub>R</sub> = 0.77	166.351 T-	5	c <sub>w</sub> vap.	L	لــــا	L	(100		<u> </u>
	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da		rams/100 gra -Calc. by for		<u> </u>
SOURCE:		AI		uc			by 101		
PURIFICAT	ION:		PI						
LITERATUE	RE REFEREI	CES	S: 3 NBS Circ.	514; 3' NFF	A C	irc.;	3 <sup>2</sup> JACS <u>76</u> ,	333 (1954)	,
Finke et al.									

2-Met  Ref.  143. 26 81. 0 53. 15 32. 2 -2. 5  6. 554 1142.  0. 713 0. 709 0. 709 1. 403 1. 400 1. 398 43. 762 1. 046 1. 969	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	ar C <sub>9</sub> H <sub>20</sub>	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	250 Ref. 52554 22425555444445555555	f   g   h   f'   g'   h   h'   m   n   o   o   o   Surfacdynes	CTURAL  CH3 CH3CH (CH  300 to 600 °K  to 600 °K  700 to 1000 °K  700 to 1000 °K  ce tension /cm. 20°C 30 40 thor [P] 20°C 30 40	2) <sub>5</sub> CH <sub>3</sub>	Ref 5 5 5 5 4 4 4 4 4 4 4 4
-80. 4  143. 26 81. 0 53. 15 32. 2 -2. 5  6. 554 1142.  0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref.   2   2   2   4   2   5   5   5   6   4   4   6   6   4   6   6   6   6	a C9H20  dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	f   g   h   f' g'   h   n   n   n   n   n   n   n   n   n	300 to 600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 20°C 30°C 30°C 30°C 30°C 30°C 30°C 30°C 3	0.0496 0.0013 -0.0 <sub>6</sub> 35 0.0294 0.0015 -0.0 <sub>6</sub> 52 0.1222 0.0013 -0.0 <sub>6</sub> 45	5 5 5 4 4 4 4 4 4 2 2
-80. 4  143. 26 81. 0 53. 15 32. 2 -2. 5  6. 554 1142.  0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref.   2   2   2   4   2   5   5   5   6   4   4   6   6   4   6   6   6   6	a C9H20  dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	f   g   h   f' g'   h   n   n   n   n   n   n   n   n   n	300 to 600 °K to 600 °K 700 to 1000 °K 700 to 1000 °K 20°C 30°C 30°C 30°C 30°C 30°C 30°C 30°C 3	0.0496 0.0013 -0.0 <sub>6</sub> 35 0.0294 0.0015 -0.0 <sub>6</sub> 52 0.1222 0.0013 -0.0 <sub>6</sub> 45	5 5 5 4 4 4 4 4 4 2 2
-80. 4  143. 26 81. 0 53. 15 32. 2 -2. 5  6. 554 1142.  0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 43. 762 1. 046	Ref.   2   2   2   4   2   5   5   5   6   4   4   6   6   4   6   6   6   6	a C9H20  dt/dP	2.527 0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 4 2 5 5 4 4 4 4 4 5 5 5 5	g   h   f' g'   h'   m   n   o   o   m'   n'   o'   Surfacdynes	600 °K   to   600 °K   700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K   1000 to	0.0496 0.0013 -0.0635 0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	5 5 5 4 4 4 4 4 4 2 2
143, 26 81, 0 53, 15 32, 2 -2, 5 6, 554 1142. 0, 713 0, 709 0, 705 -0, 03 1, 400 1, 398 43, 762 1, 046	2 2 2 4 2 5 5 5 5 5 5 5 5 5 6 4 4 9 9 9 4 4 8 1 2 2 8 8 2 8 3 3 4 4 8 3 3 4 4 8 3 3 4 4 2 2 5 5 4 4 2 2 .	dt/dP	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	5 2 5 4 2 4 2 5 5 5 4 4 4 4 4 4 4	g   h   f' g'   h'   m   n   o   o   m'   n'   o'   Surfacdynes	600 °K   to   600 °K   700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K   1000 to	0.0496 0.0013 -0.0635 0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	5 5 5 4 4 4 4 4 4 2 2
143, 26 81, 0 53, 15 32, 2 -2, 5 6, 554 1142. 0, 713 0, 709 0, 705 -0, 03 1, 400 1, 398 43, 762 1, 046	2 2 2 4 2 5 5 5 5 5 5 5 5 5 5 5 5 5 5 6 4 4 6 6 6 6	*C/mm 25*C BP te 30 mm  AHm cal/g  AHv cal/g 25*C 30 mm BP te (d, e) AHv/Te d   55 to e   160 *C d'   10 to e'   55 *C  d g/ml v ml/g t_c *G P_c mm  PV/RT 25*C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	g   h   f' g'   h'   m   n   o   o   m'   n'   o'   Surfacdynes	600 °K   to   600 °K   700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K     700 to   1000 °K   1000 to	0.0013 -0.0 <sub>6</sub> 35 0.0294 0.0015 -0.0 <sub>6</sub> 52 0.1222 0.0013 -0.0 <sub>6</sub> 45	5 5 4 4 4 4 4 4 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	25°C BP t a 30 mm  AHm cal/g  AHv cal/g 25°C 30 mm BP te te(d,e) AHv/Te d   55 to e   160 °C d'   10 to e'   55 °C  d_c g/ml v_c ml/g t_c °C P_c mm  PV/RT 25°C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	h f' g'   h'   m   n   n   n'   n'   o'   o'   Surfactynes	700 to 1000 °K  700 to 1000 °K  ce tension /cm. 20°C 30 40  thor [P] 20°C	0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	4 4 4 4 4 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	BP te 30 mm  AHm cal/g  AHv cal/g 25°C  AHv cal/g 25°C  AHv/Te  d   55 to e   160 °C d'   10 to e'   55 °C	0.0494 0.0360 0.6895 83.2 79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216	2 5 4 2 4 2 5 5 5 5 4 4 4 4 4 5 5	f' g'   g'   h'   m   o   o   m'   n'   o'   Surfact dynes	300 to 600 °K  700 to 1000 °K  ce tension /cm. 20°C 30 40  thor [P] 20°C 30	0.0294 0.0015 -0.0652 0.1222 0.0013 -0.0645	4 4 4 4 4 2 2 2
81. 0 53.15 32. 2 -2. 5 6. 554 1142. 0. 713 0. 709 0. 709 -0. 037 1. 403 1. 400 1. 398 0. 753 43. 88 43. 762 1. 046	2 4 4 2 5 5 5 5 5 5 6 4 2 2 6 6 4 4 4 6 6 6 4 4 6 6 6 4 4 6 6 6 6	30 mm  AHm cal/g  AHv cal/g  25°C  30 mm  BP  te (d,e)  AHv/Te  d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v ml/g t_c °C P_ mm  PV/RT  25°C	83. 2 79. 78 68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216	2 4 2 5 5 5 4 4 4 4 4 5 5	g'   h'   m   n   o   m'   n'   o'   Surfact dynes	300 to 600 °K  700 to 1000 °K  ce tension /cm. 20°C 30 40  thor [P] 20°C 30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
32.2 -2.5 6.554 1142. 0.713 0.705 0.705 0.725 -0.037 1.400 1.398 43.762 1.046	2 5 5 5 5 5 5 5 5 5 6 6 4 6 6 6 6 6 6 6 6	ΔHm cal/g  ΔHv cal/g  25°C  30 mm  BP  te te (d,e)  ΔHv/Te  d   55 to e   160 °C d'   10 to e'   55 °C  d_c g/ml v_c ml/g t_c °C P_mm  PV/RT 25°C	83. 2 79.78 68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	2 4 2 5 5 5 4 4 4 4 5 5	h'   m   n   o   m'   n'   o'   Surface dynes	300 to   600 °K   700 to   1000 °K   ce tension   /cm. 20°C   30   40   chor [P]   20°C   30°	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
-2.5 6.554 1142. 0.713 0.705 0.705 0.725 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	5 5 5 5 5 5 6 4 2 6 6 4 4 778 4 4 8 1 1 2 2 3 3 4 4 2 3 3 4 4 4 4 4 4 4 4 4 4 4	ΔHv cal/g 25°C 30 mm BP te (d,e) ΔHv/Te d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v ml/g t c °C P mm PV/RT 25°C	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	m   n   o   o   m'   n'   o'   o'   Surfactives	700 to   1000 °K   700 to   1000 °K   700 to   20°C   30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
0.713 0.705 0.705 0.705 -0.037 1.400 1.398 0.753 43.88 43.762	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	25°C 30 mm BP  te te (d,e)  ΔHv/Te  d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v ml/g tc °C P mm  PV/RT 25°C	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	n   o   m'   n'   o'   Surfac dynes	700 to   1000 °K   700 to   1000 °K   700 to   20°C   30	0.0015 -0.0652 0.1222 0.0013 -0.0645 21.88 20.94	4 4 4 4 2 2
0.713 0.705 0.705 0.705 -0.037 1.400 1.398 0.753 43.88 43.762	5 34 2 255 2 66 4 90 4 4778 4 31 2 38 2 33 4 40 4 2 5 5 4 2.	30 mm BP  t t e (d, e)  ΔHv/T d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v <sub>c</sub> ml/g t <sub>c</sub> °G P <sub>c</sub> mm  PV/RT 25°C	79.78 68.3 66.31 66.26 19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	4 2 5 5 5 4 4 4 4 5 5	o   m'   n'   o'   Surfac dynes	700 to   1000 °K     1000 °K     20°C   30   40   20°C   3	-0.0 <sub>6</sub> 52 0.1222 0.0013 -0.0 <sub>6</sub> 45 21.88 20.94	4 4 4 2 2
0.713 0.705 0.705 0.725 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	34 2 255 4 90 4 4778 4 31 2 28 2 33 4 30 4 2 5 54 2.	BP te (d, e)  ΔHv/Te d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml vc ml/g tc °G Pc mm  PV/RT 25°C	68. 3 66. 31 66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	2 5 5 5 4 4 4 4 5 5	Surface dynes	1000 °K ce tension /cm. 20°C 30 40 chor [P] 20°C 30	0.1222 0.0013 -0.0 <sub>6</sub> 45 21.88 20.94	4 4 2 2
0.709 0.705 0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	25	AHv/T <sub>e</sub> d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C  P <sub>c</sub> mm  PV/RT 25°C	66. 26 19. 67 86. 55 0. 1274 86. 24 0. 1216 311. 0 16595.	5 5 4 4 4 4 5 5	Surface dynes	1000 °K ce tension /cm. 20°C 30 40 chor [P] 20°C 30	0.0013 -0.0 <sub>6</sub> 45 21.88 20.94	4 4 2 2
0.709 0.705 0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	25	AHv/T <sub>e</sub> d   55 to e   160 °C d'   10 to e'   55 °C  d g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C  P <sub>c</sub> mm  PV/RT 25°C	19.67 86.55 0.1274 86.24 0.1216 311.0 16595.	5 4 4 4 4 5 5	Surfac dynes y	/cm. 20°C 30 40 thor [P] 20°C 30	21.88 20.94	2 2
0.729 -0.037 1.403 1.400 1.398 0.753 43.88 43.762 1.046	90 4 1778 4 31 2 108 2 133 4 30 4 2 5 54 2.	d   55 to e   160 °C d'   10 to e'   55 °C  d_c g/ml v_c ml/g t_c °G P_ mm  PV/RT 25°C	86.55 0.1274 86.24 0.1216 311.0 16595.	4 4 4 4 5 5	dynes	/cm. 20°C 30 40 thor [P] 20°C 30	20.94	2
-0.0 <sub>3</sub> 7 1.403 1.400 1.398 0.753 43.88 43.762 1.046	778 4 31 2 98 2 33 4 30 4 2 5 64 2.	e   160 °C d'   10 to e'   55 °C d g/ml v ml/g t e G P mm PV/RT 25°C	0.1274 86.24 0.1216 311.0 16595.	4 4 4 5 5	dynes	/cm. 20°C 30 40 thor [P] 20°C 30	20.94	2
1.403 1.400 1.398 0.753 43.88 43.762 1.046	31 2 08 2 33 4 30 4 2 5 54 2.	d'   10 to e'   55 °C  d g/ml vc ml/g tc °G  P mm  PV/RT 25°C	86. 24 0. 1216 311. 0 16595.	4 4 5 5		40 hor [P] 20°C 30		
1.400 1.398 0.753 43.88 43.762 1.046	08 2 33 4 30 4 2 2 54 2.	d g/ml vc ml/g tc *G Pc mm PV/RT 25*C	311. <b>0</b> 16595.	5 5		hor [P] 20°C 30	20.00	۴
1.400 1.398 0.753 43.88 43.762 1.046	08 2 33 4 30 4 2 2 54 2.	t <sub>c</sub> *G  P <sub>c</sub> mm  PV/RT  25°C	16595. 1.00 <b>0</b> 0	5		20°C		1
0.753 43.88 43.762 1.046	30 4 2 2 5 4 2.	P <sub>C</sub> mm PV/RT 25°C	16595. 1.00 <b>0</b> 0	5			i i	i
43.88 43.762 1.046	2 5 5 4 2.	P <sub>c</sub> mm PV/RT 25°C	16595. 1.00 <b>0</b> 0	-		40		ĺ
43.762 1.046	2 5 54 2.	PV/RT 25°C	1.0000	-	<u> </u>	Sugd	385.2	5
1.046	4 2.	25°C		=	Evn	L. 1. %/wt.	0,74	3
<del>                                     </del>		30			· ·	u.	2.9	3
1.70.	) [5]	BP	1.0000 0.9610	5 4	Dispe		98.4	2
6, 917	-	t <sub>e</sub>	0.9495	5		Point °C	32.	5
1410.0	2	tc		oxdot	Fire l		<del> </del>	-
206.0	2	ΔHc kcal/m			M Spe Ultra			
1.374		ΔHf ΔFf			X-Ray	y Dif.	ļ	
1321.0	5	Viscosity		$\vdash$	Infrar		ļI	<u> </u>
1		centistokes					_	
		η °C		1 1			80	
7 097	14 5						<b>80</b>	İ
1501.9	5			$\perp$			80	ı
214.3	5				Etha	nol	<b>∞</b>	l
							1	ĺ
		v.						
1799.9	5			<u> </u>				ĺ
255.7	5				1			İ
		c <sub>D</sub> vap.300°K	0.39649	2	1			İ
<b></b>		400			1			l
<u> </u>	5	<b>v</b> ( <b>up</b> .15.0 C	0.501	لئا	L		l	<u> </u>
								<u> </u>
ES: 1-Do			alc. from det	t. dat	a 5-C	alc. by for	mula	
		~						
E REFE	RENCES	3 ASTM Spe	c. Tech. Pub	. No.	109			
]	1501. 9 214. 3 1.554 1412. 9 7.396 1799. 9 255. 7 159. 28 T <sub>c</sub> ES: 1-Do	214.3 5 1.5590 5 1412.9 5 7.39626 5 1799.9 5 255.7 5 159.28 5 T <sub>C</sub> ES: 1-Dow 2-AF ION: AF	7.0924 5 1501.9 5 214.3 5 1.5590 5 1412.9 5 7.39626 5 1799.9 255.7 5  159.28 5  T <sub>C</sub> ES: 1-Dow 2-API 3-Lit. 4-C  API  ION: API	7 °C     7 °C       7 °C	7 °C  7, 0924 5 1501, 9 5 214, 3 5  1, 5590 5 1412, 9 5  7, 39626 5 1799, 9 5 255, 7 5 6	7. 0924 5 1501. 9 214. 3 5 1. 5590 5 1412. 9 5 1799. 9 255. 7 5   C <sub>p</sub> liq. 300°K 0.39979 5 C <sub>p</sub> vap.300°K 0.39649 2 400 0.50136 2 C <sub>p</sub> vap.300°K 0.39649 2 400 0.50136 2 C <sub>p</sub> vap.300°K 0.381 3  T <sub>C</sub> ES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-C	Centistokes	Centistokes   Acetone   Carbon tet.   Sensene   Sensen

								No. 43	
NAME _	3-Meth	yloc tane				ST	RUCTURAL	FORMUL	A
					1		ÇН <sub>3</sub>		
<u> </u>						(	CH <sub>3</sub> CH <sub>2</sub> CH (C	н.).сн.	
Mole		Molecul		Molecular	.	•	30112011 (0	72/4	
% Pur.		Formula	7 20	Weight 128.25					
	1	Ref.			Ref.	<u> </u>	<del></del>		Ref.
F, P. °C	-107.6	2	dt/dP			f	300 to	0.0496	5
F.P. 100%	<u> </u>		°C/mm 25°C	2,595	5	g	'_6 <u>0</u> 0_• <u>K</u>		5
B. P. °C 760 mm	144.18	2	BP	0.04966	4	h_	l	-0.0635	5
100	81.	2	t <sub>e</sub>	0.0360	5	f'	to		
30	53.71	4	30 mm	0.6920	4	g'	<u>*K</u>		
10	33. -2.1	2 5	∆Hm cal/g			h'	<u> </u>		l
Pressure	<del> </del>	+-	ΔHv cal/g			m	300 to	0.0294 0.0015	4
mm 25°C	6.367	5	25°C	83.4	2	0	'_000_E	-0.0652	4
te	1149.4	5	30 mm BP	79.76 68.5	4 2	<u> </u>			<del>  -</del>
Density			t <sub>a</sub>	66.51	5	m' n'	700 to	0.1222	4
g/ml 20°C	0.720 0.716		t <sub>e</sub> (d, e)	66.47	5	٥'	<u>                                   </u>	-0.0645	4
d <sub>4</sub> 25 30	0.712		ΔHv/T <sub>e</sub>	19.67	5		<u> </u>	-	$\vdash$
a	0.736	3 4	d 50 to	86.45	4		face tension es/cm. 20°C	22.34	2
ь	-0.037		160 °C	0.1245 86.57	4 4	8'	30	21.40	2
Ref. Index			e' 50 °C	0.1267	4		40	20.46	2
n <sub>D</sub> 20°C	1.406		d <sub>c</sub> g/ml			Par	rachor [P] 20°C		
30	1.401		v_ml/g		_		30		
"C"	0,750	8 4	<u>_</u> c	313.8	5		40		_
MR (Obs.)	43.73	2	P <sub>c</sub> mm	16791.	5			385.2	5
MR (Calc.	43.762	5	PV/RT 25°C	1,0000	5	Exp	o. L.1.%/wt. u.		
(nD-d/2)	1.045		30 mm	1.0000	5	Dis	persion	97.5	2
Dielectric	1.977		BP	0.9638	4	Fla	sh Point C		
A 50 to B 195 °C	6.910 1411.0	2 2 2	te tc	0.9525	"	Fir	e Point		
c 1122	206.0	2	ΔHc kcal/m	<del> </del>			Spec.		
A* 50 to	1.360	1 5	ΔHf				ra V. Ray Dif.		1
B*  170 °C		5	ΔFf	ļ			ared		
K	1		Viscosity centistokes			Sol	ubility in +		
t <sub>k</sub> – tō	•		η °C			Ac	etone	oc	1
t⊈ °C			•				rbon tet. enzene	οο οο	
A'   20 to	7.041			1		Et	her	<b>00</b>	
B' _ 50 °C	1480.2	5 5	B <sub>v</sub> to				Heptane hanol	ος ος	Ì
A'* 20 to	1,510		ĂV I °C				ater	~	l
B' ≠ 50 °C		5	(B <sup>V</sup> )  to	1		W	ater in		
Acl 195 to	7. 388		(A <sup>V</sup> )  °C						
Bc tc °C	1803.0	5 5	c liq. 300°K	0.39979	5				
Cc	256.2		P 400	0.50256	5				
Cryos. A° consts. B°	1		c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2				
t <sub>e</sub> °C	160.52	5	c vap.	0.30130	-				
$T_R = 0.7$	L		L <u>.                                      </u>	<u> </u>	Li	٠.		<u> </u>	
		2 41	DT 2 144 4	Cala from 1	- د ه		Cala ha far		ıt
REFERENC	<u>, ຄວ: 1-Do</u>		PI 3-Lit. 4- PI	Calc, from de	t. da	ua 5	-Calc. by for	IIIUI <b>A</b>	
SOURCE:			<del></del>						
PURIFICAT			PI						
LITERATU	RE REFE	RENCES	<b>:</b> :						

								No. 44	<u> </u>
NAME	4 - Methyl	octane				STF	RUCTURAL	FORMULA	
<u> </u>					$\neg$		сн3		
					$\neg$	СН	<sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн (	CH <sub>2</sub> ),CH,	
Mole		lolecul		Molecular Weight 128.2	50		3. 2.2 .	2.3 3	
% Pur.	<u></u>	ormul	*	Weight 120.2					Ref.
E D 46	-113,2	Ref.		1	Ref	<u> </u>	1 1		-
F.P. *C F.P. 100%		+-	dt/dP *C/mm	1		f g	300 to 600 °K	0.0496 0.0013	
B. P. °C	<del></del>	+	25°C	2.445	5	h		-0.0635	1
760 mm 100	142.48	2 2	BP t <sub>e</sub>	0.049 0.0360	2 5	f'	to		
30	80. 52.53	4	30 mm	0,6882	4	g'			
10 1	32. -2.80	2 5	AHm cal/g			h'	l		
Pressure	-2.80	+3	ΔHv cal/g			m	300 to	0.0294	
mm 25°C	6.756	5	25°C 30 mm	83.4 79.63	2 4	n o	_600 •K	0.0015 -0.0 <sub>6</sub> 52	
t <sub>e</sub>	1141.1	5	BP	68.2	2	m'	700 to	0,1222	4
Density g/ml 20°C	0.7199	2	t <sub>e</sub> (d, e)	66.23	5	n'	1000 K	0.0013	4
at 25	0.7160	2	ΔHv/T <sub>e</sub>	19.68	5	0'	i	-0.0 <sub>6</sub> 45	4
	0.7121	4	d   55 to	<del> </del>	4		ace tension		
a b	0.7355 -0.0 <sub>3</sub> 778	5 5	e   160 °C	0.1271	4	dyne	s/cm, 20°C 30	22.34 21.40	2
Ref. Index		$\top$	d'   10 to		4 4	L	40	20.46	2
n <sub>D</sub> 20°C	1.4061	2 2	d <sub>c</sub> g/ml			Par	achor [P]		
30	1.4015	4	ll V mi/g	1	_		20°C 30		
"C"	0.7515	4	tc °C	311.0	5		40	305 3	_
MR (Obs.)		2	P <sub>c</sub> mm	16810.	,	- FC		385.2	5
MR (Calc. (nD-d/2)	1.0462	5 2	25°C	1.0000	5	Exp	. L.1.%/wt. u.		
Dielectric	1.977	5	30 mm BP	1.0000 0.9615	5 4	Disp	persion	97.5	2
A   50 to	<del></del>	2	t <sub>e</sub>	0.9502	5		sh Point °C Point		
B <u>[ 165 °C</u>	1406.0	2	tc			M S			
C	206.0	2	ΔHc kcal/m ΔHf			Ultr	a V.		
A* 50 to B* 170 °C		5	ΔFf				ay Dif. ared		
к — — –	-		Viscosity			<b> </b>	bility in +		
t <sub>k</sub>			centistokes 7°C			Ac	etone	∞	
t <sub>x</sub> • •			<b>'</b>				rbon tet. nzene	00 00	
A'   20 to B' <u>55</u> °C		5				Eth		oc	
C, L 33 7	208.5	5	B <sup>V</sup> to				Heptane nanol	00 00	
A!# 20 to		5	AV   °C	_		Wa	ter ter in		
	7 3010	5	(B <sup>V</sup> ) to			- wa	PET III		-
Ac   165 to	7.39191 1794.2	5	(A <sup>V</sup> )  °C	<del></del>					
Cc - c-	255.6	5	c <sub>p</sub> liq. 300°K 400	0.39979 0.50256	5				
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K	0.39649 0.50136	2				
t <sub>e</sub> °C	158.42	5	c <sub>w</sub> vap.	0.50130					
$T_{\mathbf{R}} = 0.7$				J	لــــــا	+ gra	ams/100 grai	ns solveni	t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. dat				
SOURCE:		AI	PI						
PURIFICA'		AI							
LITERATU	RE REFER	ENCES	5:						

								No. 45	
NAME	3-Ethylhe	ptane	•			ST	RUCTURAL	FORMUL	A
							C <sub>2</sub> H <sub>5</sub>		
Mole % Pur.		lecul		Molecular Weight 128.25	50	C	н <sub>3</sub> сн <sub>2</sub> сн (с	H <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	
		Ref.			Ref.				Ref.
F.P. °C F.P. 1009	<u>.</u>	├	dt/dP °C/mm			f	300 to	0.0496	5
B. P. °C		<del>                                     </del>	25°C	2.457	5	g h	'_6 <u>0</u> 0_ <b>°K</b>	0.0013 -0.0 <sub>6</sub> 35	5
760 mm 100	143.0	2 2	BP t <sub>e</sub>	0.0496 0.0359	5	f'	to	0.0633	Ť
30	80.3 52.7	4	30 mm	0.6904	4	g'	'° <u>K</u>		
10	32. -2.65	5	∆Hm cal/g			h'			L_
Pressure			ΔHv cal/g 25°C	02 5		m n	300 to   600 °K	0.0294 0.0015	4
mm 25°C	6.715 1152.5	5	30 mm	83.5 79.44	2 4	0		-0.0 <sub>6</sub> 52	4
Density	1	+-	BP t_	68.5 66.55	5	m'	700 to	0.1222	4
g/ml 20°0	0.727 0.723	2 2	te te (d, e)	66.51	5	n' o'	1000 •K	0.0013 -0.0 <sub>6</sub> 45	4
d <sub>4</sub> 25 30	0.719	4	е	19.73	5	Sur	face tension	0	
a b	0.743	4	d 50 to e 160 °C	85.82 0.1211	4 4		es/cm. 20°C	22.81	2
Ref. Index	-0.03798	4	d' 10 to	87.17	5	•	30 40	21.87 20.93	2
<sup>n</sup> D 20°C	1.4093	2	d <sub>c</sub> g/ml	0.1407	+-	Par	achor [P]		
30	1.4070 1.4045	2 4	v <sub>c</sub> mi/g	212 1	5		20°C 30		
"C"	0.7496	4	t <sub>c</sub> °C P <sub>c</sub> mm	313.1 16895.	5		40 Sugd.	385 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT	1	+	Exp	. L.1.%/wt.	303.2	<u> </u>
(nD-d/2)	1.046	2	25°C 30 mm	1.0000 1.0000	5		u. persion	96.6	,
Dielectric	+ -: 700	5	<b>B</b> P	0.9682	4		sh Point C	28.	5
A 50 to B 1195 °C		2	te t <sub>c</sub>	0.9575	5	Fir	e Point		Ĺ
c	206.	2	∆Hc kcal/m		t		Spec. ra V.		
A* 50 to B* 170 °C		5	ΔHf ΔFf			X-I	Ray Dif.		
к	-		Viscosity				ability in +		_
t <sub>k</sub> — tō			centistokes 7°C			Ac	etone	<b>∞</b> 0	ĺ
'x							rbon tet.	<b>0</b> 0	
A'  25 to B' _55 °C		5 <b>5</b>			$\perp$		her Hep <b>ta</b> ne	oc oc	
C'	205.	5	B <sup>V</sup>   to A <sup>V</sup>   °C			Et	hanol ater	œ	
A'* 25 to B'* 55 °C		5	$\frac{\mathbf{B}^{\mathbf{v}}}{(\mathbf{B}^{\mathbf{v}})} - \frac{\mathbf{v}}{to}$	-			ter in		
Ac 195 to	7 37800	5	(A <sup>V</sup> )  °C						
Bc t <sub>c</sub> °C	1793.5 - 256.2	5	c <sub>p</sub> liq. 300°K	0. 39979	5				
Cryos. A°		Ť	c vap.300°K	0.50256 0.39649	5 2	1			1
consts. B		ļ_		0.50136	2				
$t_e$ °C $T_R = 0.7$	159.42	5	c vap.	1	$oxed{oxed}$	L	(100		<u> </u>
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t de		ams/100 gra		t
SOURCE:			PI	11 om de	ua		by 101		
PURIFICA	TION:	A	PI						
LITERATU	RE REFERE	NCE	5:						

								No. 46	
NAME	4-Ethyll	heptane				STR	UCTURAL 1	FORMULA	
Ī					$\neg$		C2H5		
					$\neg$	CH	<sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> с́н <sup>3</sup>	(CH <sub>2</sub> ) <sub>2</sub> CH	
Mole % Pur.	Ref.	Molecul	аг С <sub>9</sub> Н <sub>20</sub>	Molecular Weight 128.2	50	•	5. 2.2	2.2	3
<del></del>		Ref	1 20 1	Weight 120.2	Ref				Ref.
F. P. *C	T	1	dt/dP	1		ſ	300 to	0.0496	
F.P. 100%			°C/mm			g	600°K_	0.0013	
B. P. *C			25°C BP	2.3002 0.050	5 2	h		-0.0635	5
760 mm 100	141.2 79.	2 2	t.	0.0358	5	f¹	to		
30	51.4	4	30 mm	0.6864	4	g'			
10 1	31.	2 5	ΔHm cal/g			h'			<u> </u>
Pressure	<b>†</b>	-	ΔHv cal/g			m n	300 to	0.0294 0.0015	
mm 25°C	7.174	5	25°C 30 mm	83.5 79.27	2 4	0		-0.0 <sub>6</sub> 52	
Te te	1147.7	5	BP	68.3	2	m'	700 to	0, 1222	4
Density g/ml 20°C	0,730	2	te te (d, e)	66.43	5	n'	11000°K	0.0013	4
dt 25	0.726	2	ΔHv/T	19.79	5	0'		-0.0 <sub>6</sub> 45	4
	0.722	4	d   50 to		4		ace tension		
a b	0.746 -0.0 <sub>3</sub> 8	4 4	_e <u>  155</u> °C	0.1221	4	dyne	s/cm. 20°C	22.81 21.87	2 2
Ref. Index		$\neg$	d'   10 to		4 4		40	20.93	2
n <sub>D</sub> 20°C	1.4096		d g/ml			Para	chor [P]		
30	1.4049		d g/ml vc ml/g tc *C	1	ا ۔ ا		20°C 30		ĺ
"C"	0.7472	2 4	, c	311.0 17072.	5		40	205 2	_
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	17072.	l <sup>3</sup>	F	Suga. L. 1. %/wt.	385.2	5
MR (Calc. (nD-d/2)	) 43.762 1.0446	5 2	25°C	1.0000	5	Exp.	u.		
Dielectric	1,987	5	30 mm BP	1.0000 0.9676	5 4		ersion	96.6	2
A   50 to		1 2	t <sub>e</sub>	0.9582	5		h Point °C Point	26.0	5
B <u> </u>	1397.	2	t <sub>c</sub>	<u> </u>		M Sp			├
C	206.	2	ΔHc kcal/m ΔHf			Ultr	a V.		
A* 50 to B* 170 °C		5 5	ΔFf	L		X-Ra Infra	ay Dif.		
к — — -	-		Viscosity			l <del></del>	bility in +		_
t <sub>k</sub> — to	-		centistokes 7°C	.		Ace	tone	œ	ł
<b>1 x</b> 1		_	'				bon tet.	ec ec	ļ
A' 20 to B' 55 °C		5				Eth	er	œ	
c, 🗀 🛎 🗅	200.	5	B <sup>v</sup>   to		$\Box$		leptane anol	oc oc	İ
A!# 20 to		5	A	_		Wat	ter		
B'* 55 °C	+	5	(B <sup>V</sup> ) to			Wat	ter in		<u> </u>
Ac 190 to	7.3817	78   5	(A <sup>V</sup> ) °C		$\sqcup$				
Cc C- C-	256.0	5	c <sub>p</sub> liq. 300°K	0.39979 0.50256	5 5				
Cryos. A°			400 c <sub>p</sub> vap.300°K	0.39649	2				1
consts. B°	1,57 0		400 c <sub>w</sub> vap.	0.50136	2				1
t <sub>e</sub> °C	157.31	5	V F.	<u> </u>	لــــا	L			L
REFEREN		. 2 4 -	OT 2 T 44 4	C-1- 6			ms/100 gran		
SOURCE:	ν <u></u>	v 2-AF		Calc. from det	. dat	ta 5-	Calc. by form	nula	
PURIFICA:	TION:	AP	<del></del>						
LITERATU				<del></del>					
	··· NEFEF	. PHOES	•						
}									

								No. 47	
NAME	2, 2-Dime	thylh	neptane			ST	RUCTURAL	FORMUL	A
							CH <sub>3</sub>		
			T			C	н <sub>3</sub> ċ (сн <sub>2</sub> ) <sub>4</sub>	CH <sub>3</sub>	
Mole % Pur.		lecul rmuk	- C.H. 1	Molecular Weight 128.2:	50		ĊН <sub>3</sub>		
		Ref		T T T T T T T T T T T T T T T T T T T	Ref.	T			Ref.
F.P. °C	-113.00	2	dt/dP		1	f	300.45	0.040/	
F.P. 100%			°C/mm	1		g	300 to _600_°K	0.0496 0.0013	
B. P. ℃			25°C BP	1.582	5 2	h	l	-0.0635	5
760 mm 100	132.69 71.0	4	t	0.0368	5	f¹	to		
30 10	43.82	4	30 mm	0.6775	4	g'	' <u>°K</u>		İ
1	23.3	5	∆Hm cal/g		$\Box$	h'	1 200		<del> </del>
Pressure			ΔHv cal/g 25°C	70.0	,	m n	300 to	0.0294 0.0015	
mm 25°C	11.06 1097.6	5	30 mm	78.8 76.62	2 4	٥		-0.0 <sub>6</sub> 52	4
Density	1097.6	3	BP	64.8	2 5	m'	700 to	0.1222	4
g/m1 20°C	0.7105	2	te te (d, e)	63.00 62.91	5	n' o'	10 <u>00 °K</u>	0.0013	4
d <sup>t</sup> 25 4 30	0.7066 <b>0</b> .7027	2	AHV/T	19.23	5			-0.0 <sub>6</sub> 45	4
a 30	0.7027	4	d 45 to	82.44	4		face tension es/cm. 20°C	20.80	2
ь	-0.0378	4	-å, -145 % to	0.1330 81.70	4 4	الع ا	30	19.88	2
Ref. Index	. •		e' 45 °C	0.1160	4		40	18.97	2
<sup>n</sup> D 20°C	1.4016 1.399 <b>3</b>	2 2	d <sub>c</sub> g/ml			Par	achor [P]		
30	1.3964	4	v <sub>c</sub> m1/g t <sub>c</sub> °C	295, 2	5		30		
"C"	0.7527	4	P <sub>c</sub> mm	15656.	5		40 Sugd.	385.2	5
MR (Obs.) MR (Calc.		2	PV/RT		$\vdash$	Exp	L.1.%/wt.		Ī
(nD-d/2)	1.0464	5 <b>2</b>	25°C 30 mm	1.0000 1.0000	5	-	u.	00.5	
Dielectric			BP	0.9506	4		persion sh Point C	99.5	2
A 40 to	6, 8580	2	te t	0.9390	5		e Point		
B   165 °C C	_ 1355.0 208.00	2	ΔHc kcal/m		$\vdash$		Spec.		
A* 40 to	1.34316	5	ΔHf				ra V. Ray Dif.		1
B*[155°C K	1272.7	5	ΔFf	<del>                                     </del>	╁─┤		ared		
c	_		Viscosity centistokes				ubility in +		
t <sub>k</sub> to			າ •ເ				etone rbon tet.	oc oc	
t <sub>x  </sub> °C	7. 02527	5					nzene her	<b>00</b>	
B' 45 °C	1440.55	5	_v			n-	Heptané	<b>8</b> 0	
C'	215.8	5	B <sup>V</sup> to C				hanol ater	œ	
A'* 10 to B'* 45 °C	1.4984 1352.6	5	(B <sup>V</sup> )  - to	-			ater in		
Ac 165 to	7, 33533	5	(A <sup>V</sup> )  °C						
Bc tc °C	1733.9	5	c liq. 300°K	0.39979	5				
Cryos. A°	0.042	2	P 400	0.50256	5	1			
consts. B°	0.042	-	c <sub>p</sub> vap.300°K 400	0. 39649 0. 50136	2 2				
t <sub>e</sub> °C	146.94	5	c <sub>v</sub> vap.						
$T_{\mathbf{R}} = 0.7$	7 T <sub>c</sub>				'	+ g1	rams/100 gra	ms solven	t
REFEREN	CES: 1-Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			PI						
PURIFICA?			PI						
LITERATU	RE REFERE	NCES	5:						

No. 48 NAME 2, 3-Dimethylheptane STRUCTURAL FORMULA сн3 сн<sub>3</sub>сн сн - (сн<sub>2</sub>)<sub>3</sub>сн<sub>3</sub> Ċнз Mole Ref. Molecular Molecular  $C_9H_{20}$ Weight 128, 250 % Pur Formula Ref. Ref Ref. <u>•c</u> dt/dP 300 to 0.0496 f F. P. 100% \*C/mm 25\*C 600°K 0.0013 5 g 2.182 5 B. P. \*C -0.0635 5 h BP 0.0496 4 760 mm 140.5 2 t<sub>e</sub> 0.0362 5 f' to 100 78. 50.3 2 g' °K 4 30 30 mm 0.6885 4 10 29.4 5 h' ∆Hm cal/g 1 -5.18 5 300 to 0.0294 m AHv cal/g Pressure 600°K 0.0015 n 25°C 81.6 -0.0<sub>6</sub>52 mm 25°C 7.738 4 o 4 30 mm 78.51 1137.3 5 t<sub>e</sub> 67.3 ΒP 2 700to 0.1222 m۱ Density 65.44 5 te (d, e) 1000°K 0.0013 g/ml 20°C 0.7260 2 65.33 5 -0.0645 4 ٥' ď4 25 0.7221 2 ΔHv/Te 19.54 5 30 0.7182 4 Surface tension 1 84.76 50 to 4 0.7416 -0.0<sub>3</sub>78 44 dynes/cm. 20°C 22.34 2 155 0.1242 4 ١ °C ь 30 21.40 2 ď١ 10 to 84.66 4 40 20.47 2 Ref. Index 50 °C e' 0.1223 4 20°C 1.4085 [P]  $\mathbf{n}_{\mathbf{D}}$ dc Parachor g/ml 25 1.4062 2 20°C v<sub>c</sub> ml/g °C 1.4042 4 30 30 309.5 5 t<sub>c</sub> 40 "C" 0.7500 4 P<sub>c</sub> mm 16649. 5 5 Sugd. 385.2 MR (Obs.) 43.63 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 43, 762 25°C 1,0000 1.0455 2 (nD-d/2)30 mm 1.0000 5 Dispersion 97.2 2 1.984 5 Dielectric 0.9619 BP 4 Flash Point °C 5 26.0 0.9517 50 to 6.887 2 Fire Point tç 1392.0 В 2 M Spec. c 207.0 2 AHc kcal/m Ultra V ΔHf A\* 50 to 1.3421 X-Ray Dif. ΔFf 165 °C B\* 1302.4 Infrared ĸ Viscosity Viscozzi, centistokes °C Solubility in c Acetone to Carbon tet. •c Benzene œ 10 to 7,0269 Ether 1464.9 В' 50 °C 5 n-Heptane œ  $\mathbf{B}^{\tilde{\mathbf{v}}}$ Ċ١ 5 213.7 Āv Ethanol 80 °C Water A'\* 10 to 1.4968 5 Water in B'\* 50 °C (BV) 1377. 5 to Ac | 175 to 7.36579 (A<sup>V</sup>) °C Bc tc C 1782.3 5 liq. 300°K 0.39979 5 Cc 257.3 5 400 0.50256 5 Cryos. A° consts. B° vap.300°K 0.39649 0.50136 400 te °C vap. 156, 34 5  $T_R = 0.77 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 4-Calc. from det. data 5-Calc. by formula 2-API 3-Lit. SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 49 2,4-Dimethylheptane NAME STRUCTURAL FORMULA CH3CH - CH2CH (CH2)2CH3 Ċнз ĊHą Mole Ref. Molecular Molecular Weight 128, 250  $C_{9}H_{20}$ % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g \_°K 25°C 1.606 4 B. P. ℃ h BP 0.049 2 133.5 760 mm 2 0.0359 5 ſ١ to 71. 100 2 g' <u>• к</u> 44.23 .30 4 30 mm 0.6772 4 10 23.8 5 h' ∆Hm cal/g 5 -9.6 300 to 0.0294 m ∆Hv cal/g Pressure 0.0015 | 600 °K n 25°C 79.9 2 mm 25°C 10.738 4 o -0.0<sub>6</sub>52 4 30 mm 76.85 5 1131.2 t<sub>e</sub> 5 BP 2 65.9 700 to 1000 °K m' 0.1222 4 Density 64.85 5 te te (d, e) n' 0.0013 g/ml 20°C 4 0.716 2 64.07 5 ۰, -0.0645 4  $\mathbf{d_{4}^{t}}$ 25 0.712 2 AHV/Te 5 19.73 30 0.708 Surface tension d 45 82.27 4 to 0.732 a 4 dynes/cm. 20°C 21.30 1<u>50</u> °C 0.1226 4 -0.0<sub>3</sub>796 Ъ 4 20.38 2 30 ă¬` 10 to 83.87 19.46 2 40 Ref. Index e¹ 45 0.1587 5 20°C 1.4033 [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 25 1.4010 2 20°C vc ml/g tc °C 30 1.398 4 30 297.7 5 40 "C" 0.7501 4  $P_c$  mm 16179. 5 Sugd. 385.2 5 MR (Obs.) 43.7 2 PV/RT Exp. L.l.%/wt. MR (Calc.) 43.762 5 25°C 1.0000 5 (nD-d/2) 1.0453 2 30 mm 1.0000 5 97.8 2 Dispersion Dielectric BP 0.9610 5 Flash Point C 0.9642 A 40 to 6.869 2 Fire Point В 1360. 2 M. Spec. Ultra V. С 208. 2 AHc kcal/m ΔHf A\* 40 to 5 1.3129 X-Ray Dif. ΔFf B\*[160 °C 1267. Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene 25 to 6.7010 Ether В' 45 °C 1277. 5 n-Heptane Bv | Av | 200. 5 to Ethanol ۰c Water 25 to 1.1957 5 B'\* 45 °C Water in 1199. (B<sup>V</sup>)| to Acl 165 to 7.34717 5 (A<sup>V</sup>) °C Bc tc °C 1741.6 5 c<sub>p</sub> liq. ۰ĸ 5 Cc' 257 3 cp vap 300°K Cryos. A 0.39649 2 consts. B° 400 0.50136 c vap. te °C 148.44 5  $T_{R}$  $= 0.77 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 50	
NAME	2,5-Di	methylhe	ptane			STRUCTURAL	FORMULA	
						сн <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub> с	א כא כש	
	1 1		···				;н <sub>3</sub>	3.
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 128,2	50	03	3	
76 Fur.		Ref	7 20 1	weight 120,2	Ref	r		Ref.
F. P. *C	Τ	Kei	1. / 1.		Kei	<del> </del>		
F.P. 100%	<del> </del>	$\dashv$	dt/dP *C/mm	ı		f 300 to	0.0496 0.0013	
B. P. *C	†	$\dashv$	25°C	1.847	5	h	-0.0635	
760 mm 100	136.0	2	BP t <sub>e</sub>	0.049 0.0362	2	f' to	<b> </b>	$\vdash$
30	74.1 46.9	2 4	30 mm	0.6802	4	g'   '°K		
10	26.2	2	ΔHm cal/g	1,333	_	h'		
1	-8.4	5	ΔHv cal/g	<del></del>		m   300 to	0.0294	
Pressure mm 25°C	9, 338	3   5	25°C	79.9	2	n _600 °K	0. <b>00</b> 15	
t <sub>e</sub>	1119.5	5	30 mm BP	77.80 66.4	5 2	ļ <u>-</u>	-	
Density	1		t_	64.56	5	m'   700 to	0.1222 0. <b>0</b> 01 <b>3</b>	4
g/ml 20°C	0.715		t <sub>e</sub> (d, e)	64.48	5	0 1 <u>1000 K</u>	-0.0645	
dt 25 4 30	0.707		ΔHv/T <sub>e</sub>	19.52	5	Surface to add	-	├-
a	0.731	4	d   45 to	83.80	4	Surface tension dynes/cm. 20°C	21.30	2
Ъ	-0.038	3 4	150 °C		4	¥ 30	20.38	2
Ref. Index			e' 45 °C		4	40	19.46	2
n <sub>D</sub> 20°C	1.403		d g/ml v ml/g			Parachor [P] 20°C	1	1
30	1.398		vc ml/g tc C	300.7	5	30		
"C"	0.752	6 4		16142.	5	40 Sugal	385.2	5
MR (Obs.)	43.9	2	P <sub>c</sub> mm PV/RT	10112.		Exp. L.1.%/wt.	305.2	-
MR (Calc. (nD-d/2)	1.046		25°C	1.0000	5	u.		
Dielectric	1.971		30 mm BP	1.0 <b>00</b> 0 0.9591	5 4	Dispersion	97.8	2
A 45 to	+		t.	0.9484	5	Flash Point °C Fire Point	23.0	5
B 170°C	1372.0	2	t <sub>c</sub>			M Spec.	<del>                                     </del>	├
С	207.	2	ΔHc kcal/m ΔHf			Ultra V.		
A* 45 to B* 160 °C	1.347	3 5 5	ΔFf	ł		X-Ray Dif.		
K 1.00	- 1205.5		Viscosity			Infrared Solubility in +	ļ	
·	_		centistokes			Solubility in +		
t <sub>k</sub> to	:	- 1 1	<b>η °</b> €.	1		Carbon tet.	<b>60</b>	
A'   10 to	7, 251	5 5				Benzene Ether	ec ec	
B' _ 50 °C		5	B <sup>V</sup>   to			n-Heptane	<b>60</b>	
C'	224.4	5	B' to A'   °C	•		Ethanol Water	• <b>•</b>	İ
A'* 10 to B'* 50 °C		14 5 5 5 5 5 T 5 T 5 T 5 T 5 T 5 T 5 T 5	(BV) to	1		Water in		L
Ac   185 to	7, 357		(A <sup>V</sup> )  °C					
Bc t °C	1753.3	5		0.39979	5	1		1
Ce	256.0	5	400	0.50256	5			
Cryos, A° consts. B°			c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2			
t <sub>e</sub> °C	151.04	5	c <sub>w</sub> vap.					1
$T_R = 0.7$			<u> </u>	L	L	+ grams/100 gra	me solver	<u> </u>
REFERENC		w 2-AF	PI 3-Lit. 4-0	alc. from det	t. da	ta 5-Calc. by for		<u> </u>
SOURCE:		AF				2 2220, 27 101		
PURIFICAT	ION:	AF	<del></del>					
LITERATU						··_		

<del></del>								No. 51	·
NAME	2,6-1	Dimethyl	heptane			ST	RUCTURAL	FORMUL	A
<u> </u>						C	н <sub>3</sub> сн (сн <sub>2</sub> ) <sub>3</sub>	сн - сн	
Mole % Pur.	Ref.	Molecu Formul	lar C <sub>9</sub> H <sub>20</sub>	Molecular Weight 128.2	50		- 1	Ċн <sub>3</sub>	
		Rei	4		Ref.				Ref.
F. P. ℃	-102.9	2	dt/dP			ſ	300 to	0.0496	5
F.P. 100%			°C/mm			g	_6 <u>0</u> 0_ <b>°</b> K	0.0013	5
B. P. ℃	l		25°C BP	1.780	5 2	h		-0.0 <sub>6</sub> 35	5
760 mm 100	135.21		te	0.0363	5	ſ١	to		
30	46.18	4	30 mm	0.6793	4	g'	'° <u>K</u>		
10	25.5	4 5	∆Hm cal/g			h'	ļ		
Pressure	1		∆Hv cal/g			m n	300 to	0.0294 0.0015	
mm 25°C	9, 70		25°C 30 mm	79.8 77.55	2	0		-0.0652	
t <sub>e</sub>	1117.7	5	BP	66.2	2	m'	700 to		4
Density g/ml 20°C	0.70	89 2	te (3 a)	64.35	5	n'	1000 K	0.1222 0.0013	4
dt 25	0.70		te (d, e)	t	5	0'		-0.0 <sub>6</sub> 45	4
<sup>4</sup> 30	0.70	09 4	ΔHv/T <sub>e</sub>	19.49		Sur	face tension		<b>!</b>
a	0.72		d 45 to e 150 °C	83.44 0.1275	4		es/cm. 20°C	20.83	2
b	-0.03	80 4	d 10 to	82.45	4	,	30 40	19.92 19.00	2
Ref. Index		07 2	e'   45 °C	0.1061	4	Par	achor [P]		Ħ
25	1.39	83 2	d g/ml vc ml/g				20°C		
30	1.39		± °C °	298.6	5		30 <b>4</b> 0		
"C"	0.75		P <sub>c</sub> mm	15863.	5			385.2	5
MR (Obs.) MR (Calc.			PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	Die	u. persion	98.9	2
Dielectric	1.96	2 5	∥ RD	0.9599	4		sh Point C	22.0	5
A 45 to	6.87		t <sub>e</sub>	0.9488	5		e Point	22.0	
B 1_170 °C	1366.0	2 2	tc AHc kcal/m	<del></del>	-	M.	Spec.		
A* 45 to	1.33		ΔHf				ra V. Ray Dif.		
B*[160 °C		5	ΔFf	ļ.,			ared		
K	1		Viscosity centistokes			Sol	ability in +		
t <sub>k</sub> to	•		7 °C				etone rbon tet.	<b>00</b>	
t <sub>x</sub> °C			1			Be	nzene	∞ ∞	
A'   10 to B'   50 °C	7.15 1510.6	10   5					her Usatana	<b>00</b>	
c'	220.1	5	B <sub>v</sub> to				Heptane hanol	ος ος	
A** 10 to	1.61		A C	_			ter		
B'* 50 °C	1419.0	5	(B <sup>V</sup> )  to	1			ter in		├
Ac 170 to Bc t <sub>c</sub> °C	7.34 1741.9	814 5	(A <sup>V</sup> )  °C						
Cc C	255.7	5	c <sub>p</sub> liq. 300°K	0.39979	5				
Cryos. A°			c <sub>p</sub> vap300°K	0.50256 0.39649	5 2				1
consts. B°	ļ		n 400	0.50136	2				
t <sub>e</sub> °C	150.20	5	c <sub>v</sub> vap.	<u> </u>	$\perp$				<u> </u>
$T_{\mathbf{R}} = 0.7$							ams/100 gra		t
REFERENC	CES: 1-D			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			API						
PURIFICAT			API						
LITERATU	RE REF	ERENCE	S:						
1									

							No. 52	
NAME	3, 3-E	imethyll	neptane			STRUCTURAL CH3	FORMUL	A.
			<del></del>			сн <sub>3</sub> сн <sub>2</sub> с -		
Mole	Ref.	Molecu	lar C <sub>9</sub> H <sub>20</sub>	Molecular Weight 128.	350	Ċн <sub>3</sub>		
% Pur.		- T	<del></del>	Weight 128.	Ref	r	·	Ref.
F, P. *C	T	Ref	1	T	Kei	<del>                                     </del>	1	101.
F.P. 100%	<b>†</b>		dt/dP *C/mm		} }	f t	K.	1
B. P. °C			25°C BP	1.833	5 2	h ,		ĺ
760 mm 100	137.3	2 2	t.	0.0367	5	f' t		
30	46.9	4	30 mm	0.6897	4	- :	K_	l
10 1	26. -8.4	5	ΔHm cal/g			h'		<u> </u>
Pressure	† — —		ΔHv cal/g	-0.5		m   300t		
mm 25°C	9.45	3 5	25°C 30 mm	79.5 76.72	5	0	-0.0652	
t <sub>e</sub> Density	1124.		BP	65.8	2	m'   700 t		
g/ml 20°C	0.72		te te (d, e)	64.01 63.94	5	n'   1000°	K   0.0013	4
dt 25 4 30	0.72 0.71		AHv/T	19.28	5	o'	-0.0645	4
	0.71		d   45 to		4	Surface tension dynes/cm. 20°		2
ь	-0.03		150 °C		4	30	21.09	2
Ref. Index			e' 45 °C		4	40	20.17	2
<sup>n</sup> D 20°C	1.40		d <sub>c</sub> g/ml			Parachor [P]	۱.	İ
30	1.40		v <sub>c</sub> ml/g t <sub>c</sub> °C	304.5	5	30	١	l
"C"	0.75	03 4	P <sub>c</sub> mm	16081.	5	40 Sug	d. 385.2	5
MR (Obs.) MR (Calc.)	43.7 43.76	2   2	PV/RT	+	$\vdash$	Exp. L.1.%/wt		H
(nD-d/2)	1.04		25°C	1.0001	5	u.		١,
Dielectric	1.98	4 5	30 mm BP	1.0000 0.9591	5	Dispersion Flash Point °C	97.4	2
A 45 to			ţ.	0.9484	5	Fire Point		1
B <u>  170°C</u>	1385.	2 2	t <sub>c</sub>	+	$\vdash$	M Spec.		
A* 45 to	1.33	05 5	ΔHf			Ultra V. X-Ray Dif.		1
B* ∟160 °C	1296.	5	ΔFf Viscosity	+	$\vdash$	Infrared		<u> </u>
c	_	İ	centistokes			Solubility in Acetone	+	1
tk C			η •α			Carbon tet.		
t <sub>x</sub>   *C		30 5	1			Benzene Ether		
B' _ 50 ℃		5 5	B <sup>v</sup>   to	<del> </del>	$\vdash$	n-Heptane	ĺ	
A'* 10 to	1.36		B to to ℃	1		Ethanol Water		1
B'* 50 °C		5	(B <sup>V</sup> ),	-		Water in		1
Ac   170 to	7.35		(A <sup>V</sup> ) •C	:				
Bc tc_C	1779.0 260.9	5 5	cp liq. °K					
Cryos. A°	1		c <sub>p</sub> vap.300°K	0.39649	2			
consts. B°	<u> </u>		400	0.50136			}	
t <sub>e</sub> °C	152.73	5	c <sub>v</sub> vap.	1				<u></u>
$T_{R} = 0.77$						† grams/100 gr		t
REFERENC	ES: 1-D			Calc. from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE: PURIFICAT	TON:		PI					
LITERATU			PI s.					
	NE KEFI	an ence	J.					

No. 53 3, 4-Dimethylheptane NAME STRUCTURAL FORMULA CH<sub>3</sub> CH3CH2CH CH (CH2)2CH3 Mole Ref. Molecular Molecular ċнз  $C_{9}H_{20}$ % Pur Weight 128, 250 Formula Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °<u>K</u> g 25°C 2.178 5 B.P. °C h BP 0.050 760 mm 140.6 2 t<sub>e</sub> 0.0360 5 ſ١ 100 78. to 2 g' <u>°K</u> 30 50.31 4 0.6899 4 30 mm 10 29.4 4 h' AHm cal/g 1 -5.09 5 m 300 to 0.0294 4 AHv cal/g Pressure n 1\_600 °K 0.0015 0.0<sub>6</sub>52 4 4 25°C 81.7 2 mm 25°C 7.742 o 30 mm 78.35 5 1147. 5 t<sub>e</sub> ВP 67.8 2 700 to 0.1222 m 4 Density 65, 92 5 t<sub>e</sub> (d, e) 0.0013 1000 °K g/ml 20°C 0.7314 2 65.90 o' -0.0645 4 25  $\mathbf{d_4^t}$ 0.7275 2 AHv/Te 19.66 5 30 0.7236 4 Surface tension ď 50 to 84, 22 4 0.74699 a 4 dynes/cm. 20°C 22.80 2 °C 0.1168 å-1<u>55</u> 4 ь -0.0<sub>3</sub>777 4 30 21.87 2 10 to 85.01 5 20.94 2 40 Ref. Index e' 50 0.1325 5 20°C 1.4111 n<sub>D</sub> 2 [P] Parachor dc g/ml 25 1.4089 2 20°C vc ml/g t\_°C 30 30 t<sub>c</sub> 311.1 5 40 "C" 4 0.7483 P<sub>c</sub> mm 16953. 5 Sugd 385.2 5 MR (Obs.) 43.55 43.762 2 5 PV/RT Exp. L.1.%/wt. MR (Calc.) 25°C 1.0001 5 (nD-d/2)1.0454 2 u. 30 mm 1.0000 5 5 Dispersion 2 96.8 0.9692 Dielectric ВP Flash Point C A 50 to 0.9588 5 6.897 2 Fire Point B 175 °C 1400. 2 M. Spec. С 208.0 2 AHc kcal/m Ultra V. A\* 50 to ΔHf 1.3395 5 X-Ray Dif. ΔFf B\* 165 °C 1306.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet.  $^{t}x$ °C Benzene A' 25 to 6.9320 Ether B' L 50 °C 1418.1 5 n-Heptane B<sub>V</sub> C' 209.7 5 to Ethanol °C A'\* 25 to B'\* 50 °C Water 5 5 1.4076 Water in 1332.9 (B<sup>V</sup>)| to Acl 175 to 7.37833 5 (AV) °C Bc tc °C 1794.8 5 c<sub>p</sub> liq. ۰ĸ Cc 258.9 5 Cryos. A c<sub>p</sub> vap.300°K 0.39649 2 consts. B° 0.50136 2 400 c vap. te °C 156.83 5  $T_R = 0.77 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

· ·				<del></del>				No. 54	
NAME	3,5-D	imethyll	neptane			STR	UCTURAL I	FORMULA	
						CH <sub>2</sub> C	н2сн сн2с	н - сн,с	Н
Mole	Ref.	Molecu	la w	Molecular	-	,		н,	•
% Pur.		Formu		Weight 128.2	50		<b>,</b>		
		Rei	1		Ref				Ref.
F, P. *C			dt/dP			f	to		
F.P. 100%			*C/mm			g	<u>*</u> K_		
B. P. °C			25°C BP	1.812 0.049	5 2	h			
760 mm 100	136.0 73.9	2	t <sub>e</sub>	0.0363	5	f'	to		
30	46.59	4	30 mm	0.6824	4	g'	•K_		
10 1	26. -8.	5	ΔHm cal/g			h'			
Pressure	-0.	-+-	ΔHv cal/g			m	300 to	0.0294	
mm 25°C	9.49	5 5	25°C	80.1	2 5	n o	_600 °K	0.0015 -0.0 <sub>6</sub> 52	
t <sub>e</sub>	1124.	5	30 mm BP	77.40 66.4	2		1		
Density			t. (1.0)	64.57	5	m'	700 to	0.1222 0.0013	4
g/ml 20°C	0.72		'e (u, e)	64.52	5	٥'	1.2020 202	-0.0645	4
d 25 4 30	0.71		ΔHv/T <sub>e</sub>	19.51	5	Sumf	ce tension		
	0.73	9 4	d 45 t		4 4		s/cm. 20°C	21.77	2
<u>· Ъ</u>	-0.03	80 4	-  Ta' 10 t		5	*	30	20.85 19.93	2 2
Ref. Index	1.40	47 2	e'   45 °	C 0.1250	5		40	19. 73	<u> </u>
<sup>n</sup> D 20°C	1.40		d <sub>c</sub> g/ml			Para	chor [P]		
30	1.40	21 4	d g/ml vc ml/g tc °C	302.3	5	1	30		l
"C"	0.74	93 4	P <sub>c</sub> mm	16335.	5	ĺ	40 Sugd	385.2	5
MR (Obs.)	43.6	_   2	PV/RT	-	-	Evn	L. 1. %/wt.	303.2	۱
MR (Calc.) (nD-d/2)	43.76 1.04		25°C	1.0000	5		u.		
Dielectric	<u> </u>	<u> </u>	30 mm BP	1.0 <b>000</b> 0.9626	5		ersion	96.9	2
A   45 to	6,87	8 2	t <sub>e</sub>	0.9519	5		h Point °C Point		ł
B <u>[160 °C</u>	1375.0	2	tc			M Sp			-
С .	208.0	2	ΔHc kcal/m			Ultra			i
A* 45 to B* 160 °C	1.33 1286.3	769   5	ΔFf				y Dif.		ì
K Lie e	1200.5		Viscosity			Infra			-
t			centistokes	.	1		oility in +		İ
tk to		- 1	7 °C	1			bon tet.		1
A'   25 to	6.97	157 5	-∦	ł		Ben Eth	zene er		
B' ∟ <u>50</u> °C		5	B <sup>v</sup>   to	+	$\vdash$	n-H	eptane		
A'* 25 to	212.4	661 5	B' to			Wat	anol er		
B'* 50 °C		5	(B <sup>V</sup> )	-			er in		
Ac   160 to	7.27	56 5	(A <sup>V</sup> )						
Bc tc_C	1681.6	5	c <sub>p</sub> liq. •p		-				
	246.8	5	-{  '	i					
Cryos, A° consts. B°	İ		c <sub>p</sub> vap.300°F	0.39649 0.50136		į			
t <sub>e</sub> °C	151.27	5	c <sub>v</sub> vap.						Ì
$T_{\mathbf{R}} = 0.77$	T <sub>c</sub>					+ gra	ms/100 gran	ns solveni	
REFERENC	ES: 1-D	2-A	PI 3-Lit. 4-	Calc, from de	t. da				
SOURCE:		A	PI						
PURIFICAT	ION:	A	PI						
LITERATUE	E REFI	RENCE	S:						

								No. 55	·
NAME _	4, 4-Dime	thylh	eptane			STR	UCTURAL	FORMUL	A
					CH <sub>3</sub>				
<b></b>			T		$\dashv$	CH.	(СH <sub>2</sub> ) <sub>2</sub> ċ (	CH_)_CH_	
Mole	Ref. Mo	ecul	ar cu l	Molecular		3	cH <sub>3</sub>	2, 2 - 3	,
% Pur.	For	mula	C <sub>9</sub> H <sub>20</sub>	Weight 128.25	0		<u></u>		
		Ref.			Ref.				Ref.
F.P. °C			dt/dP			f	to		
F.P. 100%			°C/mm		_	gl	<u>°K</u>		
B.P. ℃		1 1	25°C BP	1.680 0.0496	5 4	h			
760 mm 100	135.2 73.	2 2	te	0.0365	5	f'	to		
30	45.16	4	30 mm	0,6865	4	g'	° <u>K</u>		
10	24.5	5	ΔHm cal/g			h'			
1	-9.	5	ΔHv cal/g			m	300 to	0.0294	4
Pressure mm 25°C	10.32	5	25°C	79.5	2	n	_600 <b>•</b> K	0.0015	
t <sub>e</sub>	1128.	5	30 mm	76.26	5			-0.0652	7
Density			BP t	65.9 64.05	2 5	m'	700 to	0.1222	4
g/ml 20°C	0.725	2	te te (d, e)	64.11	5	n'	1000 °K	0.0013 -0.0 <sub>6</sub> 45	4 4
dt 25 4 30	0.721	2	AHv/Te	19.37	5	لـــّـــا		-0.0633	╨
	0.717	4	d 45 to	81, 45	5		ce tension		1. 1
a b	0.74099 -0.0 <sub>3</sub> 797	4 4	_e_ _1 <u>50</u> °C	0.1150	5	dyne	s/cm. 20°C 30	22.01 21.09	2 2
Ref. Index	1 3.7	$\vdash$	d'   10 to	83.52 0.1609	5	] -	40	20.17	2
n <sub>D</sub> 20°C	1.4076	2		0,1007	-	Para	chor [P]		
25	1.4053	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g				20°C		
30	1.4026	4	tc°C	301.7	5		30 40		
"C"	0.7485	4	P <sub>c</sub> mm	15996.	5		-	385.2	5
MR (Obs.) MR (Calc.)	43.6 43.762	2 5	PV/RT			Exp.	L.1.%/wt.		
(nD-d/2)	1.045	2	25°C	1.0000	5	_	u.		_
Dielectric		$\vdash$	30 mm BP	1.0000 0.9673	5		ersion	97.4	2
A   45 to	6, 858	2	te	0.9562	5		h Point C Point		
B 1160 °C	1373.	2.	tc			M. S			
С	210.	2	ΔHc kcal/m ΔHf			Ultra			
A*  45 to B*  160 °C	1.30972 1281.5	5	ΔFf				y Dif.		
K C	1201.3		Viscosity			Infra			
c			centistokes				oility in <sup>T</sup> tone		
t <sub>k</sub> to			η •c	j			bon tet.		
A'   25 to	( (222)						zene		
B' 50 ℃	6.62226	5 5				Ethe	er eptane		
C'	198.8	5	B <sub>v</sub> to			Eth	anol		
A1* 25 to	1.11783	5	LA:			Wat			
B' ≠ 50 °C	1178.6	5	(B <sup>V</sup> )  to	1		Wat	er in		-
Acl 160 to	7.2592 1684.1	5	(A <sup>V</sup> )  °C						
Bc t <sub>c</sub> °C	249.6	5	c <sub>p</sub> liq. °K	1					
Cryos. A°		$\vdash$		0.39649	2	ļ			
consts. B°			c <sub>p</sub> vap.300°K 400	0.50136	2				
t <sub>e</sub> °C	150.77	5	c <sub>v</sub> vap.	1		1			
T <sub>R</sub> = 0.757	r <sub>c</sub>					+ gra	ms/100 gra	ms solven	t
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da				·
SOURCE:			PI				,		
PURIFICAT	ION:		PI					***	
	RE REFERE								
LILERATU	AL REFEKE	NUES	);						
1									
									i
L									-

									No. 56	
NAME	3-Eth	y1-2-	metl	hylhexane			STR	UCTURAL	FORMULA	4
Ī							C <sub>2</sub> H <sub>5</sub>			
							CF	<sub>13</sub> сн сн (с	H <sub>2</sub> ) <sub>2</sub> (CH <sub>3</sub>	)
Mole % Pur.	Ref.		ecul:		Molecular Weight 128.2	50		ĊН <sub>3</sub>		
70 Pur.			Ref.	1 / 20 1	weight 120.2	Ref				Ref
F,P. °C	Т		Kei.	11/15	Т	Kei	<del>                                     </del>	<del> </del>		-
F.P. 1007	1			dt/dP *C/mm			f g	to to		
B. P. *C	1			25°C	1.929	5	h			İ
760 mm 100	138.0 75.5	ļ	2	BP t <sub>e</sub>	0.050 0.0361	2	f'	to		$\vdash$
30	47.98	ł	4 4	30 mm	0,6870	4	g'	K_		
10 1	27.	1	2	ΔHm cal/g	<del>                                     </del>		h'	1		
Pressure	-6.			ΔHv cal/g	<del>                                     </del>		m	300 to	0.0294	
mm 25°C	8,74	4	5	25°C	81.7	2	n o	_600 •K	0.0015 -0.0 <sub>6</sub> 52	
t <sub>e</sub>	1140.		5	30 mm BP	77.57	5 2		1	<del></del>	
Density		, 1	$\Box$		65.40	5	m' n'	700 to	0.1222 0.0013	4
g/ml 20°C	0.73		2 2	te te (d, e) AHv/T	65.24	5	0'	1 . 4. 2. 2	-0.0 <sub>6</sub> 45	
d <sub>4</sub> 25	0.72		4	e	19.65	5	Surf	ace tension	<del>`</del> _	$\vdash$
a L	0.74		4	d   50 to		4 4		s/cm. 20°C	22.80	2
ь	-0.03	197	4	d' 10 to	86.20	5	,	30 <b>40</b>	21.87 20.94	2
Ref. Index		20	2	e' 50 °C	0.1799	5	Par	achor [P]	20.71	Ē
25	1.40	97	2	d g/ml vc ml/g	ì			20°C		
"C"	1.40	$\rightarrow$	4	tc °C	306.7	5		30 40		
	0.74	79	5	P <sub>c</sub> mm	16524.	4			385.2	5
MR (Obs.) MR (Calc.		2	2 5	PV/RT		-	Exp	L.1.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000 1.0000	5	Dist	u. ersion	96.6	2
Dielectric				BP	0.9703	5		h Point °C	70.0	<u> </u>
A 45 to		2	2 2	te t <sub>c</sub>	0.9594	5		Point		
B [160 °C	208.0	ļ	2	ΔHc kcal/m	<del> </del>	-	M S			
A* 45 to	1.31	689	5	ΔHf	i		Ultr X-R	a V. ay Dif.		[
B* 165 °C	1288.5	- 1	5	ΔFf	ļ		Infra			
c		ļ		Viscosity centistokes				bility in +		
tk To		ŀ		η •c				etone rbon tet.		
t <sub>x</sub>   °C		701	-		İ			nzene		
B' _ 50 °C		101	5		<u> </u>		Eth n-H	er Ieptane		
C'	193.1		5	B <sup>V</sup>   to	j		Eth	anol		
A'* 25 to B'* 50 °C	1.05	788	5		4		Wa Wa	ter ter in		
Ac   160 to	+	07	5		1					
Bc t °C	1691.5	۰٬	5	<u> </u>	<del> </del>	$\vdash$				
CC	247.5		5	c <sub>p</sub> liq. ∘K	1					
Cryos. A° consts. B°				c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2				
t <sub>e</sub> °C	153.99		5	c <sub>v</sub> vap.						
$T_R = 0.7$							+ gra	ms/100 grar	ns solven	t
REFEREN	CES: 1-D	ow 2	2-AF	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-	Calc. by form	mula	
SOURCE:			AF							
PURIFICA'			AF					· · · · · · · · · · · · · · · · · · ·		
LITERATU	RE REF	EREN	CES	:						
L										

r							No. 57		
NAME	4-Ethyl-2-	meth	ylhexane		STRUCTURAL FORMULA				
			-						
	T			$\neg \neg$	CH3CH-CH2CH-CH2CH3				
Mole % Pur.		lecul mula		Molecular Weight 128.25	50		2 <sup>H</sup> 5		
/0 Fur.	F0.	Ref.	<del></del>	Weight 120.2.	Ref.		Ref		
F. P. °C		IX.	dt/dP		Ker.		T		
F.P. 100%			°C/mm	İ		f ' to			
B. P. °C			25°C BP	1.60471	4 2	h			
760 mm	133.8 72.0	2 2	te	0.0490 0.03630	5	f' to			
30	44. 31	4	30 mm	0.6820	4	g' ¦°1	<u> </u>		
10	24.0	2	ΔHm cal/g			h' i			
Pressure	-7.24	-	ΔHv cal/g		!	m   300 to			
mm 25°C	10.719	4	25°C 30 mm	80.1 76.35	2 5	n [600 e	-0.0652 4		
t <sub>e</sub> _	1127.	5	BP BP	66.4	2	m1 700 te	+		
Density g/ml 20°C	0,723	2	t <sub>e</sub>	64.805 64.685	5	n'   1000 º1			
dt 25	0.719	2	te (d, e)	i	5	0'	-0.0645 4		
<sup>4</sup> 30	0.715	4	ΔHv/T <sub>e</sub>	19.675 81.27	4	Surface tension			
a b	0.73898 -0.0 <sub>3</sub> 796	4	_e _  _1 <u>50 °C</u>	0.1111	4	dynes/cm. 20°C	21.77 2		
Ref. Index	-0.03170	<u> </u>	d' 10 to e' 45 °C	84.96 0.1945	5 5	8 30 40	20.85 2 19.93 2		
n <sub>D</sub> 20°C	1.4068	2		0.1743	۲-	Parachor [P]			
25	1.4046	2	dc g/ml vc ml/g			20°0	;		
"C"	0.7486	4	vc ml/g tc °C	299.9	5	30 40			
MR (Obs.)	43.7	2	P <sub>c</sub> mm	16102.	5	Sugo	1. 385. 2 5		
MR (Calc.)	43, 762	5	PV/RT 25°C	1 0000	5	Exp. L.1.%/wt	.		
(nD-d/2)	1.045	2	30 mm	1.0000	5	u. Dispersion	96.9 2		
Dielectric			BP	0.9694 0.9591	5	Flash Point C			
A 40 to B 160 °C	6.854	2 2	t <sub>c</sub>	0. 7571		Fire Point			
c 12.3.2	209.0	2	∆Hc kcal/m			M. Spec. Ultra V.			
A* 40 to	1.30397	5	ΔHf ΔFf			X-Ray Dif.			
B*[160 °C	1270.45	5	Viscosity			Infrared.			
:			centistokes	1		Solubility in TACetone			
t <sub>k</sub> to			η °C	1		Carbon tet.			
A'  25 to	6,40991	5		1		Benzene Ether			
B' _ 45 °C	1146. 31	5	B. to	<u> </u>	$\vdash$	n-Heptane			
C' 25 45	188.1	5	B to A °C	{		Ethanol Water			
A'* 25 to B'* 45 °C	0.92150 1076.64	5	(B <sup>V</sup> )  to	1		Water in			
Ac   160 to	7, 2532	5	(A <sup>V</sup> )  °C	†					
Bc tc °C	1669.5 248.2	5 5	c <sub>p</sub> liq. °K		$\vdash$				
Cryos. A°	270,2	,	l -	0.30/45	,				
consts. B°			c <sub>p</sub> vap.300K 400	0.39649 0.50136	2				
t <sub>e</sub> °C	149.23	5	c <sub>w</sub> vap.						
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>		u	<b>L</b>	ı	grams/100 gr	ams solvent		
	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by fo			
SOURCE:		AF				<u> </u>			
PURIFICAT	ION:	AF	PI						
LITERATU	RE REFERE	NCE	S:						

								No. 58			
NAME	3-Eth	yl-3-met	hylhexane			STR	UCTURAL 1				
						CH <sub>3</sub>					
l					$\overline{}$	CH	<sub>13</sub> сн <sub>2</sub> с - (с	CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>			
Mole % Pur.	Ref.			Molecular Weight 128.2	250		Ċ <sub>2</sub> H <sub>5</sub>				
70 Fur.		Formul Ref		weight 120.2	Ref				Ref		
F.P. °C	T	IXE1.	dt/dP	T		١,	1		-		
F.P. 100%			*C/mm	}		f g	to eK		1		
B. P. *C			25°C BP	1.991 0.050	5 2	h					
760 mm 100	140.6 77.0	2 2	t <sub>e</sub>	0.0369	5	f	to				
30	48.68		30 mm	0.7007	4	g'	<u>*</u> K_		ł		
10 1	27.0 -6.75	2 5	ΔHm cal/g	Ť		h'					
Pressure	-0.73		ΔHv cal/g			m (	300 to	0.0294			
mm 25°C	8.62		25°C 30 mm	80. 2 76. 37	5	n l	6 <b>0</b> 0 °K	0.0015 -0.0 <sub>6</sub> 52	4		
t <sub>e</sub>	1143.	5	BP	66.6	2	m'	700 to		-		
Density g/ml 20°C	0.74	1 2	te te (d, e)	65.02 64.86	5 5	n'	1000 °K	0.1222 0. <b>00</b> 13	4		
dt 25	0.74		Le (d, e)		1 1	0'		-0.0 <sub>6</sub> 45			
<sup>4</sup> 4 30	0.73	3 4	ΔHv/T <sub>e</sub>	19.39 81.55	5	Surfa	ce tension				
a b	0.75		e   155 °C	0.1063	4	dyne	s/cm. 20°C	23. 22 22. 30	2 2		
Ref. Index		171   1	a' 10 to		5 5	•	30 40	21.38	2		
n <sub>D</sub> 20°C	1.41			0.1011	H	Para	chor [P]				
25 30	1.41		v ml/g	į			20°C 30				
"C"	0.74		) 'c '	312.5	5		40		ŀ		
MR (Obs.)	+	2	P <sub>c</sub> mm	16423.	5			385.2	5		
MR (Calc.	43.76	2 5	PV/RT 25°C	1,0000	5	Exp.	L.1.%/wt.		}		
(nD-d/2)	1.04	4 2	30 mm	1.0000	5	Disp	u. ersion	95.9	2		
Dielectric	4		BP	0.9654 0.9550	4	Flas	h Point °C				
A 45 to B 165 C		3 2	te tc	0.7550		Fire	Point				
c	212.0	2	ΔHc kcal/m	<u> </u>		M Sp Ultra			İ		
A* 45 to			ΔHf ΔFf				y Dif.				
B* <u> </u>	1309.87	5	Viscosity	<del> </del>	$\vdash$	Infra			<u> </u>		
¢	_	- 1	centistokes	1			ility in + tone		l		
tk   to		1	η ∘c			Car	bon tet.				
A' 25 to	6,57	634 5				Ben Ethe	zene				
B' _ 50 °C	1258, 52	5	B <sup>V</sup>   to	<del> </del>	-	n-H	eptane		İ		
C1	198.1	5	B to A °C			Eth: Wat	nol		l		
A'* 25 to B'* 50 °C	1.06		(B <sup>V</sup> ) to	1			er in				
Ac   165 to	7.26	97 5	(A <sup>V</sup> ) °C						"		
Bc tc_°C	1728.4 253.4	5	c <sub>p</sub> liq. °K	1	$\vdash \vdash$						
Cryos. A°			-11	0.30(40	,	1					
consts. B°			c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2	İ			l		
t <sub>e</sub> °C	157.00	5	c <sub>v</sub> vap.								
$T_R = 0.7$	5 T <sub>c</sub>					+ gra	ms/100 gran	ns solvent			
REFEREN	CES: 1-D			Calc, from det	t. dat						
SOURCE:		AI	PI								
PURIFICA'		AI									
LITERATU	RE REF	ERENCES	5:								

							No. 59		
NAME	4-Ethyl-3	-met	hylhexane		STRUCTURAL FORMULA				
					С <sub>2</sub> Н <sub>5</sub>				
	11				-	сн <sub>3</sub> сн <sub>2</sub> сн-сн		,	
Mole		lecul		Molecular		°с́н <sub>з</sub>	2 3	,	
% Pur.	For	mul		Veight 128.2					
	+	Ref.			Ref.		r	Ref.	
F.P. °C F.P. 100%	<u> </u>	<b>-</b>	dt/dP			f to			
	<del> </del>	-	°C/mm 25°C	2,101	5	g° <u>K</u>	1		
B. P. °C 760 mm	140.4	2	BP	<b>0</b> .05	2	h	-		
100	77.0	2	t <sub>e</sub>	0.0362	5	f <sup>†</sup> to			
30 10	50. 29.	2 2	30 mm	0. 69251	4	8   A			
1	-5.2	5	ΔHm cal/g			m   300 to	0.0294	4	
Pressure			ΔHv cal/g 25°C	81.9	2	n   600 °K		4	
mm 25°C	8.00593	5	30 mm	77.76	5	•	-0.0652	4	
t <sub>e</sub>	1147.	3	BP	67.8	2	m¹ 700 to	0.1222	4	
Density g/ml 20°C	0.742	2	t <sub>e</sub> (d, e)	66.15 66.01	5	n'    1000 °K	0.0013	4	
dt 25	0.738	2	ΔHv/T <sub>e</sub>	19.73	5	0'	-0.0645	4	
4 30	0.734	4	е	83, 22	4	Surface tension			
a b	0.758	4	d 50 to	0.1098	4	dynes/cm. 20°C	23, 27	2	
	-0.038	1	d' 10 to	86.09	5	8 30 40	22.34 21.40	2	
Ref. Index	1.416	2	e' 50 °C	0.1676	5	Parachor [P]			
25	1.414	2	d <sub>c</sub> g/ml			, 20°C	]		
30	1.412	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	312.3	5	30 40			
"C"	0.7459	4	P <sub>c</sub> mm	17008.	5		385.2	5	
MR (Obs.) MR (Calc.)	43.4 43.762	2 5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	1.045	2	25°C	1.0000	5	u.			
Dielectric			30 mm BP	1.0000 0.9694	5	Dispersion	95.5	2	
A 45 to	6, 885	2	t e t c	0.9591	5	Flash Point C Fire Point	i I	ĺ	
B (165 °C		2				M. Spec.		H	
C	209.0	2	ΔHc kcal/m ΔHf			Ultra V.			
A* 45 to B* 165 °C	1.32622	5	ΔFf			X-Ray Dif.			
к 🗀 💆			Viscosity			Infrared		-	
t, to			centistokes			Solubility in Acetone			
t <sub>k</sub> to t <sub>x</sub> °C			η °C			Carbon tet.			
A'   25 to	6, 63329	5				Benzene Ether			
B' 50 °C	1271.80	5	B <sub>w</sub> to		-	n-Heptane			
C'	197.0	5	B to C			Ethanol Water			
A'* 25 to B'* 50 °C	1.12712 1195.27	5	(B <sup>V</sup> )  - to			Water in			
Ac  165 to	7, 2865	5	(A <sup>V</sup> )  °C						
Bc <sub>l</sub> t <sub>c</sub> ℃	1716.2	5			<del>                                     </del>				
	249.3	5	P -						
Cryos, A° consts. B°			cp vap300°K	0.39649					
	154 73	5	c <sub>v</sub> vap.	0.50136	2				
$\frac{\mathbf{t_e} ^{\circ} \mathbf{C}  \mathbf{F}}{\mathbf{T_R} = 0.7}$	156.72	٠,	V	L	<u> </u>	+	L	<u> </u>	
		<u> </u>	DI 2 II.	2-1- 6 - 1		grams/100 gra		t	
	ES: 1-Dow			Jaic. from de	t. da	ta 5-Calc. by for	mula		
SOURCE:			.PI						
PURIFICAT			PI						
LITERATU:	RE REFERE	NCE	5:						
ļ									
i									
L								į.	

							<b>No.</b> 60			
NAME	2, 2, 3-Tr	imeth	ylhexane			STRUCTURAL	FORMULA			
						CH <sub>3</sub>				
					$\neg \uparrow$	СH <sub>3</sub> - С СН -	(CH <sub>2</sub> ) <sub>2</sub> CH	<sup>[</sup> 3		
Mole	Ref. Me	lecul	ar C <sub>9</sub> H <sub>20</sub>	Molecular Weight 128.2	50	Ċн <sub>3</sub> Ċн <sub>3</sub>				
% Pur.	<u></u>	Ref.		Weight 120.2	Ref	r		Ref.		
D D 46	T	Kei.		Т	Kei	<u> </u>				
F.P. °C F.P. 100%	<del> </del>	$\vdash$	dt/dP *C/mm			f to		1		
B. P. *C		†	25°C	1.551 0.0490	5 2	h .				
760 mm	133.60	2 2	BP t <sub>e</sub>	0.0369	5	f' to				
100 30	71.0 43.	2	30 mm	0.6863	4	g'   'K_				
10 1	23. -11.	2 5	ΔHm cal/g			h'				
Pressure	-11.	+-	AHv cal/g	1		m   300 to	0.0294			
mm 25°C	11.373	5	25°C 30 mm	78.1 75.47	2 5	n 600°K	0.0015 -0.0 <sub>6</sub> 52			
te	1114.	5	BP	64.8	2	70045				
Density g/ml 20°C	0.7292	2	te te (d,e) AHv/T	63.05 63.02	5	m'   700 to	0.1222 0.0013			
dt 25	0.7254	2	ΔHv/T <sub>e</sub>	1	5	0'	-0.0645	4		
<b>4</b> 30	0.7216	4	d   45 to	19.17	4	Surface tension				
a b	0.7444 -0.0 <sub>3</sub> 756	4	e   150 °C		4	dynes/cm. 20°C	21.86 20.96	2		
Ref. Index	-0.03150	+-	d'   10 to		5	40	20.06	2		
n <sub>D</sub> 20°C		2		0.1423	-	Parachor [P]				
25 30	1.4082	2	d g/ml vc ml/g			20°C 30				
"C"	0.7494	4	16 6	300.9	5	40				
MR (Obs.)	43,62	2	P <sub>c</sub> mm	16053.	5		385.2	5		
MR (Calc.)	43.762	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt. u.				
(nD-d/2)	1.0459	2	30 mm	1,0000	5	Dispersion	97.3	2		
Dielectric		<del>↓</del>	BP t <sub>e</sub>	0.9601 0.9490	5	Flash Point °C				
A   40 to		2	t <sub>c</sub>			Fire Point		<u></u>		
c ——	211.0	2	∆Hc kcal/m			M Spec. Ultra V.				
A*   40 to	1.30918 1277.4	5	ΔHf ΔFf		1	X-Ray Dif.				
B* 150 °C	-  1211.4	1	Viscosity			Infrared				
·	4		centistokes			Solubility in + Acetone				
t <sub>k</sub> to	1		η •c			Carbon tet.				
A'   25 to	6.72059	5	ľ			Benzene Ether				
B' L 45 °C	1303.5	5	B <sup>V</sup> to	<del>                                     </del>	-	n-Heptane				
A'* 25 to	<del></del>	5	AV C			Ethanol Water				
	1223.0	5	(BV) to	-		Water in				
Ac   160 to	7, 2477	4	(A <sup>V</sup> ) °C	1						
Bc tc_°C	1678.6 250.9	1	c <sub>p</sub> liq. °K							
Cryos. A°	230.7	┿	c <sub>p</sub> vap.300°K	0.39649	2					
consts. B°		<u>L</u>	- 400	0.50136	2					
t <sub>e</sub> °C	148.67	5	c <sub>v</sub> vap.							
$T_R = 0.7$	T <sub>C</sub>					+ grams/100 gran	ns solvent	t		
	ES: 1-Dow	2-A1	PI <b>3-Lit</b> . 4-0	Calc, from det	t. da	ta 5-Calc. by for				
SOURCE:		A)	PI							
PURIFICAT	'ION:	A1	PI							
LITERATU	RE REFERE	NCES	5:							
	· · · · · · · · · · · · · · · · · · ·									

										No. 61		
NAME	2, 2, 4-Trimethylhexane							STRUCTURAL FORMULA				
							CH <sub>3</sub>					
<u> </u>								С	-сн <sub>2</sub> сн <sub>3</sub>			
Mole		Ref.	Mo	ecul	ar C <sub>9</sub> H <sub>20</sub>	Molecular			сн <sub>3</sub> сн			
% Pur.			For	mula	C9 <sup>11</sup> 20	Weight 128.25	0			-3		
				Ref.			Ref.				Ref.	
F. P. ℃		-120.0		2	dt/dP			f	to			
F.P. 1009	•				°C/mm			g	<u>^K</u> _			
B.P. °C	1				25°C BP	1.17286	2	h				
760 mm	1	126.54 65.	•	2 2	t <sub>e</sub>	0.0370	5	f'	to			
30		38.		2	30 mm	0.6767	4	g'	' <u>*</u> K			
10	ı	17.		2	ΔHm cal/g		-	h'				
1	+	-16.		5	ΔHv cal/g	1	_	m	300 to	0.0294	4	
Pressure mm 25°C	ŀ	15, 53	. 0	5	25°C	75.6	2	n o	_6 <u>0</u> 0_ <u>•</u> K	0.0015		
t <sub>e</sub>	1	083.	•	5	30 mm BP	73.75	5 2			-0.0652	4	
Density	Ť				t <sub>e</sub>	63.4	5	m'		0.1222	4	
g/ml 20°C	:	0.71		2	t <sub>e</sub> (d, e)	61.76	5	n' o'	1000 •K	0.0013 -0.0 <sub>6</sub> 45	4 4	
d <sub>4</sub> 25 30		0.71 0.70		2 4	ΔHv/T <sub>e</sub>	19.14	5		L	6.3	$\vdash$	
a	+	0, 73		4	d 40 to	78.13	4		face tension	20 51	2	
b		-0.03		4	e 140 °C 10 to	0.1164 79.26	4 5	g yn	es/cm. 20°C 30	20.51 19.67	2	
Ref. Index	:				e'   40 °C	0.1463	5		40	18.83	2	
n <sub>D</sub> 20°C	;	1.40		2	d <sub>c</sub> g/ml			Par	achor [P]			
25 30		1.40 1.39		2 4	vc ml/g				20°C 30			
"C"	+	0.75		4	vc ml/g tc °C	288.7	5		40			
MR (Obs.)	+	43.76		2	P <sub>c</sub> mm	15468.	5		Sugd.	385.2	5	
MR (Calc.		43.76		5	PV/RT	1 0000	_	Exp	. L.1.%/wt.			
(nD-d/2)		1.04	55	2	25°C 30 mm	1.0000	5	Dis	u. persion	98.8	2	
Dielectric					RP	0.9601	4	<u> </u>	sh Point C	70.0	÷	
A 35 to		6, 83	91	2	t <sub>e</sub>	0.9496	5		e Point			
B 1_150 °C	-  '	344.0 213.0		2	tc ΔHc kcal/m	<del> </del>		M.	Spec.			
A* 35 to	+	1, 31	004	5	ΔHf				ra V.			
B*  150 °C		256.4	001	5	ΔFf				Ray Dif. ared			
к	1				Viscosity			í <del></del>	ability in +			
	-				centistokes 7 °C			Ac	etone			
t <sub>x</sub> °C	:				'				rbon tet. enzen <b>e</b>			
A'  25 to		6, 63	114	5					her			
B' 40°C	-   1	241.8		5 5	B <sub>v</sub> to				Heptane			
A'* 25 to	+	1, 12	724	5	B to				hanol ater			
B'* 40 °C		1, 12	1 34	5	(B <sup>V</sup> )  to	-			ter in			
Ac   150 to	+	7, 24	30	5	(A <sup>V</sup> )  °C							
Bc tc °C		651.4	-	5			<b>-</b>	1		ł		
Cc	1	252.1		5	P -						1	
Cryos, A° consts. B°		0.06	•	2	cp vap.300°K	0.39649				İ		
	-+	140 (5		$\vdash$	c <sub>v</sub> vap.	0.50136	2					
t <sub>e</sub> °C T <sub>R</sub> = 0.7		140.65		5	V	L	L	L	4100	<u> </u>	<u> </u>	
					DT 2 T::	0.1			ams/100 gra		t	
	υE	5: 1-E	ow		PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula		
SOURCE:				AP								
PURIFICA				AP								
LITERATU	RE	REF	ERE	VC ES	5:							
1												
L												

							No. 62		
NAME	2, 2, 5- <b>T</b>	rimeth	ylhexane			STRUCTURAL FORMULA			
1			· · · · · · · · · · · · · · · · · · ·			CH <sub>3</sub>			
I			·			сн <sub>3</sub> с - (сн <sub>2</sub> ) <sub>2</sub>	CH CH,		
Mole		lolecul	ar C <sub>9</sub> H <sub>20</sub>	Molecular	- 1		CH <sub>2</sub>		
% Pur.	F	ormul	a 09.120	Weight 128.2	50	J			
		Ref.		,  -	Ref			Ref.	
F.P. °C	-105.780	2	dt/dP			f   to			
F.P. 1009	•		*C/mm 25*C	1,109	5	g <u>°K</u>			
B. P. °C 760 mm	124 004	١.	BP	0.04838	2	h			
100	124. 084 63. 092	2 2	t.	0.0372	5	f' to			
30	36.37	2	30 mm	0.6676	4	g'   ' <u>•</u> K_			
10 1	16.17	2 5	ΔHm cal/g			h'			
	-18.2	-	ΔHv cal/g			m   300 to	0.0294		
Pressure mm 25°C	16, 593	5	25°C	74.86	2	n 600 K	0.0015		
te	1066.	5	30 mm BP	74.14	5 2		-0.0 <sub>6</sub> 52	-	
Density			t.	61.44	5	m'   700 to	0.1222	4	
g/ml 20°C			t (d, e)	61.24	5	n' 1000 K	0.0013	4	
dt 25	0.7032		AHv/Te	19.21	5		-0.0 <sub>6</sub> 45		
a 30	0.69922		d   40 to	78.80	4	Surface tension	30.04	_	
ь	-0.0379				4	dynes/cm. 20°C	20.04 19.60	2	
Ref. Index	<del></del>	$\neg$	d'   10 to	76.45 0.0634	5	40	18. 29	2	
n <sub>D</sub> 20°0				0.0031	۲	Parachor [P]			
25 30	1.39728		d g/ml v ml/g			20°C			
"C"		$\overline{}$	t <sub>c</sub> *C	282.1	5	30 40			
	0,7538	4	P <sub>c</sub> mm	15064.	5		385.2	5	
MR (Obs.) MR (Calc.		2 5	PV/RT		Н	Exp. L. 1. %/wt.			
(nD-d/2)	1.04612		25°C	1.0000	5	u.		_	
Dielectric			30 mm BP	1.0000 0.9458	5 4	Dispersion	99.0	2	
A 35 t	6. 83531	2	te	0.9347	5	Flash Point °C Fire Point			
B 1145 °C	1324.049	2	t <sub>c</sub>			M Spec.		<del>                                     </del>	
C	210, 737	2	ΔHc kcal/m ΔHf	1		Ultra V.		ĺ	
A* 35 to		5	ΔFf			X-Ray Dif.			
B* ∟150 °C	1244.71	"	Viscosity	<del> </del>		Infrared		<u> </u>	
° .— —.	_		centistokes			Solubility in + Acetone		l	
t <sub>x</sub>   -t			η •c			Carbon tet.		ĺ	
A'   25 to		3 5				Benzene Ether			
B' 40 °C		5		<u> </u>		n-Heptane		l	
C'	238.4	5	B <sup>V</sup> to			Ethanol		ĺ	
A1# 25 to			A <sup>V</sup> C	.]		Water Water in			
	1534.28	5	(B <sup>V</sup> )  to	}		11 4 4 4 1 1 1			
Ac 145 to	7.2339	5	(A <sup>V</sup> )  °C					ĺ	
Cc c_	248.5	5	c <sub>p</sub> liq. °K					İ	
Cryos. A		2	c <sub>p</sub> vap.300°K	0.39649	2			1	
consts. B			P 400	0.50136	2				
t <sub>e</sub> °C	137.04	5	c <sub>v</sub> vap.	1				ĺ	
$T_R = 0.7$	'5 T <sub>C</sub>					+ grams/100 gran	ns solvent		
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit, 4-0	Calc. from det	da				
SOURCE:		Al							
PURIFICA	TION:	AI	PI						
LITERATU	RE REFER	ENCES	: :						

	-							No. 63	
NAME _	2, 3, 3-Tri	meth	ylhexane			ST	RUCTURAL	FORMUL	A
							CH <sub>3</sub>		
Mole % Pur.	Ref. Mol	ecul	ar C <sub>9</sub> H <sub>20</sub>	Molecular Veight 128.2	50	c	:н <sub>3</sub> сн-с (с сн <sub>3</sub> сн <sub>3</sub>	н <sub>2</sub> ) <sub>2</sub> сн <sub>3</sub>	
<i>N</i> 1 U.1.	1 1 1 1 1 1 1	Ref.	, <u></u>		Ref.	Γ			Ref.
F. P. °C	-116,800	2	dt/dP			f	to		
F.P. 100%			°C/mm 25°C	1.749	5	g	<u>•</u> K_		
B.P. °C 760 mm	137.68	2	BP	0.0505	4	h			
100 30	74. 46.	2 2	t <sub>e</sub> 30 mm	0.0371 0.6982	5	f' g'	to		
10	25.	2	ΔHm cal/g	0.0782		h'			
Pressure	-9.	5	ΔHv cal/g		_	m	300 to	0.0294	4
mm 25°C	10.01	5	25°C 30 mm	78.7 75.38	2	n o	1_6 <u>0</u> 0_•K	0.0015 -0.0 <sub>6</sub> 52	4
t <sub>e</sub>	1130.	5	BP	65.2	2	m'	700 to	0.1222	4
Density g/ml 20°C	0.738	2	te te (d, e)	63.43 63.43	5	n' o'	1000 •K	0.0013	4
dt 25 4 30	0.734 0.730	2	AHv/Te	19.06	5	o.		-0.0 <sub>6</sub> 45	4
a 30	0.754	4	d 45 to	80.49	5		face tension es/cm. 20°C	22.41	2
ъ	-0.0 <sub>3</sub> 797	4	d 155 ℃ to	0.1110 82.65	5	8,	30	21.49	2
Ref. Index	1.4141	2	e'   45 °C	0.1581	5	Par	achor [P]	20.59	2
25 30	1.4119 1.4092	2	d g/ml vc ml/g			1 ***	20°C		
"C"	0.7467	4	tc°C	307.8	5		30 <b>4</b> 0		
MR (Obs.)	43.4	2	P <sub>c</sub> mm	16059.	5			385.2	5
MR (Calc.) (nD-d/2)		. 5 . 2	PV/RT 25°C	1.0000	5	Exp	u. L.1.%/wt.		
Dielectric	1.045	-	30 mm BP	1.0000 0.9629	5		persion	96.3	2
A   45 to	6.8474	2	t_	0.9517	5		sh Point °C e Point		
B (165 °C	1391.0 213.0	2	tc AHc kcal/m			<u> </u>	Spec.		
A* 45 to	1.29980	5	ΔHf				ra V. Ray Dif.		
B*[165 °C	1298.6	5	ΔFf Viscosity		<u> </u>		ared		
c			centistokes				ability in +		
t <sub>k</sub> to to			η °C			Ca	rbon tet.		
A'   25 to	6,55736	5					nzene her		
B' 45 °C	1244.81 199.0	5 5	B <sup>v</sup>   to A <sup>v</sup>   °C				Heptane hanol		
A!# 25 to	1.05181	5				Wa	ater		
B'* 45 °C Acl 165 to	7 2550	5	(B <sup>V</sup> )  to				ater in		├
Bc tc °C	7. 2550 1713. 7	5 5	(A <sup>V</sup> )  °C c_liq, °K						
Cc — —	254.3	5	Р.						
Cryos. A° consts. B°	0.0447	2	c vap.300°K 400	0.39649 0.50136					
t <sub>e</sub> °C	153.61	5	c <sub>y</sub> vap.	L		L		L	<u>Ļ</u>
T <sub>R</sub> = 0.77		2-4	PI 3-Lit. 4-0	Calc from de	t d=		ams/100 gra		t
SOURCE:	E.S. 1-DOW		PI	Carc. Hom de	t. ua		-Caic. by 101	IIIuIa .	
PURIFICAT	ION:		PI						
LITERATUR	E REFERE	CES	3:						
i									

							No. 64	
NAME	2, 3, 4	-Trimeth	ylhexane			STRUCTURAL	FORMULA	1
						сн <sub>3</sub>		
			T		$\neg$	сн <sub>3</sub> сн сн сн		
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 128.2	50	Ċн <sub>3</sub> сн	<sup>1</sup> 3	
- N T UI.		Ref	1 / 20 1	weight 120.2	Ref	1		Ref.
F.P. °C		- Ker	46/47	T	Ker	<del></del>	-	
F. P. 100	6		dt/dP *C/mm	l		f to		
B. P. °C			25°C	1.914	5 2	h .		
760 mm	139.0	2	BP t <sub>e</sub>	0.050 0.0365	5	f' to		
30	76. 48.	2 2	30 mm	0, 6952	4	g'   '°K_		
10 1	27.	2 5	ΔHm cal/g			h'		
Pressure	-7.		ΔHv cal/g			m   300 to	0.0294	
mm 25°C	9.00	0 5	25°C	80.0	2	n 600 °K	0.0015 -0.0 <sub>6</sub> 52	
t <sub>e</sub>	1144.	5	30 mm BP	76.57 66.5	5 2			
Density	2 22		t <sub>e</sub> (d.e)	64.75	5	m'   700 to n'   1000 °K	0.1222 0.0013	4
g/ml 20°0	0.73 0.73		e (4, 6)	64.70	5	0'	-0.0645	
dt 25 4 30	0.73		ΔHv/T <sub>e</sub>	19.38	5	Surface tension		<u> </u>
ŧ	0.75		d   50 to		4	dynes/cm. 20°C	22.80	2
Bot Inde	-0.03	76 4	d' 10 to	83.76	5	30 40	21.87 20.94	2 2
Ref. Inde		44 2	e' j 50 °C	0.1505	5	Parachor [P]		_
25	1.41	20   2	d g/ml v ml/g			20°C		
"C"	1.40	-	tc °C	310.8	5	30 40		
	0.74		P <sub>c</sub> mm	16713.	5	II .	385.2	5
MR (Obs. MR (Calc			PV/RT		_	Exp. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	96.4	2
Dielectric			BP	0.9689	4	Flash Point °C	70	Ē
A   45 t			t <sub>e</sub> t <sub>c</sub>	0.9594	5	Fire Point		
B L165°	211.0	2 2	∆Hc kcal/m	-	-	M Spec.		
A*   45 t	1,30	76 5	ΔHf	ļ		Ultra V. X-Ray Dif.		
B* 165°	<u>C</u> 1300.	5	ΔFf	<u> </u>		Infrared		
c			Viscosity centistokes			Solubility in +		
t <sub>k</sub>			η •⊂			Acetone Carbon tet.		
A'   25 t	1	57 5		İ		Benzene		
B' _ 50 °	C 1302.	5				Ether n-Heptane		
C'	202.3	5	B <sup>V</sup>   to A <sup>V</sup>   °C			Ethanol		
A'* 25 t B'* 50 °		38 5 5		-		Water Water in		
Ac   165 t	<del></del>			1				
Bc t °	C 1715.5	5		<del>                                     </del>				
Ce	251.9	5	c <sub>p</sub> liq. ∘K					
Cryos, A			c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2			
t <sub>e</sub> °C	155. 27	5	c <sub>v</sub> vap.	0.50136	<b> </b>			
$T_R = 0.5$			1	1		+ grams/100 gran	ne eclus-4	<u> </u>
REFEREN		ow 2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-Calc. by for	mula mula	
SOURCE:		AF				5 CLIC, by 1017		
PURIFICA	TION:	AF						
LITERATU								
]								
1								

NAME	2,3,5-	Trimeth	ylhexane		STRUCTURAL FORMULA			
						CH <sub>3</sub>		
	2.6			(-11		сн <sub>3</sub> сн сн сн		
Mole % Pur.		Molecula Formula		Molecular Weight 128.2	50	Ċн <sub>3</sub>	Ċн <sub>3</sub>	
		Ref.	, ————————————————————————————————————	T	Ref.			Ref.
F. P. °C	-127.8	2	dt/dP			f to		
F.P. 1009			°C/mm			g  *K		
B. P. °C			25°C	1.442	5	h i		
760 mm	131.34	2	<b>B</b> P	0.049 0.0369	2 5	f' to		
100	69. 42.	2 2	t <sub>e</sub> 30 mm	0.6814	4	g'K		
10	21.	2		0.0014		h'		
1	-12.9	5	ΔHm cal/g		-	m   300 to	0.0294	4
Pressure	12.25/		ΔHv cal/g 25°C	77.19	2	n 600 K	0.0015	4
mm 25°C	12.376 1104.	5	30 mm	75.27	5	°	-0.0652	4
Density	+	-	BP	64.9	2	m¹ 700 to	0.1222	4
g/ml 20°0	0.7219		t <sub>e</sub> (d, e)	63, 21	5	n'   1000 °K		
at 25	0.7179		AHV/Te	19.38	5		-0.0 <sub>6</sub> 45	4
	0,7139		d 40 to	80,13	4	Surface tension	.,	
a b	-0.0379		e 1 145 °C	0.1159	4	dynes/cm. 20°C	21.27	2 2
Ref. Index			d' 10 to e' 40 °C	80.03	5	40	19.49	2
n <sub>D</sub> 20°0		1 2		0.1137	3	Parachor [P]		
- 25	1.403		d <sub>c</sub> g/ml v <sub>c</sub> ml/g			20°C		
30	1,4013		vc ml/g tc °C	295.6	5	30 40		
"C"	0.7494		P <sub>c</sub> mm	15745.	5		385.2	5
MR (Obs. MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0450		25°C 30 mm	1.0000	5	u.	97.8	١,
Dielectric	:	$\neg$	BP	0.9586	5	Dispersion	97.8	2
A 40 to	6,850	5 2	te	0.9475	5	Flash Point C Fire Point		
B (155°C		2	t <sub>c</sub>			M. Spec.		<del>                                     </del>
C	211.0	2	ΔHc kcal/m ΔHf			Ultra V.		
A* 40 to B* 155 °C		39 5	ΔFf			X-Ray Dif. Infrared		
K C	=		Viscosity			Solubility in +		+-
t, to	-1		centistokes 7 °C			Acetone		
t <sub>k</sub> to			໊າ °⊂			Carbon tet.		
A'  25 to	6.9602	29 5				Benzene Ether		1
B' _45 °C	1415.10	5				n-Heptane		
C'	216.2	5	B <sup>V</sup> to oC			Ethanol Water		
A'* 25 to B'* 45 °C		69   5	(B <sup>V</sup> )  - to	1		Water in		
Acl 155 to			(A <sup>V</sup> )  °C					
Bc tc °C	1667.6	5		<del>                                     </del>		1		
Cc	250.3	5	р.				1	
Cryos. A			c <sub>p</sub> vap.300°K	0.39649				
consts. B		+_	c vap.	0.50136	2			
t <sub>e</sub> °C	145.96	5	V	L	L	1	L	<u> </u>
$T_R = 0.$						grams/100 gra		ıt
	CES: 1-Dov			Calc. from de	t. da	ata 5-Calc. by for	mula	
SOURCE:		AP						
PURIFICA		AP						
LITERATU	JRE REFER	RENCES	<b>5:</b>					
1								
L								

									No. 66	
NAME	2,4,4	-Tri	meth	ylhexane			STR	UCTURAL I	FORMULA	
								сн		
						$\neg$	CH	3CH CH2C	CH2CH3	
Mole % Pur.	Ref.		ecul: mul		Molecular Weight 128.2	250		ċн <sub>3</sub> ċн	3	
7			Ref.	<del> </del>		Ref				Ref
F.P. *C	-113.38	0	2	dt/dP	T		f	to		
F.P. 100%				°C/mm	1	1	g	<u>•</u> K_		ĺ
B. P. °C				25°C BP	1.356 0.0496	5 2	ь			
760 mm 100	130.64		2 2	t <sub>e</sub>	0.0373	5	f'	to		
30	40, 62		2	30 mm	0.6860	4	8'	•K_		İ
10 1	19.86		2	ΔHm cal/g			h'			<u> </u>
Pressure	1			ΔHv cal/g	74.		m   n	300 to	0.0294 0.0015	
mm 25°C	13.4		5	25°C 30 mm	76.1 74.15	5	ō.		-0.0 <sub>6</sub> 52	
t <sub>e</sub> Density	1096.		5	BP	63.9	2 5	m' i	700 to	0,1222	4
g/ml 20°C	0.72	381	2	te t (d, e)	62.41	5	n' i	11000 °K	0.0013	4
d <sub>4</sub> 25	0.72		2	ΔHv/T	19.14	5	6		-0.0 <sub>6</sub> 45	4
	0.71		5	d   40 to	<del> </del>	4		ce tension		
a b	-0.03		5	_e  145 °C		4 5	dyne	s/cm, 20°C	21.17 20.33	2 2
Ref. Index				d'   10 to		5		40	19.49	2
<sup>n</sup> D 20°C	1.40		2 2	d <sub>c</sub> g/ml			Para	chor [P] 20°C		İ
30	1.40		4	I V mi/g	295.6	5		30		
"C"	0.74	99	4	-	15689.	5		40 Suad	385.2	5
MR (Obs.)			2	P <sub>c</sub> mm	13007.	-	Evn	L. 1. %/wt.	305.2	Ľ
MR (Calc.) (nD-d/2)	43.76 1.04		5 <b>2</b>	25°C	1.0000	5	1	u.		
Dielectric				30 mm BP	1.0000 0.9533	5		ersion	98.2	2
A   40 to	6,85	163	2	t <sub>e</sub>	0.9421	5		h Point °C Point		
B 1155 °C			2	t <sub>c</sub>	ļ		M Sp			
A*  40 to	1.32	-	5	ΔHc kcal/m ΔHf		1	Ultra	v.		
B* 155 °C			5	ΔFf			Infra	y Dif. red		ĺ
K — — —				Viscosity centistokes			Solub	ility in +		
ել				7	1			tone bon tet.		ĺ
t <sub>x l</sub>	<u> </u>			'			Ben	sene		
A' 25 to B' 40 °C	6.79 1339.10		5				Ethe	er eptane		
<u>c'                                    </u>	211.3		5	B <sup>V</sup> to			Eth	nol		
A'* 25 to B'* 40 °C	1.27 1255.58		5 <b>5</b>	AV I C	-		Wat Wat	er er in		
Ac   155 to	<del></del>		5	(B <sup>V</sup> ) to	1					
Bc t C	1684.3	~	5		+	$\vdash$	ļ			
Cc	254.2		5	c <sub>p</sub> liq. ∘K						
Cryos. A° consts. B°	0.05		2	c <sub>p</sub> vap.300°K	0.39649 0.50136	2				
t <sub>e</sub> °C	145.03		5	c <sub>v</sub> vap.			L			
$T_R = 0.7$								ms/100 gran		<u> </u>
REFERENCE	ES: 1-D	ow :	Z-AF		Calc. from de	t. da	ta 5-0	Calc. by form	nula	
SOURCE:	TON:		AF	<del></del>						
LITERATU		CD EN	AF							
	RE REF	SK EN	ICES	•						

NAME	3, 3, 4-7	Trimeth	ylhexane			STI	RUCTURAL CH <sub>3</sub>	FORMUL.	A
Mole % Pur.		Molecula 'ormula		Molecular Weight 128.2	250	CH	<sub>13</sub> сн <sub>2</sub> с - сн сн <sub>3</sub> сн	сн <sub>2</sub> сн <sub>3</sub>	
		Ref.			Ref.				Ref.
F.P. °C F.P. 1009	-101.20	2	dt/dP °C/mm			f g	to °K		
B. P. °C 760 mm	140.46	2	25°C BP	1.983 0.049	5 2	h   f'			
100 30 10	77. 48.	2 2 2	t <sub>e</sub> 30 mm	0.0375 0.7011	5 4	g'   h'	to		
i	27. -7. 90	5	ΔHm cal/g		ļ	<b>├</b> ──	7200 4-	0.0304	
Pressure mm 25°C	8.8191 11 <b>2</b> 1.	5 5	ΔHv ca1/g 25°C 30 mm BP	78.8 76.23 65.5	2 5 2	m o	300 to	0.0294 0.0015 -0.0 <sub>6</sub> 52	4 4
Density g/ml 20°0 dt 25 d4 30	0.7454 0.7414	2	t <sub>e</sub> t <sub>e</sub> (d, e) ΔΗν/Τ <sub>e</sub>	63.85 63.67 19.07	5 5	n'   o'	700 to 1000 °K	0.1222 0.0013 -0.0 <sub>6</sub> 45	4 4
a b	0.7374 0.7614 -0.0 <sub>3</sub> 80	4	d 50 to e 155 °C	81.89 0.1167	4 4		s/cm. 20°C	23. 27 22. 31	2 2
Ref. Index		2	d' 10 to e' 50 °C	81.54 0.1094	5	Para	40 schor [P] 20°C	21.35	2
"C"	1.4130 0.7458	-	vc ml/g tc °C	312.0	5		30 40	205 2	
MR (Obs. MR (Calc.	43.762	2 5	P <sub>c</sub> mm PV/RT 25°C	1,0000	5	Exp.	L. l. %/wt.	385.2	5
(nD-d/2) Dielectric			30 mm BP	1.0000 0.9514	5 4 5		ersion h Point <sup>6</sup> C	95.3	2
A 45 to B 165 °C		2 2 2	te tc ΔHc kcal/m	0.9391	•	М. 5	Point		
A* 45 to B* 165 °C		2 5	ΔHf ΔFf Viscosity			Infra	ay Dif.		
			centistokes 7 °C			Ace Car	bility in tetone rbon tet.		
A'   25 to B'   50 °C		6 5 5 5	BV   to			Eth n-F Eth	er Ieptane anol		
A'* 25 to B'* 50 °C		6 5	(B <sup>V</sup> )  to		ļ	Wa:	ter ter in		
Acl 165 to Bc t <sub>c</sub> °C	7. 2624 7. 2624 7. 253. 4	5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K						
Cryos, A consts. B	<u>'</u>	2	c vap.300°K	0.39649 0.50136	2 2				
t <sub>e</sub> °C	156. 12	5	c <sub>v</sub> vap.	L	<u> </u>	L			<u> </u>
$T_R = 0$ .							ms/100 gra		t
	CES: 1-Dov		PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE:			PI				- 1. · · · · · · · · · · · · · · · · · ·		
PURIFICA			PI						
LITERATU	JRE REFER	ENCES	<b>:</b>						
L									

						<u> </u>	No. 68	
NAME	3, 3-Diet	hylpent	ane			STRUCTURAL	FORMULA	
					$\neg \neg$	C <sub>2</sub> H <sub>5</sub>		
			····			сн,сн,с с	H.CH	
Mole	Ref. 1	Molecul	ar a	Molecular			203	
% Pur.		Formul		Weight 128.2	50	Ċ <sub>2</sub> н <sub>5</sub>		
		Ref.			Ref.			Ref
F, P. °C	-33.110	2	dt/dP			f to		
F.P. 100%			°C/mm	1	1	g L°K		1
B. P. *C			25°C BP	2.379 0.05153	5	h .		1
760 mm 100	146.168	2 2	t <sub>e</sub>	0.0371	5	f' to		
30	52.40	2	90 mm	0.7164	4	g'   'K		
10	30.71	2	ΔHm cal/g	<del>                                     </del>	H	h'		
1	-4.20	5	ΔHv cal/g	<del> </del>		m   300 to	0.0294	4
Pressure mm 25°C	7 140	5	25°C	80.8	2	n600 °K		
t <sub>e</sub>	7.168	5	30 mm	76.44	5	° ;	-0.0652	4
Density	<del> </del>	+	BP t	67.1 65.46	5	m¹   700 to	0.1222	
g/ml 20°C	0.7535		te te (d, e)	65.38	5	n' 1000 °K		
dt 25 4 30	0.7500		ΔHv/T	19.23	5	o'	-0.0 <sub>6</sub> 45	4
	0.7464	$\overline{}$	d   50 to		4	Surface tension		
a b	0.7679		_e <u>  165</u> ℃		4	dynes/cm. 20°C		2 2
Ref. Index	1 2.03/1	-   -	d'   10 to		5	40	22.83	2
n <sub>D</sub> 20°C	1.4205	1 2		0.1591	5	Parachor [P]	<del>                                     </del>	
D 25	1.4183	7 2	d g/ml v ml/g			20°C		
30	1,4162		vc ml/g tc °C	324.3	5	30 40		
"C"	0.7419	_	P <sub>c</sub> mm	17130.	5		. 385.2	5
MR (Obs.)	43.113	2	PV/RT	<del>                                     </del>	$\vdash$	Exp. L.1.%/wt.	+ · · · ·	Ť
MR (Calc.) (nD-d/2)	43.762 1.0437	2 2	25°C	1.0000	5	u.		
Dielectric	<u> </u>	+-	30 mm BP	1.0000 0.9643	5	Dispersion	94.3	2
A 50 to	6.8926	2 2	t <sub>e</sub>	0.9526	5	Flash Point °C		
B <u>175 °C</u>		2	tc		<u>L_</u>	Fire Point	<del> </del>	$\vdash$
<u>c</u>	215.575	2	ΔHc kcal/m	1		M Spec. Ultra V.		
A*   50 to	1.3300		ΔHf ΔFf			X-Ray Dif.		
B* ∟175 °C	1353.96	5	Viscosity	<del> </del>	<u> </u>	Infrared	1	<u>L</u>
c	_		centistokes	1	1	Solubility in +		
tk to		jj	η •c			Acetone Carbon tet.		
x '	L	<del>,   _  </del>				Benzene		1
A'   25 to B'   55 °C	6.5475 1272.18	2 5 5				Ether n-Heptane		
c,	198.5	5	B <sup>V</sup> to			Ethanol		
A1* 25 to	1.0368		AV   °C	_{	1	Water		
	1194.11	5	(B <sup>V</sup> ) to		1	Water in		-
Ac   175 to	7.3074		(A <sup>V</sup> )  °C		<u>L</u> _	]		
Bc tc_°C	259.2	5	cp liq. °K					
Cryos. A°	0.0223	-+	1 -	0.30440	,			
consts. B°	"."."		c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2			
t <sub>e</sub> °C	163.42	5	c <sub>w</sub> vap.		-			
$T_{R} = 0.7$			L	L	Ь	grams/100 gra	me solve-	L
	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	t. de	ta 5-Calc. by for		•
SOURCE:		AF				5 Cuic. by 101		
PURIFICAT	ION.	AF	<del></del>					
LITERATU	KE KEFER	ENCES	<b>:</b>					

NAME	3-Ethyl-2	, 2 - di	imethylpentane		STRUCTURAL CH3			
Mole % Pur.	Ref. Mo	lecul mula		Molecular Weight 128.2	250	сн <sub>3</sub> с - сн - с сн <sub>3</sub> с <sub>2</sub> н <sub>5</sub>	сн <sub>2</sub> сн <sub>3</sub>	
		Ref.			Ref.			Ref.
F. P. °C	-99.2	2	dt/dP			f to		
F.P. 1009			°C/mm			g  °K		
B. P. ℃			25°C	1.521	5	h i	]	1
760 mm	133.83	2	<b>B</b> P	0.050	2 5			<del>  -</del>
100	71.	2	t <sub>e</sub>	0.0371	1 1	g' to	ł	
30 10	43.	2 2	30 mm	0.6905	<b>↓⁴</b> _	1 1		1
1	22. -10.86	5	∆Hm cal/g			h' i		-
Pressure			∆Hv cal/g			m   300 to	0.0294 0.0015	4
mm 25°C	11.572	5	25°C	78.3	2	"   ' <u></u>	-0.0652	4
t <sub>e</sub>	1114.	5	30 mm BP	74.88 64.9	5 2		6	
Density			t <sub>e</sub>	63.36	5	m' 700 to	0.1222	4
g/ml 20°0		2	te (d, e)	63.23	5	n' 1000 •K		4
d <sub>4</sub> 25	0.7310	2	AHV/Te	19.25	5		-0.0645	1
	0.7272	4	d 45 to	79.63	4	Surface tension		
a L	0.74998	4 4	e   150 °C	0.1101	4	dynes/cm. 20°C	22.38	2
ь	-0.03756	*	d' 10 to	83.01	5	8 30 40	21.45	2 2
Ref. Index		2	e'   45 °C	0.1883	5	<u> </u>	20.32	+
<sup>n</sup> D 20°C	1.4102	2	d <sub>c</sub> g/ml		1 1	Parachor [P] 20°C		
30	1.4077	4	V <sub>C</sub> m1/g	202.2	_	30		
"C"	0.7469	4	tc °C	302.2	5	40		
MR (Obs.		2	P <sub>c</sub> mm	16099.	5	Sugd.	385.2	5
MR (Calc.		5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0449	2	25°C 30 mm	1.0000	5	u.	0, 0	
Dielectric			BP	1.0000 0.9599	5	Dispersion	96.3	2
A 40 to	6, 8482	2	te	0.9489	5	Flash Point C		
B (160 °C		2	t c			Fire Point		<b>_</b>
c	213.0	2	AHc kcal/m			M. Spec. Ultra V.		
A* 40 to	1,31046	5	ΔHf			X-Ray Dif.		
B*[160 °C	1286.16	5	ΔFf			Infrared		
к — — - c			Viscosity centistokes			Solubility in +		
t <sub>k</sub> -tō	-		η °C			Acetone		1
t <sub>x</sub> °C	;		'		1	Carbon tet. Benzene		
A'   25 to	6. 34515	5			1	Ether		
B' 1_45 °C		5		<del>                                     </del>	+	n-Heptane		
C'	189.0	5	B <sup>V</sup> to A <sup>V</sup> °C			Ethanol		
A'* 25 to		5		.		Water Water in		
B'* 45 °C	<del></del>	5	( <b>B</b> <sup>V</sup> )  to				<del> </del>	+-
Acl 160 to		5	(A <sup>V</sup> )  °C	L				
Bc tc °C	253.8	5	c <sub>p</sub> liq. °K					
Cryos, A		$\vdash$		0.30440	,			
consts. B			c <sub>p</sub> vap.300°K	0.39649 0.50136	2			
t <sub>e</sub> °C	149.00	5	c <sub>v</sub> vap.					
$T_{R} = 0.7$		لـتــا	<u> </u>	L		+ arama/100	ma a a a l == =	<del>_</del>
	CES: 1-Dow	2 4	DT 2 T-4 4	Colo ( )		grams/100 gra		10
	CE3; 1-D0W			Caic, irom de	ει. αa	ta 5-Calc. by for	ınuıa	
SOURCE:			.PI					
PURIFICA			.PI					
LITERATU	IRE REFERE	NCES	<b>5:</b>					

							No. 70		
NAME	3-Ethyl-2	, 3-dir	nethylpentane			STRUCTURAL FORMULA			
						CH <sub>3</sub>			
<u> </u>						сн,-сн с	CH <sub>2</sub> CH <sub>3</sub>		
Mole % Pur.	Ref. M	olecul ormul	аг С <sub>9</sub> Н <sub>20</sub>	Molecular Weight 128.	250	ċн <sub>3</sub> ċ <sub>2</sub> н <sub>5</sub>			
<del></del>		Ref			Ref	r		Ref.	
F, P. *C	1	1	dt/dP	Τ		f to		$\vdash$	
F.P. 100%			°C/mm			g LK		1	
B. P. °C	1,42		25°C BP	2.0224 0.050	5 2	h			
760 mm 100	142. 77.	2 2	t <sub>e</sub>	0.0374	5	f¹ to			
30 10	49. 28.	2 2	30 mm	0.7083	4	g'   '°K_			
l i	-7.09	5	ΔHm cal/g			h'	0.0204	<u> </u>	
Pressure			ΔHv cal/g 25°C	79.3	2	m   300 to   600 °K	0.0294 0.0015		
mm 25°C	8.591 1138.	5	30 mm	75.71	5	•	-0.0 <sub>6</sub> 52	4	
Density	1130.	+	BP	65.8 64.22	2 5	m'   700 to	0,1222	4	
g/ml 20°C	0.754	2	t <sub>e</sub> (d, e)	64.05	5	n' 1000 K	0.0013		
dt 25 4 30	0.750	2 4	AHv/Te	19.08	5		-0.0645	4	
. 50	0.76999		d   50 to		4	Surface tension dynes/cm, 20°C	23.87	2	
ь	-0.03797		$\frac{160}{d}$ $\frac{1}{10}$ $\frac{160}{to}$		4 5	30	22.94	2	
Ref. Index			e'   10 to		5	40	22.00	2	
n <sub>D</sub> 20°C	1.419	2 2	d <sub>c</sub> g/ml			Parachor [P] 20°C			
30	1.415	4	t <sub>c</sub> *C	316.4	5	30		l	
"C"	0.7390	4	P <sub>c</sub> mm	16454.	5	40 Sugd.	385.2	5	
MR (Obs.) MR (Calc.)	43.0 43.762	2 5	PV/RT	<b></b>		Exp. L.1.%/wt.		<u> </u>	
(nD-d/2)	1.042	2	25°C 30 mm	1.0000 1.00 <b>0</b> 0	5	u.			
Dielectric			BP	0.9589	5	Dispersion Flash Point °C	94.6	2	
A 45 to		2	t <sub>e</sub> t <sub>c</sub>	0.9477	5	Fire Point			
B <u>170</u> ℃	214.	2 2	ΔHc kcal/m		-	M Spec.			
A*   45 to	1.30518	5	ΔHf			Ultra V. X-Ray Dif.			
B* 170 °C	1321.20	5	ΔFf		-	Infrared			
c			Viscosity centistokes		1	Solubility in +			
tk to			η •c			Acetone Carbon tet.		1	
t C A'   25 to	6,60638	5	-	1		Benzene		Ì	
B' _ 50 °C	1287.24	5		ļ		Ether n-Heptane			
C'	201.9	5	B <sup>V</sup> to °C	ĺ		Ethanol Water		ŀ	
A'* 25 to B'* 50 °C	1.09388	5	(BV) - to	-		Water in			
Ac   170 to	7, 2640	5	(A <sup>V</sup> ) °C	1					
Bc t °C	1746.3	5	c <sub>p</sub> liq. °K	<del> </del>	$\vdash$				
Cryos, A°	256.6	+ "	_	0.20(12					
consts. B°			c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2				
t <sub>e</sub> °C	158.42	5	c <sub>v</sub> vap.						
$T_{R} = 0.79$						+ grams/100 gram	ns solvent		
	ES: 1-Dow	2-AF		Calc, from de	t. dat	ta 5-Calc, by forn			
SOURCE:		AF	<del></del>						
PURIFICAT		AF							
LITERATU	RE REFERE	ENCES	i:						

No. 71 3-Ethyl-2, 4-dimethylpentane NAME STRUCTURAL FORMULA C2H5 CH3CH-CH CH CH3 Ref. Mole Molecular Molecular Ċнз C9H20 ĊH3 % Pur. Weight 128, 250 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -122.2 2 dt/dP f to °C/mm 25°C g \_°<u>K</u> 1.684 5 B. P. °C h BP 0.0503 760 mm 136,73 2 2 2 t<sub>e</sub> 0.0366 5 ſ١ to 100 73. g' °<u>K</u> 30 0.6959 45. 30 mm 4 10 24. 2 h' ∆Hm cal/g -8.42 m 300 to 0.0294 AHv cal/g Pressure n 600 °K 0.0015 25°C 80.0 2 mm 25°C 10.228 o -0.0652 4 30 mm 75.34 5 1145. t<sub>e</sub> ΒP 66.0 2 m 700 to 0.1222 4 Density 64.43 5 5 te te (d, e) 0.0013 n' [10<u>00 °</u>K g/ml 20°C 0.7379 2 64.32 ٥' -0.0645 4 0.7341  $d_4^t$ 25 2 AHv/Te 19.40 5 30 0.73031 4 Surface tension d 50 to 79.98 4 0.75308 4 dynes/cm. 20°C 22.80 2 150 ℃ 0.1022 4 ь -0.0<sub>3</sub>757 4 21.87 2 30 ă٦ 10 to 85.71 20.94 2 40 Ref. Index 50 °C 0,2286 5 20°C 1.4137 n D [P] Parachor d<sub>c</sub> g/ml 25 1.4115 2 20°C vc ml/g 30 1.4091 4 30 t<sub>c</sub> °C 307.6 5 40 "C" 0.7461 4 P<sub>c</sub> mm 16350. 5 Sugd. 385. 2 5 MR (Obs.) 43.40 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 43.762 5 25°C 1.0000 5 (nD-d/2) 1.0488 2 30 mm 1.0000 5 Dispersion 96.4 2 Dielectric ВP 0.9753 5 Flash Point C 6.8524 te tc 0.9650 A 45 to 2 Fire Point 1389.0 B ∟165 °C 2 M. Spec. 213.0 2 AHc kcal/m Ultra V. **AHf** A\* 45 to 1.2849 5 X-Ray Dif. ΔFf B\*[165 °C 1291.5 Infrared Viscosity Solubility in centistokes c Acetone to Carbon tet. œ °C t^ Benzene 00 A' 25 to 6.10396 Ether œ B١ 45 °C 1029.13 n-Heptane œ B<sup>V</sup> A 5 177.0 to Ethanol œ A'\* °C Water 25 to 0.62992 5 B'\* Water in 45 °C 965.97 5 (B<sup>V</sup>) to Acl 165 to 7,2601 5 (A<sup>V</sup>)| °C Bc tc °C 1711.9 c<sub>p</sub> liq. ۰ĸ Cc 5 254.4 Cryos. A° consts. B° c<sub>p</sub> vap300°K 0.39649 2 0.50136 2 c vap. te °C 5 153.13  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 72	
NAME	2, 2, 3, 3-7	[etra	methylpentane			STRUCTURAL	FORMULA	<b>L</b>
						CH <sub>3</sub> CH <sub>3</sub>		
						сн <sub>3</sub> с - с - с	$H_2CH_3$	
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 128.2	50	Ċн <sub>3</sub> Ċн <sub>3</sub>		
		Ref.	F		Ref.			Ref.
F.P. ℃	-9.90	2	dt/dP			f to		
F.P. 1009			*C/mm 25*C	1.893	5	g <u>*K</u> _		1
B. P. °C 760 mm	140, 274	2	BP	0.05124	2	h		_
100	75.705	2	t <sub>e</sub>	0.0372	5	f' to		
30 10	47.43 26.05	2 2	30 mm	0.7064	4	h'		İ
1	-10.3	5	ΔHm cal/g		-	m   300 to	0.0294	4
Pressure mm 25°C	0.535	_	ΔHv cal/g 25°C	76.4	2	n   _600 °K_	0.0015	4
t <sub>e</sub>	9.525 1150.3	5	30 mm BP	75.17 65.7	5 2	° .i	-0.0 <sub>6</sub> 52	4
Density			t.	63.93	5	m'   700 to n'   1000 °K	0.1222 0.0013	4
g/ml 20°C	0.75666 0.75299	2 2	l e (d, e)	63.98	5	o' 11000 X	-0.0645	4
dt 25 4 30	0.74932	4	ΔHv/T <sub>e</sub>	19.05	5	Surface tension		<del>                                     </del>
a L	0,77133	4	d   50 to e   160 °C	80.00 0.1020	4	dynes/cm. 20°C	23.38	2
b Ref. Index	-0.03731	4	d'   10 to	77.77	5	30 40	22.48 21.60	2 2
n <sub>D</sub> 20°0		2	e' j 50 °C	0.0550	5	Parachor [P]		
25 30	1.42140	2 4	d g/ml vc ml/g			20°C 30		
"C"	0,7440	4	t <sub>c</sub> *C	316.6	5	40		ĺ
MR (Obs.)	<del></del>	2	P <sub>c</sub> mm	16709.	5		385,2	5
MR (Calc. (nD-d/2)	) 43.762	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric	1.04527	2	30 mm BP	1.0000 0.9714	5	Dispersion	95.9	2
A   45 t	6. 82876	2	t <sub>e</sub>	0.9606	5	Flash Point °C Fire Point		
B [180 °C	1397.483	2	t <sub>c</sub>			M Spec.		_
C 45 45 4	213.703	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A*  45 to   B* <sub> </sub> 170 °(		5	ΔFf			X-Ray Dif. Infrared		
к ——-	-[	1	Viscosity		[ ]	Solubility in +		
k		į	centistokes 7 °C	1		Acetone		
'x		L	•		1	Carbon tet. Benzene		
A'   25 to B'   50 °C		5				Ether n-Heptane		
C'	242.0	5	B <sup>V</sup> l to			Ethanol		
A'* 25 to B'* 50 °C		5	AV I C			Water Water in		
Ac   180 to	<del></del>	5	(B <sup>V</sup> ) to					
Bc t °C	1788.9	5			-			
Cc — —	265.7	5	c <sub>p</sub> liq. °K					
Cryos. Acconsts. B		2	c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2			
te ℃	157.18	5	c <sub>v</sub> vap.					
$T_R = 0.7$						grams/100 grai		<u> </u>
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION.	AF	<del></del>					
PURIFICA	RE REFERE	AF				<del></del>		
LILERAIU	ne refekt	NUES	);					

									No. 73	<del>,</del>
NAME	2, 2,	3,4-7	Γetra	methylpentane			STRUCTURAL FORMULA			
							CH <sub>3</sub> CH <sub>3</sub>			
ļ		Γ				$\dashv$	c	н,с - сн-с	н сн.	
Mole	Ref.		lecul	ar C <sub>9</sub> H <sub>20</sub>	Molecular			-, ,	3	
% Pur.		For	mula	<del>,</del>	Weight 128.2	-		CH <sub>3</sub> CH <sub>3</sub>		_
	-		Ref.			Ref.	<u> </u>			Ref.
F.P. °C	-121.0	9	2	dt/dP			f	to		
F.P. 1009	<u> </u>			°C/mm 25°C	1.434	4	g	' <u>°K</u>	-	
B. P. °C 760 mm	133.0	16	2	BP	0.05030	2	h			_
100	69.6		2	t <sub>e</sub>	0.0374	5	f'.	to		
30	41.8		2	30 mm	0.6938	4_	g'	<u>•</u> K		
10	20.8		2 5	∆Hm cal/g			h'			$\vdash$
Pressure	1		-	ΔHv cal/g			m n	300 to	0.0294	
mm 25°C	12.5	74	4	25°C	76.4 73.89	2 5		<u></u>	-0.0652	4
t <sub>e</sub>	1114.4		5	30 mm BP	63.9	2		1 -00		$\vdash$
Density				t <sub>e</sub>	62.23	5	m'	700 to	0.1222	4 4
g/ml 20°C		3895 3524	2 2	t <sub>e</sub> (d, e)	62.23	5	ان		-0.0645	
d <sub>4</sub> 25 30		3153	4	AHv/T <sub>e</sub>	18.93	5			-	$\vdash$
a		5378	4	d 45 to	78.48	4		face tension es/cm. 20°C	21.98	2
b	-0.0		4	150 °C	0.1096 80.12	5	₹,,,,	30	21.11	2
Ref. Index				e' 45 °C	0.1488	5		40	20.25	2
<sup>n</sup> D 20°C	. 1	1472	2	d <sub>c</sub> g/ml			Par	achor [P]		
30		1246 1026	2 4	v <sub>c</sub> ml/g				20 <b>°C</b> 30		
"C"	0.7		4	tc°°C	302.3	5	ĺ	40		
MR (Obs.)			2	P <sub>c</sub> mm	15991.	5			385.2	5
MR (Calc.			5	PV/RT	, ,,,,,,	١	Exp	. L.1.%/wt.		
(nD-d/2)	1.0	4524	2	25°C 30 mm	1.0000	5	Dis	u. persion	96.8	2
Dielectric				BP	0.9613	4	I	sh Point °C	70.6	
A 40 to		3173	2	t <sub>e</sub>	0.9503	5		e Point		
B 1160 °C	1374.0 214.7		2 2	t <sub>c</sub>			М.	Spec.		
<del></del>	<del></del>			ΔHc kcal/m ΔHf			Ultz	aV.		
A*  40 to		9082	5 5	ΔFf				lay Dif. ared		
к — — -	-			Viscosity			l <del>}</del>			$\vdash$
t,to	-			centistokes n °C				ibility in <sup>T</sup> etone		
t <sub>k</sub>   to				ŋ °C		1	Ca	rbon tet.		
A'  25 to	6.5	6205	5					nz <b>e</b> ne her		
B' _ 45 °C	1239.1		5	· · ·	<del> </del>			Heptane		
C'	201.8		5	B <sup>V</sup>	i			hanol	1	
A'* 25 to B'* 45 °C		5786	5	Hv				iter iter in		
Ac 160 to			-	1						
Bc tc °C		8064	5	(A <sup>V</sup> )  °C			1			
Cc	261.7		5	c <sub>p</sub> liq. °K						
Cryos. A		027	2	c <sub>p</sub> vap.300°K	0.39649	2				
consts. B°				<del>- 4</del> 00	0.50136	2				
t <sub>e</sub> °C	148.2	9	5	c <sub>v</sub> vap.			L			
$T_{R} = 0.7$	6 Т <sub>с</sub>						† gr	ams/100 gra	ms solven	t
REFEREN	CES: 1-1	Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:			A	.PI						
PURIFICA	TION:		A	.PI						
LITERATU	RE REF	ERE	NCES	5:						
										,

						No. 74
NAME	2, 2, 4, 4-7	etra	methylpentane			STRUCTURAL FORMULA
						CH <sub>3</sub> CH <sub>3</sub>
			T		$\dashv$	сн <sub>3</sub> с - сн <sub>2</sub> с - сн <sub>3</sub>
Mole		lecul		Molecular Weight 128.	250	ch <sub>3</sub> ch <sub>3</sub>
% Pur.	FC	rmul		Weight 128.		
	-66.54	Ref.		1	Ref.	
F.P. °C F.P. 100%		+	dt/dP *C/mm	ļ		f to to
B. P. *C	<b>†</b>	<del>                                     </del>	25°C	0.968	4	h
760 mm	122. 284	2	BP t <sub>e</sub>	0.04932 0.0378	2 5	f' to
100 30	60.154 33.00	2 2	30 mm	0.6778	4	g'   ' <u>*</u> K_
10	12,50	2	ΔHm cal/g	0.01.0	÷	h'
1	-24.7	5	ΔHv cal/g	<del> </del>	$\vdash$	m   300 to 0.0294 4
Pressure mm 25°C	20.0524	5	25°C	71.0	2	n   _600 °K   0.0015 4 -0.0252 4
t <sub>e</sub>	1073.	5	30 mm BP	71.44 61.2	5 2	
Density				59.69	5	m'   700 to   0.1222 4 n' ,  1000 °K   0.0013 4
g/ml 20°C	0.71947 0.71563	2 2	te (d, e)	59.66	5	n'   1000°K   0.0013 4 0'   -0.0 <sub>6</sub> 45 4
d <sub>4</sub> 25	0.71178	4	ΔHv/T <sub>e</sub>	18.72	5	<del> </del>
a	0.73482	4	d   35 to		4	Surface tension dynes/cm. 20°C 20.37 2
ь	-0.03762	4	a 135 °C		*	30   19.46   2 40   18.55   2
Ref. Index		2	e' j °C			<del> </del>
n <sub>D</sub> 20°C	1.40459	2	d <sub>c</sub> g/ml			Parachor [P] 20°C
30			d g/ml v ml/g t °C	283.0	5	30
"C"	0.7535	4	Pcmm	14836.	5	40 Sugd. 385.2 5
MR (Obs.) MR (Calc.		2 5	PV/RT	<del> </del>		Exp. L. 1. %/wt.
(nD-d/2)	1.04720	2	25°C 30 mm	1.0000	5	u.
Dielectric			BP	1.0000 0.9538	4	Dispersion 101.0 2 Flash Point °C
A 30 to		2	t <sub>e</sub>	0.9431	5	Fire Point
B (_155 °C	216, 093	2	t <sub>c</sub>	<del> </del>	$\vdash$	M Spec.
A*  30 to	+	5	ΔHf			Ultra V. X-Ray Dif.
B* 145 °C		5	ΔFf			Infrared
K		1	Viscosity centistokes			Solubility in +
t <sub>te (</sub> to			7 °C		1	Acetone Carbon tet.
'x '	1		<b>,</b>	j		Benzene
A'   25 to B' <u>35</u> °C		5				Ether n-Heptane
c, C == =	313.4	5	B <sup>V</sup> to			Ethanol
A!# 25 to		5	AV I _ °C	_		Water
B'* 35 °C	+	5	(B <sup>V</sup> ) to	1		Water in
Ac 155 to	7.2058 1635.6	5	(A <sup>V</sup> )  °C			
Cc c-	256.0	5	c <sub>p</sub> liq. °K			
Cryos. A° consts. B°	0.0273	2	c <sub>p</sub> vap.300°K 400	0.39649 0.50136	2 2	
t <sub>e</sub> °C	135.74	5	c <sub>w</sub> vap.			
$T_R = 0.7$	5 T <sub>C</sub>					+ grams/100 grams solvent
REFERENC	CES: 1-Dow	2-AI	PI 3-Lit. 4-(	Calc, from det	da	
SOURCE:		AI	PI			
PURIFICAT	TION:	AI	PI			
LITERATU	RE REFERE	NCES	:			

									No. 75	
NAME	2, 3,	3,4-7	Γetra	methylpentane		İ	ST	RUCTURAL	FORMUL	A
1 [								СH <sub>3</sub>		
ļL	—							сн <sub>3</sub> сн с - (	-u -cu	
Mole	Ref.	Mo	ecul	ar с <sub>9</sub> н <sub>20</sub>	Molecular	- 1				
% Pur.		For	rmula	C9H <sub>20</sub>	Weight 128.2	50		ĊH <sub>3</sub> ĊH <sub>3</sub>	сн <sub>3</sub>	
			Ref.	1		Ref.				Ref.
F. P. °C	-102.12	23	2	dt/dP			f	to		
F.P. 100%				°C/mm		1	g	°K		
B. P. °C				25°C	1.997	5	h			
760 mm	141.5		2	BP +	0.04223 0.0376	5	f¹	to		-
100 30	76.98		2 2	t <sub>e</sub>			g'	! <b>°K</b>		
10	48.64		2	30 mm	0.7083	4	h'			
1	-8.4		5	ΔHm cal/g		L	m	300 to	0.0294	4
Pressure				ΔHv cal/g	1		n	600 °K	0.0274	4
mm 25°C	8.8	67	5	25°C 30 mm	77.8 75.53	5	0		-0.0652	4
t <sub>e</sub>	1131.		5	BP	65.1	2	m'	700 4-		4
Density			ا . ا	te (d.s)	63.31	5	n'	700 to	0.1222 0.0013	4
g/ml 20°C		5473 5113	2 2	e (4, 6)	63.29	5	ە'		-0.0645	4
d <sub>4</sub> 25 30		1753	4	ΔHv/T <sub>e</sub>	18.84	5				
a	0.70		4	d 50 to	81.00	4		face tension es/cm. 20°C	23.31	2
ь	-0.0		4	- 160 °C	0.1123 80.20	5	8,,,	30	22.44	2
Ref. Index				e' 50 °C	0.0959	5		40	21.58	2
n <sub>D</sub> 20°C		2222	2	d <sub>c</sub> g/ml			Par	achor [P]		
25 30		2003 1798	2 4	v <sub>c</sub> ml/g		ł		20°C		
"C"	<del></del>		4	vc ml/g tc °C	317.1	5		30 <b>4</b> 0		
	0.74			P <sub>c</sub> mm	16742.	5			385.2	5
MR (Obs.) MR (Calc.			5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04		2	25°C	1.0000	5		u.		
Dielectric	+		$\vdash$	30 mm BP	1.0000 0.9554	5 4		persion	95.5	2
A 45 to	6.85	:061	2	t_	0.9434	5		sh Point C		
B 1175 °C			2	tc				e Point		-
c	214.70	)5	2	∆Hc kcal/m				Spec. raV.		
A* 45 to			5	ΔHf ΔFf		1		lay Dif.		
B*[170 °C	1326. 25	•	5			-	Infr	ared		
c			l i	Viscosity centistokes		ł		ability in +		
t <sub>k</sub> _ to				η °c				etone rbon tet.		
t <sub>x</sub> °C				·				nzene		
A'   25 to			5					her		
B' 1_50 °C	1507.44 222.9	ŀ	5	B <sub>v</sub> to				Heptane hanol		
A'* 25 to	<del></del>	1424	5	AV   °C		1		ter		
B'* 50 °C			5	(B <sup>V</sup> )  - to	1		Wa	ter in		
Ac 175 to			5	(A <sup>V</sup> )  °C						
Bc t <sub>c</sub> °C	1751.7		5		<del>                                     </del>	t				
Cc	257.6		5	P	1					
Cryos. A°	0.03	369	2	c vap.300°K	0.39649	2				
consts. B°	<b>_</b>			1 - 400	0.50136	2				
t <sub>e</sub> °C	157.71		5	c <sub>v</sub> vap.	L	L	L			L
$T_{\mathbf{R}} = 0.7$	6 Т <sub>с</sub>						† gı	ams/100 gra	ms solven	t
REFEREN	CES: 1-I	Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			AP	I					-	
PURIFICA	TION:		AP	ľ						
LITERATU		ERF								
	Kuf			••						
L										

								No. 76	
NAME	n-Decan	e				STF	UCTURAL 1	FORMULA	
							CH <sub>3</sub> (CH <sub>2</sub> )	-CH-	
Mole	Ref. M	olecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular			0113(01127)	803	
% Pur.	F	ormul	a C <sub>10</sub> H <sub>22</sub>	Weight 142.2	276				
	<del></del>	Ref.		·	Ref		<del></del>		Ref.
F.P. *C F.P. 1007	-29.661	2	dt/dP *C/mm			f	to •K		
B. P. *C		+	25°C	10.537	5	g h	1		
760 mm 100	174.123 108.582	2 2	BP t <sub>e</sub>	0.05172 0.0364	2 5	f'	+ to		
30	79.65	4	30 mm	0.7246	4	g'			
10 1	57.56 20.59	5	ΔHm cal/g	48.24	3	h'	<u> </u>		
Pressure	1	+	ΔHv cal/g	1		m n	300 to	0.0256 0.0014	
mm 25°C		5	25°C 30 mm	86.28 79.99	2 5	0	-030 IK	-0.0647	
t <sub>e</sub> Density	1190.8	5	BP	66.0	2	m'	700 to	0.1099	4
g/ml 20°0			te te (d, e)	63.55	5 5	n'	17000 .K	0.0012	4
dt 25 4 30	0.72625 0.72245		ΔHv/T <sub>e</sub>	19.41	5	٥'	1	-0.0640	4
1 30	0.74525		d   80 to	91.79	4		ace tension ss/cm. 20°C	23.92	2
Ъ	-0.03759				5	3,,,,	30	22.96	2
Ref. Index	_ 1	,   ,	e' 80 °C		5	<u> </u>	40	22.02	2
D 25	1.41189		d g/ml v ml/g	0.236	2	Par	achor [P] 20°C		
30	1.40734		t <sub>c</sub> °C	4.231 346.	2 2		30 40		
"C"	0.7510	4	P <sub>c</sub> mm	15808.	2			424.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04686	5 2	25°C 30 mm	1.0000	5 5	Dist	u. ersion	98.0	2
Dielectric		_	BP	0.9345	5		sh Point °C		<u> </u>
A   75 t	6.95367 2   1501.268	2 2	t <sub>e</sub> t <sub>c</sub>	0.9187 0.246	2	Fire	Point		
c	194, 480	2	∆Hc kcal/m	1516.63	2	M S			
A* 75 to B* 200 °C		4	ΔHf ΔFf	[		X-R	ay Dif.	1	
K 1200	1422.53	*	Viscosity	1		<b></b>	hility in +	ļ	L
t.	<del>-</del>		centistokes 7 110 °C	0.5164	2		bility in + etone		
t <sub>k</sub> to			130	0.4502	2		rbon tet. nzene		
A'   25 to			150 170	0.3976 0.3543	2 2	Eth	er		
B' L 80 °C	2 1719.86   213.8	5	B <sup>V</sup>   70 to	462.90	4		deptane nanol	}	
A'* 25 to			A 1 125 °C	7. 50523	4	Wa	ter	!	
B'* 80 °C	1	4	(B <sup>V</sup> )  125 to	464.77	4	- Wa	ter in		_
Ac   210 to		5	(A <sup>V</sup> )  180 °C	₹.50071	4		osity tistok <b>e</b> s		
	232.0	5	c <sub>p</sub> liq. •K			η	80 <b>°</b> C	0.6544	
Cryos. Acconsts. B		3	c <sub>p</sub> vap.300°K	0.39599 0.50072	2 2		120	0.4814	2
t <sub>e</sub> °C	192.66	5	c <sub>w</sub> vap.			}			
$T_R = 0.7$	8 T <sub>c</sub>					+ gra	ams/100 gran	ns solvent	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from det	dat				
SOURCE:		AI							
PURIFICA		AI		222 (1054) =:					
LITERATU	RE REFER	ENCES	s: 3 JACS <u>76,</u>	333 (1954) Fi	nke e	tal			

							No. 77	
NAME	2-Methyli	nonar	ne			STRUCTURAL	FORMUL	A
Mole	Post Ma			4-11		CH <sub>3</sub> CH (CH	<sub>2</sub> ) <sub>6</sub> CH <sub>3</sub>	
% Pur.		lecul rmula		Molecular Veight 142.2	276	Ċн <sub>3</sub>		
		Ref.	1	1	Ref.			Ref.
F. P. °C	-74.50	2	dt/dP			f to		
F.P. 100%	· <del> </del>	<u> </u>	°C/mm 25°C	6,8858	5	g '° <u>K</u>		
B. P. °C 760 mm	166.8	2	<b>B</b> P	0.05176	5	h		
100 <b>3</b> 0	101.27 72.38	4	t <sub>e</sub>	0.0357	5	f' to		ĺ
10	50.35	5	30 mm	0.7231	5_	h'		١
1	13.55	5	ΔHm cal/g ΔHv cal/g		łН	m to		
Pressure mm 25°C	2.206	5	25°C	81.75	5	n   <u>*K</u>		
t <sub>e</sub>	122.39	5	30 mm BP	76.89 66.18	5			L_
Density	0.7301		t <sub>e</sub> t <sub>e</sub> (d, e)	63.96	5	m' to		١
g/ml 20°C	0.7281 0.7242	2 2		63.94	5	0'		
d <sub>4</sub> 25 30	0.7203	4	ΔHv/T <sub>e</sub>	19.79	5	Surface tension		
a b	0.7437 -0.0 <sub>3</sub> 78	4	d 10 to	85.10 0.1134	5	dynes/cm. 20°C	22.21 21.27	5
Ref. Index	<del></del>	Ė	d' 20 to e' 70 °C	84.32 0.1026	5	30 40	20.36	5
n <sub>D</sub> 20°C	1.4099	2	d <sub>c</sub> g/ml	0.1020	5	Parachor [P]		
25 30	1.4076 1.4053	2 4	v <sub>c</sub> mi/g	4.076	5	20°C 30		
"C"	0.7496	4		336.	5	40	424.2	_
MR (Obs.)		2	P <sub>c</sub> mm	15364.	-		424.2	5
MR (Calc. (nD-d/2)	1.0459	5 2	25°C	1.0000	5	Exp. L.1.%/wt.		
Dielectric	1.988	5	30 mm BP	1.0000 0.9693	5	Dispersion	99.	2
A 70 to	6. 93010	5	t	0.9568	5	Flash Point C Fire Point		
B   225 °C	_ 1485.28 200.	5	tc ΔHc kcal/m	0.242	5	M. Spec.		<del>                                     </del>
A* 70 to	1.3949	5	∆Hf	!		Ultra V.		
B* 190 °C		5	ΔFf			X-Ray Dif. Infrared		
K c			Viscosity centistokes			Solubility in +		
t <sub>k</sub> Toc			η °C			Acetone Carbon tet.		
t <sub>x</sub> °C	7. 2905	5				Benzene		
Bti 70 ℃	1688.1	5	B <sub>v</sub> to			Ether n-Heptane		
C'	218.	5	B' to			Ethanol Water		
A'* 10 to B'* 70 °C	1.7817 1592.7	5 5	(BV) - to	İ		Water in		
Acl 225 to	7,7412	5	(A <sup>V</sup> )  °C					
Bc tc °C	2254.1 - 298.5	5	c liq. °K					
Cryos. A°	+	Ť	li .					
consts. B°			P					
t <sub>e</sub> ℃	186.55	5	c <sub>w</sub> vap.					
$T_R = 0.8$						f grams/100 gra		it
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	PION.	AF						
PURIFICAT	RE REFERE	AF						
J. I DAR I U	ve veleve	140 E	·					

							No. 78	
NAME	3-Methyln	onane	•		_	STRUCTURAL	FORMULA	١.
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	сн <sub>3</sub> сн <sub>2</sub> сн (сн сн <sub>3</sub>	<sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>	
		Ref.			Ref			Ref
F.P. °C F.P. 100%	-84.80	2	dt/dP °C/mm 25°C	7.8111	5	f to		
B. P. °C 760 mm 100 30	167.8 103.08 74.40	2 4 4	BP t <sub>e</sub> 30 mm	0.05101 0.03565 0.7190	5 5	f' to g'		
10 1	52.48 15.75	5	ΔHm cal/g			h'		<u> </u>
Pressure mm 25°C t <sub>e</sub>	1.907 1200.1	5	ΔHv cal/g 25°C 30 mm BP	83.37 78.24 66.38	5 5 5	m to		
Density g/ml 20°C dt 25 d4 30	0.7334 0.7296 0.7258	2 2 4	te te (d,e) ΔHv/Te	64.04 64.01 19.82	5 5 5	m'   to		
a b	0.7486 -0.0 <sub>3</sub> 76	4 4	d   75 to e   185 °C d'   15 to	87.68 0.1269 85.96	5 5 5	Surface tension dynes/cm. 20°C 30 40	22.86 21.93 21.02	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30	1.4125 1.4103 1.4080	2 2 4	e' 75 °C  d g/ml vc ml/g tc °C	0.1039 0.243 4.117 336.8	5 5 5	Parachor [P] 20°C 30		
"C"	0.7486	4	P <sub>c</sub> mm	16121.	5	40 Sugd.	424.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	48.32 48.38 1.0458	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5 5	Exp. L.1.%/wt. u. Dispersion	98.	2
Dielectric	1.995	5	BP t <sub>e</sub>	0.9536 0.9384	5	Flash Point °C		$\vdash$
A 75 to B <u>1226 °C</u> C	7.0020 1516. 200.	5 5 5	t <sub>c</sub> AHc kcal/m	0.234	5	Fire Point  M Spec. Ultra V.		
A* 75 to B* 200 °C K	1.4957 1427.3	5	ΔHf ΔFf Viscosity			X-Ray Dif. Infrared Solubility in +		
t <sub>k</sub> to t <sub>x</sub> C			r centistokes			Acetone Carbon tet. Benzene		
A'   15 to B'   75 °C C' A'* 15 to	7.3644 1721. 218.	5 5 5	B <sup>V</sup>   to			Ether n-Heptane Ethanol Water		
B'* 75 °C  Ac   226 to  Bc   t <sub>c</sub> °C	7.8175 2291.	5 5 5	(B <sup>V</sup> ) to (A <sup>V</sup> ) °C			Water in		
Cryos. A°	298.	5	c <sub>p</sub> liq. °K c <sub>p</sub> vap. °K					
te °C	186.5	5	c <sub>v</sub> vap.					
$T_{R} = 0.82$		لــّــا	1			+ grams/100 gran	ne solven	<u> </u>
		2-AI	PI 3-Lit. 4-C	alc. from det	. da	ta 5-Calc. by for		
SOURCE:		A.						
PURIFICAT		A)						
LITERATUR	E REFEREI	NCES	:					
_								

							No. 79	,
NAME	4-Methylr	onane				STRUCTURAL	FORMUL	A
						CH (CH ) CH (	сн / сн	
Mole	Ref. M	lecul		Molecular		сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн ( сн <sub>3</sub>	C112/4C11	3
% Pur.	Fo		1022	Weight 142.2				D - 6
F.P. °C	-98.7	Ref.	dt/dP	<del> </del>	Ref.	- T		Ref.
F.P. 100%	-70.1		°C/mm			f to g*K_		
B. P. ℃			25°C BP	7.1296	5	h		
760 mm 100	165.7 101.18	2	te	0.03566	5	f¹ to		
<b>3</b> 0 10	72.61 50.79	4 5	30 mm	0,7158	5_	g'° <u>K</u>		
_ 10	14.25	5	ΔHm cal/g		$\vdash \dashv$	h' i		┼
Pressure			ΔHv cal/g 25°C	82.72	5	m to		
mm 25°C	2.1059 1195.8	5	30 mm	77.78	5	•		
Density		+	BP t	65.98 63.70	5	m' to		T
g/ml 20°C	0.7323	2	t <sub>e</sub> (d, e)	63.64	5	n'   <u>•K</u> _		
d 25 4 30	0.7284 0.7245	2 4	ΔHv/T <sub>e</sub>	19.82	5			<del> </del>
a	0.7479	4	d 70 to	86.99	5	Surface tension dynes/cm. 20°C	22,72	5
ь	-0.0378	4	_e180 °C 15 to	0.1268 85.31	5 5	<b>8</b> 30	21.77	5
Ref. Index	1,4123	2	e' 70 °C	0.1037	5	40	20.84	5
<sup>n</sup> D 25	1.4100	2	d <sub>c</sub> g/ml	0.242	5	Parachor [P]		
30	1.4076	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.127 333.	5 5	30 40		
"C"	0.7494	4	P <sub>c</sub> mm	15903.	5	1	424.2	5
MR (Obs.) MR (Calc.)	48.38 48.38	5	PV/RT			Exp. L. l. %/wt.		
(nD-d/2)	1.0461	2	25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric	1.994	5	BP	0.9542	5	Flash Point C	70.	+ <del>-</del>
A 70 to B 222°C	6.9904	5	t e t c	0.9399 0.247	5	Fire Point		
c '	200.	5	∆Hc kcal/m		H	M. Spec. Ultra V.		
A* 70 to	1.4844	5	ΔHf ΔFf	Ì		X-Ray Dif.		
B*[_1 <u>90</u> °C   K	1414.3	5	Viscosity			Infrared		↓
°			centistokes			Solubility in + Acetone		1
t <sub>k</sub> to t <sub>x</sub> °C			η °c			Carbon tet.		
A'  15 to	7.3544	5				Benzene Ether		
B' 1_ 70°C	1708.0 218.	5	B <sup>V</sup> to		+1	n-Heptane Ethanol		1
A'* 15 to	1,8560	5	B <sup>V</sup> to A °C			Water		
B'* 70°C	1615.6	5	(B <sup>V</sup> )  to			Water in		ـــ
Acl 222 to	7.7856	5	(A <sup>V</sup> )  °C					
Bc t <sub>c</sub> °C	2249.3	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°		$\top$	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	184.13	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.82$				1		grams/100 gra	ms solver	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		AI						
PURIFICAT		AI						
LITERATUE	RE REFERE	NCES	S:					

							No. 80		
NAME	5-Methyln	onane	•			STRUCTURAL I	FORMULA	1	
Ī									
			1		$\neg$	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>3</sub> сн (с	H <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>		
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	ĊH <sub>3</sub>			
		Ref.			Ref			Ref.	
F, P. °C	-87,70	2	dt/dP			f   to			
F.P. 100%			*C/mm	4 0051	5	gK_			
B. P. °C 760 mm	165.1	2	25°C BP	6.8851 0.0508	5	h		_	
100	100.6	4	t <sub>e</sub>	0.03560	5	f' to			
30 10	72.05 50. <b>2</b> 6	5	30 mm	0.7148	5	h' '		ĺ	
1	13.77	5	ΔHm cal/g	<del> </del>	_	m   to		┢	
Pressure mm 25°C	2 1960	_	ΔHv cal/g 25°C	82.52	5	n•K_			
t <sub>e</sub>	2.1860 1197.0	5 5	30 mm BP	77.64 65.87	5	<u>°</u>		<u></u>	
Density			te (d. e)	63.73	5	m' to	•		
g/ml 20°C	0.7326 0.7288	2 2	l e (d, e)	63.55	5	o'   '			
dt 25 4 30	0.7250	4	ΔHv/T <sub>e</sub>	19.61	5	Surface tension		$\vdash$	
<b>a</b> .	0.7478	4	d   70 to		5	dynes/cm. 20°C	22.76	5	
b Ref. Index	-0.0376	4	d'   25 to	85.11	5	30 40	21.83 20.93	5 5	
n <sub>D</sub> 20°C	1.4122	2		0.1037	5	Parachor [P]			
25 30	1.4100 1.4077	2	d g/ml vc ml/g	4.098	5	20°C 30			
"C"	0.7489	4	te o	333.	5	40	424.2	_	
MR (Obs.)	48.34	2	P <sub>c</sub> mm	16015.	5	Sugd.	424.2	5	
MR (Calc. (nD-d/2)	) 48.38 1.0459	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.			
Dielectric	1.994	5	30 mm BP	1.0000 0.9542	5	Dispersion	98.	2	
A 55 to	<del></del>	5	t_	0.9421	5	Flash Point °C Fire Point		l	
B [222 •C	1499. 200.	5	tc	0.247	5	M Spec.		$\vdash$	
A*   55 to	<del></del>	5	ΔHc kcal/m ΔHf		1	Ultra V.		İ	
B* 190 °C	1409.5	5	ΔFf		<u> </u>	X-Ray Dif. Infrared			
K		ļ .	Viscosity centistokes			Solubility in +			
t <sub>k</sub> to		[	η •c	:		Acetone Carbon tet.		ĺ	
A' 10 to		5			ł	Benzene			
B' _ 55 °C		5	<b></b>	ļ	<u> </u>	Ether n-Heptane			
C'	218.	5	B <sup>V</sup> to			Ethanol Water			
A'* 10 to B'* 55 °C		5 5	(B <sup>V</sup> ) to	-		Water in		<u> </u>	
Ac   222 to	7.7813	5	(A <sup>V</sup> )	1					
Bc Ltc_°C	2244. 294.	5	c <sub>p</sub> liq. °K			1			
Cryos. A	† <del>- / - ·</del>	۲	c <sub>p</sub> vap. °K	: 1					
consts. B°			р.					}	
t <sub>e</sub> °C	183.47	5	c <sub>v</sub> vap.	L					
$T_{R} = 0.88$						<sup>+</sup> grams/100 gran		<u>t</u>	
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by form	nula		
SOURCE:	PION.		PI PI						
PURIFICATI	RE REFERE								
LILERATU	AL REPEKE	NUES	<b>:</b>						

No. 81 3-Ethyloctane NAME STRUCTURAL FORMULA CH3CH2CH (CH2)4CH3 Ċ<sub>2</sub>н<sub>5</sub> Mole Ref. Molecular Molecular  $C_{10}H_{22}$ Weight 142, 276 % Pur Formula Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm °<u>K</u> g 25°C 7.8654 5 B.P. °C h BP 0.05101 5 760 mm 168. 2 t<sub>e</sub> 0.0356 5 f١ to 100 103.2 4 g¹ <u>°к</u> 30 74.52 4 30 mm 5 0.7192 10 52.60 5 h\* AHm cal/g 15.86 5 m to ∆Hv cal/g Pressure ۰ĸ n 25°C 83.42 5 mm 25°C 1.8929 o 30 mm 78.27 5 te 1202.5 5 ΒP 5 66.42 m to Density t<sub>e</sub> (d, e) 5 64.16 n' °K g/ml 20°C 0.740 2 64.05 5 ٥' 25  $\mathbf{d_4^t}$ 0.736 2 AHV/Te 19.85 5 30 0.732 4 Surface tension 75 to 87.72 5 0.756 dynes/cm. 20°C 23.70 5 ٠C 0.1268 5 Ъ -0.0380 4 22.69 5 30 ٦·٦ 10 86.02 5 to 21.71 5 40 Ref. Index 75 5 e' 0.1039 <sup>n</sup>D 20°C 1.416 [P] Parachor d<sub>c</sub> g/ml 0.245 5 25 1.414 2 20°C vc ml/g 4.085 30 1.411 4 30 337. 5 tc 40 "C" 0.7479 4 16170. P<sub>c</sub> mm 5 Sugd. 424.2 5 MR (Obs.) 48.3 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2)1.046 2 30 mm 1.0000 Dispersion 96. 2 Dielectric 2,005 5 BP 0.9535 5 Flash Point C 0.9398 A 75 to te tc 7.0031 5 Fire Point 0.247 5 1517. B | 227 °C 5 M. Spec. C 200. 5 AHc kcal/m Ultra V. AHf A\* 75 to 1.4945 5 X-Ray Dif. ΔFf B\*|\_200 °C 1427.6 Infrared K Viscosity Solubility in centistokes c Acetone  $\mathbf{t_{k}^{t}}$ to Carbon tet. °C Benzene A' 10 to 7.3654 Ether B' 7<u>5 °C</u> 1722.5 n-Heptane B<sub>v</sub> | 218. 5 to Ethanol °C Water A'\* 10 to 1.8550 5 Water in B'\* 75 °C (BV) 1626.7 5 to Acl 227 to 7.8210 (A<sup>V</sup>) 5 °C Bc tc °C 2294.9 5 cp liq. °K Cc 298.1 5 Cryos. A° ۰ĸ cp vap. consts. Bo c<sub>v</sub> vap. te °C 186,69  $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc, by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

							No. 82	
NAME	4-Ethylo	ctane				STRUCTURAL 1	FORMUL	A
Г								_
L			T			CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH		<sup>1</sup> 3
Mole % Pur.		olecul ormul		Molecular Weight 142.2	76	с <sub>2</sub> н <sub>5</sub>		
76 Pur.		Ref	10 22	Weight 142.2	Ref			Ref.
	T	Kei.		1	Ker			1
F.P. °C F.P. 100%	<del>                                     </del>	+	dt/dP *C/mm	1		f to		
B. P. °C		+	25°C	7.8654	5	g		i
760 mm	168.	2	BP	0.05101 0.03560	5	f' to		+
100 30	103.2 74.52	4	t <sub>e</sub> 30 mm	0.7192	5	g'   '°K		1
10	52.60	5		0.1172	-	h'		
1	15.86	5	ΔHm cal/g		-	m   to		<del>                                     </del>
Pressure		_	ΔHv cal/g 25°C	83, 42	5	n •K_		
mm 25°C	1.8929	5	30 mm	78.27	5	°		
Density	1	+	BP	66.42 64.16	5	m¹   to		T
g/ml 20°C	0.740	2	te (d, e)	64.05	5	n' •K_		
d <sub>4</sub> 25	0.736 0.732	2 4	AHV/T	19.85	5	o'		
	<b></b>		d   75 to	87.72	5	Surface tension		١.
a b	0.756 -0.0 <sub>3</sub> 80	4	_e <u>  1</u> 85 °C	0.1268	5	dynes/cm. 20°C	23.70 22.69	5
Ref. Index			d'   10 to		5	40	21.71	5
n <sub>D</sub> 20°C	1.416	2	d g/ml	0.245	5	Parachor [P]		T
25 30	1.414	2 4	V mi/g	4.085	5	20°C 30		İ
"C"	0.7479	4	tc °C	337.	5	40		
	<b></b>	_	P <sub>c</sub> mm	16170.	5	Sugd.	424.2	5
MR (Obs.) MR (Calc.)	48.3 48.38	2 5	PV/RT			Exp. L.1.%/wt.		
(D-d/2)	1.046	2	25°C 30 mm	1,0000	5	u. Dispersion	96.	2
Dielectric	2,005	5	BP	0.9535	5	Flash Point °C	/0.	+
A 75 to		5	t <sub>e</sub>	0.9398	5	Fire Point		1
B <u>  227 °C</u> C	1517. 200.	5	t <sub>c</sub>	0.247	٦	M Spec.		
A*  75 to	1.4945	5	ΔHc Real/m			Ultra V.		
B* 200 °C		5	ΔFf			X-Ray Dif. Infrared		
к — — —			Viscosity	İ		Solubility in +		+
,	-		centistokes 7°C			Acetone		
t <sub>x</sub> ¦ °C			'			Carbon tet. Benzene		1
A'   10 to	7.3654	5				Ether		1
B' ∟75 °C	1722.5	5	B <sup>V</sup> to	1	$\vdash$	n-Heptane Ethanol		
A'* 10 to	1.8550	5	AV C			Water		1
B'* 75 °C		5	(BV) to	1		Water in	<u> </u>	$\bot$
Ac   227 to	7.8210	5	(A <sup>V</sup> ) °C					
Bc t <sub>c</sub> °C	2294.9	5	cp liq. °K					1
CC	298.1	- 3	tl -					
Cryos. A° consts. B°			c <sub>p</sub> vap. *K					1
t <sub>e</sub> °C	186, 69	5	c <sub>v</sub> vap.					1
$T_{R} = 0.83$			<u> </u>	1		grams/100 grai	ne eclus-	<del></del>
REFERENC		2-AI	PI 3-Lit. 4-0	Calc. from de	t. de			
SOURCE:			PI	a		- J-Care, by for		
PURIFICAT	ION:		PI					
	RE REFERE							
KA101	rereki	NUES	) <b>:</b>					

No. 83 STRUCTURAL FORMULA

NAME	2, 2-Di	imethyloc	tane			STRUCTURAL FORMULA				
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 142.2	76	СН	с - (сн <sub>2</sub>	<sub>2</sub> ) <sub>5</sub> СН <sub>3</sub>		
		Ref.			Ref.				Ref.	
F.P. °C			dt/dP			f	to			
F.P. 100	%		°C/mm	i	1 1	gl	K		1 1	
B. P. ℃			25°C	3.864	5	h			1 1	
760 mm	155.	2	BP	0.05104 0.0366	5	f'	1		+-	
100	90.3	4	t <sub>e</sub>	ı	1 1	g'	to *K		1 1	
30 10	61.71 39.9	4 5	30 mm	0.7159	5	h'			1 1	
li	3.4	5	ΔHm cal/g			<u> </u>				
Pressure			∆Hv cal/g			m n	to °K		1 1	
mm 25°C	4.23	7   5	25°C	75.85	5	;	, <del>,-</del>	ļ	1 1	
t <sub>e</sub>	1161.	5	30 mm BP	72.94 62.29	5				+	
Density			t_	60.38	5	m'	to	1	1 1	
g/ml 20°			te (d, e)	60.32	5	n',	_ <u>•</u> K	1		
d <sub>4</sub> 25 30	0.72		ΔHv/T <sub>e</sub>	19.29	5				$\bot$	
	0.71		d 60 to	79.98	5		tension			
a b	0.73		_e_ _170 °C		5	dynes/o	m. 20°C	21.77	5	
			d' 10 to	77.84	5	•	30 40	20.89	5	
Ref. Inde:		32 2	e'   60 °C	0.0794	5	B		20.04	+	
<sup>n</sup> D 20°	1.40		d <sub>c</sub> g/ml	0.240	5	Parach	or [P] 20°C			
30	1.40	38 4	vc ml/g t °C	4.170 318.5	5		30	1	1 1	
"C"	0.750	04 4	-	i	1 1		40		1 _ 1	
MR (Obs.	) 48.47	2	P <sub>c</sub> mm	14927.	5			424.2	5	
MR (Calc	.) 48.38	5	PV/RT 25°C	1 0000	_	Exp. L	.1.%/wt.	1	1 1	
(nD-d/2)	1.04	60 2	30 mm	1.0000	5	Dispers	u. ion	100.	2	
Dielectric	1.98	3 5	BP	0.9500	5	Flash F			+	
A 60 to	6.96	21 5	t <sub>e</sub>	0.9369	5	Fire Po				
B 1_205 °C		5	t c	0.24	5	M. Spe			+-	
С	210.	5	ΔHc kcal/m ΔHf			Ultra V				
A* 60 to			ΔFf			X-Ray			1 1	
B*[180°C	1 390. 3	5	Viscosity		$\vdash$	Infrare	d 			
c			centistokes			Solubili				
t <sub>k</sub> T to			່າ °c			Acetor Carbo				
x	3			1	1	Benze			1 1	
A'  0 to						Ether				
B' 1_60 °	228.	5 5	B <sup>V</sup> to			n-Hep				
		-	AV   °C			Ethano Water				
A'* 0 to		246   5	$\frac{\overline{\mathbf{B}^{\mathbf{V}}}}{ \mathbf{B}^{\mathbf{V}}  } = \frac{1}{to}$	-[		Water		1		
Acl 205 to									77	
Bc tc °C	2175.	5		<del> </del>	$\vdash$					
Cc	<b>–</b> 296.	5	c <sub>p</sub> liq. °K	1						
Cryos, A consts, B			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	172,21	5	c <sub>v</sub> vap.							
$T_R = 0.81$			L	1	نـــــــــــــــــــــــــــــــــــــ	+ gram	s/100 gra	ms solve	nt	
		ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da					
SOURCE:		AP					,			
PURIFICA	TION	AP								
	JRE REF									
	JRE REFI	SKENCE:	•							
									}	

								N	o. 84
NAME	7	2,3-D	imet	thylo	ctane		- 1	STRUCTURAL FOR	MULA
				_				CH <sub>3</sub>	
1		1			<u>-</u>			сн <sub>3</sub> сн сн (сн <sub>2</sub> )	₄CH,
Mole		Ref.	Мо	lecul	ar c u	Molecular		ĊH <sub>3</sub>	4 )
% Pur.				rmul		Weight 142.2	76	3	
_				Ref.			Ref.		Ref
F. P. *C	Т				dt/dP			f to	
F.P. 1007	•				°C/mm	1	I. I	g  *K	
B. P. *C	1			t	25°C	5.5982	5	h	
760 mm		63.8		2	BP	0.0519 0.0367	5	f' to	
100 30		98.0 68.9		4 4	t <sub>e</sub> 30 mm	0,7287	5	g'   '°K_	
10		46.7		5		0.7201	-	h' I	1
1		9.6		5	ΔHm cal/g			<u> </u>	
Pressure	T				ΔHv cal/g	70.4/	ا ۔ ا	m to	
mm 25°C		2,82	8	5	25°C 30 mm	78.46 74.79	5	0	
t <sub>e</sub>	111	83.		5	BP	63.73	5	⊩ <del>.!</del>	
Density	.		_,		t <sub>e</sub>	61.68	5	m' to K	
g/ml 20°0	7	0.73		2 2	t <sub>e</sub> (d, e)	61.60	5	", ' <del>"</del>	
dt 25 4 30		0.72		4	ΔHv/T <sub>e</sub>	19.27	5		
	+	0.75		4	d   70 to		5	Surface tension dynes/cm. 20°C 23	. 39 5
ь		-0.03		4	180 °C		5		.41 5
Ref. Index	.   -				d'   25 to		5	40 21	.47 5
n <sub>D</sub> 20°0		1.41		2		0, 241	5	Parachor [P]	
25 30		1.41 1.41		2 4	d g/ml v ml/g	4, 154	5	20°C	
	+-				t <sub>c</sub> ml/g	331.8	5	30 40	
"C"		0.74	83	4	P <sub>c</sub> mm	15320.	5	Sugd. 424	. 2 5
MR (Obs.		48.28		2	PV/RT	<del> </del>	$\vdash$	Exp. L.1.%/wt.	
MR (Calc. (nD-d/2)	"	48.38 1.04	60	5 2	25°C	1.0000	5	u.	
Dielectric	+	2.00		5	30 mm BP	1.0000	5	Dispersion 97	. 2
A 70 t	-	6.97		5	t <sub>e</sub>	0.9480 0.9340	5	Flash Point °C	
B _ 220°			••	5	tc	0.24	5	Fire Point	
<u> </u>		08.		5	ΔHc kcal/m			M Spec. Ultra V.	
A* 70to		1.47	511	5	ΔHf			X-Ray Dif.	
B* 190°	2 14	32.2		5	ΔFf			Infrared	
K — — -					Viscosity centistokes			Solubility in +	
t <sub>k</sub>	5				7 0	:	1	Acetone	
t <sub>x</sub> i •	٦				l <b>'</b>			Carbon tet. Benzene	
A' 25to		7.33	426	5				Ether	
B' _ 70°		26.8 26.		5 5	B <sup>V</sup>   to		-	n-Heptane	
	$\rightarrow$			-	B' to			Ethanol Water	
A'* 25to	16	1.81	744	5		-		Water in	
	-		20	5		1			
Ac   220to	23	7.79 01	20	5	(A <sup>V</sup> )  °C				
Cc '— c—	- 3	06.		5	c <sub>p</sub> liq. °K		l i		
Cryos. A					c <sub>p</sub> vap. °K		i i		
consts. B	-			-	c <sub>v</sub> vap.				1
t <sub>e</sub> °C		82.10		5	ov .ap.	<u> </u>	L		
						<del></del>		grams/100 grams	
REFEREN	CES:	1-D	o <b>w</b>			Calc. from de	t. da	ta 5-Calc, by formul	a
SOURCE:	TION				PI				
LITERATU			ים פי		PI				
	<b></b>				•				

Г							No. 8	
NAME _	2,4-D	imethylo	ctane			STRUCTURAL	FORMU	LA
Mole % Pur.	Ref.	Molecu Formu		Molecular Weight 142.2	76	сн <sub>3</sub> сн-сн <sub>2</sub> сн сн <sub>3</sub> сн <sub>3</sub>		<sup>1</sup> 3
		Rei	:1		Ref.			Ref.
F. P. ℃			dt/dP			f to		
F.P. 100%	ļ		°C/mm 25°C	3.5488	5	g '° <u>K</u>	-	
B.P. °C 760 mm	153.	2	BP	0.05088	5	h	ļ	4—
10 <b>0</b>	88.51	4	t <sub>e</sub>	0.0367	5	f' to		
30 10	60.03	4 5	30 mm	0.7132	5_	g' i '° <u>K</u>	•	
1	1.9	5	ΔHm cal/g			m to	<u> </u>	+
Pressure	4 /5		ΔHv cal/g 25°C	75.24	5	n	_	
mm 25°C	4.65 1155.	14   5	30 mm	72.49	5	0	ŀ	
Density	<u> </u>		BP t <sub>e</sub>	61.90	5	m' to		
g/ml 20°C	0.72		t <sub>e</sub> (d, e)	59.97	5	n'  *K	4	
d <sub>4</sub> 25	0.72		ΔHv/T <sub>e</sub>	19.27	5	<u>-</u>		
a	0.74		d 60 to		5	Surface tension dynes/cm. 20°C	22.00	5
ь	-0.03	80 4	$\frac{1}{4}$ , $\frac{170}{25}$ °C to		5	<b>8</b> 30	21.04	5
Ref. Index np 20°C	1.40	93 2	e'   60 °C	0.0785	5	40	20.12	+ -
D 25	1.40		d <sub>c</sub> g/ml	0.237	5	Parachor [P] 20°C		
30	1.40	45 4	vc ml/g tc °C	4.214 314.7	5	30 40		
"C"	0.75		P <sub>c</sub> mm	14677.	5		424. 2	5
MR (Obs.) MR (Calc.)	48.45 48.38	2 5	PV/RT		1	Exp. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric	1.98	6 5	BP	0.9500	5	Flash Point °C	70.	+-
A 60 to	6.95		te t <sub>C</sub>	0.9370 0.24	5	Fire Point		ł
B (200 ℃ C	210.	5	ΔHc kcal/m		+	M. Spec.		
A*  60 to	1.46		ΔHf ΔFf		Ì	Ultra V. X-Ray Dif.		1
B* 180 °C   K	1390.2	5	Viscosity		├	Infrared		
c			centistokes		1	Solubility in * Acetone		
t <sub>k</sub> Toc			η °c			Carbon tet.		
t <sub>x</sub> °C	7, 32	217 5	-			Benzene Ether		
B'   60 °C		5	-v	·	+-	n-Heptane		
	1,80	936 5	B <sup>V</sup> to °C			Ethanol Water		
A'* 25 to B'* 60 ℃	1582.9	5	(B <sup>V</sup> )  to	-		Water in		
Acl 200 to	7.68		(A <sup>V</sup> )  °C					
Bc tc °C	2142. 294.	5	c <sub>p</sub> liq. °K			1		
Cryos, A°	† <del></del>	-   -	c <sub>p</sub> vap. °K					
consts. B°	<u> </u>		1					
t <sub>e</sub> ℃	169.95	5	c <sub>v</sub> vap.	<u> </u>			<u></u>	
$T_{R} = 0.817$						grams/100 grams		nt
REFERENC	ES: 1-D			-Calc. from de	t. da	ata 5-Calc. by fo	rmula	
SOURCE:			API					
PURIFICAT			API					
LITERATUI	KE REF	ERENCE	5:					

						p	No. 86	<u>,                                      </u>
NAME	2,5-Dimet	hylo	tane			STRUCTURAL	FORMUL	A
						כם כם וכם / כם	ICH \ C	u
Mole % Pur.		lecul rmul	Lucha I	Molecular Weight 142.2	76	ch <sub>3</sub> ch (ch <sub>2</sub> ) <sub>2</sub> ch	(CH <sub>2</sub> / <sub>2</sub> C	**3
		Ref.	·		Ref	1		Ref
F,P. °C	I		dt/dP			f   to		T
F.P. 100%			°C/mm		۱	g°K_		1
B. P. °C	150		25°C BP	4.3755 0.05134	5	h		1
760 mm 100	158. 92.9	2 4	te	0.0365	5	f <sup>1</sup> to		T
30	64.15	4	30 mm	0.7204	5	g'		
1 <b>0</b> 1	42.2 5.5	5	ΔHm cal/g			h'		
Pressure	<u> </u>	$\vdash$	ΔHv cal/g			m to		
mm 25°C	3.7001	5	25°C 30 mm	76.71 73.55	5			
t <sub>e</sub>	1170.	5	BP	62.81	5	m'   to		┼─
Density g/ml 20°C	0.736	2	te (d.e)	60.85	5	m' to		
t 25 d <sub>4</sub> 30	0.732	2	e (-, -,	60.79	1	o'	İ	
<sup>4</sup> 30	0.728	4	ΔHv/T <sub>e</sub>	19.29	5	Surface tension	· .	+
a L	0.752	4	d   65 to e   175 °C		5	dynes/cm. 20°C	23.19	5
b	-0.038	4	d'   25 to	78. <b>73</b>	5	30 40	22.19 21.23	5
Ref. Index	1.414	2	e'   65 °C		5	Parachor [P]		+-
25	1.412	2	d g/ml v ml/g	0.240 4.160	5 5	20°C		
30	1.409	4	tc C	323.2	5	30 40		1
"C"	0.7486	4	P <sub>c</sub> mm	15080.	5		424.2	5
MR (Obs.) MR (Calc.)	48.3 48.38	2 5	PV/RT		$\vdash$	Exp. L.1.%/wt.		T
(nD-d/2)	1.046	2	25°C	1.0000	5	u.		
Dielectric	2.000	5	30 mm BP	1.0000 0.9500	5	Dispersion	99.	2
A   65 to	6.9658	5	te	0.9366	5	Flash Point °C Fire Point		
B   210 °C	1499. 209.	5	t <sub>c</sub>	0, 24	5	M Spec.		<del> </del>
A*  65 to	1,46620	5	ΔHc kcal/m ΔHf			Ultra V.		1
B* 185 °C	1409.0	5	ΔFf			X-Ray Dif. Infrared	ĺ	1
K			Viscosity		ļ	Solubility in +	<del></del>	+
t <sub>k</sub>			centistokes 7°C		İ	Acetone	ĺ	1
tÇ i •C			'		ł	Carbon tet. Bensene		1
A'   25 to	7, 32751	5				Ether	1	
B' ∟ 65 °C	1703.2 227.	5	B <sup>V</sup> to		$\vdash$	n-Heptane, Ethanol	1	
A'* 25 to	1.81284	5	AV I °C		}	Water	1	
B'* 65 °C		5	(BV) to		l	Water in		
Ac   210 to	7.7373	5	(A <sup>V</sup> ) °C		1		ĺ	
Bc tc_°C	2217. 299.7	5	c <sub>p</sub> liq. °K			1	ĺ	
Cryos. A°		<del>                                     </del>	1 -				1	
consts. B°			c <sub>p</sub> vap. °K			1		
t <sub>e</sub> °C	175.63	5	c <sub>v</sub> vap.	į		1		
$T_{\mathbf{R}} = 0.81$	T <sub>C</sub>					grams/100 gran	ns solver	ıt.
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc, from de	t. da			
SOURCE:		A	PI					
PURIFICATI	ION:	A	PI					
	E REFERE							
		_						

No. 87 2, 6-Dimethyloctane STRUCTURAL FORMULA NAME CH3CH (CH2)3CH CH2CH3 ĊH3 ĊH, Mole Molecular Molecular Ref.  $C_{10}H_{22}$ % Pur Formula Weight 142.276 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 4.4828 B. P. °C h ВP 0.05137 5 760 mm 158.54 2 5 0.0366 f t<sub>e</sub> to 100 93.4 4 g' <u>°К</u> 30 4 64.62 30 mm 0.7210 5 10 42.6 5 h' ∆Hm cal/g 1 5.9 5 m to ∆Hv cal/g Pressure n °K 25°C 76.90 mm 25°C 3.6029 5 o 30 mm 73.69 5 1171. 5 t<sub>e</sub> ΒP 62.92 5 m' to Density 60.95 5 te (d, e) n' ۰ĸ g/ml 20°C 0.7285 2 5 60.89 o' 25 0.7245 2  $d_4^t$ AHv/Te 19.30 5 0.7205 30 4 Surface tension 70 to 81.09 d 5 0.7449 4 dynes/cm. 20°C 22,26 175 **℃** 0.1146 ь -0.0<sub>3</sub>80 4 30 21.29 5 dı آ 25 to 78.92 40 20,36 5 Ref. Index e¹ 70 °C 0.0810 5 <sup>n</sup>D 20°C 1.4113 2 [P] Parachor d<sub>c</sub> g/ml 0.237 5 2 25 1.4089 20°C vc ml/g t °C 5 4.217 30 1,4065 4 30 322.7 t<sub>c</sub> 40 "C" 0.7516 4 14866. 5  $P_c$  mm Sugd. 424. 2 5 MR (Obs.) 48.52 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 25°C 1.0000 (nD-d/2) 1.0471 2 30 mm 1.0000 5 Dispersion 99. 2 Dielectric 1.992 5 BP 0.9500 5 Flash Point C 0.9366 70 to 6.9681 5 ţe. Fire Point 0.24 5 B 210 °C 1502. 5 M. Spec. С 209. 5 AHc kcal/m Ultra V. ΔHf A\* 70 to 1.46804 X-Ray Dif. ΔFf B\*[185 °C 1411.6 Infrared ĸ Viscosity Solubility in centistokes c Acetone to tk tx °C ÷ Carbon tet. Benzene A' 25 to 7. 3295**3** 17**0**5 9 Ether B 7<u>0 °C</u> 5 n-Heptane B<sup>V</sup> | C' 227. 5 to Ethanol °C Water 25 to 1.81464 5 Water in B'\* 70 °C  $(B^{V})$ 1605.5 to Acl 210 to 7.7359 5 (AV) °C Bc tc C 2214. c<sub>p</sub> liq. °K Cc 299. 5 Cryos. Aº °K c<sub>p</sub> vap. consts. B° c<sub>v</sub> vap. te °C 176,23 5  $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

							No. 88
NAME	2,7-Din	nethylo	octane			STRUCTURAL FO	
Mole % Pur.	Ref. M	lolecul Formul	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	276	CH <sub>3</sub> CH-(CH <sub>2</sub> ) <sub>4</sub> C CH <sub>3</sub>	н сн <sub>3</sub> н <sub>3</sub>
		Ref.			Ref.		Re
F. P. *C	-54.	2	dt/dP			f to	
F.P. 100%	)		*C/mm	4 7300	5	g <u>*K</u>	
B. P. *C 760 mm	150.07		25°C BP	4.7380 0.05150		h	
100 mm	159.87 94.6	2	t <sub>e</sub>	0.0367	5	f' to	
30	65.7	4	30 mm	0.7230	5	g'K	
10 1	43.7 6.8	5	∆Hm cal/g			h'	
Pressure	+	+-	ΔHv cal/g			m to	
mm 25°C	3.3920		25°C 30 mm	77.28 73.96	5		
t <sub>e</sub>	1173.	5	BP	63.08	5	m'   to	<del></del>
Density g/ml 20°C	0.7242	2	te (d. e)	61.09 61.02	5	n' 'K	
dt 25	0.7202	2	e (-, -,	1	5	0'	
<sup>4</sup> 4 30	0.7162	4	ΔHv/T <sub>e</sub>	19.27	-	Surface tension	
<b>a</b>	0.7402	4	d   70 to		5	dynes/cm. 20°C	21.73 5
b Def Zerlee	-0.0380	4	a' _ 25 to	79.32	5		20.79   5 19.87   5
Ref. Index		2	e' j 70 °C	<del></del>	5	Parachor [P]	
25	1.4062	2	d g/ml vc ml/g	0, 235 4, 261	5	20°C	
30	1.4038	4	tc °C	323.6	5	30 40	
"C"	0.7514	4	P <sub>c</sub> mm	14735.	5	Sugd. 4	24.2 5
MR (Obs.) MR (Calc.		2 5	PV/RT	<u> </u>		Exp. L.1.%/wt.	
(nD-d/2)	1.0465	ž	25°C	1.0000	5	u.	98.6 2
Dielectric	1.984	5	30 mm BP	0.9490	5		70.0 2
A 70 to	6.9697	5	t <sub>e</sub>	0.9354	5	Flash Point °C Fire Point	
B [_2]0 °C		5	t <sub>c</sub>	0,24	1 3	M Spec.	
A* 70 to	209. 1.4701	5 5	ΔHc kcal/m ΔHf			Ultra V.	
B* 185 °C		5 5	ΔFf			X-Ray Dif. Infrared	
к — — —	7	- 1	Viscosity			Solubility in +	<del></del>
\$	<del>.</del> -		centistokes 7 °C			Acetone	:
tî t	;		'			Carbon tet. Benzene	
A'   25 to					1	Ether	
B' _ 70 °C	227.	5	B <sup>V</sup>   to	<del> </del>	$\vdash$	n-Heptane Ethanol	
A'* 25 to			ĀV ∤ °C			Water	
B'* 70 °C		5	(BV) to			Water in	
Ac   210 to		5	(A <sup>V</sup> ) °C				
Bc tc_°C	2225.	5	c <sub>p</sub> liq. °K				
Cryos. A°	1	<u> </u>	i	j			
consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	177.69	5	c <sub>v</sub> vap.				
$T_R = 0.8$	l T <sub>c</sub>		· · · · · · · · · · · · · · · · · · ·			+ grams/100 grams	solvent
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-Calc, by form	
SOURCE:		A	PI				
PURIFICAT	TION:	A	PI				
	RE REFER	ENCES	):				

NAME	3, 3-D	imet	hyloc	tane			ST	RUCTURAL	FORMU	
	·	r					c	сн <sub>3</sub> н <sub>3</sub> сн <sub>2</sub> сн (с	H-) (CH-	
Mole % Pur.	Ref.	Mo! For	ecul:		Molecular Weight 142.2	76	Ū	сн <sub>3</sub>	2/4	
			Ref.			Ref.				Ref.
F. P. ℃				dt/dP			f	to		
F.P. 100%				°C/mm			g	<u>•K</u>		
B. P. ℃	1,7,2		,	25°C BP	5.0083 0.05163	5	h	1		
760 mm 100	161.2 95.7		2	te	0.0367	5	f'	to		
30	66.78	В	4	30 mm	0.7250	5	g'	<u>K</u>		-
10	44.7		5 5	∆Hm cal/g		Γ	h'	<u>i</u>		<u> </u>
Pressure	†			∆Hv cal/g			m n	to °K		1
mm 25°C	3.19	931	5	25°C 30 mm	77.66 74.23	5	0	! ' <del></del>		
t <sub>e</sub>	1176.		5	BP	63.25	5	m'	1 1 4-		+-
Density g/ml 20°C	0.7	200	,	t <sub>e</sub>	61.24	5	n'	to °K		
dt 25 4 30	0.73		2	te (d, e)	61.16	5	0'			
<sup>d</sup> 4 30	0.7		4	ΔHv/T <sub>e</sub>	19.26	5	Sur	face tension		+
a	0.75		4	d	81.99 0.1162	5		es/cm. 20°C	23.57	5
b	-0.0	3/8	4	d 25 to	79.72	5	8	30 40	22.59	5
Ref. Index	1.41	165	2	e'   70 °C		5		rachor [P]	21.03	+-
45	1.41	142	2	d g/ml	0.242 4.136	5	F.	20°C	1	
30	1.41		4	v <sub>c</sub> ml/g t <sub>c</sub> °C	328.5	5		30 40	i	
"C"	0.74		4	P <sub>c</sub> mm	15304.	5			424. 2	5
MR (Obs.) MR (Calc.)			2 5	PV/RT	<del> </del>	1	Exp	L.1.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5		u.	07	2
Dielectric	2.00	07	5	BP BP	1.0000 0.9480	5		persion	97.	<del>  '</del> -
A 70 to	6.97	713	5	t <sub>e</sub>	0.9342	5		sh Point C e Point	ł	
B (215 °C	1512. 208.		5	t <sub>c</sub>	0, 24	5		Spec.		+-
A* 70 to	1.47	722	5	ΔHc kcal/m ΔHf			Ult	ra V.		
B* 190 °C		123	5	ΔFf				Ray Dif. rared		1
к — — —	-			Viscosity			ļ	ubility in +		+-
t <sub>k</sub> to	-			centistokes り °C			A	etone		
t <sub>x</sub>   °C				'				arbon tet. enz <b>e</b> ne		
A'   25 to B'   70 °C	7.33	308	5				Et	her		
c' - '- =	1716.0 226.		5	B <sub>v</sub> to				Heptane hanol		
A'* 25 to	1,81	491	5	A   °C			w.	ater		
B'* 70 °C			5	(B <sup>V</sup> ) to			<del></del>	ater in		+-
Ac 215 to	7.76	597	5	(A <sup>V</sup> )  °C	1					
Bc tc °C	304.		5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°				c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	179.15	;	5	c vap.						
$T_R = 0.8$			لـــــا	L			+ ~	rams/100 gra	ms solve	nt.
REFERENC		Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da				
SOURCE:				PI						
PURIFICAT	ION:			PI						
LITERATU		EREI								

							No. 90	
NAME	3, 4-D	imethylo	tane			STRUCTURAL	FORMUL	A.
						CH <sub>3</sub>		
			1			CH3CH2CH CH (	сн.).сн	•
Mole	Ref.	Molecul		Molecular Weight 142.	276	сн <sub>3</sub>	2.3	3
% Pur.		Formul		Weight 142.	Ref	<u> </u>		Ref
· · · ·	Τ	Ref.		T	Kei		Γ	Kei
F.P. °C F.P. 100%	<del> </del>		dt/dP *C/mm			f to		
B. P. °C	†		25°C	6. 1341	5	h .		İ
76 <b>0 mm</b> 10 <b>0</b>	166.	2	BP t <sub>e</sub>	0.05208 0.0367	5 5	f' to		†
30	99.9	4 4	90 mm	0.7320	5	g'   ' <u>°</u> K_		1
1 <b>0</b> 1	48.4	5	ΔHm cal/g	†		h'		
Pressure	11.1	-   3	ΔHv cal/g		$\Box$	m to		
mm 25°C	2.56	03 5	25°C	79.08 75.22	5	n   <u>*K</u> _		
t <sub>e</sub>	1190.	5	30 mm BP	64.10	5	<u> </u>		╁
Density g/ml 20°C	0.74	6 2	te (d.e)	62.01	5	m'   to		
dt 25	0.74		e (4, 5)	61.92	5	0'		1
<sup>4</sup> 30	0.73		ΔHv/T <sub>e</sub>	19.27	5	Surface tension		T
a b	0.76		d   70 to		5	dynes/cm. 20°C	24.47	5
Ref. Index	-0.03		a' 725 to	81.19	5	30 40	23.44 22.44	5
n <sub>D</sub> 20°C	1.41	82 2	<u> </u>	0.0844	5	Parachor [P]		
25 30	1.41		d g/ml vc ml/g	4, 128	5	20°C	1	
"C"	0.74		tc °C	335.8	5	30 40		
MR (Obs.)	48,1	2	P <sub>c</sub> mm	15521.	5	Sugd.	424.2	5
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	1.04		30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	2.01		BP	0.9480 0.9339	5	Flash Point °C		t
A 70 to B 225 °C		82 5 5	te t <sub>C</sub>	0.24	5	Fire Point		<u> </u>
č —	208.	5	ΔHc kcal/m			M Spec.		
A* 70 to	1.47		ΔHf ΔFf			Ultra V. X-Ray Dif.		1
B* 195 °C	1439.7	5	Viscosity		1	Infrared		<u> </u>
c			centistokes			Solubility in + Acetone		
tk   t0 t   •C			<b>η</b> γ • • • • • • • • • • • • • • • • • •	;		Carbon tet.		1
t		417 5		1		Benzene Ether		
B' _ 70 °C	1734.7	5			ļ	n-Heptane		
C'	226.	5	B <sup>V</sup> to			Ethanol Water		1
A'* 25 to B'* 70 °C		657 5		<b>-</b>		Water in		
Ac   225 to	1		(A <sup>V</sup> ) to	1				
Bc t °C	2341.	5	c <sub>p</sub> liq. °K					1
Cc	310.	5	ll .					1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K	:				1
t <sub>e</sub> °C	184.62	5	c <sub>w</sub> vap.	1			l	
$T_{R} = 0.8$			L	<u> </u>	L	+ grams/100 gran	ne solven	<del></del>
REFERENC		ow 2-AI	PI 3-Lit. 4-	Calc. from de	t. dat	ta 5-Calc. by for		<u> </u>
SOURCE:			PI					
PURIFICAT	'ION:	A	PI					
LITERATU								

No. 91 3, 5-Dimethyloctane NAME STRUCTURAL FORMULA CH3CH2CH CH2CH (CH2)2CH3 Mole Ref. Molecular Molecular CH<sub>3</sub> CH<sub>3</sub>  $C_{10}H_{22}$ % Pur Weight 142, 276 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g <u>°K</u> 25°C 9.4812 B. P. °C h ВP 0.05150 5 760 mm 160. 2 t<sub>e</sub> 0.0367 5 ſ١ 100 94.7 4 to 0.7231 5 °<u>K</u> 4 g' 65.82 30 30 mm 5 10 43.8 h١ 5 ∆Hm cal/g 6.9 to AHv cal/g Pressure n ۰ĸ 25°C 77.33 mm 25°C 3.3674 o 5 30 mm 74.00 1172. 5 te BР 63.06 5 m to Density 61.07 te te (d, e) °K g/m1 20°C 0.736 2 60.99 5 o' 25 0.732  $\mathbf{d_4^t}$ 2 ΔHv/T<sub>e</sub> 19.27 5 30 0.728 4 Surface tension d 70 to 81,65 5 0.752 44 dynes/cm. 20°C 23.19 <u>°C</u> 0.1162 5 -0.038 ь 30 22.19 āΠ 25 to 79.38 5 5 21.23 40 Ref. Index 70 °C 0.0817 5  $^{\mathbf{n}}\mathbf{D}$ 1.413 20°C [P] Parachor d<sub>c</sub> g/ml v<sub>c</sub> ml/g t<sub>c</sub> °C 5 0.240 25 1.411 2 20°C 4.170 30 1.408 4 30 t<sub>c</sub> 325.8 5 40 "C" 0.7468 4 P<sub>c</sub> mm 15110. 5 Sugd. 424.2 5 MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48 38 5 25°C 1.0000 (nD-d/2)1.045 2 30 mm 1.0000 Dispersion 97. 2 Dielectric 2.039 5 ВP 0.9480 5 Flash Point C 0.9343 5 A 70 to 6.9710 5 Fire Point 0.24 B 1\_215 °C 1508. 5 M. Spec. C 209. 5 AHc kcal/m Ultra V. ΔHf A# 70 to 1.47305 5 X-Ray Dif. ΔFf B\* 185 °C 1418.2 Infrared K Viscosity Solubility in centistokes Acetone to t<sub>k</sub> t<sub>x</sub> Carbon tet. °C Benzene A' 25 to 7.33132 Ether В¹ 70 °C 1711.9 5 n-Heptane B<sub>v</sub> | C' 227. 5 to Ethanol °C Water 1.81589 5 25 to 70 °C Water in (B<sup>V</sup>)| 5 1611.5 to Ac 215 to 7.7550 5 (A<sup>V</sup>) °C Bc tc °C 2243. 5 c<sub>p</sub> liq. °K Cc 301. 5 Cryos. Aº c<sub>p</sub> vap. °K consts. B° c vap. te °C 177.78  $= 0.81 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

·····							No. 92	
NAME	3,6-Dime	thylo	tane			STRUCTURAL F	ORMULA	
						CH CH CH (CH )	רם כם כ	u
Mole	Ref. Mo	lecul		Molecular		CH <sub>3</sub> CH <sub>2</sub> CH (CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	сн сп <sub>2</sub> с сн <sub>3</sub>	113
% Pur.		rmul	C <sub>10</sub> H <sub>22</sub>	Weight 142.	276	3		
		Ref.			Ref.			Ref
F.P. C			dt/dP			f to		
F. P. 100% B. P. °C	<b>}</b>	$\vdash$	*C/mm 25*C	4.7691	·5	g  K		
760 mm	160.	2	BP	0.05150 0.0367	5 5	f' to		-
100 30	94.7 65.8	4 4	t <sub>e</sub> 30 mm	0. 7231	5	g' to		
10	43.8	5	ΔHm cal/g	0.1251	Ť	h'		
1	6.9	5	ΔHv cal/g		_	m to		
Pressure mm 25°C	3.3674	5	25°C 30 mm	77.33 74.00	5 5	n •K		
t <sub>e</sub>	1172.	5	BP BP	63.06	5	m'   to		
Density g/ml 20°C	0.7363	2	te t (d, e)	61.07	5	n'  °K_		
dt 25 d4 30	0.7324 0.7285	2	ΔHv/T <sub>e</sub>	19.27	5	0'		
2 30	0.7285	4	d   70 to	81.65	5	Surface tension	23, 22	5
b	-0.0378	4	e   175 °C   d   25 to	0.1162 79.38	5	dynes/cm. 20°C	22.25	5
Ref. Index	1 4145		e' 70 °C	0.0817	5	40	21.31	5
n <sub>D</sub> 20°C 25	1.4145 1.4122	2 2	d <sub>c</sub> g/ml	0.241	5	Parachor [P] 20°C		
30	1.4099	4	d g/ml v ml/g tc °C	4.147 326.4	5	30 40		
"C"	0.7491	4	P <sub>c</sub> mm	15209.	5		424.2	5
MR (Obs.) MR (Calc.)	48.337 48.38	2 5	PV/RT	1 0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0463	2	25°C 30 mm	1.0000 1.0000	5	u. Dispersion	97.	2
Dielectric	.2.001	5	BP t <sub>e</sub>	0.9480	5	Flash Point °C		
A 70 to B 215 °C	6.9710 1508.	5 5	tc	0.24	5	Fire Point		-
C	209.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 70 to B* 185 °C	1.47305	5 5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
K			Viscosity			Solubility in +		<del>                                     </del>
t <sub>k</sub> to			centistokes 7°C			Acetone Carbon tet.		
<b>1</b>	2 22122	<b> </b>	·			Benzene		
A'   25 to B'   70 °C	7.33132 1711.89	5				Ether n-Heptane		
C'	227.	5	B <sup>V</sup>   to A <sup>V</sup>   °C	ļ		Ethanol Water		
A'* 25 to B'* 70 °C	1.81589	5	(B <sup>V</sup> ) to	1		Water in		
Ac   215 to	7.7581	5	(A <sup>V</sup> ) °C					
Bc t <sub>c</sub> °C	2247. 302.	5	c <sub>p</sub> liq. °K	<u> </u>		1		
Cryos. A°	†	+-	c <sub>p</sub> vap. °K					
consts. B°	<b>_</b>	$\perp$	i -					
t <sub>e</sub> °C	177.78	5	c <sub>w</sub> vap.	L				
TR = 0.81 REFERENC		2 4 -	27 2 724 4 5	- 1- <i>(</i> 1		grams/100 gran		<u>t</u>
SOURCE:	±3: 1-D0₩	2-AF		alc. from det	. da	ta 5-Calc. by form	nula	
PURIFICAT	ION:	AP						
	RE REFERE							

No. 93 4, 4-Dimethyloctane NAME STRUCTURAL FORMULA сн3  $CH_{3}(CH_{2})_{2}C - (CH_{2})_{3}CH_{3}$ Mole Ref. Molecular Molecular Weight 142.276 ĊH3  $C_{10}H_{22}$ % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g \_°<u>K</u> 4.9725 5 B. P. °C h ВP 0.05160 5 760 mm 161. 2 0.0367 5 ſ١ to 100 95.6 4 g' <u>•K</u> 30 30 mm 5 66.64 4 0.7246 10 44.6 5 h! ∆Hm cal/g 7.6 1 5 m to AHv cal/g Pressure °K n 25°C 77.62 5 mm 25°C 3,2176 o 30 mm 74.20 5 1175. 5 te BP 63.23 5 m' to Density 5 t<sub>e</sub> (d, e) 61.22 n' g/ml 20°C °K 2 0.737 61.15 ۰,  $\mathbf{d_{4}^{t}}$ 25 0.733 2 ΔHv/Te 19.27 5 30 0.729 4 Surface tension Т 70 to ď 81.95 5 0.753 a 4 180 °C 25 to dynes/cm. 20°C 23.31 5 0.1163 ь -0.038 4 22.31 30 ď٦ 79.68 5 5 21.35 40 Ref. Index e' 70 °C 0.0821 5 <sup>n</sup>D 20°C 1.414 Parachor [P] d<sub>c</sub> g/ml 0.240 5 25 2 1.412 20°C vc ml/g t °C 4.168 327.4 5 30 1.409 4 30 **t**c 5 40 "C" 0.7475 4  $P_c$  mm 15157. 5 Sugd. 424.2 5 48.2 MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 25°C 1,0000 5 (nD-d/2)1.046 2 u. 30 mm 1.0000 5 Dispersion 97. 2 2,000 Dielectric 5 BP 5 0.9480 Flash Point C 0.9342 5 A 70 to 6.9720 5 Fire Point 0.24 5 1511. 1 215 °C M. Spec. 208. 5 AHc kcal/m Ultra V. ΔHf A\* 70 to 1.47331 X-Ray Dif. ΔFf B\* 190 °C 1421.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $\mathbf{t_k} \mid \mathbf{t_x} \mid$ Carbon tet. °C Benzene 25 to 7.33180 Ether B' L 70 °C 1715.69 n-Heptane B<sub>v</sub> | 226. 5 Ethanol to °C Water A'\* 25 to 1.81600 5 Water in B'\* 70 °C (B<sup>V</sup>)| 1615.2 to Ac 215 to 7,7641 (A<sup>V</sup>) 5 °C Bc tc °C 2257. c<sub>p</sub> liq. ۰ĸ Cc 303. 5 Cryos. A° c<sub>p</sub> vap. °K consts. B° c vap. te °C 178,92  $T_{\mathbf{R}} = 0.81 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

NAME	4,5-Din	nethyloc	tane			STRUCTURAL 1		
Mole % Pur.		Molecula Formula		Molecular Weight 142.2	276	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн-сн сн <sub>3</sub>		H <sub>3</sub>
		Ref			Ref			Ref
F.P. °C	T	1		Т				1
F.P. 100%	<del> </del>	+	dt/dP *C/mm			f to		1
B. P. *C	<del> </del>	+	25°C	5,2038	5	g - 1		1
760 mm	162.13	2	BP	0.05173		h		┼
100	96.5	4	t <sub>e</sub>	0.0367	5	f' to		1
30 1 <b>0</b>	67.5	4	30 mm	0.7264	5	g'K_		
1	45.4 8.3	5 5	∆Hm cal/g			h'		
Pressure	0.3		ΔHv cal/g			m   to		1
mm 25°C	3.0629	5	25°C	77.92	5	n   *K		Į
t <sub>e</sub>	1178.	5	30 mm	74.41	5			
Density	<del> </del>		BP	63.41	5	m' to		
g/ml 20°C	0.7470	2	te (d, e)	61.31	5	n'   °K		
t 25	0.7432		AHV/T	19.26	5	0'		
<sup>4</sup> 4 30	0.7394	4	ΔHv/T <sub>e</sub>	<del></del>	5	Surface tension		$\top$
<b>a</b>	0.7622		d   70 to		5	dynes/cm. 20°C	24.60	5
ь	-0.0 <sub>3</sub> 76	4	a - 1 25 to		5	30	23.62	5
Ref. Index	1	1 1	e' 70 °C		5	40	22.66	5
<sup>n</sup> D 20°C	1.4190		d g/ml	0,246	5	Parachor [P]		1
30	1.4167		d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	4.062	5	20°C 30		1
"C"	<del></del>		tc °C °	331.6	5	40		1
	0,7459		P <sub>c</sub> mm	15661.	5	Sugd.	424.2	5
MR (Obs.) MR (Calc.)	48.10 48.38	2 5	PV/RT			Exp. L.1.%/wt.		1
(nD-d/2)	1.0455		25°C	1.0000	5	u.		
Dielectric		-	30 mm	1.0000	5	Dispersion	95.8	2
A 70 to	2.013	5	BP t <sub>e</sub>	0.9480 0.9341	5	Flash Point °C		T
B 220 °C	6.9719	5 5	tc	0.24	5	Fire Point		ــــــ
č —— =	208.	5	ΔHc kcal/m		<del>                                     </del>	M Spec.		1
A*  70 to	1,4719	3 5	ΔHf		1	Ultra V. X-Ray Dif.		1
B+ 190 °C		Ĭ   Š	ΔFf	<u> </u>		Infrared		1
к ———	l	1 1	Viscosity			Solubility in +		+
:	}		centistokes	.		Acetone		ì
tk to			η •ο	1		Carbon tet.		1
t C A'   25 to	7.3306	1 5	ŀ		1	Benzene		1
	1719.3	1 5				Ether n-Heptane		1
C'	226.	5	B <sup>V</sup> to			Ethanol		
A'* 25 to	1,8144	3 5	A <sup>V</sup>   °C	:	i	Water		1
B'* 70 °C	1618.8	5	(B <sup>V</sup> ) to		1	Water in		ـــــ
Ac   220 to	7, 7883	5	(A <sup>V</sup> ) °C	ł				1
Bc tc °C	2295.	5			-	ij .		1
Cc	307.	5	c <sub>p</sub> liq. °K	·				1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K	:	l			ł
·t <sub>e</sub> °C	180,22	5	c <sub>v</sub> vap.					Ì
$T_R = 0.81$		1	1	<u> </u>	L	1,00	<u> </u>	Щ.
REFERENC		. 2 45	N 2 7 11 4			grams/100 gran		ıt
	E3: 1-D0W			Calc. from det	. da	ta 5-Calc, by form	nula	
SOURCE:			PI					
PURIFICAT			PI					
LITERATU	RE REFER	ENCES	:					

No. 95 4-Propylheptane STRUCTURAL FORMULA NAME CH3(CH2)2CH (CH2)2CH3 Mole Ref. Molecular Molecular Weight 142, 276 Ċ<sub>3</sub>H<sub>7</sub>  $C_{10}H_{22}$ % Pur Formula Ref. Ref. Ref. <u>F.P. °C</u> dt/dP f to F.P. 100% °C/mm g <u>°K</u> 25°C 5.1849 5 B.P. °C h BP 0.05169 760 mm 162. 2 5 0.0367 ſ١ to 96.4 100 4 0.7261 g' <u>• K</u> 5 67.5 30 4 30 mm 10 45.3 5 h١ ∆Hm cal/g 8.3 5 to m AHv cal/g Pressure °K n 25°C 77.91 5 3.0744 mm 25°C o 30 mm 74.41 1178. 5 te BP 63.41 5 m' to Density te te (d, e) 5 61.38 n' g/ml 20°C °K 0.7364 2 5 61.30 ۰, ď4 25 0.7326 2 ΔHv/T<sub>e</sub> 19.27 5 30 0.7288 4 Surface tension 65 **to** 82, 26 5 0.7516 4 dynes/cm. 20°C 23, 24 180 °C 0.1164 5 ь -0.0376 4 30 22.29 5 aח 25 to 79.98 5 40 21.37 5 Ref. Index e١ 65 °C 0.0826 5 20°C 1.4150 [P] <sup>n</sup>D Parachor 5 d<sub>c</sub> g/ml 0.242 25 1.4127 2 20°C vc ml/g t °C 4.135 30 1.4105 4 30 t<sub>c</sub> 329.6 5 40 "C" 0.7498 4 P<sub>c</sub> mm 15334. 5 Sugd. 424.2 5 MR (Obs.) 48.38 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2)2 1.0468 30 mm 1.0000 5 Dispersion 96. 2 Dielectric 2,002 5 BP 0.9480 5 Flash Point C 0.9342 5 65 to 6.9734 5 Fire Point 0.24 В \_220 °C 1515. 5 M. Spec. С 208. 5 AHc kcal/m Ultra V. ΔHf A\* 65 to 1.47356 5 X-Ray Dif. ΔFf B\*| 190 °C 1425.4 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. °C t<sub>x</sub> Benzene 25 to 7.33227 Ether B' 1719.5 65 °C 5 n-Heptane Bv | Av | C 226. 5 Ethanol °C Water 25 to 65 °C 1.81612 5 Water in B'\* 1619.0 (B<sup>V</sup>)| to Acl 220 to 7,7776 5 (AV) °C 2278. Bc tc °C c<sub>p</sub> liq. °K 5 Cc 305. c<sub>p</sub> vap. Cryos. Aº °K consts. B° c<sub>v</sub> vap. te °C 180.06 5  $T_{R} = 0.81 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 96	
NAME	4-Isoprop	ylher	otane			STRUCTURAL I	ORMULA	١
						CU (CU ) CU	CH / CH	
						сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн - (	CH <sub>2</sub> / <sub>2</sub> CH	3
Mole % Pur.	Ref. Mo	rmul	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.	276	C <sub>3</sub> H <sub>7</sub>		
		Ref.	7		Ref.			Ref.
F. P. °C			dt/dP	1		f to		
F.P. 1007	<u> </u>		*C/mm 25°C	4, 7691	5	g <u>  K</u>		
B. P. °C 760 mm	160.	2	BP	0.05150	5	h		<u> </u>
100 30	94.7 65.8	4	t <sub>e</sub>	0.0367	5	f' to g'°K		
10	43.8	5	30 mm	0.7231	3	h'		
1	6.9	5	ΔHv cal/g	-	-	m to		
Pressure mm 25°C	3. 3674	5	25°C	77. 33	5	n   °K_		
t <sub>e</sub>	1172.	5	30 mm BP	74.00 63.06	5	<u>                                     </u>		
Density g/ml 20°0	0.741		t.	61.07	5 5	m' to		
dt 25	0.741 0.737	2	te (d, e)	61.00	5	0'		
	0.733	4	ΔHv/T <sub>e</sub>	19.27	5	Surface tension		
a b	0.757 -0.0 <sub>3</sub> 8	4	e   175 °C	0.1161	5	dynes/cm. 20°C	23.82 22.81	5
Ref. Index		Ė	a'   25 to		5	40	21.82	5
n <sub>D</sub> 20°0	1.417 1.415	2 2	d <sub>c</sub> g/ml	0, 242	5	Parachor [P]		
30	1.412	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.133 326.7	5	20°C 30		
"C"	0.7486	4		15270.	5	40 Sugal	424.2	5
MR (Obs.)		2	P <sub>c</sub> mm	13210.	-	Exp. L.l.%/wt.	727.2	-
MR (Calc. (nD-d/2)	) 48.38 1.047	5 2	25°C	1.0000	5	u.		_
Dielectric	2.008	5	30 mm BP	1.0000 0.9480	5	Dispersion	97.	2
A 70 t		5	t <sub>e</sub>	0.9343 0.24	5	Flash Point °C Fire Point		
B L215°C	209.	5	t <sub>c</sub> AHc kcal/m	0.21	-	M Spec.		
A* 70 to	1.47305	5	ΔHf			Ultra V. X-Ray Dif.		
B* 187°C	1418.2	5	ΔFf Viscosity	<del> </del>	-	Infrared		
c	_	ŀ	centistokes	1		Solubility in + Acetone		ł
t <sub>k</sub> to			7 ℃			Carbon tet.		
A'   25 to		5				Benzene Ether		
B' _ 70°	227.	5	B <sup>V</sup>   to		-	n-Heptane		l
A'* 25 to	<del></del>	5	Av C			Ethanol Water		
B'* 70°0		5	(BV) to			Water in		
Ac   215 to	7.7600	5	(A <sup>V</sup> )  °C					
Bctc_°C	2250. 302.	5	c <sub>p</sub> liq. °K					
Cryos. A			c <sub>p</sub> vap. °K					
consts. B°	177.78	5	c <sub>v</sub> vap.					
T <sub>R</sub> = 0.8			ļ ·	<u> </u>		+ grame/100 a	20 201	<u> </u>
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from det	dat	grams/100 gran ta 5-Calc. by form		<u> </u>
SOURCE:			PI					
PURIFICA			PI					
LITERATU	RE REFERE	NCES	:		-			

NAME	3-Ethyl-2-methylheptane							STRUCTURAL FORMULA				
								C <sub>2</sub> H <sub>5</sub>				
Mole % Pur.	Re	f. Mo	lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.	276	CH <sub>3</sub> CH CH (CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>					
-/• 1 u1.		1.0	Ref.	r — —	T = ===	Ref.	Γ			Ref.		
F.P. °C F.P. 100	%			dt/dP °C/mm 25°C	( 1241		f g	to° <u>K</u>				
B. P. °C 760 mm 100	166. 99.		2 4	BP t <sub>e</sub>	6.1341 0.05208 0.0367	5 5 5	h   f'   g'	to *K		$\vdash$		
30 10	70. 48.	4	5	30 mm	0.7320	-5-	h' i	<u>B</u>				
1	11.	1	5	ΔHm cal/g ΔHv cal/g	+	-	m	to				
Pressure mm 25°C	2. 1189.	5603	5 5	25°C 30 mm BP	79.08 75.22 64.06	5 5 5	n o	<u>•K</u>		1		
Density g/ml 20° dt 25 4 30		746 742	2 2	t <sub>e</sub> (d, e)	61.97 61.89	5	m'   n'   o'	to •K				
		738	4	ΔHv/T <sub>e</sub>	19.26	<b>5</b>		ce tension				
a b	0. -0.	762 0 <sub>3</sub> 8	4 4	e 185 °C	0.1171	<b>5</b>	dynes	/cm. 20°C	24.47 23.44	5		
Ref. Inde		410		e' 75 °C		5		40	22.43	5		
<sup>n</sup> D 20° 25 30	1. 1.	418 416 413	2 2 4	d g/ml vc ml/g tc °C	0.242 4.129 335.8	5 5 5	Parac	thor [P] 20°C 30 40				
"C"		7452	4	P <sub>c</sub> mm	15514.	5			424.2	5		
MR (Obs. MR (Calc (nD-d/2)	.) 48.		2 5 2	PV/RT 25°C 30 mm	1.0000	5	Exp.	L. l. %/wt. u. rsion	96.	2		
Dielectri	<del></del>	011	5	BP	0.9475	5		Point °C	70.	† <u>-</u> -		
A 75 to B 1 225 °C		9782	5 5 5	te tc ΔHc kcal/m	0.9333 0.24	5	Fire I M. Sp Ultra	ec.		-		
A* 75 to B* 195 °	1. 1439.	47547 9	5 <b>5</b>	ΔHf ΔFf Viscosity			X-Ray Infrar	y Dif. ed				
				centistokes 7°C			Acet Carb	on tet.				
A'   25 to B'   75°		33417 7	5 5 5	B <sup>V</sup> to		-	Benz Ethe n-He Etha	r ep <b>tane</b>				
A!* 25 to B!* 75 °	o 1.	81657 1	5 5	$ \begin{array}{c c} B^{\mathbf{V}} & \text{to} \\ A^{\mathbf{V}} & - \\ \hline (B^{\mathbf{V}}) & \text{to} \end{array} $	-1		Wate Wate	r				
Acl 225 to Bc t <sub>c</sub> °		8159	5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K	+	-						
Cryos. A	•			c <sub>p</sub> vap. °K					:			
t <sub>e</sub> °C	184.	59	5	c <sub>v</sub> vap.								
$T_R = 0.$								ns/100 gra		nt		
REFEREN	CES: 1	-Dow			-Calc. from de	et. da	ta 5-C	alc. by for	mula			
SOURCE:	TION		AI									
LITERAT		FERF		· · · · · · · · · · · · · · · · · · ·								

							<b>N</b> o. 98	
NAME	4-Ethyl-2-	meth	nylheptane			STRUCTURAL I	ORMULA	4
						C.1. C.1. C.1. C.1.	1CT \ CT	
	7.//			N-1		CH <sub>3</sub> CH-CH <sub>2</sub> CH- CH <sub>3</sub> C <sub>2</sub> H <sub>4</sub>		<sup>-1</sup> 3
Mole % Pur.	Ref. Mo	rmul	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	76	0113 02115	,	
		Ref.			Ref.			Ref.
F.P. °C			dt/dP			f to		
F.P. 100%			*C/mm 25*C	5 4013	5	g <u>*K</u> _		
B. P. °C 760 mm	160.	2	BP	5.4812 0.05146		h		1
100	94.7	4	t <sub>e</sub>	0.0367	5	f' to		l
30 10	65.9 43.8	4 5	30 mm	0.7229	5	g'K_		
ĭ	7.0	5	∆Hm cal/g			h'		$\vdash$
Pressure			ΔHv cal/g 25°C	77, 38	5	m to		
mm 25°C	3.3554 1172.	5	30 mm	74.04	5	•		
t <sub>e</sub> Density	1112.	-	BP	63.10	5 5	m'   to		
g/ml 20°C	0.736	2	te te (d, e)	61.11	5	n'  °K_		
d <sub>4</sub> 25	0.732	2 4	ΔHv/Te	19.28	5	o'		
	0, 728	4	d   70 to	<del></del>	5	Surface tension		١.
a b	-0.038	4		0.1163	5	dynes/cm. 20°C	23.19 22.19	5
Ref. Index			d'   25 to		5 5	40	21.23	5
n <sub>D</sub> 20°C	1.413	2	d g/ml	0.240	5	Parachor [P]		
30	1.411 1.408	2 4	d g/ml vc ml/g	4.163	5	20°C		
"C"	0.7468	4	1c C	325. 8	5	40	l	1
MR (Obs.)	48.2	2	P <sub>c</sub> mm	15136.	5		424.2	5
MR (Calc.)		5 2	PV/RT 25°C	1.0000	5	Exp. L.l.%/wt. u.		1
(nD-d/2) Dielectric	1.045	5	30 mm	1.0000	5	Dispersion	96.	2
A 70 to	1.996 6.9737	5	BP t <sub>e</sub>	0.9480 0.9 <b>3</b> 43	5 5	Flash Point °C		
B 1215°C	1509.	5	tc	0.24	5	Fire Point		<b>├</b> ──
С	209.	5	ΔHc kcal/m			M Spec. Ultra V.		
A* 70 to B* 185 °C	1.47578 1419.2	5	ΔHf ΔFf	ł		X-Ray Dif.		1
K Lie C	1417.2	] ]	Viscosity			Infrared Solubility in +		<del> </del>
t to	ŀ		centistokes 7°C			Solubility in †		ļ
t <sub>x</sub> to		1	7 ℃			Carbon tet.		1
A'   25 to	7.33416	5				Benzene Ether		
B' ∟ 70 °C	1713. 227.	5	B <sup>V</sup> to	<del> </del>	$\vdash$	n-Heptane		
A'* 25 to	1.81869	5	B' to	1		Ethanol Water		
B' ≠ 70 °C	1612.6	5	(B <sup>V</sup> ) to	-		Water in		<u> </u>
Ac   215 to	7.7575	5	(A <sup>V</sup> ) °C	ł				
Bc tc_°C	2243. 301.	5	c <sub>p</sub> liq. °K	1	$\vdash$			
Cryos. A°	301.	1	_					
consts. B°			c <sub>p</sub> vap. °K					1
t <sub>e</sub> °C	177.77	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.81$	T <sub>C</sub>					+ grams/100 gran	ns solven	t
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-(	Calc. from de	t. dat	a 5-Calc, by form		
SOURCE:		A.	PI					
PURIFICAT			PI					
LITERATUR	RE REFERE	CES	:					

No. 99 5-Ethyl-2-methylheptane STRUCTURAL FORMULA NAME CH3CH (CH2)2CH CH2CH3 Mole Ref. Molecular Molecular Ċнз  $C_{10}H_{22}$ Ċ<sub>2</sub>H<sub>5</sub> Weight 142, 276 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 4.6975 5 B. P. °C h BP 0.05150 5 760 mm 159.7 2 0.0367 5 ſ١ to 100 94.4 4 g' <u>×</u> 65.5 30 4 30 mm 5 0.7229 10 43.5 5 h\* ∆Hm cal/g 6.6 to m AHv cal/g Pressure ۰ĸ n 25°C 77.20 mm 25°C 3.42450 o 30 mm 73.90 1171. te 5 62.97 5 BP m' to Density 60.98 5 n' °K g/ml 20°C te (d, e) 0.736 60.91 5 o'  $\mathbf{d_4^t}$ 25 0.732 AHv/Te 19.25 5 30 0.728 4 Surface tension 70 to 81.51 5 0.752 a 4 dynes/cm. 20°C 23.19 5 75 °C 25 to 175 0.1161 5 ь -0.038 4 30 22.19 ā٦ 79.24 40 21.23 5 e¹ Ref. Index 70 0.0815 n<sub>D</sub> 20°C 1.4134 [P] Parachor d<sub>c</sub> g/ml 0.240 5 25 1.4109 2 20°C vc ml/g t\_°C 4.175 30 1.4086 4 30 <sup>t</sup>c 325.4 40 "C" 0.7475 4 P<sub>c</sub> mm 15082. 5 Sugd 424.2 5 MR (Obs.) 48.2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2)u. 1.045 2 30 mm 1.0000 5 Dispersion 96. 2 0.9480 Dielectric 1.998 5 BP 5 Flash Point C 0.9343 5 A 70 to 6.9681 5 Fire Point 0.24 5 B (215 °C 1506. M. Spec. AHc kcal/m С 209. 5 Ultra V. A\* 70 to B\* 185 °C ΔHf 1.47051 5 X-Ray Dif. ΔFf 1416.2 Infrared ĸ Viscosity Solubility in centistokes Acetone Carbon tet. °C <sup>t</sup>x\_ Benzene 25 to 7, 32861 Ether B' |\_ 70 °C 1709.8 n-Heptane B<sub>v</sub> | 227. 5 to Ethanol °C Water A'\* 25 to B'\* 70 °C 1.81332 5 1609.4 Water in (B<sup>V</sup>) to Acl 215 to 7.7503 5 (A<sup>V</sup>)| °C Bc tc °C 2238. c<sub>p</sub> liq. °K 301. 5 Cryos. A c<sub>p</sub> vap. ٩K consts. B° c vap. te °C 177.45  $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

												No. 10	0
NAME		3-Etl	hyl-	3-me	thylheptar	ıe				STR	UCTURAL		
			<u> </u>						$\dashv$		C <sub>2</sub> H <sub>5</sub>		
		Т							$\dashv$	СН	<sub>3</sub> сн <sub>2</sub> с่ - (с	H <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub>	
Mole		Ref.		lecul			Molecular		.		CH,	2.3 3	
% Pur.	_	ليبا	Fo	rmul	a -10	22	Weight 14.	<del></del> -	_				<b>.</b>
	_			Ref.			1	-	Ref	<u></u>	. 1		Ref.
F.P. °C F.P. 1009	÷			$\vdash$	dt/dP •C/mr	_		- 1		f	to*K_		ľ
B. P. °C	+			$\vdash$	25°C	••	5.581		5	g h			1
760 mm		163.8		2	BP		0.051		5	f'	to		+-
100 30		97.98 68.89		4 4	t <sub>e</sub> 30 mm	,	0.728	- 1	5	g'			
10		46.67		5	ΔHm ca				Ť	h'	1		1
1	+	9.50		3	ΔHv ca	<u> </u>	<del> </del>			m	to		
Pressure mm 25°C		2.83	8	5	25°C	-	78.41		5	n o	<u>*</u> K_		
t <sub>e</sub>	1	183.2		5	30 mm BP	1	74.75 63.70		5		!		┼
Density g/ml 20°0		0.75	Λ1		t <sub>e</sub> ,		61.65		5	m' n'	to or or or or or or or or or or or or or		l
dt 25	1	0.75 0.74		2 2	t (d, e		61.57	ĺ	5	0'	'		
<sup>u</sup> 4 30	$\perp$	0.74	25	5	ΔHv/7	-	19.26		5	Surf	ace tension		<del>                                     </del>
a b		0.76		5 5	d     e   1	68 to		4	5		s/cm. 20°C	25.01	5
Ref. Index	+	-0.03		╁┤	d'	20 to	80.49	.	5	•	30 <b>4</b> 0	24.01 23.04	5
n <sub>D</sub> 20°0		1.42		2	<del></del>		0.08	$\rightarrow$	5	Para	chor [P]		<del>                                     </del>
25 30		1.41 1.41		4 1	(v ml/		4. 045		5		20°C 30		
"C"	+	0.74		4	tc °C		334.		5		40		
MR (Obs.	+	48.08		2	P <sub>c</sub> mm		15790.		5			424.2	5
MR (Calc.		48.38		5	PV/RT 25°C		1,000	, l	5	Exp	. L.1.%/wt. u.		
(nD-d/2)	+	1.04		2	30 mm	1	1.000	00	5	Disp	ersion	97.	2
Dielectric		2.01		5	BP t <sub>e</sub>		0.948		5		h Point °C		
A 68 t		6.97 521.3	39	5	tc		0, 24	.	5		Point		ļ
С	1	208.		5	∆Hc kca	ıl/m				M S			
A* 68 to		1.47	24	5 5	ΔHf ΔFf					X-R	ay Dif.		1
B* 200 °C	-  '	431.2			Viscosi	ty	<u> </u>			Infra			-
tt					centisto	ke s	.	ŀ	l		bility in + etone		1
t <sub>k</sub> to					7	°C	1		1		rbon tet.		
A'   20 to		7.33	14	5			į	l		Eth	nzene er		1
B'   68 °		725.6 226.0		5 5	B	to	<u> </u>	-			leptane anol		
A'* 20 to	-	1.81	47	5	A	•6			1	Wa			
B'* 68 °		625.2		5	(B <sup>V</sup> )	to	-			Wa	ter in		-
Ac   223 to		7.80	61	5	(A <sup>V</sup> )	°C							
Bc tc_°	- -	323. 9 309. 6		5 5	c <sub>p</sub> liq.	°K				1			1
Cryos. A					c <sub>p</sub> vap.	°K							
consts. B	4				_					1			1
te °C		182.12		5	c <sub>v</sub> vap.		]			L,			1
$T_{R} = 0.8$											ms/100 gran		ıt
REFEREN	CES	: 1-De	)W		I 3-Lit	4-	Calc. from	det	. dat	a 5-	Calc. by for	nula	
SOURCE:				AI			····						
PURIFICA				AI									
LITERATU	RE	REFI	CRE	NCES	:								

					<del></del>		No. 10	<u>,,                                   </u>
NAME	4-Ethy	1-3-met	hylheptan <b>e</b>			STRUCTURAL	FORMUL	A
						C <sub>2</sub> H		
						сн <sub>3</sub> сн <sub>2</sub> сн сн	(CH <sub>2</sub> ) <sub>2</sub> CF	1,
Mole % Pur.	Ref.	Molect	la C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	276	с́н,		•
-/0 T U I .		Re		Weight 142.7	Ref.			Ref.
F. P. ℃			dt/dP		Ker.	1	T	101.
F.P. 100%			°C/mm			f to		
B. P. °C			25°C	6. 3987	5	h		
760 mm 100	167.	2	BP t <sub>e</sub>	0.05218 0.0367	5	f' to		$\vdash$
30	100.8	4	30 mm	0.7335	5	g'° <u>K</u>		
10	49.2	5	ΔHm cal/g	1	+ -	h'		1
1	11.7	5	ΔHv cal/g		$\dagger$	m to		
Pressure mm 25°C	2.44	54 5	25°C	79.37	5	n   •K	-	
t <sub>e</sub>	1192.	5	30 mm BP	75.43 64.27	5			↓
Density		. T	t_	62.17	5	m' to		
g/m1 20°C _t 25	0.75		t <sub>e</sub> (d, e)	62.08	5	",   <u> </u>	1	
d <sub>4</sub> 25	0.74		ΔHv/T <sub>e</sub>	19.27	5	Surface tension		+
a L	0.76		d   75 to		5	dynes/cm. 20°C	25.40	5
b D-C T-d	-0.03	8 4		81.49	5	8 30 40	24.34	5
Ref. Index	1,42	2 2	e'   75 °C	+	5	Parachor [P]	1	Ť
- 25	1.41	9 2	d g/ml vc ml/g	0.245 4.081	5	20°C		1
30	1.41		tc °C	338.4	5	30 40		
	0.74		P <sub>c</sub> mm	15764.	5		424.2	5
MR (Obs.) MR (Calc.)	48.0 48.38	5	PV/RT 25°C	1 0000	_	Exp. L.1.%/wt.		
(nD-d/2)	1.04	6 2	30 mm	1.0000	5	u. Dispersion	96.	2
Dielectric	2.02	2 5	BP	0.9480	5	Flash Point °C	100	Ť
A 75 to	6.97 1534.	94 5	t <sub>e</sub>	0.9338 0.24	5	Fire Point		ļ
B 1 230 °C_ C	207.	5	ΔHc kcal/m	+	+	M. Spec.		
A* 75 to	1.47	482 5	ΔHf		1	Ultra V. X-Ray Dif.		
<b>B</b> *  195 °C K	1443.3	5	ΔFf		┼	Infrared		
С			Viscosity centistokes		1	Solubility in +		
t <sub>k</sub> to			7 °C			Acetone Carbon tet.		
t <sub>x  </sub> °C A'  25 to	7, 33	464 5	4			Benzene		
B'	1738.4	5	<b>-</b>		1	Ether n-Heptane		
C'	225.	5	B <sup>V</sup> to A <sup>V</sup> °C			Ethanol		
A'* 25 to B'* 75 °C	1.81 1637.9	669   5		-		Water Water in		
Acl 230 to	7.83	+	-					1
Bc <sub>l</sub> t <sub>c</sub> ℃	2367.	5		<del>                                     </del>	<del> </del>	-		
Cc — —	313.	5	P -					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	185.76	5	c, vap.					
$T_{\mathbf{R}} = 0.82$			1 -		1	grams/100 gra	<b>m</b> o cola:-	<del></del>
REFERENC		ow 2-	API 3-Lit. 4	-Calc. from de	t. da	ata 5-Calc. by for		ııt
SOURCE:			PI	u		5 CLIC. D, 101		
PURIFICAT	ION:		PI					
LITERATUR								
			- <del></del>					
			T					

NAME								No. 10	·Z
	5 - E	thyl-3-	me	thylh <b>e</b> ptane			STRUCTURAL F	ORMUL	A
							CH CH CH CH C	u cu c	- 1.1
Mole % Pur.	Ref.	Mole Form		C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	76	ch <sub>3</sub> ch <sub>2</sub> ch ch <sub>2</sub> c ch <sub>3</sub> c	2 <sup>H</sup> 5	′**3
	<del></del>		Ref.			Ref			Ref.
F.P. °C				dt/dP			f   to		Ī
F.P. 1009	•			°C/mm	4 0000	ا ۔ ا	g <u>•K</u> _		1
B. P. °C 760 mm	161.		2	25°C BP	4.9880 0.05160	5	h		
100	95.6		4	t <sub>e</sub>	0.0367	5	f' to		
30 10	66.6 44.6	l	4 5	30 mm	0.7246	5	g'   'K_		
1	7.6		5	ΔHm cal/g			h¹		₩
Pressure	1	$\neg \uparrow$	$\neg$	ΔHv cal/g	77 (2	ا ۔ ا	m to to		
mm 25°C	3.20	076	5	25°C 30 mm	77.62 74.20	5 5	0		
te	1175.		5	BP	63, 23	5	m' to		+
Density g/ml 20°C	0.74	13	2	te t (d, e)	61.23	5 5	n'		
dt 25	0.73	39	2		19. 27	5	0'		1
<b>4 3</b> 0	0.73		4	ΔHv/T <sub>e</sub>	81.95	5	Surface tension		
a b	-0.03		4	_e <u> </u> 179 °C		5	dynes/cm. 20°C	24.08 23.06	5
Ref. Index		-	-	d'   25 to	79.68 0.0821	5 5	30 40	22.07	5
n <sub>D</sub> 20°0		6	2		0.0821	5	Parachor [P]		_
25 30	1.41		2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	4, 120	5	20°C		
"C"	1.41		4	tc °C	328.	5	30 40		
	0.74	149	4	P <sub>c</sub> mm	15349.	5	1 . 1	424.2	5
MR (Obs.) MR (Calc.		,	2 5	PV/RT			Exp. L.l.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5	u. Dispersion	96.	2
Dielectric	2.00	)5	5	BP	0.9480	5	Flash Point °C	70.	+-
A 67 to			5	t <sub>e</sub>	0.9342 0.24	5	Fire Point		
B   214 °C	208.		5	t <sub>c</sub> AHc kcal/m	0.24	-	M Spec.		
A*  67 to	+	-	5	ΔHc Real/H	ĺ	i i	Ultra V.		
B* 189 °C		1	5	ΔFf			X-Ray Dif. Infrared		
K	1	-		Viscosity			Solubility in +		†
the Total	5		ı	centistokes 7°C			Acetone		
t <sub>x 1</sub> *(	_ i			•	1		Carbon tet. Benzene		
A'   25 to B'   67 °C			5				Ether		
č, – <u></u>	226.	1	5	B <sup>v</sup> l to			n-Heptane Ethanol		ŀ
A'* 25 to		417	5	A <sup>V</sup> I °C			Water		
B'* 67 °C	1617.6		5	(B <sup>V</sup> ) to	1		Water in		+
Ac   214 to	7.77 C 2266.		5	(A <sup>V</sup> )  °C					
Bc tc_°C	304.		5	c <sub>p</sub> liq. °K					1
Cryos. A				c <sub>p</sub> vap. °K	}				
te °C	178.92		5	c <sub>v</sub> vap.					
$T_R = 0.8$	1				L		grams/100 gran		
REFEREN		ow 2	- AP	PI 3-Lit. 4-0	alc. from det	det			15
SOURCE:			AF		uei		a 3-care, by fort	·······a	
PURIFICA	TION:		AF	<del></del>					
LITERATU		ERENC						<del></del>	
				•					

NAME	3-Ethy	1-4-r	neth	ylheptane			STRUCTURAL CH <sub>3</sub>		LA
Mole % Pur.	Ref.	Mo! For	ecul:	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	276	сн <sub>3</sub> сн <sub>2</sub> сн сн с <sub>2</sub> н <sub>5</sub>	CH <sub>2</sub> ) <sub>2</sub> CH	<sup>[</sup> 3
			Ref.	i -		Ref.			Ref.
F.P. °C				dt/dP			f to		
F.P. 100%	1		_	°C/mm			1 1		
B.P. °C	+			25°C	6. 3987	5			
760 mm	167.		2	BP	0.05218	5	h		+
100	100.8		4	t <sub>e</sub>	0.0367	5	f' to		
<b>3</b> 0	71.5		4	30 mm	0.7335	5	g'° <u>K</u>		
10	49.2		5	∆Hm cal/g		$\vdash$	h'		
1	11.7		5		<del>                                     </del>		m to		1
Pressure				ΔHv cal/g 25°C	79.37	5	n °K		
mm 25°C	2.44	54	5	30 mm	75.43	5	• !		
t <sub>e</sub>	1191.		5	BP	64.20	5	m¹ to		+-
Density	1			t <sub>e</sub> ,	62.10	5	m' to		
g/ml 20°C			2	t <sub>e</sub> (d, e)	62.01	5	;;		
d <sub>4</sub> 25	0.74		2 4	ΔHv/T <sub>e</sub>	19.25	5			
	<del></del>			d 75 to	83,83	5	Surface tension		1
a	0.76		4	e   185 °C	0.1175	5	dynes/cm. 20°C	25.40	5
Ъ	-0.03	•	4	d 25 to	81.49	5	8 30	24.34	5
Ref. Index		_		e'   75 °C	0.0848	5	40	23.31	5
<sup>n</sup> D 20°C	1.42		2	d <sub>c</sub> g/ml	0,245	5	Parachor [P]		
30	1.41		4	v ml/g	4.084	5	20°C 30		
"C"	0.74		4	tc °C	338.3	5	40		İ
	<del></del>	:50		P <sub>c</sub> mm	15750.	5		424.2	5
MR (Obs.)			2	PV/RT			Exp. L.1.%/wt.		1
MR (Calc. (nD-d/2)			5	25°C	1,0000	5	u.	ļ	-
	1.04		2	30 mm	1.0000	5	Dispersion	96.	2
Dielectric	2.02	2	5	BP	0.9470	5	Flash Point C		_
A 75 to	6.97	94	5	t <sub>e</sub>	0.9327 0.24	5	Fire Point		
B 1230 °C			5	t <sub>c</sub>	0.24		M. Spec.		
С	207.		5	ΔHc kcal/m ΔHf			Ultra V.		
A# 75 to	1.47	661	5	ΔFf			X-Ray Dif.		
B*[19 <u>5</u> °C K	- 1443.8		5	<del> </del>	<del>                                     </del>	<del> </del>	Infrared		
C				Viscosity centistokes			Solubility in +		
t <sub>k</sub> _ to	•			η			Acetone	1	
t <sub>x</sub>   °C							Carbon tet.		
A'   25 to	7,33	464	5				Benzene Ether		
B' 75 °C	1738.4		5		···	<del> </del>	n-Heptane		
C'	225.		5	B <sup>V</sup>			Ethanol	į	
A'* 25 to	1.81	669	5	A - C			Water		
B'* 75 °C	1637.9		5	(B <sup>V</sup> )  to			Water in		Д—
Ac  230 to	7.83	25	5	(A <sup>V</sup> )  °C					
Bc t <sub>c</sub> °C	2366.		5		<b></b>	<b>—</b>	1	1	
Cc	313.		5	c <sub>p</sub> liq. °K			1	1	
Cryos. A°				c <sub>p</sub> vap. °K	}				
consts. B°	<del> </del>		_	li e					
t <sub>e</sub> °C	185.70	,	5	c <sub>v</sub> vap.		<u>L</u>	L	l	
$T_R = 0.8$							' grams/100 gra		nt
REFERENC	CES: 1-I	Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			AF	ZI					
PURIFICAT	rion:		AF	PI					
LITERATU	RE REF	EREI	NCES	5:					

							No. 10	4
NAME	4-E	thyl-4-me	thylheptane	····		STRUCTURAL		
					$\dashv$	C <sub>2</sub> H <sub>5</sub>		
ļ	T	T	T		$\dashv$	СH <sub>3</sub> (СH <sub>2</sub> ) <sub>2</sub> С - (	сна)асна	,
Mole	Ref.			Molecular Weight 142.2	76	cH <sub>3</sub>	2.2	,
% Pur.		Formul		Weight 142.2	Ref			Ref.
F.P. °C	Τ	Ref.		T	Kei		1	1
F.P. 100%	+		dt/dP *C/mm			f to		1
B. P. *C			25°C BP	6.3987 0.05218	5 5	h .	1	1
760 mm 100	167.	2 4	te	0.0367	5	f' to		
30	71.5	4	30 mm	0.7335	5	g'	1	
10	49.2	5 5	ΔHm cal/g			h'		<u> </u>
Pressure	+		ΔHv cal/g		Ι_	m to		Ì
mm 25°C	2.44		25°C 30 mm	79.37 75.43	5	"   <u></u> -	1	ĺ
Den eiter	1191.	5	BP	64.20	5	m' i to		$\vdash$
Density g/ml 20°C	0.75	52 2	te te (d, e)	62.10 62.01	5	n'  °K_		1
dt 25 4 30	0.74		ΔHv/T	19.25	5	o'		
1 30	0.74		d   72 to	83.83	5	Surface tension	25 27	١,
ь	-0.0		$\frac{e}{d}$ , $\frac{186}{25}$ $\frac{\circ}{t_0}$		5	dynes/cm. 20°C	25.27 24.21	5
Ref. Index			e'   72 °C		5	40	23.18	5
n <sub>D</sub> 20°C	1.42		d <sub>c</sub> g/ml	0.244	5	Parachor [P] 20°C		
30	1.4		tc *C	4.091 338.	5	30	İ	
"C"	0.74	143 4	P <sub>c</sub> mm	15717.	5	40 Sugd.	424.2	5
MR (Obs.) MR (Calc.		3 5	PV/RT	+	-	Exp. L.1.%/wt.		†Ť
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u.	0.7	-
Dielectric	2.0	19 5	BP	0.9470	5	Dispersion Flash Point °C	97.	2
A 72 to			t <sub>e</sub>	0.9327 0.24	5	Fire Point		
B 1 228 °C	1534. 207.	5	ΔHc kcal/m	+	H	M Spec.		
A*  72 to		7661 5	ΔHf ΔFf			Ultra V. X-Ray Dif.	1	
B* 196 °C	1443.8	5	Viscosity		<b>├</b>	Infrared	L	1
С	_		centistokes			Solubility in + Acetone	l	
t <sub>k</sub> to			የ ℃			Carbon tet.		
A'   25 to	7.33	3464 5				Benzene Ether	1	
B' ∟ 72 °C	1738.4 225.	5	B <sub>v</sub>   to		-	n-Heptane		
A'* 25 to		1669 5	A <sup>V</sup> to			Ethanol Water		
B'* 72 °C		5	(BV) to	-		Water in	ļ	+
Ac   228 to			(A <sup>V</sup> ) °C	:			[	
Bc tc_°C	2364. 312.	5	c <sub>p</sub> liq. °K				]	
Cryos. A°	1		c <sub>p</sub> vap. °K				1	
consts. B°	<del>                                     </del>		c <sub>v</sub> vap.	1			1	1
t <sub>e</sub> °C T <sub>R</sub> = 0.8	185.70	5		1		L.	L	
		)ow 2 41	PI 3-Lit. 4-	Colo form 1		grams/100 gram		ıt
SOURCE:	. 1-L		PI 3-Lit. 4-	Caic. irom del	. da	ta 5-Calc. by for	mula	
PURIFICA?	TION:		PI					
LITERATU								
1								
}								

NAME	2, 2, 3-Tı	rimethy	lheptane			STRUCTURAL FOR	MULA
Mole % Pur.		iolecula ormula		Molecular Weight 142.2	76	сн <sub>3</sub> с -сн-(сн <sub>2</sub> ) <sub>3</sub> сн <sub>3</sub>	CH <sub>3</sub>
		Ref.			Ref.		Ref.
F.P. ℃			dt/dP			f to	
F.P. 100%			°C/mm			g  °K	
B. P. °C			25°C	4. 37554	5	h	
760 mm	158.	2	BP	0.05134 0.0367	5	f' to	
100	92.9	4	t <sub>e</sub>		5	g' ' ' <u>K</u>	
10	64.2 42.2	4 5	30 mm	0.7204	⊢°–	h'=	
1	5.5	5	ΔHm cal/g	<b> </b>	ļ	m to	
Pressure			ΔHv cal/g 25°C	74 71	5	n cK	
mm 25°C	3.7001		30 mm	76.71 73.55	5	•	
t <sub>e</sub>	1167.	5	BP	62.68	5	m' to	
Density g/ml 20°C	0.743		t <sub>e</sub> (d, e)	60.72	5 5	n'   °K	
	0.742	2 2	t <sub>e</sub> (a, e)	60.65	i i	0'	
d <sub>4</sub> 30	0.734	4	е	19.25	5	Surface tension	
a	0.758	4	d 65 to	80.98	5	Surface tension dynes/cm. 20°C 23.	95 5
Ъ	-0.038	4	e 175 °C to	0.1158 78.73	5	<b>8</b> 30 22.	93 5
Ref. Index			e' 65 °C	0.0808	5	40 21.	94 5
<sup>n</sup> D 20°C		2 2	d <sub>c</sub> g/ml	0,243	5	Parachor [P]	
25 30	1.414	4	v <sub>c</sub> m1/g	4.122	5	20°C 30	
"C"	0.7476		t <sub>c</sub> °C	324. 1	5	40	
MR (Obs.)		2	P <sub>c</sub> mm	15244.	5	Sugd. 424.	2 5
MR (Calc.		5	PV/RT		_	Exp. L. l. %/wt.	
(nD-d/2)	1.046	2	25°C 30 mm	1.0000	5	u. Dispersion 97.	2
Dielectric	2.008	5	BP	0.9480	5		-+-
A 65 to	6,9658	5	te	0.9345	5	Flash Point C Fire Point	
B (210 °C		5	t <sub>c</sub>	0. 24	5	M. Spec.	
С	209.	5	ΔHc kcal/m ΔHf			Ultra V.	
A* 65 to B* 185 °C		7   5	ΔFf			X-Ray Dif.	
K 105 5	- 1410.0		Viscosity			Infrared	
°	_		centistokes			Solubility in * Acetone	
t <sub>k</sub> Too			<b>n</b> °c			Carbon tet.	
A'   25 to		1 5				Benzene	
B' 65 °C		1 5			ļ	Ether n-Heptane	
C' '	227.	5	B <sup>V</sup> to C		ļ	Ethanol	
A'* 25 to		4 5		_		Water	1
B'* 65 °C		5	(B <sup>V</sup> )  to			Water in	
Ac 210 to	7.7423		(A <sup>V</sup> )  °C			]	
Bc tc °C	2224. - 301.	5	c <sub>p</sub> liq. °K				
Cryos. A°	<del></del>	+-			1		
consts. B°			c <sub>p</sub> vap. °K		1		
t <sub>e</sub> °C	175.52	5	c <sub>v</sub> vap.				
$T_R = 0.8$	1			1		grams/100 grams s	olvent
		v 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ata 5-Calc. by formula	
SOURCE:		AF					
PURIFICA'	TION:	AF					
<del></del>	RE REFER						
1							

No. 106 STRUCTURAL FORMULA NAME 2, 2, 4-Trimethylheptane сн3 сн<sub>3</sub>с - сн<sub>2</sub>сн (сн<sub>2</sub>)<sub>2</sub>сн<sub>3</sub> Molecular C<sub>10</sub>H<sub>22</sub> Ċнз Molecular ĊH<sub>3</sub> Mole Ref. Weight 142.276 % Pur Ref. Ref Ref. F.P. °C F.P. 100% dt/dP f °C/mm °K\_ g 25°C 2.6734 5 B. P. °C h ВP 0.05125 5 760 mm 147.7 2 t<sub>e</sub> 5 0.03733 f to 82.96 100 4 g' °K 30 54.52 4 30 mm 0.7113 5 10 32.86 h AHm cal/g -3,27 1 m to AHv cal/g Pressure °K n 72.51 25°C 6.4065 mm 25°C 30 mm 70, 29 5 t<sub>e</sub> 1142.1 5 5 BP 60.0 m¹ to Density g/ml 20°C te (d, e) 58.21 °K 0.7275 2 'n 58,16 5 0 0.7237  $\mathbf{d_4^t}$ 25 2 ΔHv/T<sub>e</sub> 18.93 5 30 0.71994 Surface tension d 54 76.32 5 0.7427 to 4 dynes/cm. 20°C 22.13 0.1105 <u>l 16</u>4 ь -0.0376 4 21,22 5 30 ď to 74.39 5 5 20.33 40 Ref. Index 0.0752 5  $\mathbf{n}_{\mathbf{D}}$ 20°C 1.4092 2 [P] dc Parachor g/ml 0.231 5 25 1.4070 20°C v<sub>c</sub> ml/g 4.329 5 30 1.4046 4 30 °C 309. tc 40 "C" 0.7490 4 Pç 14147. 5 5 mm Sugd. 424.2 MR (Obs.) 48.37 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 (nD-d/2) 1.0454 2 30 mm 1.0000 5 Dispersion 100. 2 Dielectric 1.986 5 BP 0.9510 Flash Point °C 0.9383 5 6.8795 A | 54 to t<sub>e</sub> 5 Fire Point 0.24 5 1434.1 M Spec. C 211. 5 AHc kcal/m Ultra V. ΔHf 54 **to** A\* 1.3879 5 X-Ray Dif. ΔFf 174°C 1345.3 Infrared Viscosity Solubility in c centistokes Acetone to Carbon tet. ۰c Benzene 25 **to** A' 7,2458 Ether В' 1635.1 <u>54 °C</u> n-Heptane  $\mathbf{B}^{\overline{\mathbf{v}}}$ 5 229. to Ethanol A I °C 25 to Water A!\* 1.7359 5 Water in B'\* 54 °C 1535.2 5 (BV) to Ac | 195 to 7,5887 5 (AV) °C Bc tc\_°C 2068.0 liq. °K Сp 292.5 Cc Cryos. Aº cp vap. °K consts. B° t<sub>e</sub> °C c<sub>v</sub> vap. 164.30 5  $T_{R} = 0.80 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 107
NAME _	2,2,5-Trin	neth	ylheptane			STRUCTURAL	FORMULA
						CH <sub>3</sub>	
ļ			T			сн <sub>3</sub> с - (сн <sub>2</sub> ) <sub>2</sub> сі	н сн,сн,
Mole % Pur.		lecul mula	ar C <sub>10</sub> H <sub>22</sub>	Molecular Veight 142.2	76		н,
70 Fur.	F01	Ref	<del></del>	veight 142.2	Ref.		Ref.
F.P. °C	1	Kei.		<b> </b>	Rei.		1
F.P. 100%		-	dt/dP °C/mm			f to g*K_	
B. P. °C			25°C	2.900	5	h i	1
760 mm	148.	2	BP t <sub>e</sub>	0.05036 0.0366	5	f' to	
100 30	84.2 56.0	4	30 mm	0.7055	5	g'	
10	34.5	5	ΔHm cal/g	0.1033		h'	l
1	-1.5	5	ΔHv cal/g		-	m to	
Pressure mm 25°C	5.7968	5	25°C	73.88	5	n   <u>*K</u>	
t <sub>e</sub>	1144.	5	30 mm BP	71.52 61.21	5 5	LL	
Density	-	1	t.	59.42	5	m' to	1
g/ml 20°C	0.726	2	t <sub>e</sub> (d, e)	59.38	5	n'	1
dt 25 4 30	0.722 0.718	2	ΔHv/T <sub>e</sub>	19.32	5		<del>  </del>
a	0.742	4	d 56 to	77.79	5	Surface tension dynes/cm. 20°C	21.95 5
ь	-0.038	4	e 164 °C to	0.1120 75.79	5	<b>8</b> 30	20.99 5
Ref. Index	1 400	١, ١	e' 56 °C	0.0763	5	40	20.07 5
n <sub>D</sub> 20°C	1.409	2 2	d <sub>c</sub> g/ml	0.239	5	Parachor [P] 20°C	
30	1.404	4	vc ml/g tc °C	4.176 308.	5	30	
"C"	0.7502	4	P <sub>c</sub> mm	14639.	5	40 Sugd.	424.2 5
MR (Obs.)	48.5	2	PV/RT			Exp. L.1.%/wt.	
MR (Calc.) (nD-d/2)	48.38 1.046	5 2	25°C	1.0000	5	u.	
Dielectric	1.985	5	30 mm BP	1.0000 0.9520	5	Dispersion	100. 2
A   56 to	6.9532	5	t	0.9395	5	Flash Point C Fire Point	
B (192 °C		5	tc	0.24	5	M. Spec.	<del> </del>
C	211.	5	ΔHc kcal/m ΔHf	1		Ultra V.	
A* 56 to B* 174 °C	1.45980	5	ΔFf			X-Ray Dif. Infrared	
к ———			Viscosity			Solubility in +	+
t <sub>k</sub> to			centistokes り °C			Acetone	
t <sub>x</sub> °C			7			Carbon tet. Benzene	
A'   25 to	7.32254	5				Ether	
B' _56 °C	1665.3 229.	5	B <sup>v</sup> to			n-Heptane Ethanol	
A₩ 25 to	1,81153	5	AV I °C			Water	
B'* 56 °C	1565.1	5	(B <sup>V</sup> )  to			Water in	
Ac 192 to	7.6549	5	(A <sup>V</sup> )  °C				
Bc tc °C	2085. 290.	5	c <sub>p</sub> liq. °K				
Cryos, A°							
consts. B°			P				
t <sub>e</sub> °C	164.36	5	c vap.				
$T_{\mathbf{R}} = 0.80$	T <sub>c</sub>					† grams/100 gra	ms solvent
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc. by for	:mula
SOURCE:		AI	PI				
PURIFICAT	ION:	AI	PI				
LITERATU	RE REFERE	NCES	5:				
Į.							
1							

							No. 10	8
NAME	2, 2,	6-Trimet	hylheptane			STRUCTURAL		
			, ,					-
						CH <sub>3</sub>	CII CII	
Mole	Ref.	Molecul		Molecular		сн <sub>3</sub> с - (сн <sub>2</sub> ) <sub>3</sub>	LH CH3	
% Pur.		Formul		Weight 142.2	76	CH <sub>3</sub>	Ċн <sub>3</sub>	
		Ref.			Ref			Ref.
F, P. *C	T		dt/dP			f to		T
F.P. 100%			°C/mm			g LK	ĺ	
B. P. *C			25°C BP	2.9204 0.05039	5	h		İ
760 mm 100	148.2	2 4	te	0.0366	5	f' to		
30	56.2	4	30 mm	0.7059	5	g'   ' <u>•</u> K_		
10	34.6	5	ΔHm cal/g	<del>                                     </del>		h'		
1	-1.3	1 5	ΔHv cal/g	+	$\vdash$	m to		
Pressure mm 25°C	5.75	29   5	25°C	73.92	5	n   <u>*K</u>		
t <sub>e</sub>	1144.	- / 5	30 mm	71.54	5	0		
Density	<del>                                     </del>		BP te	61.22 59.43	5	m'   to		
g/ml 20°C	0.71		te (d, e)	59.39	5	n' °K	-	1
dt 25 4 30	0.71		AHv/Te	19.31	5	<u> </u>	ļ	1_
a 30	0.71		d   56 to		5	Surface tension	21 17	-
b	-0.03				5	dynes/cm. 20°C	21.17	5
Ref. Index	<u> </u>		d'   25 to		5	40	19.39	5
n <sub>D</sub> 20°C	1.40		d <sub>c</sub> g/ml	0, 237	5	Parachor [P]	Ī —	
25 30	1.40		V_ml/g	4, 214	5	20°C 30		
"C"	0.75	-	tc °C	308.	5	40		l
MR (Obs.)	+		P <sub>c</sub> mm	14510.	5	Sugd.	424.2	5
MR (Calc.)	48.56 48.38		PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	98.	2
Dielectric	1.97	6 5	BP	0.9520	5	Flash Point °C	70.	+-
A 56 to			t <sub>e</sub>	0.9395 0.24	5	Fire Point		
B <u>[182 °C</u> C	211.	5 5	t <sub>c</sub> ΔHc kcal/m	+	ļ -	M Spec.		1
A*  56 to	1.45		∆Hf			Ultra V.		
B* 175 °C		5	ΔFf			X-Ray Dif. Infrared		
к — — —	` <u> </u>		Viscosity			Solubility in +	<del> </del>	+
,	-		rentistokes °C	ļ		Acetone		}
tx C			7			Carbon tet. Benzene		1
A'   25 to	7.32					Ether		
B' ∟ 56 °C	1665.6 229.	5 5	B <sup>V</sup> to	<b>+</b>	$\vdash$	n-Heptane		
A'* 25 to	1.81		B' to			Ethanol Water		
B'* 56 °C		5	(B <sup>V</sup> ) to	-[		Water in		
Ac   182 to	7.65		(A <sup>V</sup> ) °C	1				1
Bc t <sub>c</sub> °C	2083.	5	c <sub>p</sub> liq. °K					
	289.	-   3		ŀ				
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				1	
t <sub>e</sub> °C	164.59	5	c <sub>v</sub> vap.			L <u>.</u>	L	<u>L</u>
$T_R = 0.8$						grams/100 gra		ıt
REFERENC	ES: 1-D			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		Al	<del></del>					
PURIFICAT		Al						
LITERATU	RE REFI	ERENCES	:					

r							No. 109
NAME	2, 3, 3-Trin	nethy	rlheptane		I	STRUCTURAL	FORMULA
		·				сн <sub>3</sub>	
			Т				H <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>
Mole		lecul		Molecular	_,	сн3сн3	2 3 3
% Pur.	For	mula		Weight 142.2			
		Ref.			Ref.		Ref
F.P. °C F.P. 100%	l	ļ	dt/dP		1	f to	
	<del>                                     </del>	├	°C/mm 25°C	4.76914	5	g <u>°K</u>	
B. P. °C 760 mm	160.	2	BP	0.05150	5	h	ļ
100	94.7	4	t <sub>e</sub>	0.0367	5	f' to	
30 10	65.8 43.8	5	30 mm	0.7231	5_	h'	
1	6.9	5	ΔHm cal/g	<u> </u>	↓	m l to	
Pressure			ΔHv cal/g 25°C	77.33	5	n °K	
mm 25°C	3.3674	5	30 mm	74.00	5	0	
Deneite.	1172.	)	BP	63.07	5	m' to	
Density g/ml 20°C	0.7488	2	t <sub>e</sub> (d, e)	61.08	5	n'   °K	
dt 25	0.7448	2	ΔHv/T <sub>e</sub>	19.27	5	•'	
	0,7408	4	d 66 to	81.64	5	Surface tension	
a b	0.7648	4	_e <u>  178 °C</u>	0.1161	5	dynes/cm. 20°C	24.84 5
Ref. Index	-0.0380	4	d 25 to	79.38	5	30 40	23.79 5 22.78 5
n <sub>D</sub> 20°C	1.4202	2		0.0817	5	Parachor [P]	
- 25	1.4178	2	d g/ml	0.245 4.075	5	20°C	
30	1,4153	4	t <sub>c</sub> °C	328.	5	30 40	
"C"	0.7461	4	P <sub>c</sub> mm	15519.	5		424.2 5
MR (Obs.) MR (Calc.)	48.10 48.38	2 5	PV/RT		T	Exp. L. 1. %/wt.	
(nD-d/2)	1.0458	2	25°C 30 mm	1.0000	5	u.	
Dielectric	2.017	5	BP	0.9480	5	Dispersion	96. 2
A 66 to	6,9710	5	te	0.9343	5	Flash Point C Fire Point	
B 1 214 °C		5	t <sup>e</sup> c	0.24	5	M. Spec.	
<del></del>	209.	5	ΔHc kcal/m ΔHf	!		Ultra V.	
A* 66 to B* 188 °C	1.47305 1418.2	5	ΔFf	1		X-Ray Dif. Infrared	
к			Viscosity			Solubility in +	
			centistokes 7°C	İ	Ì	Acetone	
t <sub>x</sub> °C			<sup>7</sup>			Carbon tet. Benzene	
A'   25 to	7.33132	5				Ether	
B' 1_ 66 °C	1711.9	5 5	B <sup>V</sup> to	·	<u> </u>	n-Heptane	
A'* 25 to	<del>                                     </del>		B' to		ĺ	Ethanol Water	
B'* 66 °C	1.81589	5	(B <sup>v</sup> )  to		1	Water in	
Ac 214 to	7.7679	5	(A <sup>V</sup> )  °C				
Bc tc °C	2261.	5	c <sub>p</sub> liq. °K		T		
	304.	5	1	İ			
Cryos, A° consts, B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	177.78	5	c <sub>v</sub> vap.	1			
$T_R = 0.81$			u	L	Щ.	grams/100 gra	me soleent
	ES: 1-Dow	2_4	PI 3-1.it 4.4	Calc from de	t de	ita 5-Calc. by for	
SOURCE:	,, I-DUW	AF		Care, Irom de	. ua	ica J-Caic, by for	******
PURIFICAT	TON:	AF					
LIIEKATU.	RE REFERE	NCE	<b>:</b>				

NAME	2, 3, 4	-Trimet	hylheptane			STRUCTURAL	No. 11 FORMUL	
		-				CH <sub>3</sub>		
<u></u>						сн <sub>3</sub> сн сн сн	(CH_)_CH	I.
Mole	Ref.	Molecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular	_,	ċн <sub>3</sub> ċн <sub>3</sub>		-3
% Pur.			10-22	Weight 142.2		0113		<b>5</b> (
		Ref.		1	Ref.		Г	Ref
F.P. °C F.P. 100%		-	dt/dP *C/mm			f to		
B. P. °C			25°C	5.4068	5	g <u>*K</u>		
760 mm	163.	2	BP	0.05179	5			+
100 30	97.3 68.3	4 4	t <sub>e</sub>	0.0367	5	f' to g'  *K_		
10	46.1	5	30 mm	0.7276	"	h'		1
1	9.0	5	ΔHm cal/g	ļ	_			+
Pressure			ΔHv cal/g 25°C	78,20	5	m to		
mm 25°C	2.937 1181.	2   5   5	30 mm	74.61	5	•		
t <sub>e</sub> Density	1101.		BP	63.58	5	m'   to		T
g/ml 20°C	0.751	2	te t (d, e)	61.55	5	n'  °K_		
at 25	0.747	2	ΔHv/T	19.27	5	٥'		
	0.743		d   68 to		5	Surface tension		
a b	0.767 -0.0 <sub>3</sub> 8		e   181 °C		5	dynes/cm. 20°C	25.13 24.08	5
Ref. Index	-0.030	-+	d'   25 to		5	30 40	23.05	5
n <sub>D</sub> 20°C	1.421	2	<del></del>	<del></del>	-	Parachor [P]		+
45	1.418		d g/ml vc ml/g tc °C	0.245 4.078	5	20°C		
30	1.416		tc °C	333.	5	30 40		
"C"	0.745		P <sub>c</sub> mm	15637.	5	Sugd.	424.2	5
MR (Obs.) MR (Calc.)	48.0 48.38	2 5	PV/RT	<u> </u>		Exp. L.1.%/wt.		
(nD-d/2)	1.045		25°C 30 mm	1.0000	5	u.	0.7	2
Dielectric	2.019	5	BP	0.9480	5	Dispersion Flash Point °C	97.	+-
A 68 to	6.974	6 5	te	0.9341	5	Fire Point		
B 1_224_°C		5	t <sub>c</sub>	0.24	3	M Spec.		t
A*  68 to	208.	5 82 5	ΔHc kcal/m	1		Ultra V.		
B* 191 °C	1.473 1428.9	°2   .5	ΔFf			X-Ray Dif. Infrared		
к ———			Viscosity			Solubility in +		╁
c			centistokes °C			Acetone		
t <sub>x</sub> to			η •c	1		Carbon tet. Benzene		1
A'   25 to	7.332					Ether		
B' ∟ 68 °C	1723.3	5	B <sup>V</sup> to	<u> </u>	$\vdash$	n-Heptane		
A'* 25 to	226.	23 5	B to ℃			Ethanol Water		
	1.816 1622.8	23   5	(B <sup>V</sup> ) to	1		Water in		
Ac   224 to	7.795	5 5	(A <sup>V</sup> ) °C	ł				
Bc t °C	2305.	5	c <sub>p</sub> liq. •K					
Cc	308.	5	ł					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	181.20	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.82$	Tc		L	L	L	grams/100 gran	ne enlver	
REFERENCE		w 2-AF	PI 3-Lit. 4-0	Calc. from det	. dai	ta 5-Calc, by for		
SOURCE:		AI						
PURIFICATI	ON:	Al	PI					
LITERATUR				<del></del>	-			

No. 111 2, 3, 5-Trimethylheptane NAME STRUCTURAL FORMULA CH<sub>3</sub> ch,ch ch ch,ch ch,ch, Molecular C<sub>10</sub>H<sub>22</sub> Mole Ref. Molecular Weight 142,276 Ċнз ĊH, % Pur. Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm ٩Ķ g 25°C 3.8869 5 B. P. °C h ВP 0.05219 5 760 mm 157. 91.05 2 ſ١ 0.03744 5 to 100 4 g' °K 30 62.05 4 30 mm 0.7253 5 10 39.97 5 h' ∆Hm cal/g 5 1 3.12 m to AHv cal/g Pressure n °К 25°C 75.08 mm 25°C 4.2557 5 5 o 30 mm 72.14 5 te 1164.4 BP 61.37 5 m' Density to te te (d, e) 59.43 5 n' g/ml 20°C °K 0.741 2 5 59.36 ٥' dt4 25 0.737 2 AHV/Te 18.88 5 30 0.733 4 Surface tension 62 to 79.19 d 5 0.757 4 23.82 5 dynes/cm. 20°C 175 0.1135 5 °C -0.038 4 30 22.81 5 ăח 25 to 77.06 40 21.82 5 Ref. Index e' 62 °C 0.0793 5 20°C 1.416 2 [P] <sup>n</sup>D d g/ml vc ml/g t °C Parachor 0.231 5 25 1.414 20°C 5 4.327 30 1.411 4 30  $^{\mathbf{t}}\mathbf{c}$ 323 5 40 "C" 0.7469 4  $P_c$  mm 14494. 5 5 Sugd. 424.2 MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. 48.38 MR (Calc.) 5 25°C 1.0000 5 (nD-d/2)u. 1.046 2 30 mm 1.0000 5 2 Dispersion 97. Dielectric 2,005 5 0.9480 RP Flash Point C 0.9343 5 A 62 to 6.89044 5 5 Fire Point 0.24 1468.2 B | 210 °C M. Spec. Ultra V. С 209.2 5 AHc kcal/m ΔHf A\* 62 to 1.39506 1379.2 5 X-Ray Dif. ΔFf B\* 185 °C Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet.  $\mathbf{t_{x}}$ °C Benzene 25 to 7.24970 Ether B١ <u>62 °C</u> 1669.6 5 n-Heptane B<sub>v</sub> C' 227.2 5 to Ethanol °C Water 25 to A'\* 1.73638 5 5 62 °C Water in B'\* (B<sup>V</sup>) 1569.4 to Ac 210 to 7.66447 5 (AV) °C 2190.1 Bc tc °C c<sub>p</sub> liq. °K 302.1 5 Cryos. A° c<sub>p</sub> vap. °K consts. B° c, vap. t<sub>e</sub> °C 174.74 5  $T_{R} = 0.81 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** LITERATURE REFERENCES:

								No. 11	2
NAME	2,3,	6-Tri	imet	hylheptane			STRUCTURAL	FORMUL	A.
				· · · · · · · · · · · · · · · · · · ·		$\neg \uparrow$	CH <sub>3</sub>		
				r			снзсн сн (сн	.)_CH-CH	i.
Mole	Ref.	Mol			Molecular	ا ر	ċн <sub>3</sub>	CH <sub>3</sub>	3
% Pur.			mul	10-22	Weight 142.2	-	53	33	Ref.
	1		Ref.		т	Ref	<del></del>	r	Kei.
F.P. °C F.P. 100%		$\dashv$		dt/dP •C/mm	1		f to		İ
B. P. *C	<b>-</b>	-		25°C	3.6863	5	h		İ
760 mm	155.7		2	BP t <sub>e</sub>	0.05206 0.03743		f' to	<u> </u>	+-
100 30	89.92		4	30 mm	0,7234	5	g'   'K_		
10	38.97		5 5	AHm cal/g			h'		
1	2,22		-	ΔHv cal/g	<u> </u>	Н	m   to		
Pressure mm 25°C	4.50	93	5	25°C	74.71	5	n   <u>*K</u> -	}	
t <sub>e</sub>	1161.4		5	30 mm BP	71.88 61.18	5	ļ		₩
Density g/ml 20°C	0.73	45	2	l t.	59.26	5	m'   to		
	0.73		2	te (d, e)	59.19	5	0'	1	1
dt 25 4 30	0.72		4	ΔHv/T <sub>e</sub>	18.88	5	Surface tension		T
a b	0.75		4	d   60 to		5	dynes/cm. 20°C	23.00	5
Ref. Index	-0.03	**	4	d'   25 to	76, 68	5	30 40	22.01 21.05	5
n <sub>D</sub> 20°C		25	2	1 00 -	0.0787	5	Parachor [P]		†
25 30	1.41		2 4	d g/ml vc ml/g tc °C	4.368	5	20°C 30	İ	1
"C"	0.74	$\rightarrow$	4	1 -	320.	5	40	1	
MR (Obs.)			2	P <sub>c</sub> mm	14287.	5	Sugd.	424.2	5
MR (Calc.	) 48.38	İ	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.04	-	2	30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric			5	BP	0.9485 0.9349	5 5	Flash Point °C		$\vdash$
A 60 to B   207 °C		867	5	te t <sub>c</sub>	0.24	5	Fire Point		↓
c ——	209.4		5	ΔHc kcal/m			M Spec. Ultra V.		
A*  60 to		367	5	ΔHf ΔFf			X-Ray Dif.		
B* 183 °C	- 13/4.3		,	Viscosity	<del> </del>		Infrared Solubility in +		┿
t	_	1		centistokes			Solubility in + Acetone		
t <sub>k</sub> to		- 1		ፖ °C			Carbon tet.	ĺ	
A'   25 to		889	5				Benzene Ether		
B' ∟ 60 °C	227.4	i	5	B <sup>V</sup>   to	<del>  -</del>	$\vdash$	n-Heptane		ļ
A'* 25 to	<del></del>	605	5	A <sup>V</sup> C	1		Ethanol Water		
	1564.5		5	(BV) to	1		Water in		ـــ
Ac   207 to	7.64	810	5	(A <sup>V</sup> ) °C				l	
Bc tc_C	2165.1		5	cp liq. °K					
Cryos, A		-	Ť	c <sub>p</sub> vap. *K					
consts. B°				р -					
t <sub>e</sub> °C	173.29	[	5	c <sub>v</sub> vap.					
$T_{R} = 0.8$							f grams/100 grai	ns solven	it
REFEREN	CES: 1-D	ow 2	-AF		Calc. from det	da t	ta 5-Calc, by for	mula	
SOURCE:			Al	· · · · · · · · · · · · · · · · · · ·					
PURIFICA'			A1						
LITERATU	RE REFI	EREN	CES	:					

NAME	2,4,4	4-Tri	meth	ylheptane			STRUCTURAL FORMULA			
Mole % Pur.	Ref		lecul rmula		Molecular Weight 142.2	276	сн <sub>3</sub> сн-сн <sub>2</sub> с сн <sub>3</sub> сн		H <sub>3</sub>	
			Ref.			Ref.		***************************************	Ref.	
F. P. °C				dt/dP			f to		-	
F.P. 100	6			°C/mm			g  °E			
B.P. °C				25°C BP	3.3037 0.05179	5 5	h	ŀ	1	
760 mm 100	153. 87.5	6	2 4	te	0.03743	5	f' to		T	
30	58.8		4	30 mm	0.7194	5	g' ' <u>°F</u>			
10	36.9		5	ΔHm cal/g		+	h'		1	
<u> </u>	0.3	555	5	ΔHv cal/g		+-1	m to		T	
Pressure mm 25°C	5.0	832	5	25°C	73.95	5	n   P	<u> </u>	İ	
t <sub>e</sub>	1108.	032	5	30 mm	71.33	5	Li			
Density	<b>†</b>			BP t <sub>e</sub>	60.68	5	m' to			
g/ml 20°0			2	te (d, e)	58.73	5	n'			
d <sub>4</sub> 25	0.7		2 4	AHv/Te	18.85	5				
·	0.7		4	d 59 to	77.99	5	Surface tension	22.04	_	
a b	-0.0		4	_e_ <u> _168</u>	0.1133	5	dynes/cm. 20°C	23.06	5	
Ref. Inde	,		$\vdash$	d'   20 to   e'   59 °C	75.89 0.0775	5	40	20.18	5	
n <sub>D</sub> 20°			2		0, 208	5	Parachor [P]		T	
25 30	1.4		2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	4.804	5	20°C			
"C"	0.7		4	tc °C	307.	5	30 40			
MR (Obs.			2	P <sub>c</sub> mm	12706.	5	Sugd	. 424.2	5	
MR (Calc.			5	PV/RT			Exp. L.1.%/wt.			
(nD-d/2)	1.0	46	2	25°C 30 mm	1.0000	5 5	u. Dispersion	97.	2	
Dielectric	1.9	94	5	BP	0.9485	5	Flash Point C	<del>  /</del>	+-	
A 59 to		8497	5	t <sub>e</sub>	0.9352 0.24	5	Fire Point			
B 1197 °C			5	tc ΔHc kcal/m	0.24	۲	M. Spec.	1	+	
A*  59 to	209.9	9259	5	ΔHc Real/III			Ultra V.			
B* 178 °C			5	ΔFf			X-Ray Dif. Infrared			
к ——	_			Viscosity			Solubility in +			
t <sub>k</sub> — to	-			centistokes り °C			Acetone	1		
t <sub>x</sub> °C	;			,			Carbon tet. Benzene			
A'   20 to	7.2	4720	5 5				Ether			
B'   59 °C	1654.4 228.		5	PA	<b>†</b>	+	n-Heptane	1		
A!* 20 to	<del></del>	2524	5	B <sup>V</sup>	1		Ethanol Water			
B'* 59 °C		3534	5	(B <sup>V</sup> )  to	-		Water in			
Ac  197 to	7.5	8044	5	(A <sup>V</sup> )  °C					1	
Bc tc °C	2068.0		5	c <sub>p</sub> liq. °K	T	$\vdash$				
Ce	281.1		5						-	
Cryos, A' consts, B'				c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	167.8	1	5	c <sub>v</sub> vap.						
$T_R = 0.8$			ب	<u> </u>	1		grams/100 gr	ame colsec		
REFEREN		Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by fo		111	
SOURCE:			AF		-3.0, 110m de	ua	5 Care. by 10			
PURIFICA	TION:		AF					·		
LITERATI		ERF								
				•						

						No.	o. 114
NAME	2,4,5-Tri	meth	ylheptane			STRUCTURAL FOR	MULA
						CH <sub>3</sub>	
			T			CH3CH CH2CH CH C	H <sub>2</sub> CH <sub>3</sub>
Mole		lecul		Molecular Weight 142.2	76	сн <sub>3</sub> сн <sub>3</sub>	- ,
% Pur.	1 1 10	rmul		weight 142.2	Ref	<del>r</del>	Ref
F, P. *C	Γ	Ref.		T	Kei		- Kei
F.P. 100%	<del> </del>	-	dt/dP •C/mm			f to g °K	
B. P. *C	<b>†</b>		25°C	3.8869	5	h	
760 mm 100	157. 91.05	2 4	BP t <sub>e</sub>	0.05219 0.0374	5	f' to	
30	62.05	4	30 mm	0.7253	5	g'   ' <u>°</u> K_	
10 1	39.97 3.12	5	ΔHm cal/g			h!	
Pressure	31.2	-	ΔHv cal/g			m to to	
mm 25°C	4.2557	5	25°C 30 mm	75.08 72.14	5	n   ' *K	
t <sub>e</sub>	1165.1	5	BP	61.40	5	l — !	
Density g/ml 20°C	0.741	2	t <sub>e</sub> (d, e)	59.48 59.39	5	m'   to n'  °K_	
at 25	0.737	2	ΔHv/T <sub>e</sub>	18.89	5	o'	
	0.733	4	d   62 to	79.16	5	Surface tension	
a b	0.757 -0.0 <sub>3</sub> 8	4	e   175 °C	0.1131	5		.82 5 .81 5
Ref. Index	1 3	Ť	a'   20 to		5		.82 5
n <sub>D</sub> 20°C	1.4160	2		0,231	5	Parachor [P]	
25 30	1.4137	2	d g/ml vc ml/g tc °C	4.325	5	20°C	
"C"	0.7469	4	t <sub>c</sub> ·C	323.	5	40	
MR (Obs.)	48.2	2	P <sub>c</sub> mm	14500.	5	Sugd. 424	. 2 5
MR (Calc.) (nD-d/2)		5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.	
Dielectric	1.046 2.005	5	30 mm BP	1.0000	5	Dispersion 97	. 2
A 62 to	6,89044	5	t <sub>e</sub>	0.9485 0.9348	5	Flash Point °C	
B [210 °C	1468.2	5	t <sub>c</sub>	0.24	5	Fire Point	
С	209.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.	
A*   62 to B*   185 °C	1.39418	5	ΔFf		1	X-Ray Dif. Infrared	
к — — —			Viscosity			Solubility in +	
t <sub>k</sub>	1		centistokes 7°C	1		Acetone	
tx C			"			Carbon tet. Benzene	
A'   20 to	7.24970	5		}		Ether	
B' ∟62 °C	1669.6	5	B <sup>V</sup>   to	<del></del>	1	n-Heptane Ethanol	
A** 20 to	1,73638	5	ĀV   °C	1		Water	
B'* 62 °C		5	(B <sup>V</sup> ) to			Water in	
Ac   210 to	7.6647 2190.	5	(A <sup>V</sup> )  °C				
Bc tc_°C	302.	5	c <sub>p</sub> liq. °K				
Cryos. Aº			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	174 74	5	c <sub>v</sub> vap.				
$T_{R} = 0.81$	T <sub>C</sub>	٠		l	L	+ (100	
REFERENC		2-AI	PI 3-Lit. 4-0	Calc from det	t de	grams/100 grams sta 5-Calc. by formula	
SOURCE:		Al				5 Cure. by formula	•
PURIFICAT	ION:	Al					
	E REFERE						

NAME	2,4,6-Trin	nethy	lheptane		STRUCTURAL FORMULA				
Mole % Pur.		lecul:	ar C <sub>10</sub> H <sub>22</sub>	Molecular Weight 142.2	76	сн <sub>3</sub> сн	сн <sub>2</sub> сн с		<sup>1</sup> 3
-		Ref.			Ref.	Г			Ref.
F.P. °C			dt/dP			f	1		
F.P. 100%		_	°C/mm	i	1 1	gl	to K		
B. P. °C			25°C	2.3773	5	I .			
760 mm	144.8	2	BP	0.05099	5	h			+-
100	80.41	4	t <sub>e</sub>	0.0373	5	f'	to		
30	52,13	4	<b>3</b> 0 mm	0.7071	5	g'	: <u>°K</u>		
10	30.6	5	ΔHm cal/g		$\sqcap$	h'			
	-5.3	-	ΔHv cal/g			m	to		
Pressure mm 25°C	7 2072	5	25°C	71.69	5	n	<u>•</u> K		
t <sub>e</sub>	7.2873 1133.7	5	30 mm	69.68	5	l ° ¦			
Density		-	BP	59.47	5	m'	to		
g/ml 20°C	0.7225	2	te te (d, e)	57.73 57.68	5	n'	<u>•</u> K		
at 25	0.7184	2	ΔHv/T	ŀ	i I	o'			1
4 30	0.7143	4		18.92	5	Surfac	e tension		
a	0.7389	4	d 52 to	75.43	5		cm. 20°C	21.53	5
Ъ	-0.0382	4	_e_ 161 °C 15 to	0.1102 73.53	5	8	30	20.57	5
Ref. Index			e' 52 °C	0.0739	5		40	19.63	5
n <sub>D</sub> 20°C	1.4071	2	d <sub>c</sub> g/ml	0.226	5	Parac			1
25 30	1.4046 1.4022	2 4	v ml/g	4.418	5		20°C		
			vc ml/g tc °C	303.	5		30 40		
"C"	0.7505	4	P <sub>c</sub> mm	13721.	5			424.2	5
MR (Obs.)	48.48	2	PV/RT		$\vdash$	Exp	L. 1. %/wt.		1
MR (Calc.)   (nD-d/2)	48.38 1.0459	5 2	25°C	1.0000	5	DAP.	u.		
		-	30 mm	1.0000	5	Disper		97.	2
Dielectric	1.980	5	BP	0.9510 0.9385	5	Flash	Point °C		1
A 52 to B 1188 °C	6.87377 1 <b>422</b> .6	5	t <sub>e</sub>	0.24	5	Fire F	Point		1
c   1100 U	211.5	5	ΔHc kcal/m	+		M. Sp			
A* 52 to	1.38504	5	ΔHf	1		Ultra			1
B* 171 °C	1334.4	5	ΔFf			X-Ray Infrar			
к ———			Viscosity			<u> </u>			+
°			centistokes	l	i 1	Acet	lity in <sup>†</sup>		
t <sub>k</sub> to			ŋ °C				on tet.		-
t <sub>x</sub> °C	7.2422/	-				Benz			
B' 52 °C	7.24226 1623.5	5				Ethe	r ptane		-
c' '-' = =	229.	5	B <sup>V</sup> to C			Ethan			
A** 15 to	1.73344	5	AVI °C	1	1 1	Wate	r		1
B'* 52 °C	1523.7	5	(B <sup>V</sup> )  to	1		Wate	r in		
Ac 188 to	7.55659	5	(A <sup>V</sup> )  °C						
Bc t <sub>c</sub> °C	2021.3	5			$\vdash$				
Cc — —	288. 3	5	р.						
Cryos, A° consts. B°			c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	161.00	5	c vap.						
$T_R = 0.80$	T <sub>c</sub>			•		+ gran	ns/100 gra	ms solve	nt
	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	-			
SOURCE:		AF							
	ION:	AF							
LITERATUR	E REFERE								

								<b>N</b> o. 11	6
NAME	2,5,5	-Tri	m etl	hylheptane			STRUCTURAL		A.
							CI	₹ <sub>3</sub>	
L				1		$\neg$	сн <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub> с	CH <sub>2</sub> CH	3
Mole % Pur.	Ref.		lecul rmul		Molecular Weight 142.2	76	Ċн <sub>3</sub> сі	<sup>1</sup> 3	
70.1.0.1.		-	Ref.			Ref	I		Ref.
F,P. ℃	T			dt/dP			f   to	T	1
F.P. 100%				*C/mm	1 .		f to		
B. P. *C				25°C BP	3.2598 0.05184	5 5	h		
760 mm 100	152.80 87.3	)	2 4	te	0.0375	5	f' to		
30	58.5		4	30 mm	0.7195	5	g'   ' <u>*</u> K_		
10 1	36.6		5	ΔHm cal/g			h'		
Pressure	1 0.1		-	AHv cal/g			m to	İ	
mm 25°C	5.16	521	5	25°C 30 mm	73.80	5	n •K_	1	
t <sub>e</sub>	1153.		5	BP	71.21 60.60	5	<del></del>	<del> </del>	┼
Density g/ml 20°C	0.72	40	,	te	58.74	5	m'   to		
	0.73		2 2	t (d, e)	58.67	5	0,		1
d <sup>t</sup> 25 4 30	0.72	88	4	ΔHv/T <sub>e</sub>	18.86	5	Surface tension	<u> </u>	$\vdash$
	0.75		4	d   59 to	77.80 C 0.1125	5	dynes/cm. 20°C	23.29	5
b D. C. To b	-0.03	80	4	a'   15 to	75.74	5	30 40	22.29	5
Ref. Index		36	2	e' j 59 °	<del></del>	5	Parachor [P]		+-
45	1.41	12	2	d g/ml	0.230 4.351	5	20°C		
30	1.40		4	t <sub>c</sub> m1/g	317.	5	30 40		
"C"	0.74		4	P <sub>c</sub> mm	14258.	5		424.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT		_	Exp. L.1.%/wt.		
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric				BP	0.9485	5	Flash Point °C	<del> </del>	<u>├</u> -
A 59 to		94	5	te t <sub>c</sub>	0.9351 0.24	5	Fire Point		
B (205 °C	1451. 210.		5 5	ΔHc kcal/m	+	-	M Spec.		
A*  59 to	1.38	718	5	ΔHf			Ultra V. X-Ray Dif.		
B* 180 °C			5	ΔFf			Infrared		ĺ
K — — –				Viscosity centistokes			Solubility in +		
t <sub>k</sub>   to				7 °C	;		Acetone Carbon tet.		l
t⊊   •C							Benzene		
A'   10 to B' _ 59 °C		154	5				Ether n-Heptane	ĺ	
	228.		5	B <sup>V</sup> to			Ethanol		
A!# 20 to		983	5	A <sup>V</sup>	<u>-</u>		Water Water in		
B'* 59 °C	<del></del>	27	5	(B <sup>V</sup> ) to	1				1
Bc tc °C	2130.	.21	5	(A <sup>V</sup> )  °C					
Cc	298.		5	c <sub>p</sub> liq. ∘k					1
Cryos. A° consts. B°				c <sub>p</sub> vap. °K	:				1
t <sub>e</sub> °C	170,01		5	c <sub>w</sub> vap.					
$T_R = 0.8$	1 T <sub>c</sub>			4		J	grams/100 gran	ns solven	t
REFEREN		ow	2-AI	PI 3-Lit. 4-	Calc. from det	t. da			
SOURCE:				PI					
PURIFICAT	rion:		A	PI					
LITERATU	RE REFI	EREN	CES	<b>3:</b>					

· · · · · · · · · · · · · · · · · · ·					——-		No. 1	17		
NAME _	3,3,4-T	rimet	hylheptane			STRUCTURAL FORMULA				
						CH <sub>3</sub>				
I				***************************************	$\dashv$	сн <sub>3</sub> сн <sub>2</sub> с сн	CH-)-CH	í_		
Mole	Ref. Me	lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular		с́н <sub>3</sub> с́н <sub>3</sub> с	2, 2	3		
% Pur.	Fo	rmul	a 010122 1	Weight 142.2	76	01130113				
	-	Ref.			Ref.			Ref.		
F, P, °C			dt/dP			f to				
F.P. 100%			°C/mm	5 17/0	_	g '° <u>K</u>				
B.P. °C		1	25°C BP	5.1760 0.05288	5	h				
760 mm 100	164. 97.1	2 4	t <sub>e</sub>	0.0374	5	f' to				
30	67.7	4	30 mm	0.7358	5	g' <u>°K</u>		1		
10	45.3	5	AHm cal/g	<u> </u>	†	h'				
1	7.9	5	ΔHv cal/g		+-	m to				
Pressure mm 25°C	3,1136	5	25°C	77.06	5	n   <u>*K</u>				
t <sub>e</sub>	1184.	5	30 mm	73.54 62.56	5					
Density		†	BP t	60.51	5	m' to				
g/ml 20°C		2	t <sub>e</sub> (d, e)	60, 42	5	n'   <u>•K</u> _				
d <sub>4</sub> 25	0.753	2	ΔHv/T <sub>e</sub>	18.88	5					
	0.749	4	d 68 to	81,27	5	Surface tension		1_		
a b	0.773 -0.0 <sub>3</sub> 8	4	e   183 °C	0.1141	5	dynes/cm. 20°C	25.95 24.87	5		
Ref. Index		÷	d' 20 to	79.12 0.0824	5	40	23.82	5		
n <sub>D</sub> 20°C		2		<del></del>	<del> </del>	Parachor [P]	1	+-		
45	1.422	2	d g/ml vc ml/g	0.236 4.246	5	20°C				
30	1.420	4	t <sub>c</sub> °C	336.	5	30 40		İ		
"C"	0.7444	4	P <sub>c</sub> mm	15079.	5	II .	424.2	5		
MR (Obs.)		2	PV/RT	<del> </del>	┼	Exp. L. 1. %/wt.		+-		
MR (Calc. (nD-d/2)	48.38 1.046	5 2	25°C	1.0000	5	u.				
Dielectric	2, 028	5	30 mm BP	1.0000	5	Dispersion	97.	2		
A 68 to	6.8988	5	ll t	0.9480 0.9337	5	Flash Point C				
B   227 °C	1494.	5	t <sub>c</sub>	0.24	5	Fire Point		4		
С ———	208.	5	∆Hc kcal/m			M. Spec. Ultra V.				
A* 68 to	1.39669	5	ΔHf ΔFf	1	İ	X-Ray Dif.				
<b>B</b> *[193 °C K	1403.7	5	<u> </u>	-		Infrared				
c	}		Viscosity centistokes	1	1	Solubility in +				
t <sub>k</sub> to	1.	1	η °c	Ì		Acetone Carbon tet.				
t <sub>x</sub> °C	<u> </u>			}		Benzene				
A'   20 to B'   68 °C	7. 25291	5				Ether				
C'	1695.6	5	B <sup>v</sup> to			n-Heptane Ethanol				
A** 20 to	1.73706	5	A °C			Water				
B'* 68 °C	1595.4	5	(B <sup>V</sup> )  to	1		Water in	L	$\perp$		
Ac  227 to	7.7400	5	(A <sup>V</sup> )  °C	1						
Bc t <sub>c</sub> °C	2310. - 313.	5	c <sub>p</sub> liq. °K		T	1 '	1			
Cc	313.	13	11	İ	1					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	102 72	+_	c vap.		1		1			
	182.73	5	II • .	L	<u> </u>	I	L			
$T_{R} = 0.8$						grams/100 gra		nt		
	ES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by for	mula			
SOURCE:		AI								
PURIFICAT	TION:	AF	PI							
LITERATU	RE REFERE	NCE	S:							

							No. 11	8
NAME	3, 3, 5	-Trime	thylheptane			STRUCTURAL	FORMUL.	A.
						сн,		
			T				сн сн,с	Н
Mole	Ref.	Molecu		Molecular Weight 142.2	76		Ċн <sub>3</sub>	,
% Pur.		Formu		Weight 142.2				Ref.
E D 40	T	Re	1	1	Ref.		T	Ker.
F.P. C F.P. 1007			dt/dP •C/mm		1	f to	İ	
B. P. *C	1		25°C	3. 6798	5	h .	1	
760 mm 100	155.68		BP t <sub>e</sub>	0.05207 0.0374	5	f' to	<b>-</b>	+
30	89.9 60.96	4 4	30 mm	0.7234	5	g'   ' <u>*</u> K_	1	1
1 <b>0</b> 1	38.9	5	ΔHm cal/g	1		h'		
Pressure	+		ΔHv cal/g	<del> </del>		m   to		
mm 25°C	4.51		25°C	74.69	5	n •K	┨	1
t <sub>e</sub>	1164.	5	30 mm BP	71.86 61.26	5	<u> </u>	<del> </del>	+
Density g/ml 20°C	. 0 74	20 2	te te (d, e)	59.35	5	m' to	ĺ	1
	0.74			59.28	5	o'  =	1	ļ
dt 25 4 30	0.73	48 4	ΔHv/T <sub>e</sub>	18.91	5	Surface tension	<del>                                     </del>	T
a b	0.75		d 61 to		5	dynes/cm. 20°C	24.05	5
Ref. Index	-0.03	80 4	- a' - 20 to	76.66	5	30 40	23.03 22.04	5
n <sub>D</sub> 20°C			e' j 61 °C	<del></del>	5	Parachor [P]	<del>                                     </del>	+
25 30	1.41		d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.232 4.303	5	20°C	l	l
"C"	0.74		ե <sup>©</sup> °C ຶ	322.	5	30 40	-	
MR (Obs.)		2	P <sub>c</sub> mm	14542.	5	Sugd.	424.2	5
MR (Calc.		5	PV/RT	1 0000	_	Exp. L.1.%/wt.		$T^{T}$
(nD-d/2)	1.04		25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	<u> </u>		BP	0.9500	5	Flash Point °C		╁
A 61 to B   209 °C		77   5	t <sub>e</sub>	0.9365 0.24	5	Fire Point		
C C	209.	5	ΔHc kcal/m			M Spec.	1	
A*  61 to			ΔHf ΔFf	1		Ultra V. X-Ray Dif.	ł	
B* ∟183 °C	1373.2	5	Viscosity			Infrared	<u></u>	
c	_]	- 1	centistokes	1		Solubility in +	ĺ	
t <sub>k</sub>   to		į	7 °	;		Carbon tet.		
t <sub>x</sub>   °C		789 5	4			Benzene Ether		
B' _ 61 °C	1664.2	5	<b>-</b>		-	n-Heptane		1
C'	227.	5	B <sup>V</sup> to			Ethanol Water		
A'* 20 to B'* 61 °C	1.73	506   5		-	}	Water in		1
Ac   209 to	7 65		(A <sup>V</sup> ) to	1				
Bc t °C	2176.	5	c <sub>p</sub> liq. •K	+	<del> </del>	1		
CE	301.	5	-ll <sup>-</sup>					1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K	.			}	
t <sub>e</sub> °C	173.35	5	c <sub>v</sub> vap.					
$T_R = 0.8$			ш	т	ь	+ grams/100 gra	ms solver	+
REFEREN		ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da			·*
SOURCE:			API					
PURIFICA'	TION:	1	IPI			· · · · · · · · · · · · · · · · · · ·		
LITERATU	RE REFE	ERENCE	S:					

						No. 119	<del>)</del>			
NAME	3,4,4-Trin	nethy	lheptane			STRUCTURAL	FORMULA	1		
			_			CH <sub>3</sub>				
Mole % Pur.	Ref. Mol	ecul mul		Molecular Veight 142.2	:76	Сн <sub>3</sub> сн <sub>2</sub> сн с (сн <sub>2</sub> ) <sub>2</sub> сн <sub>3</sub>				
		Ref.			Ref.		F	Ref		
F.P. °C F.P. 100%			dt/dP °C/mm	5 17(0	_	f to g				
B. P. °C 760 mm 10 <b>0</b>	164. 97.1	2	25°C BP t <sub>e</sub>	5.1760 0.05288 0.0374	5 5 5	h f <sup>1</sup> to				
30	67.7	4	30 mm	0.7358	5	g' <u>*K</u>				
10 1	45.3 7.9	5	ΔHm cal/g			h' i				
Pressure mm 25°C t <sub>e</sub>	3.1136 1184.	5	ΔHv cal/g 25°C 30 mm	77.06 73.54	5 5	m to				
Density g/ml 20°C	0.757	2	BP t <sub>e</sub> t <sub>e</sub> (d, e)	62.56 60.51 60.42	5 5 5	m' to				
d <sub>4</sub> 25	0.753 0.749	2	AHv/Te	18.88	5					
a b	0.773 -0.0 <sub>3</sub> 8	4	d 68 to e 183 °C d' 20 to	81,27 0,1141 79,12	5 5 5	Surface tension dynes/cm. 20°C 30	25.95 24.87	5		
Ref. Index n <sub>D</sub> 20°C	1,424	2	e'   68 °C	0.0824	5	Parachor [P]	23.82	5		
30	1.422 1.420	2	d g/ml vc ml/g tc °C	0.236 4.246 336.	5 5 5	20°C 30				
"C"	0.7444	4	P <sub>c</sub> mm	15079.	5	40 Sugd.	424.2	5		
MR (Obs.) MR (Calc.) (nD-d/2)	47.9 48.38 1.046	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5	Exp. L.1.%/wt. u. Dispersion	97.	2		
Dielectric	2,028	5	BP	0.9480	5	Flash Point °C	71.			
A 68 to B 1227 °C C	6.8988 1494. 208.	5 5 5	te tc AHc kcal/m	0.9337 0.24	5	Fire Point M. Spec.				
A* 68 to B* 193 °C	1.39669 1403.7	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared				
K c t <sub>k</sub> to C			Viscosity centistokes 7°C			Solubility in Acetone Carbon tet. Benzene				
A'   20 to B'   68 °C C'	7. 25291 1695. 6 226.	5 5 5	B <sub>v</sub> to			Ether n-Heptane Ethanol				
A'* 20 to B'* 68 °C	1.73706 1595.4	5	$\frac{A^{\mathbf{v}}}{ \mathbf{B}^{\mathbf{v}}  } - \frac{{}^{\mathbf{v}}C}{to}$			Water Water in				
Acl 227 to Bc t <sub>c</sub> °C	7.7400 2310. 313.	5 5 5	(A <sup>V</sup> )  °C c <sub>n</sub> liq. °K							
Cryos. A° consts. B°	3.5.	,	c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	182.73	5	c <sub>w</sub> vap.							
$T_{\mathbf{R}} = 0.82$	T <sub>c</sub>				·	grams/100 gra	ms solvent	;		
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:		AF								
PURIFICATION	ION: RE REFEREI	AF								
LITERATOR	E REFERE	NCES	o:							

							No. 120	)
NAME	3, 4, 5-Tr	imeth	ylheptane			STRUCTURAL	FORMULA	L.
						CH <sub>3</sub>		
					$\dashv$	сн <sub>3</sub> сн <sub>2</sub> сн сн сн	CH,CH,	
Mole	Ref. Mo	lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular	1	сн, сн		
% Pur.	F	rmul	a C10H22	Weight 142.2	76		3	
	r	Ref.		· · · · · · · · · · · · · · · · · · ·	Ref			Ref
F.P. °C		L	dt/dP			f to		l
F.P. 100%			*C/mm 25*C	5,1294	5	g <u>°K</u> _		1
B. P. *C 760 mm	164.	2	BP	0.05288	5	h		L
100	96.97	4	t <sub>e</sub>	0. 03729	5	f' to		ŀ
30	67.56	4	30 mm	0.7358	5	g'  K_	-	١
10 1	45.16 7.77	5	ΔHm cal/g			h'		L
Pressure		+	ΔHv cal/g			m to		
mm 25°C	3.146	5	25°C 30 mm	76.96	5		1	
t <sub>e</sub>	1190.3	5	BP BP	73.47 62.56	5	<u> </u>	<b> </b>	-
Density	0.750		te (d. a)	60.76	5	m' to		
g/ml 20°C	0.759 0.755	2 2	le (a,e)	60.44	5	", ' <u>-</u>		
d <sub>4</sub> 25 30	0.751	4	ΔHv/T <sub>e</sub>	18.96	5	Surface tension	<b> </b>	$\vdash$
a	0.775	4	d   68 to	81.12	5	dynes/cm. 20°C	26.22	1
b	-0.0 <sub>3</sub> 8	4	$\begin{bmatrix} \frac{\mathbf{e}}{\mathbf{d}} & \frac{1}{1} & \frac{183}{20} & \frac{\mathbf{c}}{\mathbf{to}} \end{bmatrix}$		5	30	25.13	1
Ref. Index			e' 68 °C		5	40	24.07	1
n <sub>D</sub> 20°C	1.424 1.422	2 2	d <sub>c</sub> g/ml	0.245	5	Parachor [P] 20°C	424.19	4
30	1.419	4	l v∵ml/g	4.080 335.9	5	30	424. 19	4
"C"	0.7424	4	-c -	15182.	5	40	424.18	4 5
MR (Obs.)	47.832	4	P <sub>c</sub> mm PV/RT	15162.	3	Sugd.	424.2	-
MR (Calc.)	48.380	5	25°C	1.0000	5	Exp. L.1.%/wt. u.		ł
(nD-d/2)	1.044	2	30 mm	1.0000	5	Dispersion		Į
Dielectric	2.028	5	BP t <sub>e</sub>	0.9480 0.9383	5	Flash Point °C		
A 68 to B 225 °C	6.89880 1494.	5	tc	0.232	5	Fire Point		↓
$\bar{c}$ ——=	208.	5	ΔHc kcal/m			M Spec.		ļ
A*   68 to	1.38938	5	ΔHf			Ultra V. X-Ray Dif.		
B* ∟ <sup>193</sup> °C	1401.63	5	ΔFf	<del> </del>	-	Infrared		
c			Viscosity centistokes		i '	Solubility in +		
tk to			η °c			Acetone Carbon tet.		1
*x 1		$\perp$				Benzene	1	
A'   15 to B'   68 °C	7.25295 1695.56	5			<u> </u>	Ether n-Heptane	[	
<u>c</u> , <u> </u>	226.	5	B <sup>v</sup>   to			Ethanol		
A!* 20 to	1.73702	5	AV   _ °C			Water Water in		
	1595.25	5	(B <sup>V</sup> ) to	1	]	Water in	<del> </del>	$\vdash$
Ac   225 to Bc   t <sub>c</sub> °C	7,74352	5	(A <sup>V</sup> )  °C			1	1	
Cc	314.01	5	c <sub>p</sub> liq. °K				ļ	1
Cryos. A°			c <sub>p</sub> vap. °K	1				l
consts. B°			_				[	
t <sub>e</sub> °C	182.78	5	c <sub>w</sub> vap.				1	
$T_{\mathbf{R}} = 0.82$	T <sub>c</sub>					grams/100 grai	ns solven	t_
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	calc, from det	da			
SOURCE:		AF	PI					
PURIFICAT:	ION:	AF	PI					
LITERATUR	E REFERE	NCES	:					

No. 121 3-isopropyl-2-methylhexane NAME STRUCTURAL FORMULA C<sub>3</sub>H<sub>7</sub> CH<sub>3</sub>CH CH (CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub> Ref. Mole Molecular Molecular ĊН<sub>3</sub>  $C_{10}H_{22}$ Weight 142.276 % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g °K 5.4068 5 B. P. °C h ВP 0.05179 5 760 mm 163. 2 0.0367 ſ١ to 100 97.3 4 g' °<u>K</u> 5 30 68.3 4 30 mm 0.7276 10 46.1 5 h! ∆Hm cal/g 9.0 m to ∆Hv cal/g Pressure °K n 25°C 78.20 mm 25°C 2.9372 o 30 mm 74.61 5 1181. 5 te BP 63.58 5 m' to Density 61.55 t<sub>e</sub> (d, e) 5 n' ۰ĸ g/ml 20°C 0.751 2 61.47 o'  $d_4^t$ 25 0.747 2 19.27 5 AHv/Te 30 0.743 4 Surface tension Т 68 to 82.56 5 0.767 a 44 dynes/cm. 20°C 25,13 ᇷᅱ 181 °C 0.1164 5 -0.038 ь 30 24.08 5 to 20 80,28 40 23.05 5 Ref. Index e' 68 0.0830 5  $^{n}D$ 20°C 1.421 Parachor [P] d<sub>c</sub> g/ml 0.245 5 25 2 1.418 20°C vc ml/g t °C 4.075 30 5 1.416 4 30 t<sub>c</sub> 5 333. 40 "C" 4 0.7453  $P_c$  mm15637. 5 Sugd. 5 424.2 MR (Obs.) 48.0 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 5 2 25°C 1,0000 5 (nD-d/2)u. 1.045 30 mm 1.0000 5 Dispersion 98. 2 Dielectric 2.019 5 BP 0.9480 5 Flash Point C 0.9341 5 A 68 to 6.9746 5 Fire Point 0.24 5 B | 224 °C 1519. 5 M. Spec. C 208. 5 AHc kcal/m Ultra V. ΔHf A\* 68 to 1.47382 5 X-Ray Dif. ΔFf B\* 191 °C 1428.9 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet.  $\mathbf{t_{x}}$ °C Benzene A' 20 to 7.33275 Ether В' 6<u>8</u> °C 1723.3 B<sub>v</sub> n-Heptane 226. 5 Ethanol to °C Water A'\* 20 to B'\* 68 °C 5 1.81623 Water in 1622.8 (B<sup>V</sup>) to Acl 224 to 7.7955 5 (A<sup>V</sup>)| °C Bc tc °C 2305. c<sub>p</sub> liq. ۰ĸ Cc 308 5 Cryos. A° consts. B° cp vap. °K c<sub>v</sub> vap. te °C 181,20 5  $T_{\mathbf{R}} = 0.82 T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

							No. 12	.2
NAME	3, 3-	Diethylhe	xane			STRUCTURAL	FORMUL	A
					$\neg \uparrow$	C <sub>2</sub> H <sub>5</sub>		
						CH <sub>3</sub> CH <sub>2</sub> c <sup>2</sup> (C	H <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub>	
Mole	Ref.			Molecular	74	́с <sub>2</sub> н <sub>5</sub>	2 2 3	
% Pur.		Formul	a 10 22	Weight 142.2	_	T		Ref.
	-T	Ref.		1	Ref.			Kei
F.P. °C F.P. 1007			dt/dP *C/mm		1	f to		İ
B. P. °C	+		25°C	6, 2085	5	h .		1
760 mm	166.3	2	BP	0.05212	5	f' to	<del></del>	+
100 30	70.9	4 4	t <sub>e</sub> 30 mm	0,7325	5	g'   '°K_		İ
10	48.6	5	ΔHm cal/g	+	1	h'		
1	11.3	5	ΔHv cal/g		-	m to	<b></b>	$\top$
Pressure mm 25°C	2.52	72 5	25°C	79.16	5	n '	4	1
t <sub>e</sub>	1190.	5	30 mm BP	75,27 64,12	5	<u> </u>		
Density	<del></del>		t.	62.03	5	m'   to	1	l
g/ml 20°C			t (d, e)	61.94	5	n'	1	
dt 25 4 30	0.76		ΔHv/T <sub>e</sub>	19.26	5			+
a .	0.78		d   71 to		5	Surface tension dynes/cm. 20°C	27.35	5
b	-0.03		d' 1 185 °C	0.1170 81.27	5	<b>y</b> 30	26,22	5
Ref. Index			e' 71 °C		5	40	25.13	5
n <sub>D</sub> 20°0	1.42		d g/ml	0.251	5	Parachor [P] 20°C		
30	1.42		d g/ml vc ml/g tc *C	3.981 340.	5	30		
"C"	0.74	12 4	1	16196.	5	40 Suad	424.2	5
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	10170.	<del>                                     </del>	Exp. L.1.%/wt.	464.6	+-
MR (Calc. (nD-d/2)	1.04		25°C	1.0000	5	u.		
Dielectric	<b></b>		30 mm BP	1.0000	5	Dispersion	95.	2
A 71 to			t <sub>e</sub>	0.9475 0.9333	5	Flash Point °C		
B 230 °C		0 5	tc	0.24	5	Fire Point	ļ	+
с	207.	5	∆Hc kcal/m			M Spec. Ultra V.		
A* 71 to		50   5	ΔHf ΔFf			X-Ray Dif.		1
B* 195 °C	1440.8	"	Viscosity			Infrared Solubility in +	ļ	+-
¢ .— —,	_		centistokes			Solubility in +		
t <sub>k</sub> t <sub>x</sub>			ነ የ ℃	·		Carbon tet.		
A'   20 to		375 5				Benzene Ether		
אי בן 11°C	225.	5 5	B <sup>V</sup>   to	<del> </del>	├	n-Heptane		ŀ
A'* 20 to			B <sup>V</sup> to			Ethanol Water		
A'* 20 to B'* 71 °C		5	(B <sup>V</sup> ) to	-1		Water in	<u> </u>	
Ac   230 to	7.84		(A <sup>V</sup> ) <sub>1</sub> °C	1				
	2379.	5	c <sub>p</sub> liq. °K	+	<del> </del>	1		
Cc — -	315.	5	1					
Cryos. Acconsts. B			c <sub>p</sub> vap. °K					İ
t <sub>e</sub> °C	184.93	5	c <sub>v</sub> vap.				ļ <u>.</u>	<u> </u>
T <sub>R</sub> = 0.8		2 47	7 2 7 1			grams/100 gra		<u>ıt</u>
REFEREN	ுக <b>ு</b> : 1-D			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION		PI					
PURIFICA			PI •	. <del> </del>				
	NO NOT	MCENCES.	•					

	2 4 51 11	••					No. 1	
NAME _	3,4-Dieth	ylhexa	ine			STRUCTURAL	FORMUI	JA
						C <sub>2</sub> F		
Mole	Ref. M	. 11		M-11		CH <sub>3</sub> CH <sub>2</sub> CH CH	сн <sub>2</sub> сн	3
% Pur.		iolecul ormula		Molecular Weight 142,27	76	С <sub>2</sub> н <sub>5</sub>		
		Ref.	<b>I</b>	T	Ref.			Ref.
F. P. °C			dt/dP			f to		
F.P. 100%			°C/mm	5 1040	١.	g  ° <u>K</u>		
B. P. °C 760 mm	162.	2	25°C BP	5.1849 0.05169	5	h		
10 <b>0</b>	96.4	4	t <sub>e</sub>	0.0367	5	f' to		
<b>30</b> 10	67.5 45.3	4 5	30 mm	0.7261	5_	g' <u>*K</u>		
ì	8.3	5	ΔHm cal/g	L		m   to		+-
Pressure			ΔHv cal/g 25°C	77.91	5	n °K		
mm 25°C	3.0744 1178.	5	30 mm	74.41	5	0		
t <sub>e</sub> Density	1	+-	BP	63.41	5	m¹ to		1
g/ml 20°C	0.754	2	te (d, e)	61.31	5	n'		
d <sub>4</sub> 25	0.750 0.746	2 4	AHv/Te	19.27	5			┼
<u>a</u>	0.770	4	d 67 to	82.25	5	Surface tension dynes/cm. 20°C	25,54	5
Ъ	-0.038	4	_e1 <u>80</u> °C d' 25 to	0.1163	5	<b>8</b> 30	24.47	5
Ref. Index	, ,,,,,		e' 67 °C	0.0826	5	40	23.43	5
<sup>n</sup> D 20°C	1.420 1.418	2 2	d <sub>c</sub> g/ml	0.247	5	Parachor [P] 20°C		
30	1.415	4	vc ml/g tc °C	4.050 332.	5	30		1
"C"	0,7407	4	P <sub>c</sub> mm	15718.	5	40 Sugd.	424.2	5
MR (Obs.) MR (Calc.)	47.8 48.38	2 5	PV/RT		†	Exp. L. l. %/wt.		1
(nD-d/2)	1.043	2	25°C 30 mm	1.0000	5	u.	04.6	2
Dielectric	2,016	5	BP	0.9480	5	Dispersion Flash Point C	94.6	+-
A 67 to	6.9734	5	ţ <sub>e</sub>	0.9342 0.24	5	Fire Point	l	1
B  _223 °C_ C	1515. 208.	5	tc ΔHc kcal/m		<del>                                     </del>	M. Spec.		
A* 67 to	1, 4735		ΔHf			Ultra V. X-Ray Dif.		
B*[_190°C K	1425.4	5	ΔFf	<del>                                     </del>	├	Infrared		
c		!	Viscosity centistokes		1	Solubility in +		
t <sub>k</sub> to °C			η °C			Acetone Carbon tet.		
t <sub>x</sub> °C A'   25 to	7. 3322	7 5				Benzene Ether		1
B' 67 °C	1719.5	5	B <sub>v</sub> to	<del> </del>		n-Heptane		
C'	226.	5	B <sup>V</sup> to ℃		1	Ethanol Water		1
A'* 25 to B'* 67 °C	1.81617 1619.0	2 5 5	$\frac{\mathbf{A}}{(\mathbf{B}^{\mathbf{V}}) } - \frac{\mathbf{C}}{\mathbf{to}}$	-		Water in		
Ac  223 to	7,7900	5	(A <sup>V</sup> )  °C					
Bc tc °C	2296.	5	c liq. °K		<b>†</b>			
Cryos. A°	307.	3	) P		}			
consts. B°	İ		c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	180.06	5	c <sub>w</sub> vap.					-
$T_{R} = 0.82$	T <sub>c</sub>		•			grams/100 gra	ms solve	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AF						
PURIFICAT	ION:	AF	PI					
LITERATUE	RE REFER	ENCES	S:					

							No. 12	24
NAME	3-Ethyl-	2, 2-dir	methylhexane			STRUCTURAL		
-	· · · · · · · · · · · · · · · · · · ·					сн <sub>3</sub>		
			T			Сн <sub>3</sub> с сн (	CH <sub>2</sub> ) <sub>2</sub> CH <sub>2</sub>	
Mole % Pur.		lolecul ormul		Molecular Weight 142.	276	сн <sub>3</sub> с <sub>2</sub> н <sub>5</sub>	2.2 3	
N Ful.		Ref		weight 112.	Ref			Ref.
F.P. °C		1	dt/dP			f to	T	1
F.P. 100%	İ		°C/mm		_	g*K	ļ	1
B. P. *C 760 mm	159.		25°C BP	4.2134 0.05239	5	h		
100	92.8	2 4	t <sub>e</sub>	0.0375	5	f' to		
30 10	63.7 41.5	4 5	30 mm	0.7284	5	g'   ' <u>°</u> K	-	1
i	4.5	5	ΔHm cal/g		<u> </u>	<u> </u>		+
Pressure			ΔHv cal/g 25°C	75.63	5	m to		1
mm 25°C	3.8974 1170.	5	30 mm	72.53	5	<u> </u>		
Density	<b>†</b>		BP te	61.67 59.70	5	m'   to		
g/ml 20°C	0.749 0.745	2 2	[ [a,e)	59.62	5	n' °K	-	
d <sub>4</sub> 25	0.741	4	AHv/T <sub>e</sub>	18,87	5	Surface tension	<u> </u>	+
•	0.765	4	d   64 to		5	dynes/cm. 20°C	24.87	5
<u> </u>	-0.038	4	d'   25 to	77.63	5	30 40	23.82 22.80	5
Ref. Index n <sub>D</sub> 20°C	1.420	2	e'   64 °C		5	Parachor [P]		+-
D 25 30	1.418 1.415	2 4	d g/ml v ml/g	0.234 4.281	5	20°C		
"C"	0.7456	4	tc °C	327.	5	30 40	-	1
MR (Obs.)	48.1	2	P <sub>c</sub> mm	14748.	5		424.2	5
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	1.046	2	30 mm	1.0000	5	Dispersion	97.	2
Dielectric A 64 to	2.016 6.8921	5	BP t <sub>e</sub>	0.9475 0.9336	5	Flash Point °C		
B ¿?13 ℃	1475.	5	t <sub>c</sub>	0.24	5	Fire Point	<del> </del>	+
C	209.	5	ΔHc kcal/m ΔHf		İ	M Spec. Ultra V.	ĺ	l
A* 64 to B* 187 °C	1.39567 1386.2	'   5     5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
к — — —			Viscosity			Solubility in +	<u> </u>	+
\$	1		centistokes °C	:		Acetone		İ
*x 1			•		1	Carbon tet. Benzene	!	
A'   25 to B'   64 °C	7.24984 1676.7	5 5				Ether n-Heptane	İ	
C'	227.	5	B <sup>V</sup> to			Ethanol		
A'* 25 to B'* 64 °C	1.73581	5	AV   °C	-		Water Water in	1	
Ac   213 to	7,6877	5	(B <sup>V</sup> )  to	1				$\top$
Bc t °C	2227.	5		<del></del>	<b></b>	1		
Cc	306.	5	•					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	177.00	5	c <sub>w</sub> vap.		L		<u> </u>	
$T_{\mathbf{R}} = 0.81$						f grams/100 gra		<u>ıt</u>
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE: PURIFICAT	ION:	AI						
LITERATUI	· · · · · · · · · · · · · · · · · · ·							
			••					

					<del></del> ,			No. 12	
NAME	4-Eth	yl-2, 2-d	imethylhexane			STRUCTU	RAL	FORMUI	-A
						CH <sub>3</sub>			
<del></del>			<del></del>				1-СН	сн <sub>2</sub> сн	_
Mole	Ref.	Molecul	ar 1	Molecular					3
% Pur.		Formul	ar C <sub>10</sub> H <sub>22</sub>	Weight 142.2	76	ĊH <sub>3</sub>	C <sub>2</sub> F	<sup>1</sup> 5	
		Ref			Ref.				Ref.
F. P. °C			dt/dP	<u> </u>		f			1
F.P. 100%	†		°C/mm			g	to _°K		İ
B. P. °C	1		25°C	2.6651	5	h i	· - <del></del>	1	1
760 mm	147.	2	<b>B</b> P	0.05084	5	<del></del>			+-
100	82.7	4	t <sub>e</sub>	0.0371	1 -	f' g'	to °K	l	
30 10	54.4 32.8	4 5	30 mm	0.7080	5_	h - 1	1		1
li	-3.2	5	ΔHm cal/g		L	]_h'			┿
Pressure	<del> </del>		AHv cal/g		]	m	to °K		İ
mm 25°C	6.40	21 5	25°C	72.79	5	"   '	- <u>-</u> -	•	
te	1140.	5	30 mm BP	70.57 60.28	5				↓
Density			t_	58.50	5	m'	to	1	1
g/ml 20°C			t <sub>e</sub> (d, e)	58.46	5	n'     -	_ <u>•</u> K_	1	1
dt 25 4 30	0.72		AHv/Te	19.07	5				₩
	0.72		d 54 to	76,62	5	Surface ten			
a b	0.74		e   163 °C	0.1112	5	dynes/cm.		22.81	5
Ref. Index	1 -0.03	<del>~                                     </del>	d' 25 to	74.67	5		30 40	21.82	5
n <sub>D</sub> 20°C	1.41	31 2	e'   54 °C	0.0752	5	Parachor			+-
D 25	1.41		d <sub>c</sub> g/ml	0.235	5	Parachor	20°C		
30	1.40	83 4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.249	5		30		
"C"	0.75	01 4	<u>_</u>	14390.	5		40	424.2	-
MR (Obs.)	48.4	2	P <sub>c</sub> mm	14370.	ļ	l		424.2	5
MR (Calc.)		5	PV/RT 25°C	1.0000	5	Exp. L.1.7	/wt.		
(nD-d/2)	1.04	66 2	30 mm	1.0000	5	u. Dispersion		97.	2
Dielectric	1.99	7 5	BP	0.9510	5	Flash Poin	°C		┿
A 54 to	6.90		t <sub>e</sub>	0.9384 0.24	5	Fire Point			
B 1_192 °C_	1441. 211.	5	t <sub>c</sub>	0.21	<u> </u>	M. Spec.			$\top$
A* 54 to	<del>                                     </del>		ΔHc kcal/m ΔHf	1		Ultra V.			
B* 173 °C	1.41	441   5	ΔFf			X-Ray Dif. Infrared			
к — — —	1.002. 1		Viscosity			h	+	<u> </u>	+
° ~		1	centistokes			Solubility i	n '		1
t <sub>k</sub> to			∥η °C			Carbon te	t.		
t <sub>x</sub> °C A'   25 to		22/ 5				Benzene		1	
B' 54 °C	7.27 1643.1	326   5 5				Ether n-Heptane		ì	
c,	229.	5	B <sup>V</sup> to C			Ethanol			
A'* 25 to	1.76	324 5	AV   °C			Water			
B'* 54 °C		5	(B <sup>V</sup> )  to		1	Water in			↓
Ac  192 to	7.60		(A <sup>V</sup> )  °C						
Bc tc °C	2068.	5	c <sub>p</sub> liq. °K	T	<b>†</b>	1			1
Cc — —	291.	5	11						1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	163.38	5	c vap.						1
$T_R = 0.80$			II • ·	Ц	<u> </u>	H +		L	<u> </u>
			TOT 2 T : :	Cala C				ms solve	nt
REFERENC	E9: 1-D	ow 2-A		Calc. from de	ec. da	ata 5-Calc.	oy 101	muia	
SOURCE:									
PURIFICAT		AI							
LITERATU	RE REF	ERENCE	5:						

							No. 126	6	
NAME	3-Ethyl-2,	3-di	methylhexane			STRUCTURAL FORMULA			
						C <sub>2</sub> H <sub>5</sub>			
Mole	Ref. Mo	lecul		Molecular			H <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>		
% Pur.		rmul		Weight 142.2	76	Ċн <sub>3</sub> Ċн <sub>3</sub>			
	,	Ref.			Ref			Ref.	
F.P. °C F.P. 100%		_	dt/dP			f to			
B. P. *C			*C/mm 25*C	6.3714	5	g <u>*K</u> _			
760 mm	169.	2	BP t <sub>e</sub>	0.05337 0.0375	5	f' to		<del>                                     </del>	
100 30	101.5 71.8	4	30 mm	0.7432	5	g'   'K_			
10	49.2 11.4	5	ΔHm cal/g			h'			
Pressure		-	ΔHv cal/g	70.53	_	m to			
mm 25°C	2.4826 1197.	5	25°C 30 mm	78.52 74.56	5	0   ==			
Density	1197.	,	BP	63.33 61.19	5	m'   to		$\vdash$	
g/ml 20°C	0.765	2 2	te te (d, e)	61.09	5	n'			
dt 25 4 30	0.760 0.755	4	ΔHv/T <sub>e</sub>	18.86	5	<u> </u>			
a	0.785	4	d   72 to e   188 °C	82.85 0.1155	5	Surface tension dynes/cm. 20°C	.27.06	5	
Ref. Index	-0.0 <sub>2</sub> 1	4	d'   25 to	80.63	5	30 40	25.67 24.34	5	
n <sub>D</sub> 20°C	1.427	2	d <sub>c</sub> g/ml	0.0846 0.225	5	Parachor [P]			
25 30	1.424 1.421	2 4	v ml/g	4.447	5	20°C			
"C"	0.7415	4	t <sub>c</sub> ·C	339.	5	40	424.2	_	
MR (Obs.)	47.8	2	P <sub>c</sub> mm	14482.	5	Sugd. Exp. L.1.%/wt.	424, 2	5	
MR (Calc.)   (nD-d/2)	48.38 1.045	5 2	25°C	1.0000	5	u.			
Dielectric	2.036	5	30 mm BP	1.0000 0.9470	5	Dispersion Flash Point °C	97.	2	
A 72 to	6.9053	5	t <sub>e</sub>	0.9323 0.24	5	Fire Point			
B 1.229 °C	1513. 207.	5	t <sub>c</sub> ΔHc kcal/m		-	M Spec.			
A*  72 to	1.40029.		ΔHf ΔFf			Ultra V. X-Ray Dif.			
B* 198 °C	1422.1	5	Viscosity		-	Infrared		<u> </u>	
t,	ļ		centistokes			Solubility in + Acetone		]	
t <sub>x</sub> to to			7 ℃			Carbon tet. Benzene			
A'   25 to	7. 25592	5				Ether			
B' _ 72 °C	1714.5 225.	5	B <sup>V</sup>   to	***************************************		n-Heptane Ethanol			
A'* 25 to	1.73827	5	AV   °C			Water Water in			
B'* 72 °C Ac   229 to	7,7601	5	(B <sup>V</sup> ) to					$\vdash$	
Bc t °C	2347.	5	(A <sup>V</sup> )  °C		_				
Cc	314.	5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	188.39	5	c <sub>v</sub> vap.						
$T_{R} = 0.82$						+ grams/100 gran	ns solven	t	
	ES: 1-Dow			alc. from det	da	ta 5-Calc, by for	nula		
SOURCE:	ION.		PI						
PURIFICAT	E REFEREN		PI						
LILERATOR	L REFERE	₹UES	••						
ŀ									
L									

Г					- 1		No. 1	٤1
NAME _	4-Ethy	1-2, 3-d	imethylhexane	STRUCTURAL FORMULA  CH 3				
<b>-</b>			T			снзсн сн сн	СН.СН	
Mole	Ref.			Molecular	ا ر	ĊH <sub>3</sub> Ċ <sub>2</sub> H		
% Pur.		Formu	<del></del>	Weight 142.	276	3 2	<u>-5</u>	
	<del> </del>	Re	<u>f.</u>		Ref.			Ref.
F.P. °C	ļ		_ dt/dP			f to		
F.P. 100%	ļ		°C/mm 25°C	5,1760	5	g '° <u>K</u>	-	
B. P. °C 760 mm	1,44	١,	BP	0.05288		h		<u> </u>
100	164. 97.1	2 4	t <sub>e</sub>	0.0374	5	f¹ to		
30	67.7	4	30 mm	0.7358	5	g' ' <u>°K</u>	ļ	
10 1	45.3 7.9	5	ΔHm cal/g			h'		↓
Pressure	···/		ΔHv cal/g			m to		1
mm 25°C	3.11	36 5	25°C	77.06	5	n <u>*K</u>	-	
t <sub>e</sub>	1184.	5	30 mm BP	73.54 62.56	5			<b>↓</b> —
Density			] t_	60.51	5	m' to		
g/ml 20°C	0.75		'e (d, e)	60.42	5	",	1	
$d_4^t$ $\frac{25}{30}$	0.75		ΔHv/T <sub>e</sub>	18.88	5		<del> </del>	+
a	0.77		d 68 to	81.27	5	Surface tension dynes/cm. 20°C	26, 22	5
ь	-0.03		183 °C 10 to	0,1141 79,12	5	<b>y</b> 30	25.13	5
Ref. Index			e'   68 °C	0.0824	5	40	24.07	5
<sup>n</sup> D 20°C	1.42		d <sub>c</sub> g/ml	0,236	2	Parachor [P]		
30	1.41		II V mile	4.233	5	20°C		
"C"	0.74	24 4	<u> </u>	336.	5	40		_
MR (Obs.)	47.8	2	P <sub>c</sub> mm	15139.	5		424.2	5
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.04		30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	2.02		BP	0.9480	5	Flash Point C		+
A 68 to B 227 °C	6.89	88 5	t <sub>e</sub> t <sub>c</sub>	0.9337 0.24	5	Fire Point		
c (227 C	208.	5	ΔHc kcal/m	+	<del>-</del>	M. Spec.		
A* 68 to	1,39	669 5	ΔHf	i		Ultra V. X-Ray Dif.		
B* 193 °C	1403.7	5	ΔFf			Infrared		
K		. [	Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			η °C			Acetone Carbon tet.		İ
t <sub>x</sub>   °C			<u></u> 1			Benzene		l
A'   25 to	7.25					Ether		
B'   _68 °C	1695.6 226.	5	B <sup>v</sup> to			n-Heptane Ethanol		
A1* 25 to	1.73		A L °C	1		Water		
B'* 68 °C	1595.4	5	(B <sup>V</sup> )  to	-		Water in		₩
Acl 227 to	7.74		(A <sup>V</sup> )  °C					
Bc t <sub>c</sub> °C	2313. 314.	5	c <sub>p</sub> liq. °K				1	
Cryos. A°	1 22.2.	+		1			Į	
consts. B°		1	P .					
te °C	182.73	5	c <sub>v</sub> vap.					
$T_{R} = 0.82$		L	и	<u> </u>		grams/100 gra	ms solver	at
REFERENC		ow 2-1	API 3-Lit. 4-	Calc, from de	t, da	ita 5-Calc. by for		
SOURCE:			PI			, ,	<del>.</del>	
PURIFICAT	ION:		PI					
LITERATUE								
TI DANI UI	KEF	LADINOP	<i>.</i>					

							No. 128	3	
NAME	3-Ethyl-2	4-di	methylhexane			STRUCTURAL FORMULA			
l f	-				$\neg$	C <sub>2</sub> H <sub>5</sub>			
<u> </u>			- <del></del>		$\dashv$	снзсн сн ст	н сн <sub>2</sub> сн		
Mole	Ref. Mo	lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular		с́н <sub>3</sub> с́н	I <sub>3</sub>		
% Pur.	Fo		a 10-22	Weight 142.2	_			Ref.	
<del></del>	<del></del>	Ref.		т	Ref	<del> </del>		Rei.	
F.P. °C F.P. 100%	+		dt/dP *C/mm	ļ		f to			
B. P. °C		-	25°C	5.1760	5	g   ' <u>*K</u>		ļ	
760 mm	164.	2	BP	0.05288 0.0374	5	f' to		<del>                                     </del>	
100 30	97.1 67.7	4	t <sub>e</sub> 30 mm	0.7358	5	g'   '°K_		ĺ	
10	45.3	5	ΔHm cal/g	0.1350	Ť	h'		ł	
11	7.9	5	<del></del>	<del> </del>		m   to			
Pressure mm 25°C	3, 1136	5	ΔHv cal/g 25°C	77.06	5	n •K_		ŀ	
t <sub>e</sub>	1184.	5	30 mm	73,54	5	<u>  °                                   </u>			
Density	1		BP t <sub>e</sub>	62.56 60.51	5	m'   to			
g/ml 20°C		2	t <sub>e</sub> (d, e)	60.42	5	n'   °K			
dt 25 4 30	0.755 0.751	2 4	AHv/T <sub>e</sub>	18.88	5	<u> </u>		ـــــ	
a.	0,775	4	d   68 to		5	Surface tension dynes/cm. 20°C	26, 22	5	
ь	-0.038	4	-a, - 183 €		5	<b>y</b> 30	25.13	5	
Ref. Index			e' 68 °C		5	40	24.07	5	
n <sub>D</sub> 20°C	1,424	2 2	d g/ml vc ml/g	0,236	5	Parachor [P]			
30	1.419	4	v <sup>c</sup> ml/g t <sub>o</sub> °C	4. 233 336.	5	30			
"C"	0.7424	4	1 6	15139.	5	40 Sugd.	424.2	5	
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	13137.	<u> </u>	Exp. L. l. %/wt.	727.2	<del>                                     </del>	
MR (Calc. (nD-d/2)	1.044	5 2	25°C	1.0000	5	u.			
Dielectric	+	5	30 mm BP	1.0000	5	Dispersion	97.	2	
A   68 to		5	te	0.9480 0.9337	5	Flash Point °C Fire Point			
B <u>227</u> °C	1494.	5	t <sub>c</sub>	0.24	5		<del> </del>	$\vdash$	
<u> </u>	208.	5	ΔHc kcal/m ΔHf	1	l	M Spec. Ultra V.			
A*  68 to   B*_193 °C		5	ΔFf			X-Ray Dif. Infrared			
к — — —			Viscosity			Solubility in +		$\vdash$	
c	<del>,  </del>		centistokes 77 °C			Acetone		1	
t <sub>x</sub>   t <sub>0</sub>		1	η °⊂	}		Carbon tet. Benzene			
A'   25 to		5				Ether			
B' L 68 °C	2 1695.6 226.	5	B <sup>V</sup>   to		$\vdash$	n-Heptane	İ		
A'* 25 to		5	Av i ℃			Ethanol Water			
B'* 68 °C		5	(B <sup>V</sup> ) to	1		Water in		<u> </u>	
Ac   227 to		5	(A <sup>V</sup> ) °C				l		
Bc tc_°C	2313. 314.	5	c <sub>p</sub> liq. °K			1			
Cryos, A°	<del></del>	<u> </u>	11						
consts. B°			Р .						
t <sub>e</sub> °C	182.73	5	c <sub>v</sub> vap.						
$T_R = 0.8$						+ grams/100 gran	ns solven	t	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc. from de	t, da	ta 5-Calc. by for			
SOURCE:		A	ΡĪ						
PURIFICA'			PĪ						
LITERATU	RE REFERE	NCES	5:		_				
}									

							No. 12	
NAME	4-Ethyl-2	4-di	methylhexane			STRUCTURAL	FORMUL	·Α
1						C <sub>2</sub> H	<sup>1</sup> 5	
			T		$\dashv$	сн,сн сн,с	сн2сн	,
Mole	Ref. Mo	lecul	ar cull	Molecular	- 1	,		5
% Pur.	Fo	rmul	a C <sub>10</sub> H <sub>22</sub>	Weight 142.2	76	CH <sub>3</sub> CH <sub>3</sub>		
		Ref.			Ref.			Ref.
F. P. ℃			dt/dP			f to		
F.P. 100%	•		°C/mm			g  °K		
B. P. ℃			25°C	4.0445	5	h		
760 mm	158.	2	BP t <sub>e</sub>	0.05230 0.0374	5	f' to		
100 30	91.9 62.9	4	30 mm	0.7269	5	g'   '° <u>K</u>		
10	40.7	5	()————————————————————————————————————	0.7209	-	h'		
1	3.8	5	ΔHm cal/g			m to		+
Pressure		1	ΔHv cal/g 25°C	75.35	5	n eK		1
mm 25°C	4.0755	5	30 mm	72.33	5	•		
t <sub>e</sub>	1168.	5	BP	61.56	5	m' to		+-
Density g/ml 20°C	0.747	2	te (d a)	59.61 59.53	5	n'°K		
	0.743	2	t <sub>e</sub> (d, e)	]	i i	o'		
d <sup>t</sup> 25 4 30	0.739	4	ΔHv/T <sub>e</sub>	18.88	5	Surface tension		+
a	0.763	4	d 63 to	79.44	5	dynes/cm. 20°C	24.60	5
Ъ	-0.038	4	-e - 176 °C to	0.1132 77.34	5	<b>8</b> 30	23.56	5
Ref. Index		_	e' 63 °C	0.0797	5	40	22.55	5
<sup>n</sup> D 20°C	1.419	2 2	d <sub>c</sub> g/ml	0.233	5	Parachor [P]		!
30	1.414	4	V <sub>C</sub> m1/g	4.292	5	20°C		
"C"	0.7459	4	t <sub>c</sub> °C	326.	5	40		
MR (Obs.)	<del></del>	2	P <sub>c</sub> mm	14685.	5	Sugd.	424.2	5
MR (Calc.		5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.045	2	25°C 30 mm	1.0000	5	u. Dispersion	97.	2
Dielectric	2.013	5	BP	0.9485	5		71.	+
A 63 to	6,8907	5	t <sub>e</sub>	0.9347	5	Flash Point C Fire Point		
B   212 °C	1472.	5	t <sub>c</sub>	0.24	5	M. Spec.		+
C	209.	5	ΔHc kcal/m ΔHf			Ultra V.		
A* 63 to B* 186 °C		5	ΔFf		1	X-Ray Dif.		
к — =	- 1302.1	1	Viscosity			Infrared		+
°	-		centistokes		i	Solubility in TACetone		
t <sub>k</sub> to			η °C	1	ĺ	Carbon tet.		
t <sub>x  </sub> °C		5		1		Benzene		
B'63 °C		5	ļ		<u> </u>	Ether n-Heptane		
C'	227.	5	B <sub>v</sub> to	1	ļ	Ethanol		
A** 25 to	1.73555	5	A C			Water Water in		
B'* 63 °C	<del></del>	5	(B <sup>V</sup> )  to	[		water III		+-
Ac 212 to		5	(A <sup>V</sup> )  °C		<u> </u>	]		
Bc t <sub>c</sub> °C	2211. - 304.	5	c <sub>p</sub> liq. °K	1				
Cryos. A°		广						
consts. B°		]	P		1			
t <sub>e</sub> °C	175.91	5	c <sub>v</sub> vap.					
$T_R = 0.8$				1		grams/100 gran	ms solver	nt.
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		AF				<del></del>		
PURIFICA	TION:	AF						
	RE REFERE							
J. I J. KAI U	NO NOTERE	. 1 C E 3	J.					

No. 130 STRUCTURAL FORMULA NAME 3-Ethyl-2, 5-dimethylhexane  $C_2H_5$ CH3CH CH CH2CH CH3 Molecular C<sub>10</sub>H<sub>22</sub> Ref. Ċнз Mole Molecular Ċна Weight 142.276 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 3.8827 B. P. \*C h BP 0.05220 5 760 mm 157. 2 5 f' 0.0374 te to 100 91.0 4 °K g' 30 4 30 mm 5 62.0 0.7254 10 39.9 5 h! ∆Hm cal/g 1 3, 1 5 to ΔHv cal/g Pressure n °K 25°C 75.06 mm 25°C 4,2614 5 o 30 mm 72.13 61.39 5 1165. 5 t<sub>e</sub> 5 BP Density g/ml 20°C to m 59.5 5 te te °K n' 0.741 2 (d, e) 59.38 5 01 25 0.737  $\mathbf{d_{4}^{t}}$ AHv/Te 18.88 5 30 0.733 4 Surface tension 79.14 62 to 5 0.757 4 dynes/cm. 20°C 23.82 175 °C 0.1131 5 ь -0.038 4 30 5 22.81 ď٠ 77.04 5 5 40 21.82 e' 62 °C Ref. Index 0.0792 5 20°C 1.416 2 [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor d g/ml vc ml/g 0.231 5 1.414 25 2 20°C 4.328 5 4 30 1.411 30 °C tc 323 5 40 "C" 0.7469 4 P<sub>c</sub> mm 14490. 5 424.2 5 Sugd. MR (Obs.) 48.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 5 1.0000 25°C (nD-d/2) 1.046 2 30 mm 1,0000 5 2 Dispersion 97. Dielectric 2,005 5 0.9485 BP 5 Flash Point °C 0.9348 A | 62 to 5 6.8894 5 Fire Point tc 0.24 5 [210 °C 1468. M Spec. C 209. 5 AHc kcal/m Ultra V. ΔHf A\* 62 to 1.39312 5 X-Ray Dif. ΔFf B\* 185 °C 1378.6 Infrared ĸ Viscosity Viscour, centistokes °C Solubility in c Acetone t<sub>x</sub> to Carbon tet. °C Benzene 25 to 7,24860 Ether B' ∟<u>62 °C</u> 1669.1 5 n-Heptane вŸ CI 227. Ethanol to  $\hat{\mathbf{A}^{\boldsymbol{V}}}$ 1.73529 °C Water 25 to 62 °C 5 Water in B'\* 1569.0 (BV) to Ac | 210 to 7.6637 (AV) ۰c 2190. 5 Bc \_tc\_ °C cp liq. ۰ĸ Cc 302 5 Cryos. A° consts. B° c<sub>p</sub> vap. ۰ĸ te °C c, vap. 174.77 5  $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

									No. 1	31	
NAME	4-Eth	yl-3,	3-di	methylhexane			STRUCTURAL FORMULA				
								CH <sub>3</sub>			
<u> </u>		r					CF	<sub>13</sub> сн <sub>2</sub> с сн	CH_CH	(_	
Mole	Ref.	Mo!	ecul	ar cu	Molecular	1	٠.			3	
% Pur.			mula		Weight 142.2	76		ĊH <sub>3</sub> Ċ <sub>2</sub> I	<sup>-1</sup> 5		
			Ref.			Ref.				Ref.	
F.P. °C				dt/dP			ſ	to			
F.P. 1009	5			°C/mm			g	<u>*</u> K	]		
B. P. °C				25°C BP	5.3948	5	h	i	l		
760 mm	165.		2	te	0.05298 0.0374	5	f'	to			
100 30	98.0		4	30 mm	0.7373	5	gʻ	<u>*K</u>	1		
10	46.1		5		0.1313	+ -	h'		Ì		
1	8.6		5	ΔHm cal/g	<del></del>	-	m	to		1	
Pressure				ΔHv cal/g 25°C	77.35	5	n	<u>°K</u>			
mm 25°C	2.97	762	5	30 mm	73.75	5	٥	1	ļ		
t <sub>e</sub>	1107.		-	BP	62.74	5	m'	l to		$\dagger \exists$	
Density g/ml 20°C	0.76	64	2	t <sub>e</sub> (d, e)	60.67	5	n'	L <u>°K</u>			
at 25	0.76	60	2			i	ە'	İ	ŀ		
<sup>4</sup> 30	0.75		4	d 69 to	18.89	5	Sur	face tension			
a	0.78		4	d   69 to e   184 °C	81.57 0.1142	5		es/cm. 20°C	26.92	5	
٠b	-0.0	38	4	d' 25 to	79.42	5	8	30 40	25.81 24.73	5	
Ref. Index		,,	2	e'   69 °C	0.0828	5	<u> </u>		47.13	+ 5	
n <sub>D</sub> 20°C	1.42		2	d <sub>c</sub> g/ml	0,238	5	Pa	rachor [P] 20°C			
. 30	1.42		4	v_ml/g	4.200	5	i	30			
"C"	0.74	124	4		1	1		40	424.2	-	
MR (Obs.)	47.8		2	P <sub>c</sub> mm	15307.	5	<u> </u>	Sugd.	424.2	5	
MR (Calc.	) 48.38		5	PV/RT 25°C	1,0000	5	Exp	b. L.1.%/wt.	ļ		
(nD-d/2)	1.04		2	30 mm	1.0000	5	Dis	u. persion	97.	2	
Dielectric		36	5	BP	0.9480	5		sh Point C		+-	
A 69 to		001	5	t <sub>e</sub>	0.9337 0.24	5		e Point			
B 1228 °C	1498.		5	t <sub>c</sub> ΔHc kcal/m	0.24	<del>                                     </del>	M.	Spec.			
A*  69 to		2704	5	∆Hf Kea1/III	1			ra V.			
B* 194 °C		706	5	ΔFf				Ray Dif. cared			
к — —	-		_	Viscosity			<b> </b>			+	
t <sub>L</sub> tō	-			centistokes	1	1		ubility in T			
t <sub>k</sub> to				η °C		1	Ca	rbon tet.			
A'   25 to		351	5					enzene :her			
B' 69 °C		,,,,,	5	L				Heptane			
C'	226.		5	B <sup>V</sup> to °C			Et	hanol			
A!* 25 to		3730	5		-			ater ater in			
B'* 69 °C			5	(B <sup>v</sup> )  to			<b></b>	B-G1 111	<del> </del>	+	
Ac 228 to		71	5	(A <sup>V</sup> )  °C			1		1		
Bc tc °C	2336. 316.		5	c <sub>p</sub> liq. °K						İ	
Cryos. A	<del></del>		<u> </u>			1			1		
consts. B°				c <sub>p</sub> vap. °K	1						
t <sub>e</sub> °C	183, 87	,	5	c <sub>w</sub> vap.							
$T_R = 0.8$			L	ш	<del></del>	1	+ ~.	rams/100 gra	me coler		
		)ov	2. 4	PI 3-Lit. 4-	Calc from 4					111	
REFEREN	ر <u>دی: ۱-۱</u>	,U#			Calc. from de	uz		-Calc. by for	u.d		
SOURCE:			AI		_						
PURIFICA			AI								
LITERATU	RE REF	ERE	NCE	S:							
Į.											

							No. 132	2		
NAME	NAME 3-Ethyl-3, 4-dimethylhexane						STRUCTURAL FORMULA			
]						C <sub>2</sub> H <sub>5</sub>				
<u> </u>	<del></del>				$\dashv$	сн,сн,с с	CH <sub>2</sub> CH	3		
Mole		lecul		Molecular	74	с́н, с́н		•		
% Pur.	Fo	rmul	<del></del>	Weight 142.2	_		<del></del>	To - C		
	<del></del>	Ref.		т	Ref	<del></del>		Ref.		
F.P. °C F.P. 100%		<del>                                     </del>	dt/dP *C/mm			f to				
B. P. *C	<b>+</b>		25°C	6.6033	5	h		l		
760 mm	170.	2	BP t	0.05354 0.0375	5	f' to		+		
100 30	102.3 72.5	4	t <sub>e</sub> 30 mm	0.7452	5	g'   'K_				
10	49.8	5	ΔHm cal/g		Н	h'		1		
1	12.0	5	ΔHv cal/g	<u> </u>	Н	m   to				
Pressure mm 25°C	2,3895	5	25°C	78.71	5	n				
t <sub>e</sub>	1200.2	5	30 mm BP	74.67 63.45	5 5	<del> </del>		├		
Density			t_	61.30	5	m' to				
g/ml 20°C	0.772 0.768	2 2	te (d, e)	61.19	5	0'				
d <sub>4</sub> 25 30	0.764	4	ΔHv/T <sub>e</sub>	18.85	5	Surface tension				
•	0.788	4	d 73 to		5	dynes/cm. 20°C	28.07	5		
Ref. Index	-0,038	4	d'   25 to	80.84	5	30 40	26.92 25.81	5 .		
n <sub>D</sub> 20°C		2	e' i 73 °C	<del> </del>	5	Parachor [P]		$\vdash$		
25 30	1.429 1.426	2 4	d g/ml vc ml/g	0.239 4.183	5	20°C				
"C"	0.7412	4	tc °C	346.	5	30 40		1		
MR (Obs.)	+	2	P <sub>c</sub> mm	15572.	5	Sugd.	424.2	5		
MR (Calc.		5	PV/RT 25°C	1 0000	,	Exp. L.1.%/wt.		1		
(nD-d/2)	1.045	2	30 mm	1.0000	5	u. Dispersion	97.	2		
Dielectric	<b>→</b>	5	BP	0.9470 0.9322	5	Flash Point °C		$\vdash$		
A 73 to B 235 °C		5	t <sub>e</sub> t <sub>c</sub>	0. 9322	5	Fire Point		<u> </u>		
c	207.	5	AHc kcal/m	<del> </del>		M Spec. Ultra V.				
A*   73 to		5	ΔHf ΔFf			X-Ray Dif.				
B* L200 °C	1423.7	5	Viscosity	<del> </del>	-	Infrared		ļ		
с	_		centistokes			Solubility in + Acetone				
tk			უ °⊂	; <b> </b>		Carbon tet.				
t   'C		5				Benzene Ether		ŀ		
B'i 73 ℃	1716.1	5			$\vdash$	n-Heptane				
C'	225.	5	B <sup>V</sup>   to			Ethanol Water		1		
A'* 25 to B'* 73 °C		5	(B <sup>V</sup> ) to	-1		Water in		<u> </u>		
Ac   235 to	7,8101	5	(A <sup>V</sup> ) <sub>1</sub> °C	ł						
Bc tc_°C	2428. 325.	5	c <sub>p</sub> liq. °K	+						
	+	+ -	il -	1						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K	1						
t <sub>e</sub> °C	189.56	5	c <sub>v</sub> vap.							
$T_R = 0.8$	2 T <sub>c</sub>		<del></del>			grams/100 gran	ns solven	t		
	CES: 1-Dow	2-A1	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for				
SOURCE:			PI							
PURIFICA'	TION:	A	PI							
LITERATU	RE REFERE	NCES	S:							
[										
L										

NAME	2, 2, 3, 3-1	etra	methylh <b>e</b> xane	STRUCTURAL FORMULA  CH <sub>3</sub> CH <sub>3</sub>				
Mole % Pur.		ecul		Molecular Weight 142.	276	сн <sub>3</sub> ċ ċ (с ċн <sub>3</sub> ċн <sub>3</sub>	H <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	
		Ref.			Ref.			Ref.
F.P. °C	-54.00	2	dt/dP			f to		} .
F.P. 1009	•		°C/mm			g  °K		
B. P. °C			25°C	4.3468	5	h i		
760 mm	160.31	2	BP	0.05280 0.0376	5	<del></del>		+
100 30	93.6 64.4	4 4	t <sub>e</sub>	ł	1 1	g' to		
10	42.1	5	30 mm	0.7322	5_	h'		
1	4.9	5	ΔHm cal/g	<u> </u>	$\perp$			+
Pressure			AHv cal/g			m to		
mm 25°C	3.7784	5	25°C 30 mm	75.62	5	<del>-</del>		
t <sub>e</sub>	1175.	5	BP	72.45 61.65	5 5			+
Density			t_	59.67	5	m' to		
g/ml 20°C		2	t <sub>e</sub> (d, e)	59.58	5	n'K_		
d <sub>4</sub> 25 30	0.76089 0.75732	2 4	ΔHv/T <sub>e</sub>	18.79	5			
		-	d 64 to	79.70	5	Surface tension	2/ 22	_
a b	0.77873 -0.0 <sub>3</sub> 713	4	_e _l1 <u>79 °C</u>	0.1126	5	dynes/cm. 20°C	26.99 25.99	5
Ref. Index	<del></del>	-	d'   25 to	77.63	5	40	25.02	5
n <sub>D</sub> 20°C		2	e'   64 °C		5	Parachor [P]		+
_D _25	1.42600	2	d <sub>c</sub> g/ml	0.241	5	20°C		
30	1.42390	4	vc ml/g t °C	4.150 334.	5	30		
"C"	0.7439	4	-	1	5	40	124 2	5
MR (Obs.)	47,900	5	P <sub>c</sub> mm	15392.	3		424.2	+ -
MR (Calc.	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.04595	2	30 mm	1,0000	5	u. Dispersion	97.	2
Dielectric	2.040	5	BP	0.9485	5	Flash Point C		+
A 64 to	6,8728	5	t <sub>e</sub>	0.9345	5	Fire Point		
B 1225 °C		5	°c .	0.24	5	M. Spec.		+
C	209.	5	ΔHc kcal/m ΔHf	!		Ultra V.		
A* 64 to B*  189 °C		5	ΔFf			X-Ray Dif.		
K K	- 1302.3	,	Viscosity			Infrared		
с	_		centistokes			Solubility in +		
t <sub>k</sub> Tto			η °C	i		Acetone Carbon tet.		
t <sub>x</sub> °C						Benzene		
A'   25 to B'   64 °C		5				Ether		İ
C'   -04-7	227.	5	B <sup>v</sup> to			n-Heptane Ethanol		
A'* 25 to	<del></del>	5	B <sup>V</sup> to C	·		Water		
B'* 64 °C		5	(B <sup>V</sup> ) to	-		Water in		<u>L</u>
Acl 225 to	<del></del>	5	(A <sup>V</sup> )  °C					
Bc tc °C	2282.	5		+	$\vdash$			
Cc	314.	5	c <sub>p</sub> liq. °K					l
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	178.66	5	c <sub>w</sub> vap.					
T <sub>R</sub> = 0.8			u			grams/100 gra	ms solve	nt.
		2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc, by for		<del></del>
SOURCE:		Al						
PURIFICA	TION:	AI						
	RE REFERE						<u> </u>	

							No. 13	4
NAME	2, 2, 3	, 4-Tetra	methylhexane			STRUCTURAL	FORMUL	——— А
<u> </u>			· · · · · · · · · · · · · · · · · · ·			CH <sub>3</sub> CI	I <sub>2</sub>	
1						сн, с сн с		ı
Mole	Ref.	Molecu		Molecular		3 сн3 сн3	2 3	,
% Pur.		Formu	1	Weight 142.2	_			
		Ref	<b></b>	·	Ref.		<del></del>	Ref.
F.P. °C F.P. 100%	<b></b>		dt/dP			f to	1	Į
B. P. *C	<del> </del>		*C/mm 25*C	3, 4772	5	g <u>*K</u>	·	1
760 mma	154.9	2	BP	0.05231	5	h	<b></b>	┼
100	88.9	4	t <sub>e</sub>	0.0377	5	f' to		
30 10	59.9 37.8	4 5	30 mm	0.7244	5	h'	1	
ì	1.1	5	ΔHm cal/g			<del> </del>	<b> </b>	+
Pressure	1		ΔHv cal/g 25°C	74.03	_	m to		
mm 25°C	4.82		30 mm	74.03 71.31	5		1	
Te	1157.	5	BP	60.57	5	m'   to	<del> </del>	†
Density g/ml 20°C	0.75	48 2	te (d, e)	58.67 58.59	5	n'	]	
t 25	0.75	08 2	ΔHv/Te	18.73	5	o'		ł
<sup>4</sup> 4 30	0.74		d   60 to		5	Surface tension		
a b	0.77		e   172 %		5	dynes/cm. 20°C	25.65	5
Ref. Index	-0.03	0 7	d 25 to	75.98	5	30 40	24.57	5
n <sub>D</sub> 20°C	1.42	26 2	e' j 60 °C	+	5	Parachor [P]		<del> </del>
2.5	1.42		d g/ml vc ml/g	0.233 4.293	5	20°C		Ì
30	1.41		tc °C	323.	5	30 40	ł	1
"C"	0.74		P <sub>c</sub> mm	14608.	5		424.2	5
MR (Obs.) MR (Calc.)	47.96		PV/RT	<del>                                     </del>		Exp. L.1.%/wt.	<del> </del>	<del>                                     </del>
(nD-d/2)	1.04		25°C	1.0000	5	u.		
Dielectric	2.02	4 5	30 mm BP	1.0000 0.9470	5	Dispersion	98.	2
A 60 to	6.86	21 5	t <sub>e</sub>	0.9333	5	Flash Point °C Fire Point	}	
B [210 °C		5	t <sub>c</sub>	0,24	5	M Spec.		+-
C	210.	5	ΔHc kcal/m	1		Ultra V.	İ	
A* 60 to B* 182 °C		036 5	ΔFf	-	1	X-Ray Dif. Infrared		
к ———			Viscosity			Solubility in +	-	+-
cto	-		centistokes °C	.		Acetone		1
tk   to			7	<b>`</b>		Carbon tet. Benzene		
A'   25 to			1			Ether		
B', ∟ 60 °C		5	B <sup>V</sup>   to		<del> </del>	n-Heptane	j	
	228.		∦ B' ∤ to			Ethanol Water	1	
A'* 25 to B'* 60 °C		956 5	(B <sup>V</sup> ) to	<b>-</b> l		Water in		
Ac   210 to	7,63	65 5	(A <sup>V</sup> )	1		H		
Bc t °C	2174.	5	c <sub>p</sub> liq. °K	<del></del>	<b>†</b> i	1	1	
Ce	304.	5	-1,				1	
Cryos. A° consts. B°			c <sub>p</sub> vap. °K	•				Ì
t <sub>e</sub> °C	172.42	5	c <sub>v</sub> vap.					
$T_R = 0.8$	l T <sub>c</sub>					+ grams/100 gra	ms solver	nt
REFERENC	ES: 1-D	ow 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICAT	ION:	A	PI					
LITERATU	RE REF	ERENCE	S:					

No. 135 2, 2, 3, 5-Tetramethylhexane STRUCTURAL FORMULA NAME сн₃с҅ CH CH2CH CH3 Mole Ref. Molecular Molecular Weight 142,276 င်မ³င္မမ³ ĊH3 C10H22 % Pur Formula Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm g °K 25°C 2,6618 5 B. P. °C h BP 5 0.05176 760 mm 148.4 2 t<sub>e</sub> 0.03783 5 f١ 100 to 83, 11 g' ۰ĸ 30 54.49 4 30 mm 0.7153 5 10 32,73 5 h! ∆Hm cal/g -3.54 5 1 m to AHv cal/g Pressure °K n 25°C 72.10 5 mm 25°C 6.4707 5 o 30 mm 69.89 5 5 te 1138.6 5 BP 59.34 m' to Density te te (d, e) 57.55 5 n' g/ml 20°C ۰ĸ 0.7378 2 57.48 ۰,  $d_4^t$ 25 0.7336 2 AHV/Te 5 18.69 30 0.7294 4 Surface tension 55 **to** 76.01 ď 5 0.755 -0.0<sub>3</sub>84 23.41 a 44 dynes/cm. 20°C 5 1<u>65</u> ℃ 0.1123 5 b ď٦ 5 30 22.36 25 to 73.98 5 5 40 21.34 Ref. Index 5 e' 55 °C 0.0750 20°C 1.4142 2 [P]  $n_{D}$ Parachor d<sub>c</sub> g/ml 0,225 5 25 1,4117 2 20°C vc ml/g tc °C 4.444 5 30 1.4092 4 30 310. 5 40 "C" 0.7471 4 P<sub>c</sub> mm 13805. 5 424.2 5 Sugd. MR (Obs.) 48,21 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 5 25°C 1.0000 5 u. (nD-d/2)1.0453 2 30 mm 1.0000 5 Dispersion 2 98. Dielectric 2,000 5 0.9470 RP 5 Flash Point °C 0.9338 5 A 55 to 6.8462 5 Fire Point 0.24 B 1199 °C 1424.4 5 M. Spec. Ultra V. С 211. 5 AHc kcal/m A\* 55 to B\* 175 °C ΔHf 5 1.36082 X-Ray Dif. ΔFf 1337.3 5 Infrared ĸ Viscosity Solubility in centistokes Acetone tk tx to Carbon tet. °C Benzene 25 to 7,21049 5 Ether B١ 5<u>5 °C</u> 1624.23 5 n-Heptane B<sup>V</sup> | C 229. 5 to Ethanol ۰c Water A'\* 25 to B'\* 55 °C 1.70077 5 Water in (B<sup>V</sup>)| 1524.4 5 to Acl 199 to 7.56018 5 (AV) °C Bc tc °C 2064.8 5 c<sub>p</sub> liq. ۰ĸ 293.8 c<sub>p</sub> vap. Cryos. A °K consts. B° t<sub>e</sub> °C c, vap. 165.04 5  $T_R = 0.81 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 136 2, 2, 4, 4-Tetramethylhexane STRUCTURAL FORMULA NAME CH<sub>3</sub> CH<sub>3</sub> сн<sub>3</sub>ċ сн<sub>2</sub>с CH2CH3 Molecular Weight 142, 276 Mole Ref Molecular ĊH3 ĊНз C10H22 Formula % Pur Ref. Ref Ref. dt/dP F.P. \*C F.P. 100% f to \*C/mm 25\*C <u>°K</u> g 3.2530 5 B, P. °C h BP 0.05218 760 mm 153.3 2 t<sub>e</sub> 0.0378 5 f to 100 87.44 4 g' °K 30 58.55 30 mm 0.7222 5 10 36.58 5 h† ∆Hm cal/g 1 -0.06 m to AHv cal/g Pressure n °K 25°C 73.54 mm 25°C 5.1912 o 30 mm 70.95 1152.6 5 t<sub>e</sub> BP 60.26 5 to m' Density te (d, e) 58.38 5 g/ml 20°C n' °K 0.7470 58.30 5 o' ď4 25 0.7428 2 ΔHv/Te 5 18.72 30 4 0.739 Surface tension d 77.56 5 59 to ℃ . 0.7638 4 dynes/cm. 20°C 24.60 \_171 °C 0.1129 e ь -0.0384 4 30 23.51 5 āï 75.47 1 5 40 22.45 Ref. Index e' 0.0772 5  $\mathbf{n}_{\mathbf{D}}$ [P] 20°C 1.4208 2 Parachor d<sub>c</sub> g/ml 0.228 5 25 1.4183 2 20°C ml/g 5 4.385 30 1.4159 4 c 30  $\mathbf{t_c}$ 318. 5 40 "C" 0.7489 4 mm 14184. 5 Sugd. 424.2 5 MR (Obs.) 48.28 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 48.38 25°C 1,0000 (nD-d/2) u. 1.0473 2 30 mm 1.0000 Dispersion 99. 2 Dielectric 2.019 5 BP 0.9470 Flash Point °C T 59 to 0.9334 t<sub>e</sub> 6.85770 5 Fire Point tc 0.24 В L206 °C 1444.2 M Spec. 5 AHc kcal/m 210. Ultra V. ΔHf A\* 59 to 1.36755 5 X-Ray Dif. ΔFf B\* 181 °C 1356.4 Infrared Viscosity Solubility in c centistokes Acetone t<sub>k</sub> to Carbon tet. •c Benzene A' 25 to 7.21851 Ether В' 59 °C 1644.5 5 n-Heptane 228. Ethanol ÁV °C Water A'\* 1.70692 to Water in (BV) B'\* 59 **°C** 1544.5 5 to (AV) Ac | 206 to 7.60949 5 °C Bc \_tc\_ 2135.9 liq. °K Cc 299.6 5 Cryos, Aº c<sub>p</sub> vap. ۰ĸ consts. B° c<sub>v</sub> vap. 170,60 5  $T_R = 0.81 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

									No. 13	31
NAME	2,2,4,	5-Te	tram	nethylhexane		- 1	ST	RUCTURAL	FORMUL	.A
								CH <sub>3</sub>	CH <sub>3</sub>	
<u> </u>							C		сн сн	
Mole	Ref.	Mo	ecul	ar	Molecular	1	Cı		_	
% Pur.		For	mula		Weight 142.2	276		Ċн <sub>3</sub> сн	3	
			Ref.			Ref.	Γ			Ref.
F. P. °C				dt/dP			f	1		T
F.P. 100%				°C/mm			g	to *K		
B. P. °C				25°C	2,6050	5	h	·		
760 mm	147.88		2	BP	0.05173 0.03784	5	f'			<del> </del>
100	82.64		4	t <sub>e</sub>	1	1 1	g'	to *K		
30 10	54.05 32.31		4	30 mm	0.7146	5	h'			
1	-3.93		5	∆Hm cal/g	<u> </u>	<u> </u>				┼
Pressure				AHv cal/g	71.04	-	m n	to to		1
mm 25°C	6, 62	67	5	25°C 30 mm	71.94 69.77	5	0			
t <sub>e</sub>	1137.1		5	BP	59.25	5	m'	A-		+-
Density				t <sub>e</sub> .	57.47	5	n'	to °K		1
g/ml 20°C	0.73 0.73		2 2	te (d, e)	57.39	5	0'	<u> </u>		1
d 25 4 30	0.72		4	ΔHv/T <sub>e</sub>	18.68	5		<u> </u>		$\vdash$
a	0.75		4	d 54 to	75.83	5		face tension es/cm. 20°C	23.12	5
Ъ	-0.03		4	e 165 °C to	73.81	5	8,	30	22.16	5
Ref. Index				e' 54 °C	0.0748	5		40	21.23	5
<sup>n</sup> D 20°C	1.41 1.41		2 2	d <sub>c</sub> g/ml	0,228	5	Par	achor [P]		
30	1.41		4	IV_mi/g	4.391	5		20 <b>°C</b> 30		
"C"	0.74		4	tc °C	310.	5		40	į	1
MR (Obs.)	48, 25		2	P <sub>c</sub> mm	13973.	5			424.2	5
MR (Calc.)	48.38		5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.04	545	2	25°C 30 mm	1.0000	5	Die	u. persion	99.	2
Dielectric	1.99	7	5	BP	0.9470	5		sh Point C	77.	+-
A 54 to	6.84	45	5	t <sub>e</sub>	0.9339	5		e Point		
B (199 °C	1422.1		5	tc	0.24	5	м	Spec.		+
C	211.		5	ΔHc kcal/m ΔHf	1		Ult	ra V.		
A*  54 to B* 175 °C	1.35 1335.1	962	5	ΔFf	ĺ			lay Dif. ared		
к — — —	1333.1			Viscosity						+-
°				centistokes				ability in Tetone		
t <sub>k</sub> to t <sub>x</sub> °C				ກ °⊂				rbon tet.		1
A'   25 to	7,20	014	5					nzene		
B' 54 °C	1621.9	714	5		<b>_</b>	$\perp$		her Heptane		
C'	229.		5	B <sub>v</sub> to			Et	hanol		
A'* 25 to	1.69	963	5	AV I C	-			ater ater in		
B'* 54 °C	1522.1		5	(B <sup>V</sup> )  to			W	PPCI III	<del> </del>	+-
Acl 199 to Bcit CC	7.56	096	5	(A <sup>V</sup> )  °C		$\sqcup$				
Bc t <sub>c</sub> °C Cc ⊏	2065.9 294.6		5	c <sub>p</sub> liq. °K		]				1
Cryos. A°			$\vdash$	l						
consts. B°				p -						
t <sub>e</sub> °C	164.45		5	c <sub>w</sub> vap.		1 1	l		1	1
$T_{\mathbf{R}} = 0.81$	T <sub>c</sub>		L	<u> </u>	1		+ 91	ams/100 gra	ms solver	 nt
REFERENCE		ow	2-A	PI 3-Lit. 4-	Calc, from de	t. da		-Calc. by for		
SOURCE:			AF					- ,		
PURIFICATI	ON:		AF							
LITERATUR		ERF								
		ابتاءت	CEG	•						

								No. 13	8
NAME	2, 2, 5	5,5-1	<b>Tetra</b>	methylhexane			STRUCTURAL	FORMUL	A
						$\neg \neg$	ÇН <sub>3</sub>	CH <sub>3</sub>	
i.							СН3С (СН2)		
Mole	Ref.	Mo	lecul		Molecular	1	Ċн <sub>3</sub>	Ċн <sub>3</sub>	
% Pur.			rmul		Weight 142.2	76	O.1. <sub>3</sub>	0113	
			Ref.	F		Ref.			Ref.
F.P. *C	-12.60		2	dt/dP			f to		T
F.P. 100%	1			°C/mm	1	H	g L _ °K		1
B. P. *C	1			25°C BP	1.7150 0.05083	5	h .		1
760 mm 100	137.46		2 4	t <sub>e</sub>	0.03782		f' to		
30	45.41		4	30 mm	0.6999	5	g'   '°K_	1	1
10	24.12		5	ΔHm cal/g			h'		1
1	-11.33		5	ΔHv cal/g	+	H	m to		
Pressure mm 25°C	10.50	0.2	_	25°C	68.95	5	n '°K_	ļ	1
t <sub>e</sub>	10.50	03	5	30 mm	67.53	5	<u> </u>	i	
Density	+		-	BP	57.58 55.96	5	m'   to		
g/ml 20°C	0.71		2	te te (d, e)	55.92	5	n' •K_		
d <sub>4</sub> 25	0.71		2	ΔHv/T <sub>e</sub>	18.68	5	o' ¦		
	0.71		4	d   45 to	<b>_</b>	5	Surface tension		
a b	0.73		4 4	e   153 °C		5	dynes/cm. 20°C	21.08	5 5
Ref. Index	-0.03	<u>·/</u>	-	d'   25 to		5	40	19.27	5
n <sub>D</sub> 20°C	1.40	550	2		<del></del>	5	Parachor [P]		+
- 25	1.40	316	2	d g/ml vc ml/g	0.222 4.504	5	20°C		
30	1.40		4	tc °C	293.	5	30 40		
"C"	0.75		4	P <sub>c</sub> mm	13225.	5	Sugd.	424.2	5
MR (Obs.)	48.57		2	PV/RT	+	$\vdash$	Exp. L.1.%/wt.		+
MR (Calc.) (nD-d/2)	48.38 1.04		5 2	25°C	1.0000	5	u.		
Dielectric	1.97		5	30 mm BP	1.0000 0.9510	5	Dispersion	102.	2
A   45 to	+		5	te	0.9389	5	Flash Point °C	İ	i
B   180 °C		,,,	5	tc	0.24	5	Fire Point		+
С	213.		5	AHc kcal/m			M Spec. Ultra V.		
A* 45 to	1.33	826	5	ΔHf ΔFf	1		X-Ray Dif.	İ	1
B* 163 ℃	1293.1		5	Viscosity	<del> </del>	-	Infrared		1
c	1			centistokes			Solubility in +		1
tk to				γ •c			Acetone Carbon tet.		
X '			Ļ				Benzene	1	
A'   25 to B'   45 °C	7.19 1579.0	208	5				Ether n-Heptane		
c, - 12 -	231.		5	B <sup>V</sup> to			Ethanol		
A'* 25 to		745	5	_AV_ °C	_		Water		1
B'* 45 °C	1479.7		5	(B <sup>V</sup> ) to	ì	1	Water in	<b></b>	+
Ac   180 to	7.46	624	5	(A <sup>V</sup> )  °C					
Bc tc °C	284.4		5	c <sub>p</sub> liq. °K					
Cryos. A°	1 -0		-	il					1
consts. B°				c <sub>p</sub> vap. °K					
te °C	152.84		5	c <sub>w</sub> vap.					
$T_R = 0.8$	OT <sub>C</sub>				<del></del>		+ grams/100 gran	me solver	<del></del>
REFERENC		ow .	2-AF	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for		
SOURCE:				PI					
PURIFICAT	'ION:			PI					
LITERATU		CR EN							
DAM I U	ADF	-REI	-CES	•					

NAME	2, 3, 3, 4-7	Cetra	methylhexane		1	STRUCTURAL	FORMUI			
IVAME						сн <sub>3</sub> сн <sub>3</sub>				
L						сн,сн с сн	3 CH-CH.			
Mole		ecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular	_,	ċн₃ċн₃	31120113	j.		
% Pur.	For	mula		Weight 142.2	+			<del></del>		
	1	Ref.		<del> </del>	Ref.			Ref.		
F.P. °C F.P. 100%			dt/dP °C/mm			f to				
B. P. °C		$\vdash$	25°C	5.2091	5	g '° <u>K</u>				
760 mm	164.59	2	BP	0.05316	5	h		+-		
100	97.44	4	t <sub>e</sub>	0.03760	5	f' to to*K				
30 10	67.92 45.45	5	30 mm	0.7381	5_	h'				
1	7.97	5	ΔHm cal/g		<b> </b>	m to		+		
Pressure		_	ΔHv cal/g 25°C	76.93	5	n				
mm 25°C	3.0990 1186.1	5	30 mm	73.40	5	0				
t <sub>e</sub> Density	-	Ť	BP	62.41	5 5	m <sup>e</sup> to				
g/ml 20°C		2	t <sub>e</sub> (d, e)	60.34 60.25	5	n'				
dt 25 4 30	0.7648 0.7602	2	ΔHv/T <sub>e</sub>	18.80	5	<u> </u>				
	0.7878	4	d 68 to	81.12	5	Surface tension	27 /0	-		
a b	-0.0392	4	e   183 °C	0.1137	5	dynes/cm. 20°C	27.69 26.39	5		
Ref. Index	<u> </u>		d 25 to	78.99 0.0824	5	40	25.13	5		
n <sub>D</sub> 20°C	1.4297	2	d <sub>c</sub> g/ml	0,231	5	Parachor [P]				
25 30	1.4269 1.4242	2	I V_mi/g	4.334	5	20°C				
"C"	0.7416	4	t <sub>c</sub> °C	336.	5	40				
MR (Obs.)	47,74	2	P <sub>c</sub> mm	14785.	5		424.2	5		
MR (Calc.)	48.38	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.				
(nD-d/2)	1.0450	2	30 mm	1.0000	5	Dispersion	97.	2		
Dielectric	1.901	5	BP	0.9480 0.9336	5	Flash Point C		1		
A 68 to B 226 °C	6.88337 1490.2	5 5	te tc	0.24	5	Fire Point				
c '	208.	5	∆Hc kcal/m			M. Spec. Ultra V.				
A* 68 to	1.38067	5	ΔHf ΔFf	1		X-Ray Dif.				
B*[193 °C K	1399.8	5	Viscosity		$\vdash$	Infrared				
c	.		centistokes			Solubility in *Acetone				
t <sub>k</sub> to			η °C		1	Carbon tet.				
t <sub>x  </sub> °C A'   25 to	7, 2364	5				Benzene Ether				
B' 68 °C	1691.2	5			<del> </del>	n-Heptane		1		
C'	226.	5	B <sup>v</sup>   to A <sup>v</sup>   °C			Ethanol Water				
A'* 25 to B'* 68 °C	1.72056 1591.0	5	(BV) - to	·		Water in				
Ac  226 to	7, 7251	5	(A <sup>V</sup> )  °C					T		
Bc <sub>l</sub> t <sub>c</sub> ℃	2307.0	5	c_liq. °K		1					
Cc — —	313.3	5 •	Р.		1					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K							
te °C	183.49	5	c <sub>v</sub> vap.							
$T_{\mathbf{R}} = 0.82$	Tc					grams/100 gra	ms solve	nt		
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc. by for				
SOURCE:		AF	PI							
PURIFICAT	ION:	AI	PI							
LITERATU	RE REFERE	NCES	5:							
j.										

·							No. 14	0		
NAME	2,3,3,	5-Tetra	methylhexan <b>e</b>			STRUCTURAL FORMULA				
l l						CH <sub>3</sub>				
h			Т			снзсн с сн				
Mole % Pur.	Ref.	Molecul		Molecular	74	Ċн <sub>3</sub> Ċн <sub>3</sub>	ĊH <sub>3</sub>			
70 Pur.		Formula Ref	10 22	Weight 142.2	Ref			Ref.		
E B &C	T	Kei.	1.715	T	Kei			Kei.		
F.P. *C F.P. 100%		$\dashv$	dt/dP *C/mm	}		f to				
B. P. *C	1		25°C	3.2148	5	h		ĺ		
760 mm 100	153.	2	BP t <sub>e</sub>	0.05215 0.03777	5	f' to		<del>                                     </del>		
30	87.19 58.31	4 4	30 mm	0.7217	5	g'   'K_		1		
10	36.35	5	ΔHm cal/g		1	h'				
1	-0.26	5	ΔHv cal/g			m to				
Pressure mm 25°C	5,258	5   5	25°C	73.46	5	n •K_		l		
te	1151.7	5	30 mm BP	70.89	5	ļ.—.! <u> —</u>		<u> </u>		
Density			t_	58.34	5	m' to		1		
g/ml 20°C	0.746 0.742	2 2	t <sub>e</sub> (d, e)	58. 26	5	n'   ' x-				
d <sub>4</sub> 25	0.738		ΔHv/T <sub>e</sub>	18.72	5	Surface tension		<del> </del>		
•	0.762	4	d   58 to		5	dynes/cm. 20°C	24.47	5		
Ь	-0.038	4	a'   25 to	75.39	5	30 40	23.43 22.43	5		
Ref. Index		6 2	e'   58 °C	<del></del>	5	Parachor [P]		<del>                                     </del>		
45	1.417	2   2	d g/ml	0.230 4.347	5	20°C				
30	1.414		tc *C	318.	5	30 40		1		
"C"	0.747	-	P <sub>c</sub> mm	14305.	5		424.2	5		
MR (Obs.) MR (Calc.		2	PV/RT	<del> </del>		Exp. L.1.%/wt.				
(nD-d/2)	1.046		25°C 30 mm	1.0000	5	u. Dispersion	97.	2		
Dielectric	2.015	5	BP	0.9470	5	Flash Point °C	77.	-		
A 58 to			te t <sub>c</sub>	0.9334 0.24	5	Fire Point				
B <u>L 206</u> °C	21443.3	5 5	ΔHc kcal/m		-	M Spec.				
A*  58 to			ΔHf	1	1	Ultra V. X-Ray Dif.				
B* 180 °C		5	ΔFf	-	<u> </u>	Infrared				
K C			Viscosity centistokes			Solubility in +				
to to			η °c	ļ		Acetone Carbon tet.				
\$ 00						Benzene				
A'   25 to B' 58 °C		58   5				Ether n-Heptane		į		
c,	228.	5	B <sup>V</sup> to			Ethanol				
A'* 25 to			AV	-		Water Water in				
B'* 58 °C	<del></del>	5 31 5	(B <sup>V</sup> ) to	1				<u> </u>		
Bc t °C	2137.7	5	(A <sup>V</sup> )  °C	+			,			
Cc	300.1	5	c <sub>p</sub> liq. °K							
Cryos, A° consts, B°			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	170, 26	5	c <sub>w</sub> vap.					1		
$T_R = 0.8$			L	I		+ grams/100 gran	ne solve-	<u> </u>		
REFEREN		w 2-AF	PI 3-Lit. 4-0	Calc. from det	da:	ta 5-Calc. by for		•		
SOURCE:		Al								
PURIFICA?	TION:	Al								
LITERATU										
	. =									

No. 141 2, 3, 4, 4-Tetramethylhexane NAME STRUCTURAL FORMULA CH<sub>3</sub>CH<sub>3</sub> CH3CH CH C CH2CH3 Mole Ref. Molecular Molecular Ċнз C10H22 ĊH % Pur Weight 142,276 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ۰<u>K</u> g 25°C 4.7322 5 B. P. ℃ h BP 0.05290 5 162.2 760 mm 2 t<sub>e</sub> 0.03758 5 ſ١ 100 95.38 4 to g' °K 30 66.01 4 30 mm 5 0.7344 10 43.66 5 h' ∆Hm cal/g 1 6.38 5 m to AHv cal/g Pressure n ۰ĸ 25°C 79.29 5 mm 25°C 3.4403 5 o 30 mm 72.95 te 1179.4 5 ΒP 62.03 5 m to Density 60.01 5 5 te te (d, e) <u>°K</u> g/ml 20°C 0.7639 2 59.93 o' 25  $\mathbf{d_{4}^{t}}$ 0.7596 2 ΔHv/Te 5 18.81 30 0.7553 4 Surface tension 66 to 80.44 5 0.7811 4 a dynes/cm. 20°C 26.91 0.1135 -0.0386 h 30 25.71 5 an) 25 5 66 °C 78.31 5 40 24.56 Ref. Index 5 e' 0.0814 20°C 1.4270 2 [P] nD Parachor 0,233 5 d<sub>c</sub> g/ml 25 1.4244 2 20°C vc ml/g t °C 5 4.290 30 1.4218 4 30 t<sub>c</sub> 333. 5 40 "C" 0.7425 4 5 14863 P<sub>c</sub> mm Sugd. 424.2 5 MR (Obs.) 47.82 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 5 1.0000 25°C (nD-d/2)1.0451 2 30 mm 1.0000 Dispersion 2 97. Dielectric 2.036 5 ВP 0.9480 5 Flash Point C 0.9338 5 A 66 to te t 6.88230 5 Fire Point 5 1482.1 0.24 B 1224 °C M. Spec. С 208. 5 AHc kcal/m Ultra V. ΔHf A\* 66 to 1.38187 5 X-Ray Dif. ΔFf B\*[191 °C 1392.1 Infrared Viscosity Solubility in centistokes Acetone to °C Carbon tet.  $\mathbf{t_{x}}$ Benzene A' | 25 to 7.23713 Ether B' \_66\_°C 1683.1 5 n-Heptane B<sup>V</sup> A C' 5 226. to Ethanol °C Water A1\* 25 to 1.72212 5 Water in B'\* 66 °C 1582.9 5 (B<sup>V</sup>)| to Ac 224 to 7.70922 5 (A<sup>V</sup>)| °C Bc t<sub>c</sub> °C 2278.0 cp liq. °K Cc 311.2 5 Cryos. A° consts. B° °K cp vap. c<sub>v</sub> vap. te °C 180.75 5  $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

· · · · · · · · · · · · · · · · · · ·							No. 14	2		
NAME	2, 3, 4, 5-	Tetra	methylhexane			STRUCTURAL FORMULA				
						CH <sub>3</sub>	CH <sub>3</sub>			
			1			сн <sub>3</sub> сн сн сн	CH CH <sub>3</sub>	i		
Mole % Pur.		lecul rmul		Molecular Weight 142.2	76	ĊН <sub>3</sub> СН	3			
		Ref.			Ref			Ref.		
F.P. ℃			dt/dP			f   to				
F.P. 100%			*C/mm 25*C	4, 4596	5	g <u>"K</u>		1		
B. P. *C 760 mm	161.	2	BP	0.05290	5	h		↓_		
100	94.20	4	t <sub>e</sub>	0. 03768		f' to		1		
30 10	64.87 42.55	4 5	30 mm	0.7334	5	h' ''				
1	5.33	5	ΔHm cal/g	ļ		m   to		+-		
Pressure mm 25°C	3.6754	5	ΔHv cal/g 25°C	75, 77	5	n K_				
mm 25°C	1176.1	5	30 mm	72.55	5	<u> </u>				
Density		$\vdash$	BP t	61.69 59.69	5	m'   to				
g/ml 20°C	0.757 0.753	2 2	te te (d, e) AHv/T	59.60	5	n'   °K_				
d <sub>4</sub> 25 30	0.749	4		18.76	5			+		
a	0.077	4	d   65 to		5	Surface tension dynes/cm. 20°C	25.95	5		
b	-0.038	4	d' _ 25 to	77.79	5	30 40	24.87 23.81	5		
Ref. Index n <sub>D</sub> 20°C	1.424	2	e'   65 °C		5	Parachor [P]	23.61	+-		
- 25	1.422	2	d g/ml	0.233 4.294	5 5	20°C				
30 "C"	1.419	4	tc °C	331.	5	30 40				
MR (Obs.)	0.7444 47.959	2	P <sub>c</sub> mm	14801.	5	н .	424.2	5		
MR (Calc.)		5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.				
(nD-d/2)	1.046	2	30 mm	1.0000	5	u. Dispersion	98.	2		
Dielectric A 65 to	2,028	5	BP t <sub>e</sub>	0.9480 0.9339	5	Flash Point °C		1		
B 222 °C		5	tc	0.24	5	Fire Point		-		
С	208.	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.				
A* 65 to B* 190 °C	1.3720 1384.4	5	ΔFf			X-Ray Dif.				
к ———	1301.1		Viscosity			Infrared Solubility in +		+		
t <sub>k</sub> - to	-		centistokes 7 °C			Acetone		1		
t <sub>X</sub> C	ļ		'	1		Carbon tet. Benzene				
A'   25 to B'   65 °C		5				Ether				
c, _ 5	226.	5	B <sup>V</sup> to			n-Heptane Ethanol				
A** 25 to		5	_AV  °C			Water		1		
	1574.6	5	(B <sup>V</sup> ) to	}		Water in		+		
Ac   222 to Bc   t <sub>c</sub> °C	7.69175 2261.0	5	(A <sup>V</sup> )  °C			4				
Cc	310.6	5	c <sub>p</sub> liq. ∘K					1		
Cryos. A° consts. B°			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	179.43	5	c <sub>v</sub> vap.							
$T_R = 0.8$						† grams/100 gran	ns solven	ıt		
	ES: 1-Dow	2-AI		Calc. from det	da:	ta 5-Calc. by for	mula			
SOURCE:			PI							
PURIFICAT	TON: RE REFEREI		PI 							
	NO NOI DIVE	VCE3	·•							
	_									

NAME	3, 3, 4, 4-	Tetra	methylhexane		STRUCTURAL FORMULA CH <sub>3</sub> CH <sub>3</sub>			
Mole % Pur.		lecul rmula		Molecular Veight 142.2	76	сн <sub>3</sub> сн <sub>2</sub> с с сн <sub>2</sub> сн сн <sub>3</sub> сн <sub>3</sub>	3	
		Ref.			Ref.		Ref.	
F.P. °C F.P. 1009	6		dt/dP °C/mm			f to g *K		
B.P. °C 760 mm 100 30 10	170.0 102.23 72.43 49.73 11.85	2 4 4 5	25°C BP t <sub>e</sub> 30 mm	6.5635 0.05361 0.03757 0.7456	5 5 5	f' to g'°K		
Pressure mm 25°C t <sub>e</sub>	2.4070 1200.2	5 5	ΔHv cal/g 25°C 30 mm BP	78.61 74.59 63.37	5 5 5	m   to n   - °K		
Density g/ml 20°0 dt 25 4 30	0.7783 0.7742	2 2 4	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub> d 72 to	61.23 61.12 18.82 82.91	5 5	n' °K_ o' Surface tension		
a b	0.7988 -0.0 <sub>3</sub> 82	4	e 190 °C to	0.1150 80.74	5	dynes/cm. 20°C 29.61 30 28.39 40 27.20	5 5 5	
Ref. Index n <sub>D</sub> 20°0 25 30		2 2 4	e'   72 °C d g/ml vc ml/g tc °C	0.0849 0.241 4.150 348.	5 5 5	Parachor [P] 20°C 30		
"C"	0.7406	4	P <sub>c</sub> mm	15745.	5	40 Sugd. 424, 2	5	
MR (Obs. MR (Calc. (nD-d/2) Dielectric	) 48.38 1.0456	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9470	5 5 5	Exp. L.1.%/wt. u. Dispersion 97.	2	
A 72 to B 236 °C	6. 89630	5 5 5	t e t c	0.9322 0.24	5	Flash Point C Fire Point M. Spec.		
A* 72 to B* 200 °C K c t <sub>k</sub> to	1421.7	5	ΔHf ΔFf Viscosity centistokes η °C			Ultra V. X-Ray Dif. Infrared Solubility in + Acetone Carbon tet.		
A'   25 to B'   72 °C C' A'* 25 to	1714.0 225.	5 5 5	B <sup>V</sup>   to A <sup>V</sup>   °C			Benzene Ether n-Heptane Ethanol Water		
B'* 72 °C  Ac 236 to  Bc t <sub>c</sub> °C	7,81479	5 5 5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C c <sub>p</sub> liq. °K			Water in		
Cryos, Acconsts, B		Ť	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C T <sub>R</sub> = 0.8	189.58	5	c <sub>v</sub> vap.			<u> </u>		
		2 ^	DI 2 124 4	Cala (many )		grams/100 grams solve	nt	
	CES: 1-Dow			Laic, irom de	t. da	ita 5-Calc, by formula		
SOURCE: PURIFICA	TION	AJ A1	PI					
	IRE REFERE							

No. 144 NAME 3-Isopropyl-2, 4-dimethylpentane STRUCTURAL FORMULA C<sub>3</sub>H<sub>7</sub> сизси си си сиз Ref. Mole Molecular Molecular ĊНą ĊН<sub>3</sub>  $C_{10}H_{22}$ % Pur. Weight 142, 276 Formula Ref. Ref Ref. F.P. C F.P. 100% -81.70 2 dt/dP to °C/mm <u>°K</u> g 25°C 3.8935 B. P. \*C 0.05219 5 BP 157.04 2 760 mm 5 0.03742 ſ١ t<sub>e</sub> to 100 91.08 4 g' °K 62.08 4 0.7254 5 30 30 mm 10 40.00 5 h' ∆Hm cal/g 5 1 3.151 to ł AHv cal/g Pressure °K n 25°C 75.09 4.2478 5 mm 25°C o 72.15 30 mm t<sub>e</sub> 1165.2 5 5 61.42 BP ١ to Density 5 m 59.50 te (d, e) °K g/ml 20°C 5 0.75830 2 59.42 I ۰, 0.75457  $\mathbf{d_{4}^{t}}$ 5 AHv/T 18.90 30 0.7508 4 Surface tension 62 to 79.17 5 0.7732 dynes/cm. 20°C 26.13 5 \_175\_ 0.1130 ь -0.0374 4 30 5 25.11 ď٠ to 5 77.08 25 5 40 24.12 Ref. Index e' °C 62 0.0793 5 n<sub>D</sub> 20°C 1.42463 [P] Parachor d<sub>c</sub> g/ml 0.241 5 25 1.42248 2 20°C ml/g 4.145 1.42017 4 30 c 30 •C 5 327. tc 40 "C" 0.7441 4 P<sub>c</sub> mm 15233. 5 Sugd. 424.2 5 MR (Obs.) 47.939 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 1.04548 u. (nD-d/2) 2 1,0000 5 97. 30 mm Dispersion 2 0.9485 Dielectric 2.029 5 BP Flash Point °C 5 0.9348 A 62 to 6.89055 5 Fire Point 0.24 L213 °C 1468.4 M Spec. Ultra V C 209. 5 ΔHc kcal/m ΔHf A\* 62 to 1.3942 5 X-Ray Dif. ΔFf B\* 185 °C 1379.1 5 Infrared ĸ Viscosity Viscos., centistokes °C Solubility in Acetone t<sub>x</sub> Carbon tet. °C Benzene 25 to 7.2498 Ether Вı 1669.7 62 °C 5 n-Heptane  $\mathbf{B}^{\mathbf{v}}$ 5 227. to Ethanol  $\overset{\checkmark}{A}^{V}$ °C Water 5 25 to 1.7364 Water in B'\* 62 °C 1569.6 5 (BV) to Ac | 213 to 7.68916 (A<sup>V</sup>) °C Bc tc\_°C 2224.8 5 cp liq. °K Cc 307.2 5 Cryos, A° °K cp vap. consts. B° c<sub>v</sub> vap. te °C 174.81 5  $T_R = 0.81 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

Г	2 2 1	Di -41-	.1 2			—Т			No. 1		
NAME _	3, 3-1	Dietny	71-2-	methylpentane	!		STRUCTURAL FORMULA				
		,						C <sub>2</sub> H <sub>5</sub>			
Mole	Ref.	Mol	1		Molecular	Ì	(	сн <sub>3</sub> <b>с</b> н с	CH <sub>2</sub> CH <sub>3</sub>		
% Pur.	Kei.	For	mula	C <sub>10</sub> H <sub>22</sub>	Weight 142.2	76		с́н <sub>3</sub> с <sub>2</sub> н <sub>5</sub>			
			Ref.			Ref.	Γ'''			Ref.	
F. P. °C				dt/dP			f	to			
F.P. 100%				°C/mm		١.	g	<u>*K</u>			
B. P. ℃	1,74		.	25°C <b>BP</b>	7.8588 0.05386	5	h				
760 mm 100	174.	,	2	t <sub>e</sub>	0.03746		f'	to			
30	75.87	7	4	30 mm	0.7507	5	g'	!° <u>K</u>			
10 1	53.01		5 5	∆Hm cal/g		Γ	_h'				
Pressure	1		-	ΔHv cal/g		T -	m	to *K			
mm 25°C	1.97	757	5	25°C 30 mm	79.99	5	0	<u>_</u>			
t <sub>e</sub>	1211.6		5	BP	75.57 64.20	5	m'			+-	
Density		.		t <sub>e</sub> ,	61.97	5	n'	to °K			
g/ml 20°C	0.78		2 2	te (d, e)	61.87	5	٥'			l	
d <sub>4</sub> 25 30	0.77		4	ΔHv/T <sub>e</sub>	18.87	5	Sur	face tension		+-	
a	0.80		4	d 76 to		5		es/cm. 20°C	29.25	5	
b	-0.00	)1	4	d' 25 to	82.16	5	8	30 40	27.78 26.36	5	
Ref. Index	1.43	35	2	e'   76 °C		5		achor [P]	20.30	+-	
D 25	1.43	32	2	d g/ml vc ml/g	0.229 4.361	5	1	20°C			
30	1.42		4	t <sub>c</sub> °C	348.	5		30 40		ļ	
"C"	0.74		4	P <sub>c</sub> mm	14985.	5			424.2	5	
MR (Obs.) MR (Calc.)	47.59		2 5	PV/RT			Exp	. L.1.%/wt.			
(nD-d/2)	1.04		2	25°C 30 mm	1.0000	5	1	u.	0.7		
Dielectric	2,05	59	5	BP	1.0000 0.9470	5		persion	97.	2	
A 76 to	6.91	118	5	te	0.9319	5		sh Point C e Point			
B 1236 ℃ C	1531.5		5	t <sub>c</sub>	0.24	5		Spec.	<b></b>	+-	
A*  76 to	1.40	1206	5	∆Hc kcal/m ∆Hf	!		Ulti	ra V.			
B* 204 °C	1439.9	200	5	ΔFf				lay Dif. ared			
к — — —				Viscosity		1	<u> </u>	ability in +		+	
t <sub>k</sub> – to	•			centistokes り °C		1	Ac	etone			
t <sub>x</sub> °C				'				rbon tet.			
A'   25 to	7. 25	589	5				Et	her			
B' 1_76 °C	1733.4		5	B <sup>V</sup> to				Heptane hanol			
A'* 25 to	1.73	394	5	AV I °C				ter			
B'* 76 ℃	1633.1		5	(B <sup>V</sup> )  to	-		W	ter in			
Acl 236 to	7.82	257	5	(A <sup>V</sup> )  °C							
Bc t <sub>c</sub> °C	2451.7		5	c liq. °K			1				
Cryos. A°	+		_	1							
consts. B°				P -							
t <sub>e</sub> °C	194.10	)	5	c <sub>w</sub> vap.							
$T_R = 0.8$	2T <sub>c</sub>						+ g1	ams/100 gra	ms solve	nt	
REFERENC		)ow	2-A	PI 3-Lit. 4	-Calc. from de	t. da	ata 5	-Calc. by for	mula		
SOURCE:			AI	PI							
PURIFICAT	ION:		ΑF	PI							
LITERATU	RE REF	EREN	CES	3:							

NAME								No. 146	<u> </u>
Mole	NAME	3-Ethyl-2	, 2, 3	-trimethylpenta	ne		STRUCTURAL I	FORMULA	1
Mole						$\neg \neg$	CH <sub>3</sub> C <sub>2</sub> H <sub>5</sub>		
Ref.   Ref.						$\neg \neg$	сн,с с	CH <sub>2</sub> CH <sub>3</sub>	
Ref.   Ref.		Ref. Mo	lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular	76	ĊH <sub>3</sub> ĊH <sub>3</sub>		
F.P. °C	% Pur.		_		weight 142.2	_	T	<del></del>	Ref
F. P. 100%	E D 86	T	Kei.			Kei			ICI.
B.P. °C   C   C   C   C   C   C   C   C   C	F.P. 100%	†	_				1 1 1 11 11 11		
Too				25°C			-		
30				1) .			<del></del>		
10					0.7429	5			
Pressure mm 25°C				ΔHm cal/g			h'		
The color of the		10.41	3	<del></del>	1				
Nemark   196.0   5		2.6452							
Solution   Solution	t <sub>e</sub>	1196.0	5	BP		5			-
At   25 to   1.046   25 to   1.046   2   25 to   20		0.781	,	te (d.s)			n'  °K_		1
Color   Colo				e (4, 5)		1			
Solubility in the control of the c		<del></del>							
Ref. Index nD 20°C 1.436 2 2.5 1.434 2 2.5 1.431 4 4				e <u>  187</u> °C	0.1140	5			
No.		<del></del>	Ė						
30 1.431 4 4 c ml/g 4.131 5 30  30  40  40  40  40  40  40  40  40	n <sub>D</sub> 20°C	1.436		<del></del>	<del>                                       </del>				
NR (Obs.)	45			V_ mi/g		5			
MR (Obs.)			-	t <sub>c</sub> ·C	1		40		
MR (Calc.)   48.38   5   1.046   2   25°C   30 mm   1.0000   5   5	MR (Obs.)	+			15744.	5	<u> </u>	424.2	5
Dispersion   97.   2   30 mm   1.0000   5   Dispersion   97.   2	MR (Calc.	48.38	5		1,0000	5			
A 71 to 6.8911 5		<del></del>		30 mm	1.0000	5		97.	2
Note			_	il .					
C				tc	0.24	5			
A*   71 to   1.38518   5   AFf			5	∆Hc kcal/m					
Viscosity   Centistokes   Viscosity   Carbon tet.   Benzene   Ether   Dethanol   Water   Viscosity   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Carbon tet.   Dethanol   Viscosity   Viscosity   Viscosity   Centistokes   Viscosity   Carbon tet.   Dethanol   Viscosity   Viscosity   Viscosity   Carbon tet.   Dethanol   Viscosity   Viscosity   Viscosity   Viscosity   Carbon tet.   Dethanol   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Carbon tet.   Dethanol   Viscosity   Viscosity   Viscosity   Viscosity   Carbon tet.   Dethanol   Viscosity							X-Ray Dif.		
Centistokes   N	K 177	- 1413.1	]	Viscosity		<u> </u>			
Carbon tet.   Benzene   Ether		-1				1			
A'   25 to   7,2419   5   Ether   1,705,4	I.K I			"		1			
C'	A'   25 to					l			
At 25 to B 1.72474 5	B' ∟ 71 °C			BV I to	<del> </del>	<u> </u>			
Bi				AV i °C					
Bc   tc °C   2410.3   5   cp liq. °K				(B <sup>V</sup> ) to	1		Water in		<u> </u>
Cryos. A° consts. B° c <sub>p</sub> vap. °K c <sub>p</sub> vap. T <sub>R</sub> = 0.82 T <sub>C</sub> + grams/100 grams solvent  REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula  SOURCE: API  PURIFICATION: API	Ac   234 to	7.79459		(A <sup>V</sup> ) °C					
Cryos. A°	Cctc°C	2410.3 324.7		c <sub>p</sub> liq. °K					
consts. B°         p				11					
T <sub>R</sub> = 0.82T <sub>C</sub> grams/100 grams solvent  REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula  SOURCE: API  PURIFICATION: API	consts. B°		ļ	-		Ì			
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula  SOURCE: API  PURIFICATION: API	t <sub>e</sub> °C		5	c <sub>v</sub> vap.					
SOURCE: API PURIFICATION: API									t
PURIFICATION: API		CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by form	nula	
LIIDRAIURD REFERENCES:									
	LITERATU	RE REFERE	NCES	i:					

г							No. 1	41	
NAME	3-Eth	yl-2, 2, 4	-trimethylpenta		STRUCTURAL FORMULA				
1						CH <sub>3</sub>	CH <sub>3</sub>		
<del> </del>						сн, с сн			
Mole	Ref.	Molecul:	ar C <sub>10</sub> H <sub>22</sub>	Molecular	İ				
% Pur.		Formula	10 <sup>11</sup> 22 V	Weight 142.2	276	ċн <sub>3</sub> ċ <sub>2</sub> н <sub>5</sub>			
		Ref.			Ref.			Ref.	
F. P. °C			dt/dP			f to			
F.P. 100%	,		°C/mm			g  ° <u>K</u>	ł		
B. P. °C			25°C	3.5358	5	h	1		
760 mm	155.3	2	BP	0.05235 0.03680	5	<del></del>	<del> </del>	+	
100 30	89.23		t <sub>e</sub>		l	g'   to		ľ	
10	60.23		30 mm	0.7250	5_	h'	· [	1	
1	1.39		AHm cal/g		<u> </u>	<u>}</u>	<del> </del>	+	
Pressure			ΔHv cal/g		l _	m to		ł	
mm 25°C	4.73	70 5	25°C 30 mm	74.15 71.40	5	<del></del>	1		
t <sub>e</sub>	1160.4	5	BP	60.74	5	<b>├</b> ─ <del>,</del>	<u> </u>	+	
Density			t	58.84	5	m' to			
g/ml 20°C			te (a, e)	58.76	5	"	1		
dt 25 4 30	0.75		AHv/T <sub>e</sub>	18.76	5	I		+	
a	0.77		d 60 to	78.15	5	Surface tension	25.96	5	
b	-0.03		e 173 °C to	0.1121	5	dynes/cm. 20°C	24.88	5	
Ref. Index			d'   25 to	76.10 0.0782	5	40	23, 82	5	
n <sub>D</sub> 20°C				0,234	5	Parachor [P]		T	
25 30	1.41		d g/ml	4, 272	5	20°C			
	1.41		tc °C	324.	5	30 40	Į.		
"C"	0,74		P <sub>c</sub> mm	14706.	5		424.2	5	
MR (Obs.)			PV/RT		<del> </del>	Exp. L.1.%/wt.		1	
MR (Calc. (nD-d/2)	1.04		25°C	1.0000	5	u.			
Dielectric	2,02		30 mm BP	1.0000 0.9485	5	Dispersion	98.	2	
A 60 to			t	0.9349	5	Flash Point °C			
B (211 °C		5 5	t <sub>c</sub>	0.24	5	Fire Point	ļ	4	
c '	209.	5	AHc kcal/m			M. Spec.			
A* 60 to	1.36	84 5	ΔHf	İ		Ultra V. X-Ray Dif.			
B*  <u>1</u> 8 <u>3 °C</u>   K	1363.7	5	ΔFf	<del></del>		Infrared	1	İ	
C			Viscosity centistokes			Solubility in +			
t <sub>k</sub> – tō	-		η °c			Acetone			
t <sub>x</sub> °C			•			Carbon tet. Benzene			
A'  25 to						Ether			
B'   _60_°C	1653.1 227.	5	B <sup>V</sup> to		t —	n-Heptane	ļ		
A'* 25 to	<del></del>		B <sup>V</sup> to C		ŀ	Ethanol Water			
B'+ 60 °C		5	(B <sup>V</sup> )  to	1	1	Water in			
Ac  211 to			(A <sup>V</sup> )  °C	1					
Bc <sub>i</sub> t <sub>c</sub> °C	2183.2	5		<del>                                     </del>	+	1			
Cc	304.6	5	c <sub>p</sub> liq. °K			1		1	
Cryos. A°			c <sub>p</sub> vap. °K	1	1			1	
consts. B°	+	-	i)	1					
t <sub>e</sub> °C	172.95	5	c <sub>w</sub> wap.	L	L	L			
$T_{R} = 0.8$	l T <sub>c</sub>					grams/100 gra	ms solve	nt	
REFEREN	CES: 1-D	ow 2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ata 5-Calc. by for	mula		
SOURCE:		AF	PI						
PURIFICA'	TION:	AF	PI						
LITERATU									
L									

No. 148 NAME 3-Ethyl-2, 3, 4-trimethylpentane STRUCTURAL FORMULA С<sub>2</sub>H<sub>5</sub> Сн<sub>3</sub>сн с с CH CH3  $\dot{c}_{H_3}\dot{c}_{H_3}\dot{c}_{H_3}$ Mole Ref. Molecular Molecular  $C_{10}H_{22}$ Weight 142.276 Formula % Pur. Ref. Ref. Ref F.P. C F.P. 100% dt/dP to °C/mm <u>•ĸ</u> g 25°C 6.3989 5 B. P. \*C h BP 0.05358 760 mm 169.44 2 4 f' 0.03759 5 to 100 101.72 °K g¹ 71.94 4 30 30 mm 0.7450 5 49, 26 5 10 h' ∆Hm cal/g 5 11.43 1 to m AHv cal/g Pressure °K n 25°C 78.42 mm 25°C 2.4751 5 5 o 74.45 63.25 5 1198.6 30 mm t<sub>e</sub> BР 5 m to Density 5 te (d, e) 61.11 °K g/ml 20°C 0.7773 2 61,00 5 ۰, 25 0.7733 2 d4 ΔHv/Te 18.81 5 4 30 0.7693 Surface tension ī 82,71 72 5 0.7933 -0.0<sub>3</sub>80 44 28.85 dynes/cm. 20°C •c 0.1149 189 ь 5 30 27.67 ď to 5 80.53 25 5 40 26.54 . Ref. Index e' 72 0.0846 5 20°C [P] n<sub>D</sub> 1,4333 2 Parachor d<sub>c</sub> g/ml 0.240 5 25 2 1.4310 20°C , c 4.168 ml/g °C 30 1.4285 4 30  $t_c$ 5 347. 40 "C" 0.7399 4 P<sub>c</sub> mm 15653. 5 5 Sugd 424.2 MR (Obs.) 47.60 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 48.38 25°C 1.0000 (nD-d/2)1.0447 2 1.0000 30 mm 2 Dispersion 97. Dielectric 2.054 5 5 RP 0 9470 Flash Point °C 72 to 0.9322 5 6.8938 5 Fire Point tc 0.24 \_235 °C 1509.9 В M Spec. C 207 5 AHc kcal/m Ultra V ΔHf A\* | 72 to 1.38832 5 X-Ray Dif. ΔFf B\* 199 °C 1419.1 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in c Acetone to Carbon tet. •c Benzene 7.24360 A 25 to 5 Ether B١ 1711.2 \_ 72 °C 5 n-Heptane 225. 5 to Ethanol °C 25 to Water A'\* 1.72596 Water in B'\* 72 °C 1610.9 (BV) to Ac | 235 to 7.80480 (AV) °C Bc \_tc\_ 2426.8 ٠c 5 c<sub>p</sub> liq. °K Cc 325.7 5 Cryos. A° c<sub>p</sub> vap. °K consts. B° c, vap. 188.95 5  $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 14	19		
NAME	2, 2, 3, 3, 4	-Per	ntamethylpentan	e		STRUCTURAL FORMULA				
						сн,сн,				
	TTT		Т		$\dashv$	сн <sub>3</sub> с с с	сн сн.			
Mole		lecul	ar C <sub>10</sub> H <sub>22</sub>	Molecular	.	ċн <sub>3</sub> ċн <sub>3</sub> ċ				
% Pur.	For	mula	1022	Weight 142.2		C113C113C	3			
	1	Ref.		<b></b>	Ref.			Ref.		
F.P. °C	-36.45	2	dt/dP			f to				
F.P. 100%	•		°C/mm 25°C	5.5392	5	g '° <u>K</u>				
B. P. °C 760 mm	166.05	2	BP	0.05328	5	h				
100	98.72	4	t <sub>e</sub>	0.03759	5	f' to				
30	69.13	4	30 mm	0.7402	5	g' :° <u>K</u>				
10	46.60 9.01	5	∆Hm cal/g		T 7	h'				
Pressure	<del>  /. 01</del>	Ť	ΔHv cal/g			m to		1		
mm 25°C	2.8978	5	25°C	77.37	5	n   <u>*K</u>				
t <sub>e</sub>	1190.4	5	30 mm BP	73.71 62.70	5			↓		
Density			t_	60.63	5	m' to				
g/ml 20°C	0.78009 0.77675	2	t <sub>e</sub> (d, e)	60.52	5	,				
d <sub>4</sub> 25 30	0.7734	4	AHv/T <sub>e</sub>	18.82	5	<u> </u>		+		
a	0.7934	4	d 69 to	81.56	5	Surface tension dynes/cm. 20°C	29.26	5		
Ъ	-0.0367	4	e 185 °C	0.1136	5	8 30	28.27	5		
Ref. Index			e' 69 °C	0.0831	5	40	27.30	5		
<sup>n</sup> D 20°C		2	d <sub>c</sub> g/ml	0.249	5	Parachor [P]				
30	1.43412 1.43210	2	v ml/g	4.012	5	20°C 30				
"C"	0.7417	4	tc °C	346.	5	40				
MR (Obs.)		2	P <sub>c</sub> mm	16236.	5	Sugd.	424.2	5		
MR (Calc.		5	PV/RT	1 0000	5	Exp. L.l.%/wt.				
(nD-d/2)	1.04602	2	25°C 30 mm	1.0000	5	u. Dispersion	97.	2		
Dielectric	2.062	5	BP	0.9480	5	Flash Point C	···	┿		
A 69 to		5	ţ <sub>e</sub>	0.9335	5	Fire Point				
B 1235 °C	1496.1	5	tc  ΔHc kcal/m	0.24		M. Spec.		1		
A* 69 to		5	ΔHc kcai/iii			Ultra V.				
B*  195 °C		5	ΔFf			X-Ray Dif. Infrared				
к	-		Viscosity			Solubility in +		+-		
	-		centistokes 7°C	!		Acetone				
t <sub>x</sub> °C			<sup>7</sup>			Carbon tet.				
A'  25 to		5				Benzene Ether				
B' 1_69 °C		5	B <sub>v</sub> to			n-Heptane				
C'	225.	-	B to			Ethanol Water				
A'* 25 to B'* 69 °C		5	(B <sup>V</sup> ) - to	·		Water in		1		
Ac   235 to		5	(A <sup>V</sup> )  °C					T		
Bc tc °C	2410.3	5	<del>                                     </del>	<del> </del>						
Cc	326.8	5	c <sub>p</sub> liq. °K							
Cryos. A°			c <sub>p</sub> vap. °K							
consts. B°	+	L	c <sub>v</sub> vap.							
t <sub>e</sub> °C	185.15	5		L	L	L	l			
$T_{\mathbf{R}} = 0.8$						grams/100 gra		nt		
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula			
SOURCE:		Al								
PURIFICA	TION:	A.	PI							
LITERATU	RE REFERE	NCE	5:							
L										

No. 150 NAME 2, 2, 3, 4, 4-Pentamethylpentane STRUCTURAL FORMULA сн3 CH<sub>3</sub> сн3с сн с CH3 Mole Ref. Molecular Molecular с҅н₃с҅н₃с҅н₃  $C_{10}H_{22}$ Weight 142.276 % Pur. Formula Ref. Ref Ref. •c -38, 75 2 dt/dP to F.P. 100% °C/mm <u>°K</u> g 25°C 4.1689 5 B.P. °C h ВP 0.05270 5 760 mm 159.29 2 te 0.03768 5 f to 100 92.74 4 g' °K 63.52 4 30 30 mm 5 0.7307 10 41.28 5 h' AHm cal/g 1 4.20 5 to m AHv cal/g Pressure °K n 25°C 75.32 mm 25°C 3.9551 5 0 5 5 30 mm 72.24 te 1170.4 5 BP 61.41 Density m to 59.45 5 te (d, e) g/ml 20°C °K n' 0.76703 2 59.36 5 0 25 0.76361 dt 2 AHV/Te 18.77 5 30 0.7602 4 Surface tension 79.43 64 to 5 0.7807 -0.0<sub>3</sub>68 27.35 4 dynes/cm. 20°C 178 °C 25 to 0.1131 Ъ 5 30 26.38 ď 77.32 5 40 25.44 5 Ref. Index e¹ 64 °C 0.0800 5 20°C 1.43069 [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor 5 d<sub>c</sub> g/ml 0.244 25 2 1.42868 20°C v<sub>c</sub> t<sub>c</sub> ml/g °C 5 4.102 30 1.42660 4 30 333 5 40 "C" 0.7455 4 P<sub>c</sub> mm 5 15544. Sugd. 424.2 5 MR (Obs.) 47.984 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 48.38 25°C 1.0000 5 1.04717 (nD-d/2)2 30 mm 1.0000 5 Dispersion 98. 2 Dielectric 2.047 5 RP 0.9475 Flash Point °C A 64 to 0.9335 6.8712 5 Fire Point t<sub>c</sub> 5 1468.6 0.24 В 1\_22<u>4 °C</u> c M Spec. 209. 5 AHc kcal/m Ultra V ΔHf A\* 64 to 1.37443 5 X-Ray Dif. ΔFf B+ 188 °C 1379.3 Infrared ĸ Viscosity Centistokes Solubility in c Acetone to •c Carbon tet. Benzene 25 to A۱ 7, 22785 5 Ether 1669.2 5 B <u>64</u> °C n-Heptane B<sup>V</sup> A<sup>V</sup> 227. 5 to Ethanol 1.71400 °C Water A'\* 25 to 5 B'\* 64 °C (BV) Water in 1569.1 5 to Ac | 224 to 7.70538 (A<sup>\*</sup>) | °C Bc tc\_ ۰c 2275.4 5 c<sub>p</sub> liq. ۰ĸ Сc 314.2 Cryos. A° c<sub>p</sub> vap. ۰ĸ consts. B° t<sub>e</sub> °C c<sub>v</sub> vap. 177.44 5  $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

										No. 15	1	
NAME		n-Ui	ndeca	ne			_	STRUCTURAL FORMULA				
Mole		Ref.	Mo	ecul	ar C H	Molecular			CH <sub>3</sub> (CH <sub>2</sub> )	<sub>9</sub> CH <sub>3</sub>		
% Pur.		لــــــــــــــــــــــــــــــــــــــ	For	mula	C <sub>11</sub> H <sub>24</sub>	Weight 156.2	92					
	-			Ref.		<b>↓</b>	Ref.		r		Ref.	
F. P. °C	,_	-25.59	4	2	dt/dP			f	to			
F.P. 1009	•				°C/mm 25°C	30.42	5	g	' <u>°K</u>			
B. P. °C 760 mm		195.89	0	2	BP	0.05356	5	h			L_	
100		127.94	3	2	t <sub>e</sub>	0.03641	5	f'	to			
30 10		97.90 74.97		4 4	30 mm	0.7526	5_	g'	° <u>K</u>			
1 i		36.60		5	ΔHm cal/g			h'	<u> </u>		<u> </u>	
Pressure					ΔHv cal/g		! !	m n	300 to	0.0255 0.0014	4	
mm 25°C		0.42	89	5	25°C 30 mm	86.4 77.55	2 5	0		-0.0647	4	
t <sub>e</sub>	1	244.8		5	BP	63.5	2	m'	700 to	<u> </u>	-	
Density g/ml 20°0	-	0.74	017	2	te te (d, e)	60.81	5	n'	1000 °K		4	
dt 25		0.73		2		60.46	5	0'		-0.0640	4	
<b>4</b> 30	$\perp$	0.73		4	ΔHv/T <sub>e</sub>	19.38	5	Sur	face tension		T	
a	į	0.75		4	d 98 to	91.59 0.1434	5		es/cm. 20°C	24.74	2	
Ь	+	-0.03	724	4	d 25 to	89.44	5	8	30 <b>4</b> 0	23.80 22.88	2 2	
Ref. Index		1.41	716	2	e'   98 °C	0.1214	5	P-	achor [P]	22.00	-	
45		1.41		2	d <sub>c</sub> g/ml	0.237	2	Par		471.0	4	
30	$\perp$	1.41	284	4	vc ml/g tc °C	4.223 367.	2 2		30	471.0	4	
"C"		0.74	97	4	P <sub>c</sub> mm	14592.	2		40 Suød.	471.0 463.2	4 5	
MR (Obs.)		53.12		2	PV/RT		H	Evr	L.1.%/wt.	103.2	+-	
MR (Calc. (nD-d/2)	"	52.99 1.04		5 2	25°C	1.0000	5		u.		ł	
Dielectric	+	2. 01		5	30 mm BP	1.0000 0.9305	5 5		persion			
A 98 to		6.97		2	t	0.9126	5		sh Point C		1	
B   258 °C		572.47		2	t <sup>e</sup> <sub>c</sub>	0.242	2		e Point		<b> </b> -	
c		188.02	2	2	ΔHc kcal/m ΔHf	1663.55	2		Spec. ra V.		1	
A*  98 to B*  227 °C	- 1	1.53	000	5	ΔFf			X-F	Ray Dif.		ļ	
к 227 5	- 1	493,5		5	Viscosity		$\Box$		ared		ļ	
·	_				centistokes				ability in Tetone			
t <sub>k</sub> to					7 110 °C	0.5977	2 2	Ca	rbon tet.	{		
A'  25 to		7.32	25	5	150	0.4522	2		nzene her			
B'   98 °C		776.4		5	170	0,4006	2		Heptane		ŀ	
C'	_	206.		5	B <sup>V</sup>   100 to	491.2	4	Et	hanol		1	
A'* 25 to B'* 98 °C	- 1	1.85 685.0	11	5	⊢.=v.— — — -	₹. 49440	4		ater ater in			
Acl 258 to	-	7.93	52	5	l 😈					<b> </b>	<u> </u>	
Bc tc °C		7.93 543.8	J <u>L</u>	5			╁					
Cc		309.3		5	c <sub>p</sub> liq. °K							
Cryos, A° consts, B°					c <sub>p</sub> vap. °K							
t <sub>e</sub> °C		217.09		5	c <sub>w</sub> vap.							
$T_R = 0.8$	3 T	c					اــــــا	+ g1	ams/100 gra	ms solven	t t	
REFEREN			ow	2-A	PI 3-Lit. 4-	Calc. from de	t. da		-Calc. by for			
SOURCE:				AF	PI							
PURIFICA	TIO	N:		AF	PI							
LITERATU	IRE	REF	ERE	NCES	S:							
1												
L												

No. 152 NAME n-Dodecane STRUCTURAL FORMULA CH3(CH2)10CH3 Mole Molecular Molecular  $C_{12}H_{26}$ Weight 170.328 % Pur Formula Ref. Ref Ref F.P. °C 2 -9.587 dt/dP f to \*C/mm 25\*C F.P. 100% °K g 89.64 B. P. \*C h BP 0.05528 760 mm 216, 278 2 t<sub>e</sub> 0.03639 5 ſ١ to 146.142 100 2 g' °K 30 115.1 4 30 mm 0.7771 5 10 91.45 5 h' ∆Hm cal/g 51.69 31 51.84 5 300 to 0.0234 m AHv cal/g Pressure \_600 °K 0.0013 4 n 25°C 86.0 mm 25°C 0.1320 5 -0.0643 4 0 30 mm 75.46 1297.3 5 ŧ, BP 2 61.3 700 to 0.1115 m' Density te (d, e) 58.38 5 5 1000 °K 0.0012 g/ml 20°C 0.74869 57.97 -0.0640 4 o' 25 0.74516 d4 19.37 5 AHv/T 30 0.74163 4 Surface tension 1 150 to 91.58 5 0.76281 4 dynes/cm. 20°C 25.44 0.1400 5 230 ℃ Ъ -0.0<sub>3</sub>706 30 4 24.51 2 ď٠ 25 88.92 5 to 40 23.60 2 Ref. Index e' 150 °C 0.1170 5 n<sub>D</sub> 20°C 1.42160 [P] 2 Parachor 0.237 2 d<sub>c</sub> g/ml 25 510.9 1.41949 2 20°C 4 v<sub>c</sub> ml/g 2 4.215 30 1.41735 4 30 511.1 4 386. t<sub>c</sub>. 511.1 4 40 "C" 0.7486 4 P<sub>c</sub> mm 2 13604 Sugd. 502.2 5 MR (Obs.) 57.76 5 PV/RT Exp. L.1.%/wt. MR (Calc.) 57.616 2 25°C 1.0000 (nD-d/2) 1.04725 5 30 mm 1.0000 5 Dispersion Dielectric 5 2.02 BP 0.9283 5 Flash Point °C 165. 3 0.9084 150 to 6.98059 2 Fire Point t<sub>c</sub> 0.237 2 В L\_280 °C 1625.928 2 M Spec. ċ 180.311 2 AHc kcal/m 1810,48 Ultra V ΔHf A\* | 150 to 1.5604 5 X-Ray Dif. ΔFf B+ 250°C 1546.9 Infrared K Viscosity Solubility in c centistokes Acetone to 150 °C 0.5128 2 Carbon tet. •c 170 0.4516 2 Benzene 190 0.4017 25 to 7.3157 5 Ether 0.3597 2 1830.0 210 Вı 150 °C n-Heptane 5 198,3 140 to 529.0 Ethanol AV i 220 °C **Z.** 46116 A'\* Water 25 to 1.8787 (BV) Water in B'\* 150 °C 5 1740.9 to Ac | 280 to 8.06653 (A<sup>V</sup>) °C 2780.0 Bc tc\_°C 5 cp liq. ۰ĸ Сс 322.0 5 Cryos. Aº c<sub>p</sub> vap. °K consts. B° te °C c<sub>v</sub> vap. 240.09  $T_{R} = 0.84 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES: 3 NFPA 325; 3' JACS 76, 333 (1954) Linke et al.

No. 153 n-Tridecane STRUCTURAL FORMULA NAME  $CH_3(CH_2)_{11}CH_3$ Mole Ref. Molecular Molecular Weight 184.354  $C_{13}H_{28}$ % Pur Formula Ref. Ref. Ref. F.P. °C -5.392 2 dt/dP f to F.P. 100% °C/mm ٥K g 25°C 249.27 5 B. P. °C h ВP 0.0568 760 mm 2**3**5.44 2 t<sub>e</sub> 0.03648 5 f١ to 100 163.3 2 <u>• к</u> g' 30 0.7994 5 131.45 4 30 mm 10 107.09 5 h! ∆Hm cal/g 5 1 66.35 300 to 0.0241 m ∆Hv cal/g Pressure 600 °K 0.0014 n 25°C 85.9 mm 25°C 0.03972 0 -0.0<sub>6</sub>48 4 30 mm 73.60 t<sub>e</sub> 1339.0 5 BP 59.1 2 700 to m' 0.1133 4 Density 55.97 5 te te (d, e) n' 1000 °K 0.0012 g/ml 20°C 0.7564 2 55.47 5 ۰, -0.0<sub>6</sub>39 4 25 0.7528  $\mathbf{d_{4}^{t}}$ ΔHv/T<sub>e</sub> 5 19.30 30 0.7492 2 Surface tension 131 to 91.92 5 0.7708 4 26.1 2 dynes/cm. 20°C 261 °C 0.1394 5 ъ -0.0372 4 25.2 2 30 ď٠ 25 to 88.78 24.3 2 40 Ref. Index e¹ 131 °C 0.1153 5 20°C 1.4256 [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 0.24 2 25 1.4234 2 550.9 4 20°C vc ml/g t °C 2 4.231 30 1.4213 4 30 551.3 4 <sup>t</sup>c 404. 2 551.6 4 40 "C" 0.7476 4 P<sub>c</sub> mm 12920. 2 541.2 5 Sugd. MR (Obs.) 62.40 5 PV/RT Exp. L. l. %/wt. 2 MR (Calc.) 62.234 25°C 1.0000 5 (nD-d/2)1.0474 5 30 mm 5 Dispersion 1.0000 Dielectric 2.03 5 BP 0.92225 5 Flash Point C 0.9001 5 A 131 to 6.9887 1677.43 2 Fire Point 2 ž 0.23 B | 302 °C M. Spec. Ultra V. С 172.90 AHc kcal/m 1957.40 2 ΔHf A\* 131 to 1.6005 5 X-Ray Dif. ΔFf B\*[ 271 °C 1600.8 Infrared Viscosity Solubility in centistokes Acetone 0.5043 to 170 °C 2  $\mathbf{t_k}$ Carbon tet. 0.4468 °C 190 2 Benzene 0.3987 2 210 A' | 25 to 7.3147 Ether 230 0.3577 2 В' 131 °C 1881.7 5 n-Heptane B<sup>v</sup> 160 C' 5 4 190.9 558.8 Ethanol V |240 °C **Z**. 44372 Water A'\* 25 to B'\* 131 °C 1.9095 5 (BV) Water in 1795.0 5 to Acl 302 to (A<sup>V</sup>)| 8.1985 5 °C Bc tc °C 3013.2 5 c<sub>p</sub> liq. ۰ĸ Cc 333.4 c<sub>p</sub> vap. Cryos. A° °K consts. B° c vap. te °C 261.5 5  $T_R = 0.85 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula

LITERATURE REFERENCES:

SOURCE:

PURIFICATION:

API

API

No. 154 n-Tetradecane STRUCTURAL FORMULA NAME  $CH_{3}(CH_{2})_{12}CH_{3}$ Mole Molecular Molecular C14H30 Weight 198.380 % Pur Formula Ref Ref. Ref. •c 5.863 2 dt/dP f to \*C/mm 25\*C F.P. 100% g °K 711.4 5 B. P. \*C h RP 0.0582 2 760 mm 253.57 2 t<sub>e</sub> 0.03632 5 ſ١ to 179.6 100 2 g¹ °K 30 146.91 4 30 mm 0.8202 5 10 121,92 5 h١ ∆Hm cal/g 1 80.13 5 300 to 0.0242 AHv cal/g m Pressure \_600 °K 0.0014 25°C n 85.7 2 5 mm 25°C 0.01172 -0.0<sub>6</sub>48 4 o 30 mm 71.85 1390.0 5 t<sub>e</sub> BP 57.5 2 700 to 0.1121 4 Density 54. [9 5 te (d, e) 11000 °K 0.0012 n' g/ml 20°C 0.7628 2 53.66 5 -0.0640 01 4 25 0.7593 dt 2 AHv/T 5 19.36 30 0.7558 4 Surface tension 0.7768 -0.0<sub>3</sub>70 ī 147 to 91.62 5 a b 44 dynes/cm. 20°C 26.6 2 <u> 282</u> 5 e d' °C 0.1346 2 30 25.7 to 88.94 5 5 ı 25 40 24.8 2 e¹ Ref. Index °C 147 0.1298  $\mathbf{n}_{\mathbf{D}}$ 20°C 1.4289 [P] 2 Parachor 0.24 d<sub>c</sub> g/ml 2 25 1.4268 2 20°C 590.6 v<sub>c</sub> ml/g \*C 4.184 2 30 1.4247 4 30 590.9 4  $t_c$ 422. 2 4 40 591.2 "C" 0.7467 4 Pc 5 2 580.2 mm 12160. Sugd. MR (Obs.) 67.03 5 PV/RT Exp. L.1.%/wt. MR (Calc.) 66.852 2 25°C 1.0000 5 u. (nD-d/2) 1.0475 5 30 mm 1.0000 5 Dispersion Dielectric 5 2.04 BP 0.9232 5 Flash Point °C 0.8996 147 to 6.9957 2 Fire Point t<sub>c</sub> 0.23 2 В \_325 °C 1725.46 M Spec. C 165.75 2 AHc kcal/m 2104.32 2 Ultra V. ΔHf A\* | 147 to 1.6252 5 X-Ray Dif. ΔFf B\* 292 °C 1647.4 Infrared Viscosity ĸ Solubility in c centistoke s Acetone to 190 °C 0.4946 2 Carbon tet. °C 2 210 0.4402 Benzene 0.3941 2 7.3143 230 25 to 5 Ether 2 250 0.3552 В' 1930.4 \_1<u>47 °C</u> n-Heptane B<sup>V</sup> A<sup>V</sup> 180 to 183.8 5 588.9 4 Ethanol Z. 42490 4 260 °C Water A'\* 25 to 1.9386 B'\* 147 °C (BV) Water in 1846.1 to Ac | 325 to 8.36982 5 (A<sup>V</sup>)1 °C Bc tc\_ °C 3312.8 liq. Cc 352.5 Cryos. A° °K cp vap. consts. B° te °C Vap. 282.11  $T_R = 0.86 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 155 n-Pentadecane STRUCTURAL FORMULA NAME  $CH_{3}(CH_{2})_{13}CH_{3}$ Molecular C<sub>15</sub>H<sub>32</sub> Mole Ref. Molecular % Pur Weight 212.406 Ref. Ref. F.P. °C F.P. 100% 9.926 2 dt/dP to °C/mm ۰ĸ g 35521. 25°C B.P. °C h 0.0595 ΒP 2 270.63 760 mm 2 5 ſ١ t<sub>e</sub> 0.03642 to 100 195.0 2 g' <u>°К</u> 30 4 30 mm 0.8390 5 161.57 10 5 136, 01 h١ ∆Hm cal/g 93.26 5 300 to 0.0239 m AHv cal/g Pressure n 600 °K 0.0014 25°C 85.7 2 mm 25°C 0.00233 5 ٥ -0.0648 30 mm 5 70. 26 te 1424.1 5 BP 55.6 2 m 700 to 0.1156 Density te (d, e) 52. 12 5 n' 1000 °K 0.0012 g/ml 20°C 0.7685 0.7650 5 2 51.51 o! -0.0639 4 25  $\mathbf{d_4^t}$ ΔHv/T<sub>e</sub> 19.28 5 30 0.7615 4 Surface tension 160 91.99 d to 5 0.7825 4 dynes/cm. 20°C 27.1 2 1 290 °C 0.1344 ь -0.0370 4 30 26.2 2 آاتa 25 86.18 to 2 40 25.3 Ref. Index e' | 160 5 0.1130 20°C 1.4319 2 Parachor [P] <sup>n</sup>D d<sub>c</sub> g/ml 0.24 2 25 1,4298 2 20°C 630.6 vc ml/g 4.190 2 30 631.1 1.4277 4 30 4  $^{\mathbf{t}}_{\mathbf{c}}$ 437. 2 40 "C" 631.4 4 0.7461 4 11400. 2  $P_c$  mm Sugd. 619.2 5 MR (Obs.) 71.67 5 PV/RT Exp. L. 1. %/wt. MR (Calc.) 71.47 25°C 1.0000 (nD-d/2)1.0476 5 30 mm 1.0000 5 Dispersion 2.05 Dielectric 5 BP 0.9170 5 Flash Point C te tc 0.8914 A 160 to 7,0017 2 Fire Point 0.23 2 B 1338 °C 1768.82 2 M. Spec. C 158.60 2 ∆Hc kcal/m 2251, 24 Ultra V ΔHf A\* 160 to 1.6613 5 X-Ray Dif. ΔFf B\* 310 °C 1693.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to 210 °C 0.4819 2 Carbon tet. °C t<sub>x</sub> 230 0.4308 2 Benzene 250 0.3878 2 25 to 7.3123 Ether 270 0.3510 В' 160 °C 1973.3 5 n-Heptane B<sup>V</sup> | 200 to A<sup>V</sup> | 280 °C T200 to 176.6 5 608.0 Ethanol Z. 42601 A'\* 25 to 4 Water 1.9646 5 B'\*160 °C Water in (BV) 1891.6 5 to Acl 338 to 8.5317 5 (A V) 3600.5 Bc tc °C 5 c<sub>p</sub> liq. ۰ĸ 5 Cc 369.1 c<sub>p</sub> vap. Cryos. Aº consts. B° c vap. te °C 301.06 5  $T_R = 0.86 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

								<b>No.</b> 156	6
NAME	n-He	exade	cane	······································		T	STRUCTURAL		
		1				$\dashv$	сн <sub>3</sub> (сн <sub>2</sub>	) <sub>14</sub> CH <sub>3</sub>	
Mole % Pur.	Ref.		lecul rmul		Molecular Weight 226.4	32			
			Ref.			Ref			Ref.
F.P. °C	18.1	65	2	dt/dP	T		f   to		
F.P. 100%				*C/mm			g   <u>.</u>		
B. P. °C	20/ 7	•		71.87°C BP	127.43	5 2	h ;		
760 mm 100	286.7 209.5		2 2	t <sub>e</sub>	0.03617		f' to		
30 10	175.2		4	30 mm	0.8596	5	g'   ' <u>°</u> I	<b>'</b>	l
l i	149.0		5	ΔHm cal/g			h'		ļ.,
Pressure				ΔHv cal/g	70.20	ا ۔ ا	m   300 to		
mm 71.8	7 0.0 1474.2	0105	5	71.87°C <b>30 mm</b>	79.39 68.43	5 5	•	-0.0648	4
Density	13/1.2		-	BP	5 <b>4.3</b> 50.76	2 5	m'   700 to	0.1230	4
g/ml 20°C		7344	2	te (d, e)	50.13	5	n'   11000 º1		
dt 25		6996 6648	2	ΔHv/T	19.38	5	0 1	-0.0640	-
1 30	0.7		4	d   175 to		5	Surface tension dynes/cm, 20°0		2
ь	-0.0		4	d 310 to		5	<b>y</b> 30	26.7	2
Ref. Index	1			e' 175 °C		5	40	25, 8	2
n <sub>D</sub> 20°C		3453 3250	2 2	d g/ml vc ml/g	0.24	2	Parachor [P]	671.0	4
30		3036	4	tc *C	4.195 452.	2 2	30	671.5	4
"C"	0.7	456	4	P <sub>c</sub> mm	10640.	2	40 Sug	672.0 d. 658.2	4 5
MR (Obs.) MR (Calc.			5	PV/RT	<del>                                     </del>	$\vdash$	Exp. L.1.%/wt		<del>                                     </del>
(nD-d/2)		88 4781	2 5	71.87°C	1.0000	5	u.		ĺ
Dielectric	2.0	6	5	30 mm BP	1.0000	5	Dispersion Flash Point °C	<del> </del>	├
A 175 to		3044	2	ţ.	0.8937 0.22	5 2	Fire Point		1
B 1_358 °C	1831.3		2	t <sub>c</sub> AHc kcal/m	2398, 17	2	M Spec.		
A*   175 to	+	0090	5	ΔHf	2370.11	ا آ ا	Ultra V. X-Ray Dif.		1
B* 330°C			5	ΔFf	-	$\vdash$	Infrared		
K C				Viscosity centistokes		1 1	Solubility in	•	
the Total				7 225 °C		2	Acetone Carbon tet.		
t 0 0		3309	5	245 265	0.4322	2 2	Benzene		
B' 175 °C	2036.4		5	285	0.3540	2	Ether n-Heptane		
C'	172.5		5	B <sup>V</sup>   220 to A <sup>V</sup>   290 °C	626.8 7.42611	4 4	Ethanol Water		
A'* 70 to B'* 175 °C		7708	5		- 2.42011	*	Water in		
Ac   358 to		725	5	(A <sup>V</sup> )  to	1				
Bc t °C	4015.9		5	c <sub>p</sub> liq. °K	<del></del>	$\vdash$	1		
Cryos. A°	397.6		5	ll .					
consts. B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	319.7		5	c <sub>v</sub> vap.					L
$T_R = 0.8$							f grams/100 gr	ams solven	t
REFEREN	CES: 1-I	ow			Calc, from det	t. dat	ta 5-Calc. by fo	rmula	
SOURCE:			Al	<del></del>					
PURIFICAT		-	Al						
LITERATU	KE REF	ERE	NCES	<b>5:</b>					

								No. 15	7
NAME	n-Heptad	ecane	•			STI	RUCTURAL	FORMUL.	A
					$\neg$		CH <sub>3</sub> (CH <sub>2</sub> )	5CH3	
Mole % Pur. 99	.0#   Ref.   Mo	lecul		Molecular Veight 240.45	8				
70 1 411. 77	. 0 1 3 1 2 0	Ref		l cignit stores	Ref.	T			Ref.
F.P. °C	21.980	2	dt/dP			f	to		
F.P. 1009			°C/mm		li	gl	°K		
B. P. °C			83,5°C <b>BP</b>	129.08 0.0619	5 2	h			
760 mm	301.82 223.2	2 2	te	0.03618	5	f¹	to		
30	188.36	4	30 mm	0.8733	5	g'	<u>*K</u>		
10 1	161.75 117.26	5	∆Hm cal/g			h'			
Pressure		1	ΔHv cal/g			m	300 to	0.0238 0.0014	4
mm 83.5	1	5	83.5°C 30 mm	81.28 67.20	5	0		-0.0 <sub>6</sub> 48	4
Density	1510.3	5	BP	52.8	2	m'	700 to	0.1161	4
g/ml 20°0	0.7780≠	2	t <sub>e</sub> (d, e)	49.09 48.38	5	n' o'	1000 •K	0.0012	4
dt 25	0.7745 0.7710	2	AHv/Te	19.36	5	0.		-0.0 <sub>6</sub> 39	4
a 30	0.7710	4	d 188 to	91.10	5		ace tension	28. 0 <sup>‡</sup>	,
ь	-0.0370	4	_e 337 °C d' 85 to	0.1269 88.3	5	8 gane	s/cm. 20°C 30	27.1	2
Ref. Index			e' 188 °C	0.1343	5	<u> </u>	40	26.2	2
<sup>n</sup> D 20°0	1.4348	2 2	d <sub>c</sub> g/ml	0.24	2	Para	achor [P] 20°C	711.0	4
70	1.4170	3	t <sub>c</sub> ml/g	4.159 462.	2 2		30	711.6	4
"C"	0.7450	4	P <sub>c</sub> mm	9880.	2			712.1 697.2	4 5
MR (Obs. MR (Calc.		2 5	PV/RT			Exp.	L. 1. %/wt.		
(nD-d/2)	1.0479	2	83.5°C 30 mm	1.0000 1.0000	5		u. ersion		
Dielectric		5	BP	0.9176	5		h Point C		
A 188 to		2	te t <sub>C</sub>	0.8902 0.22	5 2		Point		
B 1_374 °C	1847.82 145.52	2	ΔHc kcal/m	2545.09	2		Spec.		
A* 188 to		5	ΔHf ΔFf		-	Ultr X-R	a v. ay Dif.		
B* 347 °C	1771.33	5	Viscosity		$\vdash$	Infra			L
c	_	j	centistokes				bility in † etone		
t <sub>k</sub> to			<b>7</b> 240 ℃ 260	0.4804 0.4330	2 2	Ca	rbon tet.		
A'   85 to	7, 3095	5	280	0.3920	2	Ber Eth	nzene ner		
B' 188 °C		5	Bv 230 to	0.3550	2	n-F	leptane		
A'* 85 to	163.5	5	B <sup>V</sup>   230 to A <sup>V</sup>   310 °C	659.1 2.40037	4	Eth Wa	anol ter		
B'* 188 °C		5	(B <sup>V</sup> )  to				ter in		<u> </u>
Acl 374 to	8.9220	5	(A <sup>V</sup> ) °C						
Bc tc °C	414.4	5	c <sub>p</sub> liq. °K						
Cryos. A			c <sub>p</sub> vap. •K						
consts. B		_	11						
t <sub>e</sub> °C	336. 6	5	c <sub>v</sub> vap.	<u> </u>	L	L_			<u> </u>
T <sub>R</sub> = 0.8			for undercool				ams/100 gra		it
	CES: 1-DOW		PI 3-Lit. 4-0	Jaic. Irom de	t. da	ta 5-	Caic. by for	muia	
SOURCE: PURIFICA	TION		PI						
			S: 3 JACS 77 (2	019) 1955 Sch	0870	7 A A	et al		
	NE NEFERE	CE	3. 3 JACS 11 (2	.v.7] 1733 3CN	oere	. A.A.	. c. ai.		
# purity a	pplies to nD	°C							
	<u> </u>								

									No. 158	8
NAME	n-O	ctade	cane				STR	UCTURAL		
Mole % Pur. 99	. 9# Ref	. Mo	lecul		Molecular Weight 254.4	84		сн <sub>3</sub> (сн <sub>2</sub> ) <sub>1</sub>	6 <sup>CH</sup> 3	
			Ref.			Ref			· · · · · · · · · · · · · · · · · · ·	Ref.
F.P. °C	28.1	80	2	dt/dP			f	to		
F.P. 1009				°C/mm	04805 4	ا ا	g	<u>*</u> K		
B. P. °C 760 mm	316.1	,	2	25°C BP	94805.4 0.0630	5 2	h			<u>L.</u>
100	236.		2	t <sub>e</sub>	0.03608	5	f'	to		
30 10	200. 6 173. 5		4 5	30 mm	0.8890	5	g'			1
1	128.2		5	ΔHm cal/g	57.65	3	h'	1	0.000/	-
Pressure	†		$\vdash$	ΔHv cal/g	05.04		m n	300to   600°K	0.0236 0.0014	
mm 25°C	0.0	473	5 5	25°C 30 mm	85.3≠ 65.74	2 5	0		-0.0649	4
t <sub>e</sub>	1547.3	<u> </u>	13	BP	51.5	2	m'	700 <b>to</b>	0, 1163	4
Density g/ml 20°C	: 0.7	819 <sup>‡</sup>	2	te (d, e)	47.69 46.96	5	n'	1 <u>1</u> 000°K	0.0012	4
at 25	0.7	785≠	2	ΔHv/Te	19.38	5	0'	1	-0.0 <sub>6</sub> 39	4
		751	4	d   201 to	90.49	5		ace tension		
a b	-0.7	955 368	4 4	e   353 °C	0.1234	5	dyne	s/cm. 20°C 30	28.4 <sup>‡</sup> 27.5	2 2
Ref. Index		3	+	d'   25 to	88.09	5		40	26.6	2
n <sub>D</sub> 20°C	1.4	390 <sup>‡</sup>	2		0.1114	2	Para	chor [P]		
25 70		369 <sup>≠</sup>	2 3	d g/ml v ml/g t °C	4.322	2		20°C	751.3	4
"C"		191 '446	4	tc °C	477.	2		30 <b>4</b> 0	751.9 752.2	4
MR (Obs.)			5	P <sub>c</sub> mm	9880.	2		Sugd.	736.2	5
MR (Calc.			2	PV/RT		_	Exp.	L.1.%/wt.		
(nD-d/2)	1.0	480	5	25°C 30 mm	1.0000	5	Dist	u. ersion		l
Dielectric	2.0	)7	5	BP	0.9178	5		h Point °C		<del> </del>
A 201 to		156	2	te t <sub>c</sub>	0.8893 0.22	5 2		Point		l
B <u>  387</u> • C	1883.7		2 2	ΔHc kcal/m	2692.01	2	M S			
A*   201 to B* 363 °C	1.7	241	5	ΔHf ΔFf			Ultr X-R Infra	ay Dif.		
K	-			Viscosity				bility in +		╁
\$	-1		1 1	centistokes 7 190 °C	0.701	2		etone		1
-i <u>k</u>   •∂				210	0.616	2		rbon tet. nzene		
A'   25 to		094	5	230 250	0.547 0.490	2 2	Eth			1
B' 201 °C	2088.9		5 5	B <sup>V</sup>   180 to	628, 1	4		leptane anol		1
A'* 25 to		3584	5	A <sup>V</sup>   260 °C	Z. 48970	4	Wa			
B'* 201 °C			5	(BV) to	1		Wa	ter in		<u> </u>
Ac   387 to		876	5	(A <sup>V</sup> ) °C						
Bc tc_°C	449.5		5	c <sub>p</sub> liq. °K						
Cryos, A° consts, B°		<u> </u>		c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	353.0	)	5	c <sub>w</sub> vap.						
$T_{\mathbf{R}} = 0.8$	_L			for undercool	led liquid		+	ma/100 ===		<u> </u>
				PI 3-Lit. 4-0		de	gra	ms/100 gran	ns solven	<u> </u>
SOURCE:		- "-		PI		. ua		Care. by for		
PURIFICA	TION:			PI						
		FERE		: 3 JACS 77 (	2019) 1955 <b>S</b> c	hoer	er A.	A. et al.		
# purity a	pplies to	<sup>n</sup> D <sup>70</sup>	°C							

No. 159 n-Nonadecane NAME STRUCTURAL FORMULA  $CH_3(CH_2)_{17}CH_3$ Mole Ref Molecular Molecular  $C_{19}H_{40}$ % Pur. 99.5# Weight 268.510 3 Formula Ref. Ref. Ref. F.P. °C F.P. 100% 32.1 dt/dP f to °C/mm g °K 54.24°C 8682.84 5 B.P. °C h BP 0.0640 2 329.7 760 mm 2 0.03599 f١ 5 to 100 248. 2 <u>«к</u> g' 30 212.40 30 mm 0.9035 5 10 184.9 5 h! AHm cal/g 40.78 3 138.8 5 300 to 0.0237 m AHv cal/g Pressure°C 0.0014 n \_600**°K** 54. 24℃ 83,23 5 5 mm 54.24 0.001 5 -0.0649 0 4 30 mm 1586.4 64.38 5 te BP 50.3 700 to m' 0.1164 4 Density te te (d, e) 46.37 5 'n' 0.7855 1000 °K 0.1165 g/ml 20°C 2 5 45.64 ۰, -0.0639 4 dt 4 25 0.7821 2 ΔHv/T<sub>e</sub> 0.7787 19.40 5 30 4 Surface tension 212 89.88 5 to 28.7<sup>‡</sup> 27.8<sup>‡</sup> a 0.7991 4 dynes/cm. 20°C 2 <u> 369</u> °C 0.1201 5 -0.0<sub>3</sub>68 4 2 30 ă٦ to 89.7 5 40 26.9 2 Ref. Index e¹ 212 0.1192 5 20°C 1.4409 [P] <sup>n</sup>D dc g/ml vc ml/g tc °C Parachor 2 0.24 1.4388 2 25 20°C 791.2 4.097 70 1.4211 3 30 791.8 4 <sup>t</sup>c 487. 2 792.2 40 4 "C" 0.7442 4 P<sub>c</sub> mm 9120. 2 Sugd. 775.2 5 MR (Obs.) 90.25 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 89.942 54.24°C 1.0000 (nD-d/2) 1.0481 4 1.0000 30 mm 5 5 Dispersion Dielectric 2,08 5 BP 0.9184 Flash Point °C 0.8885 5 A | 212 to 7.0192 2 Fire Point 0.22 2 B | 403 °C 2 1917.0 M. Spec. Ultra V. C AHc kcal/m 2838, 94 133.5 2 2 ΔHf A# 212 to 1.7427 5 X-Ray Dif. ΔFf B\*| 379 °C 1839.43 Infrared Viscosity Solubility in centistokes Acetone 190 °C 0.757 2 Carbon tet. °C 210 0.662 2 Benzene 230 0,586 2 25 to 7.31561 Ether B' 250 0,524 2127.54 5 212 °C n-Heptane B. 200 to 5 152 641.7 4 Ethanol A 1 260 ℃ 2.49284 Water A'\* 25 to B'\* 212 °C 2.06337 5 Water in (B<sup>V</sup>) 2054.7 5 to Ac | 403 to 9.4412 5 (A<sup>V</sup>) °C Bc t<sub>c</sub>°C 5302. ۰ĸ cp liq. Cryos. A cp vap. °К consts. B° c, vap. te °C 368.49  $T_{R} = 0.89 T_{c}$ for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A.A. et al. # purity applies to nn 70°C

							<b>No.</b> 160	)
NAME	n-Eicosa	ne			_	STRUCTURAL	FORMULA	
Mole % Pur. 99.	9# Ref. Mo	olecul		Molecular Weight 282.5	36	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>18</sub>	сн <sub>3</sub>	
		Ref.			Ref			Ref
F.P. °C	36.8	2	dt/dP			f to		
F.P. 100%			°C/mm	2000 41	5	g <u>  </u>		
B. P. °C	242.7		61.35°C BP	8808.41 0.06492	5	h		
760 mm 100	342.7 260.	2 2	t <sub>e</sub>	0.03614	5	f' to		
30	223.61	4	30 mm	0.9174	5	g'   ' <u>°</u> K_		
10 1	195.7 148.9	5	ΔHm cal/g	59.11	3	h'		
Pressure	<del></del>	+ -	ΔHv cal/g			m   300 to	0.0236 0.0014	4
mm 61.35	0.001	5	61.35°C 30 mm	79.66 63.08	5 5	0   -000 11-	-0.0649	
t <sub>e</sub>	1608.	5	BP	48.8	2			├
Density g/ml 20°C	0.7887 <sup>‡</sup>	2	te (d. a)	44.80	5	m'   700 to n'   1000 °K	0.1162 0.0012	
	0.7853 <sup>‡</sup>	2	e (d, e)	43.99	5	0'	-0.0639	
d <sub>4</sub> 70	0.7550	3	ΔHv/T <sub>e</sub>	19.29	5	Surface tension		
a	0.8023	4	d 224 to	89.89 0.1199	5	dynes/cm. 20°C	29.0 <sup>#</sup>	2
ь	-0.0368	4	d'   61 to	85.93	5	30 40	28.1° 27.2	2
Ref. Index	1.4426	2	e'   224 °C	0.1022	5	Parachor [P]	27.2	<u> </u>
D 25	1.4405	2	d g/ml	0.24	2		831.3	4
70	1.4230	3	v <sup>c</sup> ml/g t <sub>c</sub> °C	4.247 502.	2 2	30	832.0	4
"C"	0.7439	4	Pcmm	8360.	2-	40 Sugd.	832.4 814.2	4 5
MR (Obs.)	94.90	4	PV/RT		$\vdash$	Exp. L.1.%/wt.		
MR (Calc.) (nD-d/2)	94.56	2	61.35°C	1.0000	5	u.		
Dielectric	2.08	5	30 mm BP	1.0000 0.9122	5 5	Dispersion		
A 224 to	<b></b>	2	te	0.8810	5	Flash Point °C Fire Point		
B 1417 °C	1948.7	2	t <sub>c</sub>	0.21	2	M Spec.	<del> </del>	<u> </u>
<u> </u>	127.8	2	ΔHc kcal/m ΔHf	2985.86	2	Ultra V.		
A* 224 to B* 393 °C	1.77166 1874.62	5	ΔFf	1		X-Ray Dif. Infrared		
K	1.0.1.02	1	Viscosity			<del></del>		-
t to	-		centistokes	0.014	١ , ا	Solubility in + Acetone		
t <sub>x</sub> to t <sub>x</sub> C			7 190 ℃ 210	0.814 0.710	2 2	Carbon tet.		
A'   25 to	7,30970	5	230	0,627	2	Benzene Ether		
B' 1224 °C		5	B <sup>V</sup>   180 to	0.560	2	n-Heptane		
	146.	5	B 180 to A 260 °C	651.4 2.50316	4 4	Ethanol Water		
A'* 25 to B'* 224 °C	2.07885 2085.2	5	(BV) to	1 2.50510	•	Water in	<u></u>	
Ac   417 to	<del></del>	5	(A <sup>V</sup> ) °C	1				
Bc t °C	5807.	5		<del> </del>				
Cc	513.	5	P					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	382,84	5	c <sub>v</sub> vap.	<u> </u>				<u></u>
$T_{R} = 0.89$			≠ for undercoo			grams/100 gran		ŧ _
	ES: 1-Dow			alc. from det	. da1	ta 5-Calc. by for	mula	
SOURCE:		AP	<del></del>					
PURIFICAT		AP						
LITERATUI	RE REFERE	NCES	5: 3 JACS 77 (	2019) 1955 <b>Sc</b> ł	noere	er A.A. et al.		
# purity app	olies to n <sub>D</sub> <sup>70</sup>	C						

										No. 16	)1
NAME		n-I	lenei	cosa	ne			ST	RUCTURAL	FORMUL	A
		<del></del> -							CH <sub>3</sub> (CH <sub>2</sub> ) <sub>1</sub>	9СН3	
Mole % Pur. 99	. 9 <sup>1</sup>	Ref.	Mo! For	ecula mula	ar C <sub>21</sub> H <sub>44</sub>	Molecular Veight 296.	562				
				Ref.			Ref.			***	Ref.
F.P. °C		40.5		2	dt/dP			f	to		
F.P. 100%	$\perp$				°C/mm	150,40	5	g	°K		
B. P. °C 760 mm	3	56.5		2	114.64°C. BP	0.0657	4	_h_	L <u> </u>		-
100 mm	2	71.52		4	t <sub>e</sub>	0.0360	5	f'	to		1
30		32.88 03.07		4 5	30 mm	0.9765	5	g'	*K		
10		52.94		5	ΔHm cal/g	38.44	3	h'			ļ
Pressure					∆Hv cal/g			m n	to °K		
mm114.6	4	0.1		5	114.64°C. 30 mm	68.7 58.50	5 5	0			
te_	16	07.4		5	BP	47.05	5	m'	**	····	-
Density g/ml 20°C		0.79	917 <sup>‡</sup>	2	te te (d, e)	43.69 43.30	5	n'	to to		ì
at 25		0.78	883₹	2				0'			
<sup>a</sup> 4 70	$\perp$	0.75	831	3	ΔHv/T <sub>e</sub>	19.33	5	Sur	face tension		
a		0.80		4	d   232 to	80.06 0.0926	5 5		es/cm. 20°C	26. 91 26. 00	5
b	+	-0,03	608	4	115 to	78.58	5	*	30 <b>4</b> 0	26.00 25.11	5
Ref. Index	:	1.44	41 <sup>‡</sup>	2	e'   232 °C	0.0862	5	Pa 7	achor [P]		+
25		1.44	120	2	d g/ml vc ml/g			Fa.	20°C		
70	$\bot$	1.42		3	t <sub>c</sub> °C				30 40		
"C"	_	0.74		4	P <sub>c</sub> mm				Sugd.	853.2	5
MR (Obs.) MR (Calc.		99.52		2 5	PV/RT		+	Exp	. L.1.%/wt.		
(nD-d/2)	'	1.04	183 <sup>≠</sup>	ž	1 14.64°C.	1.0000	5	-	u.	98. <sup>‡</sup>	2
Dielectric	+	2.09		5	30 mm BP	1.0000 0.8948	5 5		persion	98.	12
A 115 to B   492 °C	24	7.47	174	5	t t t	0.8624	5		sh Point °C e Point		
c '		71.5		5	ΔHc kcal/m		11		Spec. ra V.		
A* 115 to	Τ.,	2,24	451	5	ΔHf ΔFf			X-F	Ray Dif.		
B*[410°C	- 23	43.		ا	Viscosity		+	Infr	ared		
c					centistokes				ability in +		
t <sub>k</sub> to					η °C			Ca	etone rbon tet. nzene		
A'  to	Т						1 1	Et	her		
B'°	-				B <sup>v</sup> to		1		Heptane hanol		
A'* 110 to	+	2,15	834	5	B <sup>V</sup>   to A <sup>V</sup>   °C			Wa	ater		
B'* 233 ℃	23	09.1		5	(B <sup>V</sup> )			W <sub>4</sub>	ter in		<u> </u>
Acl to					(A <sup>V</sup> )						
Bc t <sub>c</sub> °C	1				c <sub>p</sub> liq. °K						
Cryos, A° consts. B°					c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	3	96.79	,	5	c <sub>w</sub> vap.					1	
# for unde	rco	oled l	iquid	•		<u> </u>	لــــــــــــــــــــــــــــــــــــــ	+ g1	ams/100 gra	ms solver	nt
REFEREN				2-A	PI 3-Lit. 4-0	Calc, from de	et. da		-Calc. by for		
SOURCE:			ΡI								
PURIFICA'	CION	1: A	ΡI								
LITERATU	RE	REF	ERE	NCES	S: 3 JACS 77 (2	019) 1955 <b>S</b> ci	hoere	r A. A	. et al.		
l purity app	ies	to n <sub>D</sub>	7 0°C	and	ı d <sup>t</sup> 4						

						No. 162	:
NAME	n-Do	osane	>			STRUCTURAL FORMULA	1
Mole % Pur. 99.	71 Ref. M	olecul ormul	ar C <sub>22</sub> H <sub>46</sub>	Molecular Weight 310.5	88	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>20</sub> CH <sub>3</sub>	
		Ref.			Ref		Ref.
F.P. °C	44.4	2	dt/dP			f to	
F.P. 100%			*C/mm		_	g*K_	
B. P. °C	T		122.90°C. BP	149.36 0.0666	5 4	h	
760 mm 100	368.6 282.39	2	t <sub>e</sub>	0.0359	5	f' to	
30	243.15	4	30 mm	0.9920	5	g'   ' <u>°</u> K_	
10	212.86	5	ΔHm cal/g	37,67	3	h'	l
1	161.88	5	ΔHv cal/g	<del> </del>	$\vdash$	m to	
Pressure mm122.9	0.1	5	122.90°C.	67.19	5	n °K	ĺ
t <sub>e</sub>	1637.4	5	30 mm	57.28	5	0	
Density	<del>                                     </del>		BP t <sub>e</sub>	46.00 42.60	5	m'   to	
g/ml 20°C	0.7944	2	te (d, e)	43.38	5	n'   °K	1
dt 25 4 70	0.7910 <sup>7</sup> 0.7631 <sup>1</sup>	2	AHV/Te	19.82	5	1	
		_	d   242 to	79.14	5	Surface tension	¦ _
a b	0.8080 -0.0368	4	e 1410 °C	0.0899	5	dynes/cm. 20°C 27.12 30 26.20	5
Ref. Index	<del></del>		d'   110 to		5	40 25.31	5
n <sub>D</sub> 20°C	1 4455	2	<del></del>	0.0024	+-	Parachor [P]	
25 70	1.4435	12	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	į.		20°C	
	1.4260 <sup>1</sup>		tc°°C	-	1	30 40	1
"C"	0.7431	4	P <sub>c</sub> mm	į		Sugd. 892.2	5
MR (Obs.) MR (Calc.		2	PV/RT			Exp. L.1.%/wt.	
(nD-d/2)	1.0483	5 2	122.90°C.	1.0000	5	u. #	_
Dielectric	2,09	5	30 mm BP	1.0000 0.8943	5		2
A   242 to			te	0.8609	5	Flash Point °C	ļ
B 499°C		5	tc			Fire Point	
с	169.3	5	∆Hc kcal/m			M Spec. Ultra V.	l
A*   242 to			ΔHf ΔFf		1	X-Ray Dif.	l
B* 420 °C	2401.2	5	Viscosity	<b></b>	+	Infrared	<u> </u>
c	1		centistokes	į.	]	Solubility in +	}
k to			<b>∥</b> 7 ° c			Acetone Carbon tet.	l
x	<u> </u>	-		į	1	Benzene	l
A'   B'						Ether n-Heptane	l
C,	1		B <sup>V</sup> l to			Ethanol	
A'* 110 to		5	AV I °C			Water	ļ
B'* 242 °C	2367.2	5	(B <sup>V</sup> ) to		1	Water in	
Ac  to			(A <sup>V</sup> )  °C		1	1	ŀ
Bc tc_°C	<u>-</u>		c <sub>p</sub> liq. °K			1	
Cryos. A°	1	+	F				ł
consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	410.47	5	c <sub>w</sub> vap.				
	ercooled liqu		и	·	1	grams/100 grams solven	<u>+</u>
			PI 3-Lit. 4-0	Calc. from de	t. de	ta 5-Calc. by formula	•
SOURCE:		PI	1	<u></u>	ua	J-Care. by Ionniua	
PURIFICAT		PI	<del>-</del>				
			. 2 74.00 == :	2010) 1077 -			
LILERATU	AL KEFEKE	MCES	3: 3 JACS 77 (	2019) 1955 Sc	hoere	er A. A. et al.	
1							
purity app	olies to n <sub>D</sub> <sup>70</sup>	C. an	dd4				

No. 163 NAME n-Tricosane STRUCTURAL FORMULA  $CH_{3}(CH_{2})_{21}CH_{3}$ Molecular C23H48 Mole Ref. Molecular % Pur. 99.61 Weight 324.614 Ref. Ref. Ref. F.P. °C F.P. 100% 47.6 2 dt/dP f to 130.85°C. 25°C g ١ °K 151.91 B. P. °C h 0.0675 BP 2 760 mm 380, 2 2 0.0358 5 ſ١ to 100 292.84 4 g' °К 30 1.0068 5 253.03 4 30 mm 10 222,27 5 h' 39.74 ∆Hm cal/g 3 170.48 5 m to ∆Hv cal/g Pressure°C mm130.85 ۰ĸ 65.77 n 130.85°C. 5 0.1 5 o 30 mm 56.07 5 1666.0 te 5 BP 44.99 5 m' | to Density 41.56 te te (d, e) 5 0.7969# n' °K g/ml 20°C 5 42.38 o¹  $\mathbf{d_{4}^{t}}$ 25 0.7935<sup>7</sup> 0.7641 AHv/T 19.83 5 70 d 252 Surface tension to 78.11 5 a 0.8105 dynes/cm. 20°C 27.31 e | 442 d' | 125 1 423 0.0871 <u>°C</u> 5 ь -0.0368 4 30 26.39 to 76.16 40 25.49 5 Ref. Index e' | 252 0.0794 1.4468 20°C <sup>n</sup>D Parachor [P] d<sub>c</sub> g/ml 1.4448 25 20°C 2 vc ml/g t\_°C 70 1,4276 30 <sup>t</sup>c 40 "C" 0.7428 4  $P_c$  mm Sugd. 931.2 5 MR (Obs.) 108.80# 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 108.414 5 130.85°C. 1.0000 1.0484<sup>‡</sup> (nD-d/2) u. 2 98. <sup>‡</sup> 30 mm 1.0000 2 Dispersion Dielectric 2.09 5 BP 0.8936 Flash Point °C 0.8594 A 252 to 7.51789 Fire Point 2537.9 B \_506°C M. Spec. С ΔHc kcal/m 167.1 5 Ultra V. ΔHf A\* 252 to 2.31840 5 X-Ray Dif. ΔFf B\* 433 °C 2456.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $\mathbf{t}_{\mathbf{k}}$ Carbon tet. °C Benzene ۸' Ether B١ n-Heptane B<sub>v</sub> | C' Ethanol °C Water A\*\* 125 to 2.22338 B'\* 252 °C Water in (B<sup>V</sup>)| 2420.4 Acl (A<sup>V</sup>)| to Bc °C cp liq. ۰ĸ Cc Cryos. A° c<sub>p</sub> vap. °K consts. B° c<sub>v</sub> vap. t, °C 423,63 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A. A. et al.  $^{1}$ purity applies to  $^{70^{\circ}\text{C}}$ . and  $^{4}$ 

							No. 164	
NAME	n-Te	tracc	sane			STRUCTURAL	FORMULA	1
						CH (CH ) C	·u	
Mole	1 Ref. Mo	lecul	ar	Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>22</sub> C	<sup>,11</sup> 3	
Mole % Pur. 99.	9 <sup>1</sup> 3 Fo	rmul	C <sub>24</sub> H <sub>50</sub>	Weight 338,6	40			
	·	Ref.		·	Ref			Ref.
F.P. °C F.P. 100%	50.9	2	dt/dP *C/mm			f to		
B. P. *C	<del> </del>	$\vdash$	138.4°C.	154, 37	5	g <u>*K</u>		
760 mm	391.3 302.82	2 4	BP t <sub>e</sub>	0.0683 0.0358	4 5	f' to		$\vdash$
100 30	262.46	4	30 mm	1.021	5	g'   ' <u>°</u> K_		
10 1	231.3	5	ΔHm cal/g	38,74	3	h'		<u> </u>
Pressure		<del>                                     </del>	ΔHv cal/g			m to		
mm 138, 4		5	138, 4°C. 30 mm	64.4 54.93	5 5			
t <sub>e</sub> Density	1694.4	-	BP	44.05	5	m'   to		
g/ml 20°C	0.7991	2	te te (d, e)	40.59 40.35	5	n' °K_		]
dt 25 4 70	0.7958 0.7657l	2	ΔHv/T	19.40	5	o'		L
<u> </u>	0.8123	4	d 1260 to		5	Surface tension dynes/cm. 20°C	27.47	5
ь	-0.0366	4	$\frac{e}{d}$ , $\frac{1}{135}$ $\frac{90}{to}$		5	30	26.58	5
Ref. Index	1 1 4480₹	2	e'   260 °C		5	40	25.70	5
D 25	1.4460	2	d g/ml vc ml/g			Parachor [P] 20°C		
70	1.42861	3	tc °C	Į		30 40		
"C"	0.7427	5	P <sub>c</sub> mm			Sugd.	970.2	5
MR (Obs.) MR (Calc.	113.44 <sup>‡</sup> 113.032 <sub>4</sub>	2 5	PV/RT			Exp. L.1.%/wt.		Г
(nD-d/2)	1.0484	2	138,4°C. 30 mm	1.0000	5	u. Dispersion	98. <sup>‡</sup>	2
Dielectric	2.10	5	BP	0.8933 0.8583	5	Flash Point °C		<u> </u>
A 260 to		5	t <sub>e</sub> t <sub>c</sub>	0.0505		Fire Point		├
c	165.1	<u> </u>	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 260 to B* 450 °C		5	ΔFf			X-Ray Dif. Infrared		
к ———	-		Viscosity			Solubility in +		<del> </del>
t <sub>k</sub>   -t <sub>0</sub>	-		centistokes 7 °C			Acetone		
'x '			1	İ		Carbon tet. Benzene		
A'   to						Ether		
	1		B <sup>V</sup>   to			n-Heptane Ethanol		
A'* 135 to		5	AV I °C	-		Water Water in		
B'* 260 °C	<del> </del>	5	(A <sup>V</sup> ) <sub>1</sub> to					
Bc t °C			<u> </u>	<del> </del>	$\vdash$			
Cc — —	1		c <sub>p</sub> liq. •K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	436,24	5	c <sub>w</sub> vap.					L
	rcooled liquid					+ grams/100 gran		t
	ES: 1-Dow	2-AI		Calc. from de	dat	ta 5-Calc, by for	nula	
SOURCE:	TON.	API						
PURIFICAT		API		(2010) 1055 7				
LILERATU	RE REFERE	NUES	5: 3 JACS 77	(2019) 1955 <b>S</b> c	hoer	er A.A. et al.		
purity app	lies to n <sub>D</sub>	and	ld <sup>t</sup>					

								No. 16	55
NAME	n-Pentac	osan	•			ST	RUCTURAL	FORMUI	LA
Mole % Pur. 9	9.81 Ref. Mo	lecul	C 25H52	Molecular Weight 352.6	66		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>23</sub>	CH <sub>3</sub>	
		Ref.			Ref.				Ref.
F. P. ℃	53.7	2	dt/dP			f	to		
F.P. 1009	%		°C/mm 145.7°C.	156.7	5	g	°K		1
B. P. °C 760 mm	401.9	2	BP	0.0690	2	_h_	l		
100	312.36	4	t <sub>e</sub>	0.0357	2	f'	to		
30 10	271.48	5	30 mm	1.0345	5	g'	°K		1
1	186.55	5	ΔHm cal/g	39.13	3	h'			+
Pressure			ΔHv ca1/g 145.7°C.	63.08	5	m n	to or		
mm 145.		5	30 mm	53.82	5	0			-
Density	1720.0	5	BP	43.12	5	m'	to		T
g/ml 20°	0.8012 <sup>‡</sup> 0.7979 <sup>‡</sup>	2	t <sub>e</sub> (d, e)	39.64 40.36	5 5	n'	<b>°</b> K		
dt 25 4 70	0.79791 0.7693	2	AHv/Te	19.84	5	ο'			
		3	d 270 to	<del></del>	5		face tension	27 (2	1_
a b	0.8144 -0.0366	4	_e_ <u>  447</u> °C	0.0820	5	dyn 8	es/cm. 20°C 30	27.63 26.73	5
Ref. Inde			d' 140 to		5		40	25.86	5
n <sub>D</sub> 20°0	1 1 4471 7	2 2	d <sub>c</sub> g/ml		+	Paı	achor [P]		
25 70	1.4471	3	v ml/g				20 <b>°C</b> 30		ŀ
"C"	0.7424	5	1 _				40		_
MR (Obs.		2	P <sub>c</sub> mm	ļ	$\sqcup$			1009, 2	5
MR (Calc.	.) 117.65	5	PV/RT 145.7°C.	1,0000	5	Exp	o. L.1.%/wt. u.		
(nD-d/2)	1.0485	2	30 mm	1.0000	5	Dis	persion	98. ≠	2
Dielectric		5	BP t <sub>e</sub>	0.8927 0.8570	5		sh Point °C		
A 270 to		5	t e	0,00			e Point		
с '	163.1	5	AHc kcal/m				Spec. ra V.		
A*  270 to		5	ΔHf ΔFf			X-1	Ray Dif.		
B*[460 °C	2560.0	5	Viscosity		$\dagger$		ared		ļ
°	_		centistokes				ubility in "cetone		
t <sub>k</sub> to			η °C				rbon tet.		
t <sub>x  </sub> °C		+-			1 1		nzene her		
B' i.	_		<del></del>	<del> </del>	+		Heptane		
C'	+		B <sup>V</sup>   to A <sup>V</sup>   °C				hanol ater		
A'*140 to B'*270 °C		5	-(B <sup>V</sup> )	-			ater in		
Acl to		1	(A <sup>V</sup> )						
Bc tc °C				+	+-				
Cc		<del>                                     </del>	Р.						
Cryos, A'			c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	448.26	5	c <sub>v</sub> vap.			L,_			
	rcooled liquid						rams/100 gra		nt
	CES: 1-Dow			Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:		API							
PURIFICA		API							
			5: 3 JACS 77	<b>(</b> 2019) 1955 <b>S</b> c	hoere	r A.	A. et al.		
'purity app	olies to nD	'and	ď <sub>4</sub>	•					

							No. 166	ı
NAME	n-Hexaco	sane				STRUCTURAL		
Mole	Ref. Mo	lecul	ar	Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub>	4 <sup>CH</sup> 3	
% Pur. 99.	8 <sup>1</sup> 3 Fo	rmul	ar C <sub>26</sub> H <sub>54</sub>	Weight 366.	692			
	1 5/ 4	Ref.			Ref		r	Ref.
F.P. °C F.P. 1009	56.4	2	dt/dP °C/mm			f to		
B. P. °C	<b>†</b>	<u> </u>	152.75°C. BP	158.97	5	h		
760 mm 100	412.2 321.64	2	t <sub>e</sub>	0.0698 0.0356	2 5	f' + to		
<b>3</b> 0	280.25 248.22	4 5	30 mm	1.0477	5	g' °C		
1	194.18	5	ΔHm cal/g			h'   m   to		
Pressure		5	ΔHv cal/g 152.75°C.	61.84	5	n °K		
mml 52.7	1746.1	5	30 mm BP	52.78 42.24	5	o i		
Density	+	1_	t	38.72	5	m'   to		
g/ml 20°0 dt 25 4 70	0.8032 0.7998 0.7704	2 2	'e '"' '	39, 42	5	0'		
			ΔHv/T <sub>e</sub>	19.82	5	Surface tension		
a b	0.8168 -0.0368	4	e   460 °C	0.0799	5	dynes/cm. 20°C	27.79 26.86	
Ref. Index		1	d'   140 to		5	40	25.95	
<sup>n</sup> D 20°C	1.4501 1.4481 1.4310	2 2	d <sub>c</sub> g/ml			Parachor [P]		ĺ
70	1.43101	3	vc ml/g tc ℃			30		
"C"	0.7421	5	P <sub>c</sub> mm			40 Sugd.	1048.2	5
MR (Obs.) MR (Calc.	1 122 268	5	PV/RT		<u> </u>	Exp. L.1.%/wt.		
(nD-d/2)	1.0485*	2	152.75°C. 30 mm	1.0000 1.0000	5	u. Dispersion	98. #	2
Dielectric		5	BP	0.8924 0.8560	5	Flash Point °C	-	
A   278 to	7.57689 2692.73	5	t <sub>e</sub> t <sub>c</sub>	0.0300	1	Fire Point		
C	161.2	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		İ
A* 278 to B* 470 °C	2.41467	5	ΔFf			X-Ray Dif. Infrared		
к — — -	-		Viscosity			Solubility in +		
t <sub>k</sub> — —		1	centistokes 7°C	:		Acetone Carbon tet.		
'x '		<u> </u>	•			Benzene		
A'   B'	1	1				Ether n-Heptane		
C'	<b></b>	<u> </u>	B <sup>V</sup> to			Ethanol Water		
A'* 140 to B'* 278 °(		5	(BV)	_		Water in		
Acl to	,		(A <sup>V</sup> )					
Bc tc_°C	=		c <sub>p</sub> liq. °C			1		
Cryos, A°			c <sub>p</sub> vap. °K	:				
t_ °C	459.98	5	c <sub>w</sub> vap.					
≠ for unde	rcooled liquid		IL			+ grams/100 grai	ms solveni	 t
			PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc. by for		
SOURCE:		AP						
PURIFICA		AP						
LITERATU	RE REFERE	NCES	S: 3 JACS 77 (2	019) 1955 Scho	oere	r A.A. et al.		
purity app	lies to n <sub>D</sub> <sup>70°C</sup>	• and	d <sup>t</sup> <sub>4</sub>					

No. 167 n-Heptacosane NAME STRUCTURAL FORMULA CH3(CH2)25CH3 Molecular 380.718 Molecular Mole % Pur. 99.01 C27H56 Formula Ref Ref. Ref. 59.0 F. P. °C 2 dt/dP f to F.P. 100% °C/mm g °K 159, 55°C. 161,13 B. P. °C h BP 0.070 2 760 mm 422.1 2 <sup>t</sup>e 0.03463 5 ſ١ 330,55 to 100 g' °К 288.67 4 30 30 mm 1.0544 5 256.26 5 10 h' 37.93 3 ∆Hm cal/g 201.54 5 to m ΔHv cal/g 159.55°C. Pressure°C ۰ĸ n 60.65 0.1 5 mm159.55 o 30 mm 51.8 5 1770.4 5 t<sub>e</sub> ВP 41.4 5 m' to Density 5 38.6 te te (d, e) n' ۰ĸ g/ml 20°C 0.8050 2 38.76 ۰, 0.8016 0.7732  $d_4^t$ 25 2 19.90 5 AHv/T 70 3 Surface tension d 285 74.30 5 to 0.8186 4 dynes/cm. 20°C 27.93 0.0779 470 d 150 °C 26.99 26.09 ь -0.0368 4 30 71.58 to 5 40 Ref. Index e' | 285 0.0685 5 1.4511 <sup>n</sup>D 20°C [P] 1.4491 Parachor d<sub>c</sub> g/ml 2 25 20°C 1.43211 vc ml/g 70 3 30 <sup>t</sup>c 40 "C" 0.7420 4 P<sub>c</sub> mm Sugd 1087.2 5 127.37# MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 126.886 5 1.0000 159,55°C. 1.0486 (nD-d/2) u. 2 98.≠ 1.0000 5 30 mm 2 Dispersion Dielectric 2.11 5 BP 0.8919 5 Flash Point °C 0.8548 5 A 285 to 7.59371 27**40**.08 5 Fire Point B 1520 °C 5 M. Spec. AHc kcal/m С 159.3 5 Ultra V. ΔHf A\* 285 to 2.4430 265**6.**4 5 X-Ray Dif. ΔFf B\*|5 00 °C Infrared ĸ Viscosity Solubility in centistokes Acetone to t<sub>k</sub> ∫ t<sub>x</sub> ∣ Carbon tet. °C Benzene A' Ether В' n-Heptane B<sub>v</sub> | C' to Ethanol °C Water A'\* 150 to 2.35017 5 B'\* 285 °C Water in 2614.8 (B<sup>V</sup>) Aclto (A<sup>V</sup>) Bc t<sub>c</sub> °C cp liq. °K Cc Cryos. A. c vap. °K consts. B° c<sub>w</sub> vap. te °C 471.24 5 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 JACS 77 (2019) 1955 Schoerer A.A. et al. 1 purity applies to nD 70°C. and dt

									No. 168	
NAME		n-O	ctaco	sane				STRUCTURAL		
Mole % Pur. 99	. 01	Ref.	Мо	lecul	ar C <sub>28</sub> H <sub>58</sub>	Molecular Weight 394.7	744	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>26</sub>	CH <sub>3</sub>	
W Fur.		1,	FO	Ref	20 30	weight 374.	Ref			Ref.
F. P. *C	Т	61.4		2	dt/dP	T	101			
F.P. 1009	,			$\vdash$	*C/mm			f to		ļ
B.P. °C	1			$\Box$	0.1 mm	163.21	5	h		
760 mm 100		431.6 339.1		2 4	BP t <sub>e</sub>	0.071 0.03475	5	f' + to		l
30		296.7		4	30 mm	1.0682	5	g' 'K		
10		263.9° 208.5		5	ΔHm cal/g	39.14	3	h'		
1 0.1		166.0		5	ΔHv cal/g	<del>                                     </del>		m   to		
Press. mm	+-			Н	0.1 mm	59.50	5	n °K		
t <sub>e</sub>	1	7 <b>3</b> 7.2		5	30 mm BP	50.8 40.6	<b>5</b>	ļ.—————		<u> </u>
Density	. [		~-#		t_	37.8	5	m'   to		l
g/ml 20°C	1	0.8	067‡ 033‡	2 2	, e (a, e)	37.77	5	ö'		l
d <sup>t</sup> 25 4 70	$\perp$	0.7	750 <sup>1</sup>	3	ΔHv/T <sub>e</sub>	19.79	5	Surface tension		<del>                                     </del>
8	T	0.8		4	d   295 to		5	dynes/cm, 20°C	28.06	
Ъ	+-	-0.0	368	4	d'   162 to	70.56	5	30 40	27.12 26.21	5
Ref. Index		1.4	520 <sup>‡</sup>	2	e' i 295 °C	0.0666	5	Parachor [P]	20.21	Ť
45		1.4	50 <b>0</b>	2	d g/ml			20°C		1
70	+-	1.4	330-	3	vc ml/g tc °C			30 40		
"C"	+-	0.7		4	P <sub>c</sub> mm			Sugd.	1126.2	5
MR (Obs.) MR (Calc.	٦.	1 <b>32</b> .0	Λ4	2 5	PV/RT	<b>+</b>	<u> </u>	Exp. L.1.%/wt.		
(nD-d/2)	ή :	1.0	487 <sup>≠</sup>	Ž	0.1 mm 30 mm	1.0000	5	u.	98.≠	2
Dielectric		2.1		5	BP	1.0000 0.8912	5	Dispersion Flash Point °C	70.	-
A 295 to		7.6 785.8	0972 0	5 5	t <sub>e</sub> t <sub>c</sub>	0.8535	5	Fire Point  M Spec.		_
A*   295 to B* 490 °C		2.4° 702.8	7084	5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif.		
к ·	╛	.02.0			Viscosity centistokes			Infrared Solubility in + Acetone		-
t <sub>x</sub>   to t <sub>x</sub>   to	:				7 •			Carbon tet. Benzene		
B'i °C						<u> </u>	ļ	Ether n-Heptane		1
C'					B <sup>V</sup>   to			Ethanol	İ	
A'* 162 to B'* 295 *C	20	2.3° 658.5	7498	5 5	(B <sup>V</sup> )	-		Water in		
Ac to Bc t C					(A <sup>v</sup> ) <sub> </sub> c <sub>p</sub> liq. °K	1	_			
Cryos, A <sup>o</sup>				H	c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	_	480.		5	c <sub>w</sub> vap.					
# for unde			quid	1 1	L		ı	grams/100 grai	ne solven	<b></b>
REFEREN	ES:	1-D	ow	2-AF	I 3-Lit. 4-	Calc, from det	da	ta 5-Calc, by for		*
SOURCE:				AP						
PURIFICA	CION	T:		AP						
LITERATU	RE	REF	ER EI	NCES	: 3 JACS 77 (	2019) 1955 Sch	oere	er A. A. et al.		
i I										
l purity app	lies	to n <sub>D</sub>	70°C	and o	d <sup>t</sup> 4					

No. 169 n-Nonacosane NAME STRUCTURAL FORMULA CH3(CH2)27CH3 Mole Ref 3 Molecular Molecular  $C_{29}H_{60}$ % Pur. 99.5 Formula Weight 408,770 Ref. Ref. Ref F.P. °C F.P. 100% 63.7 2 dt/dP f to °C/mm °K g 0.1 mm 165,23 5 B. P. °C h 0.071 5 BP 760 mm 440.8 2 f¹ 0.0347 5 100 347.38 4 ۰ĸ g' 1.0796 5 30 mm 30 304.59 4 10 271.43 5 h' ∆Hm cal/g 215.40 172.36 5 1 m to AHv cal/g 0.1 5 ۰ĸ n 58.40 0.1 mm 30 mm 5 Press. mm 0 49.9 t<sub>e</sub> 1816.9 5 39.8 BP 5 m' to Density te (d, e) 37.1 5 n' g/ml 20°C ۰ĸ 0.8083 2 36.94 5 ٥,  $d_4^t$ 25 0.8049 0.8015 2 ΔHv/T<sub>e</sub> 19.78 5 30 4 Surface tension 302 72.49 28. 18 \( \frac{\pmu}{27. 25} \)
26. 33 \( \frac{\pmu}{25} \) 5 to 0.8219 dynes/cm. 20°C 1 491 ٠c 0.0742 -0.0368₹ 4 h ă 🗆 169 30 to 69.47 40 Ref. Index e' | 302 °C 0.0643 1.4529 20°C [P]  $^{n}D$ Parachor dc g/ml 25 1.4508 2 20°C vc ml/g t °C 1.4488 30 4 30 t<sub>c</sub> 40 "C" 0.7418 4  $P_c$  mm Sugd. 1165.2 5 MR (Obs.) 136.65≠ 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 136,122 0.1 mm 1.0000 5 1.0487<sup>‡</sup> (nD-d/2)2 98.¥ 30 mm 1.0000 2 5 5 Dispersion Dielectric 2,11 5 BP 0.8910 Flash Point °C 0.8527 5 A 302 to 7,62529 5 Fire Point B \_500 °C 2830,55 5 M. Spec. C AHc kcal/m Ultra V. ΔHf A\* 302 to 2.49691 5 X-Ray Dif. ΔFf B\*| 501 °C 2746.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to t<sub>k</sub> Carbon tet. °C Benzene A١ to Ether B١ °C n-Heptane B. C' Ethanol to ۱ ۷ °C Water A'\* 169 to B'\* 302 °C 2.39938 5 Water in 2701.3  $(B^{\vee})$ Acl to (A<sup>V</sup>)| °C Βc ۰ĸ c<sub>p</sub> liq. Cc c<sub>p</sub> vap. Cryos. A° °K consts. B° c vap. 492.51 te °C 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES: 3 JACS 77(2019) 1955 Schoerer A. A. et al.

							···	No. 17	0
NAME		n-T	riac	ontane		- 1	STRUCTURAL	FORMULA	
						$\neg$	CH (CH )	CII	
						ᅥ	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>28</sub>	CH <sub>3</sub>	
Mole 99.	01 Ref.	Mo	lecul	ar C <sub>30</sub> H <sub>62</sub>	Molecular				
% Pur, '''	3	Fo		a 30 62	Weight 422.	-		<del></del>	5 (
	65, 8		Ref.		1	Ref		<del></del>	Ref.
F.P. C F.P. 100%		,	-	dt/dP *C/mm	1		f to		1
B. P. °C	-		$\vdash$	0.1 mm	167.20	5	] - 1	ļ	Į
760 mm	449.7	,	2	BP	0.072	5	$\left  \frac{\mathbf{h}}{\mathbf{f'}} + \frac{\mathbf{h}}{\mathbf{f}} \right $		l
100 30	355.4		4	t <sub>e</sub> 30 mm	0.03469 1.0907	5	g' to		ĺ
10	312.1 278.6		4 5		1.0,01	H	h'	1	1
1	222.0	0	5	ΔHm cal/g	<b>↓</b>	-	m l to	<b> </b>	-
0, 1	178.4	6	5	ΔHw cal/g 0.1 mm	57, 33	5	n K		l
Press. mm	1838.9	,	5	30 mm	49.0	5	0	1	l
e Density	1030.7		H	BP	39.1	5	m'   to		
g/ml 20°C	0.8	097	2	te (d, e)	36.3 36.13	5	n' 'K		
d <sub>4</sub> 25	0.8	064 <sup>≠</sup> 031	2 4	AHv/Te	19.77	5	٥'		
a 30	<del></del>	229	4	d   309 to	71,47	5	Surface tension	20.20	۔ ا
Ъ	- <b>0</b> .0		4	<u>e   510 °C</u>	0.0720	5	dynes/cm. 20°C	28.29 27.37	5 5
Ref. Index	<b>1</b>			d'   170 to		5 5	40	26.49	
n <sub>D</sub> 20°C		536 <sup>‡</sup>	2	d g/ml	0.0023	H	Parachor [P]		
25 70		516 <sup>∓</sup> 348 <sup>1</sup>	2 3	v mi/g	}		20°C	l	1
"C"	<del></del>	415	4	, *c ~		l i	40		
MR (Obs.)			2	P <sub>c</sub> mm			Sugd.	1204.2	5
MR (Calc.	140.7	4 .	5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0	487 <sup>≠</sup>	2	0.1 mm 30 mm	1.0000	5	u. Dispersion	98.≠	2
Dielectric	2,1	1	5	BP	0.8906	5	Flash Point °C	<u> </u>	┝╌
A   309 to		4044	5	t <sub>e</sub> t <sub>c</sub>	0.8517	5	Fire Point		ĺ
C Table	2874.3	4	5	ΔHc kcal/m	<del> </del>	H	M Spec.		
A*   309 to	2.5	2257	5	ΔHf			Ultra V. X-Ray Dif.	}	ļ
B* 511 °C	2789.6		5	ΔFf	ļ		Infrared		
K — — —	1			Viscosity centistokes		ĺ	Solubility in +		
to to				7 ·c	1	1 1	Acetone Carbon tet.		l
t <sub>x l</sub>				<b>!</b> •			Benzene	1	
A'   to							Ether		
č, – – –	-			B <sup>V</sup>   to			n-Heptane Ethanol	ł	
A'* 175 to	2,4	2407	5	AV I °C			Water	}	1
B'*309 °C	2743.7		5	(B <sup>V</sup> )			Water in	<del> </del>	├
Ac to				(A <sup>V</sup> )				1	1
Bc tc_C	<u>-</u>			cp liq. °K					
Cryos. A*	<b>†</b>			c_ vap. °K	]				
consts. B°				c <sub>p</sub> vap. °K				1	
t <sub>e</sub> °C	502.6	5	2	c <sub>v</sub> vap.					İ
≠ for unde	rcooled 1	iquid					grams/100 gra	ms solven	<u></u>
REFERENC	ES: 1-D	ow	2-AF	PI 3-Lit. 4-0	Calc. from det	dat			
SOURCE:			AP						
PURIFICAT	TION:		AP						
LITERATU		EREN			(2019) 1955 Sc	hoer	er A. A. et al.		
1		7000							
purity app	plies to n	D	·.						

	<del></del>							No. 17	1
NAME _	n-He	ntria	contane			ST	RUCTURAL	FORMUL	A
	<del></del>						CII (CII )	CII	
Mole % Pur.		lecul rmul	ar C <sub>31</sub> H <sub>64</sub>	Molecular Veight 436.82	22		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>29</sub>	, CH <sub>3</sub>	
		Ref.			Ref.				Ref.
F.P. C F.P. 100%	67.9	2	dt/dP °C/mm			f g	to °K		
B.P. °C 760 mm	458.	2	0.1 mm BP t <sub>e</sub>	169.01 0.073 0.03336	5 5 5	_h _f'	to		
100 30 10	362.87 319.23 285.40	4 4 5	30 mm	1.0818	5	g' h'	*K		
1 0.1	228.17 184.17	5	ΔHm cal/g ΔHv cal/g			m	to °K	-	$\vdash$
Press. mm	1859.3	5	0.1 mm 30 mm BP	56.29 48.1 38.4	5 5 5	٥			
Density g/ml 20°C dt 25 4 30	0.8111 <sup>‡</sup> 0.8078 <sup>‡</sup> 0.8045	2 2 4	te te (d, e) AHv/Te	35.6 36.69 20.52	5 5 5	m' n' o'	to •K		
a b	0.8243 -0.0366	4 4	d 320 to -e 508 °C d 180 to	70.41 0.0699 67.46	5 5 5		face tension es/cm. 20°C 30 40	28.40 27.48 26.59	5 5 5
Ref. Index nD 20°C 25 30	1.4543 <sup>‡</sup> 1.4523 <sup>‡</sup> 1.4503	2 2 4	e'   320 °C d <sub>c</sub> g/mI v <sub>c</sub> ml/g t <sub>c</sub> °C	0.0607	5	Par	achor [P] 20°C 30	20.37	
"C"	0.7413	4	P <sub>c</sub> mm				40 Sugd.	1243.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	1.0487	2 5 2	PV/RT 0.1 mm 30 mm	1.0000 1.00 <b>0</b> 0	5	_	L.1.%/wt. u. persion	98.≠	2
Dielectric  A 320 to B 485 °C	7.65344 2914.17	5 5 5	BP te tc	0.8905 0.8511	5 5		sh Point °C e Point		
A*  320 to B*  518 °C	2.54572 2829.	5 <b>5</b>	ΔHc kcal/m ΔHf ΔFf			Ult:	Spec. ra V. Ray Dif. ared		
K to to			Viscosity centistokes 7°C			Solu Ac Ca	ability in + etone rbon tet.		
A'   to B'   _ °C C'			B <sup>V</sup>			n- Et	her Heptane hanol		
A'*180 to B'*320 °C	2.44635 2782.1	<b>5</b> 5	(B <sup>V</sup> )				iter iter in		ļ
Ac to Bc tc °C Cc	-		(A <sup>V</sup> )  c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap. *K						
te °C	512.12	5	c <sub>v</sub> vap.			<u> </u>			<u></u>
	rcooled liquid		<b>T</b>				ams/100 gra		it
	ES: 1-Dow		PI 3-Lit. 4-0	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE: PURIFICAT	ION.	AP							
	RE REFERE								

·							No. 172	<u>:</u>
NAME	n-De	otria	contane			STRUCTURAL	FORMULA	
Mole	Ref. Mo	1 1		Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>30</sub> C	H <sub>3</sub>	
% Pur.	Fo Fo	rmul	ar C <sub>32</sub> H <sub>66</sub>	Weight 450.8	48			
		Ref.			Ref.			Ref.
F. P. *C	69.7	2	dt/dP			f to		
F.P. 100%	<u> </u>	_	*C/mm 0,1 mm	171.00	5	g °K		
B.P. °C 760 mm	467.	2	BP	0.073	5	<u>+</u> +		
100 30	370.97 326.88	4	t <sub>e</sub> 30 mm	0.0347	5	f' to g' K		
10	292.69	4	ΔHm cal/g	1.111/	⊢∸	h'		
1 0. 1	234.83 190.32	5	ΔHv cal/g	<del> </del>	-	m to		
Press. mm		Ť	0.1 mm	55.37	5	n °K		
t <sub>e</sub>	1884.1	5	30 mm BP	47.1 38.24	5 5			
Density g/ml 20°C	0.8124		t <sub>e (1, 1</sub> )	35.0	5	m' to	İ	
at 25	0.8091	2 2	te (d, e)	34.67 19.71	5 5	0'		
	0.8058	4	ΔHv/T <sub>e</sub>		5	Surface tension		$\overline{}$
a b	0.8256 -0.0366	4	a 1 520 °C	0.0632	5	dynes/cm. 20°C	28.50 27.58	<b>5</b>
Ref. Index	<del></del>		d' 186 to		5	40	26.69	5
n <sub>D</sub> 20°C	1.4550 <sup>#</sup> 1.4530 <sup>#</sup>	2 2	<u> </u>	1		Parachor [P]		
30	1.4516	4	d g/ml v ml/g t °C			20°C 30		l
"C"	0.7412	4	-			40		_
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	<del> </del>	-	Exp. L.1.%/wt.	1282.2	5
MR (Calc. (nD-d/2)	149.976 1.0488	5 2	0.1 mm	1.0000	5	u.		
Dielectric	2,12	5	30 mm BP	1.0000 0.8910	5	Dispersion	98.≠	2
A 324 to		5	t <sub>e</sub>	0.8500	5	Flash Point °C Fire Point		
B 1530 °C	2958.7	5	ΔHc kcal/m	<del> </del>	-	M Spec.		
A* 324 to	2.56852	5	ΔHf			Ultra V. X-Ray Dif.	j	ĺ
B* <u>5</u> 30 ℃		5	ΔFf	<b>_</b>		Infrared		
c _			Viscosity centistokes			Solubility in +		
tk Too			η •c			Acetone Carbon tet.	1	
t o		<u> </u>			į	Benzene Ether		
B' •C			V 1	<del> </del>	<b> </b>	n-Heptane		1
C'	<del>                                     </del>	<del>  -</del>	B <sup>V</sup>   to			Ethanol Water		İ
A'* 186 to B'* 324 °C		5	(BV)	-		Water in		
Ac to			(A <sup>V</sup> )					
Bc tc_C	2		c <sub>p</sub> liq. °K					
Cryos, A°		$t^-$	c <sub>p</sub> vap. °K					
consts. B			ll -				l	
t <sub>e</sub> °C	522.46	5	c <sub>v</sub> vap.					<u>L</u>
	rcooled liquid					grams/100 gran		<u> </u>
	CES: 1-Dow	2-A1		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	PION.	API						
	RE REFERE	API		<del> </del>				
	ns refere		•					

					,			No. 173	}
NAME _	n-Trit	riac	ontane			STRUCTUR	AL	FORMUL	A
Mole % Pur.	Ref. Moi	ecul muli	ar C <sub>33</sub> H <sub>68</sub>	Molecular Weight 464.8	74	СН <sub>3</sub> (С	CH <sub>2</sub> ) <sub>3</sub>	CH <sub>3</sub>	
	1 5, 4	Ref.			Ref.				Ref.
F.P. °C F.P. 100%	71.4	2	dt/dP °C/mm			f	to		İ
B. P. °C	<u> </u>	-	0.1 mm	172.76	5	g   h	°K		
760 mm	475.	2	BP	0.074 0.03462	5	- <u>r</u>			ĺ
100 30	378.17	4	t <sub>e</sub> 30 mm	1.1221	5	g'	to •K		İ
10	299.19	5	ΔHm cal/g	<del>                                     </del>		h'			İ
0.1	240.77 195.81	5	ΔHv cal/g			m	to		
Press. mm			0.1 mm 30 mm	54.42 46.5	5	n o	•K		İ
t <sub>e</sub>	1900.8	5	BP	37.1	5	m!	to		<del> </del>
Density g/ml 20°C	0.8136≠	2	te te (d, e)	34.3 33.99	5	n' i	°K		į
dt 25	0.8103	2	ΔHv/T <sub>e</sub>	19.69	5	o'			İ
	0.8070	4	d 330 to	68,70	5	Surface tens		20.75	
a b	-0.0366	4	e   529 °C	0.0665	5	dynes/cm. 2	0°C	28.59 27.67	5
Ref. Index	#		e'   330 °C	65.66 0.0574	5		0	26,78	5
n <sub>D</sub> 20°C	1.4557 1.4536	2	d <sub>c</sub> g/ml			Parachor [I	P]		
30	1.4517	. 4	V_mi/g				0		1
"C"	0.7412	4	t <sub>c</sub> °C P <sub>c</sub> mm				0 ugd.	1321,2	5
MR (Obs.) MR (Calc.)	155.21# 154.594	2 5	PV/RT	<del> </del>		Exp. L.1.%/			<u> </u>
(nD-d/2)	1.0488≠	2	0.1 mm 30 mm	1.0000	5	u. Dispersion		98.≠	2
Dielectric	2.12	5	RP	0.8895	5	Flash Point	°C	70.	<u>-</u>
A 330 to B 539 °C	7.68082 2997.60	5 5	t e t	0.8491	5	Fire Point			
C	2 50001		ΔHc kcal/m ΔHf			M. Spec. Ultra V.			
A*  330 to B*  539 °C	2.59301 2912.0	5 5	ΔFf			X-Ray Dif. Infrared			
K			Viscosity			Solubility in	+		<u> </u>
t <sub>k</sub> To			centistokes り °C			Acetone			
*x				İ		Carbon tet. Benzene	·		
A'   to B'   °C						Ether n-Heptane			
C'			B <sup>V</sup>			Ethanol			
A'* 185 to B'* 330 °C	2.48943 2862.0	5 5	$\frac{A'}{(B') } - \frac{^{\circ}C}{}$	-		Water Water in			
Acl to		Ť	(A <sup>V</sup> )						
Bc tc °C	-		c <sub>p</sub> liq. °K						
Cryos. A*			c <sub>p</sub> vap. *K						
consts. B°	531.48	5	c vap.						
te °C	cooled liquid	٠	_ <b>v</b> • • • • • • • • • • • • • • • • • • •	1	<u> </u>	+			<u> </u>
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	grams/100 ta 5-Calc. b			-
SOURCE:		AP					,		
PURIFICAT	ION:	AP						<del></del>	
	RE REFEREI								
<u> </u>									

									No. 174	
NAME		n-	Tetr	atria	contane		$\neg$	STRUCTURAL	FORMULA	١.
Mole % Pur,		Ref.	Mo Fo	lecul rmul	ar C 34 <sup>H</sup> 70	Molecular Weight 478.9	00	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>32</sub>	СН <sub>3</sub>	
				Ref.			Ref			Ref.
F. P. *C	T	73.1		2	dt/dP	1		f   to		
F.P. 1007	5				*C/mm		1 1	f to g °K		
B.P. °C	+				0.1 mm	174.29	5	h		
760 mm	-1	482.		2	BP	0.074 0.03457	5		ļ	
100	-	384.4		4	t <sub>e</sub>	1,1310	5	g' to		
30 10	ı	339.6 304.8		4 5	30 mm	1.1310	-	h'	Ì	
ì		245.9		5	AHm cal/g	<u> </u>		<u> </u>	<del> </del>	
0.1		200.6	2	5	∆Hv cal/g	52.44	ا ـ ا	m to		
Press. mn	n	1918.5		5	0, 1 mm 30 mm BP	53.44 45.7 36.4	5 5 5	0		<u> </u>
Density	T		4		te (d.e)	33.7	5	m'   to		i i
g/ml 20°0	۱:	0.8	148‡	2	t (d, e)	33.32	5	n' 'K		
dt 25 4 30		0.8	115 <sup>#</sup>	2 4	AHV/Te	19.69	5	<u> </u>		
	+				d   336 to	67.89	5	Surface tension		
a b		0.8 -0.0		4	e   537 °C		5	dynes/cm. 20°C	28.68 27.77	5
Ref. Index	.+		3-0	÷	d'   190 to		5	40	26.87	5
n <sub>D</sub> 20°0		1.4	563‡	2		0.0557	5	Parachor [P]		
25	-	1.4	5 <b>42</b>	2	d g/ml v ml/g		1	20°C		
30	$\perp$	1.4	524	4	v <sup>c</sup> ml/g t <sub>c</sub> °C			30		
"C"	$\perp$	0.74	410	4	Pcmm		ĺ	40 Sugd	1360.2	5
MR (Obs.		159.8		2	PV/RT	<del> </del>	-	Exp. L.1.%/wt.	1300.2	+
MR (Calc. $(nD-d/2)$	7	159.2	12 488 <sup>≠</sup>	5	0.1 mm	1,0000	5	u.	,	
	+			2	30 mm	1.0000	5	Dispersion	98. <sup>‡</sup>	2
Dielectric	_	2.17		5	BP	0.8895	5	Flash Point °C		
A 336 t B 547 C		7.69 3031.78	916 <b>4</b> 8	5	t <sub>e</sub> t <sub>c</sub> AHc kcal/m	0.8486	3	Fire Point M Spec.	-	
A* 336 to B* 547 °C		2.61 2945.7	1320	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
к ——	-				Viscosity			Solubility in +	<del> </del>	+
t <sub>k</sub>					centistokes 7°C			Acetone		
t i to		·						Carbon tet. Benzene		
B' °(								Ether n-Heptane		
C'	4				B <sup>V</sup> to	.		Ethanol	1	1
A'*190 to B'*336 *		2.50 2895.2	0973	5 5	(BV)	-		Water Water in		_
Ac  to					(A <sup>V</sup> )					
	7				c <sub>p</sub> liq. °K	Í				
Cryos, Acconsts, B					c <sub>p</sub> vap. °K		'			
t <sub>e</sub> °C	$\perp$	539.49		5	c <sub>v</sub> vap.	L	<u>L_</u>	L	<u> </u>	<u></u>
# for und								grams/100 gran		<u>t</u>
REFEREN	UE	ວ: 1-D	ow	2-AI		Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:				AP						
PURIFICA				AP						
	·KI	SKEF	ER EI	NCES	o:					
				_						

								No. 175	
NAME	n-Pents	triac	ontane		ST	RUCTURAL	No. 175		
Mole % Pur.		lecul		Molecular Weight 492.9	26		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3:</sub>	3CH <sub>3</sub>	
<u> </u>		Ref		T	Ref.	T	······································		Ref.
F.P. *C	74.7	2	dt/dP			f	to		
F.P. 100%			°C/mm			g	∣ °K		
B.P. °C			0.1 mm BP	176.04 0.0754	5	h	1		
760 mm	490.	2	te	0.03453	5	f'	to		
100 30	391.68 346.47	4 4	30 mm	1.141	5	g'	°K	· ·	1
10	311.36	5	ΔHm cal/g			h'			
1 0.1	251.90 206.10	5	AHv cal/g			m	to		1
Press. mm	T	+	0.1 mm	52.60	5	n o	°K		
t <sub>e</sub>	1938,5	5	30 mm BP	45.0 35.8	5 5				├-
Density	0.8157	,	t <sub>e</sub>	33.1	5	m' n'	to ı °K		
g/ml 20°C	0.8124	2 2	t <sub>e</sub> (a, e)	32.73	5	0'	 I		
d <sub>4</sub> 25 30	0.8093	4	ΔHv/T <sub>e</sub>	19.67	5	S., .	face tension		╁
a	0.8289	4	d 343 to		5 5		es/cm. 20°C	28.74	5
ь	-0.0366	4	d' 200 to		5	8	30	27.85 26.95	5
Ref. Index	1.4568	2	e'   343 °C	0.0541	5		40	20.73	<u> </u>
25	1.4548	2	d <sub>c</sub> g/ml			Par	achor [P]		
30	1.4529	4	v <sub>c</sub> ml/g t <sub>c</sub> °C				30		
"C"	0.7408	4	P <sub>c</sub> mm				40 Sugd.	1399.2	5
MR (Obs.)	164.49	2	PV/RT	+	H	Evr	L.1.%/wt.		<u> </u>
MR (Calc.) (nD-d/2)	163.83 1.0489 <sup>≠</sup>	5 2	0, 1 mm	1.0000	5		u.	4	
Dielectric	2.12	5	30 mm BP	1.0000	5	Dis	persion	98. ≠	2
A 343 to	7,70371	5	t	0.8892 0.8478	5		sh Point °C e Point		
B _ 505 °C_	3070.74	5	L c						-
С	147.	5	ΔHc kcal/m				Spec. ra V.		
A*  343 to B*  546 °C	2.63438 2984.4	5	ΔFf	1			Ray Dif.		1
K	2704.4		Viscosity			<u> </u>	ared		
°	1		centistokes				ubility in <sup>†</sup> etone		
t <sub>k</sub> to t <sub>x</sub> °C	l		η °C			Ca	rbon tet.		1
A'I to		-					nzene her		
B'°C_					-		Heptane		
C'		1	B <sup>V</sup>	1	1		hanol ster	İ	
A'* 200 to B'* 343 °C	2.52835 2932.1	5	$-\frac{\mathbf{n}}{(\mathbf{B}^{\mathbf{v}}) }$	-			ater ater in		
Acl to	- /30.1	1	(A <sup>V</sup> )						
Bc tc °C				+	$\vdash$				}
Ce — —	<u> </u>		Р -						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K						
te °C	548.62	5	c <sub>v</sub> vap.						
<u> </u>	cooled liquid			1	لـــــــا	+	rama/100 ===	m o o = 1==	<u> </u>
	ES: 1-Dow		DI 3_144 4	-Calc, from de	+ 4-		cams/100 gra		16
SOURCE:	±0. 1-D0₩	API		-Calc. Irom de	t. ua		-Carc. by for		
	ION:	API							
PURIFICAT	RE REFERE							<del></del>	
LITERATOR	CE REFERE	NCE	<b>)</b> ;						

								No. 176	,
NAME	n-Hexa	triac	ontan	le			STRUCTURAL	FORMULA	A.
Mole 99	. 0 Ref.		lecul rmul		Molecular Weight 506.	052	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>3</sub>	<sub>4</sub> CH <sub>3</sub>	
A Fui.		FO	Ref	30 74 1	weight 500.	Ref	<del></del>		Ref.
F.P. °C	76.2		2	dt/dP		1	f to	T	<del> </del>
F.P. 100%				*C/mm			f to g °K		
B.P. °C				0.1 mm BP	177.75 0.0759	5 5	h ¦		
760 mm 100	497. 397. 9	Ω	2 4	t	0.0355	5	f' to		
30	352.4	3	4	30 mm	1.155	5	g' K		
10 1	317.0 257.1		5	ΔHm cal/g			h'	ļ	1
0.1	210.9		5	ΔHv cal/g			m to		
Press.mm	1938.		5	0.1 mm 30 mm BP	43. 65 35, 2	5 5	0		_
Density g/ml 20°C	7	169‡	2	te (d.e)	31.23	5	m'   to	ì	
at 25	0.8	136°	2	e (4, 6)	31.7	5	0'	İ	
<sup>4</sup> 4 30	0.8	103	4	d   349 t	19.09	5	Surface tension	<b>†</b>	$\vdash$
a b	-0.0		4	d 554 t	0.0584	5	dynes/cm. 20°C 30 40	28.85 27.93 27.03	5 5 <b>5</b>
Ref. Index	1.4	573‡	2	<u> </u>	С	+	Parachor [P]	1 27.03	┼
D 25 30	1.4	554 <sup>r</sup>	2	d g/ml v ml/g			20°C		
"C"	0.7	406	4	t <sub>c</sub> °C			30 40 Sugd	. 1438. 2	5
MR (Obs.) MR (Calc.)	1 160 4	40	2 5	PV/RT		+	Exp. L.1.%/wt.	<del>                                     </del>	_
(nD-d/2)	1.0	489 <sup>‡</sup>	2	0.1 mm 30 mm	1.0000	5 5	u.	98. <sup>‡</sup>	2
Dielectric	2.1		5	BP	0.8852	5	Dispersion Flash Point °C	70.	+-
A   349 to B   560 °C		13 <b>44</b> 0	5 5 5	t <sub>e</sub>	0.8397	5	Fire Point M Spec.		-
A* 349 to B* 560 °C	2.6	5726	5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
K				Viscosity			Solubility in +	<del>                                     </del>	+
t  to				centistokes	:		Acetone Carbon tet.		1
'x						1 1	Benzene		1
A' to B' °C							Ether n-Heptane		
<u>c,</u> – – –				B <sup>V</sup> to			Ethanol		
A'* to B'* °C				A <sup>V</sup> — °C	<u> </u>		Water Water in		
Acl to	<del></del>		$\vdash$	(B <sup>V</sup> )  (A <sup>V</sup> )					
Bc t <sub>c</sub> °C	-			c <sub>p</sub> liq. ∘r					
Cryos. A° consts. B°				c <sub>p</sub> vap. °F	۲				
te ℃	556.3		5	c <sub>v</sub> vap.	<u> </u>	$\perp$	L <u>,                                    </u>	<u></u>	<u>L</u>
for unde			•				grams/100 gra		ı <b>t</b>
REFERENCE	, £3: 1-D	ow	2-AF		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	'ION:		AP	************					
LITERATU		ER E	API						
		<b>!•</b>							

No. 177 n-Heptatriacontane NAME STRUCTURAL FORMULA CH3(CH2)35CH3 Molecular C37H76 Ref. Mole Molecular Weight 520.978 % Pur. Ref Ref. F.P. °C F.P. 100% 77.7 2 dt/dP f to °C/mm ۰ĸ g 0.1 mm 179.09 B. P. °C h BP 0.0764 5 504. 760 mm 2 0.3449 5 ſ١ to °K 404.28 100 4 g' 1.1590 5 30 mm 30 358.39 4 10 322.74 5 h' AHm cal/g 262.32 1 215.74 5 AHv cal/g 0.1 °K 50.91 0.1 mm Press. mm o 30 mm 5 <sup>t</sup>e 43.6 1973.0 5 ΒP 34.6 5 mi Density t<sub>e</sub> (d, e) 32.0 5 n' ۰ĸ g/ml 20°C 0.8179 2 31.54 0.8146 o'  $d_4^t$ 25 19.61 5 AHv/T 30 0.8113 4 a 355 Surface tension 65.75 to 5 0.8311 dynes/cm. 20°C 28.93 °C 562 0.0618 ь -0.0366 4 30 d' 210 61.96 28,00 to 40 27.10 Ref. Index e' 355 ٠c 5 5 20°C 1.4578  $n_D$ [P] Parachor 1.4559 d<sub>c</sub> g/ml 25 2 20°C vc ml/g t\_ °C 30 1.4539 4 30 t<sub>c</sub> "C" 40 0.7405 4 1477.2 5 P<sub>c</sub> mm Sugd. MR (Obs.) 173.77≠ 2 PV/RT Exp. L.1.%/wt. 173.066 MR (Calc.) 1.0489 0.1 mm 1.0000 2 5 (nD-d/2) 98. <sup>‡</sup> 30 mm 1.0000 Dispersion 2 2.13 5 Dielectric BP 0.8888 5 Flash Point °C A 355 to t t 0.8465 7.72381 Fire Point B 1570 °C 3138.26 5 M. Spec. c 144. 5 AHc kcal/m Ultra V. ΔHf 2.67222 A\*| 355 to X-Ray Dif. B\*| 570 °C ΔFf 3051.3 Infrared Viscosity Solubility in centistokes to °C Acetone Carbon tet. °C ŧς Benzene A' to Ether B' °C n-Heptane Bv Av C to Ethanol 1 °C A\*\*210 to 2.56511 2997.5 Water B'\*355 °C Water in (B<sup>V</sup>) Acl (AV) to Bc ۰c c<sub>p</sub> liq. ۰ĸ Cc Cryos. A° c<sub>p</sub> vap. consts. B° c, vap. te °C 564.63 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

								No. 178	
NAME	n	-Octa	tria	contane			STRUCTURAL	FORMULA	L
							CH (CH )	CH	
Mole	Ref.	Mo	lecul	I	Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub>	6 <sup>CH</sup> 3	
% Pur.		Fo	rmul	ar C <sub>38</sub> H <sub>78</sub>	Weight 535.0	04			
			Ref.			Ref.			Ref.
F.P. *C F.P. 1007	79.0	)	2	dt/dP			f to		1
B. P. °C	<u>'</u>		-	*C/mm 0,1 mm	180,62	5	g °K		
760 mm	511.		2	BP	0.076	5	$\left  \frac{\mathbf{h}}{\mathbf{f'}} + \frac{\mathbf{h}}{\mathbf{f'}} \right $		
100 30	410.5 364.3		4	t <sub>e</sub> 30 mm	0.03444 1.1679	5	g' to		
10	328.4	3	5	ΔHm cal/g	+	Н	h'		ĺ
1 0. 1	267.5		5	ΔHv cal/g	<del> </del>	$\vdash$	m   to		
Press. mn				0.1 mm	50.12	5	n °K	ŀ	
t <sub>e</sub>	1990.2		5	30 mm BP	42.9 34.1	5			⊢
Density g/ml 20°0	. 0.8	188‡	2	te te (d, e)	31.4	5	m' to		ļ
at 25	0.8	155	2	ΔHv/T <sub>e</sub>	31.0 19.59	<b>5</b>	0'		1
	0.8	122	4	d   361 to		5	Surface tension		$\vdash$
a b	-0.0		4	e   570 °C		5	dynes/cm. 20°C	28.99 28.07	5
Ref. Index	1		-	d'   216 to		5 5	40	27.17	5
n <sub>D</sub> 20°0	1.4	583 <sup>‡</sup> 564 <sup>‡</sup>	2	d <sub>c</sub> g/ml	0.0302		Parachor [P]		
30	1.4	543	4	V mi/g			20°C 30	•	
"C"	0.7	404	4	tc °C	1		40		_
MR (Obs.)			2	P <sub>c</sub> mm			Exp. L.1.%/wt.	1516.2	5
MR (Calc. (nD-d/2)	177.6	84 489 <sup>‡</sup>	5 2	0.1 mm	1.0000	5	u.	1	
Dielectric			5	30 mm BP	1.0000 0.8889	5	Dispersion	98.≠	2
A   361 to		3405	5	te	0.8458	5	Flash Point °C Fire Point		
B521*0	3172.5	6	5	t <sub>c</sub>			M Spec.		$\vdash$
A*  361 to	2.6	9112	5	ΔHc kcal/m ΔHf			Ultra V.		
B* 580°C			5	ΔFf		<u> </u>	X-Ray Dif. Infrared		
K				Viscosity centistokes			Solubility in +		
tk				η •c	;		Acetone Carbon tet.		
t			ļ		1		Benzene		
B'							Ether n-Heptane	1	
C'				B <sup>V</sup>   to			Ethanol Water		
A'* 216 to B'* 361 °C		8304	5	(BV)	-		Water in		
Ac to	<del></del>			(A <sup>V</sup> )					
Bc tc_	긔			c <sub>p</sub> liq. °K	: 1				
Cryos. A	+			11					
consts. B				P	`				
t <sub>e</sub> °C	572.63	3	5	c <sub>v</sub> vap.					
# for unde							grams/100 gra	ms solven	t
REFEREN	ES: 1-E	)ow	2-AI		Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE: PURIFICA	PION:		API						
LITERATU		ERE	API						
	REF	EI	-UE	••					

<u></u>			<del></del>		- т			No. 17	9
NAME	n-Nona	riaco	ntane			ST	RUCTURAL	FORMUL	A
			т		$\dashv$		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>37</sub>	сң	
Mole % Pur.		olecul ormul		Molecular Veight 549.	030		J 2 31	3	
		Ref			Ref.				Ref.
F.P. *C	80.3	2	dt/dP			f	to		
F.P. 100	6		°C/mm		ا ۔ ا	g	*K		
B.P. *C	]		0.1 mm BP	182.18 0.076	5	_b _			
760 mm 100	518. 416.88	2	t <sub>e</sub>	0.0345	5	f'	to		İ
30	370.30	4	30 mm	1.1755	5	g'	*K		
10 1	334.10 272.69	5	∆Hm cal/g			h' j			-
0.1	225.33	5	ΔHv cal/g 0.1 mm	40.37	ا ۔ ا	m	to •K		
Press. mr	n 2007.1	5	30 mm	49.37 42.3	5	l ° ¦			
t <sub>e</sub> Density	<del> </del>	.   -	BP	33.6	5	m'	to		
g/ml 20°	0.8197	2	t <sub>e</sub> (d, e)	30.9 30.43	5	n'	°K		
dt 25 4 30	0.8164 0.8131	2	AHv/Te	19.54	5	لـنـــا			
a 30	0.8329	4	d 366 to	64.11	5		face tension es/cm. 20°C	29,07	5
Ъ	-0.0366	4	d' 220 to	0.0589 60.35	5	8,44	30	28.14	5
Ref. Index	1 4500	4	e' 366 °C	0.0488	5	<u> </u>	40	27.24	5
<sup>n</sup> D 20°0	1.4588 1.4568	2 2	d <sub>c</sub> g/ml			Par	achor [P] 20°C		
30	1.4549	4	v <sub>c</sub> ml/g t <sub>c</sub> °C				30		
"C"	0.7404	4	P <sub>c</sub> mm			ļ	40 Sugd.	1555.2	5
MR (Obs. MR (Calc.		2	PV/RT		1	Exp	. L.1.%/wt.		
(nD-d/2)	182.302 1.0490	5 2	0.1 mm	1.0000	5	_	u.	98. <sup>‡</sup>	
Dielectric		5	30 mm BP	1.0000 0.8881	5		persion	70.	2
A 366 to			t <sub>e</sub>	0.8451	5		sh Point °C e Point		
B 1523 °C	3207.87	5	tc AHc kcal/m		-	M.	Spec.		T
A* 366 to	2,7103	, 5	ΔHf				a V. Lay Dif.		
B* 587 °C	3120.5	5	ΔFf				ared		
K — —			Viscosity centistokes		ł		bility in +		
t <sub>k</sub>			η °c				etone rbon tet.		
, x	_					Ве	nzene		
A'  to B'  *C					ļ		her Hep <b>tane</b>		
C'			B <sup>V</sup> to C			Et	hanol		
A!* 220 to B!* 366 °C			$\frac{A^{\vee}}{(B^{\vee})!} - {^{\circ}C}$				ter ter in		
Acl to	<del></del>	5	(A <sup>V</sup> )						
Bc t °C					<del>                                     </del>				
			р -						
Cryos, A' consts, B'			c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	580,62	5	c <sub>v</sub> vap.						
	ercooled liqu						ams/100 gra		t
	CES: 1-Dow		PI 3-Lit. 4-0	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP							
PURIFICA		AP							
LITERAT	JRE REFER	ENCE	5:						

								No. 180	
NAME	n	-Tet	racor	ntane		_	STRUCTURAL	FORMULA	
Mole % Pur.	Ref.	Mo Fo	lecul:		Molecular Weight 563.0	56	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>38</sub>	сн <sub>3</sub>	
			Ref.	T		Ref			Ref
F.P. *C	81.5		2	dt/dP					
F.P. 100%				*C/mm			f to		
B. P. °C	<u> </u>		H	0.1 mm	183.66	5	h		
760 mm	525.		2	BP	0.077 0.03460	5	I +		
100 30	423.1 376.2		4 4	t <sub>e</sub> 30 mm	1.1871	5	g' to		
10	339.7		5		1.1011	-	h'		
1	277.9		5	ΔHm cal/g		$\vdash$	m to		<del>                                     </del>
0.1	230.1	7	5	ΔHv cal/g	48,68	5	n K		
Press. mm	2025.3		5	0.1 mm 30 mm	41.7	5	•		
e Density			-	BP	33.1	5	m'   to		
g/ml 20°C	0.8	205‡	2	te (d, e)	30.5 29.78	5	n' *K		
t 25	0.8	172*	2	ΔHv/T	19.41	5	0'		1
	0.8	139	4	d   372 to		5	Surface tension		
a b	0.8		4	e   585 °C		5	dynes/cm. 20°C	29.13	5
	-0.0	300	4	d' 225 to	59.68	5	30 40	28.20 27.30	5
Ref. Index n <sub>D</sub> 20°C	1.4	593‡	2	e'   372 °C	0.0478	5	Parachor [P]		Ť
D 25	1.4	573°	2	d <sub>c</sub> g/ml			20°C		
30	1.4	553	4	vc ml/g tc °C			30		
"C"	0.7		4	P <sub>c</sub> mm		1	40 Sugd	1594.2	5
MR (Obs.)	187.6		2	PV/RT	<del> </del>	-	Exp. L.1.%/wt.		<u> </u>
MR (Calc.) (nD-d/2)	186.9	2 490≠	5 2	0.1 mm	1.0000	5	u.	1	
Dielectric			-	30 mm	1.0000	5	Dispersion	98. <sup>‡</sup>	2
A 1372 to	2.1	5334	5	BP t <sub>e</sub>	0.8881 0.8446	5	Flash Point °C		
B 1526 °C	3240.2		5	tc		] •	Fire Point		<u> </u>
c			1 1	ΔHc kcal/m			M Spec. Ultra V.		İ
A*   372 to	2,7	2652	5	ΔHf ΔFf			X-Ray Dif.		
B* 1595 °C	3152.5		5		<del> </del>	-	Infrared		
c				Viscosity centistokes		:	Solubility in +		
t <sub>k</sub> to				7 °C			Acetone Carbon tet.		
*x 1				,			Benzene		l
A'   to B'   °C					ľ		Ether		
E; L E				B <sup>V</sup>   to	<u> </u>		n-Heptane Ethanol		
A!* 225 to	2 6	602	5	AV I C			Water	1	
B'* 372 °C	3095.5	<del>-</del>	5	(B <sup>V</sup> )	1		Water in		-
Ac  to				(A <sup>V</sup> )					
Bc tc_°C				c <sub>p</sub> liq. °K		1			
			$\vdash$						
Cryos. A° consts. B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	588.65	5	5	c <sub>w</sub> vap.					
for under	rcooled l	iquid	1				grams/100 grai	ms solven	t
REFERENC	ES: 1-D	ow	2-AF	Pl 3-Lit. 4-0	Calc. from det	t. da			
SOURCE:			API						
PURIFICAT	ON:		API						
LITERATUR	E REF	ERE	NCES	:					

No. 1 NAME Methyl fluoride STRUCTURAL FORMULA Fluoromethane CH<sub>3</sub>F Mole Ref. Molecular Molecular % Pur Formula CH<sub>3</sub>F Weight 34.034 Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm ۰ĸ 1 g 25°C B. P. °C h 0.0238 BP 2 760 mm -78.35 0.0326 5 ſ١ t<sub>e</sub> 100 -108.68 2 to g' <u>• к</u> 30 -122.19 2 30 mm 0.3392 5 10 -132.50 2 h' ∆Hm cal/g 1 -149.60 5 m to AHv cal/g Pressure n °K 25°C mm 25°C o 30 mm 130.76 511.5 5 t<sub>e</sub> BP 118.46 5 m' to Density te te (d, e) 120.56 5 5 n' ۰ĸ g/ml 20°C 120.37 ٥' 25 d t 4 ΔHv/T<sub>e</sub> 21.83 5 30 -125 Surface tension d 96.49 dynes/cm. 20°C 5 <u>-65</u> °C 0.2804 ь 30 á⊓Ì to 40 Ref. Index e¹ °C 20°C [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 0.275 5 25 20°C vc ml/g tc °C 3.63 5 30 30 3 44.6 40 "C"  $P_c$  mm 91.8 5 44080. 3 Sugd MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 6.668 5 25°C (nD-d/2)1.0000 30 mm 5 Dispersion Dielectric BP 0.9700 Flash Point C 0.9782 5 A -122 to 7.09761 2 Fire Point 0.275B 1\_-20°C 740.22 M. Spec. С 253.89 2 AHc kcal/m Ultra V. A\* - 122 to ΔHf 1.26539 5 X-Ray Dif. AFf B\*|\_-65°C 688.81 Infra red Viscosity Solubility in centistokes c Acetone to Carbon tet. 00 tx °C Benzene œ A' to Ether an B' <u>°C</u> n-Heptane 00 B<sub>v</sub> | C to Ethanol °C Water A1\* to Water in (BV) B'\* °C to Acl -20to 7.81764 (A<sup>V</sup>)| °C Bc tc °C 1113.00 cp liq. ۰ĸ Сс 306.57 4 Cryos, A° consts, B° °K cp vap. c<sub>v</sub> vap. t<sub>e</sub> °C -85.23  $T_R = 0.75T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: Literature PURIFICATION: Literature LITERATURE REFERENCES: 3 Chem. Rev. 52, No. 1, 117 (1953), Kolbe and Lynn; 3' Physia 14, 104 (1948), Nichels and Wassenaar.

							No. 2	
NAME	Perfluor	0-n-l	nexane			STRUCTURAL	FORMULA	
						an (an)	C.P.	
Mole	Ref. Mo	lecul		Molecular		CF <sub>3</sub> (CF <sub>2</sub> ) <sub>4</sub>	CF <sub>3</sub>	
% Pur.		rmul		Weight 338.0	60			
		Ref.			Ref.			Ref.
F.P. °C	-87.1	3	dt/dP			f to		
F. P. 1009 B. P. °C	<b>-</b>	├	*C/mm 25*C	0.1045	5	g <u>*K</u>		
760 mm	57.11	3	BP	0.03827 0.03573	<b>5</b>	h f' to		-
100 30	8.27 -13.52	5	t <sub>e</sub> 30 mm	0.5473	5	g' to		l
10	-30.1	5	ΔHm cal/g	0.5415	-	h'		
Pressure	-57.8	5	ΔHv cal/g	1		m to		
mm 25°C	218.9	5	25°C	22.50	5	n <u>*K</u> _		
t <sub>e</sub>	890.9	5	30 mm BP	24.14 21.00	5	<u> </u>		├
Density g/ml 20°	1,6995	3	te (d, e)	19.62 20.8	<b>5</b>	m'   to		
at 25	1.6851	4	ΔHv/Te	19.80	5	0'		
	1.6695	3	d   -14 to	<del>                                     </del>	5	Surface tension		
a b	1.7612 -0.0029	4		0.0444	5	dynes/cm. 20°C	11.92 10.93	3
Ref. Inde			e' '°C			40	10.10	3
n <sub>D</sub> 22°0	1.2515	3	d g/ml vc ml/g			Parachor [P] 20°C	369.4	4
30			v <sub>c</sub> ml/g t <sub>c</sub> °C	Ì		30	367.9	4
"C"		<u> </u>	P <sub>c</sub> mm	ł		40 Sugd.	365.6 364.8#	4 5
MR (Obs. MR (Calc.		4 5	PV/RT			Exp. L.1.%/wt.		Ť
(nD-d/2)	0.4146	4	25°C 30 mm	0.9839 1.0000	5	u. Dispersion		
Dielectric			BP	0.9600	5	Flash Point °C		
A -14 t		4	t <sub>e</sub> t <sub>c</sub>	0.9559	5	Fire Point		
B L 66_*	C 1205.37 227.0	4	ΔHc kcal/m			M Spec.		
A*   -14 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* L 65 °	<u>C</u> 1130.9	5	Viscosity	· · · · · · · · · · · · · · · · · · ·		Infrared		L
c	_	1	centistokes	0.41/0		Solubility in + Acetone		
t <sub>x</sub> t			7 20 ℃ 40	0.4160 0.3270	3	Carbon tet.		i
A' t		T	50	0.2916	3	Benzene Ether		
B' 3	드	1	B 10 to	487.4	4	n-Heptane Ethanol		
A'* t	<u>-                                    </u>		AV   60 °C	3.95674	4	Water		1
B!* °	<del>-  </del>	<u> </u>	(B <sup>V</sup> ) to			Water in		-
Ac t			(A <sup>V</sup> )  °C	<del></del>				
Cc - c-		<u> </u>	c <sub>p</sub> liq. °K					
Cryos, A consts, B			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	61.81	5	c <sub>v</sub> vap.					
						grams/100 grai	ns solven	t
REFEREN	CES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc, from det	. da			
SOURCE:		Li	t.					
PURIFICA	<del></del>	Li						
LITERAT	JRE REFERE	NCES	3: 3 JACS <u>74,</u>	3771 (1952) St	iles	and Cady		
# atomic:	ref. index for	Fluor	ine in polyfluor	ro compounds	= 1.2	22		
# atomic	P] for Fluorin	e in ]	polyfluoro comp	oounds = 24.				

No. 3 Perfluoro-2-methylpentane STRUCTURAL FORMULA NAME CF2CF (CF2)2CF3 Molecular C6F14 Mole Ref. Molecular Weight 338.060 % Pur Formula Ref. Ref. Ref. F. P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.1065 5 B.P. °C h ΒP 0.03878 5 760 mm 57.73 3 0.03613 5 f١ to 100 8.30 5 °<u>K</u> g' 30 -13.69 5 30 mm 0.5521 10 -30.5 5 h' ∆Hm cal/g -58.3 5 m to AHv cal/g Pressure °K n 25°C 22.43 5 216.8 mm 25°C 5 5 o 30 mm 23.90 5 831. t<sub>e</sub> BP 5 21.00 m' to Density te te (d, e) 19.36 5 n' g/ml 20°C °K 1.7326 3 5 20.89 ٥' ď4 25 1.7169 ΔHv/T<sub>e</sub> 19.62 5 30 1.7011 4 Surface tension -14 to 23.34 5 1.7946 3 dynes/cm. 20°C 12.12 60 ℃ 0.0406 ь -0.0031 4 11.41 30 3 ăח 40 10.49 3 Ref. Index e' °C 22°C 1.2564 3 [P] nD Parachor d<sub>c</sub> g/ml 25 20°C 364.1 vc ml/g 30 30 365.2 4 t<sub>c</sub> 364.2 40 4 "C" 0.1688 4 P<sub>c</sub> mm Sugd. 364.8 MR (Obs.) 31.544 31.588 5 PV/RT Exp. L. l. %/wt. MR (Calc.) 25°C 0.9933 5 (nD-d/2) 0.3901 5 1.0000 30 mm 5 Dispersion Dielectric BP 0.9600 Flash Point °C 0.8954 A -14 to 7.08320 Fire Point B 1 73 °C 1198.63 M. Spec. Ultra V. C 227.5 4 ∆Hc kcal/m ΔHf A\* -14 to 2.02668 5 X-Ray Dif. ΔFf B\*| 65 °C 1118.1 Infrared ĸ Viscosity Solubility in centistokes Acetone 0.4538 20 °C 3 Carbon tet. t<u>x</u> °C 30 0.3979 3 Benzene 50 0.3130 3 to Ether B١ °C n-Heptane C' 10 to Ethanol **v** 1 60 °C 3,91880 A'\* Water to Water in B'\* °C (B<sup>V</sup>) to Acl (A<sup>V</sup>)| to °C Βc ۰c ۰ĸ c<sub>p</sub> liq. Cc' Cryos. A° cp vap. °K consts. B° c, vap. te °C 5 60.39  $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula Lit. SOURCE: **PURIFICATION:** LITERATURE REFERENCES: 3 JACS 74, 3771 (1952) Stiles and Cady

# Atomic ref. index for Fluorine in polyfluoro compounds = 1.22

Mole % Pur.  F.P. °C F.P. 100%  B.P. °C 760 mm 100 30 10 1  Pressure mm 25°C te  Density g/ml-24°C dt -20 4 -10	-97.720  -24.22 -62.91 -80.03 -93.05 -114.58  4309.7 662.9  0.997 0.990 0.973	Ref.  2 2 2 2 5 5 32 32 32		Molecular Weight 50.44  0.0081 0.0305 0.0340 0.4292	Ref. 5 5 5 5 5 5	f   to %K   to g'   to g'   °K		Ref
% Pur.  F.P. °C F.P. 100%  B.P. °C 760 mm 100 30 10 1 Pressure mm 25°C te Density g/ml-24°C at -20	-97.720  -24.22 -62.91 -80.03 -93.05 -114.58  4309.7 662.9  0.997 0.990 0.973	Ref. 3' 2 2 2 2 5 5 5 5 5 32	dt/dP °C/mm 25°C BP te 30 mm  AHm cal/g AHv cal/g 25°C 30 mm	0.0081 0.0305 0.0340 0.4292	<b>Ref. 5</b> 5 5	f   to o'K   h + to		Ref
F. P. 100%  B. P. °C  760 mm  100  30  10  1  Pressure mm 25°C  te  Density g/ml-24°C at -20	-97.720  -24.22 -62.91 -80.03 -93.05 -114.58  4309.7 662.9  0.997 0.990 0.973	Ref. 3' 2 2 2 2 5 5 5	dt/dP °C/mm 25°C BP t 30 mm  AHm cal/g AHv cal/g 25°C 30 mm	0.0081 0.0305 0.0340 0.4292	<b>Ref. 5</b> 5 5	$\frac{g}{f'} + \frac{g}{to}$		Ref
F. P. 100%  B. P. °C  760 mm  100  30  10  1  Pressure mm 25°C  te  Density g/ml-24°C at -20	-24.22 -62.91 -80.03 -93.05 -114.58 4309.7 662.9 0.997 0.990 0.973	2 2 2 2 5 5	°C/mm 25°C BP t 30 mm  AHm cal/g AHv cal/g 25°C 30 mm	0.0305 0.0340 0.4292	<b>5</b> 5 5	$\frac{g}{f'} + \frac{g}{to}$		
F. P. 100%  B. P. °C  760 mm  100  30  10  1  Pressure mm 25°C  te  Density g/ml-24°C at -20	-24.22 -62.91 -80.03 -93.05 -114.58 4309.7 662.9 0.997 0.990 0.973	2 2 2 2 2 5 5	°C/mm 25°C BP t 30 mm  AHm cal/g AHv cal/g 25°C 30 mm	0.0305 0.0340 0.4292	5	$\frac{g}{f'} + \frac{g}{to}$		
B. P. °C 760 mm 100 30 10 1 Pressure mm 25°C t <sub>e</sub> Density g/ml-24°C <sub>1</sub> t -20	-62.91 -80.03 -93.05 -114.58 4309.7 662.9 0.997 0.990 0.973	2 2 2 5 5 5	25°C BP te 30 mm AHm cal/g AHv cal/g 25°C 30 mm	0.0305 0.0340 0.4292	5	$\frac{\mathbf{h}}{\mathbf{f'}} + \frac{1}{\mathbf{to}}$		
760 mm 100 30 10 1 Pressure mm 25°C te Density g/ml-24°C 1t - 20	-62.91 -80.03 -93.05 -114.58 4309.7 662.9 0.997 0.990 0.973	2 2 2 5 5 5	t <sub>e</sub> 30 mm  ΔHm cal/g  ΔHv cal/g 25°C 30 mm	0.0340 0.4292 89.54	5	$\frac{1}{\mathbf{f'}} + \frac{1}{\mathbf{to}}$		
30 10 1 Pressure mm 25°C te Density g/ml-24°C at -20	-80.03 -93.05 -114.58 4309.7 662.9 0.997 0.990 0.973	2 2 5 5 5	30 mm  AHm cal/g  AHv cal/g  25°C  30 mm	0.4292	1 1			
Pressure mm 25°C te Density g/ml-24°C at -20	-93.05 -114.58 4309.7 662.9 0.997 0.990 0.973 0.9576	2 5 5 5	ΔHm cal/g ΔHv cal/g 25°C 30 mm	89.54	5			1
Pressure mm 25°C te Density g/ml-24°C t - 20	-114.58 4309.7 662.9 0.997 0.990 0.973 0.9576	5 5 5	ΔHv cal/g 25°C 30 mm			h'		ĺ
mm 25°C te Density g/ml-24°C t -20	0.997 0.990 0.973 0.9576	5 32	25°C 30 mm					ـــ
mm 25°C te Density g/ml-24°C t -20	0.997 0.990 0.973 0.9576	5 32	30 mm			m to		1
Density g/ml-24°C t -20	0.997 0.990 0.973 0.9576	32			5	n e K		
Density g/ml-24°C t -20	0.990 0.973 0.9576	3 <sup>2</sup>		114,01	5			<u> </u>
t -20	0.990 0.973 0.9576	32	t <sub>e</sub> ,	102.04	5	m'   tó		1
	0.973 0.9576	32	te (d, e)	102.02	5	n' °K		1
	0.9576	32	ΔHv/T <sub>e</sub>	20.55	5	0,		
		-	d   -80 to	<u> </u>	5	Surface tension		T
a b		4	e   -20 °C		5	dynes/cm20°C		
	-0.0 <sub>2</sub> 184	4	_d' to	1		7 -10 40		l.
Ref. Index		ł	e' j °C	:	Ш			⊬
n <sub>D</sub> 20°C			d g/ml vc ml/g tc °C	0.353	3'	Parachor [P] 20°C		1
30			v ml/g	2.833	3'	30		
"C"			~	ı	l ľ	40		_
MR (Obs.)		$\vdash$	P <sub>c</sub> mm	50084.	3	Sugd.	110.4	5
MR (Calc.)	11.685	5	PV/RT	0 0073	ا ہ ا	Exp. L.1.%/wt.		1
(nD-d/2)			25°C 30 mm	0.9073	5	u. Dispersion		
Dielectric			BP	0.9660	5			┼
A -80 to	6.99445	2	t <sub>e</sub>	0.9692	5	Flash Point °C Fire Point		
B [_40_°C	902.45	2	tc	0.2752	4			+
С	243.60	2	AHc kcal/m	1		M Spec. Ultra V.		ì
A* -90 to	1.24262		ΔHf ΔFf			X-Ray Dif.		
B* ∟ 40 °C	841.23	5		<del></del>		Infrared		
c			Viscosity centistokes			Solubility in +		
t <sub>L</sub> to		1 1	7 °c	1	ĺĺ	Acetone	oc	1
t <mark>x</mark> i °C			•			Carbon tet. Benzene	oc oc	
A' to				1	]	Ether	∞ ∞	
B' ∟ _ ° □			B <sup>V</sup>   to	<del> </del>	$\vdash$	n-Heptane	øo.	
		$\vdash$	B' to			Ethanol Water	oc	1
A'* to B'* °C		1 1		-[	1	Water in		
	5.01140	+	(BV)		ŀ			+
Ac 40 to Bc t <sub>c</sub> °C	7.81148 1433.6	4	(A <sup>V</sup> )	ļ	Ш			
Cc	317.5	4	c <sub>p</sub> liq. °K		1			İ
Cryos, A°			1					
consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	-27.34	4	c <sub>w</sub> vap.	1	ļ			
		لــَــا	<u> </u>	<u> </u>	ii	+ (100	·	<del></del>
REFERENCE	ES: 1-Dow	2-AF	OT 2 Tit 4 4	Cala from de		grams/100 gram		ıt
SOURCE:	DUW	2-AF	PI 3-Lit. 4-0	Jaic. Irom de	. dat	a 5-Calc, by for	u.a	
	ON.							
PURIFICATION			<del></del>		115=			
LITERATUR! nans; 3º Lang	E R <b>EFER</b> E: ge	NCES	: 3 Chem. Rev	. <u>52,</u> No. 1, 117	(1953	3), Kolbe and Lynn	; 3' Timn	ner-
•	-							

							No. 5
NAME	Methylene	chlo	ride			STRUCTURAL	FORMULA
	Dichloron	netha	ne			CH CI	
Mole % Pur. 99.	Ref. Mol			Molecular Weight 84.94		сн <sub>2</sub> сі	
·	1	Ref.			Ref.	<del></del>	Ref.
F.P. °C F.P. 100%	-95.14	1	dt/dP °C/mm 25°C	0.0575	5	f to g° <u>K</u>	
B. P. °C 760 mm	39.75	1	BP	0.03692	4 5	h (	<u> </u>
100 30	-7.27 -28.18	1 4	t <sub>e</sub> 30 mm	0.03406	5	f' to	
10	-44.12	4	ΔHm cal/g	0.5248 16.89	4	h'	
1	-70.55	5	ΔHv cal/g		-	m to	
Pressure mm 25°C	435.86	5	25°C	80.55	5	n <u>•K</u>	
t <sub>e</sub>	841.21	5	30 mm BP	89.20 78.18	5	m' to	<b></b>
Density g/ml 20°C	1.32554	1	te te (d, e)	77.74	5	n'   °K	
dt 25	1.31630	1	ΔHv/T <sub>e</sub>	20.91	5	0'	
	1.30700	4	d -28 to	84.63	5	Surface tension	
a b	1.36277	4	e 43 °C to	0.1622	5	dynes/cm. 20°C 30 40	28.00 1 26.41 1 24.84 1
Ref. Index n <sub>D</sub> 20°C	1.42416	1	e'   °C	ļ		Parachor [P]	24.04
D 25	1.42115	1	d <sub>c</sub> g/ml v <sub>c</sub> ml/g			20°C	149.4 4
"C"	1.41519 0.4252	4	tc ℃	252.	5	30 40	149.6 4 149.9 4
MR (Obs.)	16, 357	4	P <sub>c</sub> mm	56400.	5	Sugd.	147.6 5
MR (Calc.)	16.552	5	PV/RT 25°C	0.9727	5	Exp. L.1.%/wt.	
(nD-d/2)	0.76139	4	30 mm	1.0000	5	Dispersion	
Dielectric A -28 to	7.07138	1	BP t	0.9600 0.9573	5	Flash Point C	
B [ 121 °C	1134.6	1 5	tc			Fire Point M. Spec.	
C A* -28 to	1,46352	5	ΔHc kcal/m ΔHf			Ultra V.	
B*[_ 53 °C	1063.0	5	ΔFf			X-Ray Dif. Infrared	
K — —	ļ		Viscosity centistokes	Ì		Solubility in +	
t <sub>k</sub> to			<b>່າ</b> 20 °C	0.3868	1	Acetone Carbon tet.	
t <sub>x</sub> °C		<u> </u>	40 60	0.3530 0.3274	1 1	Benzene	
B' ℃			80	0.3000	1	Ether n-Heptane	
C'	ļ		B <sup>V</sup>   10 to A <sup>V</sup>   50 °C	182.34 2.9656	4	Ethanol Water	
A'* to B'* °C			(B <sup>V</sup> )  50 to	223.35	4	Water in	
Ac   121 to	7.50819	5	(A <sup>V</sup> )  90 °C	₹. 8447	4		
Bc tc °C	1462.59 278.60	5 5	c <sub>p</sub> liq. °K				
Cryos. A° consts. B°	0.02286	1	c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	42.63	5	c <sub>v</sub> vap.				
$T_{R} = 0.75$		ــــــــــا	L	L	L	+ grams/100 gra	ms solvent
	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da		
SOURCE:		Ι	ow, dist.				
PURIFICAT	ION:	Ι	Dow, dist.				
LITERATU	RE REFERE	NCE	S:				

										No. 6	
NAME	Chl	lor	for	m				STR	UCTURAL 1	FORMUL/	A.
	Tri	chl	oro	meth	ane				C1		
Mole	Re	,	Mo	lecul		Molecular			сі с н		
% Pur. 99.		<u> </u>	Fo	rmul	ar CHCl3	Weight 119.	389		<b>ċ</b> ı		
				Ref.			Ref				Ref.
F.P. C F.P. 1007	-63.	59		1	dt/dP			f	to		
B. P. *C	<u>'</u>			$\vdash$	*C/mm 25*C	0.1205	5	g	<u>*K</u>		
760 mm	61.			4	BP	0.04107 0.0358	5 5	h f'	<del>                                      </del>		
100 30	9. -13.	79 06		5 <b>5</b>	t <sub>e</sub> 30 mm	0.5718	5	g'	to		
10	-30.	39		5	ΔHm cal/g	17, 62	14	h'	1		
l Pressure	-58.	93		5	ΔHv cal/g	1	H	m	l to		
mm 25°C	197.	4		5	25°C 30 mm	62.11 65.66	5 5	n o	<u></u>		
t <sub>e</sub>	906.			5	BP BP	57.30	5		l to		-
Density g/ml 20°C	1.	483	16	1	te te (d, e)	56.68 56.68	5	m' n'	*K_		
dt 25 4 30	1.	479	85	1	ΔHv/T <sub>e</sub>	19.87	5	0'	i		
4 30 a		476		4	d   -13 to		5		ace tension	27 10	
b		036		4	-d-1 -70 to		5	Jayne	ss/cm. 20°C 30	27.10 25.78	1 1
Ref. Index					e :				40	24.53	1
n <sub>D</sub> 20°C		445 442		1 1	d <sub>c</sub> g/ml			Par	chor [P] 20°C	183.7	4
50		428		1	t <sub>c</sub> *C	271.	5		30	182.2	4
"C"		398	4	4	P <sub>c</sub> mm	39741.	5		40 Sugd.	180.5 184.8	5
MR (Obs.) MR (Calc.			,	4 5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	-1	704		4	25°C 30 mm	0.9990	5 5	Dist	u. ersion		
Dielectric		639		1	BP	0.9610 0.9561	5		h Point °C		+-
A   -15 to B   135°C		903	28	4	t <sub>e</sub> t <sub>c</sub>	0.7561			Point		
c	227.			5	ΔHc kcal/m			M S			
A* -15 to		412	86	5	ΔHf ΔFf			X-R	ay Dif.		ŀ
B*	- 1000.			, ,	Viscosity		$\Box$	<b></b>	ared bility in +		-
t <sub>k</sub>					centistokes 7 20 °C	0,3789	1	Acc	etone		1
tx   °(					30	0.3481	1 1		rbon tet. nzene		
A'   to					40 50	0.3206	1 1	Eth	er		
B' L _ •	<u>- </u>				B   10 to	338.5	4		leptane anol		
A'* to					A <sup>V</sup>   60 °C	Z. 42528	4	Wa	ter ter in		
B'* *(		226	-	_	(B <sup>V</sup> ) to	ł					<del>                                     </del>
Bcit *C	1498.	336		5	(A <sup>V</sup> )  °C		-				
Cc	276.			5	c <sub>p</sub> liq. ∘K	1					
Cryos. A <sup>e</sup>		024	18	1	c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	67.	32		5	c <sub>v</sub> vap.	<u> L</u>		<u> </u>			<u>L</u>
T <sub>R</sub> = 0.7		D-		2 47	37 3 7 1 4	<del></del>			ams/100 gran		t
SOURCE:	JES: 1-	- D0	<u> </u>	Z-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula	
PURIFICA'	TION:				istillation	· · · · · · · · · · · · · · · · · · ·			<del></del>		
LITERATU		FE	REI								
L											

								No. 7	
NAME	Carbon T	etra	hloride			ST	RUCTURAL	FORMUL	A
ſ									
	TI						C C14		
Mole % Pur. 99.		ecul mul		Molecular Veight 153.83					
70 Fur. 77.	74   1   101	Ref		vergnt 155.05	Ref.				Ref.
F, P, °C		1.61	dt/dP		IXCI.		1		1
F.P. 100%	-22.99	1	°C/mm			f g	to K		
B. P. *C			25°C BP	0.1972 0.04322	5	h	I		l
760 mm 100	76.54 21.81	1 4	te	0.0360	5	f'	to		
30	-2.31	4	30 mm	0.6040	5	g'	<u>'°K</u>		ļ
10 1	-20.62 -50.82	5 <b>5</b>	ΔHm cal/g	5.04	4	h'	<u> </u>		-
Pressure			ΔHv cal/g	50.00	_	m n	to to		l
mm 25°C	115.2	5	25°C 30 mm	50.39 52.31	5	0			
Densites	960.	5	BP	46.41	5	m'	to		t
Density g/ml 20°C	1.59397	1	t <sub>e</sub> (d, e)	45.89 45.90	5	n' o'			
d <sup>t</sup> 25 4 30	1.58429	1 4	AHv/T	19.74	5	6.	L		<u> </u>
4 30 a	1.57456	4	d -5 to	52.14	5		face tension	26 04	,
b	-0.00190	4	-å-   _90_ °C to	0.0736	5	gyn ayn	es/cm. 20°C 30	26.84 26.17	1
Ref. Index			e' . °C				40	25, 51	1
<sup>n</sup> D 20°C	1.46005 1.45704	1 1	d <sub>c</sub> g/ml	0.5576	3	Par	rachor [P]	219.7	4
30	1.45409	4	vc m1/g tc °C	1.7934 283.15	3		30	221.0	4
"C"	0.3817	4	P <sub>c</sub> mm	34200.	3		40 Sugd.	222.1 222.0	5
MR (Obs.) MR (Calc.	26.44	5	PV/RT		$\vdash$	Ext	L.1.%/wt.		†
(nD-d/2)	26.286 0.66306	5	25°C 30 mm	0.9973 1.0000	5		u.		
Dielectric			BP	0.9703	5		persion sh Point C	N	┼
A -15 to	6.93390	4	te tc	0.9645 0.2718	5		e Point	None None	
B 1138 ℃	1242.43 230.0	4	ΔHc kcal/m	0.2110	<u> </u>		Spec.		1
A* -5 to	1,51283	5	ΔHf	! 		Ult:	ra V. Ray Dif.		1
B*[100 °C	1155.8	5	ΔFf				ared		1
с			Viscosity centistokes				ubility in +		
t <sub>k</sub> to			η 20 °C	0.6091	1 1		etone rbon tet.		
A' to	<del> </del>		60	0.47364 0.3825	i		nzene her		
B' °C	.				$\vdash$	n-	Heptane		
C'	<del> </del>		B <sup>V</sup>   10 to A <sup>V</sup>   70 °C	493.5 Z.10154	4		hanol ater		
A'* to B'* °C			(B <sup>V</sup> )  to		-		ater in		
Acl 138 to	7.3703	5	(A <sup>V</sup> )  °C						
Bc tc °C	1584. - 277.	5	c <sub>p</sub> liq. °K						
Cryos. A°	0.00629	1	c <sub>p</sub> vap.300°K	0,206	31				
consts. B°				0.200					
t <sub>e</sub> °C	84.39	5	c <sub>v</sub> vap.			<u> </u>			
$T_{\mathbf{R}} = 0.79$							ams/100 gra		ıt
	ES: 1-Dow		PI 3-Lit. 4-0	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			w, Lit.						
PURIFICAT			stillation, Lit.	m:					
LITERATU	KE REFERE	NCES	5: 3 Young; 3'	Timmerman	.8				

							No. 8	
NAME	Methylen	e bro	mochloride			STRUCTURAL 1	FORMULA	r
	Bromoch	loror	nethane			CH,C1 E	sr	
Mole % Pur.	Ref. Mo	lecul rmul		Molecular Weight 129.3	99	33.233		
		Ref.			Ref.			Ref.
F.P. ℃	<u> </u>		dt/dP			f to		
F.P. 100% B.P. °C	<u>'</u>	-	*C/mm 25*C	0.1514	5	g <u>*K</u> _		
760 mm	68. 11	4	BP	0.04074 0.035 <b>0</b>	5	h f' to		-
100 30	16.69 -5.89	5	t <sub>e</sub> 30 mm	0.5644	5	g' to		
10	-22.98	5	ΔHm cal/g	0.3044	-	h'		l
<u>1</u>	-51.09	3	ΔHv cal/g			m   to		
Pressure mm 25°C	147.2	5	25°C	60.68	5	n °K_		
t <sub>e</sub>	925.	5	30 mm BP	64.81 55.43	5	- <del></del>		├
Density g/ml 20°C	1.93439	1	to (d. e)	54.69 54.64	5	m' to		
at 25	1.92292	1	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	20.36	5	o'		
	1.91143	4	d   -6 to	64.06	5	Surface tension		
a b	1.98028	4	_e	0.1267	5	dynes/cm. 20°C	33.32 31.87	1
Ref. Index			d' to			40	30.39	ī
n <sub>D</sub> 20°C	1.48376 1.48076	1 1	d g/ml	0,625	5	Parachor [P]	140.7	
30	1.47761	4	v <sup>c</sup> ml/g t <sub>c</sub> °C	1.60	5	20°C 30	160.7 160.9	4
"C"	0.3297	4	11 -	297. 45600.	5	40 Sugd	161.3 161.3	4 5
MR (Obs.)		4	P <sub>c</sub> mm	45000.	-	Exp. L.1.%/wt.	101.3	-
MR (Calc. (nD-d/2)	19.450 0.51657	5 4	25°C	0.9919	5	u.		
Dielectric			30 mm BP	1.0000 0.9618	5 5	Dispersion		
A -6 to		4	te	0.956 <b>3</b> 0.265	5	Flash Point °C Fire Point		ĺ
B 1_297 °C	216.	4	t <sub>c</sub> ΔHc kcal/m	0.203	-	M Spec.		
A*  -6 to	<del></del>	5	ΔHf			Ultra V. X-Ray Dif.		
B* ∟ 84 °C	1060.4	5	ΔFf	<u> </u>	-	Infrared		
c	_		Viscosity centistokes			Solubility in + Acetone		
tk   to			7 20 °C	0.3486 0.2949	1	Carbon tet.		
A' to		-	60	0.2569	i	Benzene Ether	İ	
B' <u>°</u>	<u>:</u>	1	B <sup>V</sup>   10 to	222.0	-	n-Heptane		
A'* to			B 10 to A 70 °C	323.8 2.43796	4	Ethanol Water		
B'* °C			(BV) to			Water in		ļ
Ac  to			(A <sup>V</sup> ) °C					
Bc t <sub>c</sub> °C	-		c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	.74, 32	5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 T <sub>C</sub>		·		L	+ grams/100 gran	ns solveni	ւ t
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc. from det	da	ta 5-Calc. by for		
SOURCE:		Do	o <b>w</b>					
PURIFICAT			stillation					
LITERATU	RE REFERE	NCES	<b>3:</b>					

	<del></del>							No. 9	
NAME	Bromotric	hloro	methane	·		STI	RUCTURAL	FORMUL	.A
							CCLB	-	
Mole % Pur. 99.	Ref. Mo	ecul		Molecular Weight 198,29	7		C C1 <sub>3</sub> B		
		Ref.			Ref.				Ref
F. P. °C F. P. 100%	-5.65	1	dt/dP °C/mm		_	f g l	to		
B. P. °C 760 mm	104.70	١,	25°C BP	0.5246 0.04655	5 4	h			
100	45.93	1 1	t <sub>e</sub>	0.03682	5	f'	to		
30 10	20.13	4	30 mm	0.6450	5_	g'	<u>*K</u>		
1	0.60 -31.53	5	AHm cal/g	3.05	4	h' i			├
Pressure			ΔHv cal/g	44.21		m	to oK		l
mm 25°C	38.40	5	25°C 30 mm	44.56	5	0			
t <sub>e</sub>	1027.13	5	BP	38.61	5	m'	to		†
Density g/ml 20°C	2.01221	1	te te (d, e)	37.86 37.84	5	n'	<u>•</u> K_		
d <sub>4</sub> 25	2.00130	1	AHv/Te	19.30	5	o' <u>'</u>			
a 30	1.99037	4	d 20 to	45.97	5		ace tension	21 2/	١.
b b	2.05583 -0.00216	4	116 °C	0.0703	5	dyne	s/cm. 20°C 30	31.26 29.98	1
Ref. Index			d' to			L	40	28.72	1
<sup>n</sup> D 20°C	1.50633	1	d <sub>c</sub> g/ml			Para	chor [P]	222.0	
50	1.50342	1 1	I v mi/g	210	_		20°C 30	233.0 233.2	4
"C"	0.3308	4	, .	318.	5		40	233.3	4
MR (Obs.)	29.294	4	P <sub>c</sub> mm	30812.	5			235.7	5
MR (Calc.)		5	PV/RT 25°C	0.9996	5	Exp.	L.1.%/wt.		
(nD-d/2) Dielectric	0.50023	4	30 mm	1.0000	5	Disp	ersion		
A 20 to	6,86625	<b>.</b>	BP te	0.9580 0.9492	5		h Point C		
B 170°C	1294.08	1 1	t <sub>c</sub>				Point		-
c	220.	5	ΔHc kcal/m ΔHf			Ultr	Spec. a.V.		
A* 20 to B* 126 °C	1.55053	5 5	ΔFf				ay Dif.		
к — =	1207.0	,	Viscosity			Infra	····		$\vdash$
t, to		i	centistokes 7 20 °C	0,8138	1		bility in <sup>†</sup> etone		
t <sub>k</sub> to °C			40	0.6091	i		rbon tet.		
A'  to			60 80	0.4788 0.3893	1 1	Eth	nzene Mer		
B'  ° <u>C</u>			BV I to	577.74	4		leptane		
A'* to	<del>                                     </del>	$\vdash$	A 50 ℃	3.9400	4	Wa	anol ter		
B'* °C			(B <sup>V</sup> )  50 to	528.82	4	Wa	ter in		<u> </u>
Ac  170 to	7.29174	5	(A <sup>V</sup> )  90 °C	2. 0931	4				
Bc tc °C	1647.18	5	c <sub>p</sub> liq. °K						
Cryos. A°	0.00426	1	c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	115.72	5	c <sub>v</sub> vap.						
$T_{\mathbf{R}} = 0.75$	1	Щ	u	L	L	+ gr	ams/100 gra	ms solver	ıt.
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-	Calc. by for	mula	
SOURCE:			w, dist.						
PURIFICAT	ION:	Do	w, dist.						
	RE REFERE	NCE	5:						
		-							

NAME	Meth	vl br	omid	le			STRUCTURAL :	FORMUL	A.
	Bron	·				$\dashv$	binocionale .		-
Mole	Ref.	Mo	lecul	ar	Molecular		СН <sub>2</sub> Вг		
% Pur. 99.	71 1	Fo	rmul	a CH <sub>3</sub> Br	Weight 94.9	<del></del>			
	т		Ref.		·	Ref			Ref.
F.P. °C F.P. 100%	-94.07		1	dt/dP *C/mm	1		f to		1
B. P. *C	<del> </del>		$\vdash$	25°C	0.0187		g °K	İ	j
760 mm	+3.56	,	2	BP	0.0339 0.0346		<u>+</u> +		
100 30	-39.40		2	t <sub>e</sub>	1	1 1	f' to g' 'K		
10	-58.37		2 2	30 mm	0.4752	宀	h'		
1	-96.56		5	ΔHm cal/g	<b></b>	Ш	m l to		├
Pressure				∆Hv cal/g 25°C	56.99	5	n K		1
mm 25°C	1633.0		5	30 mm	67.72	5	•		
t <sub>e</sub> Density	137.1			BP	59.74	<b>5</b>	m' l to		$T^{-}$
g/ml-10°C	1.73	676	1	te te (d, e)	59.83 59.57	5	n' °K		
at -5	1.74	566	1	ΔHv/T <sub>e</sub>	20.49	5	o'		
	ļ			d   -58 to	<del></del>	5	Surface tension		
a b	1.73		4 4	_e _l _3_ •		5	dynes/cm10°C		1
Ref. Index	† · · · ·		$\vdash$	d'   to			40		
n <sub>D</sub> 20°C	İ		1		<u> </u>	Н	Parachor [P]		
25 30	}			d g/ml vc ml/g			20°C		
"C"	<del> </del>		-	tc °C	194.0	3'	30 40		
	<b>├</b> ──		$\vdash$	P <sub>c</sub> mm			Sugd.	124.1	5
MR (Obs.) MR (Calc.)		3	5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)				25°C 30 mm	0.9399 1.0000	5	u. Dispersion		
Dielectric				BP	0.9625	5	Flash Point °C	ļ	├
A -58 to	6.95	965	2	t <sub>e</sub>	0.9632	5	Fire Point	İ	1
B 1_53°C C			2	t <sub>c</sub>	<u> </u>	$\vdash$	M Spec,		$t^{-}$
	238.32		2	ΔHc kcal/m			Ultra V.	Ì	
A* -58 to B* 13 °C	920.93		5	ΔFf			X-Ray Dif. Infrared	1	
к — — —	, , , , , ,			Viscosity			Solubility in +	<del> </del>	╁╌
c	-			centistokes 7 -30 °C	0.2834	1	Acetone	<b></b>	
t <sub>x</sub> to	]			7 -30 ℃ -20	0.2575		Carbon tet.	oc	
A'   to	<del>                                     </del>		-	-10	0.2356		Benzene Ether	∞ ∞	1
B', ∟ _ °C	.[			B <sup>V</sup>   -40 to	0.2174	_	n-Heptane	- ×	
	<del> </del>			A   -40 to	256. 30 2. 3985	4 4	Ethanol Water	œ	1
A'* to B'* °C				(BV) -10 to	251.07	4	Water in		1
Acl to	<del></del>	_	$\vdash$	(A <sup>v</sup> ) 10°C	Z. 4183				
Bc t °C						$\vdash$		ĺ	1
	ļ			c <sub>p</sub> liq. °K	•			1	1
Cryos. A° consts. B°	0.02	120	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	2.86	•	5	c <sub>v</sub> vap.		İ	<u> </u>		<u> </u>
REFERENC	ES: 1-D		2 - A E	I 3-Lit. 4-	Cala fram da		grams/100 gran		t
SOURCE:	25. 1-2.				Cale. Irom det	. da	ta 5-Calc. by for	mula	
PURIFICAT	TON:			v, Lit.					
		en es		v, dist.	21 77 -1-	-10	Al	~1 i 1	
Technology	2, 651 (19	948)	VC ES	: 3 limmern	lans; 5' Kirk a	ina O	thmer, Encyc. of	onemicai	

							No. 11	
NAME	Ethyl chlor	ide				STRUCTURAL	FORMUL	A
ļi	Chloroetha	ne	·			a a.a		
Mole % Pur.		ecul.		Molecular Veight 64.51	7	CH₂ClCH	3	
	1	Ref.			Ref.			Ref.
F.P. °C F.P. 100%			dt/dP °C/mm			f to		
B. P. °C			25°C	0.0245	5	h		
760 mm 100	+12.27	2 2	BP t <sub>e</sub>	0.0350 0.0348	<b>5</b>	f¹ to		
30	-51.59	2	30 mm	0.4896	5	g' <u>°K</u>		
10	-66.43	2 5	ΔHm cal/g		t	h'		
Pressure	-90.93	•	ΔHv cal/g	<u> </u>	1-1	m to		
mm 25°C	1199.0	4	25°C	88.07	5	n   •K		i
te	764. 2	5	30 mm BP	102.92 90. <b>5</b> 5	5 5			<b>├</b> ─-
Density g/ml \cdot 0°C	0.03300	3	t <sub>e</sub> (d, e)	90.52	5	m' to		1
	0.92390	3		90.40	5	0'		
d <sub>4</sub> 10 25	0.87062	4	d -52 to	20.49	5	Surface tension		
a	0.92572	4	d -52 to	92.92 0.1938	5	dynes/cm. 5°C	21.20	3
Ref. Index	-0.00203	4	d' to			8 10 40	20.64	3
n <sub>D</sub> 0°C	1.3790	3	e'   °C		Н	Parachor [P]	<b></b>	
10 30	1.3738	4	d g/ml v ml/g			`.5 <b>°</b> ℃	151.2	4
"C"	0.5486	4	v <sub>c</sub> m1/g t <sub>c</sub> °C			10 40	152.3	4
MR (Obs.)	16. 137	4	P <sub>c</sub> mm			Sugd.	149.4	5
MR (Calc.)		5	PV/RT 25°C	0.0403	,	Exp. L.1.%/wt.		
(nD-d/2)	0.8171	4	30 mm	0.9492 1.0000	5	u. Dispersion		
Dielectric			BP	0.9620	5	Flash Point C		
A -50 to B 70 °C	6.94914	2	te tc	0.9618		Fire Point		
c	236.67	2	ΔHc kcal/m			M. Spec. Ultra V.		
A* -50 to	1.25525	5	ΔHf ΔFf			X-Ray Dif.		
B* <u>_40 °C</u> K	945.56	5	Viscosity	<u> </u>		Infrared		<u> </u>
c			centistokes			Solubility in + Acetone		
t <sub>k</sub> to			<b>り</b> 5 ℃	0.3196 0.3090	3 3	Carbon tet.		ļ
A' to		H	10	0.3070		Benzene Ether		
B'°C	1		B <sub>w</sub> _5 to		$\vdash$	n-Heptane		
		$\vdash$	B 1 20 °C	230.8 2.67499	4 4	Ethanol Water		
A'* to B'* °C			(B <sup>V</sup> )  = to	2.0.1,,	•	Water in		
Acl 70 to			(A <sup>V</sup> )  °C					
Bc t <sub>c</sub> °C	250.	5	c liq. °K					
Cryos, A°	250.	-"-						
consts. B°			P					1
t <sub>e</sub> °C	12.42	5	c <sub>v</sub> vap.					
$T_R = 0.75$						grams/100 gra		it
				Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	Litez							
PURIFICAT								
LITERATU	RE REFERE	NCES	3: 3 Timmerm	ans				
Į								

							No. 12	
NAME	1,2-Dicl	loro	ethane			STRUCTURAL	FORMUL	A
Mole % Pur. 99	Ref. Mo	lecul	ar C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	Molecular Weight 98.96	6	сн <sub>2</sub> с1 <b>с</b> н	1 <sub>2</sub> C1	
		Ref.			Ref			Ref
F. P. ℃	-35.66	1	dt/dP			f to		
F. P. 1009	•		°C/mm		_	gK_		ļ
B. P. °C	00.45	١.	25°C BP	0.2647 0.04192	5	h '		
760 mm 100	83.47 29.83	1 4	t <sub>e</sub>	0.0342	5	f' to		T
30	5, 83	5	30 mm	0.6038	5	g'K_		ł
10 1	-12.54 -43.11	5	∆Hm cal/g	20.03	4	h'		↓
Pressure		<del>                                     </del>	ΔHv cal/g	22.22		to °K		1
mm 25°C	80.3	5	25°C 30 mm	83.92 86.31	5	ö   <u>-</u> -		ļ
te	963.	5	BP	76.38	5	m'   to		+-
Density g/ml 20°0	1.25309	1	te te (d, e)	76.88 75.40	5	n'   K		1
<sub>d</sub> t 25	1.24579	1	ΔHv/T <sub>e</sub>	20,88	5	o'		
	1.23847	1	d   6 to	87.06	5	Surface tension		Ι.
a b	1.28209	4	_e _  _92_ °C		5	dynes/cm. 20°C	32.48 31.06	1
Ref. Index			d'   to			40	28.27	5
n <sub>D</sub> 20°0		1	<del></del>	0.517	5	Parachor [P]		Τ.
25 30	1.44210	1 4	d g/ml v ml/g t <sub>c</sub> °C	1.935	5	20°C 30	188.5 188.6	4
"C"	0.4704	4	N -	306.	5	40		
MR (Obs.	21.01	4	P <sub>c</sub> mm	49035.	5		186.6	5
MR (Calc.	21.170	5	PV/RT 25°C	0.9994	5	Exp. L.1.%/wt. u.		1
(nD-d/2)	0.81821	4	30 mm	1.0000	5	Dispersion		1
Dielectric		1	BP t <sub>e</sub>	0.9560 0.9499	5	Flash Point °C		
A 6 t		4 4	tc	0,260	5	Fire Point		₩-
C	232.	4	ΔHc kcal/m			M Spec. Ultra V.		ł
A* 6to B* 102°		5	ΔHf ΔFf	İ		X-Ray Dif.		
K LIE	12/4.2	'	Viscosity	1		Infrared Solubility in +		₩
t.	_		centistokes	0 (47)	١, ا	Solubility in + Acetone		
t <sub>k</sub>   to			7 20 °C	0.6671 0.5287	1 1	Carbon tet.		
A' to			60	0.4337	1	Benzene Ether		
B' '	일		B <sup>V</sup>   10 to	0.3982 441.1	4	n-Heptane		
A'* to	,	$\vdash$	A   80 °C	Z. 31484	4	Ethanol Water		
B'* °(			(B <sup>V</sup> ) to	1		Water in		
Ac  161 to	7.6284	5	(A <sup>V</sup> ) °C	1				
Bc tc_°	283.	5	c <sub>p</sub> liq. °K					
Cryos. Acconsts. B	0.01770	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	91.19	5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 Т <sub>с</sub>			•		grams/100 gran	ns solven	t
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	. dat	ta 5-Calc. by for		
SOURCE:		D	ow					
PURIFICA	TION:	D	istill <b>a</b> tion					
LITERATU	RE REFERE	NCES	5:					

No. 13 NAME STRUCTURAL FORMULA 1, 1, 2-Trichloroethane CHC12CH2C1 Molecular C2H3Cl3 Mole Ref. Molecular % Pur. 99.90 Weight 133.415 1 Formula Ref. Ref. Ref. F. P. °C -36.59 1 dt/dP £ to F.P. 100% °C/mm °K R 0.8090 25°C 5 B. P. °C h ВP 0.0458 5 760 mm 113.77 1 0.0356 5 ſ١ 55.78 100 4 °K g' 30.36 5 30 5 30 mm 0.6349 10 11.04 5 h' ∆Hm cal/g 20,68 4 -20.5 5 m to ∆Hv cal/g Pressure ۰ĸ n 25°C 72.75 5 mm 25°C 22.49 5 o 30 mm 72.06 5 1055.5 te 5 BP 5 61.33 m' to Density 59.70 5 n' ۰ĸ g/ml 20°C 1.43968 te (d, e) 59.8 5 01  $d_4^t$ 25 1.43194 ΔHv/T<sub>e</sub> 19.97 5 30 1.42420 Surface tension 25 d 75.96 0.1286 5 1.47061 33.61 dynes/cm. 20°C 1 126 °C -0.00154 32.24 30 1 ď to 40 30.84 1 Ref. Index e' °C 20°C 1.47124 [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 0.497 4 25 1.46868 1 20°C 223.1 vc ml/g t °C 2.012 4 30 1.46600 4 30 223.2 4 339. t<sub>c</sub> 5 40 223.1 4 "C" 0.4323 4 P<sub>c</sub> mm 36263. 5 223.8 5 Sugd. MR (Obs.) 25.91 PV/RT Exp. L.1.%/wt. MR (Calc.) (nD-d/2) 26.037 2 25°C 1.0000 5 u. 0.75140 5 1.0000 30 mm Dispersion Dielectric 7.116 1 0.9603 5 BP Flash Point °C 0.9511 5 A 30 to 6.84165 Fire Point 0.255 5 B \_186 ℃ 1262.6 M. Spec. Ultra V. 205. 4 С AHc kcal/m ΔHf A\* 30 to 1.35368 5 X-Ray Dif. ΔFf B\*[ 136 °C 1183.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to 20 °C 0.82145 t<sub>k</sub> | 1 Carbon tet. °C t<sub>x</sub>\_ 40 0.63307 1 Benzene 60 0.50894 1 ۸ı to Ether 80 0.42113 В' °C n-Heptane C' B. 10 to 489.6 4 Ethanol A | 90 °C Ž. 23823 Water A1\* to (B<sup>V</sup>)| Water in B!# °C Ac | 186 to (AV) 7, 2567 Bc tc °C 1599. cp liq. °K 254 Cryos. Aº cp vap. 0.02484 1 °K consts. B° c, vap. t, °C 125,68  $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: Dow PURIFICATION: Distillation LITERATURE REFERENCES:

							No. 14
NAME	Methyl c	nlorof	form			STRUCTURAL FO	ORMULA
1 1							
			I			CH <sub>3</sub> CCl <sub>3</sub>	
Mole % Pur. 99.	92   Ref.   Mo	olecul ormul	ar C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub>	Molecular Weight 133.4	125		
		Ref.			Ref.		Ref
F.P. °C	-30.41	1	dt/dP	T		f to	
F.P. 1009			*C/mm 25*C	0.1829	5	g <u>*K</u>	
B. P. °C 760 mm	74.096	1	BP	0.04250	4	h	
100	20.35	1	<b>t</b> .	0.03643		f' to K	
30 10	-3.301 -21.22	4	30 mm	0.5916	5	g'   '*K	
1	-50.75	5	ΔHm cal/g	4.90	4	m to	
Pressure	122 27	_	ΔHv cal/g 25°C	57.86	5	n •K	1
mm 25°C	123.37 925.86	5	30 mm	61.12	5	0	
Density	+	+	BP t <sub>e</sub>	52.44 51.74	5	m'   to	
g/ml 20°C		1	t <sub>e</sub> (d, e)	51.71	5	n'   *K	
dt 25	1.33063	1	AHv/Te	19.51	5		
•	1.37247	4	d   -3 to		5	Surface tension dynes/cm. 20°C	25.39 1
ь	-0.00164	4	-a,-  -81_ to			30	24.11 1
Ref. Index		1	e'   •C	<b></b>		Parachor [P]	22.87 1
25	1.43519	1	d g/ml v ml/g tc °C				223.7 4
50	1.42165	1	tc *C *	266.	5		223.8   4 223.9   4
"C"	0.4338	4	P <sub>c</sub> mm	30274.	5		223.8 5
MR (Obs.) MR (Calc.		4 5	PV/RT	2 22/5	_	Exp. L.1.%/wt.	
(nD-d/2)	0.7684	4	25°C 30 mm	0.9867 1.0000	5 <b>5</b>	u. Dispersion	
Dielectric		ļ	BP	0.9467 0.9405	5	Flash Point °C	
A -3 to		1	t <sub>e</sub> t <sub>c</sub>	0.7403		Fire Point	
c	225.	5	∆Hc kcal/m			M Spec. Ultra V.	
A*  -3 to B*  91 °(	1.46948	5	ΔHf ΔFf			X-Ray Dif.	
K L	-		Viscosity	·		Infrared Solubility in +	
c	-		centistokes 7 20 °C	0.6282	1	Solubility in + Acetone	
tk to			7 20 °C	0.4899	i	Carbon tet. Benzene	
A'   to			60 80	0.3962 0.3612	1	Ether	
B' - 2	4		BV   10 to	495.84	4	n-Heptane Ethanol	
A'* to	,	1	A <sup>V</sup>   50 °C	2.1070	4	Water	1
B'* *(		1	(BV) 50 to	236.35	4	Water in	
Ac   131 to	7.32705 1524.70	5	(A <sup>V</sup> )  90 °C	₹. 8886	4		
Cc Lc_	269.86	5	c <sub>p</sub> liq. °K				
Cryos, A <sup>c</sup> consts, B <sup>c</sup>		1	c <sub>p</sub> vap. ⁰K				
t <sub>e</sub> °C	80.61	5	c <sub>v</sub> vap.	<u> </u>			
$T_R = 0.7$						grams/100 gram	s solvent
	CES: 1-Dow			Calc. from det	da	ta 5-Calc, by form	
SOURCE:	PION.		ow, dist.				
PURIFICA	RE REFERE		ow, dist.				
LIIERAIU	NE KEPEKE	NCES	) <b>:</b>				
1							
1							
L							

No. 15 sec. Butylbromide STRUCTURAL FORMULA NAME сн<sub>3</sub>сн<sub>2</sub>снсн<sub>3</sub> Molecular C4H9Br Mole Ref. Molecular Weight 137.028 Вr 99.56 % Pur. 1 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -112.65 1 dt/dP f to °C/mm °K g 25°C 0.3281 5 B. P. °C h ВP 0.04510 5 5 760 mm 91.44 4 0.0368 f† to 100 34.61 5 g' °K 9.73 5 30 mm 30 0.6216 5 10 -9.08 5 h! 12,01 4 ∆Hm cal/g 5 -39.97 1 m to AHv cal/g Pressure n ۰ĸ 60.64 62.25 25°C 5 5 mm 25°C 64.65 5 o 30 mm 989. 5 te 53.67 5 BP m¹ to Density 52.73 5 te (d, e) n' °K 1.26085 g/ml 20°C 5 1 52,70 o' 25 1.25354  $d_4^t$ 5 AHv/Te 19.32 30 1.24624 4 Surface tension Т 10 to d 63.27 5 1,29004 25.01 a dynes/cm. 20°C 1 1 100 °C 0.1050 ь -0.00144 4 ăĦ 30 23.90 1 to 40 22,82 Ref. Index e' °C 20°C 1.43705 <sup>n</sup>D Parachor [P] 1 0.412 5 d<sub>c</sub> g/ml 25 1.43453 1 20°C 243.0 4 5 vc ml/g 2,425 50 1.42079 1 30 243.1 4 t<sub>c</sub> 286. 5 243.0 4 40 "C" 0.4598 4  $P_c$  mm 26756. 5 241.1 5 Sugd. MR (Obs.) 28.48 PV/RT Exp. L.1.%/wt. MR (Calc.) 28.437 25°C 0.9978 5 u. (nD-d/2)0.80662 4 30 mm 1.0000 5 Dispersion Dielectric BP 0.9579 5 Flash Point C 0.9507 5 A 10 to 6.82724 Fire Point 0,255 B 1146 °C 1229.08 M. Spec. AHc kcal/m С 220. 4 Ultra V. ΔHf A\* 10 to 1.36870 5 X-Ray Dif. ΔFf B\*[ 110 °C 1149.0 Infrared K Viscosity Solubility in centistokes Acetone to 20 °C 0.4692 1 t<sub>k</sub> | Carbon tet. ٠c 40 0.3884 1 Benzene 60 0.3293 1 A' | to Ether 80 0.28431 R! ٠c n-Heptane B<sup>V</sup> | 10 A<sup>V</sup> | 90 C' 374.6 2.39320 4 to Ethanol °C 4 AI\* Water to Water in (B<sup>V</sup>)I B'\* °C to Acl 146 to 7.2470 5 (AV) °C Bc\_tc\_°C 1556. 5 c<sub>p</sub> liq. 266. Cryos. Aº 0.03217 1 cp vap. °K consts. B° c<sub>v</sub> vap. °C 100.72  $T_{\mathbf{R}} = 0.75 T_{\mathbf{c}}$ grams/100 grams solvent 2-API 5-Calc. by formula REFERENCES: 1-Dow 3-Lit. 4-Calc, from det. data Dow SOURCE: Distillation PURIFICATION: LITERATURE REFERENCES:

No. 16 NAME 1, 2-Dibromo-1, 1-dichloroethane STRUCTURAL FORMULA  $CH_2BrCCl_2Br$ Mole Ref. Molecular Formula C2H2Cl2Br2 Molecular % Pur. 99.97 Weight 256.782 Ref Ref Ref F.P. C F.P. 100% -66.85 1 dt/dP to °C/mm °K g 25°C 11.0081 B. P. \*C h BP 0.05269 760 mm 178.30 1 5 t<sub>e</sub> 0.03573 ſ١ to 100 111.28 g' °K 30 81.54 4 30 mm 0.7460 5 10 58.79 h' 4.73 4 AHm cal/g 20.64 5 1 to m AHv cal/g Pressure n °K 25°C 46.37 1.3482 mm 25°C 5 ٥ 43.51 37.75 30 mm te 1246.14 5 5 BP Density m' to 36.55 5 te (d, e) °K g/ml 20°C 2.26216 36,50 5 ۰, 2.25230 dt4 AHV/T 5 19.87 30 2.24244 4 Surface tension T 48,38 5 d 81 2.30159 39.12 dynes/cm. 20°C 199 °C 0.0596 5 ь -0.00197 37.92 30 1 47, 63 5 to 36.83 40 1 •c Ref. Index e' 8/1 0.0504 5 20°C 1.55666 [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 25 1.55400 1 20°C 283.9 **v**c ml/g 1.54889 35 30 284.2  $t_c$ •C 420. 5 40 284.6 4 "C" 0.3215 4 32870. 5 mm 288.4 5 Sugd. MR (Obs.) 36.523 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 36,700 1.0000 25°C (nD-d/2) 0.42558 4 30 mm 1.0000 Dispersion Dielectric 0.9617 BP 5 Flash Point °C 0.9482 A 81 to 7,03778 4 Fire Point tç 1593.35 В 1246 °C M Spec. c 205. AHc kcal/m Ultra V ΔHf A\* | 81 to 1.75459 5 X-Ray Dif. ΔFf B\* 209 °C 1493.7 Infrared ĸ Viscosity Solubility in c centistokes Acetone °C 1.4680 to 20 1 Carbon tet. °C 1.0399 40 1 Benzene 0.7852 60 1 A' 25 to Ether 80 0.6209 1 B١ \_8<u>1 °C</u> n-Heptane B<sup>V</sup> A<sup>V</sup> Ċ Ī 10 687.51 4 to Ethanol °C 1 50 3.8219 4 Water A1# to (BV) 50 B'\* ٠c Water in to 599.96 4 Ac to 7.45239 (AV) 90 °C 2.0944 4 °C 1989.74 Bc cp liq. °K Cc 258.0 0.01439 Cryos. A° 1 c<sub>p</sub> vap. ۰ĸ consts. B° te °C c, vap. 199.18 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE: Dow, dist. PURIFICATION: Dow, dist. LITERATURE REFERENCES:

No. 17 NAME Ethyl bromide STRUCTURAL FORMULA Bromoethane C<sub>2</sub>H<sub>5</sub>Br Mole Molecular Molecular % Pur. 99.87 Weight 108.976 Formula C2H5Br Ref. Ref. Ref. F.P. °C F.P. 100% -119.33 1 dt/dP f to °C/mm 25°C °K g 0.0560 5 B. P. °C h BP 0.0382 5 760 mm 38.35 2 t<sub>e</sub> 0.0354 5 ſ١ 100 -10.00 2 to g' <u>°к</u> 30 -31.30 2 0.5330 5 30 mm 10 -47.45 2 h! ∆Hm cal/g -74.08 1 m to ΔHv cal/g 25°C Pressure n ۰ĸ 59.91 5 mm 25°C 468.60 o 30 mm 66.69 5 5 837.3 t<sub>e</sub> BP 58.32 5 m' to Density 57.98 5 ţ. n' °K g/ml 20°C te (d, e) 1.45939 5 1 57.73 ٥' d<sub>4</sub> 25 1.44917 ΔHv/T 20.01 5 30 1.43889 4 Surface tension T\_32 62.93 5 to 1,50075 4 23.70 dynes/cm. 20°C 1\_41 5 °C 0.1203 å h -0.00195 4 30 22.46 1 to 40 21.45 5 Ref. Index e' nD 20°C 1.42416 [P] Parachor d<sub>c</sub> g/ml 0.507 3 25 1.42104 1 20°C 165.1 4 vc ml/g 1.9724 3 30 1.41776 4 30 165.3 4 <sup>t</sup>c 230.7 3 40 "C" 0.3862 4 P<sub>c</sub> mm 42000. 3 Sugd. 163.1 5 MR (Obs.) 19.060 PV/RT Exp. L.1.%/wt. MR (Calc.) 19.201 5 25°C 0.9716 (nD-d/2) 0.6945 4 30 mm 1.0000 Dispersion Dielectric 9.112 1 BP 5 0.9600 Flash Point C A -32 to 0.9574 5 6.91995 2 Fire Point 0.2873 B (110 °C 1090.81 2 M. Spec. С 231.71 2 AHc kcal/m Ultra V A\* -32 to **AHf** 1.42140 5 X-Ray Dif. ΔFf B\* 51 °C 1019.18 Infrared ĸ Viscosity Solubility in centistoke s Acetone to 0.3191 tk tx o °C Carbon tet. °C 0.2914 œ 1 10 Benzene m 20 0.2690 1 A' to Ether œ 0.2503 1 30 B١ °C n-Heptane BV | œ -10 to C' 291.22 4 Ethanol 00 40 °C 2. 43797 Water A1# to Water in B'\* (BV) °C to Ac 110 to 7.59098 5 (A<sup>V</sup>) °C Bc tc °C 1545.3 5 c<sub>p</sub> liq. ۰ĸ Cc 5 290.0 Cryos. A° cp vap. °K consts. B° c<sub>v</sub> vap. te °C 41.19 5  $T_{\mathbf{R}} = 0.75 T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: Dow, literature **PURIFICATION:** Distillation, literature LITERATURE REFERENCES: 3 Chem. Rev. 52, No. 1, 117 (1953) Kolbe and Lynn

														<b>No.</b> 18	
NAME		1,2-	Dibr	omoe	thane					$\Box$	STR	UCT	JRAL 1	FORMUL	A.
		Ethy	lene	libro	mide							CI	H,BrCI	L Br	
Mole % Pur. 99	. 91	Ref.	Mo Fo	lecul rmul	ar C	H <sub>4</sub> Br		Molecular Weight 18		84		•		-2	
				Ref.						Ref					Ref
F, P. *C	Ι	9.79		1	dt/	dP					f	$\sqcap$	to		
F.P. 100%		9.85		4		/mm		١.,,		_	g	i	_ <u>*</u> K_		
B. P. *C 760 mm	Ι.	21 2/		١. ا	B1	*C P		1.53 0.04		5	h	I			
100 mm	1 '	31.36		1 4	t <sub>e</sub>			0.03		5	f¹		to		1
30	1	43.04		5		mm		0.68	20	5	g'	! '_	<b>.</b> K		1
10 1	١.	22,22 12,74		5	ΔH	m cal/	g	12.76		4	h'	<u> </u>			<u> </u>
Pressure	T					v cal/g	3				m n	(	to °K		1
mm 25°C	١.,	11.70	)	5		*C		52.48 51.70		5	0	-			1
te	111	12.		5	BI	Ρ.		45.6		5	m'	<u> </u>	to		<u> </u>
Density g/ml 20°C	.	2, 17	920	1	ţ.	(d, e)		44.58		5	n'	¦ i_	*K		
at 25		2.16		i	· e	(4,0)		44.60	- 1	l II	٥'	, - 1			
<b>4</b> 30	$\perp$	2.15		4		Hv/T <sub>e</sub>		19.99		5	Surf	ace to	nsion		T
a b		2.22		4	d e		to °C	54.67 0.06		5			. 20°C	38.51	1
	+	-0.00	208	4	a'_	7 25	to	53.58	.	5	•		30 40	37.22 35.13	5
Ref. Index		1.53	868	1	e'		•c	0.04	-	5	Para	chor			H
- 25	}	1.53		1	d <sub>C</sub>	g/ml ml/g		0.77 1.28		5			20°C		4
"C"	╀	1.53		4	t <sub>c</sub>	•C		377.	Ĭ	5			30 40	215.0	4
	+-	0.32		4		mm		42731.		5			Sugd.	214.0	5
MR (Obs.) MR (Calc.		27.00 26.96		<b>4</b> 5	PV	/RT				$\neg \neg$	Exp.	L. 1.	%/wt.		
(nD-d/2)	Ί	0.44		4	25	°C		1.00		5	D	u,			
Dielectric	Т	4.77	1	1	BI			0.96		5		ersio h Poi			$\vdash$
A 43 to		7.06	245	4	ţ.			0.95 0.25		5		Poin			
B [_215*C		69.7 20.		4	t <sub>c</sub>	kcal/	·	0.23	-		M S				
A*  43 to	+-	1.67	253	5	ΔH		111	ł			Ultr				
B* 156°C		74.2	033	5	ΔF	<u> </u>					Infra	ay Di: red	•		
K – –	1					cosity	_			1	Solu	bility	in +		T
t <sub>k</sub>   to					7	tistoke 20		0.79	558	1	Ace	tone			1
t <u>x</u> ; •€					'	40		0.60		1		rbon t nzene	et.		
A'   to						60 80		0.48		1	Eth	er			1
c,	-			1	B	1 10	to	503.1		4		leptar anol	ie		1
A'* to	1				ĀV	90	•C	2.17	512	4	Wa	ter			
B'* *(	:				(BV	$\overline{\gamma}_{\parallel}$ – -	to				Wa	ter in		0.071	1_
Ac   215 to	١,,	7.49	59	5	(AV	) į	°C			į					1
Bc tc_°C	- 1 2	71. 76.		5	Сp	liq.	°К								ļ
Cryos, A° consts, B°	-	0.01	508	1	ll .	vap.	°K								
t <sub>e</sub> °C	1	45.84		5	c^	vap.									
$T_R = 0.7$	5 T <sub>c</sub>										+ gra	ms/l	.00 gran	ns solven	t
REFEREN	ES	1-D	ow	2-AF	PI 3	-Lit.	4-0	alc. from	det	. dat					
SOURCE:				Do	w										
PURIFICA'	1017	1:		Di	stilla	tion									
LITERATU	RE	REF	ERE	NCES	:										

No. 19 NAME 1, 1, 2-Tribromoethane STRUCTURAL FORMULA Br Br в снсн, Molecular C2H3Br3 Mole Ref. Molecular % Pur. 99.96 1 Weight 266.792 Ref Ref. F.P. °C -29.21 1 dt/dP f to F.P. 100% °C/mm 25°C -29.19 4 g <u>°K</u> 18.078 B.P. °C h BP 0.05408 5 760 mm 188.93 1 0.0364 5 ſ١ to 100 120.43 4 g' <u>•к</u> 5 30 90.21 30 mm 0.7567 5 10 67.16 5 h١ AHm cal/g 8.1637 4 5 28.62 m to AHv cal/g Pressure ۰ĸ n 25°C 47.27 5 mm 25°C 0.7751 o 30 mm 43.33 1263. te 5 ΒP 36.76 5 m' to Density 5 5 35.43 n' °K g/ml 20°C te (d, e) 2,62111 1 35.29 o'  $\mathbf{d_{4}^{t}}$ 25 2.61007 1 ΔHv/T<sub>e</sub> 19.52 5 30 2.59903 4 Surface tension 90 to 49.33 5 2.66526 dynes/cm. 20°C a 44.36 1 211 °C 0.0665 ь -0.00221 30 43.12 1 aי٦ to 41.68 40 1 Ref. Index e' <sup>n</sup>D 20°C 1.59336 Parachor [P] dc g/ml 25 1.59076 1 20°C 262.7 vc ml/g t °C 30 1.58797 4 30 263.0 4 <sup>t</sup>c 451. 5 40 263.1 "C" 4 4 0.2945 P<sub>c</sub> mm 35925. 5 Sugd. 264. 9 5 MR (Obs.) 34.51 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.731 25°C 1.0000 (nD-d/2)0.28278 4 30 mm 1.0000 Dispersion 0.9529 Dielectric 5.609 1 ВP Flash Point C 90 to 0.9373 5 6.94373 1562.12 Fire Point В \_270°C M. Spec. С 196. 4 AHc kcal/m Ultra V. ΔHf A\* 90 to 1.68974 5 X-Ray Dif. ΔFf B\* 221 °C 1469.3 Infrared ĸ Viscosity Solubility in c centistoke Acetone to 20 1.4784 Carbon tet. t<sub>x</sub> °C 1.0190 40 1 Benzene 0.7616 60 1 A'ı to Ether 80 0,5984 В' °C n-Heptane B<sup>V</sup> | 10 A<sup>V</sup> | 90 c' 639.3 4 to Ethanol A'\* °C 3.96698 Water to (BV) Water in B'\* °C to Acl 270 to (AV) 7.3511 5 °C Bc tc °C 1964. c<sub>p</sub> liq. ۰ĸ Сc 5 252. Cryos. A° 0.01843 1 ۰ĸ cp vap.

grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula Dow SOURCE: Distillation PURIFICATION:

c<sub>v</sub> vap.

LITERATURE REFERENCES:

211.00

consts. Bo

te °C  $T_R = 0.75 T_C$ 

r							No. 20	
NAME	Isopropyl	chlo	ride			STRUCTURAL I	FORMULA	4
	2-Chlorop	ropa	ne			(СН <sub>3</sub> ) <sub>2</sub> СН	Cl	
Mole % Pur. 99	Ref. Mo	lecul rmul		Molecular Weight 78,543	3			
		Ref.			Ref			Ref.
F.P. °C F.P. 1009	-117.18 -117.15	1	dt/dP *C/mm 25*C	0.04031	5	f to to		
B. P. °C 760 mm	34.8	4	BP	0.04931 0.03704	4	h		ļ
100 30	-12.17 -32.93	4	t <sub>e</sub>	0.03466	5	f' to		
10	-48.69	5	30 mm	0.5198 22.48	4	h'		
1	-74, 72	5	ΔHv cal/g	22.40	-	m to		
Pressure mm 25°C	529.5 827.2	5 5	25°C 30 mm BP	83.27 93.66 81.53	5 5 5	n   °K		
Density	0.0/357		t_	81.11	5	m' to		
g/ml 20°0 d <sup>t</sup> 25 4 30	0.85599	1 1	le (a, e)	81.10	5	0'		
	0.84936	4	ΔHv/T <sub>e</sub>	20.52 87.76	5	Surface tension		T
a b	0.88926 -0.00124	4	_e,_  _40 °C to	0.1791	5	dynes/cm. 20°C 30 40	18.16 17.01 15.87	5 5 5
Ref. Index		1	e' i °C	2 242	Ļ	Parachor [P]	13.07	+-
D 25	1.37464 1.37149	1	d g/ml vc ml/g	0.341 2.934	5 5	20°C		
"C"	0.5855	4	, c	212.	5	30 40		
MR (Obs.	<del></del>	4	P <sub>c</sub> mm	35451.	5	Sugd.	188.4	5
MR (Calc.	20,921	5	PV/RT 25°C	0.9690	5	Exp. L.1.%/wt. u.		
(nD-d/2) Dielectric	9.52	1	30 mm BP	1.0000 0.9601	5 5	Dispersion		
A   -32 t		5	te	0.9578	5	Flash Point °C Fire Point		İ
B ∟90°9	230.	5	t <sub>c</sub> AHc kcal/m	0.27	5	M Spec.		1
A*  -32 to B*  47 °0	1.33241	5 5	ΔHf ΔHf ΔFf			Ultra V. X-Ray Dif.		
к ——-	_		Viscosity			Infrared Solubility in +		
t <sub>k</sub>			7 0 °C	0.4499 0.4079	1 1	Acetone Carbon tet.		
t v		-	20	0.3724	1	Benzene Ether		
B' '	2		B <sup>V</sup>   0 to	0.3410 334.0	4	n-Heptane		
A'* to	,	-	A'   40 °C	2.43017	4	Ethanol Water		
B'* °	3		(B <sup>V</sup> ) to			Water in		-
Ac 90 to		5 5 5	(A <sup>v</sup> )   °C c <sub>p</sub> liq. °K		-			
Cryos. Acconsts. B	0.03661	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	37.21	5	c <sub>w</sub> vap.					
$T_R = 0.7$				•		+ grams/100 gran	ns sol <b>ve</b> n	t
	CES: 1-Dow			alc. from det	. da	ta 5-Calc. by form	nula	
SOURCE:		Do						
PURIFICA	FION: RE REFERE		stillation					
LIIERAIC	ne refekel	NUES	:					

							No. 21				
NAME	Propylene	dich	nloride		l	STRUCTURAL FORMULA					
	l, 2-Dichl	orop	ropane			CH3CHC1C	H_Cl	ļ			
Mole % Pur. 99.	76 Ref. Mo	ecul mula		Molecular Veight 112.9	92						
	1	Ref.			Ref.			Ref.			
F.P. °C F.P. 100%	-100,53 -100,44.	1 4	dt/dP °C/mm			f to					
B.P. °C			25°C BP	0.4082 0.04442	4	h					
760 mm	96.37 40.05	1 1	t	0.03573	5	f' to					
30	15.18	4	30 mm	0, 6231	4	g' <u>°K</u>					
10	-3.7 -35.0?	5	ΔHm cal/g	13.53	4	h'					
Pressure		-	∆Hv cal/g			m to					
mm 25°C	49.67	4	25°C 30 mm	77.00 78.23	5	"   <u></u>					
t <sub>e</sub>	1003.0	5	BP	67.70	5	m' to		$\vdash$			
Density g/ml 20°C	1, 15597	1	te te (d, e)	66.76	5	n' K					
dt 25	1.14936	i		66.45	i I	0'					
4 30	1.14273	4	ΔHv/T <sub>e</sub>	19.89 80.17	5	Surface tension					
a b	1.18241	4	d 15 to	0.1294	5	dynes/cm. 20°C	28.65	1			
Ref. Index	-0.00131	+	d' to			8 30 40	27.37 26.15	1			
n <sub>D</sub> 20°C		1	<u> </u>	0.41	5	Parachor [P]					
25 30	1.43679	1	d g/ml vc ml/g	0.41 2.44	5	20°C	226.2	4			
"C"	1.42389	4	vc ml/g tc °C	304.3	5	30 40	226.2 226.3	4			
MR (Obs.)	0,5041	4	P <sub>c</sub> mm	33300.	5	Sugd.	225.6	5			
MR (Calc.		5	PV/RT	0.0070	_	Exp. L.1.%/wt.					
(nD-d/2)	0.86138	4	25°C 30 mm	0.9978 1.0000	5	u. Dispersion					
Dielectric	8.96	1	BP	0.9550	5	Flash Point °C		-			
A 15 to B 160 °C	6. 96395 1295. 9	4	te tc	0.9506 0.255	5	Fire Point					
c	221.	5	ΔHc kcal/m		$\vdash$	M. Spec.					
A* 15 to	1.41927	5	ΔHf ΔFf			Ultra V. X-Ray Dif.					
B*[115°C	- 1214.9	5	Viscosity		-	Infrared					
c	_		centistokes			Solubility in +					
t <sub>k</sub> to			η 20 °C	0.7415	1	Acetone Carbon tet.					
t <sub>x</sub> °C		-	40 60	0.5830 0.4747	1 1	Benzene					
B'i °C	_		80	0,3978	1	Ether n-Heptane					
C'			B <sup>V</sup>   10 to A <sup>V</sup>   90 °C	459.1	4	Ethanol	0 275	١,			
A'* to B'* °C			$\frac{A^{\vee}}{ B^{\vee} } = \frac{90 \text{ °C}}{\text{to}}$	₹. 29983	*	Water Water in	0.275 0.132	1			
Acl 160 to	7, 35053	5	(A <sup>V</sup> )  °C								
Bc <sub>i</sub> t <sub>c</sub> ℃	1610.6	5			-						
Cc	265.2	5	Р.								
Cryos. A° consts. B°	0.02592	1	c <sub>p</sub> vap. °K				1				
t <sub>e</sub> °C	106.03	5	c <sub>v</sub> vap.				l				
$T_R = 0.7$	5 T <sub>c</sub>					grams/100 gra	ms solven	it			
REFEREN	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc. by for	mula				
SOURCE:		D	)ow								
PURIFICA'	rion:	D	ow, Distillation	1							
LITERATU	RE REFERE	NCES	<b>5</b> :								

							<b>No.</b> 22	
NAME	Trimethylene chloro bromide					STRUCTURAL FORMULA		
Mole Ref. Molecular C <sub>3</sub> H <sub>6</sub> Cl Br Weight 157, 451					СН <sub>2</sub> С1СН <sub>2</sub> СН <sub>2</sub> Вғ			
70 2 41. 77.	· · / 1 · · 1 · ·	Ref.		Weight 131.1	Ref			Ref.
F.P. °C	-58.87	1	dt/dP	T	1	f   to	Γ	
F.P. 100%			°C/mm			g L K	İ	}
B. P. *C	142.24	Τ,	25°C BP	2.582 0.04891	5 4	h	L	1
760 mm 100	143.36 81.16	1 1	t <sub>e</sub>	0.0359	5	f' to		
30 10	53.57 32.4	4 5	30 mm	0.6921	4	g'	1	
1	-3.0	5	ΔHm cal/g	14.00	4	h'		<del> </del>
Pressure	,	T .	ΔHv cal/g 25°C	66.92	5	m to		
mm 25°C	6.495 1127.0	5	30 mm	64.89	5	0		1
Density	+	+-	BP te	55.78 54.29	5	m¹   to		
g/ml 20°C			te (d, e)	54. 23	5	n'   <u>*K</u>	1	1
d <sub>4</sub> 25	1.58925	1	AHv/T <sub>e</sub>	19.79	5	<u> </u>		-
8	1.62775	4	d   53 to	70.32	5	Surface tension dynes/cm. 20°C	36.11	1
ь	-0.0 <sub>3</sub> 154	4	_a, -		5	<b>3</b> 0	34.81	1 1
Ref. Index		1	e'			Parachor [P]	33, 72	+-
D 25	1.48414	1	d g/ml v ml/g	0.522 1.914	5	20°C	241.7	4
"C"	1.47260	-	tc °C	379. 2	5	30 40	241.8	4
	0,4014	4	P <sub>c</sub> mm	36000.	5		239.3	5
MR (Obs.) MR (Calc.		4 5	PV/RT 25°C	1 0000	_	Exp. L.1.%/wt.		
(nD-d/2)	0.68788	4	30 mm	1.0000 1.0000	5	u. Dispersion		1
Dielectric	<del></del>	1	BP t <sub>e</sub>	0.9500 0.9380	5	Flash Point °C	<b></b>	+-
A 53 to B <u>215</u> °C		1 1 1	t <sub>c</sub> AHc kcal/m	0.255	5	Fire Point M Spec.		╁
A*   53 to B* 175 °C	1,59125	+	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
K — — —	]		Viscosity centistokes			Solubility in +		T
t <sub>k</sub> to t <sub>x</sub> to	:		7 20 °C 40 60	0.9148 0.7042 0.5659	1 1 1	Carbon tet. Benzene		
B' ∟ °C			80	0.4685	1	Ether n-Heptane	}	1
C'	<del> </del>	1	B <sup>V</sup>   10 to A <sup>V</sup>   90 °C	489.5 7.28480	4	Ethanol Water	1.84	1
A!* to B!* °C	;	1_	(B <sup>V</sup> ) to	2.20400	•	Water in		┼-
Ac   215 to Bc   tc_°C Cc		5 5 5	c <sub>p</sub> liq. °K		-			
Cryos. A° consts. B°	0.02418	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	158.64	5	c <sub>v</sub> vap.					
$T_R = 0.75 T_C$ + grams/100 grams solvent								
	CES: 1-Dow			Calc. from det	da:	ta 5-Calc. by for	mula	
SOURCE:			ow					
PURIFICATION: Dow, Distillation  LITERATURE REFERENCES:								
	RE REFERE	inc es	:					

1 1							No. 2		
NAME	ME 1,3-Dibromopropane Trimethylene dibromide					STRUCTURAL FORMULA Br(CH <sub>2</sub> ) <sub>3</sub> Br			
Mole % Pur. 99	Ref. M	olecul ormul		Molecular Weight 201.9	10	2. (0	2/3		
		Ref.			Ref.			Ref.	
F.P. °C F.P. 1009	-34.20	1	dt/dP °C/mm			f to			
B. P. °C			25°C	6.1946	5	h (		1	
760 mm	166.67	1	BP t <sub>e</sub>	0.05144 0.0356	5 5	f¹ to		$\top$	
100 30	100, 82	4 5	30 mm	0.7421	5	g'° <u>K</u>			
10	48.64	5	ΔHm cal/g	16.10	4	h'	1		
1	10.41	5	ΔHv cal/g	<u> </u>		m to			
Pressure mm 25°C	2.604	5	25°C	54.25	5	n •K	-		
t <sub>e</sub>	1206.5	5	30 mm BP	52.47 46.52	5		ļ	+-	
Density			t_	45.30	5	m' to		İ	
g/ml 20°0	1.98009		te (d, e)	45.34	5	<del>                                    </del>			
d <sub>4</sub> 30	1.96229		ΔHv/T <sub>e</sub>	19.93	5	Surface tension		+	
a	2.01568		d 71 to	56.92 0.0624	5 5	dynes/cm. 20°C	39.93	1	
ь	-0.00178	3 4	Tai Time to	0.0021	ا آ ا	30 40	38.60 37.34	1 1	
Ref. Index		1	e'   °C			Parachor [P]	37.34	+-	
D 25	1.5207	1	d <sub>c</sub> g/ml	0.732 1.367	5 5	20°C	256.3	4	
30	1,5180	4	vc ml/g tc °C	414.	5	30	256.5	4	
"C"	0.3467	4	P <sub>c</sub> mm	38811.	5	40 Sugd	257.5 . 253.0	5	
MR (Obs.) MR (Calc.		4 5	PV/RT			Exp. L. 1. %/wt.	<b>†</b>		
(nD-d/2)	0.53314		25°C 30 mm	1.0000	5	u.	Ì		
Dielectric		1	BP	0.9583	5	Dispersion Flash Point C	ļ	+	
A 71 to			te t	0.9451	5	Fire Point			
B (242 °C	1678.3	4	ΔHc kcal/m	0,25		M. Spec.	<b>—</b>	+-	
A*  71 to	1,81483		∆Hf Keal/III	1		Ultra V.	ŀ		
B*  196 °C		<b>5</b>	ΔFf			X-Ray Dif. Infrared			
K		1	Viscosity centistokes			Solubility in +		_	
t <sub>L</sub> _ to			າງ 20 °C	1.0431	1	Acetone Carbon tet.			
x			' 40 60	0.7838 0.6192	1 1	Benzene	ł		
A'   to			80	0.5068	i	Ether			
č, ' <sup>s</sup>	-1		B <sub>v</sub> 10 to	523.7	4	n-Heptane Ethanol			
A!* to			A'   90 °C	2. 22212	4	Water Water in			
B'* °C		+_	(B <sup>V</sup> ) to			Water in	+	+	
Acl 242 to Bc t <sub>c</sub> °C		5	(A <sup>V</sup> )  °C	ļ	$\square$	4			
Cc C	278.	5	c <sub>p</sub> liq. °K						
Cryos, Acconsts, Bo	0.02865	1	c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	185.62	5	c <sub>v</sub> vap.						
$T_R = 0.7$	5 T <sub>C</sub>			1	اــــــا	grams/100 grams/	ams solve	nt	
	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	<u></u>			
SOURCE:		Do	w						
PURIFICA	TION:	Di	stillation						
LITERATU	RE REFER	ENCE	S:			,			

F							No. 24	:
NAME	Propylene dibromide					STRUCTURAL FORMULA		
	1,2-Dibromopropane					CH, Br CH Br CH,		
Mole Ref. Molecular C3H6Br2 Molecular Weight 201.910						CH <sub>2</sub> Br CH B	· CH <sub>3</sub>	
% Pur. 99.	86   1   Fo		a C <sub>3</sub> H <sub>6</sub> BF <sub>2</sub>	Weight 201.9	_	r		<b>1</b> 2 (
<u> </u>		Ref.			Ref.	<u> </u>	r	Ref.
F.P. °C F.P. 100%	-55.50	1	dt/dP *C/mm			f to		
B. P. *C	1		25°C	2.167 0.05044	5	h .	ŀ	ļ
760 mm 100	141.99 78.24	4 5	BP t <sub>e</sub>	0.0371	5	f' to		
30	50.21	5	30 mm	0.7011	5	g'   ' <u>*</u> K_	-	İ
10	28.86 -6.78	5	ΔHm cal/g	10.58	4	h'		<u> </u>
Pressure		۰	ΔHv cal/g			m to		1
mm 25°C	8.0368	5	25°C 30 mm	50.24 48.94	5 5			1
Density	1128.	,	BP	42.04	5	m¹   to		$\vdash$
g/ml 20°C	1.93268	1	te (d, e)	40.90 40.85	5	n'   <u>*K</u> -		İ
dt 25	1.92344	1 4	AHV/Te	19.16	5			<u> </u>
a 30	1.96962	4	d   50 to	52.72	5	Surface tension dynes/cm. 20°C	34.14	1
ь	-0.00184	4			5 5	<b>3</b> 0	32.95	1
Ref. Index	1 52004	١. ا	e' °C		5	40	31.80	1
n <sub>D</sub> 20°C	1.52004	1 1	d g/ml v ml/g	0.605	5	Parachor [P] 20°C	252.5	4
50	1.50420	1	vc ml/g tc °C	1.652 371.	5	30 40	252.7	4
"C"	0.3532	4	P <sub>c</sub> mm	30719.	5	Sugd.	252.9 253.0	4 5
MR (Obs.) MR (Calc.)	31.76 31.584	4 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	0.55370	4	25°C 30 mm	1.0000	5 5	u. Dispersion		İ
Dielectric	4.369	1	BP	0.9534	5	Flash Point °C		<del> </del>
A 50 to B 210 °C		4 4	te tc	0.9409 0.255	5	Fire Point		
_ c	212.	5	∆Hc kcal/m			M Spec. Ultra V.		
A*   50 to B* 168 °C	1.55314 1330.9	5	ΔHf ΔFf			X-Ray Dif.		ļ
K L S	. 1330. /		Viscosity			Infrared Solubility in +		₩.
cto	-		centistokes 7 20 °C	0.8456	1	Solubility in + Acetone		1
t <sub>k</sub> to	ì		40	0.6462	1	Carbon tet. Benzene		
A'  25 to			60 80	0.5168 0.4267	1 1	Ether		
B' L 50 °C	•		B <sup>V</sup>   10 to	498.5	4	n-Heptane Ethanol		
A¹* to			AV   90 °C	₹. 21874	4	Water	0.143	1
B'* °C	<del> </del>	_	(B <sup>V</sup> )  to	]		Water in	0.052	1
Ac   210 to		5	(A <sup>V</sup> )  °C		$\sqcup$			
Cc C	265.	5	c <sub>p</sub> liq. °K	}				
Cryos. A° consts. B°	0.02272	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	157.85	5	c <sub>v</sub> vap.			L		
TR = 0.75 T <sub>C</sub> + grams/100 grams solvent  REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det. data 5-Calc, by formula								
SOURCE:			ow	Calc. from det	. dat	ta 5-Calc, by for	muia	
PURIFICATION: Distillation								
LITERATURE REFERENCES:								
•								
<del></del>								

							No. 25	
NAME	1,2,3-Tribromopropane					STRUCTURAL FORMULA		
					Br CH <sub>2</sub> CH Br	·CH <sub>2</sub> Br		
Mole % Pur. 99.	Ref. Mo	lecul		Molecular Veight 280.81				
W 1 u1. //:	71   1   FO.	Ref.		l 200.01	Ref.	T	Ref	
F. P. °C	16.19	1	dt/dP		Kei.	- T 1	I Kei.	
F.P. 100%	10.17	÷	°C/mm			f to g*K_		
B. P. °C			25°C	74.77	5	h	1	
760 mm 100	222.16	4	BP t <sub>e</sub>	0.0572 0.03669	4 5	f' to		
30	149.18 116.68	5	30 mm	0,8160	5	g' <u>*K</u>		
10	91.79	5	ΔHm cal/g	20,24	4	h'	l	
<u>1</u>	50.01	5	ΔHv cal/g		-	m to		
Pressure mm 25°C	0.1726	5	25°C	48.76	5	n   <u>*K</u>		
t <sub>e</sub>	1336.5	5	30 mm BP	43.94 37.27	5	<u> </u>		
Density			t_	35.83	5	m' to		
g/ml 20°C	2.42086 2.41104	1 1	te (d, e)	35.70	5	0'	1	
d <sub>4</sub> 25 30	2.40122	4	ΔHv/T <sub>e</sub>	19.34	5	Surface tension		
a	2.46014	4	d 117 to e 247 °C	51.31 0.0632	5	dynes/cm, 20°C	45.86 1	
b	-0.00196	4	d' to	1 3.0052		30 40	44.59 1 43.28 1	
Ref. Index	1.58621	1	e' °C	ļ		Parachor [P]	43.28 1	
25	1.58356	1	dc g/ml	1.2887 0.776	5	20°C	301.9 4	
50	1.57120	1	vc ml/g tc °C	550.	5	30 40	302.2 4 302.4 4	
"C"	0.3153	4	Pcmm	56554.	5		302.4 4 303.9 5	
MR (Obs.) MR (Calc.)	38.941 39.349	5	PV/RT			Exp. L. l. %/wt.		
(nD-d/2)	0.3758	4	25°C 30 mm	1.0000	5	u.	i	
Dielectric	6.256	1	BP	0.9400	5	Dispersion Flash Point C		
A 117 to	7.09534	4	t <sub>e</sub>	0.9234 0.24	5	Fire Point	None None	
B 1 322 °C	1779.2 200.	4	tc ΔHc kcal/m	0.24	-	M. Spec.		
A* 117 to	1.84916	5	∆Hf			Ultra V. X-Ray Dif.		
B* 257 °C		5	ΔFf			Infrared		
K			Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			7 20 °C	3.1764	1	Acetone Carbon tet.		
t <sub>x</sub>			40 60	1.8776	1	Benzene		
A'   to B' °C			80	1.2651 0.9249	1	Ether n-Heptane		
c'			Bv 35 to	850.07	4	Ethanol	1	
A'* to			A 85 °C	3.55946	4	Water Water in		
B'* °C		_	(B <sup>V</sup> )  to			water in	<del>                                     </del>	
Ac 322 to Bc t <sub>c</sub> °C	7.5173 2246.	5	(A <sup>V</sup> )  °C		-			
Cc Cc	264.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°	0.02811	1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	247.0	5	c <sub>v</sub> vap.					
T <sub>R</sub> = 0.75 T <sub>C</sub> grams/100 grams solvent								
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula								
SOURCE:		Do	w					
PURIFICAT	'ION:	Di	stillation					
LITERATURE REFERENCES:								
I								
L								

								·····	No. 26	
NAME		n- <b>B</b>	utyl	chlor	ide			STRUCTURAL :	FORMUL/	A.
		1-C	hlor	obuta	ne					
Mole % Pur. 99.	81 F	Ref.	Mo Fo	lecul rmul	ar C <sub>4</sub> H <sub>9</sub> C1	Molecular Weight 92.	569	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> Cl		
-/ <u>/</u>				Ref.	<u>-                                      </u>		Ref	<u> </u>		Ref
F.P. °C	-12	2 2		1	1. (15)	T		<u> </u>	1	
F.P. 100%	1-12	J. L		┢╧┪	dt/dP *C/mm			f to	1	1
B. P. °C	†			$\vdash$	25°C	0.2143	5	h .		1
760 mm		8.44		2	BP	0.0426 0.0358	5 5	II +		1
100 <b>30</b>		4.47 0.67		2 2	t <sub>e</sub> 30 mm	0.5959	5	g' to		ĺ
10		7.39		2		0.3737	۲	h'	1	1
1	-4	7.18		5	ΔHm cal/g	<b></b>		m   to	<del> </del>	<del>                                     </del>
Pressure					ΔHv cal/g 25°C	86.31	5	n °K		
mm 25°C		2.43 0.6	,	5	30 mm	90.03	5	•		ł
Den situ	73	0.0		3	BP	78.18	5	m' l to		$\vdash$
Density g/ml 20°C	Ι.	0.88	621	1	t <sub>e</sub> (d, e)	77.07 76.95	5 5	n' 'K	Ì	
t 25		0.88		l i l	e (", ",			0'		1
<sup>4</sup> 4 30	<u> </u>	0.87	5 39	4	ΔHv/T <sub>e</sub>	19.84	5	Surface tension	<del> </del>	<del>                                     </del>
a		0.90	790	4	d   0 to		5	dynes/cm. 20°C	22.42	5
ь		0.02	106	4	-i,-  -° - ;			30	21.32	5
Ref. Index	1	1 40	211	١, ا	e' i °C			40	20.25	5
<sup>n</sup> D <sup>20°C</sup>		1.40 1.39		1 1	d g/ml vc ml/g	0.207	5	Parachor [P] 20°C		1
30		1.39		4	t <sub>c</sub> °C	3.370	5	30	1	1
"C"		0.60	48	4		269.0	5	40	225.4	_
MR (Obs.)	2	5.53	9	5	P <sub>c</sub> mm	27641.	5	Sugd.	227.4	5
MR (Calc.)	2	5.44	0	4	PV/RT 25°C	0.9936	5	Exp. L.1.%/wt.	,	ł
(nD-d/2)	1	0.95	901	4	30 mm	1.0000	5	u. Dispersion	l	1
Dielectric A 0 to	+	7.39		3'	BP t <sub>e</sub>	0.9580 0.9517	5	Flash Point °C		$t^-$
B 1_123 °C	122	6.93 7.43 4.10		2 2	t <sub>c</sub> AHc kcal/m	0.255	5	Fire Point M Spec.		<del> </del>
A* 0 to		1.32	214	5	ΔHf ΔFf			Ultra V. X-Ray Dif.	}	
B* 100 °C	1148	8.80		5	Viscosity	<u> </u>		Infrared Solubility in +		-
t <sub>k</sub>	1				centistokes 7 15°C	0.5260	3	Acetone	<b>∞</b>	
t C A'   to	ļ			ļ	30	0.4626	3	Carbon tet. Benzene	ø0 ø0	
B' C								Ether n-Heptane	90 90	
c'					B <sup>v</sup> 5 to	324.9	4	Ethanol	∞ ∞	1
A'* to B'* °C					A <sup>V</sup>   40 °C	₹.59365	4	Water Water in		
Ac   123to		7.13	392	5	(A <sup>V</sup> )	1			1	1
Bc tc_°C	1410			5	c <sub>p</sub> liq. °K	<u> </u>				1
	455	5.0		5	l.		1			
Cryos, A° consts, B°	<u> </u>				c <sub>p</sub> vap. °K					
$t_e$ °C $T_R = 0.75 T$		5.87		5	c <sub>v</sub> vap.	<u> </u>		+ (100	L	_
		1 - D	.w	2 - AT	OT 3_1;+ 4 :	Cala from de		grams/100 grants  ta 5-Calc. by for		t
SOURCE:					1 3-111, 4-1	Care. Hom de	. ua	ua 5-Caic. by for	muia	
	TON	Do		- 4:						
PURIFICAT					: 3 Timmerm	21 NDC				
	KLJ K	.DF I	JK E.	NCES	: J Immerm	ans, 5 NDS				

	-							No. 1	-
NAME	Ethene (E	thyler	ne)			ST	RUCTURAL	FORMUL	A
							CH2=CH2		
Mole % Pur.		lecul		Molecular Weight 28.0	52		33.2		
	=	Ref	<del></del>	i organ	Ref.				Ref
D. D. 00	-169.15 <sup>‡</sup>	2			Kei.			<u></u>	Kei.
F.P. °C F.P. 100%	-107.15	+-	dt/dP *C/mm	}		f g	to °K		
B. P. °C			25°C			h	I		
760 mm	-103.71	2	BP	0.0224 0.03495	5	_ <u>_</u> ,_	<del> </del>		1
100 30	-131.78 -144.00	2 2	t <sub>e</sub> 30 mm	0.3049	5	g'	to K		
10	-153.22	2		0.3047		h'	! !		
1	-168.30	5	ΔHm cal/g	ļ	1	m	300 to	0.0698	4
Pressure			ΔHv cal/g 25°C			n	600 °K	0.0011	4
mm 25°C	429.3	5	30 mm	129.08	5	٥	i i	-0.0634	4
Density	+ /	+-	BP	115.39 119.91	5	m'	700 to	0.1300	4
g/ml-10°	0.384	31	te (d, e)	118.53	5	n'	1000 °K	0.0010	4
at 0	0.345	3'	ΔHv/T <sub>e</sub>	20.97	5	0'	Ì	-0.0632	4
	ļ	1	d -144 to	80.15	5	Sur	face tension		
a b	0.433	4	_e_ -115 °C	0.3398	5	dyn	es/cm. 20°C		İ
Ref. Index		+*	d' to			•	30 40		l
n <sub>D</sub> 20°C		1	<u> </u>	0.21		Par	achor [P]		$\vdash$
- 25			d <sub>c</sub> g/ml	0.21 4.76	31		20°C		
30		+	v <sup>c</sup> m1/g t <sub>c</sub> °C	9.90	2		30 40		l
"C"	ļ	4	P <sub>c</sub> mm	38380.	2			101.2	5
MR (Obs.) MR (Calc.		5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	, 10.,0,	1	25°C 30 mm	1.0000	5	_	u.		
Dielectric			BP	0.9681	5		persion		
A -144to	6.74756	2	t e t	0.9622	5 3'		sh Point °C e Point		İ
B (- 61°C		2	t <sub>c</sub>	0, 290			Spec.		<del>                                     </del>
A* -144to	255.	2	ΔHc kcal/m ΔHf	316.20	2	Ult	ra V.		
B* - 80°C	0.81634 528.72	4	ΔFf				Ray Dif. ared		
к ———	23.1	4	Viscosity				ubility in +		<del> </del>
t <sub>k</sub> [-80 to	-0.21634 -7 <b>0.</b> 0	4	centistokes	0.33	١, ١		etone		
t <sub>x</sub>   -40 °C	38.2	4	7 -125 °C	0.23	2 2		rbon tet.		
A'   to		†	-105	0.16	2		nzen <b>e</b> her		İ
B'°C	-1	1	-v	<del></del>	$\vdash$	n-	Heptane	{	1
	<del>-</del>	+-	B <sup>V</sup> to °C		i i		hanol ater		
A'* to B'* °C		1	(B <sup>V</sup> )	-			ater in		
Ac  -61 to	7,3063	5	(A <sup>V</sup> )						
Bc tc °C	1096.	5			H				
Cc	397.	5	Р.						
Cryos. A° consts. B°	0.03725 0.0130	2 2	c <sub>p</sub> vap.300°K 400	0.37252 0.45986	2 2				
t <sub>e</sub> °C	-112.83	5	c <sub>v</sub> vap.						
	T <sub>c</sub> · # at s	turat	ion pressure (t	riple point)		+ g	rams/100 gra	ms solver	ıt
	CES: 1-Dow			Calc. from de	t. da		-Calc. by for		
SOURCE:		API	I						
PURIFICA?	TION:	API							
		NCE	S: 3 Comm. L	eiden 7° 189 a	, b. c	(Tim	mermans) Ms	thias et a	1:
31 Lange				/-	, -, -	,			-,

								No. 2	
NAME	Prop	ene (	Proj	pylene)			STRUCTURAL	FORMUL!	
Mole	Ref.	Mol	ecul	C <sub>3</sub> H <sub>6</sub>	Molecular		С <b>н</b> <sub>3</sub> Сн=0	CH <sub>2</sub>	
% Pur.		For		a 03116	Weight 42.0	_			
	т	#	Ref.			Ref.		<del></del>	Ref
F.P. °C F.P. 100%	-185.25	-	2	dt/dP	1		f to		ĺ
B. P. *C	<del> </del>		-	*C/mm 25*C			gK_		
760 mm	-47.70	)	2	BP	0.0289	2	h		-
100 30	-84.12		2	t <sub>e</sub>	0.0351	5	f' to		
10	-100.06		2	30 mm	0.3982	5	h'		
1	-131.89	)	5	ΔHm cal/g	-	<u> </u>	m   300to	0.0622	4
Pressure				ΔHv cal/g 25°C	140.78	5	n 600°K	0.0011	
mm 25°C	8582. 593.97	,	5	30 mm	118.24	5	° '	-0.0629	4
Density	<del> </del>			BP	104.62 106.03	2 5	m'   700to	0.0970	
g/ml 20°C	0.51	394	2	te te (d, e)	105.99	5	n' 11000°K	0.0011	4
dt 25 4 30	0.50	53"	2	ΔHv/T <sub>e</sub>	20.31	5	0'	-0.0 <sub>6</sub> 34	<u> </u>
	0.55	0.5	4	d   -100 to	<del></del>	5	Surface tension		
a b	-0.00		4	e   -53 °C	0.2600	5	dynes/cm. 20°C		
Ref. Index				e' ' °C			40		
<sup>n</sup> D 20°C	1			d g/ml vc ml/g	0,233	2	Parachor [P]		
30	1			vc ml/g	4.302	2	20°C		
"C"	1			t <sub>c</sub> °C	91.9	2	40		
MR (Obs.)	<b>†</b>			P <sub>c</sub> mm	34504.	2		140.2	5
MR (Calc.) (nD-d/2)	15.58	37	5	PV/RT 25°C 30 mm	1,0000	5	Exp. L.1.%/wt. u. Dispersion		
Dielectric				BP	0.9632	5	Flash Point °C		├
A  -100 to		960	2	t <sub>e</sub>	0.9694 0.274	5 2	Fire Point	i	
B 1-0.€°C	785. 247.		2 2	t <sub>c</sub>	460.43	2	M Spec.		
A* -100 to B*43 °C	1.03	651	5	ΔHf ΔFf	100, 13		Ultra V. X-Ray Dif. Infrared		
K -====	'27.7		١	Viscosity			<del></del>		
t to	-			centistokes	0.40	١, ١	Solubility in + Acetone		
t <sub>k</sub> to				7 -115°C -105	0.48 0.40	2 2	Carbon tet.		
A'   to	<del>                                     </del>			- 95	0.35	2	Benzene Ether		
B' °C	.			B <sup>V</sup>   to			n-Heptane		
	<del>                                     </del>		$\vdash$	B' to			Ethanol Water		
A'* to B'* °C				(B <sup>V</sup> ) to	<b>-</b>		Water in		
Ac  -0.6 to	7.26	58	5	(A <sup>V</sup> ) <sub>1</sub> °C	1				
Bc tc C	1020.		5	c <sub>p</sub> liq. °K	<del></del>	<del>  </del>			
Cc — -	282.	(72	5	1	1			İ	
Cryos, A°	0.04	54	2	c <sub>p</sub> vap 300 °K 400	0. 36456 0. 45392				
t <sub>e</sub> *C	-52.97		5	c <sub>w</sub> vap.	L	$\bot$	L <u>.                                    </u>		<u> </u>
$T_R = 0.75$							grams/100 gra		t
REFERENC	.ES: 1-D	ow	2-AF		Calc, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	TON		AI						
PURIFICAT			AI						
LITERATU			ICES	:	-				
at saturat	-								
at saturat	ion press	sure (	(trip)	e point)					

No. 3 l - Butene (Butylene) NAME STRUCTURAL FORMULA CH3CH2CH=CH2 Mole Ref. Molecular Molecular C4H8 Weight 56.104 % Pur Formula Ref. Reί. Ref. -1**85.3**5# F.P. °C F.P. 100% 2 dt/dP f to °C/mm °K g 25°C 0.01483 5 B. P. °C h 0.0337 2 BP 760 mm -6.26 2 0.0358 5 f† 100 -48.77 2 to g¹ -67.41 <u>°К</u> 30 30 mm 0.4556 5 10 -81.5 2 h! ∆Hm cal/g -104.7 5 1 0.0121 300 to m ΔHv cal/g Pressure n |\_600 °K 0.0013 4 25°C 86.8 2 mm 25°C 2217. o -0.0<sub>6</sub>52 4 30 mm 107.38 5 te 5 703. BP 93.36 2 m 700 to 0.1051 4 Density 93.80 5 te (d, e) n' 1000 °K 0.0011 4 4 g/ml 20°C 0.5951 5 2 93.81 o' 0.5888<sup>‡</sup> -0.0<sub>6</sub>36 25  $\mathbf{d_{4}^{t}}$ ΔHv/Te 19.86 5 30 Surface tension d -67 to 91.92 5 0.6227 dynes/cm. 20°C 12.50 5 <u>-9 ℃</u> 5 0.2293 ᇷᅴ ь -0.0011 4 30 5 11.29 to 10.11 5 40 Ref. Index e' 20°C [P]  $^{n}D$ Parachor d<sub>c</sub> g/ml 0.233 2 25 20°C vc ml/g t\_°C 2 4.296 30 30 t<sub>c</sub> 146. 2 40 "C" 2  $P_c$  mm 30172. Sugd. 179. 2 5 MR (Obs.) PV/RT Exp. L. l. %/wt. MR (Calc.) 20,205 5 25°C 0.9177 5 (nD-d/2)30 mm 1.0000 5 Dispersion Dielectric ВP 0.9518 5 Flash Point C t<sub>e</sub> 0.9540 5 -67 to 6.84290 2 Fire Point 2 926. 1 0.277 B |\_40°C 2 M. Spec. С 240. 2 AHc kcal/m 607.37 2 Ultra V. ΔHf A# -67 to 1.13369 5 X-Ray Dif. ΔFf B\*| \_ <u>2°C</u> 865.72 Infrared ĸ Viscosity Solubility in c centistokes Acetone to -55°C 2 η 0.32 ٠c Carbon tet. -45 0.28 2 Benzene A' -35 0.25 2 to Ether °C B n-Heptane B<sub>v</sub> | C' to Ethanol °C Water A1# to Water in B'\* (B<sup>V</sup>) °C to Acl 40 to 7.2793 5 (A<sup>V</sup>) °C Bc tc °C 1186. cp liq. ۰ĸ Cc 5 277 cp vap.300 K 0.06002 Cryos. A° 2 0.36664 2 consts. B° 0.0045 2 2 400 0.46414 c<sub>w</sub> vap. te °C -8, 229 5  $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES: # at saturation pressure # at saturation pressure (triple point)

							No. 4	
NAME	cis-2-But	ene				STRUCTURAL		
Mole % Pur.	Ref. Mo	lecul	ar C <sub>4</sub> H <sub>8</sub>	Molecular Weight 56.10	04	CH <sub>3</sub> CH=CH	сн <sub>3</sub>	
		Ref.			Ref		-	Ref
F.P. °C	-138.910	2	dt/dP	I		f to		
F.P. 100%			*C/mm			g <u>*K</u> _		
B. P. *C			25°C BP	0.0194 0.0345	5 2	h ;	· ·	
760 mm 100	3.720 -39.824	2 2	t	0.0352	5	f' to		
30	-58.94	2	30 mm	0,4780	5	g'   ' <u>*</u> K_		
10 1	-73.42 -97.2	2 5	ΔHm cal/g			h'		
Pressure	-/2	<del>  </del>	ΔHv cal/g			m   300 to	0.0176	
mm 25°C	1604.0	5	25°C	94.5 113. <b>3</b> 8	2 5	n 600 °K/	0.0012 -0.0 <sub>6</sub> 28	
t <sub>e</sub>	741.2	5	30 mm BP	99.46	2	ļ		
Density			te (d. a)	99.61	5	m'   700 to	0.0424 0.0012	4
g/ml 20°C	0.6213 <sup>‡</sup> 0.6154 <sup>‡</sup>	2 2	e (a, e)	99.61	5	0' 1000	-0.0 <sub>6</sub> 39	
d <sub>4</sub> 25	0.6094	4	ΔHv/T <sub>e</sub>	20.23	5	Surface tension	-	$\vdash$
a .	0.6464	4	d   -59 to	100.29 0.2221	5 5	dynes/cm. 20°C	15.07	5
ь	-0.0010	4	-a,-  - 3 °€	0.2221	"	30	13.80	5
Ref. Index			e' ' °C			40	12.55	5
<sup>n</sup> D 20°C		1	d g/ml vc ml/g	0.238	2	Parachor [P] 20°C		
30			t <sub>c</sub> *C	4.206 155.	2 2	30		İ
"C"			l -	31160.	2	40 Sugar	179.2	5
MR (Obs.)			P <sub>c</sub> mm	31100.	-	Exp. L.1.%/wt.	117.2	-
MR (Calc.) (nD-d/2)	20,205	5	25°C	0.8766	4	u.	1	
Dielectric	<del> </del>	$\vdash$	30 mm	1.0000	5	Dispersion		
A -59 to	6.86926	2	BP t <sub>e</sub>	0.9637 0.9643	4 5	Flash Point °C		
B 541 °C	960.1	2	tc	0.276	2	Fire Point		
с —— —	237.	2	ΔHc kcal/m	605.73	2	M Spec. Ultra V.		
A*  -59 to	1.1261	5	ΔHf ΔFf		į	X-Ray Dif.		
B*	894.9	5	Viscosity		Н	Infrared		<u> </u>
c	.]		centistokes	}		Solubility in +		
tk ToC			η ∘c			Carbon tet.		
t <sub>x</sub>   *C	<del> </del>	$\vdash$		)	١,	Benzene Ether	ļ	
B'i °C	Į					n-Heptane		
C'			B <sup>V</sup>   to A <sup>V</sup>   °C	1		Ethanol		
A'* to B'* °C	1			ļ		Water Water in		
Ac   541 to	7 2022	╁╤┤	(B <sup>V</sup> )  to	1				
Bc t <sub>c</sub> °C	7.3023 1229.	5	(A <sup>V</sup> )  °C		$\vdash$			1
Cc'	275.	5	c <sub>p</sub> liq. ∘K	1				1
Cryos. A°	0.04877	2 2	cp vap.300 °K	0.33794				
te °C	0.0052	5	c <sub>v</sub> vap.	0.43366	2			1
$T_{R} = 0.75$	3.07	1 - 1		L	L	I	L	L
		2	# at saturation	pressure		grams/100 gran		<u> </u>
	ES: 1-Dow	Z-AL AJ		aic. from def	t. dat	ta 5-Calc, by for	mula	
SOURCE:			·					
PURIFICAT		AI						
LIIERAIUI	RE REFERE	NCES	·:					

No. 5 trans-2-Butene NAME STRUCTURAL FORMULA CH<sub>2</sub>CH=CHCH<sub>2</sub> Mole Ref. Molecular Molecular C<sub>4</sub>H<sub>8</sub> % Pur Weight 56.104 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -105.550 2 dt/dP f to °C/mm ۰ĸ g 25°C 0.0181 B. P. ℃ h ВP 0.0345 2 760 mm 2 0.88 t<sub>e</sub> 0.0356 5 f١ 100 -42.69 to 2 g' °Κ 30 0.4783 5 -61.82 30 mm 10 -76.3 2 h! AHm cal/g 1 -100.1 5 300 to 0.0816 m AHv cal/g Pressure 600 °K 0.0011 n 25°C 91.8 2 mm 25°C 1753. ٥ -0.0625 4 30 mm 110.28 5  $\mathbf{t_e}$ 729.7 5 96.94 ВP 2 700 to 0.0851 Density m 4 97.17 5 te te (d, e) n' 1000 °K 0.0011 g/ml 20°C 0.6042 2 97.17 0.5984 ٥' -0.0637 4 25 2  $d_4^t$ ΔHv/Te 19.97 5 30 0.5925 4 Surface tension -62.0 to 97.13 5 0,6291 4 13.43 5 dynes/cm. 20°C -0, 2°C 0,2128 5 ᇷᅱ -0.0010 h 4 30 12.27 5 to 5 40 11.13 Ref. Index e' °C nD 20°C [P] Parachor d<sub>c</sub> g/ml 0.238 2 25 20°C vc ml/g t °C 4.206 30 30 t<sub>c</sub> 155. 2 40 "C" P<sub>c</sub> mm 31160. 2 Sugd. 179.2 5 MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 20,205 5 25°C 0.9333 (nD-d/2) 30 mm 1.0000 5 Dispersion Dielectric BP 0.9595 4 Flash Point C 0.9606 5 A -62 to 6.86952 2 Fire Point 0.276 2 960.8 B 1 49°C 2 M. Spec. C 240. 2 AHc kcal/m 604,73 Ultra V. ΔHf A\* -62 to 1.13404 5 X-Ray Dif. ΔFf B\*| \_ <u>10 °C</u> 895.9 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t<sub>x</sub> °C Benzene ۸' to Ether В' °C B<sup>V</sup> A n-Heptane to Ethanol °C Water A'\* to Water in B'\* (B<sup>V</sup>)| °C to (A<sup>V</sup>)| Acl 49 to 7.3082 5 °C Bc t<sub>c</sub> °C 1231. cp liq. °K Cc 5 278 c<sub>p</sub> vap.300 K Cryos. A° 0.04177 0.37573 2 consts. B° 2 0.0058 2 400 0.46378 c vap. te °C -0.183 5  $T_{\mathbf{R}} = 0.75 \, \mathbf{T}$ at saturation pressure grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

	_								<b>No.</b> 6	
NAME	2-M	ethyl	prop	ene			STI	RUCTURAL	FORMULA	1
	Isob	utene	,					CH <sub>3</sub> C = C	н,	
Mole	Ref.	Mo	lecul		Molecular	ļ		с́н <sub>3</sub>	2	
% Pur.			rmul		Weight 56, 1	04				
			Ref.			Ref.				Ref
F.P. °C	-140.35	0	2	dt/dP			f	to		
F.P. 100%	<b></b>			*C/mm 25*C	0,0145	4	g	<u>*K</u>		
B. P. *C 760 mm	-6.90	00	2	BP	0.0336	2	h	<del>                                     </del>		<u> </u>
100	-49.30	9	2	t <sub>e</sub>	0.0355	4	f' g'	to *K		
30 10	-67.90 -81.95		2 2	30 mm	0.4645	5	h'	'		
1	-105.06		5	ΔHm cal/g		-	m	1 300 to	0.0498	4
Pressure	T			ΔHv cal/g 25°C	87.7	2	n'	_600 °K_	0.0012	4
mm 25°C	2278. 710.1		4 5	30 mm	107.14	5	٥	i	-0.0 <sub>6</sub> 45	4
Density	+		-	BP	94.22 94.59	2	m'	700 to	0.1217	4
g/ml 20°C	0.59	)42 <sup>‡</sup>	2	te te (d, e)	94.59	4	n' o'	1 1000 .K	0.0011 -0.0 <sub>6</sub> 34	4
d <sub>4</sub> 25	0.58	379 °	2 4	AHv/T	20.06	5		<u> </u>	-0.0634	1
	0.62		4	d   -68 to		5		face tension es/cm. 20°C	12.42	_
ъ ~	-0.00		4	<u></u>		5	3,11	30	11.22	5
Ref. Index				e'   %				40	10.04	5
<sup>n</sup> D 20°C				d <sub>c</sub> g/ml	0.234	2	Par	achor [P] 20°C		
30				V_ mi/g	4.278 144.73	2 2	1	30		
"C"				t <sub>c</sub> •C	29982.	2		40 Suad	179.2	5
MR (Obs.)	I	_		PV/RT	27702.	-	Fyn	. L.1.%/wt.	117.2	-
MR (Calc.) (nD-d/2)	20.20	15	5	25°C	0.9314	4	-	u.		Ì
Dielectric	<del>                                     </del>		Н	30 mm BP	1.0000	5		persion		
A -68 to	6, 84	134	2	t.	0.9646	5		sh Point °C e Point		i
B  39°C	923.2		2	t <sub>c</sub>	0. 276	2		pec.		├
<u>C</u>	240.		2	ΔHc kcal/m	603.36	2	Ultr	a V.		
A*   -68 to B*1 °C	861.1	.232	5	ΔFf	ł	1		ay Dif. ared		
к — — —	1			Viscosity	ļ.		<b></b>	bility in +	<u> </u>	
t <sub>k</sub>   to	-{			centistokes 7°C	:	1	Ac	etone		1
t C				'		1		rbon tet. nzene		
A'   to						}	Etl	her		
B' •C	·			B <sup>V</sup>   to	<b>†</b>			Heptane nanol		
A¹* to	1			AV I °C			Wa	ter		
B'* °C				(B <sup>V</sup> ) to	7		W a	ter in		-
Ac 39 to	7.27	77	5	(A <sup>V</sup> )  ∘c	:					
Bc _tc_*C	1183.		5	c <sub>p</sub> liq. °K						
Cryos. A*	0.04	044	2	cp vap.300°K	0.38126	2				
consts. B°	0.00	)5	2		0.47358	2				
t <sub>e</sub> °C	-8.62	3	4	c <sub>v</sub> vap.			L			<u></u>
$T_R = 0.7$				≠ at saturation				ams/100 gran	ns solven	t
REFERENC	ES: 1-D	ow	2-AF		Calc. from de	t. da	a 5	Calc. by for	mula	
SOURCE:				PI						
PURIFICAT				PI						
LITERATU	RE REF	EREI	NCES	:					-	

F*								No. 7	
NAME	l-Penten	B		·		ST	RUCTURAL	FORMUL.	A
							CH (CH ) CH	1-611	
Mole % Pur.		lecul mula		Molecular Weight 70.13	0		сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сі	n=CH <sub>2</sub>	
	+	Ref.	l		Ref.				Ref.
F.P. °C F.P. 100%	-165, 220	2	dt/dP °C/mm 25°C	0.04297	5	f g	to		
B.P. °C 760 mm	29, 968	2	BP	0.03797	5	h			
100 30	-17.92 -38.91	2 2	t <sub>e</sub> 30 mm	0.0360	5	f' g'	to •K		
10	-54.8	2	ΔHm cal/g	+	+ -	h'			
l Pressure	-80.9	5	ΔHv cal/g	<del> </del>	-	m	300 to	0.0283	
mm 25°C	648.8	4	25°C 30 mm	86.93 98.81	5	n o	_6 <u>0</u> 0_•K	0.0013 -0.0 <sub>6</sub> 47	
t <sub>e</sub>	813.4	5	BP	86.24	5	m'	700 to	0, 1035	4
Density g/ml 20°C	0.64050	2	te te (d, e)	85.88 85.88	5	n' o'	1000 °K	0.0011	4
d <sub>4</sub> 30	0.63533	2 4	AHV/Te	19.74	5			-0.0 <sub>6</sub> 37	4
a 30	0.66168	4	d -39 to	91.71	5		face tension es/cm. 20°C	15.57	5
Ъ	-0.03954	4	e 32 °C		5	8,	30	14.50	5
Ref. Index n <sub>D</sub> 20°C	1.37148	2	e'	+		Pa 7	40 achor [P]	13.45	5
D 25	1.36835	2	d g/ml v ml/g	0.227 4.409	5		20°C		
30	1.36512 0.7764	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	188.	5		30 40		
MR (Obs.)	24.854	2	P <sub>c</sub> mm	25113.	5			218.2	5
MR (Calc.)		5	PV/RT 25°C	0.9657	5	Exp	o. L.1.%/wt.		
Dielectric	1,05123	-	30 mm BP	1.0000	5	Dis	persion	126.1	2
A -39 to	6,84650	2	t	0.9600 0.9581	5		sh Point C e Point		İ
B   73 °C	1044.9 234.	2 2	t <sub>c</sub> ΔHc kcal/m	0.270	5		Spec.		
A* -39 to	1.16738	5	ΔHf	754.25	-		ra V. Ray Dif.		
B* _ 42 °C	974.8	5	ΔFf	<del> </del>			ared		
c			Viscosity centistokes				ubility in +		
t <sub>k</sub> to			7 -20 °C -10	0.30 0.27	2 2	Ca	rbon tet.		
A'   to	<del> </del>		ō	0.24	2		nzene her		
B'  °C			B <sub>v</sub> to	·			Heptane hanol		
A¹* to			LA`	_		W	ter		
B1* °C			(B <sup>V</sup> )  to			- W:	ter in		-
Acl 73 to Bc t <sub>c</sub> °C	7.2751 1324.	5	(A <sup>V</sup> )  °C	<del></del>	$\vdash$				
Cc	272.	5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°	0.05995 0.0048	2	c vap.300°K	0.37516 0.47198	2 2				
t <sub>e</sub> °C T <sub>R</sub> = 0.75	31.94	5	c <sub>v</sub> vap.	<u></u>	Ll	L	(100	<u> </u>	<u> </u>
	ES: 1-Dow	2-A	PI 3-Lit. 4	-Calc, from de	t da		ams/100 gra		t
SOURCE:		AF		care, mom de	., ua		Carc. by 101		
PURIFICAT	ION:	AF							
	RE REFERE	NCES	S:						

								No. 8	
NAME	cis-	2-Pent	ene				STRUCTURAL	FORMULA	<b>.</b>
							CH <sub>3</sub> CH <sub>2</sub> CH=	сн сн.	
Mole % Pur.	Ref.	Mole Form			Molecular Weight 70.13	.	03020	3	
	·····		lef.			Ref			Ref
F.P. °C	-151.39		2	dt/dP	T		f   to		
F.P. 1007				°C/mm			g L°K		ļ
B. P. °C				25°C BP	0.0537 0.03830	5 2	h ,	1	
760 mm 100	36.94		2 2	te	0.0358	5	f' to		
30	-32.66		2	30 mm	0.5310	5	g'   'K_	l	ł
10 1	-48.7 -75.2		2 5	ΔHm cal/g			h'		L
Pressure	+	-		ΔHv cal/g			m   300 to	-0.0106 0.0013	
mm 25°C	494.6		5	25°C 30 mm	91.48	5 5	n   _600°K	-0.0 <sub>6</sub> 46	
t <sub>e</sub>	829.6		5	BP	89.12	5	70045		├
Density g/ml 20°C	0.65	56	2	te (d. a)	88.62	5 5	m'   700 to n'   1000 K	0.0568	
at 25	0.65		2	te (d, e)	88.61	1 1	0'	-0.0641	4
<b>4</b> 30	0.64	52	4	ΔHv/T <sub>e</sub>	19.87	5	Surface tension		$\vdash$
a b	0.67		4	d   -33 to		5	dynes/cm. 20°C	17.14	5
Ref. Index	-0.0	70	-	d'			30 40	16.01 14.89	5
n <sub>D</sub> 20°C		30	2		<del> </del>	<u> </u>	Parachor [P]		一
25 30	1.37		2	d g/ml vc ml/g	0.232 4.303	5 5	20°C		1
"C"	1.37	-	4	tc °C	210.	5	30 40		İ
	0.78		4	P <sub>c</sub> mm	26762.	5		218.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT	0.0(7)		Exp. L.1.%/wt.		
(nD-d/2)	1.05	52	2	25°C 30 mm	0.9671 1.0000	5	u. Dispersion	130.	2
Dielectric				BP	0.9560	5	Flash Point °C		┢▔
A   -33 to		274	2	te t <sub>c</sub>	0.9535 0.268	5	Fire Point		
B [_82*9	231.		2 2	ΔHc kcal/m	752,74	2	M Spec.		
A*   -33 to		088	5	ΔHf			Ultra V. X-Ray Dif.		
B* ∟ 50 °C	1002.4		5	ΔFf	<u> </u>	$\vdash$	Infrared		
c		l		Viscosity centistokes			Solubility in +		
tk   to				η ∘c			Acetone Carbon tet.	1	l
A' to		_	_				Benzene		
B' •(							Ether n-Heptane		
C1				B <sup>V</sup> to	j		Ethanol		ļ
A'* to B'* *0				AV   °C	-		Water Water in		
Ac   82 to	<del></del>	73	5	(B <sup>V</sup> ) to	1	i		<u> </u>	$\vdash$
Bc tc °C	1350.	"	5	(A <sup>V</sup> )  °C	<b></b>	L			ł
Cc — —	270.		5	c <sub>p</sub> liq. °K					ļ
Cryos. A° consts. B°	0.05		2 2	c <sub>p</sub> vap 300 °K 400	0.34864 0.45016	2 2			
t <sub>e</sub> °C	39.52		5	c <sub>v</sub> vap.	0,15010				
$T_R = 0.7$	5 T <sub>C</sub>						+ grams/100 grai	ns solven	<u> </u>
REFEREN	CES: 1-D	ow 2	-AP		Calc. from det	dat	ta 5-Calc. by for		
SOURCE:			AF	PI					
PURIFICA'			AF						
LITERATU	RE REF	ERENC	CES	:					
		<u> </u>							

No. 9 trans-2-Pentene STRUCTURAL FORMULA NAME CH3CH2CH=CHCH3 Molecular Weight 70.130 Mole Ref. Molecular  $C_5H_{10}$ % Pur. Formula Ref Ref. Ref. F.P. °C F.P. 100% -140.244 2 dt/dP f to °C/mm 25°C g <u>°K</u> 0.0527 5 B.P. °C h ВP 0.03824 2 760 mm 36.353 2 0.0358 5 ſ١ te to 100 -12.00 2 g' <u>°К</u> 30 30 mm -33. 28 2 0.5325 5 10 -49.4 2 h! ∆Hm cal/g 5 -76.0 m 300 to 0.0406 AHv cal/g Pressure |\_600<u>°</u>K n 0.0012 25°C 91.02 mm 25°C 505.5 5 -0.0639 o 4 30 mm 102.10 828.0 5 te BP 5 88.92 m' 700 to 0.0980 Density 4 5 88.44 n' 1000 °K 0.0011 g/ml 20°C te (d, e) 4 0.6482 5 2 88.44 ۰' -0.0636 4  $\mathbf{d_4^t}$ 25 0.6431 ΔHv/Te 5 19.88 30 0.6379 4 Surface tension d -33 to 95,80 5 a 0.6690 dynes/cm. 20°C 5 16.38 0.1893 ᇷᅴ <u>39 ℃</u> Ъ -0.0396 4 30 15.29 5 40 14.23 5 Ref. Index e' <sup>n</sup>D 1.3793 20°C [P] Parachor d<sub>c</sub> g/ml 0.237 4 25 1.3761 2 20°C v<sub>c</sub> ... ml/g 4,216 30 1.3729 4 30 t<sub>c</sub> 199. 5 40 "C" 0.7825 4  $P_c$  mm 26693. 5 Sugd. 218.2 5 MR (Obs.) 25.02 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 24.36 5 25°C 0.9653 (nD-d/2)1.0552 2 u. 30 mm 1.0000 Dispersion 132. 2 Dielectric ВP 0.9560 5 Flash Point C 0.9536 5 A -33 to 6.90575 2 Fire Point 0.268 5 B |\_ 81 °C 1084.0 M. Spec. Ultra V. C 233. 2 AHc kcal/m 751.66 2 ΔHf A\* -33 to 1.22421 5 X-Ray Dif. ΔFf B\*[\_ 49 °C 1013.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to t<sub>x</sub> Carbon tet. °C Benzene A' | to Ether В' °C  $\overline{\overset{B^{V}}{A^{V}}}$ n-Heptane C to Ethanol °C Water A'\* to B'\* Water in °C (B<sup>V</sup>) to Acl 81 to 7.3347 5 (A<sup>V</sup>)| °C 1370. Bc tc °C c<sub>p</sub> liq. ۰ĸ 272. Cc' 5 cp vap.300K Cryos. A 0.05685 2 0.37131 2 consts. B° 0.0052 0.46585 2 400 c<sub>v</sub> vap. te °C 38.86  $T_{R} = 0.75 T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 10	
NAME	2-Methy	l-l-bu	tene			STRUCTURAL	FORMULA	¥.
						CH CH C	CU	
			T			сн <sub>3</sub> сн <sub>2</sub> с сн	CH <sub>2</sub>	
Mole % Pur.		iolecul formul		Molecular Weight 70,13	ا ه	On	3	
		Ref.	řeni i i i i i i i i i i i i i i i i i i		Ref	l l		Ref
F.P. °C	-137.560	2	dt/dP			f   to		厂
F.P. 100%			°C/mm	l		g L		1
B. P. °C			25°C BP	0.0449 0.03778	5 2	h ,		ĺ
760 mm 100	31.163	2 2	t <sub>e</sub>	0.0358	5	f' to		
30	-37.50	2	30 mm	0.5239	5	g'	4	
10 1	-53.4 -79.5	2 5	ΔHm cal/g			h'	ļ.,	L
Pressure	1	+	ΔHv cal/g			m   300 to n   600 °K	0.0201	
mm 25°C	610.0	4	25°C 30 mm	88.15 100.15	5 5	-020 21.	-0.0656	
t <sub>e</sub>	813.	5	BP	86.99	5	m'   700 to	0,1147	
Density g/ml 20°C	0.6504	2	te (d, e)	86.62 86.62	5 5	n' 1000°K	0.0011	4
at 25	0.6451	2	ΔHv/T <sub>e</sub>	19.83	5	0'	-0.0 <sub>6</sub> 37	4
	0.6397	4	d   -38 to	<del>                                     </del>	5	Surface tension		
a b	0.6721 -0.0 <sub>3</sub> 98	4 4	_e _ 3 <u>3</u> °C	0.1917	5	dynes/cm. 20°C	16.56 15.42	5
Ref. Index	1		d'   to			40	14.30	5
n <sub>D</sub> 20°C	1.3778	2	d <sub>c</sub> g/ml	0, 237	5	Parachor [P]		
25 30	1.3746	2 4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.224	5	20°C 30		ļ
"C"	0.7769	4		191.	5	40		_
MR (Obs.)	24, 85	2	P <sub>c</sub> mm	26188.	5		218.2	5
MR (Calc.)		5	PV/RT 25°C	0.9618	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.0526	2	30 mm	1.0000	5	Dispersion	133.	2
Dielectric A -38 to	4 0721	4 2	BP t <sub>e</sub>	0.9560 0.9541	5	Flash Point °C		
B75 °C		2	tc	0, 268	5	Fire Point	ļ <u> </u>	├
С	233.	2	AHc kcal/m	750.57	2	M Spec. Ultra V.		
A*   -38 to	1.2040 986.94	4   5	ΔHf ΔFf	[		X-Ray Dif.		
B*	. 700.71	'	Viscosity			Infrared Solubility in +		┢
c tto			centistokes 7°C			Solubility in + Acetone		
tk to	Ì		η •c			Carbon tet.		
A'   to	<b>†</b>					Benzene Ether		
B' L _ <u>*C</u>			B <sup>V</sup>   to	<del> </del>		n-Heptane		
A¹* to	<u> </u>	+	B' to C			Ethanol Water		
B'* °C			(B <sup>V</sup> ) to	-		Water in		<u> </u>
Ac   75 to	7.3005	5	(A <sup>V</sup> ) °C					
Bc tc °C	1334. 271.	5	c <sub>p</sub> liq. °K					
Cryos. A°	0.0516		c <sub>p</sub> vap.300°K	0.38243	2			
consts. B°	0.0043	2	400	0.48068				
t <sub>e</sub> °C	33.12	5	c <sub>v</sub> vap.					1
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>					+ grams/100 gra	ms sol <b>ve</b> n	t
REFERENC	ES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat			
SOURCE:		AI	PI					
PURIFICAT		AI						
LITERATUI	RE REFER	ENCES	i:					

					$\overline{}$			
NAME _	3-Methyl	-1-b	utene			STRUCTURAL	FORMUL	A
						сн <sub>3</sub> сн сн	=CH <sub>2</sub>	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 70.13	0	CH <sub>3</sub>	2	
		Ref.		T T	Ref.	T''		Ref.
F.P. °C F.P. 100%	-168.495	2	dt/dP °C/mm			f to g <b>*</b> K		
B. P. °C 760 mm 100 30 10 1	20.061 -26.82 -47.3 -62.9 -88.3	2 2 2 2 5	25°C BP te 30 mm  AHm cal/g  AHv cal/g	0.0326 0.03721 0.0363 0.5128	5 2 5 5	h   to g'  *K   h'   m   300 to n   600 *K	0.0231 0.0015	4 4
mm 25°C	902. 784.	5 5	25°C 30 mm BP	81.51 93.96 82.20	5 5 5	0	-0.0677	4
Density g/ml 20°C dt 25 4 30	0.6272 0.6219 <sup>‡</sup> 0.6168 <sup>‡</sup>	2 2 4	t <sub>e</sub> (d, e)  ΔHv/T <sub>e</sub>	81.96 82.05 19.54	5 5 5	m' 700 to n' 1000 °K o' Surface tension	0.1380 0.0011 -0.0 <sub>6</sub> 37	4 4
a b	0.6490 -0.00094	4	d -47 to e -21 °C to	85.70 0.1744	5	dynes/cm. 20°C 30 40	15.57 14.50 13.45	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30	1.3643 1.3611 <sup>‡</sup> 1.3587 <sup>‡</sup>	2 2 4	e'   °C  d g/ml vc ml/g t °C	0.219 4.568 170.	5 5 5	Parachor [P] 20°C 30	13.43	
"C"	0.7784	4	t <sub>c</sub> °C P <sub>c</sub> mm	23294.	5	40 Sugd.	218.2	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	24.94 24.36 1.0507	2 5 2	PV/RT 25°C 30 mm BP	0.9518 1.0000 0.9550	5 5 5	Exp. L.1.%/wt. u. Dispersion Flash Point °C		
A -47 to B   60 °C C	6.82618 1013.474 237.	2 2 2	te tc	0,9576 0,270 752,33	5 5 2	Fire Point M. Spec.		
A* -47 to B* 31 °C K c t <sub>k</sub> to	1.16759 946.98	5	ΔHf ΔFf  Viscosity centistokes η °C			Ultra V. X-Ray Dif. Infrared  Solubility in + Acetone		
A'   to B'   _ °C C'			-V			Carbon tet. Benzene Ether n-Heptane Ethanol		
A¹* to B¹* °C			$\frac{A'}{(B') } - \frac{^{\circ}C}{to}$			Water Water in		
Acl 60 to Bc t <sub>c</sub> °C	7, 2588 1286. - 274.	5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K					
Cryos, A° consts, B°	0.05888 0.0047	2 2	c vap.300°K 400	0.40596 0.50278	2 2			
t <sub>e</sub> °C	20.93 T ≠	5	c <sub>v</sub> vap.	L	L	L		
	T <sub>c</sub> <sup>#</sup> at satu			C-1- ( · ·	. ,	grams/100 gra		t .
SOURCE:	E3: 1-DOW	Z-A AF		Caic. Irom de	ı. da	ta 5-Calc. by for	muia	
PURIFICAT	'ION:	AF		·				
	RE REFERE							

						No. 12	
NAME	2-Methyl	-2-b	utene			STRUCTURAL FORMULA	
						CH CH-C CH	
			T			сн <sub>3</sub> сн=с сн <sub>3</sub> сн <sub>3</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 70,13		03	
		Ref.	Ī		Ref	]	Ref.
F.P. °C	-133.768	2	dt/dP			f   to	
F.P. 1007	6		°C/mm	0.05((	5	g <u>*K</u>	
B. P. °C 760 mm	38,568	2	25°C BP	0.0566 0.03844	2	h	
100	-10.06	2	t <sub>e</sub>	0.0 <b>3</b> 57	5	f' to	
30 10	-31.48 -47.7	2 2	30 mm	0.5360	5	g'  K_	
ĭ	-74.5	5	ΔHm cal/g			J	4
Pressure			ΔHv cal/g 25°C	91.86	5	n 600 °K 0.0012	4
mm 25°C	466.1 834.3	5	30 mm	102.96	5	0 -0.0641	4
Density	+	Ť	BP	89.73 89.21	5	m'   700 to 0.0692	4
g/ml 20°0		2	te te (d, e)	89.21	5	n'   1000 °K   0.0011   -0.0 <sub>6</sub> 39	4
dt 25	0.6570 0.6517	2	AHv/Te	19.89	5		
	0,6838	4	d   -31 to	97.01	5	Surface tension dynes/cm, 20°C 17.87	5
Ъ	-0. <b>0</b> 210	4		0.1889	5	30   16.68	5
Ref. Index			e'   •C			40 15.51	5
n <sub>D</sub> 20°0	1.3874	2 2	d g/ml vc ml/g t °C	0.242	5	Parachor [P] 20°C	
30	1,3809	4	t <sub>c</sub> *C	4.128 204.	5	30	
"C"	0.7813	4	P <sub>c</sub> mm	27547.	5	40 Sugd. 218. 2	5
MR (Obs. MR (Calc.		2 5	PV/RT			Exp. L.1,%/wt.	
(nD-d/2)	1.0563	2	25°C 30 mm	0.9645 1.0 <b>0</b> 00	5	u.	•
Dielectric			BP	0.9560	5	Dispersion 135. Flash Point °C	2
A -31 t	6.91562	2	t <sub>e</sub> t <sub>c</sub>	0.9534 0.268	5	Fire Point	
B L 85 °	233.	2	ΔHc kcal/m	749.08	2	M Spec.	
A*   -31 to	1.24011	5	ΔHf			Ultra V. X-Ray Dif.	
B* ∟ 51 °	2 1026.4	5	ΔFf Vices sites	<b></b>	-	Infrared	
c	_		Viscosity centistokes	,		Solubility in +	
tk t			η •c	]		Acetone Carbon tet.	
A' to		<u> </u>				Benzene Ether	
B' L _ •	<u> </u>		B <sup>V</sup>   to			n-Heptane	
C'		-	BV to			Ethanol Water	
A'* to B'* **			(B <sup>V</sup> ) to			Water in	
Ac   85 to	7.3453	5	(A <sup>V</sup> ) •C				
Bc tc °	C 1385. 272.	5	c <sub>p</sub> liq. °K			1	
Cryos. A	<del></del>	2	c <sub>p</sub> vap.300°K	0.35962	2		
consts. B		2	400	0.35962	2		
t <sub>e</sub> °C	41.32	5	c <sub>v</sub> vap.				
$T_R = 0.7$						grams/100 grams solvent	
	CES: 1-Dow	2-AI		alc. from det	da	ata 5-Calc. by formula	
SOURCE:			PI				
PURIFICA			PI				
LITERATU	RE REFERE	NCES	<b>5:</b>				

No. 13 1-Hexene NAME STRUCTURAL FORMULA CH3(CH2)3CH=CH2 Mole Ref. Molecular Molecular C 6H12 Weight 84.156 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% -139.819 2 dt/dP f to °C/mm 25°C g °Κ 0.1267 5 B. P. ℃ h ВP 0.04149 760 mm 63.485 2 2 <sup>t</sup>e 0.0362 5 ſ١ 11.109 to 100 g' <u>°К</u> -11.88 30 30 mm 5 0.5748 2 10 -29.3 h١ ∆Hm cal/g -57.9 5 300 to 0.0258 m AHv cal/g Pressure 1 600 °K 0.0013 n 25°C 88,33 mm 25°C 187.2 5 o -0.0<sub>6</sub>48 4 30 mm 93.50 5 5 905. t<sub>e</sub> ВP 80.66 5 m' 700 to 0.1031 4 Density 79.72 5 te te (d, e) 1000 °K 0.0011 n' g/ml 20°C 0.67317 2 5 79.70 ٥' -0.0638 4  $d_4^t$ 25 0.66848 2 AHV/T 5 19.60 30 0.66377 Surface tension d -12 91.48 5 to a 0.69199 dynes/cm. 20°C 17.87 ᇷᅴ <u>69</u> °C 0.1705 5 -0.03906 ь 4 30 16.86 5 to 15.86 5 40 Ref. Index e' <sup>n</sup>D 20°C 1.38788 [P] Parachor d<sub>c</sub> g/ml 0.230 25 1.38502 2 20°C vc ml/g t\_°C 4.342 5 30 1.38219 4 30 t<sub>c</sub> 5 228. 40 "C" 0.7696 4 P<sub>c</sub> mm 22668. 5 Sugd. 257. 3 5 MR (Obs.) 29.492 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 28.978 25°C 0.9857 (nD-d/2)u. 1.05130 2 30 mm 1.0000 5 Dispersion 121.9 2 Dielectric BP 0.9550 5 Flash Point C 0.9499 A -12 to 6.86572 2 Fire Point 0.265 5 103 °C В 1152.971 2 M. Spec. AHc kcal/m С 226. 2 901.14 2 Ultra V. ΔHf A\* -12 to 1.23301 5 X-Ray Dif. ΔFf B\*[\_ 79 °C 1079. Infra red ĸ Viscosity Solubility in centistokes Acetone to tk tx -10°C 0.37 2 Carbon tet. °C 20 0.26 2 Benzene 40 0.22 2 A' to Ether 60 0.19 В' °C n-Heptane B<sup>V</sup> | C to Ethanol °C Water A'\* to Water in (B<sup>V</sup>)| B'\* °C to Acl 103 to 7,2845 (A V) 5 °C Bc tc " ٠c 1445. 5 c<sub>p</sub> liq. ۰ĸ Cc 265. 5 cp vap.300K Cryos, A 0.37763 2 consts. Bº 400 0.47566 2 c<sub>v</sub> vap. t<sub>e</sub> °C 69.08 5  $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 14	
NAME	cis-2-He	xene				STRUCTURAL	FORMULA	
	D-6 No	1 1		Mala -ula m		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH	=CHCH <sub>3</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 84.1	56			
		Ref.			Ref.			Ref.
F.P. C	-141.135	2	dt/dP			f to		
F.P. 1007	<u> </u>	<u> </u>	*C/mm 25*C	0,1530	5	g <u>°K</u> _		
B. P. *C 760 mm	68.84	2	BP	0.041	2	h		
100 30	15.85 -7.46	4	t <sub>e</sub>	0.0361	5	f' to g'K		
10	-25.13	5	30 mm ΔHm cal/g	0.5832	5	h'		
1	-54.23	5	ΔHv cal/g			m   300 to	0,0155	4
Pressure mm 25°C	151.	5	25°C	89.40	5	n 600 °K	0.0013 -0.0 <sub>6</sub> 36	4
t <sub>e</sub>	919.	5	30 mm BP	95.30 82.33	5	L		
Density	0 (0(0		t_	81.31	-5	m'   700 to n'   1000 °K	0.0978 0.0011	4
g/ml 20°0 dt 25 4 30	0.6869	2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	81.28	5	0, 1	-0.0 <sub>6</sub> 36	4
	0.6777	4	d   -7 to	19.65	5	Surface tension		
a b	0.7053 -0.0 <sub>3</sub> 89	4	<u>e l _75 °C</u>	94.03 0.1700	5	dynes/cm. 20°C	19.39 18.34	5 5
Ref. Index		Ė	d'   to			40	17.31	5
n <sub>D</sub> 20°0	1,3977	2	d <sub>c</sub> g/ml	0, 238	5	Parachor [P]		
25 30	1.3948 1.3920	2 4	11 V m1/g	4.202	5	20°C 30		l
, "C"	0.7722	4	tc °C	238.	5	40 Sund	257.2	_
MR (Obs.		2	P <sub>c</sub> mm	23801.	3	Exp. L.1.%/wt.	257.3	5
MR (Calc.   (nD-d/2)	) 28.978 1.0542	5 2	25°C	0.9847	5	u.		
Dielectric	<del></del>	亡	30 mm BP	1.0000 0.9540	5	Dispersion	125.	2
A   -7 t	6.89962	5	t <sub>e</sub>	0.9485	5	Flash Point °C Fire Point		
B 1_110.6	21184.6 226.	5	t <sub>c</sub> AHc kcal/m	0.264	-	M Spec.		
A*  -7 to	<del></del>	5	ΔHf	899.54	2	Ultra V. X-Ray Dif.		
B* L 85 °C		5	ΔFf		_	Infrared		
c c			Viscosity centistokes			Solubility in +		
tk   to		į	η ℃	ļ		Acetone Carbon tet.		
t of		-				Benzene Ether		
B' •		1	B <sup>V</sup>   to		<b> </b> i	n-Heptane		
C' to			B' to			Ethanol Water		
A'* to			(BV) to			Water in		
Ac   110 to	7, 32029	5	(A <sup>V</sup> ) °C					
Bc tc_"	C 1484. - 266.	5	c <sub>p</sub> liq. °K					
Cryos. A		Ť	c <sub>p</sub> vap.300°K	0.35874	2			
consts. B		L_	400	0.45867	2			
t <sub>e</sub> °C	75.01	5	c <sub>v</sub> vap.			l <u>, </u>		L
T <sub>R</sub> = 0.7						grams/100 gram	ns solvent	<u> </u>
	CES: 1-Dow			alc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION:	A1	<del></del>					
PURIFICA	RE REFERE	Al						
	NE REFERE		•					

								No. 15	
NAME	trans-2-H	Iexen	e			ST	RUCTURAL	FORMUL	A
						С	н <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн	ECH CH <sub>3</sub>	
Mole % Pur.	Ref. Mo	lecul rmula		Molecular Veight 84.150	6				
		Ref.		l	Ref.	Γ			Ref.
F. P. °C	-132.970	2	dt/dP			f	to		
F.P. 100%		<u> </u>	°C/mm 25°C	0.1400		gl	<u>*K</u>		İ
B.P. °C 760 mm	67.87	2	BP	0.1480 0.041	5 2	h			<u> </u>
100	15.01	4	t <sub>e</sub>	0.0361	5	f' g	to °K		
30 10	-8.25 -25.87	4 5	30 mm	0.5818	5_	h'	=		
<u> </u>	-54.90	5	ΔHm cal/g ΔHv cal/g		$\vdash$ $\dashv$	m	300 to	0.0582	4
Pressure mm 25°C	157.	5	25°C	89.70	5	n	_6 <u>0</u> 0_• <u>K</u>	0.0012	4
t <sub>e</sub>	916.	5	30 mm BP	94.98 82.05	5 5			-0.0631	ļ
Density	0 (70)		te te (d, e)	81.04	5	m'	700 to	0.0760 0.0012	4
g/ml 20°C dt 25 d4 30	0.6784 0.6738	2 2	t <sub>e</sub> (d, e)	81.02	5	0'	<u> </u>	-0.0642	
	0.6692	4	ΔHv/T <sub>e</sub>	19.65	5		face tension		
a b	0.6968 -0.0 <sub>3</sub> 89	4 4	_e _  _74_ °C	0.1698	5	dyn	es/cm. 20°C 30	18.44 17.43	5
Ref. Index	<del></del>	Ť	d'   to				40	16.45	5
<sup>n</sup> D 20°C	1.3935	2 2	d <sub>c</sub> g/ml	0,235	5	Par	achor [P]		
30	1.3879	4	v <sub>c</sub> ml/g	4.257	5 5		20 <b>°C</b> 30		
	0.7741	4	t <sub>c</sub> °C P <sub>c</sub> mm	236. 23399.	5		40 Sugd	257.3	5
MR (Obs.)		2	PV/RT			Exp	. L. 1. %/wt.	251.5	<u> </u>
MR (Calc. (nD-d/2)	) 28.978 1.0543	5 2	25°C 30 mm	0.9937	5		u.		
Dielectric		1	BP	1.0000 0.9540	5 5		persion sh Point °C	127.	2
A -8 to	6.89830	5	t e t c	0.9485 0.264	5		e Point		
B   108 °C C	_ 1181.0 226.	5	ΔHc kcal/m	898.54	2		Spec.		
A*  -8 to	1.26070	5	ΔHf ΔFf	i			ra V. lay Dif.		
B* _84°C	- 1106.0	5	Viscosity		$\vdash$		ared		_
t,	-		centistokes				ibility in <sup>T</sup> etone		
t <sub>k</sub> to			η °c			Ca	rbon tet.		
A'   to						Et	nzene h <b>er</b>		
B'°C	-		B <sub>v</sub> to				Heptane hanol		
A¹* to			A, C	1		Wa	ter		
B'* °C	<del> </del>	igspace	(B <sup>V</sup> )  to			₩ª	ter in		
Acl 108 to Bc t <sub>c</sub> °C	7.3188 1479.	5	(A <sup>V</sup> )  °C		$\vdash$				
Ce	266.	5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap.300K 400	0.37763 0.47174	2 2				
t <sub>e</sub> °C	73, 93	5	c <sub>v</sub> vap.	0.41174	-				
$T_R = 0.7$		<u> </u>	u	1	اا	+ gr	ams/100 gra	ms solven	t
		2-A	PI 3-Lit. 4-0	Calc. from de	t. da				
SOURCE:		AF	PI						
PURIFICAT		AF							
LITERATU	RE REFERE	NCE	5:						

							No. 16	
NAME	cis-3	-Hexene				STRUCTURAL	L FORMULA	۸.
						CH <sub>3</sub> CH <sub>2</sub> CH=	כחכח כח	
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 84.1	56	011301120110	01120113	
		Ref			Ref			Ref.
F.P. °C	-137.82		dt/dP	T T		f	to	
F.P. 100%			°C/mm	0.1410	_		<u>K</u>	ļ
B. P. °C 760 mm	1	١,	25°C BP	0.1410 0.041	5 2	h		
100	66.44 13.75		t <sub>e</sub>	0.0361	5		to	1
30 10	-9.43		30 mm	0.5797	5		<u>'K</u>	ļ
1	-26.98 -55.90		ΔHm cal/g		$\sqcup$	h'	0.0337	<u> </u>
Pressure			ΔHv cal/g 25°C	00.53	ا ۽ ا	m   300		
mm 25°C	166. 912.	5	30 mm	88.52 94.47	5 5	0	-0.0 <sub>6</sub> 53	4
Density	1,12.		BP	81.62	5	m'   700	0.0653	4
g/ml 20°C	0.67	96 2	te te (d, e)	80.64 80.62	5	n'   11000'	<b>'K</b> 」 0. <b>0</b> 012	4
dt 25 4 30	0.67		ΔHv/Te	19.64	5	0'	-0.0 <sub>6</sub> 42	4
a 30	0, 67		d   -94 to		5	Surface tensio		
ь	-0.03				5	dynes/cm. 20°	C   18.57   17.53	5
Ref. Index			d' to			40	16.52	5
n <sub>D</sub> 20°C			d_ g/ml	0,236	5	Parachor [P]		ĺ
30	1.39		d g/ml vc ml/g tc °C	4.245	5	20° 30		l
"C"	0.77	49 4	II -	233.	5	40	,	١_
MR (Obs.)		2	P <sub>c</sub> mm	23329.	3		gd. 257.3	5
MR (Calc. (nD-d/2)			25°C	0.9879	5	Exp. L.1.%/w u.	τ.	l
Dielectric	1.05	49 2	30 mm BP	1.0000	5	Dispersion	126.	2
A -94 to	6.89	493 5	t <sub>e</sub>	0.9540 0.9487	5 5	Flash Point °C		
B 1_107°C		193   5	tc	0.264	5	Fire Point		<u> </u>
С	226.	5	ΔHc kcal/m	899.54	2	M Spec. Ultra V.		1
A*  -94 to B* 82 °C		194 5	ΔFf			X-Ray Dif. Infrared		ļ
к — —			Viscosity				+	
t to	-	l	centistokes 7°C	1		Solubility in Acetone	·	1
t <sub>k</sub> to		ļ	η •c			Carbon tet. Benzene		
A' to			1			Ether		
B' L _ *	2	i	B <sup>V</sup> to	<del>                                     </del>		n-Heptane Ethanol	ŀ	
A¹* to			A C			Water		1
B1* °C			(B <sup>V</sup> ) to	1	1 1	Water in		<b>├</b>
Ac   107 to			(A <sup>V</sup> )					
Bc Ltc_°C	266.	5 5	cp liq. °K					Ì
Cryos. A°		1	c <sub>p</sub> vap.300°K	0.35303	2			1
te °C	72.33	5	c <sub>v</sub> vap.	0.45748	2			
$T_{\mathbf{R}} = 0.7$			ш	1		+ grams/100 g	<b>***</b>	<u> </u>
REFEREN		ow 2-A1	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-Calc. by f		<u> </u>
SOURCE:			PI	ue	-, was	J-Care. by I	111W1G	
PURIFICAT	TION:		PI					
LITERATU		ERENCES	S:	<del> </del>			······································	

No. 17

			2 11							No. 17	
NAME		trans-	-3-H	exen	e			ST	RUCTURAL	FORMUL	A
Mole		Ref.	Mol	ecul	ar	Molecular		c	н <sub>3</sub> сн <sub>2</sub> сн=сі	нсн <sub>2</sub> сн <sub>3</sub>	
% Pur.				mula		Weight 84.1	56				
				Ref.			Ref.				Ref.
F.P. °C		-113.43	0	2	dt/dP			f	l to		
F.P. 100	*			$\sqcup$	*C/mm 25*C	0 1441	_	g	<u>*K</u> _		1
B. P. °C 760 mm	ı	67.08		2	BP	0.1441 0.041	5 2	h			
100		14.30		4	t <sub>e</sub>	0.0361	5	f'	to		
30 10	- 1	-8.91 -26.50		4 5	30 mm	0.5807	5_	g'	<u>•K</u>		
1		-55.47		5	ΔHm cal/g			_h'		0.0101	-
Pressure		W.F			ΔHv cal/g		_	m n	300 to	0.0191 0.0014	4
mm 25°C	;	162.		5	25°C 30 mm	88.86 94.68	5	0		+0.0 <sub>6</sub> 53	4
t <sub>e</sub>	+	914.		5	BP	81.80	5	m'	700 to	0,0469	4
Density g/ml 20°	c	0.67	72	2	t <sub>e</sub> (d, e)	80.81 80.78	5	n'	1000 °K	0.0013	4
dt 25		0.67	25	2	ΔHv/T	19.64	5	٥'	ĺ	-0.0 <sub>6</sub> 47	4
<u> </u>	4	0.66		4	d -9 to		5		face tension		
a b	ł	0.69 -0.0 <sub>3</sub>		4	e   73 °C		5	dyn X	es/cm. 20°C 30	18.31	5
Ref. Inde	_	-0.03	71	-	d' to			"	40	17.28 16.27	5
n <sub>D</sub> 20°		1.39		2		0.234	5	Par	achor [P]		
25		1.39 1.38		2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	4.270	5		20°C 30		
"C"	$\dashv$	0.77		4	tc°°C	234.	5		40		
MR (Obs.	+	29.75		2	P <sub>c</sub> mm	23237.	5		<u> </u>	257.3	5
MR (Calc		28.97	8	5	PV/RT 25°C	0,9888	5	Exp	L.1.%/wt.		
(nD-d/2)	_	1.05	57	2	30 mm	1.0000	5	Dis	u. persion	128.	2
Dielectric					BP	0.9540	5		sh Point C		<u> </u>
A -9 to B   107 °c		6.89 1177 7	575	5	te tc	0.9486 0.264	5	Fir	e Point		
c		226.		5	ΔHc kcal/m	898,54	2		Spec.		
A* -9 to		1.25	454	5	ΔHf ΔFf	İ	İ		ra V. Ray Dif.		
B*[_83 °	<u>-</u>	100.5		5	Viscosity	<del> </del>	+	Infr	ared		<u> </u>
c	_				centistokes		1		ubility in +		
t <sub>k</sub> to					η °c	}			etone rbon tet.		
A'   to	_								nzene		
B' °							<u> </u>		her Heptane		
C'	_			Ш.	B <sup>V</sup>   to A <sup>V</sup>   °C			Et	hanol		
A'* to B'* **						-			ater ater in		
Ac  107 to	-+	7, 31	62	5							
Bc tc °		475.		5		<del>                                     </del>	1	1			
Cc —	1	266.		5	c <sub>p</sub> liq. °K						
Cryos, A consts, B					c vap,300°K	0.37906					
te °C	+	72 05	·	5	c <sub>v</sub> vap.	0,47768	2				
$T_{\mathbf{R}} = 0$ .	75.°	73,05 r_		)	L • -		<b></b>	+		L	
			OW	2-4	PI 3-Lit. 4	-Calc. from de	+ 4-		rams/100 gra		1£
SOURCE:	.CE	I-D		AP		-Caic, Irom de	ua	a. 3	-Carc. by for		
PURIFICA	TI	ON:		AP					·····		
LITERAT			ים פי								
_ mierai	J.A.	C REFI	ere!	NOE:	<b>.</b>						

NAME	2	-Me	thyl-	-l-pe	ntene		j	STR	UCTURAL	FORMULA	4
MAME			,-	- P			$\dashv$	011			-
Mole	J.	Ref.		lecul		Molecular			CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> C	= CH <sub>2</sub> H <sub>3</sub>	
% Pur.			F o	rmul	a 6-12	Weight 84.1					Ref
E B 46	1.2	5.72		Ref.	1.1.	1	Ref	<del></del>	1 1		Kei
F.P. °C F.P. 100%		5. 12	<u> </u>	1	dt/dP *C/mm			f g	to K		
B. P. °C	+			$\vdash$	25°C	0.1163	5	h	( <del>-</del> -		
760 mm		0.7		2	BP	0.0413 0.0361	5	f'	to to		一
100 30		8.77 4.06		4 4	t <sub>e</sub> 30 mm	0.5710	5	g'			
10	-3	1.35		5	ΔHm cal/g	0.5120	╅	h'	ŀ		ļ
	-5	9.83		5	ΔHv cal/g	<del> </del>	$\vdash$	m	300 to	0.0262	4
Pressure mm 25°C	20	6		5	25°C	85.91	5	n	_600 °K_	0.0014	
t <sub>e</sub>	89			5	30 mm BP	92.57 80.01	5 5	<u> </u>	1	-0.0 <sub>6</sub> 56	4
Density					t_	79.15	5	m'	700 to	0.1283	
g/ml 20°C		0.67		2	t <sub>e</sub> (d, e)	79.13	5	n' o'	11000 °K	0.0011 -0.0 <sub>6</sub> 36	4
dt 25 4 30		0.67 0.67		2 4	ΔHv/T <sub>e</sub>	19.64	5		1	11.5650	Ě
a	+	0,70		4	d   -14 te		5		ace tension s/cm, 20°C	18.78	5
Ъ		0.03		4	$-\frac{66}{4}$ , $-\frac{66}{4}$		5	8	30	17.70	5
Ref. Index			20		e'   °(			<u> </u>	40	16.64	5
n <sub>D</sub> 20°C		1.39 1.38		2 2	d g/ml vc ml/g	0.237	5	Para	chor [P] 20°C		
30		1.38		4	v ml/g	4. 223 225.	5		30		1
"C"		0.76	66	4	`		5		40 5d	257 2	_
MR (Obs.)		9.47		2	P <sub>c</sub> mm	23164.	-	F		257.3	5
MR (Calc. (nD-d/2)		8.97 1.05		5 2	25°C	0.9817	5	Exp.	L.1.%/wt. u.		
Dielectric	+	1.05	00	+	30 mm BP	1.0000	5	Disp	ersion	129.	2
A -14 to	+-	4 00	773	<del>  _  </del>	t <sub>e</sub>	0.9540 0.9492	5 5		h Point °C		
B 100°C		6.88 4.7	112	5	tc	0.265	5		Point		1
<u>c</u>	22	7.		5	AHc kcal/m	897.54	2	M S			
A*   -14 to		1.25	060	5	ΔHf ΔFf			X-R	ay Dif.		
B*	1076	b. 8		5	Viscosity		$\vdash$	Infra			ļ
°	_				centistokes	.			bility in + etone		
t <sub>x</sub>   to				] [	7 .	; [		Car	rbon tet.		
A' to				$\vdash$				Ber Eth	nzene		
B' °C						<u> </u>	$\vdash$	n-F	leptane		
C'				$\vdash \vdash$	B <sup>V</sup>   to			Eth Wat	anol		
A'* to						-1			ter in		
Ac   100 to	-	7.30	907	5							Г
Bc t °C	144	7.	/01	5							
Cc	26	7		5	c <sub>p</sub> liq. ∘K	i					
Cryos. A° consts. B°					c <sub>p</sub> vap,300 °K	0.38690 0.48481					
t <sub>e</sub> °C		5.92		5	c <sub>v</sub> vap.	1	$oxed{L}$	L		l	L_
$T_{\mathbf{R}} = 0.7$									ms/100 gran		t
	ES:	1 - De	w			Calc. from de	t. dat	a 5-	Calc. by for	mula	
SOURCE:				AI							
PURIFICAT				AF							
LITERATU	RE P	EFE	ERE	CES	:						

No. 19 3-Methyl-1-pentene STRUCTURAL FORMULA NAME CH3CH2CH CH=CH2 Ċн<sub>3</sub> Mole Ref. Molecular Molecular C6H12 Weight 84, 156 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% -153.0 2 dt/dP f to °C/mm g °K 25°C 0.0936 5 B.P. °C h ВP 2 0.040 760 mm 54.14 2 t<sub>e</sub> 0.0362 5 ſ١ to 100 3.06 4 g' <u>°к</u> 30 -19.38 4 30 mm 0.5612 5 10 -36.38 5 h' ∆Hm cal/g 5 1 -64.34 300 to -0.0261 m AHv cal/g Pressure \_600<u>°</u>K 0.0017 n 25°C 83,10 5 mm 25°C 264. 5 5 o -0.0682 4 30 mm 90.36 5 877. te BP 5 78.99 m' 700 to 0.1155 4 Density te (d, e) 77.39 5 0.0012 n' 1000 °K g/ml 20°C 4 0.6675 2 5 77.38 ٥' -0.0642 4  $\mathbf{d_{4}^{t}}$ 25 0.6628 2 AHV/T 19.63 5 30 0.6581 4 Surface tension ď -19 5 to 87.36 a 0.6865 4 dynes/cm. 20°C 17.24 5 ᇷᅴ <u>59 ℃</u> 0.1546 5 ь 4 -0.0390 30 16.24 5 40 15.25 5 Ref. Index e' <sup>n</sup>D 20°C 1.3842 [P] Parachor d<sub>c</sub> g/ml 0.234 5 25 1.3814 2 20°C vc ml/g t °C 4,276 30 1.3786 4 30 <sup>t</sup>c 213. 5 40 "C" 0.7691 4 P<sub>c</sub> mm 22329. 5 Sugd. 257.3 5 MR (Obs.) 29.49 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 28.978 5 0.9798 25°C 5 (nD-d/2)u. 1.0504 2 30 mm 1. 6000 Dispersion 2 5 5 124 Dielectric BP 0.9540 Flash Point C 0.9498 5 ţe. -19 to 6,87729 Fire Point 0.265 5 1130.4 5 1\_ 92 °C M. Spec. Ultra V. C 229. 5 AHc kcal/m 900.08 2 ΔHf A\* -19 to 5 1.26723 X-Ray Dif. ΔFf B\* 69 °C 1062.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $t_{\mathbf{x}}^{\mathbf{t}}$ Carbon tet. °C Benzene A'ı Ether B' °C n-Heptane Bv C' to Ethanol °C Water A'\* to Water in ۰c (B<sup>V</sup>) B'\* to Acl 92 to (A<sup>V</sup>)| 7.2989 5 °C Bc ۰c 1417. t<sub>c\_</sub>° c<sub>p</sub> liq. ۰ĸ Сс 267. 5 c<sub>p</sub> vap.300°K Cryos. A 0.41376 2 consts. B° 0.52759 2 400 c vap. te °C 5 58,61  $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

									No. 20	
NAME	4-M	ethyl	-1-p	ent <b>e</b> ne			STF	UCTURAL 1	FORMULA	
ŀ								сизси сиз	сн=сн-	
					- ·	$\neg$		ch <sub>3</sub> ch ch <sub>2</sub>	011-0112	
Mole % Pur.	Ref.	Mo	lecul rmul	ar C <sub>6</sub> H <sub>12</sub>	Molecular Weight 84.1	56		Cn <sub>3</sub>		
7 Pur.		FO	_		Weight 84.1	Ref				Ref
	T 152 (		Ref.		T	I.c.	_			IX.
F.P. C F.P. 100%	-153, 63		2	dt/dP *C/mm	Ĭ	1 1	f	to K		1
B. P. *C	+		$\vdash$	25°C	0.0928	5	g h	<del>-</del>		
760 mm	53.88	3	2	BP	0.0404 0.0361	2 5	f'	<del>                                     </del>		
100 30	2.84		4	t <sub>e</sub>	1	5	g'	to to		1
10	-19.58 -36.57		4 5	30 mm	0.5607	3	h'			
1	-64.51		5	ΔHm cal/g	<b>_</b>		<u> </u>	1 300 to	-0,0583	4
Pressure				ΔHv cal/g 25°C	82.95	5	m n	600 °K	0.0015	4
mm 25°C	267.		5	30 mm	90.29	5	0		- <b>0.</b> 0 <sub>6</sub> 57	4
t <sub>e</sub>	877.		5	BP	78.14	5	m'	700 to	0,0700	4
Density g/ml 20°C	0.66	42	2	te te (d, e)	77.41 77.40	5 5	n'	1000 °K	0.0012	4
dt 25 4 30	0.65		2	t <sub>e</sub> (d, e)	1		0'		-0.0 <sub>6</sub> 42	4
<sup>4</sup> 30	0.65	46	4	ΔHv/T <sub>e</sub>	19.65	5	Surf	ace tension	<del></del>	
	0.68		4	d   -20 to		5		s/cm. 20°C	16.90	5
ь	-0.03	92	4	-a'' &			,	30 40	15.90	5
Ref. Index	1.38	20	١, ١	e' i •(			<u> </u>		14.91	5
<sup>n</sup> D 20°C	1.37		2 2	d g/ml vc ml/g	0.233	5	Par	achor [P] 20°C		l
30	1.37	70	4	tc °C	4.295	5		30		
"C"	0.77	03	4	·	22185.	5		40 Su - d	257.3	-5
MR (Obs.)	29.54	1	2	P <sub>c</sub> mm	22165.	-			251.3	-
MR (Calc.)			5 2	25°C	0.9807	5	Exp	. L.1.%/wt. u.		
(nD-d/2)	1.05		-	30 mm	1.0000	5	Disp	ersion	124.	2
Dielectric	<del> </del>			BP	0.9550 0.9509	5		h Point °C		
A   -20 to B   91°C		757	5	te t <sub>c</sub>	0. 265	5	Fire	Point		<u> </u>
č – Æ º	229.		5	ΔHc kcal/m	899.44	2	M S			
A*   -20 to	1,25	996	5	ΔHf			Ultr X-R	a v. ay Dif.		
B+ <u>68 °C</u>	1059.0		5	ΔFf	<del> </del>		Infr			
K — — —	İ			Viscosity centistokes	1	1 1	Solu	bility in +		
ել	1			7 0	:			etone		l
t				l '		1 1		rbon tet. nzene		
A'   to						ì	Eth			
B' L _ •C	·			B <sup>V</sup>   to		1		leptane anol		l
A!* to	1.			A <sup>V</sup>   •C			Wa			[
B'* °C				(B <sup>V</sup> ) to	-		Wa	ter in		
Ac   91 to	7.29	91	5	(A <sup>V</sup> ) •c	ŀ					1
Bc Ltc_°C			5	c <sub>p</sub> liq. °K	<del> </del>	t d				[
	267.		5	11 -			Ì			
Cryos, A° consts, B°				c <sub>p</sub> vap.300°K 400	0.36171 0.46224	2 2				
t <sub>e</sub> °C	58, 35		5	c <sub>v</sub> vap.						[
$T_{R} = 0.7$			L	L	<del></del>		+ ~~	ms/100 gran		<u> </u>
REFERENC		~	2 - A E	OT 3_1.ie 4	Calc, from de	+ 4-4				τ
SOURCE:				PI	Care. Irom de	c. ual	J-	Care, by for		
PURIFICAT	ION:			PI						
LITERATU		CR EN								
KA I U	NE REF	-REI	·CES	•						

No. 21 2-Methyl-2-pentene NAME STRUCTURAL FORMULA CH3CH2CH=C CH3 ĊН Mole Ref. Molecular Molecular C6H12 % Pur Weight 84.156 Formula Ref. Ref. Ref. F.P. °C F.P. 100% -135.070 2 dt/dP f to °C/mm <u>\_°K</u> g 25°C 0.1451 5 5 B. P. °C h ВP 0.0418 760 mm 67, 29 2 t<sub>e</sub> 0.0362 5 f¹ 100 14.48 to 4 4 g' <u>°К</u> -8.75 30 5 30 mm 0.5811 5 10 -26.36 h' ∆Hm cal/g -55.34 5 300 to -0.0144 AHv cal/g 0.0014 -0.0<sub>6</sub>58 Pressure n <u>60</u>0\_°K 4 4 25°C 88.87 5 mm 25°C 160. 5 o 30 mm 94.73 5 914. 5 t<sub>e</sub> ΒP 81.84 5 m 700 to 0.0644 4 Density te (d, e) 5 80.84 0.0012 n' 1000 °K g/ml 20°C 0.6863 2 -0.0642 80.82 o' 4  $d_4^t$ 25 0.6815 2 AHv/Te 5 19.64 30 0.6767 4 Surface tension d -9 to 93.24 5 0.7055 4 dynes/cm. 20°C 19.32 7<u>3</u> °C 0.1695 5 -0.0393 ь 4 5 30 18.22 ā٦ to 40 5 17.16 Ref. Index e' °C <sup>n</sup>D [P] 20°C 1.4004 Parachor d<sub>c</sub> g/ml 5 0.237 25 1.3976 2 20°C vc ml/g t °C 4.216 5 30 1.3945 4 30 t<sub>c</sub> 235. 5 40 "C" 0.7778 4 P<sub>c</sub> mm 23580. 5 Sugd. 257.3 5 MR (Obs.) 29.75 PV/RT Exp. L. l. %/wt. MR (Calc.) 28.978 25°C 0.9837 5 (nD-d/2)1.0573 4 30 mm 2 1.0000 Dispersion 131. 5 Dielectric ВP 0.9540 Flash Point C 0.9486 5 -9 to 6.89488 5 Fire Point 0.264 5 B [\_108 °C 1178.1 5 M. Spec. AHc kcal/m С 5 226. 896.14 Ultra V. ΔHf A\*l -9 to 1.25698 5 X-Ray Dif. ΔFf B\*| 83 °C 1101.9 Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene A' to Ether В' ٠c n-Heptane B<sup>V</sup> | C' to Ethanol •c Water A'\* to Water in B'\* °C (B<sup>V</sup>)| to Acl 108 to (A<sup>V</sup>)| 7.3157 5 5 °C Bc tc C 1476. c<sub>p</sub> liq. ۰ĸ Cc 266. 5 c<sub>p</sub> vap.300 K Cryos. A 0.36147 2 consts. B° 0.46343 2 c, vap. te °C 73.28 5  $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{C}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 22	
NAME	3-Methyl	-cis-	-2-pentene			STRUCTURAL I	FORMULA	
Γ						CV CV C - C		
	1 .1					CH <sub>3</sub> CH <sub>2</sub> C = C	нсн <sub>3</sub>	
Mole % Pur.		lecul: rmul		Molecular Weight 84.1!	56	Ċн <sub>3</sub>		
		Ref.	l l		Ref	T		Ref.
F.P. *C	-138.445	2	dt/dP			f to		
F.P. 100%			°C/mm	0.40-	_	g°K		
B. P. °C	70.45		25°C BP	0.1627 0.0421	5 4	h ;		
760 mm 100	70.45 17.21	2 4	te	0.0362	5	f <sup>1</sup> to		
30	-6.21 -23.95	4	30 mm	0.5859	5	g'K_		
10 1	-53.19	4	ΔHm cal/g			h'		
Pressure	<b>†</b>		ΔHv cal/g	00.77	5	m   300 to	-0.0144 0.0014	
mm 25°C	140.97	4	25°C 30 mm	89.77 95.77	5	0	-0.0 <sub>6</sub> 58	
Donaltu	923.5	4	BP	82.73	5	m¹   700 to	0.0644	4
Density g/ml 20°C	0.6986	2	te te (d, e)	81.67 81.64	5	n'   11000 °K	0.0012	4
dt 25 4 30	0.6942	2	AHV/T	19.64	5	01	-0.0 <sub>6</sub> 42	4
a 30	0.6898	4	d   -6 to	94.71	5	Surface tension	20.75	5
ь	-0.03855	4	-å,-¦77- °€	0.1701	5	dynes/cm. 20°C	20.75 19.69	5
Ref. Index			d'   to			40	18.65	5
<sup>n</sup> D 20°C	1.4045 1.4018	2 2	d g/ml	0.242	5	Parachor [P] 20°C		
30	1.3990	4	d g/ml vc ml/g tc °C	4.140	5	30		
"C"	0.7715	5	-	243.	5	40	257.2	_ ا
MR (Obs.)	29.49	2	P <sub>c</sub> mm	24391.	"	Exp. L.1.%/wt.	<b>2</b> 57.3	5
MR (Calc. (nD-d/2)	28.978 1.0552	5 2	25°C	0.9815	5	u.		ļ
Dielectric	1.0332	Ť	30 mm BP	1.0000 0.9540	5	Dispersion	131.	2
A -6 to	6, 8985	5	te	0.9483	5	Flash Point °C Fire Point		ĺ
B 1114°C	1189.5	5	tc	0.264	5	M Spec.		-
. C	225.6	5	ΔHc kcal/m ΔHf	896.78	2	Ultra V.		
A*  -6 to B*  87 °C	1.25784	5	ΔFf			X-Ray Dif. Infrared		
к ———			Viscosity			Solubility in +		<del>                                     </del>
c k  to	-		centistokes 7°C			Acetone		
t <sub>x</sub> '°C	1		'			Carbon tet. Benzene		
A'   to				İ		Ether		
B' L _ *C	-1		B <sup>V</sup>   to			n-Heptane Ethanol		
A¹+ to			ĂV I C			Water		
B'* °C		L	(B <sup>V</sup> ) to			Water in		_
Ac   114 to		5	(A <sup>V</sup> )  °C					
Cc c_	266.4	5	c <sub>p</sub> liq. °K					
Cryos, A°			cp vap.300°K	0.36147				
consts. B°	<b> </b>	<b> </b>	400	0.46343				
t <sub>e</sub> °C	76.82	5	c <sub>v</sub> vap.			L <u>.</u>		L
TR = 0.7		• •				grams/100 gram	ns solvent	t
	ES: 1-Dow	2-AI		alc. from det	da	ta 5-Calc. by for	mula	
SOURCE:	TON.	AI	<del></del>					
PURIFICAT			PI					
LIIEKATU	RE REFERE	NCES	<b>:</b>					
l								

No. 23 3-Methyl-trans-2-pentene NAME STRUCTURAL FORMULA  $CH_3CH_2C = CHCH_3$ Ċн<sub>3</sub> Mole Ref. Molecular Molecular C6H12 % Pur Formula Weight 84.156 Ref. Ref. Ref. F.P. °C F.P. 100% -134.840 2 dt/dP f to °C/mm °K g 25°C 0.1468 5 B. P. °C h ВP 5 5 0.0418 760 mm 67.63 2 0.0362 ſ١ to 100 14.77 4 g' ۰ĸ 30 -8.48 4 30 mm 0.5816 5 10 -26.09 5 h' ∆Hm cal/g 1 -55.11 300 to -0.0144 m AHv cal/g Pressure |\_600<u>°K</u> n 0.0014 4 25°C 89.07 5 mm 25°C 158. 5 o -0.0658 4 30 mm 94.85 5 915. 5 t<sub>e</sub> BP 5 82.38 m' 700 to 0.0644 4 Density 80.94 5 t<sub>e</sub> (d, e) n' 1000 °K 0.0012 4 g/ml 20°C 0.6942 2 5 80.92 ٥' -0.0642 4 d<sub>4</sub> 25 0.6898 2 AHv/Te 19.64 5 30 0.6854 4 Surface tension -8 to 93.46 5 a 0.7119 -0.0<sub>3</sub>86 4 4 20.22 dynes/cm. 20°C 5 1 <u>74 ℃</u> 0.1638 5 h 30 19.18 5 to 40 18.15 5 Ref. Index e' | ٠c 20°C 1.4016 [P]  $n_D$ Parachor dc g/ml 0.241 5 25 1.3989 2 20°C 5 vc ml/g 4.151 30 1.3962 4 30 <sup>t</sup>c 238. 5 40 "C" 0.7712 4  $P_c$  mm 24093. 5 5 Sugd. 257.3 MR (Obs.) 29.49 28.978 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 25°C 0.9823 1.0545 5 (nD-d/2)4 30 mm 2 1.0000 5 Dispersion 131. Dielectric BP 0.9540 5 Flash Point C 0.9486 5 A -8 to 6,89552 Fire Point 5 0.264B [110°C 1179.4 5 M. Spec. 5 С 226. AHc kcal/m 896.78 Ultra V ΔHf A\* -8 to 1.25531 5 X-Ray Dif. ΔFf B\* 84 °C 1103.0 Infra red K Viscosity Solubility in c centistokes Acetone tk tx to Carbon tet. °C Benzene ۸' to Ether B °C n-Heptane  $\mathbf{B}^{\mathbf{v}}$ C to Ethanol Ăv i °C Water A'\* to Water in (BV) B!# °C to Ac 110 to 7.3167 5 (A<sup>V</sup>) °C Bc tc °C °C 1480. 5 ۰ĸ cp liq. 266. c<sub>p</sub> vap.300°K Cryos. A 0.36147 2 consts. B° 400 0.46343 2 c, vap. te °C 5 73.66  $= 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 5-Calc. by formula 4-Calc. from det. data API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 24	
NAME	4-Methy	l-cis	-2-pentene			STRUCTURAL	FORMULA	<b>L</b>
						сн <sub>3</sub> сн сн=с	нсн.	
M-1-	2.6	1		Molecular		ċн <sub>3</sub>	3	
Mole % Pur.	Ref. Mo	rmul	ar C <sub>6</sub> H <sub>12</sub>	Weight 84.1	56	3		
		Ref.			Ref			Ref
F.P. °C	-134.430	2	dt/dP			f to		
F.P. 100%			*C/mm 25*C	0.1005	5	g <u>*K</u>		
B. P. °C 760 mm	56.30	2	BP	0.0406	5	h	ļ	<u> </u>
100 30	4.96 -17.60	4	t <sub>e</sub>	0.0361	5	f' to	j	
10	-34.69	5	30 mm ΔHm cal/g	0.5643	-	h'	1	i
1	-62.82	5	ΔHv cal/g	<del></del>	$\vdash$	m   300 to	-0.0000	
Pressure mm 25°C	243.	5	25°C	83.81	5	n 600 K	0.0015	
te	883.	5	30 mm BP	91.13 78.78	5 5		-0.0667	-
Density g/ml 20°C	0,6690	,	t <sub>e</sub> ,	78.01	5	m'   700 to n'   11000 °K	0.1236	
at 25	0.6642	2 2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	77.99	5	0'	-0.0 <sub>6</sub> 36	
	0.6593	4	d   -18 to	19.64 88.19	5	Surface tension		
a b	0.6884	4		0.1671	5	dynes/cm. 20°C	17.41 16.38	<b>5</b>
Ref. Index	3	†	d'   to		1	40	15.37	5
n <sub>D</sub> 20°C	1.3880 1.3849	2 2	d g/ml	0,234	5	Parachor [P]		
30	1.3820	4	vc ml/g tc °C	4. 265	5	20°C 30		ł
"C"	0.7746	4	P mm	217 <u>.</u> 22487.	5	40 Suad	257 2	5
MR (Obs.)	29.68	2	P <sub>c</sub> mm PV/RT	22401.	-	Exp. L. 1. %/wt.	257.3	-
MR (Calc.) (nD-d/2)	28.978 1.0535	5 4	25°C	0.9764	5	u.		_
Dielectric	1		30 mm BP	1.0000 0.9540	5	Dispersion	126.	2
A   -18 to		5	t <sub>e</sub>	0.9496 0.264	5 5	Flash Point °C Fire Point		}
B ∟94°C	1139.0	5	t <sub>c</sub> AHc kcal/m	897.84	2	M Spec.		
A*   -18 to	1,25914	5	ΔHf	071.04		Ultra V. X-Ray Dif.		
B* <u> </u>	1064.5	5	ΔFf	<u> </u>	$\vdash$	Infrared		
С	]		Viscosity centistokes			Solubility in + Acetone		
t <sub>x</sub> to	ł		າ ℃	-		Carbon tet.		
A' to	<del> </del>	-				Benzene Ether		
B' <u>°</u> C			B <sup>V</sup> l to	<u> </u>	$\vdash$	n-Heptane		
A'* to		-	A to			Ethanol Water		
B'* *C			(BV) to	-i		Water in		<u> </u>
Ac  94 to Bc t °C	7.3041	5	(A <sup>V</sup> ) °C					l
Bc Ltc_C	1427. 267.	5	c <sub>p</sub> liq. •K					
Cryos. A*			cp vap300°K	0.38108				
consts. B°		<b> </b>	] <del>4</del> 00	0.48125	2			
t <sub>e</sub> °C	61.01	5	c <sub>v</sub> vap.	L		L	<u> </u>	L
T <sub>R</sub> = 0.75	ES: 1-Dow	2 47	OT 2 124 4 4	Tala for		grams/100 gram		<u> </u>
SOURCE:		Al		Calc. from de	. dai	ta 5-Calc. by for	mula	
PURIFICAT	ION:	A.						
	RE REFERE							
			·•					

No. 25 4-Methyl-trans-2-pentene NAME STRUCTURAL FORMULA CH3CH CH=CH CH3 Molecular C6H12 Ċн<sub>3</sub> Mole Ref. Molecular Weight 84.156 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% -140.810 2 dt/dP f to °C/mm ٩K g 25°C 0.1083 5 B. P. °C h ВP 0.0409 4 760 mm 58.55 2 f 0.0361 5 to 100 6.91 4 g' °К 30 -15.78 4 30 mm 0.5677 5 10 -32.98 5 h! AHm cal/g 5 1 -61,28 0.0441 300 to m AHv cal/g Pressure n 600 °K 0.0014 25°C 84.88 5 mm 25°C 224. 5 o 4 -0.0663 30 mm 91.88 5 889. 5 t<sub>e</sub> BP 79.41 m' 700 to 0.1393 Density 4 78,60 5 n' g/ml 20°C 1000 °K 0.0011 te (d, e) 0.6686 2 5 78.58 ٥, -0.0636 4  $d_4^t$ 25 0.6638 2 AHv/Te 19.64 5 30 0.6590 4 Surface tension T-10 d 89.23 to 5 0.6879 4 dynes/cm. 20°C 17.37 \_7<u>0</u>\_ 0.1677 ъ -0.0392 4 ăΠ 30 16.35 5 to 40 15.35 5 Ref. Index e¹ °C 20°C 1.3889 [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 0.234 5 25 2 1.3859 20°C vc ml/g t °C 5 4.28 30 1.2831 4 30 t<sub>c</sub> 220. 5 40 "C" 0.7768 4 P<sub>c</sub> mm 22557. 5 5 Sugd. 257.3 MR (Obs.) 29.76 2 PV/RT Exp. L. l. %/wt. 29.441 MR (Calc.) 5 25°C 0.9814 (nD-d/2)1.0546 5 4 30 mm 2 1.0000 5 Dispersion 128. Dielectric BP 0.9540 Flash Point °C 0.9494 5 A -20 to 6.88584 Fire Point 0,264 5 В \_ <u>97 °C</u> 1147.1 5 M. Spec. Ultra V. C 227.9 5 ∆Hc kcal/m 896,84 ΔHf A\* -20 to 1.25965 5 X-Ray Dif. ΔFf B\*| 70 °C 1073.8 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C tk tx Carbon tet. °C Benzene A۱ to Ether В' °C n-Heptane B<sub>v</sub> | C' Ethanol °C Water A'\* to Water in B'\* °C (B<sup>V</sup>) to 97 to Acl (A<sup>V</sup>)| 7.3069 5 5 5 °C Βc 1437. t<sub>c</sub>°C cp liq. °K Cc 267. c<sub>p</sub> vap.300°K Cryos. A° 0.40330 2 consts. B° 0.49788 c<sub>v</sub> vap. te °C 63.52 5  $\overline{T_R}$  $= 0.75 T_{c}$ grams/100 grams solvent 2-API REFERENCES: 1-Dow 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 26	
NAME	2-Ethyl-	l -but	ene			STRUCTURAL I	ORMULA	
						CH (CH ) C	- CH	
						сн <sub>3</sub> (сн <sub>2</sub> ) с	- CH <sub>2</sub>	
Mole % Pur.		lecul		Molecular Weight 84.15	6	J.	2**5	
		Ref.			Ref	<u> </u>		Ref.
F.P. °C	-131.530	2	dt/dP	Ī		f to		
F.P. 100%			°C/mm		_	g K		
B, P. °C 760 mm	64,66	2	25°C BP	0.1328 0.0415	5 4	h		ļ
100	12.22	4	t <sub>e</sub>	0.0361	5	f' to		ĺ
30 10	-10.84 -28.32	4 5	30 mm	0.5769	5	g'   'K_		ĺ
ľi	-57.10	5	ΔHm cal/g			m   300 to	-0.0060	4
Pressure			ΔHv cal/g 25°C	87.83	5	n 600°K	0.0015	4
mm 25°C	177. 907.	5 5	30 mm	93.91	5	° ;	-0.0 <sub>6</sub> 67	4
Density	+	<del>                                     </del>	BP te	81.15 80.21	5	m'   700 to	0.1283	4
g/ml 20°C		2 2		80.19	5	n'   11000°K	0.0011 -0.0 <sub>6</sub> 36	4
dt 25 4 30	0.6847 0.6800	4	AHv/T <sub>e</sub>	19.65	5		- 650	Ĺ
a	0.7083	4	d   -5 to	92.08	5	Surface tension dynes/cm. 20°C	19.66	5
ь	-0.0391	4	-å, - 75 °C to	1	5	30 40	18.57 17.50	5
Ref. Index		2	e' i °C			Parachor [P]	17.50	<u>-</u> -
45	1.3941	2	d g/ml v ml/g	0.240 4.15	5	20°C		İ
30	1.3912	4	tc °C	232.	5	30 40		
MR (Obs.)	0.7680	4	P <sub>c</sub> mm	23699.	5		257.3	5
MR (Calc.		5	PV/RT 25°C	0.9844	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0522	2	30 mm	1.0000	5	u. Dispersion	128.	2
Dielectric			BP t <sub>e</sub>	0.9540 0.9488	5	Flash Point °C		
A -5 to B 106°C		5	tc	0.263	5	Fire Point		
c	226.7	5	AHc kcal/m	898.18	2	M Spec. Ultra V.		l
A*  -5 to   B*		5	ΔHf ΔFf			X-Ray Dif.		
к — — -		١	Viscosity			Infrared Solubility in +		-
's	<del>,</del> -		centistokes 7 °C			Acetone		
t <sub>x</sub> °C	7		'			Carbon tet. Benzene		
A' to						Ether		
c	<del>-</del>		B <sup>V</sup>   to			n-Heptane Ethanol		
A!* to			A <sup>V</sup> I C			Water		
B'* *(		-	(B <sup>V</sup> )  to	]		Water in		
Ac   106 to	7.3153 1466.	5	(A <sup>V</sup> )  °C		<u> </u>			
Cc —	267.	5	c <sub>p</sub> liq. °K					
Cryos, A° consts, B°			cp vap 300 °K	0.38120 0.48363				
t <sub>e</sub> °C	70,34	5	f 400 c, vap.	0.48363	۱'			
$T_{\mathbf{R}} = 0.7$		1 2	<u> </u>	L	<u> </u>	+ grams /100 s===	na aalaa	<u> </u>
	CES: 1-Dow	2-AI	PI 3-Lit. 4-(	Calc. from det	da	grams/100 grar ta 5-Calc, by for		<u> </u>
SOURCE:	<del></del>		PI					
PURIFICA	TION:	A	PI				·····	
LITERATU	RE REFERE	NCES	3:					
}								
L		_						

N4.145	2, 3-Dime	thvl-	l -butene		T	CARDICATION I	No. 27	
NAME	-,	,-				STRUCTURAL CH <sub>3</sub>	FORMUL.	A.
<u>-</u>						сн <sub>3</sub> сн с = с	·u	
Mole		lecul	ar C <sub>6</sub> H <sub>12</sub>	Molecular		сн <sub>3</sub> сн с - с	, <b>11</b> 2	
% Pur.	Fo	rmula		Weight 84.15				
	1 157 27	Ref.	1	<del> </del>	Ref.	<u> </u>		Ref
F.P. °C F.P. 100%	-157.27	2	dt/dP °C/mm		1 1	f to g °K		1
B. P. °C		$\vdash$	25°C	0.0985	5	g ' <u>*K</u>		ĺ
760 mm	55.67	2	BP	0.0406	4	<del></del>		_
100 30	4.41	4	t <sub>e</sub>	0.0361	5	g' to		
10	-35.17	5	30 mm	0.5055	<b>⊢</b> ″-H	h'		1
<u>l</u>	-63.25	5	ΔHm cal/g	<del> </del>	+ $-i$	m   300 to	0.0285	4
Pressure	1	_	ΔHv cal/g 25°C	84.74	5	n 600 °K	0.0015	4
mm 25°C t <sub>e</sub>	249. 881.	5 5	30 mm	90.92	5	0	-0.0675	4
Density	+	Ť	BP	78.61	5	m' 700 to	0.1113	4
g/ml 20°C	0.6779	2	t <sub>e</sub> t <sub>e</sub> (d, e)	77.85 77.84	5	n'   1000 °K	0.0012	4
t 25	0.6731	2	ΔHv/T	19.64	5	o'	-0.0642	4
	0.6682	4	d -15 to	87.90	5	Surface tension		
a b	0.6973 -0.0392	4	_e	0.1669	5	dynes/cm. 20°C	18.35	5
Ref. Index	+	÷	d' to e' °C			8 30 40	17.28 16.23	5
n <sub>D</sub> 20°C		2	<b></b>	0.330	+	Parachor [P]		
45	1.3874	2	d <sub>c</sub> g/ml	0.238 4.20	5	20°C		ĺ
30	1.3845	4	vc ml/g tc °C	217.	5	30 40		l
"C"	0.7689	4	P <sub>c</sub> mm	22833.	5	Sugd.	257.3	5
MR (Obs.) MR (Calc.		5	PV/RT		$\vdash$	Exp. L.1.%/wt.		
(nD-d/2)	1.0515	2	25°C 30 mm	0.9919	5	u.	120	2
Dielectric			BP	0.9540	5	Dispersion Flash Point C	129.	Ľ
A -15 to	6.88200	5	t t c	0.9497	5	Fire Point		ĺ
B   94 °C	1136.7	5 5		0.264	2	M. Spec.		
A* -15 to	1, 25939	5	ΔHc kcal/m ΔHf	896.32	'	Ultra V.		1
B*  80 ℃		5	ΔFf			X-Ray Dif. Infrared		ĺ
ĸ	-		Viscosity		1	Solubility in +		
t <sub>k</sub> – to	-		centistokes 7°C		1	Acetone		l
t <sub>x</sub> °C			,			Carbon tet. Benzene		
A'   to						Ether		
B'°C	-		B <sup>V</sup> to		T	n-Heptane Ethanol		
A¹* to	+	<del>                                     </del>	A I C			Water		
B¹* °C			(B <sup>V</sup> )  to			Water in		
Acl 94 to	7.3038	5	(A <sup>V</sup> )  °C					
Bc tc °C	1425. - 267.	5	c <sub>p</sub> liq. °K					
	201.	-	łı -	0.40034				ĺ
Cryos. A° consts. B°			c <sub>p</sub> vap,300°K 400	0.40924 0.50620	2 2			ı
t <sub>e</sub> °C	60.31	5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 Т <sub>с</sub>					grams/100 grai	ms solven	t
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		AJ	PI					
PURIFICAT	TION:	Al	PI					
LITERATU	RE REFERE	NCE	S:					

									No. 28	
NAME	3, 3-	Dime	thyl	-1-butene		_	STRU	JCTURAL I	FORMULA	1
								CH <sub>3</sub>		
	7.		,		Mal			CH <sub>3</sub> ¢ CH=	:СН <sub>2</sub>	
Mole % Pur.	Ref.		mul		Molecular Weight 84.1	56		Ċн <sub>3</sub>		
		_	Ref	r — —		Ref				Ref
F, P. *C	-115.20		2	dt/dP			f	1		$\vdash$
F.P. 100%	1		ᅴ	*C/mm		1 1	g	to °K		
B. P. *C			$\neg$	25°C BP	0.0619 0.0392	5 4	h			1
760 mm 100	41.24		2 4	t <sub>e</sub>	0.0363	5	f'	to		
30	-29.96		4	30 mm	0.5424	5	g'	•K_		
10 1	-46.38		5	ΔHm cal/g	1		h'			
Pressure	-73.36	<del>'</del> +	5	ΔHv cal/g		$\vdash$	m (	300 to	0.0071	4
mm 25°C	425.9		5	25°C	77.42	5	n	_600°K_	0.0013 -0.0 <sub>6</sub> 42	
te	841.0		5	30 mm BP	85.87 74.36	5	1			├
Density				l t_	73.87	5	m'	700 to	-0.0237 0.0014	4
g/ml 20°C	0.65		2 2	t <sub>e</sub> (d, e)	73.86	5	0'	1_1000	-0.0659	
d <sub>4</sub> 25 30	0.64		4	AHv/T <sub>e</sub>	19.58	5	Surfa	ce tension		-
	0.67	32	4	d   -30 to		5 5		/cm. 20°C	15.72	5
_ь	-0.03	93	4		-		3	30 40	14.71	5
Ref. Index	1.37	60	2	e' i °C	;	Ш	Para		13.71	-
<sup>n</sup> D 20°C	1.37		2	d g/ml	0.230	5 5	Para	20°C		
30	1.36	99	4	tc *C	4.348 192.	5		30		
"C"	0.77	05	4	P <sub>c</sub> mm	21009.	5	ļ	40 Sugd.	257.3	5
MR (Obs.) MR (Calc.)	29.58		2	PV/RT	<del> </del>	$\vdash$	Exp.	L.1.%/wt.		-
(nD-d/2)	29.44		5 2	25°C	0.9752	5	_	u.		
Dielectric	T		$\neg$	30 mm BP	1.0000 0.9550	5 5		rsion	124.	2
A   -30 to		763	5	te	0.9521	5	Fire	Point °C		
B <u> </u>	108 <b>0</b> .6	- 1	5 5	tc	0.265	5	M Spe			
C Atl son	+	1/2		ΔHc kcal/m ΔHf	896.85	2	Ultra	v.		
A* -30 to B* 54°C	1.24	101	5	ΔFf			Infra:	y Dif. red		
K		l	_	Viscosity				ility in +		-
\$k   −	-	- 1	- 1	centistokes 7°C	ł	1 1	Acet	one		
t <sub>x</sub> C		- 1		'		1 1	Carl Ben	on tet.		
A'   to			$\neg \neg$		1		Ethe	r		
B', ∟ _ <u>•</u> ⊆	· <b>{</b>		1	B <sup>V</sup>   to	<del> </del>	$\vdash$	n-He Etha	eptane		
A'* to	<del> </del>	$\rightarrow$	-	AV C		1 1	Wate			
B'* *C	ł	- 1	- 1	(BV) to	1		Wate	er in		
Ac   76 to	7,27	04	5	(A <sup>V</sup> ) °C	1					
Bc tc_C	1356. 268.		5	cp liq. °K	<del></del>					
Cryos. A*	† <del></del>	-	-			2				
consts. B°				c <sub>p</sub> vap.300°K 400	0.36112 0.46224					
t <sub>e</sub> °C	44.29		5	c <sub>v</sub> vap.						
$T_{R} = 0.79$							+ grai	ms/100 gran	ns solven	t
REFERENC	ES: 1-D	ow 2	2-AF	PI 3-Lit. 4-0	Calc. from det	t. dat	ta 5-C	alc. by form	nula	
SOURCE:			A	PI						
PURIFICAT				PI						
LITERATU	RE REF	EREN	CES	:						

			·				No. 29
NAME	2,3-Dime	thyl-	2-butene				L FORMULA
						c •	<sup>:</sup> H <sub>3</sub>
Mole	Ref. Mo	ecul	AT	Molecular		сн <sub>3</sub> с = с	- СН <sub>.3</sub>
% Pur.	For	mul		Weight 84.15	6	Ċн <sub>3</sub>	
	1	Ref.		1	Ref.		Re
F.P. °C F.P. 100%	-74.280	2	dt/dP *C/mm	0.155(	_		to °K
B.P. °C 760 mm	73.21	2	25°C BP	0.1776 0.0424	5 4	h	
100	19.63	4	t <sub>e</sub>	0.0361	5		to °K
30 10	-3.95 -21.82	4 5	30 mm	0.5899	5_	h'	4
1	-51.26	5	ΔHm cal/g	+	<del>├</del> ─	m   300	to 0.0464 4
Pressure mm 25°C	127.	5	ΔHv cal/g 25°C	90.14	5	<u> </u>	°K 0.0011 4
t <sub>e</sub>	931.	5	30 mm BP	96.73 83.55	5		
Density	0.7090	,	t_	82.44	5	m' 700 n' 1000	
g/ml 20°C dt 25 4 30	0.7080 0.7034	2	te (d, e)	82.41	5	0'	-0.0 <sub>6</sub> 48 4
	0.6988	4	ΔHv/T <sub>e</sub>	19.65 96. <b>0</b> 6	5	Surface tension	
a b	0.7264 -0.0 <sub>3</sub> 90	4	e   90 °C	0.1708	5	dynes/cm. 20	
Ref. Index	1		d' to °C			40	19.63 5
<sup>n</sup> D 20°C	1.4122	2 2	d <sub>c</sub> g/ml	0.246	5	Parachor [P]	
30	1.4065	4	V mi/g	4.06 248.	5	30	
"C"	0.7750	4	t <sub>c</sub> °C P <sub>c</sub> mm	24865.	5	40 Su	gd. 257.3 5
MR (Obs.) MR (Calc.)	29. 59 29. 441	2 5	PV/RT			Exp. L.1.%/v	
(nD-d/2)	1.0582	2	25°C 30 mm	0.9725 1.0000	5	u. Dispersion	132. 2
Dielectric			BP	0.9540	5	Flash Point 6	
A -10 to B   118 °C	6. 90 <b>3</b> 93	5	te tc	0.9481 0.264	5	Fire Point	
с — — —	225.	5	AHc kcal/m	895.19	2	M. Spec. Ultra V.	
A* -10 to B* 90 °C	1.25993	5 5	ΔHf ΔFf			X-Ray Dif.	
к — = =			Viscosity			Infrared Solubility in	+
t <sub>k</sub>			centistokes η °C		'	Acetone Carbon tet.	
t <sub>x  </sub> °C	-	_				Benzene Ether	
B'°C			B <sub>v</sub> to		<u> </u>	n-Heptane	
A¹* to	<del> </del>		B to C			Ethanol Water	
B'* °C			(B <sup>V</sup> )  to	-]		Water in	
Ac 118 to Bc t <sub>c</sub> °C	7.3249 1506,	5	(A <sup>V</sup> )  °C	<u> </u>	<u> </u>		
Bc∟t <sub>c</sub> °C Cc	266.	5	c <sub>p</sub> liq. °K				
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.36397 0.45867	2 2		
t <sub>e</sub> °C	79.91	5	c <sub>v</sub> vap.				
$T_{\mathbf{R}} = 0.75$						·	grams solvent
	ES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by	formula
SOURCE:	TOM		PI	· · · · · · · · · · · · · · · · · · ·			
PURIFICAT	TON: RE REFERE		PI 				
LI LERA I U	NE REFERE	AC ES	<b>)</b> ;				

NAME	1-H	ept <b>e</b> n							
Ì		cpten					STRUCTURAL	FORMULA	4
Mole % Pur.	Ref.	Mo	lecul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	2	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>4</sub> (	:H=CH <sub>2</sub>	
			Ref.	<del></del>		Ref			Ref
F. P. *C	-119.0	29	2	dt/dP			4   1		T
F.P. 1007			H	*C/mm	ŀ		f to	1	ļ
B. P. *C	<del>†</del>		$\vdash$	25°C	0.3655	5	h .	]	Į
760 mm	93.6		2	BP	0.04447 0.03624	4 5	f' to	+	+-
100 30	37.4 12.6		2 2	t <sub>e</sub> 30 mm	1	5	g'   'K		1
10	-6.0		2		0.6189	۲	h' i	1	l
1	- <b>3</b> 6.9	5	5	ΔHm cal/g	ļ	<u> </u>	m   300 to	0,0251	4
Pressure	-/-	_		ΔHv cal/g 25°C	87.79	5	n 600 °K	0.0013	4
mm 25°C	56.3 987.5	5	5	30 mm	89.09	5	0	-0.0 <sub>6</sub> 48	4
Density	+ /313		┵┤	BP	76.31	5	m'   700 to	0,1118	4
Density g/ml 20°C	0.6	9698	2	te t_ (d, e)	74.93 74.87	5	n'  1000 °K	0.0011	4
at 25	0.6	9267	2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>		5	o'	-0.0 <sub>6</sub> 37	4
<sup>4</sup> 4 30	0.6	8829	4		19.57		Surface tension		T
•		1432	4	d   10 to		5	dynes/cm. 20°C	19.52	5
b	-0.0	3855	4	_d'to	1		30 40	18.55 17.60	5
Ref. Index		9980	2	e' i *C				17.00	۲-
n <sub>D</sub> 20°C		9713	2	d g/ml vc ml/g	0.235	5	Parachor [P] 20°C		1
30	1.3	9451	4	t <sub>c</sub> mi/g	4. 247 262.	5 5	30		l
"C"	0.7	648	4	P mm	1	5	40 Suad	206.2	5
MR (Obs.)	34.1	35	2	P <sub>c</sub> mm PV/RT	20813.	1		. 296. 2	-
MR (Calc.			5	25°C	0.9966	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0	5131	2	30 mm	1,0000	5	Dispersion	118.2	2
Dielectric	<u> </u>		Ш	BP	0.9520 0.9441	5	Flash Point °C		
A 10 to		0069	2	te t <sub>c</sub>	0.260	5	Fire Point		
B 1_128_*C	219.1		2 2	ΔHc kcal/m	1048,05	2	M Spec.		
A* 10 to		0572	5	ΔHf			Ultra V. X-Ray Dif.	1	
B* 113 °C			5	ΔFf	ļ	Ш	Infrared	1	
c c				Viscosity	ļ		Solubility in +	†	
i <sub>k</sub>	-1			centistokes   7 0°C	0.44	2	Acetone		
t⊈i °C	7			20	0.35	2	Carbon tet. Benzene		1
A' to			$\Box$	40 80	0.29 0.22	2 2	Ether	1	1
B' •	4				+ ··	┝╧┤	n-Heptane		
A'* to	+		$\vdash\vdash$	B to A C	j		Ethanol Water		
B'* °(				(B <sup>V</sup> ) to	1		Water in		L
Ac   128 to	7.3	119	5	(A <sup>V</sup> ) °C	ļ				
Bc t *C	1560.		5		<del> </del>	$\vdash \vdash \vdash$	1		
Cc — —	259.		5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°				c <sub>p</sub> vap.300°K 400	0.37960 0.47840	2			
t <sub>e</sub> °C	102.7	5	5	c <sub>v</sub> vap.					L
$T_R = 0.7$							† grams/100 gra	ms solven	t
REFEREN	CES: 1-D	)ow	2-AF	PI 3-Lit. 4-(	Calc. from det	. dat	ta 5-Calc, by for		
SOURCE:			AI	PI					
PURIFICA'	rion:		AF	PI					
LITERATU	RE REF	ERE	CES	:					

<del></del>							No. 31	l
NAME	cis-2-He	ptene	<b>)</b>			STRUCTURAL	FORMUL	.A
	<del></del>					CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH	=СНСН.	
Mole % Pur.	Ref. Mo	ecul		Molecular Veight 98.18	2	=		
	1	Ref.			Ref.		т	Ref.
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0.4366	5	f to to° <u>K</u>		
B. P. °C 760 mm 100 <b>3</b> 0	98.5 41.62 16.53	2 4 4	BP t <sub>e</sub> 30 mm	0.0449 0.0362 0.6280	4 5 5	f' to		
10 1	-2.50 -33.90	5 5	ΔHm cal/g	1,0200	+ -	h'		
Pressure mm 25°C t <sub>e</sub>	45.64 1001.	5	ΔHv cal/g 25°C 30 mm	90.00 90.19	5	m to		
Density g/ml 20°C dt 25 d4 30	0.708 0.704 0.700	2 2 4	BP t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	77.53 76.09 76.04 19.59	5 5 5	m' to		
a b	0.724 -0.0 <sub>3</sub> 8	4 4	d 12 to e 115 °C d' to	92.74 0.1544	5 5	Surface tension dynes/cm. 20°C 30 40	20.79 19.84	5 5 5
Ref. Index <sup>n</sup> D 20°C 25 30	1.406 1.403 1.401	2 2 4	e'   °C dcg/ml vcml/g tc°C	0.247 4.054 272.	5 5 5	Parachor [P] 20°C 30	18.91	-
"C"	0.7640	4	P <sub>c</sub> mm	21783.	5	40 Sugd.	296.2	5
MR (Obs.) MR (Calc.) (nD-d/2)	34.0 34.059 1.052	2 5 2	PV/RT 25°C 30 mm	0.9976 1.0000	5	Exp. L.1.%/wt. u. Dispersion	122.	2
Dielectric  A 12 to B 135 °C	6.93647 1292.9	5	BP t e t c	0.9515 0.9432 0.255	5 5 5	Flash Point C Fire Point M. Spec.		
C A* 12 to B* 118 °C	220. 1.33477 1213.1	5 5 5	ΔHc kcal/m ΔHf ΔFf	1046.45	2	Ultra V. X-Ray Dif. Infrared		
K			Viscosity centistokes n °C			Solubility in + Acetone Carbon tet, Benzene Ether		
B'   _ °C C' A'* to			B <sup>V</sup> to			n-Heptane Ethanol Water		
B'* °C  Ac 135 to  Bc tc °C  Cc	7.3513 1605. 261.	5 5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C c <sub>p</sub> liq. °K			Water in		
Cryos, A°	201.	3	c <sub>p</sub> vap. °K					
te °C	108 19	5	c <sub>v</sub> vap.					
$\frac{\mathbf{T_e} \cdot \mathbf{C}}{\mathbf{T_R} = 0.75}$	108.18 T <sub>C</sub>		L v -	L	Ll	† grams/100 gra	me solva-	<u></u>
REFERENC		2-A	PI 3-Lit. 4-0	Calc. from de	t. da			
SOURCE:		Al						
PURIFICAT		Al						
LITERATUR	RE REFERE	NCES	<b>5:</b>					

						No. 32						
NAME	trans-2-	Hepte	$\Box$	STRUCTURAL FORMULA								
			T.		$\neg$	$CH_3(CH_2)_3CH=CHCH_3$						
Mole % Pur.		lecul		Molecular Weight 98.18	2							
		Ref.			Ref	Re						
F.P. °C	-109.480	2	dt/dP			f to						
F.P. 1009			°C/mm	0 4354	ا ۔ ا	g <u>*K</u>						
B. P. *C 760 mm	07.05		25°C BP	0.4274 0.0449	5 4	h ,						
100	97.95 41.12	2 4	t.	0.0362	5	f' to						
30 10	16.06 -2.96	4 5	30 mm	0.6273	5	g'   ' <u>°</u> K_						
1	-34.31	5	∆Hm cal/g		Ш	h <sup>1</sup>						
Pressure		<del>                                     </del>	ΔHv cal/g	00.10	ا ۽ ا	m to r						
mm 25°C	46.55 999.3	5	25°C 30 mm	90.18	5 5	•  =-						
t <sub>e</sub> Density	799.3	3	BP	77.35	5	m' to						
g/ml 20°C	0.7012	2	te (d, e)	75.91 75.87	5 5	n'  •K_						
dt 25 4 30	0.6969	2	ΔHv/T	19.57	5	0'						
	0.6926	4	d   16 to	<del></del>	5	Surface tension						
a b	0.7183 -0.0 <sub>3</sub> 85	4	<u>•   108 °C</u>	0.1543	5	dynes/cm. 20°C   20.00   5						
Ref. Index			d'   to			40 18.08						
<sup>n</sup> D 20°C	1.4045 1.4020	2 2	d <sub>c</sub> g/ml	0, 243	5	Parachor [P]						
30	1.3992	4	A	4,120	5	20°C						
"C"	0.7687	4	, ~	269.	5	40						
MR (Obs.)	34.28	2	P <sub>c</sub> mm	21316.	5	Sugd. 296. 2						
MR (Calc. (nD-d/2)	) 34.059 1.0539	5 2	25°C	0.9972	5	Exp. L.1.%/wt.						
Dielectric	<del></del>	-	30 mm	1.0000	5	Dispersion 124.						
A 16 to	<del></del>	5	BP t <sub>e</sub>	0.9515 0.9433	5 5	Flash Point °C						
B 134 °C		5	tc	0, 255	5	Fire Point						
С	220.	5	ΔHc kcal/m	1045.45	2	M Spec. Ultra V.						
A*   16 to B* 118 °C		5	ΔHf ΔFf	ļ		X-Ray Dif.						
K LIO	- 1210.3	1	Viscosity		$\vdash$	Infrared Solubility in +						
t - to	_		centistokes 7 °C			Solubility in +						
5k   to		ĺ	7 ℃			Carbon tet.						
A' to		$\vdash$				Benzene Ether						
B' '	긱	ł	B <sup>V</sup>   to	<del> </del>	$\vdash$	n-Heptane						
A'* to	<del>,  </del>	<del> </del>	A C			Ethanol Water						
B'* *(		1	(BV) to	1		Water in						
Ac   134 to	7.3482	5	(A <sup>V</sup> ) °C									
Bc tc °C	261.	5	c <sub>p</sub> liq. °K	1	П							
Cryos, A		Ť	c <sub>p</sub> vap. °K									
consts. B			11 -									
t <sub>e</sub> °C	107.57	5	c <sub>v</sub> vap.									
$T_R = 0.7$						grams/100 grams solvent						
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from de	t. dat	ta 5-Calc. by formula						
SOURCE:			PI									
PURIFICATION: API												
LITERATURE REFERENCES:												

No. 33 cis-3-Heptene NAME STRUCTURAL FORMULA CH3CH2CH=CH(CH2)2CH3 Mole Ref. Molecular Molecular Weight 98,182 C7H14 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g °K 25°C 0.3945 5 B. P. °C h ВP 0.0447 4 760 mm 95.75 2 t<sub>e</sub> 0.0362 5 ft to 100 39.21 4 <u>°К</u> g' 30 14.29 4 30 mm 0.6239 5 10 -4,62 5 h! ∆Hm cal/g 5 1 -35.81 to m AHv cal/g Pressure ۰ĸ n 25°C 87.77 5 mm 25°C 52.36 5 0 30 mm 89.36 5 993.0 te 5 BP 5 76.83 m' to Density te (d, e) 75.44 5 n' g/ml 20°C °K 0.7030 2 75.39 5 ۰,  $d_4^t$ 25 0.6987 2 AHv/Te 19.58 5 30 0.6944 4 Surface tension 14 91.56 d 5 a 0.7202 4 20,21 dynes/cm, 20°C 105 °C 0.1539 5 ь -0.0385 4 19.22 5 30 ă٦ 5 40 18.26 Ref. Index e١ °C 20°C 1.4059  $^{n}D$ [P] Parachor d<sub>c</sub> g/ml 0.244 5 25 1.4033 2 20°C vc ml/g 4.096 5 30 4 1.4006 30 <sup>t</sup>c 5 266. 40 "C" 0.7692 4 P<sub>c</sub> mm 21322. 5 Sugd. 296.2 5 MR (Obs.) 34.30 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9983 5 (nD-d/2) 1.0544 2 30 mm 1.0000 5 Dispersion 122. 2 Dielectric BP 0.9515 5 Flash Point C 0.9435 5 A 14 to 6.93161 1282.3 Fire Point 0.255 B 1 131 °C 5 M. Spec. Ultra V. C 221. 5 AHc kcal/m 1046.45 ΔHf A\* 14 to 1.33298 5 X-Ray Dif. ΔFf B\*[ 115°C 1203.1 Infrared ĸ Viscosity Solubility in centistokes Acetone  $\mathbf{t}_{\mathbf{k}}$ to Carbon tet. ٠c t^ Benzene A١ to Ether B' °C n-Heptane B<sub>v</sub> | C Ethanol °C Water A'\* to Water in °C B'\* (B<sup>V</sup>) to Ac 131 to 7.3464 5 (A<sup>V</sup>) °C 1591. 5 Bc\_tc\_° °C cp liq. °K Cc 261. 5 c<sub>p</sub> vap. Cryos. A °K consts. B° c vap. t<sub>e</sub> °C 105.09 5 TR  $= 0.75 \, T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 34						
NAME	trans-3-	Hept	ene		STRUCTURAL FORMULA								
Mole	D-4 14-	11		Malasulas		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH=C	HCH <sub>2</sub> CH <sub>3</sub>	3					
% Pur.		lecul rmul		Molecular Weight 98.18	82								
		Ref.			Ref.			Ref.					
F.P. °C	-136.63	2	dt/dP			f to							
F. P. 1009	•	<u> </u>	*C/mm 25*C	0.3933	5	g <u>°K</u> _							
B. P. °C 760 mm	95.67	2	BP	0.0447	4	h -							
100 30	39.13 14.04	4	t <sub>e</sub>	0.0362	5	f' to							
10	-4.70	5	30 mm	0.6239	-	h'							
11	-35.88	5	ΔHm cal/g ΔHv cal/g	<del>                                     </del>	-	m to		<u> </u>					
Pressure mm 25°C	52,50	5	25°C	83.77	5	n <u>*K</u> -							
te	993.	5	30 mm BP	89. 22 76. 79	5	<u> </u>		<u> </u>					
Density g/ml 20°6	0 (00:		t_	75.40	5	m' to		l					
dt 25	0.6938	2 2	te (d, e) ΔHv/Te	75.37	5	0'							
	0.6895	4	d   10 to	19.57 91.36	5	Surface tension							
a b	0.7152 -0.0 <sub>3</sub> 85	4 4	110°C	0.1523	5	dynes/cm. 20°C	19.65 18.68	5					
Ref. Index		Ť	d'   to			40	17.79	5					
n <sub>D</sub> 20°0	1.4043 1.4017	2 2	<del></del>			Parachor [P]							
30	1.3989	4	d g/ml vc ml/g tc °C	2//	_	20°C 30							
"C"	0.7717	4	tc °C P <sub>c</sub> mm	266. 21116.	5	40 Sugd.	296.2	5					
MR (Obs.		2	PV/RT	21110.	<u> </u>	Exp. L.1.%/wt.	270.2	-					
MR (Calc. (nD-d/2)	34.059 1.0553	5 2	25°C	0.9984	5	u.		_					
Dielectric			30 mm BP	1.0000 0.9515	5	Dispersion Flash Point °C	124.	2					
A 10 t		5	t <sub>e</sub>	0.9435 0.255	5	Fire Point							
B 1_131*	C 1281.8 221.	5	t <sub>c</sub> AHc kcal/m	1045.45	2	M Spec.							
A* 10 to	1.32366	5	ΔHf		_	Ultra V. X-Ray Dif.							
B* L115 °	<u>C</u> 1200.5	5	ΔFf Viscosity	<del> </del>	-	Infrared							
c		1	centistokes			Solubility in + Acetone							
t <sub>x</sub>   -t			η •c			Carbon tet.							
A' t	<del>-</del>	$\vdash$				Benzene Ether							
B' 1	드		B <sup>V</sup>   to		-	n-Heptane							
A'* t		╁	Av i c	1	į	Ethanol Water							
B'* •			(B <sup>V</sup> ) to	]		Water in							
Ac   131 to		5 5	(A <sup>v</sup> ) i ∘c										
Bc tc_	261.	5	c <sub>p</sub> liq. °K										
Cryos. A			c <sub>p</sub> wap. °K										
consts. B		<del>  _</del>	c, vap.										
t <sub>e</sub> °C	105.03	5	T 4 F.	L		+							
T <sub>R</sub> = 0.	CES: 1-Dow	2-41	PI 3-Lit. 4-(	alo from de		grams/100 gran		<u> </u>					
SOURCE:			PI 3-Lit. 4-0	e.c. from det	. aa	ta 5-Calc. by for	BIDIT						
PURIFICA	TION:		PI			<del> </del>							
LITERATURE REFERENCES:													
L													

No. 35 2-Methyl-1-hexene STRUCTURAL FORMULA NAME  $CH_3(CH_2)_3C = CH_2$ Molecular C7H14 Mole Ref. Molecular Weight 98.182 % Pur. Ref. Ref. Ref. F.P. °C F.P. 100% -102.840 2 dt/dP f to °C/mm 25°C g °K 0.3442 B. P. ℃ h ВP 0.0443 4 760 mm 92.00 2 t<sub>e</sub> 0.03624 5 ſ١ to 100 35.93 11.22 4 g' <u>°K</u> 30 5 4 30 mm 0.6184 10 -7.52 5 h' ∆Hm cal/g -38.42 5 m to AHv cal/g Pressure n °K 25°C 89.66 5 mm 25°C 60.48 o 30 mm 88.25 5 982.3 5 te BP 75.88 5 m to Density 74.56 5 te te (d, e) °K g/ml 20°C 0.7030 2 74.52 o' 25 0,6986  $d_4^t$ 2 AHV/T 5 19.57 30 0.6942 4 Surface tension to 89.97 5 0.7206 a dynes/cm. 20°C 20, 20 5 1110 0.1531 <u>°C</u> -0.0387 4 ь 19.19 5 30 ăח to 18,20 5 40 Ref. Index e' °C <sup>n</sup>D 1.4034 [P] 20°C 2 Parachor d<sub>c</sub> g/ml 25 1.4007 2 20°C vc ml/g t °C 30 4 1.3979 30 t<sub>c</sub> 261. 5 40 "C" 0.7647 4 P<sub>c</sub> mm 21115. 5 Sugd. 296.2 5 MR (Obs.) 34.11 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9962 5 (nD-d/2)1.0520 2 30 mm 1.0000 2 5 Dispersion 125. Dielectric ВP 0.9515 0.9438 5 Flash Point C A 5 to B 127 °C t<sub>e</sub> 6.92518 5 Fire Point 5 0.255 1268.0 M. Spec. 221.52 5 AHc kcal/m 1044.44 2 Ultra V. ΔHf A\* 5 to 1.33076 5 X-Ray Dif. ΔFf B\*|110 °C 1189.4 Infrared ĸ Viscosity Solubility in centistokes Acetone to η Carbon tet.  $\mathbf{t_{\underline{x}}}$ °C Benzene A' to Ether B' °C n-Heptane Bv | Av | to Ethanol °C A'\* Water to Water in B'\* °C (B<sup>V</sup>) to Ac | 127 to 7.3403 5 (A V) °C  $\underline{\mathbf{B}}\mathbf{c} \, \underline{\hspace{1em}} \, \mathbf{t}_{\underline{\mathbf{c}}}$ ۰c 1574. cp liq. ۰ĸ Cc 5 261. Cryos. Aº °K c<sub>p</sub> vap. consts. B° c<sub>v</sub> vap. t<sub>e</sub> °C 100.88  $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc, by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

NAME		3-Me	thyl.	-1-he	xene						STR	UCTU	RAL :	FORM	ULA	
							Т				C	:н <sub>3</sub> (сн	2)2CH	CH=C	<sup>:H</sup> 2	
Mole % Pur.		Ref.		lecul: rmul:		C7H14		Molecul Weight		82			0	3		
				Ref.	Ī				,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Ref						Ref
F.P. °C	$\top$			-	dt/	'AD		Γ			f	IT	to			
F.P. 100	-				•	/mm		1			g	! ∟	_ <u>*K</u>			ļ
B. P. *C	1				25 B	S°C			2587 0435	5 4	h					
760 mm 100		84. 28.99		2 4	t <sub>e</sub>				0362	5	f'	1	to			
30		4.76		4		mm		0.	6063	5	g'	'_	_ <u>_</u> K_			
10 1		13.61 43.90		5 5	ΔH	m cal	/g			$\Box$	h'	<u> </u>				L
Pressure	┿	73.70		۲	ΔН	v cal/	g	†			m	(	to	ļ		
mm 25°C	1 :	83.37		5		°C	•	83.		5	n o	! <b>'</b> -	_ <u>.</u> K	ł		1
t <sub>e</sub>	9	60.		5	B	) mm P		85. 73.		5		<u> </u>				<u> </u>
Density	.T	0 / 0	-		te			72.	82	5	m'	! !	to °K	ľ		
g/ml 20°0	1	0.69 0.69		2 2	1 t <sub>a</sub>	(d, e)		72.		5	0,	i .–	,-	1		
d 25 4 30		0.68		4		Hv/Te		19.		5	S	ace ter	eion.	<del> </del>		<del> </del>
	1	0.71		4	d		5 to		69 1512	5		s/cm.		19.2	9	5
ь		-0.03	77	4	-å,-	$\frac{1}{1} - \frac{92}{1}$	_ to	1	1314	3	*		30 40	18.4 17.5		5
Ref. Index		1.39	7	2	e'	<u>i                                     </u>	•c	<u> </u>			D			17.5	-	
<sup>n</sup> D 25	1	1.39		2	d <sub>c</sub>	g/ml			241	5	Par	achor	[P] 20°C			
30		1.39	2	4	t <sub>c</sub>	g/ml ml/g *C		249.	142	5			30			1
"C"		0.76	19	4	-	mm		20818.		5			40 Sugd.	296. 2	:	5
MR (Obs. MR (Calc.		34.0 34.05	^	2 5		/RT				$\vdash$	Exp	L.1.9				<del>-</del>
(nD-d/2)	"	1.04		2	25	°C			9956	5	_	u.				
Dielectric	1			$\Box$	30 B1	mm P			0000 9520	5		ersion		120.		2
A 5 t	•	6.91	598	4	te			0.	9450	5		h Poin Point	t °C			
B [118°		39.0		4	t <sub>c</sub>				260	5	M S				-	$\vdash$
C	_	23.04		4	ΔH	c kcal, f	/m	1046.	98	2	Ultr	a V.				
A*   5 to	2 111	1.32 61.66		5	ΔF					1 1		ay Dif. ared				
K	-					cosity					<u> </u>	bility i	n +	-		├─
c	-					tistok	°C				Ace	etone	-			
t <sub>x</sub>					7		·	ļ		1 1		rbon te nzene	t.			ĺ
A'   t				$\Box$							Eth					
B' '	의			l î	В <sup>V</sup>	T	to	<del> </del>		+		leptane	:			
A'* t	+-			$\vdash$	AV	i	.c				Wa	anol ter				
B'*					(B <sup>V</sup>		to	-				ter in				<u> </u>
Ac   118 t	>	7.33	18	5	(A <sup>V</sup>	) i	°C									
Bc tc	15	39. 62.		5 5	Сp	liq.	°K			$\Box$						
Cryos. A	_	·		$\dashv$	1		•ĸ									
consts. B					, F	vap.	r									
t <sub>e</sub> °C		91.92		5	c^	vap.										
$T_R = 0$ .												ms/10	0 gran	ns sol	vent	t
REFEREN	CES:	1 - De	w	2-AF		-Lit.	4-0	Calc. fr	om de	t. dat	a 5-	Calc.	by for	mula		
SOURCE:				AI												
PURIFICA				AF												
LITERATU	RE	REFE	ERE	NCES	:											

							No. 37	
NAME	4-Methyl	-1-h	exene			STRUCTURAL	FORMULA	4
			_			сн <sub>3</sub> сн <sub>2</sub> сн сн	-CH=CH-	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 98.18	32	³ <sup>2</sup> сн <sub>3</sub>	2 2	
		Ref.			Ref.		]	Ref
F.P. °C F.P. 100%	-141.45	2	dt/dP °C/mm			f to		
B. P. °C 760 mm	86.73	2 4	25°C BP t <sub>e</sub>	0. 2851 0. 0437 0. 0362	5 4 5	h   ft   to		
1 <b>0</b> 0 <b>3</b> 0	31.38 6.99	4	30 mm	0.6103	5	g'° <u>K</u>		
10 1	-11.51 -42.00	5 <b>5</b>	ΔHm cal/g		† –	h'		
Pressure mm 25°C	74.69	5	ΔHv cal/g 25°C 30 mm	84.06 86.78	5	m to		
t <sub>e</sub>	967.8	5	BP	74.68	5	m' to	<del>                                     </del>	
Density g/ml 20°C dt 25 d4 30	0.6985 0.6942	2 2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	73.46 73.43	5	n'  *K_		
	0.6899	4	d 7 to	19.59 87.84	5	Surface tension		
a b	0.7156 -0.0 <sub>3</sub> 84	4	_e   95 °C d' to		5	dynes/cm. 20°C 8 30 40	19.68 18.71 17.76	5 5 5
Ref. Index n <sub>D</sub> 20°C	1.4000	2	d <sub>c</sub> g/ml	0, 241	5	Parachor [P]		
25 30	1.397 <b>3</b> 1.3947	2	v_mi/g	4, 152	5	20°C 30		
"C"	0, 7635	4	t <sub>c</sub> °C	252. 2 <b>0</b> 890.	5	40	296.2	5
MR (Obs.) MR (Calc.)	34.08 34.059	2 5	PV/RT 25°C	0.9945	5	Exp. L.1.%/wt.		
(nD-d/2) Dielectric	1.0508	2	30 mm	1.0000	5	Dispersion	120.	2
A 7 to	6.92084	4	BP t <sub>e</sub>	0.9520 0.9448	5	Flash Point C Fire Point		
B  _121 °C_ C	1249.38 222.52	4	t <sub>c</sub> ΔHc kcal/m	0.260	5	M. Spec. Ultra V.		
A* 7 to B* 105 °C	1.33154 1171.6	5 5	ΔHf ΔFf			X-Ray Dif. Infrared		
c t,to			Viscosity centistokes n °C			Solubility in + Acetone		
t <sub>k</sub> to t <sub>x</sub> °C			<i>"</i> 7 °C			Carbon tet. Benzene		
B'°C			B <sub>v</sub> to	<u> </u>		Ether n-Heptane		
A'* to B'* °C			$\begin{bmatrix} \mathbf{B}^{v} & to \\ \mathbf{A}^{v} & l \\ \mathbf{B}^{v} l \end{bmatrix} = \frac{C}{to}$	-		Ethanol Water Water in		
Ac 121 to Bc tc °C	7.3364 1551,	5	(A <sup>V</sup> )  °C					_
Cc — — —	262.	5	c <sub>p</sub> liq. °K					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C T <sub>R</sub> = 0.75	94.98 T-	5	L		<u> </u>	+ ~~~~ (100		
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	† grams/100 gra		<u>.                                    </u>
SOURCE:		AF		u	,,			
PURIFICAT	ION:	AF						
LITERATUE	RE REFERE	NCES	S:					

							No. 38	
NAME	5-Methy	l-1-h	exen <b>e</b>			STRUCTURAL F	ORMULA	
Γ						כם כם (כם / כ	и-си	
						сн <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub> с сн <sub>3</sub>	л-сп <sub>2</sub>	
Mole % Pur.	Ref. Me	olecul rmul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98,18	82	3		
		Ref.		•	Ref	<u> </u>	P	Ref.
F.P. ℃	1		dt/dP	1	1	f to		
F.P. 100%			°C/mm			g°K_		
B. P. °C	05.01		25°C BP	0.2710 0.0436	5 4	h '		
760 mm 100	85.31 30.13	2 4	t.	0.0362	5	f¹ to		
30	5.82	4	30 mm	0.6083	5	g'   'K_	}	
10 1	-12.61 -42.99	5	∆Hm cal/g			h'		
Pressure	<u> </u>	$\vdash$	ΔHv cal/g	00.45	ا ـ ا	m to to		
mm 25°C	79.11	5	25°C 30 mm	83.45 86.35	5 5			
t <sub>e</sub>	963.7	5	BP	74.31	5	m'   to		
Density g/ml 20°C	0.6920	2	te te (d, e)	73.11	5	n'  °K_		
at 25	0.6876	2	ΔHv/T	19.58	5	o'		
	0.6832	4	d   0 to	<del></del>	5	Surface tension		_
a b	0.7095	4	e   100 °C	0.1515	5	dynes/cm. 20°C	18.96 17.99	5 5
Ref. Index		1	d'   to			40	17.05	5
n <sub>D</sub> 20°C	1.3966	2		0,249	5	Parachor [P]		
25 30	1.3940	2	d g/ml vc ml/g tc °C	4.202	5	20°C		
"C"	0.7645	4	11 -	249.	5	40		_
MR (Obs.)	34.12	2	P <sub>c</sub> mm	20522.	5	Sugd.	296.2	5
MR (Calc.)	,	5	PV/RT 25°C	0.9941	5	Exp. L.1.%/wt. u.		
(nD-d/2) Dielectric	1.0506	2	30 mm	1.0000	5		120.	2
	4 01701	+-	BP t <sub>e</sub>	0.9520	5	Flash Point °C		
A 0 to B 119 °C		4	tc	0.260	5	Fire Point		
c ———	222.79	4	ΔHc kcal/m	1046.34	2	M Spec. Ultra V.		
A*  0 to	1.33013	5	ΔHf ΔFf			X-Ray Dif.		
B* ∟100 °C	1166.27	] ]	Viscosity			Infrared		
¢ .—	=		centistokes			Solubility in + Acetone	İ	
t <sub>k</sub> to t <sub>x</sub> °C			η ∘c			Carbon tet.	l	
A' to		+-				Benzene Ether		
B' ∟ _ °C			B <sup>V</sup>   to		-	n-Heptane		
A¹* to		+	B' to			Ethanol Water	İ	
B'* °C			(B <sup>V</sup> ) to	1		Water in		
Ac   119 to	7.3332	5	(A <sup>V</sup> ) °C	1				
Bc tc_°C	1544. 262.	5	c <sub>p</sub> liq. °K					
Cryos. A°		+-		1			į	
consts. B°		L	Р •					
te °C	93.39	5	c <sub>v</sub> vap.	1				
$T_R = 0.7$	5 T <sub>c</sub>			•		+ grams/100 gram	s solvent	
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from de	t. da			
SOURCE:		A1	PI					
PURIFICAT			PI					
LITERATU	RE REFERE	NCES	<b>5:</b>					

						No. 39
NAME	2-Methy	-2-h	exene			STRUCTURAL FORMULA
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH=C CH <sub>3</sub>
Mole % Pur.	Ref. Mo	lecul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.13	82	ch <sub>3</sub>
		Ref.		T	Ref.	Re
F.P. °C	-130.350	2	dt/dP			f to
F.P. 1009	6	<b>├</b> ──	°C/mm 25°C	0.3897	5	g <u>*K</u>
B. P. °C 760 mm	95.41	2	BP	0.0446	4	h
1 <b>0</b> 0	38.91	4	t <sub>e</sub>	0.0362	5	f' to to "K
<b>3</b> 0 10	14.01 -48.87	5	30 mm	0.6234	5_	h! 1
1	-36.05	5	ΔHm cal/g	<del> </del>	<b>⊹</b> ⊸¦	m to
Pressure mm 25°C	53.40	_	∆Hv cal/g 25°C	87.56	5	n
mm 25°C	52.60 991.4	5	30 mm	89.26	5	0
Density		Ť	BP t <sub>e</sub>	76.71 75.32	5	m' to
g/ml 20°C		2	t <sub>e</sub> (d, e)	75, 27	5	n' <u>°K</u> _
d <sub>4</sub> 25	0.7038 0.6994	2 4	ΔHv/T <sub>e</sub>	19.57	5	
a	0.7257	4	d 14 to		5	Surface tension dynes/cm. 20°C 20.81 5
ь	-0.0386	4			5	<b>8</b> 30   19.79   5
Ref. Index		2	e' °C	;		40 18.79 5
<sup>n</sup> D 20°C	1.4079	2	d <sub>c</sub> g/ml	0.242	5	Parachor [P] 20°C
30	1.4052	4	v <sub>c</sub> m1/g t <sub>c</sub> °C	4.13 266.	5 5	30
"C"	0.7719	4	P <sub>c</sub> mm	21467.	5	40 Sugd, 296. 2 5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L. 1. %/wt.
(nD-d/2)	1.0565	2	25°C 30 mm	0.9972 1.0000	5 5	u. Dispersion 127. 2
Dielectric			BP	0.9510	5	Flash Point *C
A 14 to		4	t <sub>e</sub> t <sub>c</sub>	0.9430 0.259	5	Fire Point
B (_132 °C	220.87	4	ΔHc kcal/m		2	M. Spec.
A* 14 to	1.33367	5	ΔHf ΔFf	1	-	Ultra V. X-Ray Dif.
B* _11 <u>0</u> °⊆   K	1202.03	5	Viscosity	-	+-	Infrared
с	_		centistokes			Solubility in +
t <sub>k</sub> to			″າ °c			Acetone Carbon tet.
t <sub>x</sub> °C		<del> </del>				Benzene Ether
B'°C				<del> </del>	+	n-Heptane
C' to			B <sup>V</sup> to C			Ethanol Water
A'* to B'* °C			(B <sup>V</sup> )  - to	-j		Water in
Ac  132 to		5	(A <sup>V</sup> )  °C			
Bc tc °C	1590. 261.	5	c liq. °K	1		
Cryos. A°		-	c <sub>p</sub> vap. °K			
consts. B°		ļ	c vap.			
t <sub>e</sub> °C T <sub>R</sub> = 0.7	104.69	5	vb.		لــــــــــــــــــــــــــــــــــــــ	+ 4100
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc from de	+ d-	† grams/100 grams solvent ta 5-Calc, by formula
SOURCE:	C25. 1-D0#	AF		- Carc. Irom de	ua	a J-Carc, by formula
PURIFICA	TION:	AF		<del></del>		
	RE REFERE					
L						

							No. 40
NAME	3-Methyl	-cis-	2-hexene			STRUCTURAL F	ORMULA
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> C =	сиси.
Mole	Ref. Mo	11		Malasslan		сн.	,
% Pur.	Kei. Mo	rmul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	82		<b>,</b> 
		Ref.			Ref		Re
F.P. *C			dt/dP			f to	
F. P. 1007	<u> </u>	<u> </u>	*C/mm 25*C	0.3700	5	g <u>°K</u> _	
B. P. *C 760 mm	94.	2	BP	0.0445	4	h	
100 30	37.68 12.85	4	t <sub>e</sub> 30 mm	0.0363 0.6214	5 5	f' to to	
10	-5.98	5	ΔHm cal/g	0.0214	۲	h'	
1	-37.04	5	ΔHv cal/g		$\vdash$	m to	
Pressure mm 25°C	55.75	5	25°C	86.96	5	n •K	
t <sub>e</sub>	987.4	5	30 mm BP	88.84 76.35	5 5	<b> </b>	
Density	0.7136		te te (d, e)	74.99	5	m' to	
g/ml 20°0 dt 25 d4 30	0.7120 0.7080	2 2	t <sub>e</sub> (d, e)	74.95	5	0	
<sup>4</sup> 30	0.7039	4	ΔHv/T <sub>e</sub>	19.57	5	Surface tension	
a b	0.7282 -0.0 <sub>3</sub> 80	4	e   115 °C		5	dynes/cm. 20°C	21.26 5
Ref. Index		+	d'   to			30 40	20.29   5   19.34   5
n <sub>D</sub> 20°0	1.410	2	d <sub>c</sub> g/ml	0,245	5	Parachor [P]	
25 30	1.407 1.405	2	V mi/g	4.08	5	20°C	
"C"	0.7667	4	1 -	266.	5	40	
MR (Obs.)	34.2	2	P <sub>c</sub> mm PV/RT	21745.	5	Sugd.	296.2 5
MR (Calc. (nD-d/2)	) 34.059 1.054	5 2	25°C	0.9967	5	Exp. L.1.%/wt. u.	
Dielectric	<del></del>	+-	30 mm BP	1.0000 0.9510	5	Dispersion	127. 2
A   5 t		4	te	0.9431	5	Flash Point °C Fire Point	
B [131 °C	221.14	4	t <sub>c.</sub>	0.259	5	M Spec.	
A*  5 to	<del></del>	5	ΔHc kcal/m ΔHf	1043.68	2	Ultra V.	
B* 115 °C		5	ΔFf			X-Ray Dif. Infrared	
K — — —	1		Viscosity centistokes	l		Solubility in +	
ւել			η °c			Acetone Carbon tet.	
t <sub>x</sub> i °C		<b>├</b>		1		Benzene	
B'					$\sqcup$	Ether n-Heptane	
C'	<b>_</b>		B <sup>V</sup>   to A <sup>V</sup>   °C			Ethanol Water	
A'* to B'* °C			(B <sup>V</sup> ) to			Water in	
Ac   131 to	7.3437	5	(A <sup>V</sup> )  °C				
Bc tc_°C	1585.	5	c <sub>p</sub> liq. °K	<del>                                     </del>	$\vdash$		
Cryos. A	202.	1,	-				
consts. B			р				
t <sub>e</sub> °C	103.11	5	c <sub>v</sub> vap.				
$T_{\mathbf{R}} = 0.7$						+ grams/100 gran	ns solvent
	CES: 1-Dew	2-AI		alc. from det	t. dat	ta 5-Calc. by form	nula
SOURCE:		AI	PI				
PURIFICA		AI				· · · · · · · · · · · · · · · · · · ·	
LITERATU	RE REFERE	NCES	<b>:</b>				

							No. 41	
NAME	3-Methyl	-tran	s-2-hexene			STRUCTURAL	FORMUL	.A
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> C	= CHCH	
Mole % Pur.	Ref. Mo	lecul rmuk	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	2	3, 2,2,	н <sub>3</sub>	
		Ref.			Ref.			Ref.
F.P. °C		<u> </u>	dt/dP			f to		
F.P. 100%	·	-	°C/mm 25°C	0.3700	5	g '° <u>K</u>	1	1
B. P. °C 760 mm	94.	2	BP	0.0445	4	h	<del> </del>	├
100 30	37.68	4	t <sub>e</sub>	0.0363	5	f' to		
10	12.85 -5.98	5	30 mm	0.6214	5	h'=		
1	-37.04	5	ΔHm cal/g	+		m to	<u> </u>	$\vdash$
Pressure mm 25°C	55.75	5	ΔHv cal/g 25°C	86.96	5	n  •K		1
t <sub>e</sub>	987.4	5	30 mm	88.84	5	<u> </u>		
Density	<del> </del>	$\vdash$	BP t <sub>e</sub>	76.35 74.99	5	m' to		1
g/ml 20°C		2 2	te (d, e)	74.95	5	n'	┧	
d <sub>4</sub> 30	0.7080 0.7039	4	ΔHv/T <sub>e</sub>	19.57	5		<del> </del>	├—
a	0,7282	4	d 5 to	90.81	5	Surface tension dynes/cm. 20°C	21.26	5
ь	-0.0 <sub>3</sub> 80	4	-å, -115 😤		5	<b>8</b> 30	20.29	5
Ref. Index		2	e' °C	L		40	19.34	5
<sup>n</sup> D 25	1.407	2	d <sub>c</sub> g/ml	0.245	5	Parachor [P] 20°C		
30	1.405	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.08 266.	5	30		
"C"	0.7667	4	P <sub>c</sub> mm	21745.	5	40 Sugd	. 296. 2	5
MR (Obs.) MR (Calc.		2 5	PV/RT		+	Exp. L. l. %/wt.		$\vdash$
(nD-d/2)	1.054	2	25°C 30 mm	0.9967	5 5	u.	1	
Dielectric			BP	1.0000 0.9510	5	Dispersion	127.	2
A 5 to	6. 92821	4	te	0.9431	5	Flash Point C Fire Point	1	1
B (131 °C	_ 1275.50 221.14	4	tc  ΔHc kcal/m	0.259	2	M. Spec.		$\vdash$
A* 5 to	1.33242	5	ΔHf	1045.00	-	Ultra V. X-Ray Dif.		İ
B*[115 °C		5	ΔFf	ļ	igsquare	Infrared		
K c			Viscosity centistokes			Solubility in +		
t <sub>k</sub> to	-	1	η °c			Acetone Carbon tet.	1	l
t <sub>x</sub>						Benzene		
A'   to B' C					Ш	Ether n-Heptane		ļ
C' '			B <sup>V</sup> to C	1		Ethanol		1
A'* to				_		Water Water in		ŀ
B'* °C	<del></del>	+	(B <sup>V</sup> )  to			11202 111	<del>                                     </del>	$\vdash$
Ac 131 to	7.3437 1585.	5	(A <sup>V</sup> )  °C	+	$\vdash$			
Ce	262.	5	c <sub>p</sub> liq. °K					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	103.11	5	c <sub>v</sub> vap.				1	
$T_R = 0.7$	5 T <sub>C</sub>					grams/100 gra	ms solver	ıt
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by fo	rmula	
SOURCE:			PI					
PURIFICA'			PI					
LITERATU	RE REFERE	NCES	5:					
L								

							No. 42	
NAME	4-Methyl	-cis-	-2-hexene			STRUCTURAL	FORMULA	
						сн <sub>3</sub> сн <sub>2</sub> сн сн	=CHCH	
						Ċн <sub>3</sub>	3	
Mole % Pur.	Ref. Mo	rmul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	32	3		
		Ref.			Ref			Ref
F.P. °C			dt/dP			f to		
F.P. 1009	6		*C/mm 25*C	0,2900	5	g <u> </u>	İ	l
B, P, *C 760 mm	87.37	2	BP	0.0440	4	h		<u> </u>
100	31.74	4	t <sub>e</sub>	0.0364	5	f' to	İ	ļ
30 10	7.27 -11.3	4 5	30 mm	0.6122	5	h'		
1	-41.8	5	ΔHm cal/g ΔHv cal/g	<del> </del>	<del> </del>	m   to		
Pressure mm 25°C	73.49	5	25°C	84.01	5	n ' <u>*K</u> _		
t <sub>e</sub>	969.6	5	30 mm BP	86.69 74.50	5	<u>°</u>		
Density			,	73.26	5	m' to		
g/ml 20°0	0.6996	2 2	t <sub>e</sub> (d, e)	73.23	5	", ' <u>-</u> "-		
dt 25 4 30	0.6910	4	ΔHv/T <sub>e</sub>	19.50	5	Surface tension		
a b	0.7168	4	_e _l _l10 °C	0.1521	5	dynes/cm. 20°C	19.81 18.83	5 5
Ref. Index	-0.0385	-	d' to	l .		40	17.87	5
n <sub>D</sub> 20°0	1.4024	2	<del></del>	0,238	5	Parachor [P]		
25 30	1.3997	2	d g/ml vc ml/g tc °C	4.20	5	20°C		l
"C"	0.7667	4	,, ~	254.	5	40	206.2	_
MR (Obs.	34.20	2	P <sub>c</sub> mm	20630.	3	Exp. L.1.%/wt.	296. 2	5
MR (Calc. (nD-d/2)	) 34.059 1.0526	5 2	25°C	0.9948	5	u.		
Dielectric		<u> </u>	30 mm BP	1.0000	5	Dispersion	122.	2
A 0 t	6.89863	4	t <sub>e</sub>	0.9446	5	Flash Point °C Fire Point		
B 1_122*9	222.	4	t <sub>c</sub>	0.260	5	M Spec.		
A*  0 to		5	ΔHf	1045.36	-	Ultra V. X-Ray Dif.	ļ	
B* 110°		5	ΔFf		ļ	Infrared		
K — —		1	Viscosity centistokes		Ì	Solubility in +		
tk   to			η •c		ļ	Acetone Carbon tet.		
t i		-				Benzene Ether		İ
B' •			B <sup>V</sup> l to		├	n-Heptane		
A!* to			B' to A' °C			Ethanol Water		
B'* °			(B <sup>V</sup> ) to	-	l	Water in		<u> </u>
Ac   122 to		5	(A <sup>V</sup> ) °C					
Bc tc_°	261.	5	c <sub>p</sub> liq. °K					
Cryos. A		Ė	c <sub>p</sub> vap. °K					
consts. B		<u> </u>	łi <sup>–</sup>		1			
te °C	95.74	5	c <sub>v</sub> vap.		L	<u> </u>	<u> </u>	<u> </u>
T <sub>R</sub> = 0.7			DI 2 II			grams/100 gram	ns solven	t
SOURCE:	CE3: 1-DOW			aic, from de	t. da	ta 5-Calc. by for	mula	
PURIFICA	TION:		PI PI					
	RE REFERE							
			•					

							No. 43	
NAME _	4-Methyl-	tran	s-2-hexene			STRUCTURAL	FORMUL	A
						сн <sub>3</sub> сн=сн сн	CH <sub>2</sub> CH <sub>3</sub>	
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 98.18	82	Сн	3	
		Ref.			Ref.			Ref
F.P. °C F.P. 100%	-126.5	2	dt/dP °C/mm	0.2020		f to g° <u>K</u>		
B.P. °C 760 mm 100	87.6 31.9	2	25°C BP t <sub>e</sub>	0.2920 0.0441 0.0364	5 4 5	h to		
30 10	7.40 -11.2	4 5	30 mm	0.6129	5	g' :° <u>K</u>		ĺ
1	-41.7	5	ΔHm cal/g		<b>├</b> —¦	m   to		
Pressure mm 25°C t <sub>e</sub>	72.99 970.3	5	AHv cal/g 25°C 30 mm BP	83.92 86.67 74.49	5 5 5	n		
Density g/ml 20°C	0.6975 0.6932	2 2	te te (d, e)	73.25 73.21	5	m' to		
d <sup>t</sup> 25 4 30	0.6889	4	ΔHv/T <sub>e</sub>	19.48	5	Surface tension		
a b	0.7146 -0.0 <sub>3</sub> 84	4	e 96 °C to	87.80 0.1519	5	dynes/cm. 20°C 30 40	19.57 18.61 17.66	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30	1.4023 1.3997 1.3970	2 2 4	e'   °C dc g/ml vc ml/g tc °C	0.236 4.244	5	Parachor [P] 20°C 30		
"C"	0.7688	4	1 -	254.	5	40		
MR (Obs.) MR (Calc.)	34.30 34.059	<b>2</b> 5	P <sub>c</sub> mm PV/RT 25°C	0. 9943	5	Sugd. Exp. L.1.%/wt.	296.2	5
(nD-d/2) Dielectric	1.0536	2	30 mm BP	1.0000	5	Dispersion	124.	2
A 7 to B   122 °C	6.89566 1243.0	4	t e t c	0.9446 0.260	5	Flash Point °C Fire Point M. Spec.		
C A*  7 to B*  106 °C	1.30573 1165.24	5 5	ΔHc kcal/m ΔHf ΔFf	1044.38	2	Ultra V. X-Ray Dif. Infrared		
c t <sub>k</sub> to c t <sub>x</sub> °C			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet. Benzene		
A'   to B'   °C C'			B <sup>V</sup> to			Ether n-Heptane Ethanol		
A¹* to B¹* °C			$ \begin{array}{c c} B^{\mathbf{v}} & \text{to} \\ A^{\mathbf{v}} & & {}^{\circ}C \\ \hline (B^{\mathbf{v}}) & & & \\ \hline \end{array} $			Water Water in		
Acl 122 to Bc tc °C Cc	7.3106 1545. 261.	5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	96.01	5	c <sub>w</sub> vap.	1				
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>					grams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP		-				
PURIFICAT		AP		<del></del>				
LITERATOR	RE REFERE	NCES	<b>5</b> :					
						<del></del>		

									No. 44	
NAME	5 <b>- M</b>	ethyl-ci	s-2-hexene				STR	UCTURAL		
Mole	Ref.			Molecu			CI	н <sub>3</sub> сн сн <sub>2</sub> сн сн <sub>3</sub>	=CH CH <sub>3</sub>	
% Pur.		Form		Weight	98.18					<b>5</b> (
	<del></del>	Re	ef.			Ref			1	Ref.
F.P. *C F.P. 100%	<u> </u>	-+	dt/dP				f	to		
B. P. *C	<del> </del>	-+	*C/mm 25*C	0	. 3119	5	g	i		
760 mm	91.		BP		. 0462	4 5	h f'	<del>                                     </del>		+
100 30	33.0 7.77		t <sub>e</sub>	1	. 0379	5	g'	to *K	1	
10	-11.3	5	5 ATT 1/2		. 6301	-	h'	 i	1	
1	-42.5					$\vdash$	m	l to	ļ	+
Pressure	70.05	.   .	ΔHv cal/g		. 01	5	n	<u>*</u> K_		
mm 25°C	70.05	,   5	30 mm	84	. 52	5	0	i		İ
Density	1		BP		. 46 . 17	5	m'	l to		
g/ml 20°C					. 13	5	n' o'	<u>*</u> K	ĺ	
dt 25 4 30	0.69		4 LT/T	18	. 71	5	٥.	<u> </u>		<u> </u>
a 30	0. 71		<b>—</b>	to 85	. 65	5		ace tension	10.05	١,
ь	-0.03			∴⊆∣∘	. 1449	5	gyne	s/cm. 20°C 30	19.85 18.94	5
Ref. Index			<b>-                                    </b>	to °C		i i		40	18.04	5
n <sub>D</sub> 20°C				0	. 215	5	Para	chor [P]		1
25 30	1.39		II v mi/g	4	. 64	5		20°C 30		
"C"	0.76		<b>⊣</b> եւ Վ	260		5		40		
MR (Obs.)	34.0	2	P <sub>c</sub> mm	18614	•	5			296.2	5
MR (Calc.)	34.05	9   5	PV/RT		. 9956	5	Exp.	L.1.%/wt.		
(nD-d/2)	1.05	0 Z	30 mm	1	.0000	5	Disp	u. ersion	122.	2
Dielectric	<u> </u>		BP		.9520 .9441	5	Flas	h Point °C		<u> </u>
A 0 to B 1127 °C					. 255	5	Fire	Point		
c Living	222.	'   4		m 1044	. 74	2	M S			
A*   0 to	1,16	082 5	ΔHf		• -		Ultr.	a V. ay Dif.		
B* 110 °C	1133.96	,   5				$\vdash$	Infra			<u> L</u>
С	}		Viscosity centistoke	.				bility in +		
t <sub>k</sub> to	]		7	°C				tone rbon tet.		1
x I			<b>」</b> `				Ber	nzene		
A' to		-					Eth	er Ieptane		
c, – – –	-	į	B <sub>v</sub>	to				anol		
A'* to			Av i	•c			Wa	ter ter in		
B'* °C	<del></del>		(B <sup>V</sup> )	to			W &	ter III	<b></b>	+
Ac   127 to Bc   t <sub>c</sub> *C	7.17 1522.	33   5		•c		Ш			1	-
Cc	264.	5		°K		1 1				1
Cryos. A°			c <sub>p</sub> vap.	•ĸ						1
consts. B°	ļ		11 -							
t <sub>e</sub> °C	100.21	5	c <sub>v</sub> vap.	L						<u></u>
$T_{R} = 0.7$							+ gra	ms/100 gran	ms solver	nt
REFERENC	ES: 1-D	ow 2-	API 3-Lit.	4-Calc. f	rom de	t. dat	a 5-	Calc. by for	mula	
SOURCE:			API							
PURIFICAT			API							
LITERATU	RE REF	ERENC	ES:							

							No. 45
NAME	5-Methy	l-trai	ns-2-hexene			STRUCTURAL	FORMULA
<b> </b>						сн <sub>3</sub> сн сн <sub>2</sub> сн	I=CH CH <sub>3</sub>
Mole % Pur.		lecul		Molecular Weight 98,18	2	Ċн <sub>3</sub>	
		Ref.			Ref.		Ref
F.P. °C F.P. 100%			dt/dP °C/mm			f to g*K	
B. P. ℃	04		25°C BP	0.2618 0.0458	5 4	h	
760 mm 100	86. 28.55	2 4	t <sub>e</sub>	0.0381	5	f' to	
30 10	3.58 -15.2	4 5	30 mm	0.6226	5_	g' <u>K</u>	
1	-46.0	5	ΔHm cal/g		<u> </u>	h' i	<del> </del>
Pressure			ΔHv cal/g 25°C	79.92	5	n	
mm 25°C	85.52 966.5	5	30 mm BP	83.02 71.01	5	0	
Density			t <sub>e</sub> (d, e)	69.81	5	m' to	
g/ml 20°C	0.700	2		69.76	5	ō'	1
dt 25 4 30	0.692	4	ΔHv/T <sub>e</sub>	18.63	5	Surface tension	
a b	0.7161 -0.0 <sub>3</sub> 79	4 4	d 0 to e 110 °C d' to	83.54 0.1456	5	dynes/cm. 20°C 8 30	19.85 5 18.93 5
Ref. Index		2	e'		<u> </u>	Parachor [P]	18.02 5
25	1.397	2	d <sub>c</sub> g/ml	0.211 4.74	5	Parachor [P] 20°C	
30	1. 395	4	vc ml/g tc °C	253.	5	30 40	
"C"	0.7619	4	P <sub>c</sub> mm	18186.	5		. 296.2 5
MR (Obs.) MR (Calc.		5	PV/RT	0.0043	_	Exp. L.1.%/wt.	
(nD-d/2)	1.050	2	25°C 30 mm	0.9942 1.0000	5	u. Dispersion	124. 2
Dielectric		ļ	BP	0.9520 0.9445	5	Flash Point °C	
A 0 to B (121°C	1183.25	4	t e t c	0.258	5	Fire Point M. Spec.	
A* 0 to	1,13430	5	ΔHc kcal/m ΔHf	1043.74	2	Ultra V.	
B*[110°C		5	ΔFf			X-Ray Dif. Infrared	
c	_	İ	Viscosity centistokes		1	Solubility in + Acetone	
t <sub>k</sub> to			η °C			Carbon tet.	
A' to	<del></del>	<del>                                     </del>		}		Benzene Ether	
B'°			B <sup>V</sup> to		<del>                                     </del>	n-Heptane	
A¹* to	<del>                                     </del>	+	B <sup>V</sup> to C	1	l	Ethanol Water	
B'* °C		L	(B <sup>V</sup> )  to	1		Water in	<del>                                     </del>
Acl 121 to	7.1390	5	(A <sup>V</sup> )  °C				
Bc t <sub>c</sub> °C	1486. 263.	5 5	c <sub>p</sub> liq. °K				
Cryos, A° consts, B°			c <sub>p</sub> vap. °K				
te °C	94.60	5	c <sub>v</sub> vap.		L		
$T_{\mathbf{R}} = 0.7$						† grams/100 gra	
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	rmula
SOURCE:		AF					
PURIFICA		AI				<del></del>	
LILEKATU	RE REFERE	NCE	<b>:</b>				

							No. 46	
NAME	2-Me	thyl-cis-	3-hexene			STRUCTURAL	FORMUL	A.
						сн <sub>3</sub> сн сн=сн	сн.сн.	
Mole	1,			Malanda		Ċн <sub>3</sub>	22	
Mole % Pur.	Ref.	Formul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.1	82	3		
		Ref.			Ref			Ref
F.P. C			dt/dP			f to		T
F.P. 100%			*C/mm 25*C	0.2410	ا ۔ ا	g <u>"K</u>		
B. P. °C 760 mm	86.	2	BP	0.2618 0.0458	5 4	h		-
100	28.6	4	t <sub>e</sub>	0.0381	5	f' to g'°K		ì
30 10	3.58 -15.2	4 5	30 mm	0.6226	5	p <sub>1</sub>	1	1
1	-46.0	5	ΔHm cal/g	<del> </del>	$\vdash$	m to		+-
Pressure	05.53		ΔHv cal/g 25°C	79.89	5	n   •K	]	
mm 25°C	85.52 966.5	5 5	30 mm	83.02	5	<u> </u>		
Density	<b>-</b>		BP te	71.01 69.80	5 5	m'   to		
g/ml 20°C	0.694		t <sub>e</sub> (d, e)	69.76	5	n'   <u>*K</u> _	1	
dt 25 4 30	0.690		ΔHv/T <sub>e</sub>	18.63	5	<u> </u>		+
	0.710		d   -5 to		5	Surface tension dynes/cm. 20°C	19.17	5
ь	-0.037		-a,  -110- <b>*</b>		"	30	18.28	5
Ref. Index	1 200		e' i °C	1		40	17.40	5
<sup>n</sup> D 20°C	1.399		d <sub>c</sub> g/ml			Parachor [P] 20°C	Į	
30	1.394		d g/ml vc ml/g tc °C	252.	5	30		1
"C"	0.766	-	Pcmm	17992.	5	40 Sugd	296.2	5
MR (Obs.) MR (Calc.)	34.2 34.059	) 2	PV/RT	<del>                                     </del>	1	Exp. L.1.%/wt.		†
(nD-d/2)	1.052		25°C 30 mm	0.9938	5	u.	122.	2
Dielectric			BP	0.9520	5	Dispersion Flash Point °C	122.	+-
A -5 to	6.722		t <sub>e</sub>	0.9445	5	Fire Point		
B 1_121 °C	222.	4 4	t <sub>c</sub>	1044.74	2	M Spec.		
A*  -5 to	1,134	30 5	ΔHf		-	Ultra V. X-Ray Dif.		İ
B*	1105.83	5	ΔFf	<del> </del>	├	Infrared		
c			Viscosity centistokes			Solubility in +		
tk to			η •α			Acetone Carbon tet.		1
t <sub>x</sub> i °C		-+		1		Benzene	1	
B' ∟ °C	ļ			<b>_</b>		Ether n-Heptane		
C'		-	B <sup>V</sup>   to			Ethanol Water		1
A'* to B'* °C	1			-		Water in		
Ac! 121 to	7,138	8 5	(A <sup>V</sup> )  to	1				
Bc t C	1486.	5	c <sub>p</sub> liq. °K		+-			
Ce	263.	5						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	94.60	5	c <sub>w</sub> vap.				<u> </u>	
$T_{R} = 0.75$						grams/100 gra		ı <b>t</b>
REFERENC	ES: 1-Do			Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE: PURIFICAT	ION.	AI AI						
LITERATUE								
			•					

No. 47 2-Methyl-trans-3-hexene NAME STRUCTURAL FORMULA CH3CH CH=CHCH2CH3 Ċн<sub>3</sub> Mole Ref. Molecular Molecular Weight 98,182 C7H14 % Pur. Formula Ref. Ref. Ref. F.P. °C dt/dP f. to F.P. 100% °C/mm °K g 25°C 0.2618 5 B. P. °C h ВP 0.0458 4 760 mm 86. 2 0.0381 5 f١ to 28.6 100 4 <u>°к</u> g¹ 30 3.58 4 0.6226 5 30 mm 10 -15.2 5 h! ∆Hm cal/g 5 1 -46.0m to AHv cal/g Pressure n °K 25°C 79.89 5 mm 25°C 85.52 5 o 30 mm 83.02 5 966.5 5 te 5 BP 71.01 m' to Density t<sub>e</sub> (d, e) 69.80 5 g/ml 20°C n' °K 0.694 2 69.76 5 o'  $d_4^t$ 25 0.690 2 ΔHv/T<sub>e</sub> 18.63 5 0.686 4 Surface tension 83.54 5 a 0.710 4 19.17 5 dynes/cm. 20°C 110 °C 0.1457 5 -0.0379 h 4 18.28 5 30 to 40 17.40 5 Ref. Index e' °C 20°C 1.399 2 [P]  $n_D$ Parachor dc g/ml 25 1.396 2 20°C vc ml/g t\_ °C 30 1.394 4 30 252. 5 tc 40 "C" 0.7667 4  $P_c$  mm 17992. 5 Sugd. 296.2 5 MR (Obs.) 34.2 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.059 5 25°C 0.9938 5 u. (nD-d/2) 1.052 2 30 mm 1.0000 5 Dispersion 2 124. Dielectric BP 0.9520 Flash Point C 0.9445 5 A -5 to 6.72252 Fire Point B 121 °C 1183.25 M. Spec. С 222. 4 AHc kcal/m 1044.74 Ultra V. ΔHf A\* -5 to 1.13430 5 X-Ray Dif. ΔFf B\* 110 °C 1105.83 Infra red K Viscosity Solubility in c centistokes Acetone  $\mathbf{t_k}$ to Carbon tet. °C Benzene A to Ether B °C n-Heptane B<sup>V</sup> A C' to Ethanol °C Water A'\* to Water in (B<sup>V</sup>)I R'\* °C to Ac| 121 to 7.1388 (A<sup>V</sup>)| 5 °C Bc tc °C 1486. 5 c<sub>p</sub> liq. ۰ĸ 5 263. Cryos, A° c<sub>p</sub> vap. °K consts. B° c<sub>v</sub> vap. te °C 94.60 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

						No. 4	8
NAME	3-Methyl-	cis-	3-hexene			STRUCTURAL FORMUL	.A
						CH3CH2CH=C CH2CH3	
Mole % Pur.	Ref. Mo	lecul rmul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	32	с́н <sub>3</sub>	
76 F u I .	1 1 1 1	Ref.		weight /o.re	Ref		Ref
F.P. °C		1.01.	dt/dP	T	1		<del>                                     </del>
F.P. 100%			*C/mm			f to g K	
B. P. *C			25 <b>°C</b> BP	0.3640 0.0465	5	h .	
760 mm 100	95.35 36.9	2 4	t <sub>e</sub>	0.0378	4 5	f' to	$\top$
30	11.46	4	30 mm	0.6358	5	g'   'K_	
10 1	-7.8 -39.3	5	ΔHm cal/g			h'	
Pressure	-37.3	<u> </u>	ΔHv cal/g			m to	l
mm 25°C	58.69	5	25°C 30 mm	83.98	5	n °K	
t <sub>e</sub>	993.	5	BP BP	85.99 73.66	5	h	┿
Density g/ml 20°C	0.7122	2	t <sub>e</sub>	72.28	5	m' to to	
dt 25	0.7132 0.7089	2	te (d, e)	72.23	5	o'	1
<sup>4</sup> 30	0.7046	4	ΔHv/T <sub>e</sub>	18.76	5	Surface tension	+
a b	0.7303	4	d 0 to e 1 115 ℃	87.67 0.1469	5	dynes/cm. 20°C 21.40	5
Ref. Index	-0.0 <sub>3</sub> 84	*	d' l to			30 20.38 40 19.37	5
n <sub>D</sub> 20°C	1.4123	2	e' j •c		├	Parachor [P]	+-
25 30	1.4096	2	d g/ml v ml/g		l	20°C	
"C"	1.4070 0.7695	4	d g/ml vc ml/g tc °C	268.	5	30 40	
MR (Obs.)	34.28	2	P <sub>c</sub> mm	19203.	5	Sugd. 296.2	5
MR (Calc.)		5	PV/RT 25°C	0.00/0	_	Exp. L.1.%/wt.	
(nD-d/2)	1.0557	2	30 mm	0.9968 1.0000	5	u. Dispersion 127.	2
Dielectric			BP	0.9515	5	Flash Point °C	+
A 0 to B 133 °C	6.77018 1230.40	4	te t <sub>c</sub>	0.9432	5	Fire Point	
c 135 0	221.	4	∆Hc kcal/m	1043.68	2	M Spec.	
A* 0 to	1.17140	5	ΔHf ΔFf	İ	1	Ultra V. X-Ray Dif.	
B* ∟115 °C	1151.0	5	Viscosity		_	Infrared	
с			centistokes		ĺ	Solubility in + Acetone	
tk to			η ∘c		ļ	Carbon tet.	
t <sub>x</sub>   *C	<u> </u>	-			1	Benzene Ether	
B' *C			L	ļ	—	n-Heptane	
C'		ļ	B <sup>V</sup>   to A <sup>V</sup>   *C			Ethanol Water	
A'* to B'* *C			7-5-		ļ	Water in	
Ac   133 to	7,1872	5	(A <sup>V</sup> )  to	1			
Bc t C	1543.	5	c <sub>p</sub> liq. °K	<u> </u>	+-	1	1
Cc	263.	5	· -				
Cryos. A° consts. B°			c <sub>p</sub> .vap. °K				
t <sub>e</sub> °C	105.07	5	c <sub>w</sub> vap.	1			
$T_{R} = 0.75$		Ь	IL			+ grams/100 grams solve	 nt
REFERENC		2-AI	PI 3-Lit. 4-0	alc. from de	t. da	ta 5-Calc. by formula	
SOURCE:			PI				
PURIFICAT	ION:	A.	PI				
LITERATUE	RE REFERE	ICES	: :				

No. 49 3-Methyl-trans-3-hexene STRUCTURAL FORMULA NAME CH3CH2CH=C CH2CH3 ĊН<sub>3</sub> Ref. Mole Molecular Molecular  $C_7H_{14}$ Weight 98.182 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g \_°K 25°C 0.3413 5 B. P. °C h BP 0.0464 760 mm 93.55 2 0.0379 5 ſ١ to 100 35.3 4 g' <u>°К</u> 30 9.95 4 30 mm 0.6330 5 10 -9.2 5 h' ∆Hm cal/g 5 40.5 to m AHv cal/g Pressure °K n 25°C 83,22 5 mm 25°C 63.13 5 o 30 mm 85.44 987.5 5 t<sub>e</sub> BP 73.14 5 m+ to Density 71.79 5 te te (d, e) n' °K g/ml 20°C 0.7099 2 71.74 o' d<sub>4</sub> 25 0.7056 2 AHv/Te 5 18,73 30 0.7013 4 Surface tension ď 0 86.91 5 to а 0.7271 4 dynes/cm. 20°C 21.01 ᇷᅱ 110 °C 0.1472 5 -0.0385 19.99 19.00 Ъ 4 5 30 5 40 Ref. Index e' 20°C 1.4107 2 <sup>n</sup>D [P] Parachor d<sub>c</sub> g/ml 25 1.4080 2 20°C vc ml/g t\_°C 30 1.4053 4 30 t<sub>c</sub> 265. 5 40 "C" 0.7702 4  $P_c$  mm 18933. 5 Sugd. 5 296.2 MR (Obs.) 34.32 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 34.059 5 25°C 0.9963 5 (nD-d/2)1.0558 2 30 mm 1.0000 2 5 Dispersion 127. Dielectric BP 0.9515 Flash Point C 0.9433 5 A 0 to 6.75851 4 Fire Point B 130 °C 1219.73 M. Spec. Ultra V. ΔHc kcal/m C 4 221. 1043,68 2 ΔHf A\* 0 to 5 1.16211 X-Ray Dif. ΔFf B\*[110 °C 1140.86 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $\mathbf{t_k}$ Carbon tet. °C Benzene to Ether В' °C n-Heptane B<sub>v</sub> | C' Ethanol °C Water A'\* to Water in ٠c (B<sup>V</sup>)| to Ac 130 to 7,1751 5  $(A^{V})$ °C Bc tc °C °C 1530. 5 c<sub>p</sub> liq. ۰ĸ Cc 263. 5 Cryos. A c<sub>p</sub> vap. °K consts. B° c vap. te °C 103.05 5  $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

						No. 50
NAME	2-Ethyl-	l-pen	tene			STRUCTURAL FORMULA
						CH (CH ) C = CH
						$CH_3(CH_2)_2C = CH_2$ $C_2H_5$
Mole % Pur.	Ref. Me	olecul ormul		Molecular Weight 98.18	2	2115
		Ref.			Ref	R
F.P. °C			dt/dP	1		f to
F.P. 100%			*C/mm			g   <u>*K</u>
B. P. ℃	0.4		25°C BP	0.3468 0.0464	5 4	h
760 mm 100	94. 35.7	2 4	te	0.0379	5	f! to
30	10.33	4	30 mm	0.6337	5	g'   <u>°</u> K_
10 1	-8.8 -40.2	5 5	ΔHm cal/g			h!
Pressure	· · · · · · · · · · · · · · · · · · ·	T	ΔHv cal/g	00.41	ا ۔ ا	m to to
mm 25°C	61.99	5	25°C 30 mm	83.41 85.58	5 5	0
t <sub>e</sub> Density	988.8	5	BP	73.27	5	m'   to
g/ml 20°C	0.708	2	te t_ (d, e)	71.91	5 5	n' ' K
dt 25 4 30	0.704 0.700	2 4	ΔHv/Te	18.74	5	0'
	0.700	-	d 0 to	<del></del>	5	Surface tension
a b	-0.038	4	d, 110 °C		5	dynes/cm. 20°C   20.78   5
Ref. Index			d'   to		1 1	40 18.93 5
n <sub>D</sub> 20°C	1.405 1.402	2 2	d <sub>c</sub> g/ml			Parachor [P]
30	1.399	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	266.	5	30
"C"	0.7622	4		19065.	5	40 Sugd. 296. 2
MR (Obs.)	34.0	2	P <sub>c</sub> mm PV/RT	17005.	┝┵┤	Exp. L.1.%/wt.
MR (Calc.) (nD-d/2)	34.059 1.051	5 2	25°C	0.9963	5	u.
Dielectric	1	╁╾┤	30 mm BP	1.0000 0.9515	5	Dispersion 125.
A 0 to	6, 76143	4	te	0.9433	5	Flash Point °C Fire Point
B <u>[131 °C</u>	1222.4	4	t <sub>c</sub>			M Spec.
C	221.	5	ΔHc kcal/m ΔHf	1045.08	2	Ultra V.
A* 0 to B* 110 °C	1.16444 1143.39	5	ΔFf			X-Ray Dif. Infrared
к — — —			Viscosity			Solubility in +
\$	-	1	r centistokes	}		Acetone
x I			<b>'</b>			Carbon tet. Benzene
A'   to B' ∟ °C						Ether
c,	1		B <sup>V</sup> to			n-Heptane Ethanol
A¹* to			AV I C	_		Water Water in
B'* °C		$\downarrow \downarrow$	(B <sup>V</sup> ) to			Water in
Ac   131 to	7.1782 1534.	5	(A <sup>V</sup> ) °C		$\sqcup$	
Cc Cc	263.	5	c <sub>p</sub> liq. °K			
Cryos. A° consts. B°			c <sub>p</sub> vap. °K			
t <sub>e</sub> °C	103.56	5	c <sub>v</sub> vap.			
$T_{R} = 0.75$	Tc		L	L		grams/100 grams solvent
	ES: 1-Dow	2-AF	I 3-Lit. 4-0	Calc. from de	t. dat	a 5-Calc, by formula
SOURCE:		AI				
PURIFICAT	ION:	AI	PI			
LITERATUI	RE REFERE	NCES	:			

No. 51

							No. 51
NAME	3-Ethyl-	l-per	itene			STRUCTURAL	FORMULA
						си си си с	тн=сн
						сн <sub>3</sub> сн <sub>2</sub> сн с	
Mole % Pur.	Ref. Mo	lecul rmula		Molecular Weight 98.18	32	Ċ <sub>2</sub> H <sub>5</sub>	
<del>/0 1 G1.</del>		Ref.	<u> </u>		Ref.	T	Ref
F. P. °C	-127.4	2	dt/dP		-	f to	
F.P. 1009			°C/mm			g  °K	
B.P. °C			25°C BP	0.2520 0.0461	5 4	h	
760 mm 100	85.13 27.4	2 4	t <sub>e</sub>	0.0384	5	f' to	
30	2.34	4	30 mm	0.6244	5	g' <u>°K</u>	
10 1	-16.5 -47.3	5	∆Hm cal/g			h'	
Pressure		Ť	ΔHv cal/g			m to	
mm 25°C	90.06	5	25°C 30 mm	78.81 82.03	5	0	
t <sub>e</sub>	964.1	5	BP	70.24	5	m' to	
Density g/ml 20°0	0. 6962	2	t <sub>e</sub> (d, e)	69.06 69.02	5 5	n'	
dt 25	0.6917	2	ΔHv/T <sub>e</sub>	18.48	5	0'	
	0.6872	4	d -5 to	82, 36	5	Surface tension	
a b	0.7142 -0.0 <sub>3</sub> 88	4	_e_ _1 <u>05</u> °C	0.1423	5	dynes/cm. 20°C	19.42 5 18.41 5
Ref. Index	<del></del>	$\Box$	d'  to  e'  °C			40	17.42 5
n <sub>D</sub> 20°C	1.3980	2	d g/ml			Parachor [P]	
25 30	1.3954 1.3928	2 4	v <sub>c</sub> ml/g t <sub>c</sub> °C			20°C 30	
"C"	0.7624	4		250.	5	40	
MR (Obs.)	34.04	2	P <sub>c</sub> mm	17476.	5		296. 2 5
MR (Calc.		5 2	PV/RT 25°C	0.9938	5	Exp. L.1.%/wt.	
(nD-d/2) Dielectric	1.0499	-	30 mm BP	1.0000	5	Dispersion	121. 2
A -5 to	6,70122	4	t	0.9520 0.9445	5	Flash Point C	
B (119 °C	1177.18	4	°C			Fire Point	
С	223.	4	ΔHc kcal/m ΔHf	1047.60	2	M. Spec. Ultra V.	
A* -5 to B* 105 °C		5	ΔFf			X-Ray Dif. Infrared	
к — — -	-		Viscosity			Solubility in +	
t <sub>k</sub> – tõ	-		centistokes 7°C			Acetone	
t <sub>x</sub> °C			7			Carbon tet. Benzene	
A'   to					i	Ether	
B'°	-		B <sup>v</sup> to A <sup>v</sup> °C			n-Heptane Ethanol	
A¹* to	1					Water	
B'* °C			(B <sup>V</sup> )  to			Water in	
Acl 119 to	7, 1192	5	(A <sup>V</sup> )  °C		ļ		
Bc tc °C	1481. 265.	5	c <sub>p</sub> liq. °K				
Cryos. A°			c <sub>p</sub> vap. °K				
consts. B	<del></del>	1	1				
t <sub>e</sub> °C	93.70	5	c <sub>v</sub> vap.	L		L	
$T_{\mathbf{R}} = 0.7$						grams/100 gra	
	CES: 1-Dow			Calc. from de	t. da	ita 5-Calc. by for	mula
SOURCE:			API				
PURIFICA			API				
LILEKATU	RE REFERE	NCES	<b>:</b>				

No. 52 NAME 2, 3-Dimethyl-1-pentene STRUCTURAL FORMULA  $CH_3CH_2CH C = CH_2$ ĊH<sub>3</sub>ĊH<sub>3</sub> Mole Ref. Molecular Molecular C7H14 % Pur. Formula Weight 98.182 Ref. Ref Ref. F.P. °C F.P. 100% -134.8 2 dt/dP to °C/mm °К g 25°C 0,2466 5 B. P. \*C h BP 0.0457 4 2 760 mm 84.26 f' 0.0381 5 to 100 27.0 4 g¹ °K 2.07 4 0.6209 30 30 mm 5 -16.7 5 10 h' ∆Hm cal/g -47.4 5 1 to AHv cal/g Pressure °K n 25°C 79.05 mm 25°C 91.70 5 o 30 mm 82.33 5 te 961.5 5 BP 70,53 5 Density g/ml 20°C m to 69.37 te (d, e) °K 0.7051 2 69.33 5 o† 25 0.7008 2 dt AHV/T 18,62 5 30 0.6965 4 Surface tension 82,63 -5 5 0.7223 4 20.43 dynes/cm. 20°C 105 °C 0.1435 ь -0.0384 4 30 19.43 5 ٦ď to 40 18.45 5 Ref. Index e' •c n<sub>D</sub> 20°C 1.4033 [P] Parachor d<sub>c</sub> g/ml 25 1.4007 2 4 20°C v<sub>c</sub> ml/g 30 1.3981 30 •C 250. 5 ŧ<sub>c</sub> 40 "C" 0.7623 4 mm 18171. 5 5 Sugd. 296.2 MR (Obs.) 34.00 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 34.059 25°C 0.9933 (nD-d/2)1.0508 2 30 mm Dispersion 1.0000 5 125. 2 Dielectric BP 0.9520 5 Flash Point °C te 0.9446 A -5 to 6,72442 4 Fire Point 1\_12<u>0 °C</u> 1181.0 M Spec C ΔHc kcal/m 223. 4 1043,42 Ultra V. ΔHf A\*| -5 to 1.13745 X-Ray Dif. ΔFf B\* 105 °C 1103.49 Infrared ĸ Viscosity Viscos..., centistokes °C Solubility in Acetone t<sub>x</sub> | to Carbon tet. •c Benzene to Ether В' •c n-Heptane Вv C' Ethanol  $\vec{A}^{\boldsymbol{v}}$ °C Water AI\* to Water in B'\* ۰c (BV) to Ac | 120 to 7.1424 (AV) °C 1485. Bc tc °C cp liq. °K 265. 5 Cc Cryos. Aº c<sub>p</sub> vap. ۰ĸ consts. B° c<sub>v</sub> vap. te °C 92.65 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 53

							No. 53	
NAME	2, 4-Dim	ethyl	-1-pentene		STRUCTURAL FORMULA			
1 1						C11 C11 C11 C		
					一	CH <sub>3</sub> CH CH <sub>2</sub> C	_	
Mole % Pur.		lecul rmuk		Molecular Weight 98,182	,	Ċн <sub>3</sub> с	:н <sub>3</sub>	
76 Fur.	1 1 1 1 1	Ref	<del></del>	Weight 98, 182	Ref.	T	Ref	
F, P, °C	-123,8	2	dt/dP	<del>                                     </del>	Kei.			
F. P. 100		+-	°C/mm			f to g &K		
B. P. ℃			25°C	0. 2261	5	h		
760 mm	81.64 24.7	2	BP t <sub>e</sub>	0.0454 0.0381	5	f' to		
30	-0.04	4	30 mm	0,6173	5	g' <u>°K</u>		
10 1	-18.7 -49.2	5	ΔHm cal/g		T -	h'		
Pressure	-47.2	+-	ΔHv cal/g			m   to		
mm 25°C	101.2	5	25°C	78.00	5	n <u>•K</u>		
t <sub>e</sub>	954.0	5	30 mm BP	81.55 69.95	5	- <u></u>	ļ <u>-</u>	
Density	0 (043	2	t <sub>e</sub> (d, e)	68.84	5	m' to		
g/ml 20°0	0.6943 0.6898	2		68.81	5	0'		
d <sub>4</sub> 25 30	0.6853	4	ΔHv/T <sub>e</sub>	18.62	5	Surface tension		
a	0.7124	4	d 0 to e 95 °C	81.54 0.1420	5	dynes/cm. 20°C	19.20 5	
b	-0.0388	4	d' l to	0.1420		30 40	18.20 5 17.21 5	
Ref. Index		2	e'   °C	ļ	-	Parachor [P]	1	
D 25	1.3959	2	d g/ml			20°C		
30	1.3952	4	vc m1/g tc °C	245.	5	30 40		
"C"	0.7656	4	P <sub>c</sub> mm	17719.	5		296.2 5	
MR (Obs. MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0514	2	25°C 30 mm	0.9917 1.0000	5	u. Dispersion	,,,,	
Dielectric	:	T	BP	0.9520	5	Flash Point °C	125. 2	
A 0 to		4	t e t c	0.9449	5	. Fire Point		
B 1_115 °C	224.	4	ΔHc kcal/m	1042,74	2	M. Spec.		
A* 0 to		5	ΔHf	1042.74	-	Ultra V.		
B*[ 100 °C	1098.95	5	ΔFf			X-Ray Dif. Infrared		
K c		İ	Viscosity centistokes			Solubility in +		
t <sub>k</sub> to		1	η °C			Acetone Carbon tet.		
t			,			Benzene		
A'   to		1				Ether		
č,' 3	<u>^</u>		B <sup>v</sup> to			n-Heptane Ethanol		
A'+ to			A I °C	.		Water		
B'* °C		1	(B <sup>V</sup> )  to			Water in	-	
Ac 115 to	7.1479	5	(A <sup>V</sup> )  °C	<b>_</b>		1		
Bc tc °C	265.	5	c <sub>p</sub> liq. °K	1				
Cryos. A		1	c <sub>p</sub> vap. °K		1			
consts. B	<u>`</u>	<del> </del>	<b>!!</b>					
t <sub>e</sub> °C	89.69	5	c <sub>v</sub> vap.	1	l	L		
$T_R = 0.7$						† grams/100 gra		
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICA			PI					
LITERATI	JRE REFERE	NCE	S:					
1								
L								

								No. 54	ł
NAME	3, 3-	Dim	ethyl-	-1-pentene			STRUCTURAL	FORMUL	A
							CH3		
		Γ					сн <sub>3</sub> сн <sub>2</sub> с сі	H=CH <sub>2</sub>	
Mole % Pur.	Ref.	Mo	lecul rmul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.1	82	с́н <sub>3</sub>	-	
-/0_1 u1.		1 10	Ref		weight /o.1	Ref	<del></del>		Ref
F.P. °C	-134.3		2	dt/dP		-	f   to	T	
F.P. 100%				°C/mm			g  *K_		
B. P. ℃				25°C BP	0.1965 0.0452	5 4	h ¦		
760 mm 100	77.54		2 4	t <sub>e</sub>	0.0383	5	f¹ to		Т
30 10	-3.5		4	30 mm	0.6116	5	g'   '°K_	ł	
1	-22.1 -52.2		5 5	ΔHm cal/g			h'		-
Pressure				ΔHv cal/g 25°C	7/ 14	ا ۔ ا	m to		
mm 25°C	119.2		5 5	30 mm	76.14 80.18	5		1	
t <sub>e</sub> Density	942.9		1 3	BP	68.68	5	m'   to		1
g/ml 20°C			2	te (d, e)	67.63 67.60	5	n'  °K_		
dt 25 4 30	0.6		2 4	AHv/Te	18.53	5			<u> </u>
	0.7		4	d   -10 to		5	Surface tension dynes/cm, 20°C	19.54	5
ь	-0.0		4	- d - 100 %	0.1418	5	30	18.58	5
Ref. Index		204		e' °			40	17.64	5
n <sub>D</sub> 20°C	1.39		2 2	d g/ml vc ml/g			Parachor [P] 20°C		
30	1.3	932	4	t <sub>c</sub> °C	240.	5	30		1
"C"	0.7		4	P <sub>c</sub> mm	17499.	5	40 Sugd.	296.2	5
MR (Obs.) MR (Calc.	34.0		5	PV/RT		$\vdash$	Exp. L.1.%/wt.		†
(nD-d/2)	1.04		2	25°C 30 mm	0.9909 1.0000	5	u. Dispersion	120.	2
Dielectric				BP	0.9525	5	Flash Point °C	120.	+
A -10 to		9410	4 4	te t <sub>c</sub>	0.9458	5	Fire Point		<u> </u>
B [_1,1, _6]	224.	0	4	∆Hc kcal/m	1044.77	2	M Spec.		
A*  -10 to		1439	5	ΔHf ΔFf	1		Ultra V. X-Ray Dif.		
B* 100 °C	1073.5	В	5	Viscosity	-	-	Infrared		<u> </u>
c	_			centistokes			Solubility in + Acetone		
t <sub>x</sub>   to				7 °	7		Carbon tet.		
A' to			$\vdash$		1		Benzene Ether		
B' °	<u>.</u>			B <sup>V</sup> to	-	<del> </del>	n-Heptane		
A¹* to	<del> </del>			B to			Ethanol Water		
B'* °C				(BV) to	<u>-</u>		Water in		-
Ac   111 to	7.1	122	5	(A <sup>V</sup> ) •c					
Bc tc_°C	1447.		5	cp liq. °K					
Cryos. A°				c <sub>p</sub> vap. °K	:				
te °C	85.1	3	5	c <sub>v</sub> vap.					
$T_{R} = 0.7$	1			1			grams/100 grai	ns solver	ıt
		)ow	2-AF	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			Al						
PURIFICAT			AI						
LITERATU	RE REF	ERE	NCES	<b>3:</b>					

NAME

3, 4-Dimethyl-1-pentene

271 No. 55 STRUCTURAL FORMULA

		T T							сн <sub>3</sub> сн сн сн=сн <sub>2</sub> сн <sub>3</sub> сн <sub>3</sub>			
Mole % Pur.		Ref.	Mo! For	ecula mula	C7H14		olecular eight 98.	182		с́н <sub>3</sub> с́н <sub>3</sub>	-	
				Ref.		Ť		Ref.	r			Ref.
F.P. °C F.P. 100%					dt/dP °C/mm				f g	to		
B. P. °C 760 mm 100 30 10		81. 24.1 -0.69 -19.3		2 4 4 5	25°C BP t <sub>e</sub> 30 mm		0.2208 0.0454 0.0382 0.6168	5 4 5 5	f' g' h'	to *K		
Pressure mm 25°C t <sub>e</sub>	,	04. 2 952. 2		5 5	ΔHv cal/g 25°C 30 mm BP		78.34 81.22 69.65	5 5 5	m n o	to *K		
Density g/ml 20°C dt 25 d <sub>4</sub> 30		0.70 0.69 0.69	7	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>		68.55 68.51 18.58	5 5	m' n' o'	to °K		
a b		0.71 -0.0 <sub>3</sub>		4 4	e   89 °c	C C	81.13 0.1417	5 5		face tension es/cm. 20°C 30 40	19.95	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30		1.39 1.39 1.39	69	2 2 4	e'   °c  d g/ml vc ml/g t °C	c	0.210 4.769 246.	5 5 5	Par	achor [P] 20°C 30	18.12	
"C"		0.75	99	4	P <sub>c</sub> mm	١,	17980.	5		40 Sugd.	296.2	5
MR (Obs.) MR (Calc. (nD-d/2)	)	33.9 34.05 1.04		2 5 2	PV/RT 25°C 30 mm	-	0.9907 1.0000	5 5		b. L.l.%/wt. u. persion	120.	2
Dielectric					BP	l	0.9520	5		sh Point °C	120.	+-
A -1 to B   116 °C C	112	6, 71 170, 33 224,		4 4 4	t <sub>e</sub> t <sub>c</sub> ΔHc kcal/m	n	0.9449	2	Fir M.	e Point Spec.		-
A*  -1 to B*  99 °C K	- 10	1,13 93,21	434	5 5	ΔHf ΔFf Viscosity	-		-	X-I Infr	Ray Dif.		
c t <sub>k</sub> to C A' to	-				centistokes 7 °C				Ac Ca Be	ubility in † etone arbon tet. enzene her		
B'   _ °C C'	+				B <sup>V</sup> to				n- Et Wa	Heptane hanol ater ater in		
B'* °C  Ac 116 to  Bc t <sub>c</sub> °C  Cc	14	7.13 172. 265.	70	5 5 <b>5</b>	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C c <sub>p</sub> liq. °P	С						
Cryos. A° consts. B°					c <sub>p</sub> vap. °F	к						
t <sub>e</sub> °C		88.99		5	c <sub>v</sub> vap.	$\perp$			L			
$\frac{T_{R} = 0.7}{REFERENC}$			ow	2-A	PI 3-Lit. 4	4-C	alc. from d	et. da		rams/100 gra -Calc. by for		nt
SOURCE:				A	PI							
PURIFICAT					PI							
LITERATU	RE	REF	EREI	NCES	<b>:</b>							

							No. 56	
NAME	4,4-Dim	ethyl	-1-pentene			STRUCTURAL	FORMULA	<b>L</b>
						СН3		
						CH <sub>3</sub> C CH <sub>2</sub> C	H=CH <sub>2</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 98.18	2	Ċн <sub>3</sub>		
	<del></del>	Ref.	Ī		Ref	<u> </u>		Ref.
F.P. °C	-136.600	2	dt/dP	I		f to		
F.P. 1009	6		°C/mm		_	g LK		
B. P. °C 760 mm	72.49	2	25°C BP	0.1663 0.0448	5 4	h		
100	16.43	4	t <sub>e</sub>	0.0384	5	f' to		
30 10	-7.87 -26.13	4 5	30 mm	0.6049	5	g'   'K_ h'		
1	-55.93	5	ΔHm cal/g			m   to		-
Pressure	144 (		ΔHv cal/g 25°C	73.95	5	n 'K_		
mm 25°C	144.6 928.4	5	30 mm	78.52	5	<u> </u>		
Density		_	BP te	67.25 66.30	5	m'   to		
g/ml 20°0	0.6827 0.6785	2 2	'e (u, e)	66.27	5	n'   °K-		
d <sub>4</sub> 25	0.6743	4	ΔHv/T <sub>e</sub>	18.46	5	Surface tension		$\vdash$
	0.6995	4	d   -15 to e   90 °C	77.42 0.1403	5	dynes/cm. 20°C	17.93	5
Ref. Index	-0.0381	4	d' - to	1		30 40	17.03 16.15	5
n <sub>D</sub> 20°0	1.3918	2	e'   °C		-	Parachor [P]		Ť
25 30	1.3892 1.3867	2 4	d g/ml vc ml/g	0.201 4.985	5	20°C		
"C"	0.7661	4	16 °C	230.	5	30 40		
MR (Obs.	<del></del>	2	P <sub>c</sub> mm	16735.	5		296.2	5
MR (Calc.	34.059	5	PV/RT 25°C	0.9880	5	Exp. L.1.%/wt.		
(nD-d/2) Dielectric	1.0504	2	30 mm	1.0000	5	Dispersion	120.	2
A -15 t		4	BP t <sub>e</sub>	0.9525	5	Flash Point °C		
B 1105°	C 1128.36	4	t <sub>c</sub>	0.261	5	Fire Point M Spec.		
C	225.	4	ΔHc kcal/m ΔHf	1043.65	2	Ultra V.		
A*  -15 to B* 90 °c		5 5	ΔFf			X-Ray Dif. Infrared		
K — — -	_	ļ	Viscosity			Solubility in +		
t <sub>k  </sub>		į	centistokes 7 °C			Acetone Carbon tet.		
'x '		L	•			Benzene		
A' to						Ether n-Heptane		
C'	<b>-</b>		B <sup>V</sup> to			Ethanol		
A'* to B'* **			$A^{\mathbf{v}} \mid - \mathbf{c}$			Water Water in		
Ac   105 to	<u> </u>	5	1					
Bc t <sub>c</sub> °	C 1421.	5	c <sub>p</sub> liq. °K					
Ce	265.	5		1				
Cryos. A'			c <sub>p</sub> vap. °K					1
te °C	79.47	5	c <sub>v</sub> vap.					
$T_R = 0.7$				•	•	+ grams/100 gran	ns solvent	
	CES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-Cafic, by form	nula	
SOURCE:			PI	<del> </del>				
PURIFICA	- · · · · · · · · · · · · · · · · · · ·		PI					
LITERATU	IRE REFEREI	NCES	<b>5:</b>					

							No. 5	7		
NAME _	3-Ethyl-2	-pen	tene	<del></del>		STRUCTURAL FORMULA				
						$CH_3CH_2C = C$	H=CH,			
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 98.182	2	Ċ <sub>2</sub> H <sub>5</sub>	,			
		Ref.			Ref.			Ref		
F.P. °C F.P. 100%			dt/dP °C/mm			f to g				
B. P. °C 760 mm 100 30 10 1	96. 01 37. 51 11. 98 -7. 27 -38. 83	2 4 4 5	25°C BP te 30 mm  AHm cal/g  AHv cal/g	0.3725 0.0465 0.0378 0.6369	5 4 5 5	h   to g'   to o'K h'   to o'K n   to o'K				
mm 25°C	57.19 993.9	5 5	25°C 30 mm BP	84.20 86.15 73.79	5 5	0		<u> </u>		
Density g/ml 20°C dt 25 4 30	0.7204 0.7159 0.7114	2 2 4	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	72.40 72.35 18.76	5 5 5	m' to n' °K o' Surface tension				
a b	0.7384 -0.0 <sub>3</sub> 89	4	d 12 to e 106 °C d to	87.92 0.1472	5	dynes/cm. 20°C 30 40	22.28 21.17 20.09	5 5		
Ref. Index n <sub>D</sub> 20°C 25 30	1.4148 1.4122 1.4094	2 2 4	e'   °C  d g/ml vc ml/g t °C	0.217 4.598 270.	5 5 5	Parachor [P] 20°C 30	20.07			
"C"	0,7662	4	t <sub>c</sub> °C P <sub>c</sub> mm	19361.	5	40 Sugd.	296. 2	5		
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	34.11 34.059 1.0546	2 5 2	PV/RT 25°C 30 mm BP	0.9967 1.0000	5 5 5	Exp. L.1.%/wt. u. Dispersion	127.	2		
A 5 to B 134 °C	6.77291 1233.83	4 4	t <sub>e</sub> t <sub>c</sub>	0.9510 0.9426 0.258	5	Flash Point C Fire Point		_		
C A* 5 to B* 116°C	1.17411 1154.43	4 5 5	ΔHc kcal/m ΔHf ΔFf	1044.30	2	M. Spec. Ultra V. X-Ray Dif. Infrared				
K c t <sub>k</sub> to tx C			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet. Benzene Ether				
B'   _ °C C'			$ \begin{array}{c c} B^{\mathbf{v}} & \text{to} \\ A^{\mathbf{v}} & & {}^{\mathbf{c}} \\ \hline (B^{\mathbf{v}}) & & {}^{\mathbf{c}} \end{array} $	-		n-Heptane Ethanol Water Water in				
Ac 134 to Bc t <sub>c</sub> °C Cc	7.1903 1548. 264.	5 5 5	(B')  to (A')  °C c <sub>p</sub> liq. °K							
Cryos. A° consts. B°			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	105.79	5	c <sub>v</sub> vap.			L		<u>L.</u>		
T <sub>R</sub> = 0.75		2 4	PI 3-Lit. 4-	Cala ( 1	د ه	grams/100 gra		ıt		
SOURCE:	ES: 1-Dow	Z-A AI		Calc. from de	t. da	ta 5-Calc. by for	muia			
PURIFICAT	ION:	AI								
	RE REFERE	_		-						

							No. 58	
NAME	•2,3-	Dimethyl	-2-pentene			STRUCTURAL	FORMUL	A
1_						сн <sub>3</sub> сн <sub>2</sub> с = с сн <sub>3</sub>	с сн3	
Mole % Pur.	Ref.	Molecul Formul	ar C <sub>7</sub> H <sub>14</sub>	Molecular Weight 98.18	12	CH <sub>3</sub>	CH <sub>3</sub>	
		Ref.		*************	Ref.			Ref.
F.P. °C	-118.3	2	dt/dP			f to		
F.P. 100%	<u> </u>		*C/mm 25*C	0.3878	5	g <u>  •K</u> _		l
B. P. *C 760 mm	97.46	. 2	BP	0.0470	4	h		<u> </u>
100	38.46	4	t <sub>e</sub>	0.0381	5	f' to		ļ
30 10	12.75		30 mm	0.6412	5	g'	4	l
ĭ	-38.37		∆Hm cal/g	<u> </u>		h'	ļ	—
Pressure	<b>†</b>		ΔHv cal/g	24.24	-	m to		1
mm 25°C	54.94		25°C 30 mm	84.24 86.03	5	0		1
t <sub>e</sub>	998.2	5	BP	73.68	5	m' to		$\vdash$
Density g/ml 20°C	0.72	77 2	te te (d, e)	72.27	5	n'  °K		1
at 25	0.72	34 2	ΔHv/T <sub>e</sub>	18.64	5	0'		1
<b>4</b> 30	0.71		d   5 to		5	Surface tension		
a b	0.74		115 °C		5	dynes/cm. 20°C	23.20	5
Ref. Index	1 -0.03	05 4	d'   to			30 40	21.03	5
n <sub>D</sub> 20°C	1.42	08 2		<b>'</b>	$\vdash$	Parachor [P]		1
25 30	1.41		d g/ml vc ml/g	1		20°C		]
"C"	1.41		tc °C	273.	5	30 40		1
	0.76		P <sub>c</sub> mm	19410.	5		296.2	5
MR (Obs.) MR (Calc.)	34.20 34.05		PV/RT			Exp. L. l. %/wt.		
(nD-d/2)	1.05		25°C 30 mm	0.9970 1.0000	5	u. Dispersion	130.	2
Dielectric			BP	0.9510	5	Flash Point °C	130.	+-
A 5 to			t <sub>e</sub>	0.9424	5	Fire Point		
B 1.137.℃	1233.55 221.	4 4	t <sub>c</sub> ΔHc kcal/m	1041 03	2	M Spec.		1
A*  5 to	1.15		ΔHf	1041.82	1	Ultra V.		ļ
B* 115 °C	1153.70		ΔFf			X-Ray Dif. Infrared		
K	İ	- 1	Viscosity	1		Solubility in +		
t <sub>k</sub>   to	1		centistokes 7 °C			Acetone		
t <sub>x l</sub>	<u> </u>		•	1		Carbon tet. Benzene		
A' to B' C	1					Ether		
č, – – <u>–</u>	1		B <sup>V</sup>   to			n-Heptane Ethanol		1
A'* to			AV I °C	j		Water		1
Bi* °C	<del> </del>		(B <sup>V</sup> ) to	1		Water in		+
Ac   137 to Bc   t °C		26 5	(A <sup>V</sup> )   °C					
Bc _ tc_°C	264.	5	c <sub>p</sub> liq. °K	1		1		
Cryos. A°			c <sub>p</sub> vap. °K					
consts. B°	107.55	_   _	c <sub>v</sub> vap.					
t <sub>e</sub> °C	107.50	5	A	<u> </u>		+ ~~~ (100	<u></u>	
REFERENC		ow 2-AT	PI 3-1.it 4-6	Calc from de	+ da	grams/100 grants  ta 5-Calc. by for		it
SOURCE:			PI	<u> ue</u>	ua	- J-Carc. by for		
PURIFICAT	ION:		PI					
LITERATU								

							<b>No.</b> 5	9
NAME	2, 4-Dim	ethyl	-2-pentene			STRUCTURAL	FORMUI	A
ľ						CH C - CHC	ים כם	
Mole % Pur.		lecul		Molecular Weight 98.18	2	сн <sub>3</sub> с = сн с сн <sub>3</sub> с	H <sub>3</sub>	
•		Ref.	,	1	Ref.			Ref
F. P. °C			dt/dP			f to		
F.P. 100%			°C/mm	I		g  ° <u>K</u>		
B. P. °C			25°C BP	0.2398 0.0456	5 4	h	İ	
760 mm 100	83.44 26.25	2 4	t <sub>e</sub>	0.0381	5	f' to		Ì
30	1.39	4	30 mm	0.6196	5	g' <u>*K</u>		
10 1	-17.33 -47.94	5	∆Hm cal/g			h'	<u> </u>	<u> </u>
Pressure	1	Ť	ΔHv cal/g			m to		
mm 25°C	94.68	5	25°C 30 mm	78.72	5	"   ' <u>r</u>		
t <sub>e</sub>	959.2	5	BP	82.09 70.30	5	m' to		┼
Density	0 (055		te,	69.15	5	n' l °K		1
g/ml 20°C	0.6955 0.6912	2	te (a, e)	69.11	5	0'	1	
d <sub>4</sub> 25 30	0.6869	4	ΔHv/T <sub>e</sub>	18.60	5	Surface tension		<del> </del>
a	0.7126	4	d -10 to e 105 °C	82.29 0.1437	5 5	dynes/cm. 20°C	19.34	5
ь	-0.0384	4	a to	0.1451		30 40	18.38 17.44	5
Ref. Index		2	e' ºC		$\sqcup$		17.44	+
<sup>n</sup> D 20°C	1.4013	2	d <sub>c</sub> g/ml	0.208	5	Parachor [P] 20°C		
30	1.3986	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.81	5	30		
"C"	0.7741	4	P <sub>c</sub> mm	17809.	5	40 Sugd	296.2	5
MR (Obs.)		2	PV/RT	1	H	Exp. L.1.%/wt.	270.2	+
MR (Calc. (nD-d/2)	34.059 1.0562	5 2	25°C	0.9930	5	u.		
Dielectric	1.0302	Ė	30 mm BP	1.0000 0.9520	5	Dispersion	128.	2
A -10 to	6.71947	4	t	0.9447	5	Flash Point C Fire Point		
B 1118 °C	_ 1176.32	4	<sup>t</sup> c	0.259	5	M. Spec.		+
C	223.	4	ΔHc kcal/m ΔHf	1041.34	2	Ultra V.		1
A* -10 to B* 105 °C	1.13362	5	ΔFf			X-Ray Dif. Infrared		
к — <u>-</u> -	-		Viscosity			<u> </u>	<del> </del>	<del> </del>
t	-		centistokes 7°C			Solubility in TACetone	į	
t <sub>k</sub> to			η °C			Carbon tet.		
A'   to		<b>—</b>				Benzene Ether		
B'   _ °C	<u>-</u>		B <sup>V</sup> to		+	n-Heptane		
		-	B to			Ethanol Water		
A'* to B'* °C	:		(B <sup>V</sup> )  to	1		Water in		
Ac   118 to	7.1369	5	(A <sup>V</sup> )  °C					
Bc tc °C	1478.	5	c <sub>p</sub> liq. °K		$\Box$		1	
Cc	264.	5	<b>{</b> }				1	1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	91.73	5	c <sub>v</sub> vap.	I			L	
$T_{R} = 0.7$						grams/100 gra		nt
	CES: 1-Dow	2-A		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		AI						
PURIFICA		AI						
LITERATU	RE REFERE	NCES	5:					

								No. 60	
NAME	3,4	-Dime	thyl-	-cis-2-pentene			STRUCTURAL 1	FORMULA	4
							сн <sub>3</sub> сн с = 0	сн сн.	
Mole	Ref	Mo	lecul	27	Molecular		ċн <sub>3</sub> ċн <sub>3</sub>	3	
% Pur.			rmul		Weight 98.18	2			
			Ref.			Ref			Ref
F.P. °C			$\sqcup$	dt/dP			f to		
F. P. 1007 B. P. °C	<u>'</u>		$\vdash$	*C/mm 25*C	0,2711	5	g <u>*K</u> _		
760 mm	87.		2	BP	0.0459	4	h		⊢
100 30	29.4 4.3		4 4	t <sub>e</sub> 30 mm	0.0380	5	f' to		
10	-14.5		5	ΔHm cal/g	0.6251	-	h'		
1	-45.4		5	ΔHv cal/g	<del> </del>	$\vdash$	m to		
Pressure mm 25°C	82.2	8	5	25°C	80.23	5	n °K		
t <sub>e</sub>	969.4		5	30 mm BP	83.15	5 5			ـــ
Density		1 2		l t_	70.12	5	m'   to		
g/ml 20°0	0.7		2 2	t <sub>e</sub> (d, e)	70.09	5	o'   ' = -		
d <sub>4</sub> 25	0.7	05	4	ΔHv/T <sub>e</sub>	18.66	5	Surface tension		
a b	0.7		4	d   -5 to		5	dynes/cm. 20°C	21.37	5
Ref. Index		310	-	d' to			30 40	20.40 19.45	5 <b>5</b>
n <sub>D</sub> 20°0	1.4		2		0,216	5	Parachor [P]		
25 30	1.4		2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	4.63	5	20°C		İ
"C"	0.7		4		256.	5	40		
MR (Obs.	33.9		2	P <sub>c</sub> mm	18790.	5		296.2	5
MR (Calc. (nD-d/2)	34.0		5 2	25°C	0.9944	5	Exp. L.1.%/wt. u.		
Dielectric			┝╧╢	30 mm BP	1.000 <b>0</b> 0.9520	5 5	Dispersion	127.	2
A   -5 t		4284	4	te	0.9444	5	Flash Point °C Fire Point		
B (_124 •9			4	tc	0.259	5	M Spec.		┢
A*  -5 to		<b>5</b> 215	5	ΔHc kcal/m ΔHf	1042.02	2	Ultra V.		
B* 105 °	1118.9		5	ΔFf		Ш	X-Ray Dif. Infrared		
K C				Viscosity centistokes			Solubility in +		
the   Te				η °c			Acetone Carbon tet.		
tÇ i °C			$\sqcup$				Benzene		
B'i °							Ether n-Heptane		
C'				B <sup>V</sup>   to A <sup>V</sup>   °C	İ	1 1	Etha nol		
A'* to					-		Water Water in		
Ac   124 to	7.1	618	5	(A <sup>V</sup> )  to	1				
Bc t *	C 1506.		5	c <sub>p</sub> liq. °K	·	$\vdash$			
Cryos. A	265.		5	11 -	1				
consts. B				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	95.7	2	5	c <sub>v</sub> vap.					
$T_R = 0.7$							f grams/100 gran	ns solven	t
REFEREN	CES: 1-I	Dow	2-AF		Calc. from de	t. dat	ta 5-Calc. by form	nula	
SOURCE:				PI					
PURIFICA				PI	· · · · · · · · · · · · · · · · · · ·				
LITERATU	KE REF	ERE	CES	<b>5:</b>					

No. 61 43, 4-Dimethyl-trans-2-pentene NAME STRUCTURAL FORMULA CH3CH C = CHCH3 Ċн<sub>3</sub>Ċн<sub>3</sub> Ref. Molecular Mole Molecular C7H14 Weight 98.182 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 0.2711 5 B.P. °C h BP 0.0459 87. 760 mm 2 0.03801 5 f١ t<sub>e</sub> to 100 29.43 4 g' <u>°К</u> 30 4.36 5 30 mm 0.6251 5 10 -14:53 4 h' ∆Hm cal/g 4 -45.44 m to AHv cal/g Pressure °K n 25°C 80.23 5 mm 25°C 82.28 4 o 30 mm 83.15 5 969.36 te 4 BP 5 71.33 m' to Density 70.12 5 n' °K g/ml 20°C te (d, e) 0.713 2 70.09 5 o' ď4 25 0.709 2 AHv/T 18.66 5 30 0.7050 4 Surface tension d -5 83,78 5 to a 0.7290 4 dynes/cm. 20°C 21.37 5 ᇷᅴ <u>96</u> °C 0.1430 5 ь -0.0378 4 30 20.40 to 19.45 5 40 Ref. Index e' 20°C [P] <sup>n</sup>D 1.407 2 Parachor d<sub>c</sub> g/ml 0.216 5 25 1.404 20°C vc ml/g t °C 5 4.633 30 1,402 4 30 t<sub>c</sub> 256. 5 40 "C" 0.7604 5  $P_c$  mm 18790. 5 Sugd. 296.2 5 MR (Obs.) 33.9 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9939 5 (nD-d/2)1,050 2 u. 30 mm 1.0000 Dispersion 127. 5 2 Dielectric ВP 0.9520 5 Flash Point C 0.9444 5 A -5 to 6.74284 Fire Point 0.259 5 B | 124 °C 1197.23 M. Spec. Ultra V. С 4 223. AHc kcal/m 1042.02 2 ΔHf A\* -5 to 5 1,15215 X-Ray Dif. ΔFf B\* 105 °C 1118,91 Infrared ĸ Viscosity Solubility in centistokes Acetone to t<sub>k</sub> Carbon tet. °C Benzene ۸' to Ether В' °C n-Heptane  $\mathbf{B}^{\mathbf{v}^{\top}}$ C' Ethanol 'v | °C Water A'\* to Water in B'\* °C (B<sup>V</sup>) to Ac | 124 to 7.1618 5 (A<sup>V</sup>)| °C Bc \_tc\_ 1505.5 °C 5 c<sub>p</sub> liq. °K Cc 265.11 5 Cryos. Aº c<sub>p</sub> vap. °K consts. B° c, vap. te °C 95.72 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

			. 41 1	-i- 2		1			No. 62	
NAME		Dime	tnyı	-cis-2-pentene	:		STRUC	TURAL 1	r OR MUL.	A
								СН <sub>3</sub>		
Mole	Ref.	Ma	lecul		Molecular	1	C	н <sub>3</sub> ċ сн=0	CH CH <sub>3</sub>	
% Pur.	Ker.	Fo	rmul	ar C <sub>7</sub> H <sub>14</sub>	Weight 98.18	82		Ċн <sub>3</sub>		
			Ref.			Ref				Ref
F.P. °C	-135.46	,	2	dt/dP	T		f	to		
F.P. 100%				*C/mm	1	1.	g	<u>•</u> K_		
B. P. °C				25°C BP	0.2165 0.0454	5 4	h .		1	
760 mm 100	80.42 23.53		2	t	0.0382	5	f'	to		T
30	-1.19	)	4	30 mm	0,6160	5	g'	•K_		
10 1	-19.80		5	AHm cal/g			h'			
	-50, 21		2	ΔHv cal/g		t	m	to		
Pressure mm 25°C	106.6		5	25°C	77.16	5	n	<u>_ K</u> _	l	
t <sub>e</sub>	950.6		5	30 mm BP	81.03 69.46	5				
Density				t_	68.37	5	m'	to		
g/ml 20°C	0.69		2 2	te (d, e)	68.34	5	n' o'	•K_		
dt 25 4 30	0.69		4	AHv/T <sub>e</sub>	18.57	5				-
a	0,71		4	d   -10 to		5		tension cm. 20°C	19.79	5
ь	-0.0		4	-a,  -88 %		5	3,	30	18.79	5
Ref. Index				e' i				40	17.80	5
n <sub>D</sub> 20°C	1.40		2 2	d g/ml v ml/g	0,208	5	Parach	or [P] 20°C		1
30	1.39		4	vc ml/g	4.819	5		30		
"C"	0.76	67	4	-	244.	5		40		_
MR (Obs.)	34.20	)	2	P <sub>c</sub> mm	17725.	5	<u> </u>	Sugd.	296.2	5
MR (Calc.)			5 2	25°C	0.9915	5	Exp. L	.1.%/wt. u.		
(nD-d/2)	1.05	20	-	30 mm	1.0000	5	Disper		122.	2
Dielectric	<del> </del>		-	BP t <sub>e</sub>	0.9520 0.9450	5		Point °C		
A   -10 to B   115 °C			4	tc	0. 260	5	Fire Po	oint		ـــــ
c ——	224.		4	ΔHc kcal/m	1042.05	2	M Spec			
A* -10 to	1.13	040	5	ΔHf ΔFf			Ultra V X-Ray			
B*	1089.71		5	<del></del>		<del>-</del>	Infrare	d		
c				Viscosity centistokes			Solubili			
tk   to				η •	;		Acetor Carbo		ĺ	1
х '			$\sqcup$		1	1	Benze			
A'   to B'   °C							Ether n-Hep	tane		1
<u>c'</u>				B <sup>V</sup> l to			Ethano	ol		
A¹* to				A <sup>V</sup> O	<u>:</u> ]		Water Water			
B'* °C	ļ		Ш	(B <sup>V</sup> ) to	•		water	ın		+
Ac   115 to Bc   t °C	7.13	19	5	(A <sup>V</sup> )  °C	;					
Cc	265.		5	c <sub>p</sub> liq. ∘K	:				İ	
Cryos. A°				c <sub>p</sub> vap. °K						1
consts. B°	<u> </u>			· -		1			ļ	
t <sub>e</sub> °C	88.34		5	c <sub>v</sub> vap.	}					
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>						+ gram	s/100 gran	ns solven	ıt
REFERENC	ES: 1-D	ow	2 - AF	PI 3-Lit. 4-	Calc. from de	t. dat				
SOURCE:			Al	PI						
PURIFICAT	ION:		A1	PI						
LITERATUE	RE REF	CREN	ICES	:						

No. 63 4, 4-Dimethyl-trans-2-pentene NAME STRUCTURAL FORMULA снзс сн=снсн3 Molecular C7H14 Mole Ref. Molecular Ċн<sub>3</sub> % Pur Weight 98.182 Ref. Ref. F.P. °C F.P. 100% -115.235 2 dt/dP f to °C/mm <u>°K</u> g 25°C 0.1913 5 B. P. °C h BP 0.0451 4 760 mm 76.75 2 t<sub>e</sub> 0.0383 5 f† 100 to 20.24 4 4 g' °<u>K</u> 30 -4.27 30 mm 0.6105 10 -22.7 5 h' ∆Hm cal/g 1 -52.8 5 to AHv cal/g Pressure °K n 25°C 75.80 5 5 mm 25°C 122.9 o 30 mm 79.93 940.6 5 t<sub>e</sub> BP 68.43 5 m to Density te te (d, e) 67.40 g/ml 20°C 5 0.6889 2 67.37 ٥' 0.6845  $\mathbf{d_{4}^{t}}$ 25 2 ΔHv/T<sub>e</sub> 18.51 5 30 0.6801 4 Surface tension -10 to 79.33 5 0.7066 4 a dynes/cm. 20°C 18.60 ᇷᅱ 95 0.1420 5 Ъ -0.0386 4 17.64 5 30 to 5 40 16.69 Ref. Index e' 20°C 1.3982 2 <sup>n</sup>D [P] Parachor d<sub>c</sub> g/ml 5 0.203 25 1.3953 2 20°C vc ml/g t °C 4.928 5 30 4 1.3928 30 <sup>t</sup>c 237. 5 40 "C" 0.7709 4 P<sub>c</sub> mm 17097. 5 Sugd. 296.2 5 MR (Obs.) 34.41 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 34.059 5 25°C 0.9908 (nD-d/2)1.0538 2 30 mm 1.0000 5 124. 2 Dispersion Dielectric ВP 0.9525 5 Flash Point C A -10 to B | 110 °C 0.9458 6.68799 4 Fire Point 0.260 5 1145.0 4 M. Spec. С 224. 4 ∆Hc kcal/m 1041.05 Ultra V. AHf A\* -10 to 1.10939 5 X-Ray Dif. ΔFf **B**\*|\_95 °C 1069.0 Infra red ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. °C t^ Benzene A١ to Ether В' °C n-Heptane BV AV C' to Ethanol °C Water Water in B'\* °C (BV) to Ac 110 to 7,1055 5 5 (AV) °C Bc tc °С 1440. c<sub>p</sub> liq. °К 5 265. Cryos. A° c<sub>p</sub> vap. °K consts. Bo c<sub>v</sub> vap. 5 84, 25  $= 0.75 T_{\rm c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 64	
NAME	2-Et	hyl-3-me	thyl-l-butene			STRUCTURAL		
-			,		$\dashv$	C <sub>2</sub> H		-
			<del></del>			сн <sub>з</sub> сн с =		
Mole	Ref.	Molecul	27	Molecular	- 1		CH <sub>2</sub>	
% Pur.	Mer.	Formul	ar C <sub>7</sub> H <sub>14</sub>	Weight 98.18	32	ĊН <sub>3</sub>		
		Ref.			Ref			Ref
F.P. °C	1		dt/dP	T		f   to		t
F.P. 100%	<del>                                     </del>		*C/mm			f to		1
B, P, *C	†		25°C	0.2906	5	h	1	
760 mm	89.	2	BP	0.0460 0.0380	4 5	f' to	<del> </del>	$\vdash$
100 30	31.23		t <sub>e</sub> 30 mm	0.6272	5	g'   ' °K		
10	-12.9	5		0.0272	<del>                                     </del>	h'	1	1
1	-43.9	5	ΔHm cal/g		<del> </del>	m   to	<b></b>	<del>                                     </del>
Pressure			ΔHv cal/g 25°C	81.09	5	n '*K	_	1
mm 25°C	75.95 975.1	5 5	30 mm	83.90	5	•		•
Pomeite	713.1		BP	71.87	5	m'   to		
Density g/ml 20°C	0.71	5 2	te te (d, e)	70.61	5	n' 'K		ľ
<sub>a</sub> t 25	0.71		e (a, c)	1	5	o'		
<sup>4</sup> 4 30	0.70		ΔHv/T <sub>e</sub>	18.68		Surface tension	†	
a	0.73		d   -5 to		5	dynes/cm. 20°C	21.61	5
Ъ	-0.03	8 4	1-a,; 65			30 40	20.64 19.68	5
Ref. Index	1 , 4,	.0 2	e' i °C			l	17.00	1-
n <sub>D</sub> 20°C	1.41		d <sub>c</sub> g/ml	0.216	5	Parachor [P] 20°C	ŀ	
30	1.40	5 4	V_ mi/g	4.624 259.	5	30	1	1
"C"	0.76	35 4	11 -	ł	1 1	40		_
MR (Obs.)	34.0	2	P <sub>c</sub> mm	18932.	5	Sugd	296.2	5
MR (Calc.	34.05		PV/RT 25°C	0.9949	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.05	2 2	30 mm	1.0000	5	Dispersion	125.	2
Dielectric			BP	0.9520	5	Flash Point °C	ļ	<del> </del>
A -5 to			te te	0.9442 0.259	5	Fire Point		[
B <u>1.125 °C</u> C	1200.64	4	t <sub>c</sub> ΔHc kcal/m	1043, 42	2	M Spec.		
A*  -5 to	1.14		AHf	1045.42	-	Ultra V.	j	1
B* 110 °C			ΔFf			X-Ray Dif. Infrared		
к — — —	1	i	Viscosity			Solubility in +	<del> </del>	一
t <sub>k</sub>	-	}	centistokes 7°C		1 1	Acetone	į.	
tx c	: ]	ł	l 1			Carbon tet.	ļ	
A' to	†	<del></del>	1		1	Benzene Ether	Ì	1
B' L _ *	<u>.</u>	ŀ	B <sup>V</sup>   to	<del> </del>	1-	n-Heptane	ł	
C'	<del>                                     </del>		B <sup>V</sup> to A <sup>V</sup> C	1		Ethanol.		1
A'* to B'* °C	.1	[		-		Water Water in		
Ac   125 to		91 5	1					1
Bc t °C	1509.	91   5	(A <sup>V</sup> )  °C	<b></b>	$\perp$			]
Cc	264.	5	c <sub>p</sub> liq. °K					1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t, °C	97.97	5	c <sub>w</sub> vap.					
$T_{R} = 0.7$			<u> </u>	L		+ arame/100 c==	ma acles	
REFERENC		ow 2-A1	PI 3-Lit. 4-(	Calc from de	+ 4-	grams/100 gra		<u>.                                    </u>
SOURCE:	D		PI	Jane, Moni de	ua	ta 5-Calc. by for	u.a	
PURIFICAT	'ION:							
LITERATU			PI					
	ALFI	MENCES	··					

								No. 6	5
NAME _	2, 3, 3-Tr	imet	hyl-1-butene		STRU	CTURAL	FORMUI	LA	
					}		CH <sub>3</sub>		
Mole % Pur.	Ref. Mo	lecul rmuk		Molecular Weight 98.1	82	C	н <sub>3</sub> с с = сн <sub>3</sub> сн		
	1	Ref.			Ref.				Ref.
F. P. °C	-109.85	2	dt/dP			£	to		
F.P. 100%	<del>                                     </del>	├—	°C/mm 25°C	0.1987	5	g	<u>•</u> K		1
B. P. °C 760 mm	77.87	2	BP	0.0452	4	h			<del> </del>
100 30	21.26	4	t <sub>e</sub>	0.0383	5	f' g	to •K		
10	-3.31 -21.79	5	30 mm	0.6121	5	h'			
1	-51.98	5	ΔHm cal/g	<del> </del>	+	m	to		+
Pressure mm 25°C	117 4	5	ΔHv cal/g 25°C	76, 23	5	n	•K		1
t <sub>e</sub>	117.6 943.3	5	30 mm	80.29	5	°			⊥
Density			BP te	68.75 67.70	5 5	m'	to		
g/ml 20°C		2	e (4, 6)	67.67	5	n'   o'	_ <u>•</u> K		1
dt 25 4 30	0.7005 0.6960	2	ΔHv/T <sub>e</sub>	18.53	5				+
a	0.7230	4	d -10 to	79.82	5		tension cm. 20°C	20.40	5
ь	-0.0388	4		0.1422	"	8,	30	19.35	5
Ref. Index	1.4029	2	e' °C		1	D	40	18.32	5
D 25	1.4000	2	d <sub>c</sub> g/ml	0.209 4.789	5 5	Parach	20°C		
30	1.3973	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	241.	5		30 40		
"C"	0.7617	4	P <sub>c</sub> mm	17662.	5			296.2	5
MR (Obs.) MR (Calc.)	33.98 34.059	<b>2</b> 5	PV/RT			Exp. L	. 1.%/wt.		
(nD-d/2)	1.0504	Ž	25°C 30 mm	0.9903	5 5	Disper	u.	124.	2
Dielectric			BP	0.9520	5		Point °C	121.	+-
A -10 to B   115 °C	6.69701 115 <b>2.</b> 0	4	te tc	0.9452 0.259	5	Fire P			1
c (113 C)	224.	4	ΔHc kcal/m	1042,08	2	M. Spe			
A* -10 to	1.11770	5	ΔHf ΔFf			Ultra V X-Ray			
B*[_95 °C K	1075.8	5	Viscosity	+	+-1	Infrare	d		
c			centistokes			Solubil:			
t <sub>k</sub>   to t_ ⊢ °C			η <b>°</b> C			Carbo		ł	
t <sub>x</sub> °C A'   to		<del> </del>				Benze Ether	ne		
B' i°C	.	i i	B <sub>v</sub> to	<del> </del>	+	n-Hep		İ	1
	<b>+</b>	-	B to			Ethan Water			
A'* to B'* °C			(B <sup>V</sup> )  - to	-		Water			
Ac 115 to	7.1153	5	(A <sup>V</sup> )  °C						
Bc tc °C	1450. 265.	5 <b>5</b>	c liq. °K	1	$\top$	ĺ			
Cryos, A°	203.	1	P	1					
consts, B°			P						
t <sub>e</sub> °C	85.48	5	c <sub>v</sub> vap.	1				1	1
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>					† gram	s/100 gra	ms solve	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Ca	lc. by for	mula	
SOURCE:		AF	PI .						
PURIFICAT		AF							
LITERATU	RE REFERE	NCES	<b>5:</b>		•				

										No. 66	
NAME	1-00	tene						STR	UCTURAL 1	FORMULA	
Mole % Pur.	Ref.	Mo Fo	lecul	ar C <sub>8</sub> H <sub>16</sub>		Molecular Veight 112.20	08	С	н <sub>3</sub> (сн <sub>2</sub> ) <sub>5</sub> сн	=CH <sub>2</sub>	
			Ref.				Ref				Ref.
F. P. °C	-101.73	16	2	dt/dP				f	to		
F.P. 100%				°C/mm			_	g	°K		
B. P. °C	121 20			25°C BP		1.0436 0.04711	5 4	h	1	•	
760 mm 100	121.28		2 2	t <sub>e</sub>		0.03622	5	f'	to		
30	35.33	,	4	30 mm	- [	0.6583	5	g'	°K		
10 1	15.38 -17.5	3	5	ΔHm cal/g	g			h'			
Pressure	+		+	AHv cal/g				m	300 to 600 °K	0.0250	
mm 25°C	17.38	,	5	25°C	ı	86.82 85.36	5	n o	800 K	0.0013 -0.0 <sub>6</sub> 49	
t <sub>e</sub>	1064.		5	30 mm BP	- 1	72.70	5		700 .		
Density	0.71	402	,	te (d.e)	- 1	70.95	5	m'	700 to	0.1070	4
g/ml 20°C	0.71 0.71		2 2	e (4, 5)	ŀ	70.86	5	ان		-0.0638	
d <sub>4</sub> 25 30	0.70		4	ΔHv/T <sub>e</sub>		19.56	5	5,, +6.	ace tension		-
a	0.73		4	d   25 e   134	to	90.56 0.1473	5		s/cm. 20°C	20.79	5
_ b	-0.03	809	4	_a,	to	0.1473		8	30 40	19.86 18.94	5
Ref. Index		870	2	e' i	°C			B		10.74	-
<sup>n</sup> D 25	1.40		2	d <sub>c</sub> g/ml	I	0.240 4.173	5	Para	chor [P]		
30	1.40	376	4	t <sub>c</sub> *C	l.	292.	5		30		
"C"	0.76	13	4	P <sub>c</sub> mm		19197.	5		40 Sugd.	335.2	5
MR (Obs.)			2	PV/RT	$\dashv$	-,-,-	H	Evn	L. 1. %/wt.		-
MR (Calc. $(nD-d/2)$	38.67		5	25°C		1.0000	5	_	u.		
Dielectric	+		Ė	30 mm BP	- 1	1.0000 0.9500	5		ersion	116.8	2
A   0 to	6, 93	263	2	t_		0.9396	5		h Point °C Point		
B [151 °C	1353.5		2	tc		0.255	5	<b></b>			
<u> </u>	212.76		2	ΔHc kcal/i	m	1194.97	2	M Sp Ultra		<u> </u>	
A* 25 to B* 140 °C	1.37	023	5	ΔFf					ay Dif.		
K Line			"	Viscosity			$\vdash$	Infra			
·	_			centistoke	8	0.000			oility in + tone		
t <sub>x</sub>   to			1	7 40	°C	0.382 0.316	2 2	Car	bon tet.		
A' to	<del>,  </del>		-	80	- 1	0.270	2	Ben Eth	zene		ļ
B'			į į	100		0.234	2		leptane		
C'			ļ.,	B <sup>V</sup>   20 A <sup>V</sup>   60		429.70 2.20988	5 5	Eth Wat	anol		
A'* to B'* °C				(BV) -60		409.49	5		er ter in		
Ac   151 to	<b>——</b>	70	5	(A <sup>V</sup> ) 110		2,27184	5				
Bc t °C	1664.		5			2,21104	۲				
Cc	252.		5	•	°K						1
Cryos, A° consts, B°				c <sub>p</sub> vap.300	°K	0.38117 0.48045	2 2				
t <sub>e</sub> °C	133.78		5	c <sub>w</sub> vap.	l	•					
$T_{R} = 0.75$	T <sub>c</sub>			L				+ gra	ms/100 grai	ns solven	 t
		ow	2-AF	PI 3-Lit.	4-C	alc. from det	. dat				
SOURCE:			PI								
PURIFICAT	TION:	A	PI		_						
LITERATU	RE REF	ERE	NCES	3:							

No. 67 cis-2-Octene NAME STRUCTURAL FORMULA CH3(CH2)4CH=CHCH3 Molecular C8H16 Mole cular Mole Ref. Weight 112.208 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% -100.2 2 dt/dP f to °C/mm ۰ĸ g 25°C 1.1514 B. P. ℃ h ΒP 0.04870 5 760 mm 125.64 2 5 te 0.03714 ſ١ to 100 64.12 4 g¹ 30 37.93 4 30 mm <u>°К</u> 0.6758 5 10 16.52 5 h! ∆Hm cal/g 1 -17.84 5 m to AHv cal/g Pressure n °K 25°C 85.12 mm 25°C 16.067 5 o 30 mm 5 84.10  $\mathbf{t_e}$ 1076.47 5 BP 71.86 5 m' Density to 5 70.08 te (d, e) n' <u>°K</u> g/ml 20°C 0.7243 2 70.01 5 ۰' 25 d<sub>4</sub> 0.7201 ΔHv/T<sub>e</sub> 19.08 5 30 0.7159 4 d Surface tension 38 89.23 to 5 5 0.7411 -0.0<sub>3</sub>84 44 dynes/cm. 20°C 21.91 1 140 °C 0.1382 ъ ā 30 20.90 5 20 to 87.21 40 19.93 1 Ref. Index e' 5 38 0.0838 20°C 1.4150 2 [P]  $^{n}D$ Parachor d<sub>c</sub> g/ml 0.25 5 25 1.4125 2 20°C 335, 21 4 5 vc ml/g t °C 4.006 30 1.4099 4 30 335.23 4 <sup>t</sup>c 300. 5 "C" 40 335.21 4 0.7624 4  $P_c$  mm 18399. 5 Sugd 335.2 5 MR (Obs.) 38.794 38.677 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 25°C 1.0000 1.0529 2 (nD-d/2) 30 mm 1.0000 5 Dispersion 2 118. Dielectric BP 0.9495 5 Flash Point C t<sub>e</sub> 0.9386 5 A 38 to 6.87711 Fire Point 0.254 5 B 1157 °C 1361.3 M. Spec. C 215. 4 ∆Hc kcal/m Ultra V ΔHf 1.30761 1277.39 A\* 38 to X-Ray Dif. ΔFf B\*[150 °C Infra red ĸ Viscosity Solubility in centistokes Acetone to °C °C Carbon tet. t<sub>x\_</sub> Benzene 20 to 7, 26268 Ether 38\_**°C** 1562.64 n-Heptane B<sub>v</sub> | C' 5 233. to Ethanol °C A!\* 20 to 1.65772 Water B'\* 38 °C Water in  $(\mathbf{B}^{\mathbf{v}})$ 1463.16 5 to Acl 157 to 7.28738 (AV) 5 °C Bc tc °C 1683.60 c<sub>p</sub> liq. °K Cc 256.69 5 Cryos. A° consts. B° c<sub>p</sub> vap. °K te °C c, vap. 139.03 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 68	
NAME	trans-2-	Octer	ıe			STRUCTURAL 1	FORMULA	L
						СН <sub>3</sub> (СН <sub>2</sub> ) <sub>4</sub> СН	=CHCH.	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	208	31-31-21-4	3	
		Ref.			Ref.			Ref
F.P. °C	-87.7	2	dt/dP			f to		
F. P. 100%	ļ		*C/mm 25*C	1.1227	· <sub>5</sub>	g <u>*K</u> _		
B. P. °C 760 mm	125.0	2	BP	0.0487	4 5	h f' to		-
100 30	63.55	4	t <sub>e</sub> 30 mm	0.0372	5	g' to		
10	16.02	5	ΔHm cal/g	0.0740	<del>                                     </del>	h'		
1	-18, 28	5	ΔHv cal/g		┢	m to		
Pressure mm 25°C	16.52	5	25°C 30 mm	84.90	5 5	n •K		
t <sub>e</sub>	1075.	5	BP	83.94 71.69	5	m'   to		
Density g/ml 20°C	0.7199	2	te te (d, e)	69.92 69.84	5	m'   to		
dt 25	0.7157	2	ΔHv/T <sub>e</sub>	19.06	5	o'		L_
4 30	0.7115	4	d   37 to	<u> </u>	5	Surface tension	21 20	_
ь	-0.0384	4	e 1 138 °C		5	dynes/cm. 20°C	21.38 20.39	<b>5</b>
Ref. Index			e' 37 °C		5	40	19.44	5
n <sub>D</sub> 20°C	1.4132	2 2	d g/ml v ml/g	0.225	5	Parachor [P] 20°C		
30	1.4081	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.44 299.	5	30 40		İ
"C"	0.7639	4	Pcmm	18206.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.)	38.88 38.677	2 5	PV/RT		1_	Exp. L.1.%/wt.		
(nD-d/2)	1.0533	2	25°C 30 mm	1.0000	5	u. Dispersion	120.	2
Dielectric	<u> </u>		BP	0.9495 0.9386	5	Flash Point °C		Ť
A 37 to B 156 °C		4	t <sub>e</sub> t <sub>c</sub>	0.254	5	Fire Point		_
<u>c — — — — — — — — — — — — — — — — — — —</u>	215.	4	ΔHc kcal/m		Ī	M Spec. Ultra V.		
A* 37 to B* 148 °C	1.30491	5	ΔHf ΔFf		l	X-Ray Dif.		1
к ———	12.3.0		Viscosity			Infrared Solubility in +		┢
t <sub>k</sub>	-		centistokes 7 °C			Acetone		
'x						Carbon tet. Benzene		
A'   25 to B' ∟ 37 °C	7.25977 1558.8	5				Ether n-Heptane		i
C'	233.	5	B <sup>V</sup>   to	1		Ethanol		
A'* 25 to B'* 37 °C	1.66269	5 5		-[		Water Water in		İ
Ac   156 to	7, 2835	5	(A <sup>V</sup> ) to					
Bc tc_°C	1678. 256.	5	c <sub>p</sub> liq. °K	<u> </u>	$\vdash$	1		
Cryos, A*	230.	,	c <sub>p</sub> vap. °K					
consts. B°			P					
t <sub>e</sub> °C	138.31	5	c <sub>v</sub> vap.		<u> </u>	<u> </u>		L
T <sub>R</sub> = 0.75		•				† grams/100 gran		t
SOURCE:	ES: 1-Dow		PI 3-Lit, 4-0 PI	Calc. from de	t. da	ta 5-Calc, by for	nula	
PURIFICAT	ION:		PI					
	RE REFERE							

							No. 6	9
NAME	cis-3-	Octene				STRUCTURAL	FORMUL	A
					}	ou (ou ) ou (		,
Mole % Pur.		Molecul Formula		Molecular Weight 112.2	08	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH=0	HCH <sub>2</sub> CH	3
	<del></del>	Ref.			Ref.	<del>                                     </del>		Ref
F. P. ℃			dt/dP			f to		
F.P. 1009	•		°C/mm 25°C		1 - 1	g  ° <u>K</u>		
B. P. °C 760 mm	122.9	2	BP	1.0317 0.0485	5 4	h		
100	61.66	4	t <sub>e</sub>	0.0372	5	f' to		ł
30 10	34.76 14.28	4 5	30 mm	0,6726	5	g'° <u>K</u>		
1	-19.91	5	ΔHm cal/g		1_1	h' to		+-
Pressure			ΔHv cal/g 25°C	84.02	5	m to		
mm 25°C	18.16 1069.	5	30 mm	83. 24	5	•		
Density	1007.	-   -	BP	71.17	5	m¹ to		
g/ml 20°C	0.721	2	t <sub>e</sub> (d, e)	69.39	5	n'   <u>*K</u>		
dt 25 4 30	0.717		AHv/Te	19.05	5			
a	0.737		d 35 to	87.99	5	Surface tension dynes/cm. 20°C	21.51	5
ь	-0.038		136 °C d' 25 to	0.1369 86.04	5	<b>8</b> 30	20.56	5
Ref. Index		_   _ ]	e'   35 °C	0.0807	5	40	19.64	5
<sup>n</sup> D 20°C	1.41 <b>3</b> 1.411		d <sub>c</sub> g/ml	0,228	5	Parachor [P] 20°C		
30	1.408		v <sub>c</sub> ml/g t <sub>c</sub> °C	4.393	5 5	30		
"C"	0.763	3 4	P <sub>c</sub> mm	18324.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.		2	PV/RT		+-	Exp. L.1.%/wt.	333.1	Ť
(nD-d/2)	38.677		25°C	1.0000	5	u.		1
Dielectric			30 mm BP	1.0000 0.9495	5 5	Dispersion Flash Point *C	119.	2
A 35 to			t <sub>e</sub>	0.9388	5	Fire Point		
B (155 °C	1353.4 216.	4 4	tc ΔHc kcal/m	0, 254	13	M. Spec.		†
A* 35 to			∆Hf			Ultra V. X-Ray Dif.		
B* 146 °C		5	ΔFf	<b></b>	$\perp$	Infrared		
K — — –			Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			η °c			Acetone Carbon tet.		
,x						Benzene		ŀ
A'   25 to B'   35 °C		64   5     5			$\sqcup$	Ether n-Heptane		
c' '	234.	5	B <sup>V</sup> to A <sup>V</sup> C			Ethanol	[	
A1# 25 to			<del> </del>	-		Water Water in		
B'* 35 °C		5	(B <sup>V</sup> )  to		1			+
Ac 155 to Bc t <sub>c</sub> °C	7.285	6   5	(A <sup>V</sup> )  °C	<del> </del>	$\vdash$			
Cc — -	258.	5	c <sub>p</sub> liq. °K				}	
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	135.95	5	c <sub>v</sub> vap.				ļ	
$T_{\mathbf{R}} = 0.7$	'5 <b>T</b> <sub>c</sub>					+ grams/100 gra	ms solve	nt
REFEREN	CES: 1-Do	w 2-A	PI 3-Lit. 4-	Calc, from de	et. da	ta 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICA			PI					
LITERATU	RE REFE	RENCES	5:					

							No. 70	
NAME	trans-3	-Octer	ne			STRUCTURAL	FORMUL	A.
Mole % Pur.	Ref. M	olecul		Molecular Weight 112.2	208	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH=C	нсн <sub>2</sub> сн	3
		Ref.			Ref			Ref
F,P. ℃	-110.	2	dt/dP			f   to		
F.P. 100%			*C/mm		ا ۔ ا	g <u>•K</u> _		
B. P. °C 760 mm	122.2		25°C BP	1.0480 0.0485	5 4	h		
100 mm	123.3 62.01	2 4	t <sub>e</sub>	0.0372	5	f' to		
30	35.09	4	30 mm	0.6732	5	g'   '*K_		
10	14.59 -19.63	5	∆Hm cal/g			h'		ļ
Pressure			ΔHv cal/g			m   to		
mm 25°C	17.85	5	25°C 30 mm	84.17 83.34	5 5	0	1	
t <sub>e</sub>	1070.	5	BP	71.27	5	m¹ to		_
Density g/ml 20°C	0.7152	2	te t (d, e)	69.54 69.47	5	n'	]	
dt 25	0.7110	2	ΔHv/Te		5	o'		
4 30	0.7068	4	d   35 to	19.05 88.14	5	Surface tension		
a b	0.7320	4	e   140 °C		5	dynes/cm. 20°C	20.83	5
Ref. Index	-0.0384	+	a' _ 25 to	86.17	5	30 40	19.86 18.92	5
n <sub>D</sub> 20°C	1.4126	2			5	Parachor [P]		1
- 25	1.4102	2	d g/ml vc ml/g	0.225 4.44	5	20°C		
30	1.4075	4	tc. °C	295.	5	30 40		
"C"	0.7679	4	P <sub>c</sub> mm	17998.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.)	39.09 38.677	2 5	PV/RT		$\Box$	Exp. L.1.%/wt.		
(nD-d/2)	1.0550	2	25°C 30 mm	1.0000	5 5	u. Dispersion	, , ,	2
Dielectric			BP	0.9495	5	Flash Point °C	121.	1-
A 35 to	6.8763		t <sub>e</sub>	0.9388 0.254	5 5	Fire Point	ł	
B (_153 °C	1355.7 216.	4	t <sub>c</sub> ΔHc kcal/m	0.234	+-	M Spec.		
A*  35 to	1,3086	_	ΔHf			Ultra V.		1
B* 146 °C	1271.9	5	ΔFf		$\sqcup$	X-Ray Dif. Infrared	i	1
K c			Viscosity centistokes			Solubility in +	ļ	T
t to			7 °C		1	Acetone		İ
t <sub>x</sub> l *C		_	•	]		Carbon tet. Benzene	ļ	1
A'   25 to B'   35 °C	7.2633 1557.0	5 5				Ether		
c, - 32 -	234.	5	B <sup>V</sup>   to			n-Heptane Ethanol		ļ
A'* 25 to	1.6585	3 5	A <sup>V</sup>   °C	}		Water		1
B'* 35 °C	1457.2	5	(B <sup>V</sup> ) to	1		Water in		<del> </del>
Ac   153 to	7,2873	5	(A <sup>V</sup> ) °C				1	
Bc tc °C	1676. 257.	5	c <sub>p</sub> liq. °K				ł	
Cryos. A°			c <sub>p</sub> vap. °K					
consts. B°			1 -					
t <sub>e</sub> °C	136.40	5	c <sub>v</sub> vap.					
$T_{R} = 0.75 T_{0}$	<u> </u>					+ grams/100 grai	ns solven	t
REFERENCE	CS: 1-Dow	2-AF	PI 3-Lit. 4-0	calc, from de	t. dat	ta 5-Calc, by for:		
SOURCE:		A	PI					
PURIFICATI	ON:	A	PI					
LITERATUR	E REFERI	ENCES	):					

							No. 71	
NAME _	cis-4-Oct	ene				STRUCTURAL	FORMUL	A
						CV 4CV \ CV-C		•••
Mole % Pur.		ecul mula		Molecular Weight 112,20	08	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн=0	H(CH <sub>2</sub> ) <sub>2</sub> C	<sup>н</sup> з
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%	-118,7	2	dt/dP °C/mm		_	f to		}
B. P. °C 760 mm	122.54	2	25°C BP	1.0168 0.0485	5 4	h		
100	61.33	4	t <sub>e</sub>	0.0372	5	f' to		
30 10	34.45 13.98	4 5	30 mm	0.6721	5	g' '° <u>K</u>		İ
1	-20.18	5	ΔHm cal/g ΔHv cal/g		<del> </del>	m to		_
Pressure mm 25°C	18.46	5	25°C	83.89	5	n		
t <sub>e</sub>	1068.	5	30 mm BP	83.13 71.07	5	<u> </u>		ļ
Density g/ml 20°C	0,7212	2	te te (d, e)	69.36	5	m' to		
dt 25	0.7170	2	ΔHv/T <sub>e</sub>	69.29 19.04	5	o'		
a 30	0.7128	4	d 35 to	87,85	5	Surface tension	3,	_
b b	-0.0 <sub>3</sub> 83	4	e136 °C d' 25 to	0.1369	5	dynes/cm. 20°C	21.53 20.54	5
Ref. Index	1 41 40		e'   35 °C	85.91 0.0806	5	40	19.58	5
<sup>n</sup> D 20°C	1.4148 1.4124	2	d g/ml	0.226	5	Parachor [P] 20°C		
30	1.4097	4	vc ml/g tc °C	4.42 295.	5	30 40		ļ
"C" MR (Obs.)	0,7653	2	P <sub>c</sub> mm	18155.	5		335.2	5
MR (Calc.)		5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0542	2	30 mm	1.0000	5	u. Dispersion	120.	2
Dielectric A 35 to	6.87149	4	BP t <sub>e</sub>	0.9495 0.9388	5	Flash Point C		
B   153 °C	1351.0	4	t <sub>c</sub> _	0.254	5	M. Spec.		
C A* 35 to	1,30477	5	∆Hc kcal/m ∆Hf			Ultra V.		İ
B*  146 °C	1267.4	5	ΔFf		-	X-Ray Dif. Infrared		
c			Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			η °C			Acetone Carbon tet.		
t <sub>x</sub> °C A'  25 to	7.25919	5				Benzene Ether		
B' _35 °C	1552, 2 234.	5 5	B <sub>v</sub> to		-	n-Heptane Ethanol		
A'* 25 to	1.65489	5	A to			Water		
B'* 35 ℃	1452.5	5	(B <sup>V</sup> )  to			Water in		
Acl 153 to Bc t <sub>c</sub> °C	7.2826 1672.	5	(A <sup>V</sup> )  °C		-	1		
Cc	257.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	135.54	5	c <sub>v</sub> vap.		<u></u>			
$T_{R} = 0.75$						grams/100 gra		t
REFERENC	ES: 1-Dow	2-A AF		Calc. from de	t. da	ata 5-Calc. by for	mula	
SOURCE: PURIFICAT	ION:	AI						
	RE REFEREI							

							No. 72	
NAME	trans-4-	Octen	e			STRUCTURAL		
						( )	\ CI	
Mole	Ref. M	olecul		Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH=CH	1(CH <sub>2</sub> /2C	<sup>11</sup> 3
% Pur.	F	_	W	Weight 112.2	-			Ref.
- D 00	-93,810	Ref.		r	Ref	<u> </u>	1	Ker.
F.P. °C F.P. 100%	-93.810	+-	dt/dP *C/mm		1 1	f to		
B. P. *C	<b>†</b>	$\vdash$	25°C	1.0047	5	h .	1	İ
760 mm	122.25	2	BP t <sub>e</sub>	0.0485 0.0372	4 5	f' to	-	$\vdash$
100 30	61.06 34.19	4	30 mm	0.6717	5	g'   'K_		1
10	13.74	5	ΔHm cal/g			h'		İ
Pressure	-20.40	1 3	ΔHv cal/g			m   to		
mm 25°C	18.71	5	25°C	83.78	5	n <u>*K</u> _	ļ	1
t <sub>e</sub>	1067.	5	30 mm BP	83.04 70.98	5 5	ļ <u> </u>		
Density	2 5141		1 1	69.27	5	m'   to	]	1
g/ml 20°C	0.7141 0.7099	2 2	t <sub>e</sub> (d, e)	69.20	5	"   '		1
d <sub>4</sub> 25 30	0.7057	4	ΔHv/T <sub>e</sub>	19.03	5	Surface tension		$\vdash$
	0.7309	4	d   34 to	87.72 0.1370	5 5	dynes/cm. 20°C	20.70	5
b D.C. Tuli	-0.0383	4	d'   20 to	85.79	5	30 40	19.73 18.80	5
Ref. Index n <sub>D</sub> 20°C	1,4118	2	e' 34 °C	<del> </del>	5	Parachor [P]	10.00	+
45	1.4093	2	d g/ml vc ml/g	0.223 4.48	5	20°C		
30	1.4068	4	tc °C	294.	5	30 40		
"C"	0.7676	4	P <sub>c</sub> mm	17879.	5		335.2	5
MR (Obs.) MR (Calc.)	39.08 38.677	2 5	PV/RT	· · · · · · · · · · · · · · · · · · ·	$\Box$	Exp. L.1.%/wt.		
(nD-d/2)	1.0548	Ž	25°C 30 mm	1.0000	5	u. Dispersion	122.	2
Dielectric			BP	0.9495	5	Flash Point °C	122.	+
A 34 to			te te	0.9389 0.254	5 5	Fire Point		1
B 1_152 °C C	1349.0 216.	4	t <sub>c</sub>	0.234	+-	M Spec.		
A*   34 to	1,30268	+-	ΔHf			Ultra V. X-Ray Dif.		
B* _145 °C		5	ΔFf	<b>.</b>		Infrared		
K ———	1		Viscosity centistokes			Solubility in +		1
t <sub>k</sub>  to_	1		7 °c			Acetone Carbon tet.		l
t <sub>x 1</sub>	5 35/0/	<u> </u>	,			Benzene		
A'   20 to B'   34 °C	7.25696 1550,1	5 5				Ether n-Heptane		
c'	234.	5	B <sup>V</sup> to			Ethanol	l	
A!* 20 to	1.65288		AV I °C			Water Water in	1	
B'* 34 °C		5	(B <sup>V</sup> ) to			Water in		1
Ac   152 to	7.2798	5	(A <sup>v</sup> )  °C					
Cc ' c-	257.	5	c <sub>p</sub> liq. °K					1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	135,22	5	c <sub>w</sub> vap.					
$T_{\mathbf{R}} = 0.79$	T <sub>c</sub>					grams/100 grai	ms solven	t
REFERENC	ES: 1-Dow	2-A1	PI 3-Lit. 4-0	Calc, from de	t, da	ta 5-Calc. by for	mula	
SOURCE:		<b>A</b> :	PI					
PURIFICAT	ION:	A.	PI					
LITERATU	RE REFERE	NCES	S:					

No. 73 2-Methyl-1-heptene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_4C = CH_2$ ĊH3 Mole Ref. Molecular Molecular C8H16 Weight 112, 208 % Pur Formula Ref Ref. F. P. °C -90.0 2 dt/dP f to F.P. 100% °C/mm 25°C g °K 0.8953 5 4 B. P. °C h BP 0.0482 760 mm 119.3 2 t<sub>e</sub> 0.0373 5 f 58.44 to 100 4 g' °<u>K</u> 30 31.74 4 30 mm 0.6672 5 10 11.43 5 h' AHm cal/g -22.45 5 m to ∆Hv cal/g Pressure °K 'n 25°C 82.81 5 mm 25°C 21.24 5 o 30 mm 82,27 5 5 1058. t<sub>e</sub> 5 BP 70,25 m' to Density te (d, e) 68.59 5 n' ۰ĸ g/ml 20°C 0.7205 2 5 68.52 o'  $d_4^t$ 25 0.7164 2 AHv/Te 19.00 5 30 0.7123 4 Surface tension d 31 to 86,63 5 0.7369 dynes/cm. 20°C 5 21.45 å °C 0.1373 5 Ъ -0.0381 4 30 20.48 5 20 to 84.80 40 19.54 5 Ref. Index e' 31 °C 0.0796 5 n<sub>D</sub> 20°C 1.4123 [P] Parachor d<sub>c</sub> g/ml 5 0.226 25 2 1.4098 20°C vc ml/g t\_ °C 30 4.43 1.4073 4 30  $t_c$ 5 291. 40 "C" 0.7617 4 P<sub>c</sub> mm 17989. 5 Sugd. 335.2 5 MR (Obs.) 38.78 2 PV/RT Exp. L. 1. %/wt. MR (Calc.) 38,677 5 25°C 1.0000 5 (nD-d/2)2 u. 1.0521 30 mm 1.0000 5 Dispersion 2 122. Dielectric BP 0.9495 Flash Point C 0.9391 5 A 31 to 6.85239 Fire Point 0.254 B (150 °C 1331.7 M. Spec. Ultra V. AHc kcal/m C 216. 4 ΔHf A\* 31 to 1.28966 X-Ray Dif. ΔFf B\*[135 °C 1249.0 Infrared ĸ Viscosity Solubility in centistokes Acetone to t<sub>k</sub> Carbon tet. °C Benzene 20 to 7.24294 Ether B' 3<u>1</u> °C 1532.2 5 n-Heptane B<sub>v</sub> | 234 5 Ethanol °C Water A'\* 20 to B'\* 31 °C 1.64083 5 Water in (B<sup>V</sup>)| 1433.1 5 to Ac | 150 to 7,2629 5 (A<sup>V</sup>)| °C 1649. Bc\_tc\_°C 5 cp liq. °K Cc 257. 5 c<sub>p</sub> vap. Cryos. A °K consts. Bº c, vap. te °C 131.90 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 74	
NAME	3-Met	hyl-l-he	pten <b>e</b>		- 1	STRUCTURAL	FORMUL	A
L					$\dashv$	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH C	H=CH <sub>2</sub>	
Mole	Ref.	Molecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2		CH <sub>3</sub>		
% Pur.				weight 112.2	Ref			Ref.
	т	Ref.		T	Kei		т —	1.01.
F.P. °C F.P. 100%	+		dt/dP *C/mm		1 1	f to	l	
B. P. °C	+		25°C	0.6480	5	h	1	
760 mm	111.	2	BP	0.0476	4	f' to	<del>                                     </del>	+
100 30	50.98 24.69		t <sub>e</sub> 30 mm	0.6566	5	g'   '°K		
10	4.82	5	ΔHm cal/g	0.0500	╅	h'		
1	-27.81	5	ΔHv cal/g		-	m   to		1
Pressure mm 25°C	30.47	5	25°C	79.74	5	n °K	1	
t <sub>e</sub>	1035.	5	30 mm	79.78	5 5	<u> </u>		
Density	1		BP t <sub>e</sub> ,	68.20 66.70	5	m'   to		
g/ml 20°C			t <sub>e</sub> (a, e)	66.64	5	n'   <u>°K</u>	4	1
d <sub>4</sub> 25	0.70		AHv/T <sub>e</sub>	18.91	5		<del> </del>	
	0,72		d   20 to		5	Surface tension dynes/cm. 20°C	20.34	5
ъ	-0.03		1		5	30	19.41	5
Ref. Index			•' · · · · · · · · · · · · · · · · · · ·			40	18.50	5
n <sub>D</sub> 20°C			d g/ml v ml/g	0.220	5	Parachor [P] 20°C		
30	1.40 1.40			4.54	5 5	30		
"C"	0.76		*c \	278.	5	40	l	1_
MR (Obs.)	+	2	P <sub>c</sub> mm	17269.	3		335.2	5
MR (Calc.	38.67	7   5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.05	1 2	30 mm	1.0000	5	Dispersion	117.	2
Dielectric	<del>                                     </del>		BP t <sub>e</sub>	0.9500	5 5	Flash Point °C		
A 20 to B 140 °C		779 4	tc	0.256	5	Fire Point		—
<u> </u>	218.	4	ΔHc kcal/m			M Spec. Ultra V.		
A*   20 to			ΔHf ΔFf			X-Ray Dif.		1
B* L125 °C	1216.9	5	Viscosity	<u> </u>	+	Infrared		
c	_		centistokes			Solubility in + Acetone		1
t <sub>k</sub>   to			η ∘c			Carbon tet.		
A' to	1					Benzene Ether		1
B' •C				<u> </u>	Щ	n-Heptane	1	1
C'			B <sup>V</sup>   to			Ethanol		
A'* to B'* °C			<u> </u>	-		Water Water in		
Ac   140 to		96 5					T	1
Bc t °C	1610.	5			$\vdash$		1	
<u>Cc</u>	259.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	122.59	5	c <sub>v</sub> vap.					
$T_R = 0.7$						grams/100 gra	ms solver	nt
REFEREN	CES: 1-De	ow 2-AI	PI 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc, by for	mula	
SOURCE:			PI					
PURIFICAT			PI					
LITERATU	RE REFE	ERENCES	3:			· · · · · · · · · · · · · · · · · · ·		

No. 75 4-Methyl-1-heptene STRUCTURAL FORMULA NAME CH3(CH2)2CH CH2CH=CH2 Molecular C8H16 ĊHą Mole Ref. Molecular Weight 112.208 % Pur Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm 25°C g \_°K 0.6943 5 B. P. °C h BP 0.0478 4 760 mm 112.8 2 0.0374 5 ſ١ 52.59 te to 100 4 g' °<u>K</u> 30 26, 20 4 30 mm 0.6593 5 10 6.13 5 h١ ∆Hm cal/g -27.34 5 m to AHv cal/g Pressure ۰ĸ n 25°C 80.33 5 28.23 mm 25°C 5 o 30 mm 80,26 5 te 1041. 5 BP 5 68. **6**6 m' to Density te te (d, e) 5 67.14 n' g/ml 20°C °K 0.717 0.7132 67.08 5 o'  $\mathbf{d_4^t}$ 25 AHV/Te 5 18.94 30 0.709 4 Surface tension d 20 to 83,77 5 0.7330 dynes/cm. 20°C 4 21.03 ᇷᅴ 1<u>30 °C</u> 0.1339 5 ь -0.0<sub>3</sub>794 20.10 30 to 19.19 5 40 Ref. Index e' <sup>n</sup>D 20°C 1.410 [P] Parachor d<sub>c</sub> g/ml 0.224 5 25 1.408 2 20°C vc ml/g t\_°C 5 4.47 30 1.406 4 30 <sup>t</sup>c 281. 5 40 "C" 0.7614 4 P<sub>c</sub> mm 17639. 5 Sugd. 335.2 5 MR (Obs.) 38.8 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1,0000 5 (nD-d/2)1.052 2 u. 30 mm Dispersion 1,0000 5 117. 2 Dielectric BP 0.9500 5 Flash Point C 0.9401 A 20 to 6.88388 Fire Point 0.256 B 1143 °C 1309.3 4 M. Spec. Ultra V. AHc kcal/m C 218. 4 ΔHf A\* 20 to 5 1.28143 X-Ray Dif. ΔFf B\* 130 °C 1227.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $\mathbf{t_k^t_x}$ Carbon tet. °C Benzene A' to Ether В' °C n-Heptane B<sub>v</sub> | C to Ethanol °C Water A1\* to Water in B'\* °C (B<sup>V</sup>)| to Ac 143 to (AV) 7.2514 5 °C Bc tc °C 1624. 5 5 c<sub>p</sub> liq. °K Cc 259. Cryos. A c<sub>p</sub> vap. °K consts. B° c<sub>w</sub> vap. te °C 5 124.61  $T_{R} = 0.75 T_{C}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

								No. 76	
NAME	5- <b>M</b>	ethyl-	l -he	ptene			STRUCTURAL	FORMUL	A
							כט כט כט וכט	\ CU-CU	
				·		$\neg$	CH <sub>3</sub> CH <sub>2</sub> CH (CH <sub>2</sub>	1/2CH-CH	2
Mole % Pur.	Ref.	Mole Form	cula mula	с <sub>8</sub> н <sub>16</sub>	Molecular Weight 112.2	08	CH <sub>3</sub>		
			Ref.			Ref	T		Ref
F.P. °C				dt/dP			f to		Г
F.P. 1009	•			°C/mm	0.700/	. 5	g <u>"K</u>		1
B. P. *C 760 mm	113.3		2	25°C BP	0.7096 0.0478	4	h		<u> </u>
100	53.09		4	t <sub>e</sub>	0.0374	5	f' to		1
30 10	26.70		4 5	30 mm	0.6593	5	h'		
1	-26.83		5	ΔHm cal/g		$\vdash$	m l to		$\vdash$
Pressure mm 25°C	37.5		_	ΔHv cal/g 25°C	80.76	5	n K		
t <sub>e</sub>	27.51 1042.0	·	5	30 mm	80.53	5	°		
Density	<del>                                     </del>			BP te	68.84 67.29	5	m'   to		
g/ml 20°0	0.71		2 2	t <sub>e</sub> (a, e)	67.24	5	n'   ' <u>°K</u> -	-	l
d <sub>4</sub> 25	0.70		4	ΔHv/T <sub>e</sub>	18.96	5	Surface tension		╁
	0.73		4	d 15 to		5	dynes/cm. 20°C	21.03	5
b D. C. T. I.	-0.0	83	4	d' i to	5		30 40	20.10 19.19	5
Ref. Index		94	2	e' ' •(		+	Parachor [P]		Ť
D 25 30	1.40		2	d g/ml vc ml/g tc °C	0.222 4.50	5	20°C		1
"C"	0.76		4	-	281.	5	30 40		ł
MR (Obs.		_	2	P <sub>c</sub> mm	17537.	5	Sugd.	335.2	5
MR (Calc.	38.67	77	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		1
(nD-d/2)	1.05	12	2	30 mm	1.0000	5	Dispersion	117.	2
Dielectric		1061	4	BP t <sub>e</sub>	0.9500 0.9401	5	Flash Point °C		
B 1143°		,,,,,	4	tc	0.256	5	Fire Point		┼
<u>C</u>	217.47		4	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 20 to B* 130 °C		3101	5	ΔFf			X-Ray Dif. Infrared		
K	-			Viscosity			Solubility in +		╁
t <sub>k</sub>			i	r centistokes			Acetone		
'x 1					1		Carbon tet. Benzene		
A' to							Ether		
č, – – -	<u>-</u>		1	B <sup>V</sup> l to			n-Heptane Ethanol		
A'* to				A <sup>V</sup> I °C	_		Water Water in		
B'* *(		-	_	(B <sup>V</sup> ) to	1		Water III		$\vdash$
Ac   143 to	7.25 C 1622.	01	5	(A <sup>V</sup> )  °C	<del></del>	-			
Ce	436.		5	c <sub>p</sub> liq. °K					1
Cryos. Acconsts. B				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	125.16	· [	5	c <sub>v</sub> vap.					
$T_R = 0.7$							f grams/100 grai		t
REFEREN	CES: 1-D	ow 2	-AP		Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:	TION		AF		······································				
LITERATU		ERENC	AF CES						
		JI DIN	-	•					

									No. 7	
NAME	6-M	lethyl	-1-h	epten <b>e</b>			STRUCTURAL FORMULA			
ļ		т		····	<del></del>	-	С	H <sub>3</sub> CH (CH <sub>2</sub> ) <sub>3</sub>	CH=CH <sub>2</sub>	
Mole	Ref	. Mo	lecul	ar Cu li	Molecular			ĊН <sub>3</sub>		
% Pur.		For	mul	C <sub>8</sub> H <sub>16</sub>	Weight 112.2	08				
			Ref.			Ref.				Ref.
F.P. °C				dt/dP			f	to		
F.P. 100%				°C/mm	0.7511	_	g	<u>*K</u>		
B. P. ℃				25°C BP	0.7511 0.0465	5 4	h			$\perp$
760 mm	113.2		2	t <sub>e</sub>	0.0363	5	f¹	to		
30	28.3	6	4	30 mm	0.6505	5	g'	' <u>*K</u>		
10	8.3		5	ΔHm cal/g			h'			
Pressure	-24.8		-	ΔHv cal/g			m	to		İ
mm 25°C	25.2	5	5	25°C	83.00	5	" l	•K	ł	1
t <sub>e</sub>	1041.		5	30 mm BP	82.53 70.73	5 5				┼
Density				t.	69.19	5	m'	to *K	İ	1
g/ml 20°C		120 079	2	te (a, e)	69.13	5	٠,		1	Ì
d <sup>t</sup> 25 4 30		038	4	ΔHv/T <sub>e</sub>	19.51	5	F	face tending		+-
a	0.7	284	4	d 20 to	86.48	5		face tension es/cm. 20°C	20.45	5
ь	-0.0		4	Tai To	0,1391		8	30	19.52	5
Ref. Index			١ . ا	e' °C			<u> </u>	40	18.61	<b>↓</b> "
n <sub>D</sub> 20°C		070 045	2	d <sub>c</sub> g/ml	0.240	5	Par	achor [P] 20°C		
30		018	4	v <sub>c</sub> mi/g	4.167	5		30		
"C"	0.7	610	4	tc°C Pcmm	18887.	5		40 Sugd	335.2	5
MR (Obs.)			2	PV/RT		+	Exp	. L.1.%/wt.	333.2	+-
MR (Calc. (nD-d/2)		77 510	5 2	25°C	1.0000	5	•	u.	ļ	
Dielectric			-	30 mm BP	1.0000 0.9500	5		persion	117.	2
A 20 to		4874	5	t_	0.9404	5		sh Point C Point		
B 142 °C	1345.2		5	t <sub>c</sub>	0.256	5		····		+
<u> </u>	217.4		5	ΔHc kcal/m ΔHf				Spec. ra V.		
A*  20 to B*  135 °C		9165	5 <b>5</b>	ΔFf				lay Dif.		
K LISS S	- 1203.4		'	Viscosity				ared		+-
t,tō	-			centistokes				ibility in <sup>†</sup>		1
t <sub>k</sub> to				ŋ •c			Ca	rbon tet.		
A'   10 to	7.3	7919	5		į			nzene her	1	
B'i_20 °C	1554.3	7	5	B <sub>v</sub> to	<b> </b>	+	n-:	Heptane		
C'	235.4		5	B' to				hanol iter		
A'* 10 to B'* 20 °C		8823	5	$\frac{1}{(\mathbf{B}^{\mathbf{v}})} = \frac{1}{\mathbf{t}_0}$	-			ter in	1	
Ac 142 to		591	5	(A <sup>V</sup> )  °C						
Bc <sub>l</sub> t <sub>c</sub> ℃	1656.	- / -	5	<del></del>	<del> </del>	-	1			
Cc	257.		5	P						
Cryos, A° consts, B°				c <sub>p</sub> vap. °K						
te °C	124.7		5	c <sub>v</sub> vap.		1				
$T_{\mathbf{R}} = 0.7$	5 T <sub>C</sub>						+ gz	ams/100 gra	ms solve	nt
REFEREN	CES: 1-	Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:	-		A	PI						
PURIFICA	TION:		А	PI						
LITERATU	RE REI	ERE	NCE	S:						

							No. 78
NAME	2-Methy	/1-2-l	eptene			STRUCTURAL	FORMULA
						СН <sub>3</sub> (СН <sub>2</sub> ) <sub>3</sub> СН	=C CH <sub>2</sub>
Mole	Ref. M	olecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular		3 23	Ċнз
% Pur.	F	ormul	a C8H16	Weight 112.2	08		
	<del></del>	Ref.		<b>,</b>	Ref.		Rei
F.P. C F.P. 100%		—	dt/dP			f to	1
B. P. *C	<u>'</u>	╁	*C/mm 25°C	1.0857	· 5	g	
76 <b>0 mm</b>	122.6	2	BP	0.0474 0.0363	4 5	f' to	<del> </del>
100 30	62.51 35.98	4	t <sub>e</sub> 30 mm	0.6646	5	g'  K_	
10	15.72	5	ΔHm cal/g	0.0010	<del>                                     </del>	h'	
1	-18,21	5	ΔHv cal/g		-	m to	
Pressure mm 25°C	16.90	5	25°C	85.82	5	n •K_	1
t <sub>e</sub>	1067.3	5	30 mm BP	84.91 72.71	5	<u> </u>	
Density g/ml 20°C	0.7341	١,	t <sub>e</sub>	70.99	5	m' to	
at 25	0.7241 0.7200	2 2	te (d, e)	70.92	5	o' ' ====	
	0.7159	4	ΔHv/T <sub>e</sub>	19.50 89.89	5	Surface tension	
a b	0.7405 -0.0 <sub>3</sub> 82	4	e   135 °C	0.1408	5	dynes/cm. 20°C	21.88 5 20.90 5
Ref. Index		+-	d'   20 to	87.89 0.0829	5	40	19.95 5
n <sub>D</sub> 20°C	1.4170	2	<del> </del>	0,242	5	Parachor [P]	
25 30	1.4145	2 4	d g/ml vc ml/g	4.125	5	20°C 30	
"C"	0.7661	4	1 6	296. 19555.	5	40	
MR (Obs.)		2	P <sub>c</sub> mm	17555.	-	Exp. L.1.%/wt.	335,2 5
MR (Calc. (nD-d/2)	38.677 1.0550	5 2	25°C	1.0000	5	u.	
Dielectric	1.0350	+-	30 mm BP	1.0000	5	Dispersion	124. 2
A   36 to	6.95927	5	te	0.9391	5	Flash Point °C Fire Point	] ]
B (154 °C	215.7	5	t <sub>c</sub>	0,255	5	M Spec.	
A*  36 to		5	∆Hc kcal/m ∆Hf		ŀ	Ultra V. X-Ray Dif.	<u> </u>
B* 145 °C		5	ΔFf		<u> </u>	Infrared	
K — — -	i		Viscosity centistokes		1	Solubility in +	
t <sub>k</sub> Tto		1	7 °℃	İ		Acetone Carbon tet.	1 1
A' 20 to		5				Benzene	
B' _ 36 °C		5				Ether n-Heptane	
C'	233.7	5	B <sup>V</sup>   to A <sup>V</sup>   °C		Ì	Ethanol Water	
A'* 20 to B'* 36 °C		5	(B <sup>V</sup> )to			Water in	
Ac   154 to	7.3692	5	(A <sup>V</sup> ) °C				
Bc tc_°C	1	5	c <sub>p</sub> liq. °K		<u> </u>		
Cryos, A°	256.	+ -	11				
consts. B°			ъ				
t <sub>e</sub> °C	135.29	5	c <sub>v</sub> vap.				
$T_R = 0.7$						grams/100 gran	ns solvent
	CES: 1-Dow			alc. from de	t. da	ta 5-Calc. by for	mula
SOURCE:			PI				
PURIFICAT			.PI				
₩TERATU	RE REFERE	NCES	5:				

	3-M	ethyl.	-cie-	2-hantana				No. 7			
NAME		einyi	-018-	2-heptene			STRUCTURAL FORMULA				
L	<del></del>	T					СН <sub>3</sub> (СН <sub>2</sub> ) <sub>3</sub> С		3		
Mole % Pur.	Ref.	Mo!	ecula		Molecular Veight 112,20	۱۵	c	<sup>:H</sup> 3			
70 FUI.			Ref.	0 10 1	veight 112.20	Ref.	т		Ref.		
F.P. °C			101.	dt/dP		IXEI.		T	1		
F.P. 100%				°C/mm		1	f to				
B. P. °C				25°C BP	1.0610 0.0473	5	h i				
760 mm 100	122. 62.5		2	te	0.0363	5	f' to				
30	35.50		4	30 mm	0.6637	5	g' ' <u>°</u> <u>F</u>	2			
10 1	15.27		5	ΔHm cal/g			h'				
Pressure	1 20.0		-	ΔHv cal/g			m to				
mm 25°C	17.33	3	5	25°C 30 mm	85.63 84.76	5	<del>-</del>	1			
t <sub>e</sub>	1066.		_5	BP	72.59	5	m' to		+-		
Density g/ml 20°C	0.72	29	2	te te (d, e)	70.89 70.82	5	n'   °F				
dt 25	0.72	25	2	ΔHv/T <sub>e</sub>	19.50	5	o'				
	0.72	+	4	d 35 to	89.76	5	Surface tension				
a b	-0.0		4	_e_ <u> _135 °C</u>	0.1407	5	dynes/cm. 20°C	22.48	5		
Ref. Index		,		d'   25 to e'   35 °C	87.69 0.0826	5	40	20.55	5		
n <sub>D</sub> 20°C	1.4		2	d <sub>c</sub> g/ml	0,246	5	Parachor [P]				
25 30	1.41		2	A - m1/6	4.072	5	20°C				
"C"	0.76		4	tc°℃	296.	5	40				
MR (Obs.)			2	P <sub>c</sub> mm	19812.	5		335.2	5		
MR (Calc. (nD-d/2)	38.6		5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.	·			
Dielectric	1.05	55	2	30 mm BP	1.0000	5	Dispersion	124.	2		
A 35 to	6 04	5905	5	t <sub>e</sub>	0.9495 0.9391	5	Flash Point C				
B 1154 °C	1377.7	1	5	т <sub>с</sub>	0.255	5	Fire Point		-		
C	215.82		5	ΔHc kcal/m ΔHf	!		M. Spec. Ultra V.				
A* 35 to B* 145 °C		354	5	ΔFf			X-Ray Dif. Infrared				
к — — –	_			Viscosity			Solubility in +		+-		
t <sub>k</sub> – tō	-			centistokes 7°C			Acetone		Ì		
<u>'x   </u>				•			Carbon tet. Benzene	}			
A'   25 to B'   35 °C		168	5				Ether				
C' -35	233.82	2	5	B <sup>V</sup>			n-Heptane Ethanol		ĺ		
A!* 25 to		1680	5				Water				
B'* 35 °C			5	(B <sup>V</sup> )  to			Water in	+	+-		
Acl 154 to Bc t <sub>c</sub> °C	7.36	93	5	(A <sup>V</sup> )  °C		ļ	4				
Cc C	256.		5	c <sub>p</sub> liq. °K							
Cryos. A° consts. B°				c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	134.6		5	c <sub>w</sub> vap.							
$T_R = 0.7$	5 Т <sub>с</sub>						grams/100 gr	ams solve	nt		
REFEREN	CES: 1-I	Dow	2-A1	PI 3-Lit. 4-0	Calc. from de	t. da	ata 5-Calc. by fo	rmula			
SOURCE:				PI	<del>,</del>						
PURIFICA:				PI			····				
LITERATU	RE REF	EREN	ICES	:							

							No. 80	
NAME	3-Methyl-	tran	s-2-heptene			STRUCTURAL F	ORMULA	
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> C =	снсн.	
Mole	7.4			M-11		CH <sub>3</sub>	3	
% Pur.	Ref. Mo	rmul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	:08			
		Ref.			Ref			Ref.
F.P. C			dt/dP			f to		
F.P. 1009			*C/mm 25*C	1,0610	. 5	g <u>*K</u>		
B. P. °C 760 mm	122.	2	BP	0.0473	4	h		
100 30	62.5	4	t <sub>e</sub>	0.0363	5	f' to to		l
10	35.50 15.27	5	30 mm	0.6637	5	h'		
1	-18.61	5	ΔHm cal/g ΔHv cal/g	<b></b>	-	m to		-
Pressure mm 25°C	17.33	5	25°C	85.63	5	n   °K		
t <sub>e</sub>	1066.	5	30 mm BP	84.76 72.59	5	<u> </u>		
Density		_	t_	70.89	5	m'   to		
g/ml 20°C	0.729 0.725	2 2	14,07	70.82	5	"·   ' <del>-</del> -		
dt 25 4 30	0.721	4	ΔHv/T <sub>e</sub>	19.50	5	Surface tension		
a b	0.745 -0.0 <sub>3</sub> 8	4	d 35 to	89.76 0.1407	5 5	dynes/cm. 20°C	22.48	5
Ref. Index		<del> </del>	d' 25 to e' 35 °C	87.69 0.0826	5	30 40	21.50 20.55	5
n <sub>D</sub> 20°C	1.419	2		0.0826	5	Parachor [P]	i	
25 30	1.417 1.414	2 4	d g/ml v ml/g	4.072	5	20°C		
"C"	0.7643	4	ic C	296.	5	40		l
MR (Obs.)	38.9	2	P <sub>c</sub> mm	19812.	5	Sugd.	335.2	5
MR (Calc. (nD-d/2)	) 38.677 1.055	5 2	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric	1.033	-	30 mm BP	1.0000	5	Dispersion	124.	2
A   35 to	6.95905	5	t <sub>e</sub>	0.9495 0.9391	5	Flash Point °C Fire Point		
B [154 °C	1377.7	5	t <sub>c</sub>	0.255	5	M Spec.		
C	215.82	5	ΔHc kcal/m ΔHf	}		Ultra V.		
A*   35 to B* 145 °C		5	ΔFf		L	X-Ray Dif. Infrared		
K — — -	-		Viscosity			Solubility in +		
the Total			centistokes °C		ĺ	Acetone		
x l			•		1	Carbon tet. Benzene		
A'   25 to B' _ 35 °C		5				Ether n-Heptane		
C'	233.82	5	B <sup>V</sup> to			Ethanol		
A'* 25 to B'* 35 °C		5 5	$\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})_1} - \frac{{}^{\bullet}C}{t_0}$			Water Water in		
Ac   154 to		5						
Bc t °C	1698.	5	<del>- :</del>	ļ	-	1 1		
Ce	256.	5	P -					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
te °C	134.61	5	c <sub>w</sub> vap.					
$T_R = 0.7$						grams/100 gran	ns solvent	
REFEREN	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	t, da	ta 5-Calc. by forr	nula	
SOURCE:		A	PI					
PURIFICA			PI					
LITERATU	RE REFERE	NCES	<b>:</b>					

							No. 81	
NAME	4-Methyl	-cis-	2-heptene	_		STRUCTURAL	FORMUL	A
						сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн с	:H=CH CH.	_
Mole % Pur.	Ref. Mo	lecul rmul	ar С <sub>8</sub> н <sub>16</sub>	Molecular Weight 112,20	8	CH <sub>3</sub>		3
	_	Ref.			Ref.			Ref.
F. P. °C			dt/dP			f to		
F.P. 100% B.P. °C	<u>'</u>	<del> </del>	°C/mm 25°C	0,7791	5	g '° <u>K</u>		
760 mm	114.	2	BP	0.0465 0.0363	4 5	h		-
100 30	55.05 29.03	4	t <sub>e</sub> 30 mm	0.6516	5	f' to		
10	9.17	5	ΔHm cal/g	0.0310	+ -	h'		
1	-24.09	5	ΔHv cal/g	+	$\vdash$	m to		
Pressure mm 25°C	24.33	5	25°C	83.06	5	n <u>°K</u>		
t <sub>e</sub>	1043.	5	30 mm BP	82.76 70.93	5			├
Density g/ml 20°C	0.71/		te (d.s)	69.37	5	m' to		
at 25	0.716 0.712	2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	69.32	5.	0'		
	0.708	4	d 29 to	19.52 86.80	5	Surface tension		
a b	0.732 -0.0 <sub>3</sub> 8	4	_e   126 °C	0.1392	5	dynes/cm. 20°C	20.92	5
Ref. Index		1	d' 15 to	84.95 0.0754	5	40	19.99 19.08	5
<sup>n</sup> D 20°C	1.410	2	d g/ml	0, 242	5	Parachor [P]		
30	1.408 1.405	2 4	v ml/g	4. 126	5	20°C 30	·	
"C"	0.7624	4	tc°C P <sub>c</sub> mm	282. 19144.	5	40 Sund	225 2	_
MR (Obs.)		2	PV/RT	17144.	3	Exp. L. 1. %/wt.	335.2	5
MR (Calc. (nD-d/2)	38.677 1.052	5 2	25°C	1.0000	5	u.	ļ	
Dielectric		<u> </u>	30 mm BP	1.0000 0.9500	5	Dispersion	119.	2
A 29 to	6.95106	5	t <sub>e</sub>	0.9403	5	Flash Point C Fire Point		
B (144 °C	1348.6 217.34	5	tc ΔHc kcal/m	0.256	-	M. Spec.		
A* 29 to	1.39312	5	ΔHf	1		Ultra V. X-Ray Dif.		
B*[136 °C	_ 1266.7	5	ΔFf	<u> </u>	-	Infrared		<u>L</u> _
c	_		Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			η °C			Acetone Carbon tet.		
t   *C	7.35098	5				Benzene Ether		
B' 1_29 °C	1552.9	5 <b>5</b>	B <sup>V</sup> to	<del>                                     </del>		n-Heptane		
A'* 15 to	1,74916	5	B to ℃	1		Ethanol Water	{	
B'* 29 °C		5	(B <sup>V</sup> )  to	1		Water in		<u> </u>
Ac 144 to	7. 3615	5	(A <sup>V</sup> )  °C					
Bc t <sub>c</sub> °C	1661. 257.	5	c <sub>p</sub> liq. °K					
Cryos, A° consts, B°		Г	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	125.59	5	c <sub>w</sub> vap.					
$T_R = 0.7$	5 T <sub>C</sub>					grams/100 gra	ms solven	t
REFEREN	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICA'			PI					
LITERATU	RE REFERE	NCES	S:					

								No. 82	
NAME	4-M	ethyl	-tran	s-2-heptene			STRUCTURAL	FORMUL	A.
Mole % Pur.	Ref.	Mo Fo	lecul	ar . a C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	208	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн с сн <sub>3</sub>	н=СНСН	3
			Ref.			Ref			Ref.
F. P. *C	T		1	dt/dP	T.		f   to		$\dagger$
F.P. 1009				*C/mm	1	1	f to		
B. P. *C	†			25°C	0.7791	5	h .		
760 mm	114.		2	BP	0. 0465 0. 0363	4 5	f' to		1
100 30	55.0 29.0		4 4	t <sub>e</sub> 30 mm		5	g'   'K_		
10	9.1		5		0.6516	+	h'		
1	-24.0	9	5	ΔHm cal/g		$\vdash$	m l to		T
Pressure	1	_		ΔHv cal/g 25°C	83.06	5	n °K_		
mm 25°C	24.3 1043.	3	5	30 mm	82.76	5	0		
Density	1045.		1	BP	70.93	5 5	m¹ to		
g/ml 20°C	0.7	16	2	te (d, e)	69.37 69.32	5	n'  K_		
at 25	0.7	12	2	ΔHv/T <sub>e</sub>	19.52	5	0'		
	0.7		4	d   29 to		5	Surface tension		
a b	0.7		4 4	e   126 °C	0.1392	5	dynes/cm. 20°C	20.92	5
Ref. Index	-0.0	30	+ +	d' 15 to	84.95	5	30 40	19.99 19.08	5
n <sub>D</sub> 20°C		10	2		<del></del>		Parachor [P]		†
25	1.4	08	2	d g/ml vc ml/g	0.242 4.126	5 5	20°C		
30	1.4		4	tc °C	282.	5	30 40		
"C"	0.7	624	4	P <sub>c</sub> mm	19144.	5		335.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT	<del> </del>	$\vdash$	Exp. L.1.%/wt.		
(nD-d/2)	38.6		2	25°C	1.0000	5	u.		
Dielectric				30 mm BP	1.0000 0.9500	5 5	Dispersion	121.	2
A   29 to	6.9	5106	5	t <sub>e</sub>	0.9403	5	Flash Point °C Fire Point		
B 1144°C	1348.6		5	tc	0.256	5	M Spec.		┼─
<u> </u>	217.3	4	5	ΔHc kcal/m ΔHf			Ultra V.		]
A*  29 to   B* <sub> </sub> 136 °C		9312	5	ΔFf			X-Ray Dif.		1
K 1.30 7	1200. /			Viscosity			Infrared		-
° .— —.	_			centistokes			Solubility in + Acetone		1
tk to				η •ο			Carbon tet.		1
t <sub>x</sub>   *C		5098	5		1		Benzene Ether		
B' 29 °C			5			$\perp$	n-Heptane		
C'	235.3	4	5	B <sup>V</sup> to			Ethanol		
A'* 15 to		4916	5	A <sup>V</sup> C	_		Water Water in		
B'* 29 °C			5	(B <sup>V</sup> )  to					$\vdash$
Ac   144 to	7.3	615	5	(A <sup>V</sup> )  °C					
Cc - c-	257.		5	c <sub>p</sub> liq. ∘K					
Cryos. A° consts. B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	125.5	9	5	c <sub>w</sub> vap.					1
$T_R = 0.7$			نـــــــا	<u> </u>	<u> </u>		grams/100 grai	ne solven	.+
REFEREN		)ow	2-AF	PI 3-Lit. 4-	Calc. from de	t. dai	ta 5-Calc. by for		
SOURCE:			AI				3 Oute, by 101.		
PURIFICA'	TION:		AI						
LITERATU		ERE							

						- <u> </u>	No. 83
NAME	5-Methyl	-cis	-2-heptene			STRUCTURAL	FORMULA
Mole		lecul		Molecular	200	сн <sub>3</sub> сн <sub>2</sub> сн сн <sub>2</sub> сн <sub>3</sub>	сн=снсн <sub>3</sub>
% Pur.	For	mula		Weight 112.		<del></del>	Ref.
F.P. °C F.P. 100%		Ref.	dt/dP		Ref.	f to	Kei.
B. P. °C			°C/mm 25°C BP	0.9078 0.0469	5 4	g	
760 mm	118. 58.51	2 4	te	0.0363	5	f <sup>1</sup> to	
30	32.24	4	30 mm	0.6577	5	g'° <u>K</u>	
10	12.19 -21.38	5	ΔHm cal/g		$\perp$	h'	
Pressure			ΔHv cal/g 25°C	94 21	5	m to	
mm 25°C	20.57 1055.	5	30 mm	84.31 83.73	5	0	
Density	1055.	<u> </u>	BP	71.76	5	m' to	
g/ml 20°C		2	t <sub>e</sub> (d, e)	70.06	5	n'   <u>°K</u> _	
d <sub>4</sub> 30	0.719 0.715	2	ΔHv/T <sub>e</sub>	19.51	5		
a	0.739	4	d 32 to	88.24	5	Surface tension dynes/cm, 20°C	21.75 5
ь	-0.038	4	e 130 °C	0.1397 86.30	5	<b>8</b> 30	20.79 5 19.86 5
Ref. Index		2	e'   32 °C	0.0797	5	Parachor [P]	17.80
25	1.412	2	d g/ml	0.244 4.106	5 5	20°C	
30	1.409	4	vc ml/g tc °C	289.	5	30 40	
"C"	0.7620	2	P <sub>c</sub> mm	19479.	5	5	335.2 5
MR (Obs.) MR (Calc.		5	PV/RT	1 0000	_	Exp. L.1.%/wt.	
(nD-d/2)	1.053	2	25°C 30 mm	1.0000	5 5	u. Dispersion	119. 2
Dielectric	<del>                                     </del>	L_	BP	0.9500 0.9400	5	Flash Point C	
A 32 to B 149 °C		5	t e t c	0.256	5	Fire Point	
С	216.58	5	ΔHc kcal/m ΔHf			M. Spec. Ultra V.	
A* 32 to B* 135 °C	1.39140 1280.0	5	ΔFf	İ		X-Ray Dif. Infrared	
к — — –	-		Viscosity			Solubility in +	
t <sub>k</sub> tō	-		centistokes 7°C	1		Acetone	
t <sub>x</sub> °C			<b>'</b>			Carbon tet. Benzene	
A'   20 to	7.34977 _ 1567.0	5 5				Ether	1
C' -32 9	234.58	5	B <sup>V</sup> to C			n-Heptane Ethanol	
A'* 20 to		5	ı⊢—			Water Water in	
B'* 32 °C	<del></del>	5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C				
Bc tc °C	1679.	5		-	+-		
Cc -	257.	5	P ·				
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	130.12	5	c <sub>v</sub> vap.				
$T_R = 0.7$						grams/100 gra	
	CES: 1-Dow			Calc. from de	et. da	ta 5-Calc. by for	mula
SOURCE:	TION.		PI				
PURIFICA	RE REFERE		PI s.				
LILERATU	RE REFERE	NCE	<b>:</b>				

							No. 84	
NAME	5- <b>M</b> e	thyl-tran	s-2-heptene		- 1	STRUCTURAL	FORMUL	A
Γ								
	T		·		-	CH <sub>3</sub> CH <sub>2</sub> CHCH <sub>2</sub> C	CH=CH C	<sup>1</sup> 3
Mole	Ref.	Molecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular		ĊН <sub>3</sub>		
% Pur,			a 08-16	Weight 112.2	_			-
		Ref.		γ	Ref		<del>,                                    </del>	Ref
F.P. °C	<del>                                     </del>		dt/dP	1		f to		1
F.P. 100%			*C/mm 25*C	0.9078	5	g <u>K</u>		
B. P. *C 760 mm	118.	2	BP	0. 0469	4	h		<u> </u>
100	58.51	4	t <sub>e</sub>	0.0363	5	f' to		
30	32.24	4 5	30 mm	0.6577	5	g'  K_	1	1
10 1	12.19 -21.38	5	∆Hm cal/g			h'	ļ	ــــــــــــــــــــــــــــــــــــــ
Pressure	1		ΔHv cal/g			m to		
mm 25°C	20.57	5	25°C	84.31	5	n •K	1	
t <sub>e</sub>	1055.	5	30 mm BP	83.73 71.76	5 5	ļ		╄
Density	T		t_	70.12	5	m' to		
g/ml 20°C			t <sub>e</sub> (d, e)	70.06	5	n'   ' •K	1	
d <sub>4</sub> 25	0.71		AHv/T <sub>e</sub>	19.51	5		<u> </u>	+-
	0,73		d   32 to		5	Surface tension dynes/cm. 20°C	21.75	5
b	-0.03		a, 1 130 °C		5 5	30	20.79	5
Ref. Index			e' 32 °C		5	40	19.86	5
n <sub>D</sub> 20°C			d g/ml	0,244	5	Parachor [P]		
25 30	1.41		d g/ml vc ml/g	4, 106	5	20°C 30		
"C"	0.76		t <sub>c</sub> •C	289.	5	40		1
	+		P <sub>c</sub> mm	19479.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.	38.8 38.67	7 2 5	PV/RT		$\vdash$	Exp. L.1.%/wt.		
(nD-d/2)	1.05		25°C	1.0000	5	u.		1.
Dielectric			30 mm BP	1.0000	5 5	Dispersion	121.	2
A 32 to	6.95	360 5	te	0.9400	5	Flash Point °C Fire Point		İ
B 1149 °C	1362.7	5	t <sub>c</sub>	0.256	5			+
С	216.58	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1
A*  32 to	1.39	140 5	ΔFf	İ		X-Ray Dif.		1
B* 135 ℃	- 1280.0	"	Viscosity		+	Infrared		
c	_		centistokes			Solubility in + Acetone		1
tk to			η •c			Carbon tet.		Ī
		077			1 1	Benzene		
A'   20 to B'   32 °C		977   5				Ether n-Heptane		
č, – – –	234.58	5	B <sup>V</sup> to			Ethanol		
A!* 20 to			AV I _ °C	.]		Water		
B'* 32 °C	+	5	(B <sup>V</sup> ) to			Water in	<del> </del>	+
Ac   149 to	7.36		(A <sup>V</sup> )  °C					
Bc tc_°C	1679. 257.	5 5	c <sub>p</sub> liq. °K					1
Cryos, A°	1 23	-   -	_	ļ.				
consts. B°	1		c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	130.12	5	c <sub>w</sub> vap.					1
$T_{\mathbf{R}} = 0.7$	5 T <sub>c</sub>		L	<del></del>	ــــــــــــــــــــــــــــــــــــــ	+ grams/100 gra	me es\	
REFERENC		ow 2-AF	PI 3-Lit. 4-0	Calc. from de	t d=0			16
SOURCE:		AP		de	ual	- J-Care, by for		
PURIFICAT	TION:	AP				<del></del>		
LITERATU				<del></del>				
MIERAIU	RE REFI	ERENCES	:					

No. 85 6-Methyl-cis-2-heptene STRUCTURAL FORMULA NAME CH3CH (CH2)2CH=CHCH3 Molecular C8H16 ĊН3 Mole Ref. Molecular % Pur Weight 112,208 Ref. Ref Ref. **F**, P, °C dt/dP f to F.P. 100% \*C/mm 25\*C °K g 0.8733 5 B. P. C h BP 0.0468 4 760 mm 117. 2 0.0363 5 f١ to 100 57.63 4 <u>°К</u> g' 30 30 mm 31.43 4 0.6563 5 10 5 11.43 h! ∆Hm cal/g 5 1 -22.07 to m AHv cal/g Pressure °K n 25°C 83.98 mm 25°C 21.47 5 o 30 mm 83.48 1052. te 5 BP 5 71.53 m' to Density te (d, e) 5 69.92 n' g/ml 20°C °K 0.718 2 69.86 5 01 25 0.714  $d_4^t$ ΔHv/T<sub>e</sub> 19.51 5 30 0.710 4 Surface tension 31 to 87,86 5 0.734 44 21.15 5 dynes/cm. 20°C 129 °C 0.1396 5 -0.0<sub>3</sub>8 ь 20.22 5 to 30 d٦ 20 85.95 40 19.31 5 Ref. Index e' 31 °C 0.0788 5 20°C 1.412 1.410 2 [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 5 0.242 25 20°C vc ml/g 5 4.138 30 1.407 4 30 5 287. t<sub>c</sub> 40 "C" 0.7638 4  $P_c$  mm 19262. 5 335.2 5 Sugd. MR (Obs.) 38.9 38.677 2 PV/RT Exp. L. l. %/wt. MR (Calc.) (nD-d/2) 25°C 1,0000 u. 1.053 2 30 mm 5 119. 2 1.0000 Dispersion Dielectric BP 0.9500 Flash Point C 0.9401 6.95222 A 31 to Fire Point 5 1358.9 0.256B 147 °C M. Spec. С 216.77 5 AHc kcal/m Ultra V. A\* 31 to B\* 139 °C ΔHf 1.39108 5 X-Ray Dif. ΔFf 1276.4 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. ليكا °C Benzene 20 to 7.34929 5 Ether 1563. 2 В' 31 °C n-Heptane B<sub>v</sub> C 234.77 5 to Ethanol °C A'\* 20 to Water 1.74633 5 Water in B'\* 31 °C (B<sup>V</sup>) 1463.5 to Ac 147 to 7.3625 (A<sup>V</sup>) 5 °C Bc tc °C °C c<sub>p</sub> liq. ۰ĸ 257. 5 Cryos, Aº c<sub>p</sub> vap. °K consts. B° te °C c<sub>v</sub> vap. 128, 99 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

							No. 86	
NAME	6-Me	thyl-tra	ans-2-heptene			STRUCTURAL	FORMUL	A.
					1	сн <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub> с	н=сн сн	ſ_
34-1-	2.4		1	Malamilan		сн <sub>3</sub>		<b>'3</b>
Mole % Pur.	Ref.	Molec Form		Molecular Weight 112.	208	3		
		Re			Ref			Ref.
F. P. *C			dt/dP			f to		T
F.P. 100%			*C/mm		ا ۔ ا	g <u>*K</u> _		ı
B. P. °C	l		25°C BP	0.8733 0.0468	5 4	h ;		
760 mm 100	117. 57.63	2	- 11 .	0.0363	5	f' to		I
30	31.43	4	1 30 mm	0.6563	5	g'	ļ	
10 1	11.43					h'		
Pressure	<del> </del>		ΔHv cal/g			m to		
mm 25°C	21.47	5	25°C	83.98	5	n   ' *K-	ł	İ
te	1052.	5	30 mm BP	83.48 71.53	5 5	<del></del>		₩
Density			t <sub>e</sub> .	69.92	5	m'   to		
g/ml 20°C	0.71		t (d, e)	69.86	5	0'	1	
d <sub>4</sub> 30	0.71		ΔHV/T <sub>e</sub>	19.51	5	Surface tension	<b></b>	+-
	0.73	4 4		to 87.86 °C 0.1396	5	dynes/cm. 20°C	21.15	5
ь	-0.03	8 4		0.1396 to 85.95	5 5	30	20.22	5
Ref. Index	1 , 4,	,  ,	e' 31	°C 0.0788	5	40	19.31	5
<sup>n</sup> D 20°C	1.41		d <sub>c</sub> g/ml	0.242	5	Parachor [P] 20°C	i	
30	1.40	7 4	d g/ml v <sub>c</sub> ml/g t <sub>c</sub> •C	4.138 287.	5 5	30		
"C"	0.76	38 4	P <sub>c</sub> mm	19262.	5	40 Sugd	335.2	5
MR (Obs.)	38.9	_   2	DV/PT		+	Exp. L.1.%/wt.	333.2	+
MR (Calc.) (nD-d/2)	38.67		25°C	1.0000	5	u.		
Dielectric			30 mm BP	1.0000	5 5	Dispersion	121.	2
A 31 to	6, 95	222 5	t_	0.9401	5	Flash Point °C Fire Point		
B 1147 °C		. 5		0,256	5	M Spec.		+
C	216.77		- Luc	n		Ultra V.		
A*   31 to B* 139 °C	1.39 1276.4	108   5	AFF			X-Ray Dif. Infrared		
K F3Z =	1.0.1	آ ا	Viscosity					+
tto	.		centistokes	c		Solubility in + Acetone		
t <sub>x</sub>   to			7 '			Carbon tet.		
A'   20 to	7.34	929 5	<del>,  </del>			Benzene Ether		
B' ∟31 °C	1563.2	5	·		+	n-Heptane		
	234, 77			to °C		Ethanol Water	İ	1
A'* 20 to B'* 31 °C	1.74 1463.5	633   5	? <del>                                    </del>	to	1 1	Water in		
Ac   147 to	7, 36	25 5		·c				1
Bc t °C	1674.	5	5 la 15a a	rk	$\vdash$			1
	257.	5	<del> </del>    -					
Cryos. A° consts. B°			c <sub>p</sub> vap.	YK				
t <sub>e</sub> °C	128.99	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.75$	T <sub>c</sub>					grams/100 gran	ns solven	ıt
REFERENC	ES: 1-D	ow 2	API 3-Lit. 4	-Calc. from de	t. dat	ta 5-Calc, by for	mula	
SOURCE:			API					
PURIFICAT			API					
LITERATU	RE REFI	ERENC	ES:					

NAME   2-Methyl-cis-3-heptene   STRUCTU CH3(CH2)		снсн сн сн <sub>3</sub>	
Mole	to °K	Ċн <sub>3</sub>	
Mole	to °K	Ċн <sub>3</sub>	
Ref.	to °K		Ref
F.P. 100%  B.P. °C 760 mm 112. 100 53.31 4 30 27.41 4 30 mm 0.6486 5 1 -25.47 5  AHm cal/g  Pressure mm 25°C 26.48 5 te 1038.  BP 30 mm 30 mm 0.6486 5 k' 30 mm 0.6486 5 k' 30 mm 0.6486 5 k' 1 -25.47 5  AHv cal/g 25°C 82.42 5 0 1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -25°C 82.42 5 0 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1	to °K		
B.P. °C 760 mm 112.	to °K		
760 mm 112. 2 te 0.0463 4 h 1 100 53.31 4 te 0.0363 5 g'   30 27.41 4 30 mm 0.6486 5 h'   1 -25.47 5 AHm cal/g	to °K		
100	to °K		-
10 7.64 5 AHm cal/g	to °K		
1 -25.47 5 AHm cal/g  Pressure mm 25°C 26.48 5 5 30 mm 82.25 5 0  Density g/ml 20°C 0.706 2 d t 25 0.702 2 d 4 30 0.698 4 4 Ahv/T <sub>e</sub> 19.52 5  a 0.722 4 e 123 °C 0.1390 5 d d 27 to 84.19 5 Ref. Index p. 20°C 1.407 2 e 1.23 °C 0.0708 5	to		$\dagger$
Pressure mm 25°C 26.48 5 25°C 82.42 5 0 0  te 1038. 5 BP 70.49 5 m'	to	-	
The 1038. 5 BP 70.49 5 To 1038. To 1			1
Density g/ml 20°C 0.706 2 t e (d, e) 68.91 5 0' 19.52 5 Surface ter dynes/cm.  Ref. Index  Ref. Index  20°C 1.407 2 1.407 2 2 1.407 2			
g/ml 20°C   0.706   2   te (d,e)   68.91   5   0'	_ •ĸ		1
d4 30 0.698 4 DHv/Te 19.52 5 Surface ter dynes/cm.  Ref. Index P. 20°C 1.407 2 e' 27 °C 0.0708 5		-	
a 0.722 4 d 27 to 86.05 5 Surface tell 4 d 15 to 84.19 5 Ref. Index p. 20°C 1.407 2 e' 27 °C 0.0708 5		<u> </u>	4-
b -0.038 4 e 123 °C 0.1390 5 g/103/cm.  Ref. Index e' 27 °C 0.0708 5		19.77	5
Ref. Index	30	18.88	5
<sup>n</sup> D 20°C 1.407 2 d g/ml 0.230 5 Parachor	40	18.01	5
25   1 405   2   4 g/m   0.237   3	[P] 20°C		
30 1.402 4 v <sub>c</sub> ml/g 4.190 5 c °C 278.	30		
"C" 0.7679 4 Pa mm 18716. 5	40 Sugd.	. 335.2	5
MR (Obs.) 39.1 2 PV/RT Exp. L.1.		1555.2	+-
(nD-d/2) 1.054 2 25°C 1.0000 5 u.			١.
30 mm		119.	2
A   27 to   6 0/026   E    to     0.7402   2    _, _ , .		}	1
B   140 °C   1341.1   5   t <sub>c</sub>   0.256   5   Fire Point   C   217.72   5   ΔHc kcal/m   M. Spec.			+-
A + 27 to 1 39246 5 AHf			
B* 130 °C 1259.5 5 AFf Infrared	•		
K Viscosity Centistokes Solubility	in +		T
t <sub>k</sub> to   n ·c   Acetone	et.		ļ
Benzene	•		
B' 27 °C 1545 3	e		
C' 235.72 5 B' to Ethanol	•		-
The street of th			
Ac 140 to 7.3585 5 (A <sup>V</sup> ) °C		1	†
B- 4 0C 1451			
CC 251. 5 P		ĺ	ļ
Cryos. A° consts. B° c vap. °K			
t <sub>e</sub> °C 123.34 5 c <sub>v</sub> vap.		<u> </u>	
		ms solve	nt
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc.	by for	rmula	
SOURCE: API			
PURIFICATION: API LITERATURE REFERENCES:			

			•,-	CLWII	s-3-heptene			STRUCTURAL I	ORMUL	A.
								СН <sub>3</sub> (СН <sub>2</sub> ) <sub>2</sub> СН=С	нсн сн	
Mole		Ref.		ecul		Molecular Weight 112.2	08	0113(0112/2011-01	ċн <sub>3</sub>	3
<del>/************************************</del>	_			Ref			Ref			Ref
F. P. *C	Т		$\dashv$	****	44/45	Τ	-	( )		<del>                                     </del>
F.P. 100	5		_	$\dashv$	dt/dP *C/mm	1	l i	f to g *K		
B. P. *C	+				25°C	0.7214	5	h		
760 mm 100	-	112.	- 1	2	BP t <sub>e</sub>	0.0463 0.0363	4 5	f' to		+
30	ı	53.31 27.41	ı	4	30 mm	0.6486	5	g'   '°K_		1
10		7.64		5	ΔHm cal/g	1		h'		
1	+	-25.47		5	ΔHv cal/g	<del>                                     </del>	$\vdash$	m to		
Pressure mm 25°C	: 1	26, 48	ı	5	25°C	82.42	5	n •K		1
te		038.		5	30 mm BP	82.25 70.49	5 5			ــــــ
Density	$\top$				t	68.97	5	m'   to		
g/ml 20°0		0.700		2	t <sub>e</sub> (d, e)	68.91	5	n'   •K		İ
dt 25 4 30		0.69		2 4	AHv/Te	19.52	5	<u> </u>		
8	+	0,72		4	d 27 to		5	Surface tension dynes/cm, 20°C	19.77	5
Ъ		-0.03		4	$\frac{1}{a} - \frac{123}{15} = \frac{6}{15}$		5	y 30	18.88	5
Ref. Inde:					e' 27 °C		5	40	18.01	5
<sup>n</sup> D 20°		1.40		2 2	d <sub>c</sub> g/ml	0.239	5	Parachor [P] 20°C		
30		1.40		4	d g/ml vc ml/g tc °C	4.190	5	30		
"C"	T	0.76	79	4	, c	278.	5	40		1_
MR (Obs.	)	39.1	T	2	P <sub>c</sub> mm PV/RT	18716.	-	Sugd.	335, 2	5
MR (Calc (nD-d/2)	.)	38.67		5	25°C	1.0000	5	Exp. L.1.%/wt. u.		1
	+	1.054	*		30 mm	1.0000	5		121.	2
Dielectric		<u> </u>			BP t <sub>e</sub>	0.9500 0.9405	5 5	Flash Point °C		T
A 27 t		6.948 341.1	826	5	tc	0.256	5	Fire Point		╄
<u>c                                    </u>		217.72	l	5	ΔHc kcal/m			M Spec. Ultra V.		1
A*   27 t	0	1.39	246	5	ΔHf ΔFf			X-Ray Dif.		ł
B* 130 °	<u>C   17</u>	259.5	[	5	Viscosity	<del> </del>	╌┤	Infrared		
ċ	╝				centistokes	1		Solubility in +		1
k	c				η •c			Acetone Carbon tet.		
t⊊ i °		7, 350				1		Bensene		}
B' 27		7.350 545.3	,,,	5		<u> </u>		Ether n-Heptane		1
C'	7	235.72		5	B <sup>V</sup> to			Ethanol		
A'* 15 t B'* 27 *		1.748	894	5	A <sup>V</sup>   •C	_{		Water Water in		1
		445.8	<del>.  </del>	5	(B <sup>V</sup> ) to	1				+
Ac   140 t	cli	7.358 651.	85	5	(A <sup>V</sup> )  °C	<del> </del>				1
Cc	- 3	257.		5	c <sub>p</sub> liq. °K	1				ì
Cryos. A consts. B					c <sub>p</sub> vap. °K					
t <sub>e</sub> °C		123.34		5	c <sub>v</sub> vap.	-				
$T_R = 0$ .					·			grams/100 gran	ns solven	ıt
REFEREN	UES	s: 1-De	<b>w</b> 2	2-AF		Calc. from de	t. dat	a 5-Calc, by forr	nula	
SOURCE:				AP						
PURIFICA				AF						
LITERAT	JRE	REFE	EREN	CES	:					

							<b>N</b> o. 89	,
NAME _	3-Methyl-	cis-	3-heptene			STRUCTURAL	FORMUL	.A
						сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> сн=	с сн-сн	
Mole % Pur.		lecul rmuk		Molecular Weight 112.2	08	3. 2.2	Ċн <sub>3</sub>	3
	<del></del>	Ref.			Ref.			Ref
F.P. °C			dt/dP			f to		
F.P. 100%	<del> </del>	-	*C/mm 25*C	1.0203	5	g° <u>K</u>	-	
B. P. °C 760 mm	121.	2	BP	0.0472 0.0363	4 5	h	<del> </del>	₩
100 30	61.12 34.68	4	t <sub>e</sub>	0.6622	5	f' to		
10	14.50	5	30 mm ΔHm cal/g	0.0022	<del> </del>	h'		
1	-19.31	5	ΔHv cal/g	<del> </del>	<del>                                     </del>	m to		T
Pressure mm 25°C	18.09	5	25°C	85.30	5	n   •K	<u>-</u>	
t <sub>e</sub>	1062.7	5	30 mm BP	84.51 72.37	5 5			<u> </u>
Density			t.	70.69	5	m' to		
g/ml 20°C _t 25	0.728 0.724	2	Le (4, 6)	70.62	5	#,	1	
d <sup>t</sup> 25 4 30	0.720	4	ΔHv/T <sub>e</sub>	19.50	5	Surface tension	<del> </del>	╁
a	0.744	4	d 35 to	89.38 0.1405	5	dynes/cm. 20°C	22.36	5
ь	-0.038	4	d'   25 to	87.35	5	30 40	21.38	5
Ref. Index n <sub>D</sub> 20°C	1.418	2	e'   35 °C	0.0819	5	Parachor [P]	20.44	┼
25	1.416	2	d g/ml	0.246 4.071	5	20°C		
30	1.413	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	294.	5	30 40		1
"C"	0.7637	4	P <sub>c</sub> mm	19746.	5		. 335.2	5
MR (Obs.) MR (Calc.)	38.9 38.677	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.054	2	25°C 30 mm	1.0000 1.000 <b>0</b>	5	u. Dispersion	124.	2
Dielectric			BP	0.9495	5	Flash Point °C	+	┼╌
A 35 to B 153 °C	6.95770 1374.0	5	t c	0.9392 0.255	5	Fire Point		İ
c (133 0	216.	5	ΔHc kcal/m	<u> </u>	$\vdash$	M. Spec.		
A* 35 to	1.39323	5	ΔHf ΔFf	İ		Ultra V. X-Ray Dif.		ļ
B*[143 °C K	1290.9	5	Viscosity		-	Infrared		
c			centistokes			Solubility in + Acetone		
t <sub>k</sub> to			η •c			Carbon tet.		
A'   25 to	7, 35121	5			١.	Benzene Ether		
B' i_35 °C	1578.3	5	B <sub>v</sub> to			n-Heptane		
A'* 25 to	1,74671	5	B to ℃			Ethanol Water		
B' ≠ 35 °C	1478.6	5	(B <sup>V</sup> ) to	-		Water in	<u> </u>	
Ac   153 to	7.3680	5	(A <sup>V</sup> )  °C					
Bc tc °C	1693. 257.	5	c <sub>p</sub> liq. °K					
Cryos. A°	+ = = = = = = = = = = = = = = = = = = =	<u> </u>	1 -	1				
consts. B°	<u> </u>		P					
t <sub>e</sub> °C	1 <b>3</b> 3.48	5	c <sub>v</sub> vap.					
$T_{R} = 0.79$						grams/100 grams/		nt
	ES: 1-Dow	2-A		Calc. from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE:			PI.					
PURIFICAT			PI					
LITERATU	RE REFERE	NCE	S:					

							No. 90	)
NAME	3-м	ethyl-tra	ns-3-heptene			STRUCTURAL	FORMUL	A
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH=C	СН-СН	١.
14-1-	Ref.	14-1		Malagulag			H <sub>3</sub>	.3
Mole % Pur.	Rei.	Molecul Formu		Molecular Weight 112.2	80		3	
		Ref			Ref.			Ref
F.P. °C			dt/dP			f to		
F.P. 1009	-		*C/mm 25*C	1.0203	5	g <u> </u>		1
B. P. °C 760 mm	121.	2	BP	0.0472	4	h		↓
100	61.12	2 4	t <sub>e</sub>	0.0363	5	f' to		1
30 10	34.68 14.50		30 mm	0.6622	5	b		ļ
1	-19.31		ΔHm cal/g	<del></del>	$\sqcup$	m   to		+
Pressure			ΔHv cal/g 25°C	85.30	5	n   °K		
mm 25°C	18.09	5	30 mm	84.51	5	0		1
Density	1002.1	<del>-   -</del>	BP	72.37 70.69	5 5	m'   to		
g/ml 20°C			te te (d, e)	70.62	5	n'   <u>°K</u> _		İ
dt 25 4 30	0.72		AHV/Te	19.50	5	0 1		—
a	0.74		d   35 to		5	Surface tension dynes/cm. 20°C	22.36	5
b	-0.0		$\frac{1}{a}$ $\frac{1}{1}$ $\frac{133}{25}$ $\frac{9}{10}$		5 5	30	21.38	5
Ref. Index			e' 35 °C		5	40	20.44	5
n <sub>D</sub> 20°0	1.41		d g/ml	0.246	5	Parachor [P] 20°C		Ì
30	1.41		d g/ml vc ml/g tc °C	4.071 294.	5 5	30		
"C"	0.76	37 4	t <sub>c</sub> *C P <sub>c</sub> mm	19746.	5	40 Sugd	335.2	5
MR (Obs.	:	2	PV/RT	17740.	1-1	Exp. L.1.%/wt.	333.2	╁
MR (Calc. (nD-d/2)	38.67		25°C	1.0000	5	u.		
Dielectric		<u>,                                    </u>	30 mm BP	1.0000 0.9495	5 5	Dispersion	124.	2
A   35 t	6, 95	770 5	t <sub>e</sub>	0.9392	5	Flash Point °C Fire Point	ļ	
B 1153°9	1374.0	5	t <sub>c</sub>	0.255	5	M Spec.		+
C	216.	5	ΔHc kcal/m			Ultra V.	1	1
A*  35 to B*_143 °C		323   5	ΔFf	<u> </u>		X-Ray Dif. Infrared		
к — —	-		Viscosity			Solubility in +		+
¢	<u>-</u>		r centistokes	:		Acetone		1
t <sub>x</sub> ; •(			'			Carbon tet. Benzene		
A'   25 to			1			Ether		
B' L 35 °C	234.	5 5	B <sup>V</sup>   to			n-Heptane Ethanol		1
A'* 25 to		671 5	_AV_I°C			Water		
B'* 35 °C		5	(B <sup>V</sup> ) to			Water in	<b> </b>	+
Ac   153 to		5 5	(A <sup>V</sup> )  °C	:				1
Bc tc_	257.	5	c <sub>p</sub> liq. °K					
Cryos, A			c <sub>p</sub> vap. °K	:				-
consts. B			Ji -					
t <sub>e</sub> °C	133.48	5	c <sub>w</sub> vap.		Ш	1	<u> </u>	1
T <sub>R</sub> = 0.7		10m 2 4	DT 2 124 4	C-1- 6		grams/100 gran		<u>ıt</u>
SOURCE:	CEG: 1-D		PI <b>3-Lit. 4-</b> PI	Caic. Irom de	τ. dai	ta 5-Calc, by for	mula	
PURIFICA:	TION		PI					
LITERATU								
MAI	LE REF	-rence:	٠.					

No. 91 4-Methyl-cis-3-heptene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_2C = CHCH_2CH_3$ ċн<sub>3</sub> Mole Molecular Molecular C8H16 Weight 112, 208 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 1.0610 5 B. P. °C h BP 0.0473 122. 62.00 760 mm 2 0.0363 5 ſ١ to 100 4 g' <u>°К</u> 30 35.50 4 30 mm 0.6637 5 10 15.27 5 h' ∆Hm cal/g 5 -18.61 to m AHv cal/g Pressure ۰ĸ n 25°C 85.63 5 mm 25°C 17.33 5 o 84.76 30 mm 1066. 5 te ВP 72.59 5 m' | to Density 5 5 70.88 n' ۰ĸ g/ml 20°C te (d, e) 0.725 2 70.82 o'  $\mathbf{d_4^t}$ 25 0.721 2 5 AHV/Te 19.50 30 0.717 4 Surface tension d T 36 89.76 5 to 0.741 a 4 dynes/cm. 20°C 21.99 5 5 135 25 °C 0.1407 ь -0.038 4 30 21.03 5 ăΠ to 87.69 5 40 20.09 5 e' Ref. Index 36 0.0826 5 20°C [P] 1.417  $^{n}D$ Parachor 5 d<sub>c</sub> g/ml 0.244 25 1.415 2 20°C vc ml/g t\_ °C 4.096 5 30 1.412 4 30 <sup>t</sup>c 295. 5 40 "C" 0.7651 4 P<sub>c</sub> mm 19660. 5 Sugd. 5 335.2 MR (Obs.) 38.9 2 PV/RT Exp. L. l. %/wt. 38.677 MR (Calc.) 5 25°C 1,0000 (nD-d/2) u. 1.055 2 30 mm 1,0000 5 Dispersion 124. 2 Dielectric BP 0.9495 0.9391 5 Flash Point C 5 6.95905 A 36 to Fire Point 0.255 5 B (153 ℃ 1377.7 5 M. Spec. С 5 AHc kcal/m 215.82 Ultra V. ΔHf A\* 36 to B\* 145 °C 1.39354 5 X-Ray Dif. ΔFf 1294.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. °C  $^{t}_{\underline{\mathbf{x}}}$ Benzene 25 to 7.35168 Ether B١ 36 °C 1582.1 n-Heptane B<sub>v</sub> 5 C' 233.82 to Ethanol °C A'\* 25 to B'\* 36 °C Water 1.74680 5 Water in 1482.4 (B<sup>V</sup>) to Ac | 153 to 7,3692 5 (A<sup>V</sup>)| °C Bc tc 1697. °C c<sub>p</sub> liq. ۰ĸ Cc 256. 5 Cryos. Aº c vap. °K consts. B° c<sub>v</sub> vap. te °C 134.61 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

							No. 92	
NAME	4-Methy	l-trar	ns-3-heptene			STRUCTURAL		
					_	$CH_3(CH_2)_2C =$	сн <mark>сн</mark> 2сі	1 <sub>3</sub>
Mole % Pur.		olecul ormul		Molecular Weight 112.2	.08	ċн <sub>3</sub>		
		Ref.			Ref			Ref.
F, P. /C		1	dt/dP			f   to		
F.P. 100%			°C/mm		ا ۔ ا	g <u>°K</u> _	Į	
B. P. *C			25°C BP	1.0610 0.0473	5 4	h '	1	<u>L</u>
760 mm 100	122. 62.00	2 4	t	0.0363	5	f' to		
30	35.50	4	30 mm	0.6637	5	g'	ł	1
10 1	15.27 -18.61	5	∆Hm cal/g			h'		ـــــ
Pressure		+	ΔHv cal/g			m to	].	
mm 25°C	17.33	5	25°C 30 mm	85.63 84.76	5	"	İ	
t <sub>e</sub>	1066.	5	BP	72.59	5	m' l to	<del></del>	╁
Density g/ml 20°C	0.725	2	t <sub>e</sub>	70.88	5	m' to	}	]
dt 25	0.721	2	te (d, e)	70.82	5	0'	1	
<b>4</b> 30	0.717	4	ΔHv/T <sub>e</sub>	19.50	5	Surface tension	<b></b>	<b>—</b>
	0.741	4	d   36 to		5	dynes/cm. 20°C	21.99	5
ь	-0.038	4	d'   25 to	87.69	5	30 40	21.03 20.09	5
Ref. Index	1.417	2	e'   36 °C	+	5	Parachor [P]	20.07	+
D 25	1.415	2	d g/ml v ml/g	0.244 4.096	5	20°C		
30	1.412	4	tc •C	295.	5	30 40	<b>!</b>	1
"C"	0.7651	4	P <sub>c</sub> mm	19660.	5		335.2	5
MR (Obs.) MR (Calc.	38.9	2 5	PV/RT		$\vdash$	Exp. L.1.%/wt.		
(nD-d/2)	38.677 1.055	2	25°C	1.0000	5	u.	l	
Dielectric			30 mm BP	1.0000 0.9495	5	Dispersion	124.	2
A   36 to	6,95905	5	t <sub>e</sub>	0.9391	5	Flash Point °C Fire Point	l	1
B 1153_°C	1377.7	5	tc	0.255	5	M Spec.	<u> </u>	<del> </del>
C	215.82	5	ΔHc kcal/m ΔHf	j		Ultra V.	1	1
A*  36 to B* <sub>1</sub> 145 °C		5	ΔFf			X-Ray Dif. Infrared		1
к — — —	1-75		Viscosity			Solubility in +	<del> </del>	╁
t <sub>k</sub>  to	-		centistokes 7°C		١.,	Acetone		ł
tx c			7 °⊂	1		Carbon tet. Benzene	İ	1
A'   25 to		5		1		Ether		i
B' ∟36 °C	1582.1 233.82	5	B <sup>V</sup>   to	<u> </u>	-	n-Heptane Ethanol	ł	
A'* 25 to		5	A <sup>v</sup> C			Water		1
B' # 36 °C		5	(BV) to	1	1 :	Water in		<u> </u>
Ac   153 to		5	(A <sup>V</sup> ) °C					1
Bc tc C	1	5	c <sub>p</sub> liq. •K	1				1
Cryos, A*	256.	+-	11	i	i i		İ	Ì
consts. B	1		c <sub>p</sub> vap. *K					Ì
t <sub>e</sub> °C	134.61	5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 T <sub>C</sub>			4		grams/100 grai	ns solven	t
		2-AI	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc. by for		
SOURCE:		AI						
PURIFICAT	ION:	AF	PI					
	RE REFERE	NCES	:					

							No. 93
NAME _	5-Methyl-	cis-	3-heptene			STRUCTURAL	FORMULA
			-			сн <sub>3</sub> сн <sub>2</sub> сн сн=	сн сн <sub>2</sub> сн <sub>3</sub>
Mole % Pur.		lecul mula	ar C <sub>8</sub> H <sub>16</sub>	Molecular Veight 112,20	08	сн <sub>3.</sub>	
		Ref.			Ref.		Res
F. P. °C			dt/dP			f to	
F.P. 100%			°C/mm		ا ۔ ا	g° <u>K</u>	]
B. P. °C	١.,,		25°C BP	0.7214 0.0463	5 4	h	L
760 mm 100	112. 53.31	2 4	te	0.0363	5	f' to	
30	27.41	4	30 mm	0.6486	5	g' <u>*</u>	
10 1	7.64 -25.47	5	ΔHm cal/g			h'	
Pressure			ΔHv cal/g		_	m to	
mm 25°C	26.48	5	25°C 30 mm	82.42 82.25	5	0	
t <sub>e</sub>	1038.	5	BP	70.49	5	m' to	
Density g/ml 20°C	0.713	2	t <sub>e</sub> (d, e)	68.97 68.92	5	n'ı ek	
at 25	0.709	2	ΔHv/T <sub>e</sub>	19.52	5	o'	1
<b>4 3</b> 0	0. <b>70</b> 5	4			5	Surface tension	
a b	0.729	4	d 27 to	86.05 0.1389	5	dynes/cm. 20°C	20.57 5
Ref. Index	-0.038	4	d' 15 to	84.19	5	8 30 40	19.65 5
n <sub>D</sub> 20°C	1.410	2	e!   27 °C	0.0709	5	Parachor [P]	
20	1.408	2	d g/ml vc ml/g	0.242 4.140	5	20°C	
30 "C"	1.405	4	tc °C	279.	5	30 40	
	0.7657	4	P <sub>c</sub> mm	18976.	5	11	335.2 5
MR (Obs.) MR (Calc.)	39.0 38.677	5	PV/RT			Exp. L.1.%/wt.	
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion	119. 2
Dielectric			BP	0.9500	5	Flash Point °C	-
A 27 to	6.94826	5	t <sub>e</sub>	0.9405 0.256	5	Fire Point	
B (141 °C	1341.1 217.72	5	tc ΔHc kcal/m	0.230	-	M. Spec.	
A* 27 to	1.39246	5	ΔHf	i		Ultra V. X-Ray Dif.	
B* 133 °C	1259.5	5	ΔFf			Infrared	
K ———			Viscosity centistokes			Solubility in +	
t <sub>k</sub> Tto	İ		η °C			Acetone Carbon tet.	
* L			•			Benzene	
A'   15 to B'   27 °C	7.35001 1545.3	5				Ether n-Heptane	
č, '	235.72	5	B <sub>v</sub> to			Ethanol	
A'* 15 to	1.74894	5	ĂV I C	}		Water	
B'* 27 °C	1445.8	5	(B <sup>V</sup> )  to		1	Water in	+
Ac 141 to Bc t <sub>c</sub> °C	7.3588 1652.	5	(A <sup>V</sup> )  °C		<u> </u>		
Cere-	257.	5	c <sub>p</sub> liq. °K				
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	123,34	5	c vap.				
$T_{\mathbf{p}} = 0.75$			<u> </u>	L		grams/100 gra	ms solvent
<del></del>		2-A	PI 3-Lit. 4-	Calc. from de	t. da	ita 5-Calc. by for	
SOURCE:			PI				
PURIFICAT	ION:		PI				
	RE REFERE	NCES	5:				
1							

							No. 94	
NAME	5-Methyl	-tran	ns-3-heptene			STRUCTURAL	F <b>ORM</b> UL/	A.
						CH3CH2CH CH=C	н сн,сн	,
Mole	Ref. Mo	lecul		Molecular		сн <sub>з</sub>	2	3
% Pur.		rmul		Weight 112.2	08			
		Ref.		,	Ref			Ref.
F.P. °C	<b></b>		dt/dP			f to		
F. P. 100% B. P. °C	<del> </del>	ļ	*C/mm 25*C	0.7214	5	g <u>•K</u> _		
760 mm	112.	2	BP	0.0463	4	h f' to		╁
100 30	53.31 27.41	4	t <sub>e</sub> 30 mm	0.0363	5 5	f' to g'°K_		
10	7.64	5	ΔHm cal/g	0.6486	-	h'		
1	-25.47	5	ΔHv cal/g		$\vdash$	m to		
Pressure mm 25°C	26.48	5	25°C	82.42	5	n •K		
t <sub>e</sub>	1038.	5	30 mm BP	82.25 70.49	5 5			
Density g/ml 20°C	0,713	2	t_	68.97	5	m'   to		
at 25	0.709	2	te (d, e)	68.92	5	0'   = -		
	0.705	4	ΔHv/T <sub>e</sub>	19.52	5	Surface tension		$\vdash$
a b	0.729 -0.0 <sub>3</sub> 8	4	e   123 °C	86.05 0.1389	5	dynes/cm. 20°C	20.57	5
Ref. Index	1 -0.030	Ť	d'   15 to e'   27 °C	84.19 0.0709	5 5	30 40	19.65 18.76	5
n <sub>D</sub> 20°C	1	2	d g/ml	0.0707	5	Parachor [P]		
25 30	1.408	2 4	V mi/g	4.140	5	20°C 30		
"C"	0.7657	4	°C	279.	5	40		
MR (Obs.)	39.0	2	P <sub>c</sub> mm	18976.	5	Sugd.	335.2	5
MR (Calc.) (nD-d/2)	38.677 1.054	5 2	25°C	1.0000	5	Exp. L.1.%/wt.		
Dielectric	1.054	-	30 mm BP	1.0000 0.9500	5	Dispersion	121.	2
A   27 to	6.94826	5	te	0.9405	5	Flash Point °C Fire Point		
B [ 141 °C	1341.1	5	t <sub>c</sub>	0.256	5	M Spec.	<u> </u>	╁──
C 27.45	217.72	5	ΔHc kcal/m ΔHf			Ultra V.	ĺ	l
A* 27 to B* 133 °C	1.39246 1259.5	5	ΔFf			X-Ray Dif. Infrared		
K ———	1		Viscosity			Solubility in +	l	$\dagger$
t <sub>k</sub> to	-		7 °C			Acetone Carbon tet.		1
x I			'			Benzene	ļ	
A'   15 to B' <u>27</u> °C		5				Ether n-Heptane		-
C'	235.72	5	.B <sup>V</sup>   to			Ethanol		1
A'* 15 to B'* 27 °C	1.74894	5	AV I C			Water Water in		
B'* 27 °C	7, 3588	5	(B <sup>V</sup> ) to					1
Bc t °C	1652.	5		ļ	$\vdash$			
Cc — —	257.	5	c <sub>p</sub> liq. ∘K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	123. 34	5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 T <sub>c</sub>			L		f grams/100 gran	ns solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc. from det	t. dat	ta 5-Calc. by for		
SOURCE:		Al						
PURIFICAT	'ION:	Al	PI					
LITERATU	RE REFERE	VCES	3:					

No. 95 6-Methyl-cis-3-heptene NAME STRUCTURAL FORMULA CH3CH CH2CH=CHCH2CH3 ĊН3 Mole Ref. Molecular Molecular C8H16 % Pur Formula Weight 112.208 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.8098 5 B. P. ℃ h ВP 4 0.04662 760 mm 115. 55.93 2 t<sub>e</sub> 0.03630 5 f١ to 100 4 g' <u>°К</u> 30 29.85 4 30 mm 0.6530 5 10 9.94 5 h' ∆Hm cal/g 5 -23.40 1 m to AHv cal/g Pressure n °K 25°C 83.39 5 5 mm 25°C 23.32 5 5 0 30 mm 83.02 te 1046. BP 71.15 5 m' Density to 69.57 5 n' °K g/ml 20°C 0.713 2 te (d, e) 5 69.51 o'  $\mathbf{d_{4}^{t}}$ 25 0.709 2 ΔHv/T<sub>e</sub> 5 19.52 30 0.705 4 Surface tension 30 d 87.18 5 a 0.729 4 dynes/cm. 20°C 20.57 127 °C 0.1394 5 ь 4 -0.038 30 19.65 5 ăΠ 20 to °C 85.31 5 40 18.76 5 e' | Ref. Index 30 0.0768 5 20°C 1.410 2  $n_{D}$ Parachor [P] d<sub>c</sub> g/ml 0.241 5 25 1.408 2o°C vc m1/g t °C 4.152 30 1.403 4 30  $t_c$ 283. 5 "C" 40 0.7657 4 P<sub>c</sub> mm 5 19060. Sugd. 335.2 5 MR (Obs.) 39.0 2 PV/RT Exp. L. l. %/wt. MR (Calc.) (nD-d/2) 38.677 5 25°C 1.0000 5 u. 1.054 2 30 mm 1.0000 5 Dispersion 2 119. Dielectric BP 0.9500 Flash Point C te tc 0.9402 A 30 to 6.95245 5 Fire Point 0.256 B 1144 °C 5 1352.4 M. Spec. С 217.15 5 AHc kcal/m Ultra V. ΔHf A\* 30 to B\* 134 °C 1.39344 5 X-Ray Dif. ΔFf 1270,2 5 Infra red ĸ Viscosity Solubility in c centistokes Acetone to tk tx ٠c Carbon tet. Benzene 20 to 7.35146 Ether B١ 3<u>0 °C</u> 5 1556.7 n-Heptane B<sub>v</sub> | C 235.15 5 to Ethanol °C 1.74923 Water A1\* 20 to 30 °C 5 Water in B'\* (BV) 1457.1 5 to Ac | 144 to 7.3627 5 (A<sup>V</sup>)| °C 1665. Bc tc °C 5 °C ۰ĸ c<sub>p</sub> liq. 257. 5 Cryos. A° c<sub>p</sub> vap. °K consts. B° c<sub>v</sub> vap. te °C 126,72 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							<b>io.</b> 96
NAME	6-Methyl	-trar	ns-3-heptene			STRUCTURAL FO	RMULA
						CH <sub>3</sub> CH CH <sub>2</sub> CH=CH	CH <sub>2</sub> CH <sub>3</sub>
Mole % Pur.		lecul rmul		Molecular Weight 112.2	08	Ċн <sub>3</sub>	
/ Fui.	1 1 1 1 1	Ref.		weight 112.2	Ref	1	Ref.
F.P. °C	T		dt/dP	1		f to	
F.P. 1007	6		*C/mm		l	g L _ °K	
B. P. °C	f		25°C BP	0.8098 0.04662	5 4	h	
760 mm 100	115. 55.93	2	t,	0.0363	5	f' to	
30	29.85	4	30 mm	0.6530	5	g'	
10 1	9.94	5	ΔHm cal/g			h'	
Pressure	-23.10	<u> </u>	ΔHv cal/g			m to	
mm 25°C		5	25°C 30 mm	83.39 83.02	5	ö	
t <sub>e</sub>	1046.	5	BP	71.15	5	m'   to	<del></del>
Density g/ml 20°0	0.713	2	te te (d, e)	69.57 69.51	5	n'   oK	
dt 25 4 30	0.709	2	ΔHv/T	19.52	5	0'	
a 30	0.705	4	d   30 to	87.18	5	Surface tension	0.57 5
Ъ	-0.038	4	$\frac{127}{a}$ , $\frac{1}{20}$ to		5		0.57   5 9.65   5
Ref. Index		T	d' 20 to		5	40 1	8.76 5
n <sub>D</sub> 20°0	1.410 1.408	2 2	d <sub>c</sub> g/ml	0.241	5	Parachor [P] 20°C	
30	1.403	4	V mi/g	4.152 283.	5	30	
"C"	0.7657	4	-	19 <b>0</b> 60.	5	40 Sugd. 33	5.2 5
MR (Obs.		2	P <sub>c</sub> mm PV/RT	1 7000.	ļ	Exp. L.1.%/wt.	5.2
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.	
Dielectric		Ť	30 mm BP	1.0000 0.950 <b>0</b>	5	Dispersion 12	1. Z
A 30 t		5	te	0.9402	5	Flash Point °C Fire Point	
B 1144*	217.15	5	t <sub>c</sub>	0,256	5	M Spec.	
A*  30 to	<del></del>	5	ΔHc kcal/m ΔHf			Ultra V.	
B* 134 °		5	ΔFf			X-Ray Dif. Infrared	
K C			Viscosity centistokes	1		Solubility in +	
t <sub>k</sub>  t		1	7 °C			Acetone Carbon tet.	
t <sub>x</sub> i		<u> </u>	<b>,</b> '			Benzene	
A'   20 to		5			<u></u>	Ether n-Heptane	
C' - 3	235.15	5	B <sup>V</sup> to		1	Ethanol	
A'* 20 to B'* 30 °		5	ĂV C	- -		Water Water in	
B'* 30 °	·	5	(B <sup>V</sup> ) to				
Bc tc	C 1665.	5	(A <sup>V</sup> )  °C	<del> </del>	-	4	
Cc — —	257.	5	c <sub>p</sub> liq. °K				
Cryos. A'			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	126.72	5	c <sub>v</sub> vap.				
$T_R = 0.7$	75 <b>T</b> c					+ grams/100 grams	solvent
REFEREN	CES: 1-Dow			Calc, from de	t. da	ta 5-Calc, by formu	
SOURCE:		A.	PI				
PURIFICA			PI				
LITERATU	IRE REFERE	NCES	5:				

No. 97 2-Ethyl-1-hexene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_3C = CH_2$ Molecular C8H16  $C_2H_5$ Mole Ref. Molecular Weight 112,208 % Pur. Ref Ref. Ref. F.P. °C F.P. 100% dt/d₽ f to •c 1 g °Κ 25% 0.9813 5 B. P. ℃ h BP 0.0471 4 760 mm 120. 60.25 2 te 0.0363 5 f١ 4 to 100 33.87 g' <u>°К</u> 4 30 30 mm 0.6607 5 13.73 5 10 h' ∆Hm cal/g -20.00 5 m to AHv cal/g Pressure ۰ĸ n 25°C 84.97 5 mm 25°C 18,88 5 o 30 mm 84.25 5 1060. 5 t<sub>e</sub> ВP 5 72.15 m to Density te te (d, e) 70.48 n' ۰ĸ g/ml 20°C 0.7270 2 70.42 5 o' 25  $d_4^t$ 0.7228 2 AHv/Te 19.50 5 30 0.7186 4 Surface tension d 34 to 89.00 5 0.7438 a 4 dynes/cm. 20°C 22,24 132 °C 5 å 0.1404 ь -0.0383 4 5 25 to 30 21.22 5 87.00 40 20.23 5 Ref. Index e' 34 °C 0.0812 5 <sup>n</sup>D 20°C 1.4157 [P] Parachor d<sub>c</sub> g/ml 0.244 5 25 2 1.4132 20°C vc ml/g t\_ °C 5 4.101 30 1.4106 4 30 <sup>t</sup>c 292. 5 40 "C" 0.7608 4  $P_c$  mm 19527. 5 Sugd. 335.2 5 MR (Obs.) 38,71 PV/RT Exp. L.1.%/wt. MR (Calc.) 5 38,677 25°C 1.0000 5 (nD-d/2)1.0522 2 30 mm 1.0000 Dispersion 121. 2 Dielectric ВP 0.9495 5 Flash Point C 5 0.9393 A 34 to 6.95634 Fire Point 0.255 5 B (\_151 ℃ 1370.2 5 M. Spec. С 5 AHc kcal/m 216.2 Ultra V. ΔHf A\* 34 to 5 1.39292 X-Ray Dif. ΔFf B\* 142 °C 1287.3 Infrared Viscosity Solubility in centistokes c Acetone to Carbon tet. t<u>x</u>\_ °C Benzene A' 7.35072 25 to Ether В' 34 °C 1575. 5 5 n-Heptane B<sub>v</sub> | C' 234.2 to Ethanol °C Water A1\* 25 to 1.74662 5 Water in B'\* 5 (B<sup>V</sup>) 34 °C 1474.8 to Ac | 151 to 7.3666 5 (A<sup>V</sup>)| °C Bc tc °C 1688. 5 c<sub>p</sub> liq. °K 257. 5 Cryos. A° °K c<sub>p</sub> vap. consts. B° c<sub>v</sub> vap. te °C 132.35 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

						1	No. 98
NAME	3-Ethyl-l	-hex	ene			STRUCTURAL FO	RMULA
						כם (כם ) כם כו	u-Cn
Mala	7-6 14-	lecul		Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH Cl	11-0112
Mole % Pur.		rmul		Molecular Weight 112.2	08	25	
		Ref.	E		Ref		Ref
F.P. *C			dt/dP			f to	
F.P. 100%			*C/mm 25*C	0.6758	5	g <u>*K</u> _	
B. P. °C 760 mm	110.3	2	BP	0.0462	4	h	
100	51.82	4	t <sub>e</sub>	0.0363	5	f' to g' 'K	
30 10	26.02 6.32	4 5	30 mm	0.6461	5	g'   'K_	
1	-26.66	5	ΔHm cal/g			<del> </del>	
Pressure			ΔHv cal/g 25°C	81.86	5	m to	
mm 25°C	28.46 1033.	5	30 mm	81.80	5	0	
t <sub>e</sub> Density	1033.	-	BP	70.12 68.63	5 5	m'   to	
g/ml 20°C	0.715	2	te te (d, e)	68.58	5	n'   <u>*</u> K	
dt 25 4 30	0.711 0.707	2 4	AHv/Te	19.51	5		
4 JU	0.731	4	d   26 to	85.41	5	Surface tension dynes/cm, 20°C 2	0.80 5
Ъ	-0.038	4	e   121 °C d'   5 to		5 5	30 1	9.87 5
Ref. Index			e' 26 °C		5		8.97 5
n <sub>D</sub> 20°C	1.407 1.405	2 2	d <sub>c</sub> g/ml	0.243	5	Parachor [P]	
30	1.402	4	d g/ml vc ml/g tc °C	4.121 277.	5 5	30	
"C"	0.7582	4	P <sub>c</sub> mm	18997.	5	40 Sugd. 33	5.2 5
MR (Obs.)	38.7	2	PV/RT	10771.	<del>                                     </del>	Exp. L.1.%/wt.	3.2
MR (Calc.) (nD-d/2)	38.677 1.050	5 2	25°C	1.0000	5	u.	İ
Dielectric		╅	30 mm BP	1.0000	5 5		7. 2
A 26 to	6.94556	5	te	0.9406	5	Flash Point °C Fire Point	
B 1140°C	1334.6 218.	5	t <sub>c</sub>	0.256	5	M Spec.	
C A*  26 to	1.39159	5	ΔHc kcal/m ΔHf			Ultra V.	
B* 131°C		5	ΔFf			X-Ray Dif. Infrared	
к ———			Viscosity			Solubility in +	
t <sub>k</sub>			centistokes 7 °C			Acetone	
t <sub>x</sub> i *C						Carbon tet. Benzene	
A'   5 to B' <u>26 °C</u>	7.34887	5				Ether	
c, _ 20 _	1538.8 236.	5	B <sup>V</sup>   to			n-Heptane Ethanol	
A** 5 to	1.74852	5	AV   °C			Water Water in	
B'* 26 °C		5	(B <sup>V</sup> ) to		1 1	water in	<del></del>
Ac   140 to	7.35634 1644.	5	(A <sup>V</sup> )  °C		$oxed{oxed}$		
Cc	257.	5	c <sub>p</sub> liq. ∘K	ļ			
Cryos, A° consts, B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	121.42	5	c <sub>w</sub> vap.				
$T_R = 0.75$	Tc		<b>-</b>	<del></del>		+ grams/100 grams	solvent
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from de	t. dat	ta 5-Calc, by formu	la
SOURCE:		A.	PI				
PURIFICAT			PI				
LITERATUR	E REFERE	NCES	3:				

							No. 99	
NAME	4-Ethyl-	l-he	kene			STRUCTURAL	FORMUL	A.
	-					сн <sub>3</sub> сн <sub>2</sub> сн сн	-CH=CH-	
Mole % Pur.		ecul:		Molecular Weight 112,20	08	c <sub>2</sub> H <sub>5</sub>	22	
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%			dt/dP °C/mm			f to		
B.P. °C 760 mm 100 30 10	113. 54.18 28.22 8.40 -24.78	2 4 4 5 5	25°C BP te 30 mm  ΔHm cal/g	0.7497 0.0464 0.0363 0.6501	5 4 5 5	g		
Pressure mm 25°C t <sub>e</sub>	25.38 1040.	5 5	25°C 30 mm BP	82.74 82.50 70.74	5 5 5	n•K		
Density g/ml 20°C dt 25 d4 30	0.726 0.722 0.718	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	69. 20 69. 14 14. 69	5 5 5	m' to n' K o' Surface tension		
a b	0.742 -0.0 <sub>3</sub> 8	4	d 28 to e 110 °C d' 15 to	86.51 0.1396 84.58	5 5	dynes/cm. 20°C 30 40	22.11 21.14 20.19	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30	1.412 1.410 1.407	2 2 4	e' 28 °C d g/ml vc ml/g tc °C	0.0737 0.201 4.985 264.	5 5 5	Parachor [P] 20°C 30		
"C"	0.7554	4	P <sub>c</sub> mm	15331.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	38.5 38.677 1.049	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9500	5 5 5	Exp. L.1.%/wt. u. Dispersion	117.	2
A 28 to B   130 °C C	6.94966 1344.9 217.53	5 5 5	t e t c	0. 9400 0. 256	5	Flash Point C Fire Point M. Spec.		
A* 28 to B* 120 °C K c tk to c tx °C	1.39289	5 5	ΔHc kcal/m ΔHf ΔFf  Viscosity centistokes γ °C			Ultra V. X-Ray Dif. Infrared  Solubility in + Acetone Carbon tet.		
A'   15 to B'   28 °C C'	7.35049 1549.1 235.53	5 5 5	B <sup>V</sup> to			Benzene Ether n-Heptane Ethanol		
A'* 15 to B'* 28 °C	1.74901 1450.0	5 5	$\frac{\mathbf{A}^{\mathbf{V}}}{(\mathbf{B}^{\mathbf{V}}) } - \frac{\mathbf{C}}{\mathbf{to}}$			Water Water in		
Ac 130 to Bc t <sub>c</sub> °C Cc	7.3559 1642. 254.	5 5 5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	124.46	5	c <sub>w</sub> vap.					
T <sub>R</sub> = 0.75		2 **	DI 2 I': .	C-1- ( :		grams/100 gra		t
SOURCE:	ES: I-Dow		PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	mula	
PURIFICAT	ION:		PI					
	RE REFERE							

							No. 10	0
NAME	2, 3-Dime	thyl-	l-hexene			STRUCTURAL 1	FORMULA	
							CH <sub>3</sub>	
24-1-	24 14			Malandan		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH	Ċ = CH <sub>2</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	80	Ċн <sub>3</sub>		
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		
F.P. 100%			*C/mm 25*C	0.6817	5	8		
B. P. °C 760 mm	110.5	2	BP	0.0462	4	h		
100	52.02	4	t <sub>e</sub>	0.0363	5	f' to		
30 10	26, 21 6, 51	4 5	30 mm	0.6463	5	h'		
1	-26.48	5	ΔHm cal/g		L	m   to		-
Pressure			ΔHv cal/g 25°C	81.96	5	n  *K_		
mm 25°C	28.18 1033.	5	30 mm	81.89	5	° '		
Density		1	BP te	70.20 68.70	5 <b>5</b>	m'   to		
g/ml 20°C	0.7214	2	le (a, e)	68.65	5	n'   ' *K-		
dt 25 4 30	0.7172 0.713 <b>0</b>	2 4	ΔHv/T <sub>e</sub>	19.52	5	<u> </u>		
	0.7382	4	d   26 to	85.52	5	Surface tension dynes/cm. 20°C	21.55	5
ъ	-0.0 <sub>3</sub> 83	4	-•  -122 °C  -•  -122 ℃	0.1387 83.50	5	30	20.56	5
Ref. Index	1.4113	2	e' 26 °C	0.0617	5	40	19.58	5
n <sub>D</sub> 20°C 25	1.4089	2	d <sub>c</sub> g/ml	0.244	5	Parachor [P] 20°C		ļ
30	1,4063	4	v <sub>c</sub> ml/g t <sub>c</sub> C	4.099 278.	5	30		
"C"	0.7590	4	P <sub>c</sub> mm	19130.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.)	38.64 38.677	2 5	PV/RT		<del>                                     </del>	Exp. L.1.%/wt.		
(nD-d/2)	1.0506	2	25°C 30 mm	1.0000	5	u. Dispersion	122.	2
Dielectric			BP	0.9500	5	Flash Point °C	122.	-
A 26 to	6.94766	5	te t <sub>c</sub>	0.9406 0.256	5 <b>5</b>	Fire Point		
B <u> </u>	1336.0 218.	5	ΔHc kcal/m		<del>                                     </del>	M Spec.		
A*   26to	1,39348	5	ΔHf		}	Ultra V. X-Ray Dif.		
B* 130°C	1254.6	5	ΔFf		├	Infrared		
c			Viscosity centistokes		Ì	Solubility in +		
tk to C			η •c			Acetone Carbon tet.		ĺ
A' 15to	7, 35087	5	ŀ			Bensene	1	
B' _ 26°C	1540.2	5			<u> </u>	Ether n-Heptane		
C'	236.	5	B <sup>V</sup>   to A <sup>V</sup>   *C			Ethanol Water		
A'* 15 to B'* 26°C	1.750 <b>34</b> 1440.7	5	17.V	1		Water in		L
Ac   140to	7.3585	5	(A <sup>v</sup> )  •C	1				
Bc t C	1646.	5	c <sub>p</sub> liq. °K	<del> </del>	<del>                                     </del>			
Cc	257.	5		1	1			
Cryos. A° consts. B°			c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	121.64	5	c <sub>v</sub> vap.					
$T_R = 0.75$	T <sub>c</sub>			<u> </u>		+ grams/100 gran	ns solven	
		2-AI	PI 3-Lit. 4-0	alc, from det		ta 5-Calc, by for		
SOURCE:			PI					
PURIFICAT			PI					
LITERATUR	E REFERE	CES	:					

No. 101 2, 4-Dimethyl-1-hexene STRUCTURAL FORMULA NAME  $CH_3CH_2CH CH_2C = CH_2$ CH<sub>3</sub> ĊH<sub>3</sub> Mole **Molecular** Ref. Molecular C8H16 % Pur Formula Weight 112.208 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ۰ĸ g 25°C 0.6991 5 B. P. °C BP 0.0463 **4** 5 760 mm 111.2 2 0.0363 f١ to 100 52.60 4 g' •<u>к</u> 30 26.74 4 30 mm 0.6475 5 10 7.00 **5** h' ∆Hm cal/g 1 -**26.** 05 m to ΔHv cal/g Pressure n ۰ĸ 25°C 82,13 27.42 mm 25°C 5 o 30 mm 82.02 5 1035. 5 t<sub>e</sub> 70.30 5 BP m' to Density 68.80 5 te te (d, e) n' <u>°K</u> g/ml 20°C 5 0.720 68.75 ۰, 25 ž 0.716  $d_{\mathbf{4}}^{\mathbf{t}}$ ΔHv/T<sub>e</sub> 19.51 5 30 0.712 4 Surface tension 27 85.73 5 0.736 4 dynes/cm. 20°C 21.39 5 122 0.1387 5 •C ь -0.038 4 20.44 30 a 15 83.82 to 40 19.52 5 e¹ Ref. Index •c 0.0672 5 27 20°C 1.411 [P] nD Parachor d<sub>c</sub> g/ml 0.244 5 25 1.409 2 20°C vc ml/g t °C 4.093 30 1.406 4 30 279. 5 40 "C" 4 0.7600 P<sub>c</sub> mm 5 19195. 5 Sugd. 335.2 MR (Obs.) 38.7 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2) 1.051 2 30 mm 2 1.0000 5 Dispersion 122. Dielectric BP 0.9500 Flash Point C 0.9406 A 27 to 6.94593 Fire Point 0.256 B (141 °C 1337.7 M. Spec. C 217,87 AHc kcal/m Ultra V ΔHf A\* 27 to 1.39098 X-Ray Dif. ΔFf B\*[130 °C 1256.3 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. °C Benzene A' 15 to 7.34836 Ether B 27 °C 1541.8 5 n-Heptane B<sup>V</sup> A C' 235.87 5 to Ethanol °C Water A'\* 15 to 1.74767 Water in (BV) B'\* 27 °C 1442.4 5 to Ac | 141 to 7.3568 5 (A<sup>V</sup>)| °C Bc Ltc 1649. 5 c<sub>p</sub> liq. ۰ĸ Cc 5 257. Cryos, A° c<sub>p</sub> vap. ۰ĸ consts, B° c<sub>v</sub> vap. te °C 5 122.44  $T_R = 0.75 T_c$ grams/100 grams solvent 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula REFERENCES: 1-Dow SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 102
NAME	2, 5-Dim	ethyl	-1-hexene			STRUCTURAL FO	
						сн <sub>3</sub> сн (сн <sub>2</sub> ) <sub>2</sub> с	= CH
Mole % Pur.	Ref. Mo	lecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.	208		H <sub>3</sub>
	نضاسا	Ref.			Ref		Re
F.P. °C			dt/dP			f to	
F.P. 100%			*C/mm	0.7103	5	g <u>*K</u>	
B. P. *C 760 mm	111 4	2	25°C BP	0.7102 0.0463	4	h	
100	111.6 52.95	4	t <sub>e</sub>	0.0363	5	f' to	
30 10	27.07	4 5	30 mm	0.6480	5	g'K_	
i	7.32 -25.76	5	ΔHm cal/g			h¹	
Pressure			ΔHv cal/g		5	m to	
mm 25°C	26.95	5	25°C 30 mm	82.28 82.13	5	•	
Donaltu	1036.	5	BP	70.40	5	m'   to	
Density g/ml 20°C	0.7172	2	te (d, e)	68.88 68.83	5	n'  °K_	
dt 25 d4 30	0.7129	2	ΔHv/T <sub>e</sub>	19.51	5	o'	
	0.7086	4	d   27 to		5	Surface tension	
a b	0.7344 -0.0 <sub>3</sub> 85	4	<u>• 1 _123 °C</u>	0.1388	5	dynes/cm. 20°C	21.06   5
Ref. Index		<b>†</b>	d'   10 to		5	40	19.08 5
n <sub>D</sub> 20°C	1.4105	2		0.241	5	Parachor [P]	
25 30	1.4080 1.4057	2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	4.148	5	20°C	
"C"	0.7620	4		278.	5	40	
MR (Obs.)	38,80	2	P <sub>c</sub> mm	18908.	5		35.2 5
MR (Calc.)	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	
(nD-d/2)	1.0519	2	30 mm	1.0000	5		22. 2
Dielectric A 27 to	,	-	BP t <sub>e</sub>	0.9500 0.9405	5	Flash Point °C	
B 140°C	6.94709 1339.4	5	tc	0.256	5	Fire Point	
<u>c</u>	217.80	5	∆Hc kcal/m			M Spec. Ultra V.	
A* 27 to	1.39172	5	ΔHf ΔFf			X-Ray Dif.	
B* 130 ℃	1257.9	5	Viscosity	<del></del>	-	Infrared	
·			centistokes	1		Solubility in + Acetone	
t <sub>k</sub> to	1		η •c	Ì		Carbon tet.	
A'   10 to	7,34919	5				Benzene Ether	
B' 27 °C	1543.6	5	B <sup>V</sup>   to	<u> </u>		n-Heptane	
A'* 10 to	235.80	5	B to A C		l	Ethanol Water	
B'* 27 °C	1.74826 1444.1	5	(B <sup>V</sup> )  to	-[		Water in	
Ac   140 to	7.3576	5	(A <sup>V</sup> ) °C	1			
Bc tc_C	1650.	5	c <sub>p</sub> liq. °K	<del></del>		}	
Cryos. A*	257.	5	1				1
consts. B°		L	c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	122.89	5	c <sub>w</sub> vap.				
$T_{\mathbf{R}} = 0.75$						+ grams/100 gram	s solvent
	ES: 1-Dow			Calc, from de	t. da	ta 5-Calc, by form	ula
SOURCE:		A.	PI				
PURIFICAT			PI				
LITERATUE	RE REFERE	NCES	<b>5:</b>				

						Ne	o. 103	
NAME _	3, 3-Dimethyl-1-hexene					STRUCTURAL FORMULA		
						CH <sub>3</sub>		
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 112.2	208	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> ċ сн= ċн <sub>3</sub>	CH <sub>2</sub>	
		Ref.	1	T	Ref.		Ref	
F. P. °C			dt/dP			f to		
F.P. 100%		L	°C/mm 25°C	0.5340	_	g° <u>K</u>		
B.P. °C 760 mm	104.	2	BP	0.5340 0.0455	5 4	h		
100	46.36	4	t <sub>e</sub>	0.0363	5	f' to		
30 10	20.94 1.65	4 5	30 mm	0.6365	5	g' '° <u>K</u>		
ı	-30.18	5	∆Hm cal/g		ļ	h' to		
Pressure			ΔHv cal/g 25°C	70.70	5	m to		
mm 25°C	36.97 1016.	5	30 mm	79.70 80.24	5	•		
t <sub>e</sub> Density	1010.	-	BP	68.88	5	m' to		
g/ml 20°C		2	t <sub>e</sub> (d, e)	67.50 67.46	5	n' <u>*K</u>		
d <sub>4</sub> 25	0.7099	2	AHv/Te	19.54	5			
a 30	0,7303	4	d 21 to	83.11	5	Surface tension dynes/cm. 20°C 20.	68 5	
ь	-0.0381	4	_e	0.1368	5	8 30 19.		
Ref. Index			e' '°C			40 18.	81 5	
<sup>n</sup> D 20°C	1.4070 1.4046	2 2	d <sub>c</sub> g/ml	0.243	5	Parachor [P] 20°C		
30	1.4021	4	V_ mi/g	4.115 268.	5	30		
"C"	0.7593	4	t <sub>c</sub> °C	18786.	5	40 Sugd. 335.	2 5	
MR (Obs.)	38.68	2	PV/RT	10700.	<del>                                     </del>	Exp. L. l. %/wt.	2   3	
MR (Calc.) (nD-d/2)	38.677 1.0500	5 2	25°C	0.9991	5	u.		
Dielectric	+	-	30 mm BP	1.0000 0.9510	5	Dispersion 117.	2	
A 21 to	6.93989	5	t	0.9422	5	Flash Point C Fire Point		
B (132 °C	1312.1	5	t <sub>c</sub>	0.257	5	M. Spec.		
C A* 21 to	219, 24	5	ΔHc kcal/m ΔHf			Ultra V.		
B* 124 °C	1.39099 1231.5	5	ΔFf			X-Ray Dif. Infrared		
к ———	1		Viscosity			Solubility in +		
	.		centistokes り °C			Acetone		
t <sub>x</sub> °C			'		1	Carbon tet. Benzene		
A'  to B'  °C						Ether		
c' '	•		B <sup>V</sup> to C			n-Heptane Ethanol		
A¹* to				_	1	Water		
B'* °C			(B <sup>V</sup> )  to			Water in	-+	
Ac 132 to Bc t °C	7.3512	5	(A <sup>V</sup> )  °C	<del> </del>	<u> </u>			
Bc_tc_°C	1617. 258.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	114.36	5	c <sub>v</sub> vap.					
$T_R = 0.7$		L	L	1	<b>_</b>	grams/100 grams s	olvent	
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:		A	PI					
PURIFICAT	TION:	A	PI					
LITERATU	RE REFERE	NCE	S:					

NAME	2 / 5								4
r	NAME 3, 4-Dimethyl-1-hexene					- 1	STRUCTURAL FORMULA		
i							CH <sub>3</sub>		
							сизсизси си	сн=сн,	
Mole	Ref.		ecul		Molecular	200	с́н,	•	
% Pur.			muk	8 16	Weight 112.	_			Ref
	1		Ref.		<u> </u>	Ref			TKE!
F.P. C F.P. 1007		$\dashv$	$\dashv$	dt/dP *C/mm	Ì	П	f to		
B. P. *C	+			25°C	0.7214	5	g   <u>°K</u>		
76 <b>0 mm</b>	112.		2	BP	0.0463 0.0363	5	f! to		+
100 30	53.31 27.41		4	t <sub>e</sub> 30 mm	0.6486	5	g'    °K		
10	7.64	. 1	5	ΔHm cal/g	0.0100	╁	h'		
11	-25.47		5	ΔHv cal/g	<del> </del>	$\vdash$	m   to	····	
Pressure mm 25°C	26,48	. 1	5	25°C	82.42	5	n •K_		
t <sub>e</sub>	1038.	.	5	30 mm BP	82.25	5	<u> </u>		<u> </u>
Density	<b>†</b>		_	t <sub>e</sub>	70.50 68.98	5 5	m'   to		
g/ml 20°C			2	te (d, e)	68.93	5	n'  *K		1
d <sub>4</sub> 25	0.72 0.71		2	ΔHv/T <sub>e</sub>	19.52	5			—
	0,74	-	4	d   27 to		5	Surface tension dynes/cm, 20°C	21.87	5
Ъ	-0.03		4	d 123 to		5 5	30	20.91	5
Ref. Index	. 1	. 1	_	e' 27 °C		5	40	19.97	5
n <sub>D</sub> 20°C	1.41		2 2	d g/ml	0.246	5	Parachor [P] 20°C		1
30	1.40		4	d g/ml vc ml/g tc °C	4.068 281.	5 5	30		
"C"	0.75	92	4	-	19384.	5	40 Sugal	225 2	5
MR (Obs.)			2	P <sub>c</sub> mm PV/RT	17304.	H	Exp. L.1.%/wt.	335.2	+-
MR (Calc. (nD-d/2)	38.67		5 2	25°C	1.0000	5	u.		1
Dielectric		-		30 mm BP	1.00 <b>00</b> 0.9500	5 5	Dispersion	117.	2
A   27 to	<del></del>	826	5	te	0.9405	5	Flash Point °C		
B 142°C	1341.1	ŀ	5	tc	0.256	5	Fire Point		$\vdash$
<u> </u>	217.72	$\overline{}$	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1
A*  27 to B*  133 °C		246	5	ΔFf		1	X-Ray Dif.		1
K 133				Viscosity	† · · · · · · · · · · · · · · · · · · ·		Infrared Solubility in +		┼
t to	_	1		centistokes			Solubility in +		
t <sub>x</sub>   to			- 1	η •c			Carbon tet.		
A'   15 to	7.35	00	5				Benzene Ether		
B' <u>27 °C</u>	1545.3 236.		5	B <sup>V</sup>   to	<del> </del>	-	n-Heptane		1
	<del></del>	004	5	B' to	ł	1 1	Ethanol Water		
A'* 15 to B'* 27 °C		1074	5	(B <sup>V</sup> )  to	1	1 1	Water in		L
Ac   142 to	7, 35	92	5	(A <sup>V</sup> ) <sub>1</sub> °C	1				
Bc tc_	1653.		5	c <sub>p</sub> liq. °K	<del> </del>	$\vdash$			1
Cc	257.	-+	5	-					
Cryos. A° consts. B°				c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	123.34		5	c <sub>v</sub> vap.	ļ	) ]			
$T_{R} = 0.7$	5 <b>T</b> <sub>c</sub>						+ grams/100 gran	ns solven	t
REFEREN	CES: 1-D	ow 2	2-AP	I 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by form	nula	
SOURCE:			AF	<del></del>					
PURIFICA'			AF						
LITERATU	RE REFI	EREN	CES	:					

							No. 10	) <u>5</u>
NAME	3,5-Dimethyl-1-hexene					STRUCTURAL FORMULA		
						сн <sub>3</sub> сн сн <sub>2</sub> с	н сн=сн.	
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 112.2	08		H <sub>3</sub>	٤
		Ref.			Ref.			Ref
F, P. °C			dt/dP			f to		
F.P. 100%	ļ	<u> </u>	°C/mm 25°C	0.5340	ا ۔ ا	g   '° <u>K</u>	1	
B. P. °C 760 mm	104.	١,	BP	0.5340 0.0455	5 4	h		L
100	46.36	2	t <sub>e</sub>	0.0363	5	f' to		
30	20.94	4	30 mm	0.6365	5	g'   '° <u>K</u>		
1 <b>0</b> 1	1.65 -30.18	5	ΔHm cal/g			h' i		-
Pressure			ΔHv cal/g			m to		
mm 25°C	36.97	5	25°C 30 mm	79.70 80.24	5	<del>-</del>		
t <sub>e</sub>	1016.0	5	BP	68.88	-5	m¹ to	<del> </del>	+-
Density g/ml 20°C	0.709		te (d. a)	67.50	5	n'ı eK		
	0.708 0.704	2	te (d, e)	67.46	5	0'	1	
dt 25 4 30	0.700	4	ΔHv/T <sub>e</sub>	19.54	5	Surface tension		+-
<b>a</b>	0.724	4	d 21 to	83.11 0.1369	5	dynes/cm. 20°C	19.99	5
ь	-0.038	4	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	0.1369	"	<b>8</b> 30	19.09	5
Ref. Index	1.404	2	e'   °C		$oxed{oxed}$	40	18.22	5
<sup>n</sup> D 20°C	1.402	2	d <sub>c</sub> g/ml	0.241	5	Parachor [P] 20°C		1
30	1.399	4	vc ml/g tc °C	4.142 267.	5 5	30		1
"C"	0.7604	4	P <sub>c</sub> mm	18629.	5	40 Suga	3 <b>3</b> 5.2	5
MR (Obs.)	38.8	2	PV/RT	18027.	+-1	Exp. L.1.%/wt.	333.2	+-
MR (Calc.) (nD-d/2)	38.677 1.050	5 2	25°C	0.9991	5	u.		
Dielectric	1.050	-	30 mm BP	1.0000	5	Dispersion	117.	2
A 21 to	6, 93989	5	t	0.9510 0.9422	5 5	Flash Point C		
B   132 °C	1312.1	5	t c	0.257	5	Fire Point		
с	219.24	5	∆Hc kcal/m			M. Spec. Ultra V.		İ
A# 21 to	1.39099	5	ΔHf ΔFf	1		X-Ray Dif.		
B* 124 °C	1231.5	5	Viscosity		$\vdash$	Infrared		_
°	l		centistokes			Solubility in TACetone		1
tk to			η °C			Carbon tet.		
A'   to		-				Benzene	1	
B'	1				1	Ether n-Heptane		
<u>c'</u> — —			B <sup>V</sup> to A <sup>V</sup> °C	1		Ethanol		
A <sup>1</sup> * to B <sup>1</sup> * °C			III. TV	]		Water Water in		
	7 2510	-	(B <sup>V</sup> )  to					T
Ac 132 to Bc t <sub>c</sub> °C	7.3510 1617.	5	(A <sup>V</sup> )  °C		$\vdash$		ļ	
Cc	258.	5	c <sub>p</sub> liq. °K				1	
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	114.36	5	c <sub>v</sub> vap.					
$T_{\mathbf{R}} = 0.75$	Tc					grams/100 gra	ms solver	ıt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:		AI	PI					
PURIFICAT	ION:	AI	PI					
LITERATU	RE REFERE	NCES	S:					

							No. 106	
NAME 4,4-Dimethyl-1-hexene						STRUCTURAL FORMULA		
						CH <sub>3</sub>		
						сн <sub>3</sub> сн <sub>2</sub> с сн <sub>2</sub> с	CH=CH <sub>2</sub>	
Mole % Pur.		olecul ormul		Molecular Weight 112.2	08	CH <sub>3</sub>		
<u> </u>		Ref		Weight 112,2	Ref	<u> </u>	Re	
	T	Wei.		T	1.01			
F.P. °C F.P. 100%	<del> </del>	+-	dt/dP *C/mm	1		f to K		
B, P, *C	<del>}</del>	+	25°C	0.6016	5	g   <u>*K</u>		
760 mm	107.2	2	BP	0.0458 0.0363	5			
100	49.14	4	t <sub>e</sub>		1 1	f' to g'*K		
30 10	23.53 4.09	4 5	30 mm	0.6413	5	h'	1	
ì	-27.99	5	∆Hm cal/g			ļ		
Pressure	1		ΔHv cal/g	7. 10	_	m to		
mm 25°C	32.37	5	25°C 30 mm	71.19 81.04	5		-	
t <sub>e</sub>	1025.	5	BP	69.56	5	m'   to		
Density g/ml 20°C	0.7100	,	te te (d, e)	68.12 68.08	5	n' ' °K		
	0.7198 0.7157	2 2	(a, e)		1	0'		
d <sub>4</sub> 25 30	0.7116	4	ΔHv/T <sub>e</sub>	19.54	5	Surface tension		
8	0.7361	4	d   24 to		5	dynes/cm. 20°C	21.36 5	
Ъ	-0.0381	4	-a, - 118 °C			30	20.40 5	
Ref. Index			e' 'c			40	19.46 5	
n <sub>D</sub> 20°C		2	d <sub>c</sub> g/ml	0.245	5	Parachor [P] 20°C		
30	1.4078 1.4053	2 4	II A . 1111/16	4.089	5	30		
"C"	0.7588	4	1 -	273.	5	40		
MR (Obs.)	38,64	2	P <sub>c</sub> mm	19079.	5	Sugd.	335,2 5	
MR (Calc.)		5	PV/RT 25°C	0.0007	_	Exp. L.1.%/wt.		
(nD-d/2)	1.0503	2	30 mm	0.8807 1.0000	5	u. Dispersion	117. 2	
Dielectric			BP	0.9510	5	Flash Point °C		
A 24 to			ţe.	0.9419 0.257	5	Fire Point	1	
B [136 °C	1323.7	5	t <sub>c</sub>	0, 257	1 -	M Spec.		
	<del> </del>		ΔHc kcal/m ΔHf			Ultra V.		
A* 24 to B* 128 °C	1.39088	5	ΔFf			X-Ray Dif. Infrared		
K	.		Viscosity			Solubility in +		
tto	-		centistokes			Acetone	ļ	
t <sub>x</sub>   to			የ ℃	'		Carbon tet.		
A' to	<del> </del>	+				Benzene Ether		
B'	.}		<del></del>	ļ	<u> </u>	n-Heptane		
С'	1		B <sup>V</sup> to		1	Ethanol		
A'* to			A <sup>V</sup>   - °C	4		Water Water in		
B'* °C	<del> </del>	+	(B <sup>V</sup> ) to	i				
Ac 136 to	7,3545	5	(A <sup>V</sup> )  °C		<u> </u>		[	
Cc	258.	5	c <sub>p</sub> liq. °K					
Cryos. A°		1	c <sub>p</sub> vap. °K					
consts. B°			<u>-</u>					
t <sub>e</sub> °C	117.97	5	c <sub>w</sub> vap.	1				
$T_{\mathbf{R}} = 0.79$						<sup>+</sup> grams/100 gram	s solvent	
	ES: 1-Dow	2-AI		Calc, from de	t. da	ta 5-Calc. by form	ula	
SOURCE:	<del></del>	AF	PI					
PURIFICAT		AF						
LITERATU	RE REFERI	ENCES	5:					

No. 107 4,5-Dimethyl-1-hexene STRUCTURAL FORMULA NAME CH<sub>3</sub> CH3CH CH CH2CH=CH2 Mole Ref. Molecular Molecular Weight 112.208 Ċн<sub>3</sub> C8H16 % Pur. Formula Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm g °K 25°C 0.6431 5 B. P. °C h ΒP 0.0460 109. 50.69 760 mm 24 0.0363 5 ſ١ to 100 g' <u>°К</u> 24.96 30 4 30 mm 5 0.6441 10 5.44 5 h! AHm cal/g -26.78 5 m to AHv cal/g Pressure ۰ĸ n 25°C 79.03 mm 25°C 30.06 5 o 30 mm 81.47 5 te 1030. 5 BP 69.92 5 m' to Density 5 68.46 n' °K g/ml 20°C te (d, e) 0.728 2 68.41 5 o' d<sub>4</sub> 25 0.724 2 ΔHv/T<sub>e</sub> 19.54 5 30 0.720 4 Surface tension 25 to 84.91 5 0.744 4 dynes/cm. 20°C 22.35 1<u>20 °C</u> 0.1374 5 -0.038 4 30 21.37 ď 20.42 5 40 Ref. Index e' 20°C 1.414 [P] <sup>n</sup>D Parachor d<sub>c</sub> g/ml 0.248 5 25 1.412 2 20°C vc ml/g t °C 4.038 4 30 1.409 30 <sup>t</sup>c 277. 5 40 "C" 0.7568 4 P<sub>c</sub> mm 19460. 5 Sugd. 335.2 5 MR (Obs.) 38.5 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 38.677 5 25°C 0.9703 5 (nD-d/2) 1.050 2 30 mm 1.0000 Dispersion 2 5 117. Dielectric BP 5 0.9510 Flash Point C 0.9418 A 25 to 6.94402 Fire Point 0.257 B (\_140 °C 1329.8 5 M. Spec. Ultra V. 218, 29 5 AHc kcal/m ΔHf A≠ 25 to 5 1.38967 X-Ray Dif. ΔFf B\*|\_130 °C 1248.4 Infrared ĸ Viscosity Solubility in centistokes Acetone to  $\mathbf{t_k}$ Carbon tet. °C Benzene A١ to Ether В' °C n-Heptane B<sub>v</sub> | C Ethanol °C Water A'\* to Water in B'\* °C (B<sup>V</sup>)| to Ac | 140 to 7.3554 5 (AV) °C Bc tc °C 1641. 5 c liq. ۰ĸ 258. 5 Cryos. A c<sub>p</sub> vap. °K consts. B° c vap. te °C 5 120,00  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

NAME	5,5-1	Dimethyl	-l-hexene			STRUCTURAL F	ORMUL	A.
Mole % Pur.	Ref.	Molecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112,2	208	сн <sub>3</sub> с (сн <sub>2</sub> ) <sub>2</sub> с	H=CH <sub>2</sub>	
,0 1 u1.		Ref.		Welght 112.	Ref	<u> </u>		Ref
	г	Kei.		<del></del>	Kei	····		IVE:
F.P. °C F.P. 100%	<u> </u>		dt/dP	1		f to		1
			*C/mm 25*C	0.5054	5	g <u>*K</u>		1
B. P. *C 760 mm	102.5	2	BP	0.0453	4	h		┞-
100	45.07	4	t <sub>e</sub>	0.0362	5	f' to		ł
30	19.74	4	30 mm	0.6341	5	g'		1
10 1	0.52 -31.19	5	AHm cal/g			h'		<u></u>
	-31,19	3	ΔHv cal/g	†		m   to		1
Pressure mm 25°C	39.31	5	25°C	79.19	5	n   •K_		l
t <sub>e</sub>	1013.	5	30 mm	79.89	5	°		L
Density	· · · · ·	<del>-   ´</del> -	BP t	68.65 67.30	5	m'   to		
g/ml 20°C	0.70	9 2	te (d, e)	67.26	5	n'*K_		1
dt 25 4 30	0.70	5 2	ΔHv/T	19.57	5	01		L
	0.70		d   20 to		5	Surface tension		
<b>a</b> b	0.72		_ i _113 °C		5	dynes/cm. 20°C	20.10	5
	-0.03	8 4	d' to	51		30 40	19.21 18.33	5
Ref. Index	1.40	49 2	e' ' °(		$\perp \perp$	Parachor [P]	10.33	1-
<sup>n</sup> D 25 25	1.40		d <sub>c</sub> g/ml	0, 244	5	20°C		
30	1.40	01 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	4.101 265.	5 5	30		į.
"C"	0.76	09 4	1 -	1	5	40	225 2	5
MR (Obs.)	38.8	2	P <sub>c</sub> mm	18670.	1.1	Sugd.	335. 4	<b>↓</b> ³
MR (Calc.)	38.67		PV/RT 25°C	0.9991	5	Exp. L.1.%/wt. u.		Ì
(nD-d/2)	1.05	01 2	30 mm	1.0000	5		117.	2
Dielectric		]	BP	0.9520	5	Flash Point °C		+
A 20 to	6.93		te	0.9434 0.256	5 5	Fire Point		1
B 1130 ℃		5	tc	0.230	1-1	M Spec.		T
C	219.52	5	ΔHc kcal/m			Ultra V.		ı
A* 20 to B* 120 °C	1.390 1226.2	022 5	ΔFf			X-Ray Dif.		
B* ∟120 °C	1220.2	٦	Viscosity		$\vdash$	Infrared		ـ
c			centistokes			Solubility in +		
t <sub>k</sub> to			η •α	:	1 1	Acetone Carbon tet.		1
* <u>*</u> •C				1		Benzene		
A' to B' C			1		1	Ether		
5, L – E		1	B <sup>V</sup>   to		$\sqcap$	n-Heptane Ethanol		
A¹* to			AV   °C			Water		
B'* *C			(BV) to	7		Water in		1_
Ac   130 to	7.350	05 5	(A <sup>V</sup> )					1
Bc t °C	1610.6	5	c <sub>p</sub> liq. °K	<del></del>	$\vdash$			
Cc	258.	5	P4. *K	1	1 1			
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	112,71	5	c <sub>v</sub> vap.	<u> </u>				L
$T_R = 0.75$						grams/100 gram		t
REFERENC	೬S: 1-Do			Calc. from de	t. dat	ta 5-Calc, by forn	nula	
SOURCE:		A1						
PURIFICAT		Al						
LITERATUR	E REFE	RENCES	3:					

No. 109 3-Ethyl-cis-2-hexene STRUCTURAL FORMULA NAME CH3CH=C (CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub> Molecular C8H16 C<sub>2</sub>H<sub>5</sub> Mole Ref. Molecular Weight 112.208 % Pur. Formula Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm 25°C ٩K g 1.020 5 B. P. °C h BP 0.0472 4 760 mm 121. 24 t<sub>e</sub> 0.0363 5 ſ١ to 61.12 100 g' ٩K 34.68 4 30 5 30 mm 0.6622 14.50 5 10 h' ∆Hm cal/g -19.31 5 to m AHv cal/g Pressure n ۰K 25°C 85.30 5 mm 25°C 18.09 30 mm 84.51 5 t<sub>e</sub> 1063. 5 ВP 72,42 5 m' to Density 70.73 5 te te (d, e) n' °K g/ml 20°C 0.737 2 70.67 o' 25 30  $\mathbf{d_{4}^{t}}$ 0.733 2 AHv/Te 5 19.52 0.729 4 Surface tension 35 to 89.36 5 0.753 -0.0<sub>3</sub>8 23.49 a 44 dynes/cm. 20°C 5 1<u>34 °C</u> 0.1400 5 ь 30 22.47 5 25 to d' 5 87.35 21.49 5 40 Ref. Index e' 35 °C 0.0819 5 <sup>n</sup>D 20°C 1.424 Parachor [P] 2 d<sub>c</sub> g/ml 0.248 5 25 1.422 20°C vc ml/g t °C 4.030 5 30 1.419 4 30 t<sub>c</sub> 296. 5 40 "C" 0.7646 4 P<sub>c</sub> mm 20096. 5 Sugd. 335.2 5 38.9 MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 25°C 1.0000 5 (nD-d/2) 1.056 30 mm 2 1.0000 5 Dispersion 124. Dielectric ВP 0.9500 5 Flash Point C 0.9397 5 A 35 to ţe. 6.95770 5 Fire Point 0,256 5 1374.0 B 1154°C M. Spec. 216. 5 AHc kcal/m Ultra V ΔHf A\* 35 to 1.39234 X-Ray Dif. ΔFf B\*[144°C 1290.7 Infrared Viscosity Solubility in centistokes Acetone to Carbon tet. °C **t**x\_\_ Benzene A' 25 to 7.35121 Ether В' 35 °C 1578.3 n-Heptane B<sub>V</sub> | C' 234. to Ethanol °C Water A'\* 25 to 1.74671 5 Water in B'\* 35 °C 1478.6 (BV) to Acl 154 to (A<sup>V</sup>)| 7.3684 5 °C Bc tc °C 1694. c<sub>p</sub> liq. °K Cc<sup>1</sup> 257. 5 Cryos. A° consts. B° ۰ĸ c<sub>p</sub> vap. c<sub>v</sub> vap. te °C 133.50  $T_{\mathbf{R}} = 0.75 \, T_{\mathbf{c}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							<b>N</b> o. 11	0
NAME	3-Ethvl-t	rans-	-2-hexene		T	STRUCTURAL		
````								
						С <sub>2</sub> н	5	
34-1-	200			Malala	- 1	$CH_3(CH_2)_2C =$	CH-CH <sub>3</sub>	
Mole % Pur.	Ref. Mo	rmul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	80:			
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		Ref.			Ref			Ref.
		Kei.		T****	I.c.			1
F.P. °C F.P. 100%	<u> </u>	+	dt/dP *C/mm	i		f to		1
B. P. *C	-	1	25°C	1.0203	5	g <u>*K</u> _		
760 mm	121.	2	BP	0.0472	4	h		ـــــ
100	61.12	4	t <sub>e</sub>	0.0363	5	f' to		1
30	34.68	4	30 mm	0.6622	5	g'  K_	ł	1
10 1	14.50 -19.31	5	AHm cal/g			h¹		<b>⊥</b>
	-17.31	<del>                                     </del>	ΔHv cal/g	<u> </u>		m   to		1
Pressure mm 25°C	18.09	5	25°C	85.30	5	n •K_		ł
t <sub>e</sub>	1063.	5	30 mm	84.51	5	L		Ь.
Density		1	BP t	72.42 70.73	5 5	m'   to		1
g/ml 20°C		2	te (d, e)	70.67	5	n'   <u>*K</u> _		1
dt 25 4 30	0.733	2	ΔHv/T	19.52	5	0.		
	0.729	4	d   35 to	<del></del>	5	Surface tension		
a b	0.753	4	e   134 °C		5	dynes/cm. 20°C	23.49	5
	-0.038	4	d'   25 to	87.35	5	30 40	22.47 21.49	5
Ref. Index		2	e' j 35 °C	0.0819	5	Parachor [P]		Ť
<sup>n</sup> D 25	1.422	2	d g/ml	0.248	5	20°C		1
30	1.419	4	V_ 1111/K	4.030 296.	5	30		
"C"	0.7646	4	·	20096.	5	40	225	١,
MR (Obs.)	38.9	2	P <sub>c</sub> mm	20096.	3	Sugd.	335.	2
MR (Calc.		5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt. u.		l
(nD-d/2)	1.056	2	30 mm	1.0000	5	Dispersion	124.	2
Dielectric	İ	1	BP	0.9500	5	Flash Point °C		+
A 35 to		5	t <sub>e</sub>	0.9397 0.256	5	Fire Point	ĺ	
B [_154°C		5	t <sub>c</sub>	0.256	1 3	M Spec.		+
C	216.		ΔHc kcal/m ΔHf			Ultra V.		
A* 35 to B* 144 °C		5	ΔFf			X-Ray Dif.		
K L.== 5	- 1270.7	'	Viscosity	†		Infrared		┼
c		1	centistokes			Solubility in + Acetone	}	
tk to			∥າ °c	j		Carbon tet.		
<u> </u>	.1	L_				Benzene		
A'   25 to B'   35 °C		5				Ether		
c, 🗀 🛎 🖹	234.	5	B <sup>V</sup> to			n-Heptane Ethanol	}	
A'* 25 to	1.74671	5	A <sup>V</sup>   °C			Water		1
B'* 35 °C		5	(B <sup>V</sup> ) to	-		Water in	<u> </u>	4
Ac   154 to	7.3684	5	(A <sup>V</sup> ) °C	1		ł		1
Bc t °C	1694.	5	c <sub>p</sub> liq. °K	<del> </del>	$\vdash$		[	1
Ce	257.	5	ll <sup>-</sup>				ĺ	
Cryos. A°			c <sub>p</sub> vap. °K	İ			ļ	
consts. B°		-	c, vap.	1				
t <sub>e</sub> °C	133.50	5	- V P.	<u> </u>	<u></u>	L		
$T_R = 0.7$						grams/100 grai		1t
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICAT			PI 					
LITERATU	RE REFERE	NCES	<b>5:</b>					

							No. 11	.1
NAME	4-Ethyl-	cis-2	-hexene			STRUCTURAL	FORMUL	A
						C <sub>2</sub> H <sub>5</sub>		
Mole % Pur.	Ref. Mo	ecul		Molecular Weight 112,20	08	CH <sub>3</sub> CH <sub>2</sub> CH C	H=CH CH <sub>3</sub>	
		Ref.		ļ	Ref.		T	Ref.
F.P. °C F.P. 100%			dt/dP °C/mm 25°C	0.7500	_	f to		
B. P. °C 760 mm 100 30	113. 54.18 28.22	2 4 4	BP t <sub>e</sub> 30 mm	0.7500 0.0464 0.0363 0.6501	5 4 5 5	h   to g'  °K		
10 1	8.40 -24.78	5	ΔHm cal/g			h'		
Pressure mm 25°C t <sub>e</sub>	25.38 1040.	5 5	AHv cal/g 25°C 30 mm BP	82.95 82.50 70.72	5 5 5	m to		
Density g/ml 20°C dt 25 d4 30	0.725 0.721 0.717	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	69. 18 69. 13 19. 52	5 5	m' to		
a b	0.741 -0.0 <sub>3</sub> 8	4 4	d 20 to e 124 °C d' to	86.42 0.1390	5	Surface tension dynes/cm. 20°C 30 40	21.99 21.02 20.08	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30		2 2 4	e'   °C d g/ml vc ml/g t °C	0.246 4.061 282.	5 5 5	Parachor [P] 20°C 30	20.08	
"C"	0.7564	4	t <sub>c</sub> °C P <sub>c</sub> mm	19451.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc. (nD-d/2) Dielectric		2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9500	5 5 5	Exp. L.1.%/wt. u. Dispersion	119.	2
A 20 to B 1144 °C		5	te tc	0.9404 0.256	5	Flash Point <sup>6</sup> C Fire Point M. Spec.		
A* 20 to B* 134 °C K	1.39279 1263.1	5 5 5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared		
c t <sub>k</sub> — tō t <sub>x</sub>   °C			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet. Benzene		
A'   15 to B'   20 °C C' A'* 15 to	7. 3505 1549. 1 235. 53 1. 74901	5 5 5	B <sup>v</sup> to A <sup>v</sup> °C			Ether n-Heptane Ethanol Water		
B'* 20 °C	1450.	5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C			Water in		-
Act 144 to Bc t <sub>c</sub> °C Cc Cryos. A°	1658. - 257.	5	c <sub>p</sub> liq. °K					
consts. B°			р.					
t <sub>e</sub> °C T <sub>R</sub> = 0.7	124.47 5 T	5	c <sub>v</sub> vap.			+ ~ ma m = /100		<u></u>
	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da	grams/100 gra nta 5-Calc. by for		it.
SOURCE:			PI	de	J. 40	5 02.0. 5, 10.		
PURIFICA'	TION:	A	.PI					
LITERATU	RE REFERE	NCE	5:					

							<b>No.</b> 11	,
NAME	4-Ethyl-	trans	-2-hexene			STRUCTURAL		
						сн <sub>з</sub> сн <sub>з</sub> сн сн	=СНСН.	
Mole % Pur.	Ref. Me	olecul ormul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.	208	с <sub>2</sub> н <sub>5</sub>	3	
,, <u>, , , , , , , , , , , , , , , , , , </u>		Ref.			Ref			Ref.
F.P. *C			dt/dP	1		f   to		
F.P. 100%			°C/mm	0.750		gK_		1
B. P. *C	112		25°C BP	0.750 0.0464	5	h		
760 mm 100	113. 54.18	2 4	te	0.0363	5	f' to		
30	28.22	4	30 mm	0.6501	5	g'	-	ĺ
10 1	8.40 -24.78	5	∆Hm cal/g			h'		<u> </u>
Pressure		+	ΔHv cal/g			m to		}
mm 25°C	25.38	5	25°C 30 mm	82.95 82.50	5	" <u></u> -	1	
t <sub>e</sub>	1040.	5	BP	70.72	5	m'   to	<b></b>	+
Density g/ml 20°C	0.725	2	te te (d, e)	69.18 69.13	5	n'°K	j	
dt 25	0.721	2	ΔHv/T	19.52	5	o'	}	
4 30	0.717	4	· · · · · · · · · · · · · · · · · · ·	<del></del>	-	Surface tension		
a b	0.741	4	d 28 to		5	dynes/cm. 20°C	21.99	5
Ref. Index	-0.038	+	d'   to	5		30 40	21.02	5
n <sub>D</sub> 20°C		2	e' j •C	+	<del>  _</del>	Parachor [P]		+
25 30	1.410	2	d g/ml v ml/g	0.246 4.061	5	20°C	l	İ
"C"	1.407	4	tc °C	282.	5	30 40		
	0.7564	4	P <sub>c</sub> mm	19451.	5	10	335.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.050	2	25°C 30 mm	1.0000	5	u. Dispersion	121.	2
Dielectric			BP	0.9500	5	Flash Point °C	121.	۴
A 20 to	6.94966	5	te	0.9404 0.256	5	Fire Point	1	1
B 144 °C	217.53	5	t <sub>c</sub>	0.250	<u> </u>	M Spec.		1
A*   20 to	+	5	ΔHf		i	Ultra V. X-Ray Dif.	1	1
B* 134 °C		5	ΔFf	ļ	<u> </u>	Infrared		1
K — — —	1		Viscosity centistokes		ł	Solubility in +		$\vdash$
t <sub>k</sub> to			7	:		Acetone Carbon tet.		1
'x '			,			Benzene		i
A'   15 to B'   20 °C		5				Ether		l
c, - = =	235.53	5	B <sup>V</sup>   to			n-Heptane Ethanol		
A'* 15 to		5	AV - °C		l	Water	1	1
B'* 20 °C	<del></del>	5	(B <sup>V</sup> ) to			Water in		+-
Ac   144 to	7.3606 1658.	5	(A <sup>V</sup> )  °C		L			1
Cc - c -	257.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	124.47	5	c <sub>w</sub> wap.					
$T_{\mathbf{R}} = 0.7$						grams/100 grai		t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP	I					
PURIFICAT		AP						
LITERATU	RE REFERE	NCES	5:					

No. 113 2, 3-Dimethyl-2-hexene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_2C = C CH_3$ Molecular C8H16 ĊH<sub>3</sub>ĊH<sub>3</sub> Mole Ref. Molecular Weight 112, 208 % Pur Ref. Ref. F.P. °C F.P. 100% -115.1 2 dt/dP f to °C/mm g °Κ 25°C 1.0500 5 B. P. °C h RP 0.0473 4 760 mm 121.77 2 t<sub>e</sub> 0.0363 5 ſ١ 100 61.77 4 to 4 g' <u>• к</u> 30 35.28 30 mm 5 0.6635 15.05 5 10 h' ∆Hm cal/g 1 -18.81 to ΔHv cal/g Pressure n °K 25°C 85.51 mm 25°C 17.54 0 5 30 mm 84.67 t<sub>e</sub> 1066. 5 ВP 72.55 5 m to Density 70.85 5 te (d, e) °K g/m1 20°C 0.7408 0.7366 2 5 70.78 o' 25 ž  $\mathbf{d_{4}^{t}}$ AHv/Te 5 19.51 30 0.7324 4 Surface tension  $\top$ 35 to 89, 61 5 0.7576 4 dynes/cm. 20°C 23.97 °C 0.1401 5 134 ь -0.0384 4 5 30 22.90 aח 25 to 87.57 5 Ref. Index 40 21.85 0.0824 35 5 20°C 1.4268 nD [P] Parachor d<sub>c</sub> g/ml 0.248 5 25 1.4244 2 20°C vc ml/g tc °C 4.040 5 30 1.4217 4 30 297. 5 40 "C" 0.7654 4 P<sub>c</sub> mm 20079. 5 Sugd. 335.2 5 MR (Obs.) 38.87 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1.0000 5 (nD-d/2) 1.0564 2 30 mm 1,0000 2 Dispersion 127. Dielectric BP 0.9500 5 Flash Point C A 35 to 0.9397 6.95670 5 Fire Point 0.256 B (154 °C 1376.2 5 M. Spec. 215,86 5 AHc kcal/m Ultra V. A\* 35 to ΔHf 1.39053 5 X-Ray Dif. ΔFf B\*[144\_°C 1292.8 Infrared Viscosity Solubility in centistokes Acetone t<sub>k</sub> to Carbon tet. °C Benzene A' | 25 to 7.34943 Ether B١ 35 °C 1580.5 5 n-Heptane B<sup>V</sup> | 5 233.86 to Ethanol °C Water A'\* 25 to 1.74467 5 Water in B'\* 35 °C 5 (B<sup>V</sup>) 1480.7 to Ac 154 to 7.36734 5 (A<sup>V</sup>)| °C  $\underline{\underline{B}}c$  $\underline{\underline{t}}c$ •c 1697. c<sub>p</sub> liq. °K Cc 5 257 Cryos. Aº ۰ĸ c<sub>p</sub> vap. consts. B° c<sub>v</sub> vap. te °C 134.38 = 0.75 T<sub>c</sub> grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

·							No. 114	1
NAME	2, 4-Dim	ethy	l-2-hexene			STRUCTURAL	FORMULA	١.
						כא כט כט כי	1-C CH	
.,,						сн <sub>3</sub> сн <sub>2</sub> сн сі сн <sub>3</sub>	сн <sub>3</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	208	3	3	
		Ref.			Ref.			Ref.
F.P. °C			dt/dP			f to		
F.P. 1007	6		°C/mm 25°C	0.6835	5	g <u>  °K</u>		ļ
B. P. *C 760 mm	110.6	2	BP	0.0462	4	h		<u> </u>
100 30	52.08	4	t <sub>e</sub>	0.0363	5	f' to		İ
10	26, 26 6, 55	5	30 mm	0.6466	5	h'		
1	-26.45	5	ΔHm cal/g ΔHv cal/g	<del> </del>	-	m to		
Pressure mm 25°C	28.11	5	25°C	81.95	5	n		
te	1034.	5	30 mm BP	81.87 70.18	5	<u> </u>		<u> </u>
Density	0.7312	١,	t_	68.69	5	m'   to		
g/ml 20°0 dt 25 d4 30	0.7171	2	te (d, e) ΔHv/Te	68.64	5	o'		
	0,7129	4	d   20 to	19.51 85.51	5	Surface tension		
a b	0.7381 -0.0 <sub>3</sub> 83	4	<u>e   122</u> °C	0.1386	5	dynes/cm. 20°C	21.54 20.54	5
Ref. Index			d'   to			40	19.57	5
n <sub>D</sub> 20°0		2	<del></del>	0.244	5	Parachor [P] 20°C		
30	1.4094 1.4067	2 4	d g/ml vc ml/g tc °C	4.106 278.	5	30		
"C"	0.7600	4	tc °C P <sub>c</sub> mm	19100.	5	40 Sugd	335.2	5
MR (Obs.		2	PV/RT	17100.	1	Exp. L.1.%/wt.	333.2	Ť
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.		
Dielectric		m	30 mm BP	1.0000 0.9500	5	Dispersion Flash Point °C	124.	2
A 20 t		5	te t	0.9406 0.256	5	Fire Point		
B (_140_*)	2   1335.7   218.	5	t <sub>c</sub>		۰	M Spec.		
A* 20 to	1.39139	5	ΔHf ΔFf		l	Ultra V. X-Ray Dif.		
B* 130 °	<u>C</u> 1254. 3	5	Viscosity		<del> </del>	Infrared		<u> </u>
c	_	1	centistokes		ŀ	Solubility in + Acetone		1
			η •c	l		Carbon tet.		
A' t	<del>,</del>	<del>                                     </del>				Benzene Ether		
B', L _ º	<u> </u>	i	B <sup>V</sup>   to	<del> </del>	<del> </del>	n-Heptane Ethanol		
A'* t	,	<del> </del>	A <sup>v</sup> i °C		1	Water		
B1* °		L	(B <sup>V</sup> ) to			Water in		-
Ac   140 to		5 5	(A <sup>V</sup> ) °C					
Bc  _tc_°	257.	5	c <sub>p</sub> liq. °K					
Cryos, A			c <sub>p</sub> vap. °K					
te °C	121.76	5	c, vap.					
$T_R = 0.7$		1 3	ш	1	L	f grams/100 gran	ne solve-	<u> </u>
	CES: 1-Dow	2 - AI	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-Calc. by for		
SOURCE:			PI		· - <u>-</u>			
PURIFICA	TION:	A	PI					
LITERATU	RE REFERE	NCES	<b>3</b> :					

[ T	2 5-Dim	ethyl-	-2-hexene			CORDICONIO A I	No. 11	
NAME		Cury	- nexene		$\dashv$	STRUCTURAL	FORMUL	,A
			· · · · · · · · · · · · · · · · · · ·		$\dashv$	сн <sub>3</sub> сн сн <sub>2</sub> с		3
Mole	Ref. Mo	lecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular		Ċн <sub>3</sub>	Ċн <sub>3</sub>	
% Pur.	FC	Ref		Veight 112.2	Ref.	T .		Ref.
F.P. °C		Ken	dt/dP		Ker.			1
F.P. 1009	•		°C/mm		1	f to		İ
B. P. °C			25°C BP	0.7265 0.0464	5 4	h		
760 mm 100	112.2 53.47	2	te	0.0363	5	f' to		$\prod$
30	27.55	4	30 mm	0.6490	5	g' '° <u>K</u>		
10 1	7.77 -25.36	5	△Hm cal/g		<u> </u>	h'		-
Pressure			ΔHv cal/g 25°C	02 (2	_	m to		
mm 25°C	26.28 1038.	5	30 mm	82.63 82.27	5			İ
Density	1050.	+-	BP t	70.52 69.00	5	m¹ to		
g/m1 20°C	0.720	2	te te (d, e)	68.94	5	n' _ <u>°K</u>		
d <sub>4</sub> 25	0.716 0.712	2 4	ΔHv/T <sub>e</sub>	19.51	5			$\vdash$
a	0.736	4	d 20 to e 124 °C	86.10 0.1388	<b>5</b>	Surface tension dynes/cm. 20°C	21.39	5
ь	-0.038	4	d' To	0.1366		8 30 40	20.44 19.52	5
Ref. Index		2	e'		<u> </u>	Parachor [P]	19.52	<del>                                     </del>
25	1.4115	2	d <sub>c</sub> g/ml	0.244 4.094	5	20°C		
30	1.4091	4	vc m1/g tc °C	280.	5	30 40		
MR (Obs.)	0.7652	2	P <sub>c</sub> mm	19225.	5		335.2	5
MR (Calc.	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.054	2	30 mm	1.0000	5	Dispersion	124.	2
Dielectric A 20 to		-	BP t <sub>e</sub>	0.9500 0.9405	5	Flash Point C		
B 1142 °C		5	t <sub>c</sub>	0.256	5	Fire Point M. Spec.		-
C	217.68	5	ΔHc kcal/m ΔHf			Ultra V.		
A* 20 to B* 134 °C		5	ΔFf		ļ	X-Ray Dif. Infrared		1
K — — —	1		Viscosity centistokes		1	Solubility in +		T
t <sub>k</sub> to			η •c			Acetone Carbon tet.		
t <sub>x</sub> °C		ļ			l	Benzene		
B' ℃					<b> </b>	Ether n-Heptane		
C'	<del></del>	4—	B <sup>V</sup> to C			Ethanol Water		ĺ
A'* to B'* °C			$\frac{\mathbf{B}\mathbf{v}_{\parallel}}{(\mathbf{B}\mathbf{v}_{\parallel})} - \frac{\mathbf{v}_{\parallel}}{\mathbf{t}_{0}}$			Water in		
Ac  142 to	7.3581	5	(A <sup>V</sup> )  °C					
Bc tc °C	1653. - 257.	5	c <sub>p</sub> liq. °K					
Cryos. A°		<del>  _</del>	c <sub>p</sub> vap. °K					
consts. B		1-	c vap.					
t <sub>e</sub> °C T <sub>R</sub> = 0.7	123, 57	5		l		grams/100 gra	ma aclu	<u> </u>
	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ata 5-Calc. by for		16
SOURCE:			PI					
PURIFICA	TION:	А	.PI					
LITERATU	RE REFERE	NCE	S:					

							No. 11	6
NAME	3,4-1	Dimethy	-cis-2-hexene			STRUCTURAL		
Mole	Ref.	Molecu	lar C8H16	Molecular Weight 112.	208	сн <sub>3</sub> сн <sub>2</sub> сн с сн <sub>3</sub> сн		
A Pui.	ليبلب	Res		weight 112.	Ref			Ref.
F.P. °C	Ι	I.C.		T	IV61	l	1	1
F.P. 100%			dt/dP *C/mm			f to		
B. P. °C			25°C	0.8401	5	h .		1
760 mm 100	116.	2	BP t <sub>e</sub>	0.0467	5	f' to		<del>                                     </del>
30	56.76 30.61	4 4	30 mm	0.6548	5	g'   '°K_		
10 1	10.66	5	ΔHm cal/g			h'		
Pressure	-22.77	5	ΔHv cal/g	<u> </u>	<del>                                     </del>	m to		
mm 25°C	22,40	5	25°C	83.66	5	n <u>*K</u> _	ł	l
t <sub>e</sub>	1049.	5	30 mm BP	83.22 71.33	5			ـــ
Density			] t	69.74	5	m' to		1
g/ml 20°C	0.73		te (a, e)	69.68	5	" <del>-</del>	1	
dt 25 4 30	0.72		AHv/T <sub>e</sub>	19.51	5	Surface tension	<del></del>	╁
	0.75		d   31 to		<b>5</b>	dynes/cm. 20°C	23.48	5
ь	-0.03	8 4	d'   20 to	85.60	5	30 40	22.47	5
Ref. Index n <sub>D</sub> 20°C	1.41	8 2	e'   31 °C	<del></del>	5	Parachor [P]		+-
25	1.41	6 2	d g/ml vc ml/g	0.249 4.010	5 5	20°C		
30	1.41		tc °C	289.	5	30 40		1
"C"	0.75	$-\!\!\!\!-\!\!\!\!\!-$	P <sub>c</sub> mm	19946.	5		335.2	5
MR (Obs.) MR (Calc.)	38.4 38.67	7 5	PV/RT	<del></del>		Exp. L.1.%/wt.		
(nD-d/2)	1.05		25°C 30 mm	1.0000 1.0000	5 5	u. Dispersion	1,24	2
Dielectric			BP	0.9500	5	Flash Point °C	124.	+-
A 31 to	6.95		t <sub>e</sub>	0.9401 0.256	5 <b>5</b>	Fire Point		
B <u>  148 °C</u> C	1355.2 216.96	5	t <sub>c</sub>	0.250	Ť	M Spec.		
A*   31 to	1.39		ΔHf			Ultra V. X-Ray Dif.		
B* 138 °C	1272.8	5	ΔFf	<b>_</b>	ļ	Infrared	j	
K		ĺ	Viscosity centistokes			Solubility in +		
t <sub>k</sub>   to	1		7 .c			Acetone Carbon tet.		1
tÇ °C			<b>∄</b> `		. '	Benzene		1
A'   20 to   B'   31 ℃	7.34 1559.4	881   5				Ether n-Heptane		i
c' -	234.96	5	B <sup>v</sup> l to			Ethanol		1
A'* 20 to	1.74		A <sup>V</sup>   - °C	_		Water Water in		1
B'* 31 °C		5	(B <sup>V</sup> ) to	1				+
Ac   148 to Bc   t C	7.36 1672.	20   5	(A <sup>V</sup> )  °C	<del></del>				
Cc'	257.	5	c <sub>p</sub> liq. °K				1	
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	127.86	5	c <sub>v</sub> vap.					
$T_{R} = 0.75$	T <sub>c</sub>					grams/100 gram	ms solven	t
REFERENC		ow 2-A	PI 3-Lit. 4-	Calc, from det	t. da	ta 5-Calc. by for		
SOURCE:		A	PI					
PURIFICAT			PI					
LITERATUR	E REFE	ERENCE	S:					
		_						

							No. 117
NAME	3, 4-Dim	ethyl	-trans-2-hexene			STRUCTURAL	FORMULA
						CII CII CII C	- CH CH
	T. / .,				$\neg$	CH <sub>3</sub> CH <sub>2</sub> CH C	
Mole % Pur.	Ref. Me	rmul		Molecular Weight 112.2	08	ċн <sub>3</sub> ċн	<sup>1</sup> 3
		Ref			Ref.	<u> </u>	Ref
F.P. ℃			dt/dP			f to	
F.P. 100%		1	°C/mm 25°C	0.8401	5	g   <u>°K</u>	<b>↓</b>
B. P. °C 760 mm	116.	2	BP	0.0467	4	h	L
100	56.76	4	t <sub>e</sub>	0.0363	5	f' to	
30 10	30.61 10.66	5	30 mm	0,6548	5	g' ' <u>°K</u> h'	
1	-22.77	5	ΔHm cal/g	<u> </u>	<u> </u>	m   to	
Pressure			ΔHv cal/g 25°C	83.66	5	n	1
mm 25°C	22.40 1049.	5	30 mm	83.22	5	0	
Density		+-	BP t <sub>e</sub>	71.33 69.74	5	m' to	
g/m1 20°C		2	'e (u, e)	69.68	5	n'   <u>*K</u>	1 1
d <sub>4</sub> 30	0.733 0.729	2 4	ΔHv/T <sub>e</sub>	19.51	5		<del> </del>
a .	0.753	4	d 31 to e 128 °C	87.48	5 5	Surface tension dynes/cm. 20°C	23.48 5
ь	-0.038	4	d' 20 to	0.1393 85.60	5	ک 30 ا	22.47 5
Ref. Index		2	e' 31 °C	0.0779	5	Parachor [P]	21.48 5
25	1.416	2	d <sub>c</sub> g/ml	0.249 4.010	5	20°C	
"C"	1.413	4	vc ml/g tc °C	289.	5	30 40	
	0.7543	4	P <sub>c</sub> mm	19946.	5		<b>335.2</b> 5
MR (Obs.) MR (Calc.		2 5	PV/RT 25°C	1 0000	_	Exp. L.1.%/wt.	
(nD-d/2)	1.050	2	30 mm	1.0000	5	u. Dispersion	124. 2
Dielectric		1_	BP	0.9500	5	Flash Point C	
A 31 to B 148 °C	6.95084 1355.2	5	te t <sub>C</sub>	0.9401 0.256	5	Fire Point	
c	216.96	5	∆Hc kcal/m			M. Spec. Ultra V.	1
A* 31 to	1.39076	5	ΔHf ΔFf			X-Ray Dif.	1
B*[138 °C	- 1272.8	) J.	Viscosity			Infrared	-
t, to	-		centistokes	i		Solubility in Acetone	
			η °C			Carbon tet. Benzene	]
A'   20 to	7.34881					Ether	
B' 1_31_°C	1559.4 234.96	5	B <sup>V</sup> to			n-Heptane Ethanol	1 1
A1* 20 to	1.74623	+	A'   °C			Water	
B'* 31 °C	1459.8	5	(B <sup>v</sup> )  to			Water in	
Ac 148 to	7.3620 1672.	5	(A <sup>V</sup> )  °C				
Bc t <sub>c</sub> °C	257.	5	c <sub>p</sub> liq. °K				
Cryos. A°			c <sub>p</sub> vap. *K	}			
consts. B°	+	+	c <sub>v</sub> vap.				
t <sub>e</sub> °C	127.86	5	-v -z.	l	L	L + /200	<u> </u>
T <sub>R</sub> = 0.7		2 - A	PI 3_1.i+ 4 /	Calc from de	+ 4-	grams/100 grants 5-Calc. by for	
SOURCE:	75-3: 1-DOW		PI 3-Lit. 4-	Care, from de	, ua	ica J-Calc. Dy 101	
PURIFICA?	TION:		.PI				
	RE REFERI					· · · · · · · · · · · · · · · · · · ·	

					—-т		No. 11	
NAME _	3,5-	Dimethy	l-cis-2-hexene		_	STRUCTURAL	FORMUL.	A
Mole % Pur.	Ref.	Molecu Formu	lar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	08	сн <sub>3</sub> сн сн <sub>2</sub> с = сн <sub>3</sub> сн	CHCH <sub>3</sub>	
<u> </u>		Ref		Weight 115,5	Ref			Ref
F.P. °C	Ī	1	dt/dP	1	1		1	-
F.P. 100%	<b>-</b>		*C/mm	l		f to	l	
B. P. *C			25°C	0.7214	5	h .		1
760 mm	112.	2	BP t <sub>e</sub>	0.0463	5	f' to		$\vdash$
100 30	53.31 27.41		30 mm	0.6486	5	g'   'K_		1
10	7.64	.   5	ΔHm cal/g			h'	l	İ.
1	-25.47	5	ΔHv cal/g	<del></del>	$\vdash$	m   to		
Pressure mm 25°C	26.48	. 5	25°C	82.42	5	n   <u>*K</u> -		ł
t <sub>e</sub>	1038.	5	30 mm BP	82.25 70.50	5 5			ļ
Density			1	68.98	5	m'   to		İ
g/ml 20°C t 25	0.72		te (d, e)	68.93	5	n'   °K_		l
d <sup>t</sup> 25 4 30	0.72		AHv/Te	19.52	5			₩
	0.74		d   27 to		5	Surface tension dynes/cm. 20°C	21.99	5
ь	-0.03		$\frac{1}{a}$ , $\frac{123}{15}$ $\frac{6}{6}$		5 5	30	21.02	5
Ref. Index			e'   15 to		5	40	20.08	5
<sup>n</sup> D 20°C	1.41		d g/ml	0.246	5	Parachor [P]		1
30	1.41		d g/ml v ml/g tc °C	4.060	5	20°C 30		
"C"	0.76	$\overline{}$	11 -	281.	5	40		_
MR (Obs.)	38.9	2	P <sub>c</sub> mm	19421.	5		335.2	5
MR (Calc.)	38.67	7 5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.05	4 2	30 mm	1.0000	5	Dispersion	124.	2
Dielectric	ļ		BP	0.9500 0.9405	5 5	Flash Point °C		
A 27 to B 142 °C	6.94	826   5	t <sub>e</sub> t <sub>c</sub>	0.256	5	Fire Point		<u> </u>
c — =	217.72		ΔHc kcal/m	<b>†</b>	<b>†</b>	M Spec.		1
A*  27 to	1.39		ΔHf		1 1	Ultra V. X-Ray Dif.		
B* 133 °C	1259.5	5	ΔFf	- <b> </b>		Infrared		
c		ŀ	Viscosity centistokes			Solubility in +		
t <sub>k</sub>   to			7 °C	:		Acetone Carbon tet.		
t <sub>x</sub> i °C	ļ		4	ŀ	1	Benzene		
A'   15 to B' 27 °C	7.35 1545.3	001   5				Ether n-Heptane		
c'	235.72		B <sup>V</sup> to			Ethanol		
A'* 15 to	1.74		AV	<u>:  </u>		Water Water in		
B'* 27 °C		5	(B <sup>V</sup> ) to			water in		+
Ac 142 to	7.35 1653.	93   5	(A <sup>V</sup> )  •C	; [				
Cc	257.	5	c <sub>p</sub> liq. °K					
Cryos. A°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	122 24		c, vap.					
$T_{\mathbf{R}} = 0.75$	123.34	5	•	1		+	<u> </u>	
REFERENC		OW 2 A	PI 3-Lit. 4-	Cala from de		grams/100 gran		<u>.t</u>
SOURCE:		AI		Calc. Irom de	dal	ta 5-Calc. by for	IIIUIA	
PURIFICAT	ION:	AI						
LITERATUE				<del></del>				

NAME	
Mole	
Mole	
Ref.   Ref.	Rei
Ref.	Ref
F. P. °C	
F.P. 100%   C/mm   25°C   0.7214   5	
B. P. °C 760 mm 112. 100 53.31 4 30 27.41 10 7.64 5 1 -25.47 5  AHm cal/g  AHv cal/g  Density g/ml 20°C dt 25 0.721 24 0.0363 5 6'	
The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The	
30	
The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same	
Pressure mm 25°C 26.48 5 82.62 5 7 82.42 5 7 82.42 5 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7 82.42 7	
mm 25°C   26.48   5   30 mm   82.25   5   0	_
Density   g/ml 20°C   0.725   2   d   27 to   86.05   d   123 °C   0.738   4   d   15 to   15 to   84.19   5   30   20°C   1.414   2   30   1.411   4   t   c   c   C   230   1.411   4   t   c   c   C   21.02   21.02   21.02   22.08   C   C   281.   5   C   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   21.02   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   20°C   2	+-
Density g/mi 20°C   0.725   2   te (d, e)   68.98   5   n'     to n'     - *K   0	
The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The column   The	- 1
a 0.741 4 0 27 to 86.05 5 0.1388 5 84.19 5 20°C 1.416 2 30°C 1.414 2 30°C 21.411 4 1 2 30°C 281. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81. 5 81.	-
a	+-
Ref. Index nD 20°C 1.416 2 dc g/ml 25 1.414 2 vc ml/g 4.060 5 30 21.02 30 1.411 4 t °C 281. 5 30	5
n <sub>D</sub> 20°C 1.416 2 dc g/ml 0.246 5 20°C 30 1.411 4 t °C 281. 5 30	5
25 1.414 2 d g/ml v <sub>c</sub> ml/g 4.060 5 20°C 30 1.411 4 t °C 2810 5	+-
30 1.411 4 t °C 281 5 30	į
"C" 0.7424 4 C 40	
U. 1634 4 Pc mm 19421. 5 Sugd. 335.2	5
MR (Obs.) 38.9 2 PV/RT Exp. L.1.%/wt.	
(nD-d/2) 1.054 2 25°C 1.0000 5 u. 1.0000 5 Dispersion 124.	2
Dielectric BP 0.9500 5 Plant Point 8C	+-
4   27 4     / 0402/   7    6     0    9405   5	
C 1341.1 5 C M. Spec.	1
A# 27 to 1 39246 5 AHf	
B*[133 °C   1259.5   5   AFI   Infrared	
K Viscosity centistokes Solubility in +	
t <sub>k</sub> to Acetone Contains to	
Benzene Benzene	
A'   15 to 7,35001 5 B'   27 °C   1545.3 5	
C! 235 72   5   BV   +0	
A' 15 to 1.74894 5 RV C Water in Water in	
27 9 1993.0 5 10 10	+
Ac  142 to 7,3593   5   (A <sup>V</sup> )  °C   Bc  tc °C   1653.   5   C lig °K	
Cc — 257. 5 p · · · · ·	
Cryos. A° c <sub>p</sub> vap. °K	
Consts. B	
	Ц.
TR = 0.75 T <sub>C</sub>	ent
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE:  API	
LITERATURE REFERENCES:	
1	

							No. 12	0
NAME	4, 4	-Dimethy	l-cis-2-hexene			STRUCTURAL		
-						CH <sub>3</sub>		
		<del></del>				сн <sub>3</sub> сн <sub>2</sub> с сн	=CHCH	
Mole	Ref.			Molecular	- 1	Ċн <sub>3</sub>	3	
% Pur.		Formu		Weight 112.	_			
		Ref	<b> </b>	·	Ref.			Ref.
F.P. C F.P. 100%	<del> </del>		dt/dP	ļ.		f to		
B, P. *C	<del> </del>		*C/mm 25*C	0.5755	5	g <u>*K</u> _		
760 mm	106.	2	BP	0.0457	4	h		-
100	48.11	4	t.	0.0363	5	f' to	ł	1
30 10	22.57 3.19		30 mm	0.6395	5	h' '^	1	
ī	-28.80		ΔHm cal/g	<u> </u>		<del></del>	<del> </del>	┼
Pressure	1		ΔHv cal/g 25°C	00.43	5	m to		
mm 25°C	34.01	l 5	30 mm	80.42 80.76	5	•   <b>-</b>	1	1
T <sub>e</sub>	1020.		BP	69.25	5	m'   to	<u> </u>	<del>                                     </del>
Density g/ml 20°C	0.72	22 2	te (d, e)	67.83 67.79	5	n'  K		
at 25	0.71	18 2	ΔHv/Te	19.53	5	o'		1
<b>4</b> 30	0.71		d   22 to	<del></del>	5	Surface tension		1
a b	0.73		e i 130 °C	0,1380	5	dynes/cm. 20°C	21.62	5
Ref. Index	-0.0	30   4	- to	5		30 40	20.68	5
n <sub>D</sub> 20°C	1.41	13 2	e' i •c		<del>  </del>	Parachor [P]	<del> </del>	Ť
- 25	1.41	11 2	d g/ml vc ml/g	0.248 4.038	5	20°C		
30	1.40		tc °C	272.	5	30 40		
"C"	0.76		P <sub>c</sub> mm	19212.	5		335.2	5
MR (Obs.) MR (Calc.)	38.8 38.67	77   2	PV/RT	<del> </del>		Exp. L.1.%/wt.		1
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	1	1.
Dielectric			BP BP	1.0000 0.9500	<b>5</b>	Dispersion	119.	2
A 22 to	6.94	277 5	t <sub>e</sub>	0.9410	5	Flash Point °C Fire Point		ì
B <u>  136 °C</u> C		. 5	tc	0.256	5	M Spec.	<u> </u>	+
	218.86		ΔHc kcal/m	1		Ultra V.		
A*  22 to B*  136 °C	1.39	347 5	ΔFf			X-Ray Dif. Infrared		1
к — — —			Viscosity			Solubility in +		╁
\$	-	Ì	r centistokes	. ]		Acetone	1	
tx C		ŀ	7			Carbon tet. Benzene	1	
A'   to			1			Ether		
B' ∟ _ •⊆		ł	B <sup>V</sup>   to	<del> </del>	-	n-Heptane		
A¹* to	<del> </del>		B to	: ]		Ethanol Water	1	
B'* *C		l	(BV) to	-1		Water in		
Ac   136 to	7.35		(A <sup>V</sup> )	1				
Bc t C	1627. 258.	5	c <sub>p</sub> liq. °K	<del></del>	$\vdash$			1
CE	238.	-   5	-11	i				1
Cryos, A° consts, B°			c <sub>p</sub> vap. *K	•				
t <sub>e</sub> °C	116.57	7 5	c <sub>v</sub> vap.	1				
$T_R = 0.7$			<u> </u>	1	Ц	+ grame/100 c==	no ectri	<del></del>
REFERENC		ow 2-A	PI 3-Lit, 4-	Calc from de	t de	'grams/100 gra ta 5-Calc, by for		16
SOURCE:			PI	ue	uel	_ J-Care, by for		
PURIFICAT	ION:		PI					
LITERATU								
MAIU.	REF	RENUE	J.					

	<del></del>						No. 1	<del></del> -
NAME	4, 4-Dim	ethyl	-trans-2-hexer	ie		STRUCTURAL	FORMUL	.A
						CH <sub>3</sub>		
						сн <sub>3</sub> сн <sub>2</sub> с сн	=CH CH <sub>2</sub>	
Mole		lecul		Molecular		с́н <sub>3</sub>	•	
% Pur.	Fo	rmula		Weight 112.20				
		Ref.			Ref.			Ref.
F.P. °C F.P. 1009		<del> </del>	dt/dP	1		f to		
	•	┢	°C/mm 25°C	0.5755	5	g '° <u>K</u>		
B. P. °C 760 mm	106.	2	BP	0.0457	4	h		—
100	48.11	4	t <sub>e</sub>	0.0363	5	f' to		
30 10	22.57 3.19	4 5	30 mm	0.6395	5	h'		
1	-28.80	5	ΔHm cal/g			m to		+-
Pressure			ΔHv cal/g 25°C	80,42	5	n   °K		1
mm 25°C	34.01 1020.	5	30 mm	80.76	5	0		
Density	1.020.	<del>  _</del>	BP	69.25	5	m' to		
g/m1 20°0	0.722	2	te te (d, e)	67.83	5	n'  K_		1
dt 25	0.718	2	ΔHv/T <sub>e</sub>	19.53	5			
<u> </u>	0.714	4	d 22 to	83,87	5	Surface tension	21. /2	-
a b	0.738 -0.0 <sub>3</sub> 8	4	<u>e   130 °C</u>	0.1380	5	dynes/cm. 20°C	21.62	5
Ref. Index	:		d'   to e'   °C	1		40	19.76	5
n <sub>D</sub> 20°0		2	d g/ml	0,248	5	Parachor [P]		
25 30	1.411	2 4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.038	5	20°C 30		
"C"	0.7613	4	* °C	272.	5	40		
MR (Obs.		2	P <sub>c</sub> mm	19212.	5	Sugd.	335.2	5
MR (Calc.	38.677	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.		
(nD-d/2)	1.052	2	30 mm	1.0000	5	Dispersion	121.	2
Dielectric	+	<u> </u>	BP	0.9500 0.9410	5	Flash Point °C		T
A 22 to B   136 °C		5	t e t c	0.256	5	Fire Point		
c '	218.86	5	∆Hc kcal/m			M. Spec. Ultra V.		
A*  22 to		5	ΔHf ΔFf			X-Ray Dif.		
B*[136 °C	1239.1	5	Viscosity		-	Infrared		<u> </u>
° – – -	_		centistokes	1	1	Solubility in TACETONE		
t <sub>k</sub> T to			η °C			Carbon tet.		
t <sub>x</sub> °C						Benzene Ether		
B' °C			F		-	n-Heptane		
C'	<del>-</del>	<b> </b>	B <sup>V</sup> to A <sup>V</sup> l °C	ŀ		Ethanol		
A'* to B'* °C			$\frac{A}{ B^{V} } - \frac{C}{to}$	1		Water Water in		
Ac   136 to		5	(A <sup>V</sup> )  °C	1				
Bc tc °C	1627.	5	- · · · · · · · · · · · · · · · · · · ·	<del>                                     </del>	$\vdash$			
Cc — -	258.	5	P	1				
Cryos, A <sup>c</sup> consts, B <sup>c</sup>			c <sub>p</sub> vap. °K	1				
t <sub>e</sub> °C	116.57	5	c vap.					
$T_{R} = 0.7$		ـــــــــــــــــــــــــــــــــــــــ	<u> </u>	L	Ц	+ arama/100	ma asla::	<del>_</del>
		2-A	PI 3-1.it 4-	Calc. from de	t de	grams/100 grants 5-Calc. by for		16
SOURCE:			PI	care. Irom de	ua	J-Guic, by lor		
PURIFICA	TION:		PI					
	RE REFERE							
	KEFERE	· · CE	<b>.</b> .					

No. 122 NAME 4,5-Dimethyl-cis-2-hexene STRUCTURAL FORMULA CH3CH CH CH=CHCH3 Molecular C8H16 сн,сн, Mole Ref. Molecular Weight 112, 208 % Pur Ref. Ref. Ref F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.6682 5 B. P. \*C h ВP 0.0461 4 760 mm 110. 2 ſ١ t<sub>e</sub> 0.0363 5 to 100 51.56 4 °K g' 30 25.78 4 30 mm 5 0.6456 10 6.10 5 h' ∆Hm cal/g 5 1 -26.86 to m AHv cal/g Pressure °K n 25°C 81.77 5 5 mm 25°C 28.82 o 30 mm 81.73 1032. 5 t<sub>e</sub> ВP 70.07 5 Density g/ml 20°C m to 5 te (d, e) 68.58 °K 0.725 2 68.53 5 ٥' 25 0.721  $\mathbf{d_{4}^{t}}$ AHv/Te 5 19.52 0.717 4 30 1 20 Surface tension 85.30 5 0.741 4 dynes/cm. 20°C 21.99 °C 0.1385 <u>| 121</u> ь 4 -0.038 5 30 21.02 ď to 5 40 20.08 Ref. Index e' °C 20°C 1.413 2 [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 0.247 5 25 24 1.411 20°C v<sub>c</sub> ml/g 4.051 5 1.408 30 30 °C 278. 5 ŧ<sub>c</sub> 40 "C" 0.7582 4 Sugd. 335.2 P<sub>c</sub> mm 19361. 5 5 MR (Obs.) 38.6 2 PV/RT Exp. L.1.%/wt. 38. 677 MR (Calc.) 25°C 1.0000 11. (nD-d/2) 1.051 2 30 mm 1,0000 Dispersion 119. 2 Dielectric 0.9500 5 BP Flash Point °C t<sub>e</sub> 0.9407 A | 20 to 6.94544 5 Fire Point tc 0.256 5 B 1140 °C 1333.6 M Spec. Ċ 218. 5 AHc kcal/m Ultra V. ΔHf A\* 20 to 1.39179 5 X-Ray Dif. ΔFf B\* 131 °C 1252.4 Infrared ĸ Viscosity Viscozi, centistokes °C Solubility in Acetone ţ ţ Carbon tet. •c Benzene to Ether В' •c n-Heptane B C١  $\tilde{\mathbf{A}}^{\mathbf{v}}$ Ethanol °C Water A'\* to Water in B'\* °C (BV) to Ac | 140 to 7.3566 (AV) °C Bc tc\_° °C 1645. 5 cp liq. °K 5 Сc 258. Cryos. Aº c<sub>p</sub> vap. ۰ĸ consts. B° c<sub>v</sub> vap. te °C 121.08 5  $T_{R} = 0.75 T_{c}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME									No. 1			
Molecular   Formula	NAME	4,5-Dime	thyl-	trans-2-hexene			STRUCTURAL FORMULA					
Molecular   Formula							CI	н сн сн сн	H=CH CH.			
Formula	Mole	Pof Mo		,	Malaawla m		-		3			
Ref.						8		30113				
F. P. 100%   S. P. ° C   T. 100			Ref.	r	Γ		Γ			Ref.		
F. P.   100%	F. P. °C			dt/dP			f	to				
The content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the	F.P. 1009	6		°C/mm								
100		1,,,					h					
1				1		5						
The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same   The same		25.78	4	30 mm	0.6456	5	- 1	; <mark>•</mark> K				
Pressure mm 25°C				∆Hm cal/g			<del>:</del>					
mm 25°C	Pressure											
Density g/ml 20°C   0,725   2   4   25   0,721   2   4   20 to   68.58   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   5   68.53   69   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70.07   70	1							<u> </u>				
Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Selection   Sele		1032.	5		70.07	5	m'	to		+		
d	g/ml 20°C	0.725	2	t <sub>e</sub> (d. e)			n'					
Ref. Index		0.721	2		1	i i	o'			<u></u>		
S		<del></del>			<u> </u>	$\perp$						
Ref. Index nD 20°C				_e <u>  121 °C</u>			dyne					
No.   20°C   1.413   2   2   30   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   4   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408   1.408			1						20.08	5		
1.41	n <sub>D</sub> 20°0	1.413		<del></del>	0.247	-	Para	chor [P]				
NR (Obs.)   38.6   2   PV/RT   25°C   1.0000   5   MR (Calc.)   1.051   2   30 mm   1.0000   5   MR (Dobs.)   333.6   2   PV/RT   25°C   30 mm   1.0000   5   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.   2   Ms.	45			V <sub>C</sub> m1/g		5						
MR (Obs.)   38.6   2			-	t <sub>c</sub> °C		1 1						
MR (Calc.)   38, 677   5   25°C   1,0000   5   1,0000   5			-		19361.	5			335.2	5		
Dielectric	MR (Calc.	38.677	5		1 0000	_	Exp.					
A 20 to B 1333.6			2	30 mm			Disp		121.	2		
R   20 to   1.333.6   5   C   218.   5   AHc   kcal/m   AHf   AFf		1					Flas	h Point C		†		
A	A 20 to	6.94544		te			Fire	Point				
A*   20 to B*   131 °C   1252.4   5   AFf				ΔHc kcal/m		$\Box$						
Note					1							
Centistokes   Centistokes   Centistokes   Acetone   Carbon tet.		1252.4	5		<del> </del>		Infra	red		<u> </u>		
Carbon tet.   Benzene   Ether   n-Heptane   Ethanol   Water   water in	с	_		centistokes	1							
Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Reference   Refe	t <sub>k</sub> to			η °C								
B'   - °C	X											
B'* °C   (B')   to   Water in	lB'i °C			<u> </u>	<del></del>							
B'* °C   (B')   to   Water in		-		By to								
Ac   140 to   7.3566   5   (A <sup>V</sup> )  °C     C <sub>C</sub>   t <sub>C</sub> °C   1645.   5   5   5   C <sub>p</sub> liq. °K   C <sub>p</sub> vap. °K   C <sub>p</sub> vap. °K   C <sub>p</sub> vap. T <sub>R</sub> = 0.75 T <sub>C</sub>				<del></del>	1							
Bc   tc   °C   1645.   5   cp   liq. °K	Ac  140 to		5	1	1							
Cryos. A°	Bc t <sub>c</sub> °C	1645.	5			$\vdash$						
consts. B°         P           te °C         121.08         5         cv vap.           TR = 0.75 Tc         * grams/100 grams solvent           REFERENCES: 1-Dow         2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula           SOURCE:         API           PURIFICATION:         API			5	ll	1							
te °C         121.08         5         c vap.         f grams/100 grams solvent           TR = 0.75 Tc         * grams/100 grams solvent           REFERENCES: 1-Dow         2-API         3-Lit.         4-Calc. from det. data         5-Calc. by formula           SOURCE:         API           PURIFICATION:         API				c <sub>p</sub> vap. °K								
T <sub>R</sub> = 0.75 T <sub>C</sub>		+	5	c <sub>w</sub> vap.								
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula  SOURCE: API  PURIFICATION: API			L	<u> </u>	L	اـــــا	+ 07:	ams/100 gra	ms solve	nt		
SOURCE: API PURIFICATION: API			2-A	PI 3-Lit. 4-0	Calc, from de	t. da						
PURIFICATION: API								<b></b>				
		TION:										
							***					

								4
NAME	5,5-	Dimethyl	-cis-2-hexene	·		STRUCTURAL	FORMUL	A
						CH <sub>3</sub>		
	T					сн, с сн, сн	=CHCH,	
Mole	Ref.			Molecular		с́н <sub>3</sub>	,	
% Pur.		Formul	a 8-16	Weight 112.2		3		-
		Ref.		<del></del>	Ref			Rei
F.P. *C	ļ		dt/dP			f to		ł
F.P. 100%			*C/mm 25*C	0.5950	· <sub>5</sub>	g <u>°K</u> _		
B. P. *C 760 mm	106.9	2	BP	0.0458	4	h		┷
100	48.88		t <sub>e</sub>	0.0363	5	f' to		
30	23.29		30 mm	0.6409	5	g'   'K_	1	1
10 1	3.86		ΔHm cal/g			h!	L	ــــــــــــــــــــــــــــــــــــــ
Pressure	-20.17		ΔHv cal/g			m   to	l	ì
mm 25°C	32.77	5	25°C	80.74	5	n   •K_	1	
t <sub>e</sub>	1023.	5	30 mm BP	80.97 69.42	5 5	ļ <u>-</u> -		₩
Density			t	67.99	5	m'   to		
g/ml 20°C	0.71		'e (u, e)	67.94	5	n'   •K_	1	1
dt 25 4 30	0.71		AHv/T <sub>e</sub>	19.52	5		<b> </b>	+
1	0.73		d   23 to		5	Surface tension dynes/cm, 20°C	21.02	5
ь	-0.03		<b>-</b> a,-¦ -1¹8- <b>:</b> 3		5	30 30	20.00	5
Ref. Index		1	e'   *C			40	19. <b>0</b> 0	5
n <sub>D</sub> 20°C	1.41		d <sub>c</sub> g/ml	0, 242	5	Parachor [P]		
25 30	1.40		V mi/g	4.128	5	20°C 30		
"C"	<del> </del>	-	t <sub>c</sub> ·C	271.	5	40	1	1
	0.76	+-	P <sub>c</sub> mm	18758.	5		335.2	5
MR (Obs.) MR (Calc.)	38.89 38.67		PV/RT			Exp. L.1.%/wt.		Г
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	1110	,
Dielectric			BP	0.9500	5	Dispersion	119.	2
A   23 to		315 5	t <sub>e</sub>	0.9409	5	Flash Point °C Fire Point		1
B (135 °C	1322.7	5	tc	0.256	5	M Spec.	<del> </del>	+-
C	218.69	$\longrightarrow$	ΔHc kcal/m ΔHf			Ultra V.		
A* 23 to B* 126 °C	1.39	286   5	ΔFf	1		X-Ray Dif.	1	
K 1120 C		١	Viscosity		$\vdash$	Infrared	<b> </b>	-
c	_		centistokes	1		Solubility in + Acetone	}	]
tk to	1		η •α			Carbon tet.		1
t c	<del> </del>		l			Benzene		
B' C	1			1		Ether n-Heptane		
C,	<u></u>		B <sup>V</sup> to			Ethanol	ļ	
A¹* to			A <sup>V</sup> C	_}		Water	ļ	
B'* *C	+		(B <sup>V</sup> ) to	1		Water in	<del> </del>	+
Ac   135 to			(A <sup>V</sup> ) °C		L I		1	
Bc i_tc_°C	1629. 258.	5	c <sub>p</sub> liq. °K					
Cryos. A°	1200.		ll -					
consts. B°	1		c <sub>p</sub> vap. °K	1		1		
t, °C	117.58	5	c <sub>w</sub> vap.					
$\mathbf{T}_{\mathbf{R}} = 0.79$			<u> </u>	1	<u></u> i	+ ===== (100	<u> </u>	
		OW 2_AT	PI 3-Lit. 4-	Calc from de		grams/100 gran		10
SOURCE:			PI	Calc. Irom de	. dat	ta 5-Calc. by for	muta	
	TON		<del></del>					
PURIFICAT			PI					
LITERATU	RE REFI	ERENCES	<b>i:</b>					

							No. 1	25
NAME _	5,5-Dim	ethyl-	trans-2-hexene			STRUCTURAL	FORMUI	LA
1						CH <sub>3</sub>		
<b></b>	T 1					CH <sub>3</sub> C CH <sub>2</sub> CH	=CHCH2	
Mole	Ref. M	olecul		Molecular		с́н,	,	
% Pur.	F			Weight 112.20				-
	<del></del>	Ref.			Ref.			Ref.
F. P. °C	ļ	-	dt/dP			f to		
F.P. 100%	+	┼	*C/mm 25*C	0.5363	5	g <u>'°K</u>		
B. P. °C 760 mm	104.1	2	BP	0.0455	4	h		+
100	46.46	4	t <sub>e</sub>	0.0363	5	f' to		
30	21.04	4	30 mm	0.6366	5	g'° <u>K</u>		1
10 1	1.74	5	∆Hm cal/g			h'		
Pressure	1	1	ΔHv cal/g	ļ		m to		
mm 25°C	36.79	5	25°C 30 mm	79.74 80.29	5			
t <sub>e</sub>	1015.	5	BP	68.84	5	m¹ to		+-
Density		_	t <sub>e</sub> ,	67.45	5	m' to		
g/ml 20°C	0.7066 0.7023	2 2	te (d, e)	67.41	5	0'		
d <sub>4</sub> 25	0.6980	4	ΔHv/T <sub>e</sub>	19.52	5	Surface tension		+-
a	0,7237	4	d 20 to	83.19	5	dynes/cm. 20°C	19.83	5
ъ	-0.0385	4	-å- 120 °C	0.1379	5	<b>8</b> 30	18.87	5
Ref. Index	1 4055	,	e' °C			40	17.94	5
<sup>n</sup> D 20°C	1.4055 1.4030	2 2	d <sub>c</sub> g/ml	0.24	5	Parachor [P]	İ	
30	1.4002	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.17 266.	5	30		
"C"	0.7646	4	tc°C Pc mm	18389.	5	40 Sugd	335.2	5
MR (Obs.)		2	PV/RT	10307.	-	Exp. L.1.%/wt.	333.2	+-
MR (Calc. (nD-d/2)	38.677	5 2	25°C	1.0000	5	u.		
Dielectric	1.0322	+-	30 mm BP	1.0000 <b>0.</b> 9500	5	Dispersion	121.	2
A 20 to	6. 94095	5		0.9412	5	Flash Point C		
B 131 °C		5	t e t c	0.256	.5	Fire Point		
с — — —	219.22	5	AHc kcal/m			M. Spec. Ultra V.		
A* 20 to	1.39372		ΔHf ΔFf			X-Ray Dif.		
B*[122 °C K	- 1232.6	5	Viscosity		<b></b> -	Infrared		+
°		-	centistokes			Solubility in Acetone		
t <sub>k</sub> to			η °C		l	Carbon tet.		
A'   to	<del></del>	+			l	Benzene Ether		
B'	_		<del></del>		ļ	n-Heptane		
			B <sup>V</sup>   to A <sup>V</sup>   °C		l	Ethanol	İ	
A'* to B'* °C			<b>⊩ ⊸</b>	.		Water Water in		
Ac 131 to	7 2510	-	11					1
Bc tc °C		5		<del> </del>	├	1	-	
Cc	258.	5	c <sub>p</sub> liq. °K	1				
Cryos. A°			c <sub>p</sub> vap. °K			1		
consts. B°	1	+	c <sub>v</sub> vap.					i
t <sub>e</sub> °C	114.43	5	1 -v F.	L	<u>L</u>	L	1	
$T_{\mathbf{R}} = 0.7$						grams/100 gra		nt
	ES: 1-Dow			Calc. from de	t. da	ata 5-Calc. by for	mula	
SOURCE:			PI					
PURIFICAT	TION:	A	PI					
LITERATU	RE REFER	ENCE	S:					
]								

							<b>N</b> o. 12	6
NAME	3-Ethyl-	3-hex	cene			STRUCTURAL	FORMULA	A.
						CH CH C - C	ucu cu	
	2614	••		M-11		$CH_3CH_2C = C$ $C_2H_5$	n Cn <sub>2</sub> Cn <sub>3</sub>	3
Mole % Pur.	Ref. Mo	rmul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	208	25		
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f to		1
F. P. 1007	<b>'</b>	ļ	*C/mm 25*C	0.8401	5	g <u>*K</u>		
B. P. °C 760 mm	116.	2	BP	0.0467	4	h		┼
100 30	56.76 30.61	4	t <sub>e</sub> 30 mm	0.0363 0.6548	5 5	f' to		1
10	10.66	5	ΔHm cal/g	0.0340	╁┤	h'		
1	-22.77	5	ΔHv cal/g	<u> </u>	$\vdash$	m to		
Pressure mm 25°C	22.40	5	25°C	83.66 83.22	5	n •K	İ	
t <sub>e</sub>	1049.	5	30 mm BP	71.32	5	m' to		├
Density g/ml 20°0	0.729	2	te te (d, e)	69.73 69.67	5 5	n'  °K_		
dt 25	0.725	2	ΔHv/T	19.51	5	o' '		
4 30	0.721	4	d   31 to	87.48	5	Surface tension dynes/cm, 20°C	22.48	5
ь	-0.038	4	d' 1 128 ℃		5 5	30	21.50	5
Ref. Index		,	e'   15 to		5	40	20.54	5
D 25	1.416	2	d g/ml	0.246	5	Parachor [P] 20°C		
30	1.413	4	tc *C	4.057 287.	5	30 40		1
"C"	0.7626	4	P <sub>c</sub> mm	19646.	5		335.2	5
MR (Obs. MR (Calc.		5	PV/RT	1,0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.054	2	25°C 30 mm	1.0000	5	u. Dispersion	124.	2
Dielectric			BP t <sub>e</sub>	0.9500 0.9401	5 5	Flash Point °C		
B 147°		5	tc	0.256	5	Fire Point		┼
C	217.	5	ΔHc kcal/m ΔHf	1		M Spec. Ultra V.		
A* 31 to B* 138 °C		5	ΔFf			X-Ray Dif. Infrared		
K C	_		Viscosity			Solubility in +		$\vdash$
t <sub>k</sub> Te			centistokes 7 °C	1		Acetone Carbon tet.		1
t			•			Benzene		
B' _ 31 °		5			$\perp$	Ether n-Heptane		
C'	235.	5	B <sup>V</sup>   to			Ethanol Water		1
A'* 15 to B'* 31 °c		5 5	$\frac{1}{(\mathbf{B}^{\mathbf{V}})} - \frac{1}{\mathbf{t}_0}$	-	1 1	Water in		
Ac   147 to		5	(A <sup>V</sup> ) °C	1				
Bc tc_	C 1671. - 257.	5	c <sub>p</sub> liq. °K					
Cryos. A	,	Ė	c <sub>p</sub> vap. °K					
consts. B			c <sub>v</sub> vap.					
t <sub>e</sub> °C T <sub>R</sub> = 0.7	127.86	5	~ ·-P·	<u> </u>	$\perp$	L		J
	CES: 1-Dow	2-AI	PI 3-Lit, 4-0	Calc from de	t de	grams/100 grar ta 5-Calc. by for		<u>.t</u>
SOURCE:			PI	oute. Hom de	t. da	ia 3-Caic, by lori	IIuia	
PURIFICA	TION:	Al	PI					
LITERATU	RE REFERE	NCES	5:			-		

							<b>N</b> o. 12	27
NAME	2, 2-Dime	hyl-	cis-3-hexene			STRUCTURAL	FORMUL	·Α
					Ì	CH <sub>3</sub>		
Mole % Pur.	Ref. Mo	lecul mula	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	208	сн <sub>3</sub> с сн=сн сн <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	
		Ref.	u	T	Ref.			Ref.
F.P. °C F.P. 1009	-137,350	2	dt/dP °C/mm			f to g*K		
B. P. °C 760 mm	105.43	2	25°C <b>B</b> P	0.5627 0.0457	5 4	h		_
100	47.59	4	t <sub>e</sub>	0.0363	5	f' to		
30 10	22.07 2.71	4 5	30 mm	0.6388	5	h'=		1
1	-29.23	5	ΔHm cal/g	-	$\vdash$	m to		+
Pressure			ΔHv cal/g 25°C	80, 17	5	n   °K		
mm 25°C	34.89 1019.	5	30 mm	80.57	5	•		
t <sub>e</sub> Density	1017.	3	BP	69.08	5	m' to		<b>†</b>
g/ml 20°C	0.7128	2	t <sub>e</sub> (d, e)	67.67	5	n'   <u>•</u> K_		
dt 25	0.7086	2	ΔHv/T <sub>e</sub>	19.51	5	0'		
	0.7044	4	d 22 to	83,62	5	Surface tension		Τ
a b	0.7295 -0.0 <sub>3</sub> 83	4	_e <u>  125 °C</u>	0.1379	5	dynes/cm. 20°C	20.54	5
Ref. Index	<del></del>	1	d' to e' C			30 40	19.58 18.64	5
n <sub>D</sub> 20°C		2	<u> </u>		+	Parachor [P]		$\vdash$
- 25	1.4074	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	0.242 4.130	5 5	20°C		
30	1.4049	4	t <sub>c</sub> °C	269.	5	30 40		1
	0.7657	4	P <sub>c</sub> mm	18677.	5	Sugd.	335.2	5
MR (Obs.) MR (Calc.		2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0535	2	25°C 30 mm	1.0000	5	u. Dispersion	110	2
Dielectric			<b>B</b> P	0.9500	5	Flash Point °C	119.	+-
A 22 to		5	t t	0.9411 0.256	5	Fire Point		
B   134 °C	1316.7 218.97	5	ΔHc kcal/m	0.256	13	M. Spec.		
A* 22 to		5	ΔHf			Ultra V.		
B*  126 °C		5	ΔFf			X-Ray Dif. Infrared		
к ——	-		Viscosity			Solubility in +		+-
t <sub>k</sub> – tõ	- [		centistokes 7°C			Acetone		
t <sub>x</sub> °C	:		'			Carbon tet. Benzene		
A'   to						Ether		
B'°			Bv to		1	n-Heptane Ethanol		
A¹* to	1		AV I °C			Water		
B'* °C			(B <sup>V</sup> )  to			Water in		<u> </u>
Acl 134 to		5	(A <sup>V</sup> )  °C					
Bc tc °C	1622. - 258.	5	c <sub>p</sub> liq. °K					
Cryos. A°	<del></del>	-	c vap. °K					
t <sub>e</sub> °C	115 02	5	c <sub>v</sub> vap.					
$\frac{t_e}{T_R} = 0.7$	115.93		<u> </u>	L	لــــــــــــــــــــــــــــــــــــــ	+ ~~~~ (100		<u> </u>
	CES: 1-Dow	2-A	DI 3_I++ 4	Calc from 3	+ 4-	grams/100 gra		11
SOURCE:	515; I-DOW			Calc. from de	t. da	ta 5-Calc. by for	u.a	
	TION.		PI					
PURIFICA	RE REFERE		.PI					
2121110	ND NEI ERE	IVO E.	<b>.</b>					

· · · · · · · · · · · · · · · · · · ·						No.	128
NAME	2, 2-Di	imethyl	-trans-3-hexen	e		STRUCTURAL FORMU	LA
l	****					CH <sub>3</sub>	
						сн <sub>3</sub> с сн=сн сн <sub>2</sub> сн	
Mole	Ref. 1	Molecul		Molecular		CH <sub>3</sub>	•
% Pur.				Weight 112.2	_	- 3	<del>- E -</del>
		Ref.		r	Ref.		Ref
F.P. °C F.P. 100%	<u> </u>		dt/dP	1	i i	f to	ŀ
B. P. *C	<u> </u>	$\dashv$	*C/mm 25*C	0.4755	5	g <u>*K</u>	1
760 mm	100.85	2	BP	0.0452	4	h	
100 30	43.64	4	t <sub>e</sub>	0.0363	5	f' to to	
10	18.40	4 5	30 mm	0.6317	5	h'	
1	-32.33	5	ΔHm ca1/g	<b> </b>	-	m   to	
Pressure			∆Hv cal/g 25°C	78.56	5	n   <u>*K</u> ]	
mm 25°C	42.09 1006.	5 5	30 mm	79.47	5	0	ł
Density	1.000.	+-	BP	68.14 66.81	5	m¹   to	
g/ml 20°C	0.7039	2	te (d, e)	66.78	5	n'   <u>*</u> K_	
dt 25 4 30	0.6995		AHV/Te	19.53	5	0'	
	0.6949		d   15 to		5	Surface tension	
a b	0.7218 -0.0 <sub>3</sub> 88			0.1374	5	dynes/cm. 20°C   19.53 30   18.54	
Ref. Index		$\vdash$	d'   to		1 1	40 17.58	
n <sub>D</sub> 20°C				0, 239	5	Parachor [P]	
25 30	1.4037		d g/ml vc ml/g tc °C	4.189	5	20°C	-
"C"	0,7689			260.	5	40	1
MR (Obs.)	+	2	P <sub>c</sub> mm	18108.	5	Sugd. 335.2	5
MR (Calc.	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	-
(nD-d/2)	1.0544	2	30 mm	1.0000	5	u. Dispersion 121.	2
Dielectric		$\perp$	BP	0.9500	5	Flash Point °C	
A 15 to B 127 °C		9   5	t <sub>e</sub> t <sub>c</sub>	0.9415 0.256	5	Fire Point	
c E	219.84	5	AHc kcal/m	<del> </del>	<del>                                     </del>	M Spec.	1
A* 15 to	1,3933	4 5	ΔHf	ŀ		Ultra V. X-Ray Dif.	1
B*   120 °C	1221.2	5	ΔFf	}	-	Infrared	
c		1 1	Viscosity centistokes			Solubility in +	
tk			η ∘c	}	ļ .	Acetone Carbon tet.	
'x						Benzene	ĺ
A'   to				L		Ether n-Heptane	
C'			B <sup>V</sup> to			Ethanol	
A'* to			A <sup>V</sup> °C	_		Water Water in	
B'* °C	<del></del>		(B <sup>V</sup> ) to				+
Ac   127 to		5 5	(A <sup>V</sup> )  °C	L	ļ!		
Cc - c-	258.	5	c <sub>p</sub> liq. °K				
Cryos. A			c <sub>p</sub> vap. °K				
consts. B°	<del>                                     </del>	$\dashv$	c <sub>v</sub> vap.				
t <sub>e</sub> *C	110.77	5	-v	L	<u> </u>		
$T_R = 0.7$		•				grams/100 grams solve	ent
REFERENC	ES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by formula	
SOURCE:			PI				
PURIFICAT			PI	·			
LITERATU	RE REFER	ENCES	:				

No. 129 2, 3-Dimethyl-cis-3-hexene NAME STRUCTURAL FORMULA CH3CH C = CHCH2CH3 ҁҥӡҁҥ Mole Ref. Molecular Molecular C8H16 . % Pur. Weight 112.208 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm °K g 25°C 0.7791 5 B. P. °C h BP 0.0465 760 mm 114. 2 t<sub>e</sub> 0.0363 5 f١ 55.05 to 100 4 g† •<u>K</u> 30 29.03 4 30 mm 0.6516 5 10 9.17 5 h١ ∆Hm cal/g 5 -24.09 to m ΔHv cal/g 25°C Pressure n °K 83.06 mm 25°C 24, 33 5 o 30 mm 82.76 5 1043. t<sub>e</sub> BP 70.94 5 m' to Density 5 69.38 te (d, e) n' °K g/ml 20°C 0,728 2 69.33 ۰,  $d_4^t$ 25 0.724 2 AHv/T 5 19.52 30 0.720 4 Surface tension 29 86, 80 5 to 0.744 4 dynes/cm. 20°C 22.25 5 <u>.с</u> å, 135 0.1391 5 h -0.038 4 30 21.28 5 to 84.95 40 20.41 5 Ref. Index 29 0.0755 [P] nD 20°C 1.416 Parachor d<sub>c</sub> g/ml 0.247 5 25 2 1.414 20°C vc ml/g 4.05 30 1.402 4 30 t<sub>c</sub> 284. 5 40 "C" 0.7602 4 P<sub>c</sub> mm 19592. 5 Sugd 335.2 5 MR (Obs.) 38.7 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38,677 25°C 1.0000 5 (nD-d/2) 1.052 2 30 mm 1,0000 5 Dispersion 124. 2 0.9500 Dielectric BP 5 Flash Point C 0.9403 A 29 to 6.95106 1**348**.6 Fire Point 0.256 5 B 145 °C M. Spec. C 5 AHc kcal/m 217.34 Ultra V. **AHf** A\* 29 to 1.39312 5 X-Ray Dif. ΔFf B\* 140 °C 1266.7 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t<sub>x</sub> °C Benzene 15 to 7.35098 Ether В' 29 °C 1552.9 n-Heptane B<sup>V</sup> A 235.34 5 C to Ethanol °C Water A'\* 15 to 1.74916 5 Water in B'\* 29 °C (B<sup>V</sup>) 1453.3 5 to Ac 145 to (AV) 7.3620 5 °C Bc tc °C 1663. 5 c<sub>p</sub> liq. ۰ĸ Cc 257. 5 Cryos. A c<sub>p</sub> vap. ۰ĸ consts. B° c, vap. te °C 125.59 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

								No. 13	0
NAME	2,3-	Dime	thyl	-trans-3-hexen	e	_	STRUCTURAL	FORMULA	4
							CH <sub>3</sub> CH C = CH	сн,сн,	
Mole	Ref.	Mol	ecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular		¸ċн₃ċн₃	2 3	
% Pur.					Weight 112.2	-			<del></del>
	T		Ref.	ļ		Ref			Ref
F.P. *C F.P. 100%	<u> </u>			dt/dP *C/mm	1		f to		
B. P. *C	<del> </del>		$\vdash$	25°C	0.7791	5	g <u>*K</u>		ì
760 mm	114.		2	BP	0.0465	4	h		┼
100 30	55.05		4	t <sub>e</sub>	0.0363	5	f' to		
10	29.03 9.17		4 5	30 mm	0.6516	5	h'		l
1	-24.09		5	ΔHm cal/g	<b>_</b>		m   to		<del>                                     </del>
Pressure				ΔHv cal/g 25°C	83.06	5	n K		
mm 25°C	24.33		5	30 mm	82.76	5	•		
Te Te	1043.		5	BP	70.94	5	m'   to		
Density g/ml 20°C	0.72	8	2	te (d, e)	69. 38 69. 33	5 5	n'  °K_		
at 25	0.72	4	2	ΔHv/T	1 .	5	0'		į
4 30	0.72		4		19.52	5	Surface tension		Г
a b	0.74		4	d 29 to		5	dynes/cm. 20°C	22.25	5
	-0.03	<u>-                                    </u>	4	d'   20 to	84.95	5	30 40	21.28 20.41	5
Ref. Index	1.41	6	2	e' i 29 °C	<del></del>	5	Parachor [P]		+-
25	1.41	4	2	d g/ml	0. 247 4. 05	5 5	20°C		
30	1.40		4	t <sub>c</sub> *C	284.	5	30 40		1
"C"	0.76	02	4	P <sub>c</sub> mm	19592.	5		335.2	5
MR (Obs.)	38.7	,	2 5	PV/RT	<u> </u>	$\vdash$	Exp. L.1.%/wt.		<del>                                     </del>
MR (Calc.) (nD-d/2)	38.67 1.05		2	25°C	1.0000	5	u.		_
Dielectric				30 mm BP	1.0000 0.9500	5 5	Dispersion	124.	2
A   29 to	6.95	106	5	t <sub>e</sub>	0.9403	5	Flash Point °C Fire Point		
B 1145 °C	1348.6		5	t <sub>c</sub>	0,256	5	M Spec.	<b></b>	-
C	217.34	$\overline{}$	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A* 29 to B* 140 °C	1.39 1266.7	312	5	ΔFf			X-Ray Dif. Infrared		
K				Viscosity			<b></b>		╁
t.	ļ			centistokes	ł		Solubility in + Acetone		
t <sub>x</sub>   to t <sub>x</sub>   *C	}			η •c			Carbon tet.		1
A'   15 to	7,35	098	5			1 1	Benzene Ether		
B' _ 29 °C	1552.9		5	<del></del>		$\vdash$	n-Heptane		
C'	235.34	$\overline{}$	5	B <sup>V</sup> to	1		Ethanol Water		1
A'* 15 to B'* 29 °C	1.74 1453.3	916	5		-Í		Water in	İ	
Ac   145 to	7.36	20	5						
Bc t C	1663.		5		ļ				
Cc	257.		5	c <sub>p</sub> liq. °K					
Cryos, A° consts, B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	125.59		5	c <sub>w</sub> vap.					
$T_{R} = 0.75$							grams/100 gran	ns sol <b>ve</b> n	t
REFERENC	ES: 1-D	ow :	2-AF		Calc. from de	t. dat	ta 5-Calc, by for	nula	
SOURCE:			AP	PI .				<del>-</del>	
PURIFICAT			AP						
LITERATUI	E REFI	EREN	CES	:					

NAME	2,4	-Dim	ethyl	-cis-3-hexene			STRUCTURAL FORMULA				
								нзсн сн=с			
14-1-	<b>D</b>				M-11		•		H <sub>3</sub>	3	
Mole % Pur.	Res		lecul rmula		Molecular Weight 112.2	08		3	-3		
			Ref.			Ref.				Ref.	
F.P. °C				dt/dP			f	to			
B. P. °C	•		-	°C/mm 25°C	0.6431	5	g	<u>°K</u> _			
760 mm	109.0		2	BP	0.0460	4 5	h f'			+-	
100 30	50.6		4	t <sub>e</sub> 30 mm	0.0363	5	g'	to •K			
10	5.4	14	5	ΔHm cal/g	0.0441	<u> </u>	h'				
1	-26.7	8	5	ΔHv cal/g	<u> </u>	<del>                                     </del>	m	to			
Pressure mm 25°C	30.0	)6	5	25°C	81.47	5	n o	<u>•K</u>			
t <sub>e</sub>	1029.		5	30 mm BP	81.47 69.84	5	m'			+	
Density g/ml 20°	. 0.7	7178	2	t <sub>e</sub> (d, e)	68.38	5	n'	to K_			
dt 25	0.7	1135	2	ΔHv/T <sub>e</sub>	68.32 19.51	5	٥'				
		7095	4	d 20 to	84.93	5		face tension			
a b	-0.0	7343 7382	4 4	_e_ _130 °C		5	dyn 8	es/cm. 20°C 30	21.12	5	
Ref. Inde				d' to				40	19.21	5	
<sup>n</sup> D 20°0		1140 1114	2 2	d <sub>c</sub> g/ml	0.244	5	Par	achor [P]			
30		088	4	V mi/g	4.109 275.	5		30			
"C"	0.7	675	4	tc°C P <sub>c</sub> mm	18980.	5		40 Sugd.	335.2	5	
MR (Obs. MR (Calc.			2 5	PV/RT		<u> </u>	Exp	L.1.%/wt.		+-	
(nD-d/2)		551	2	25°C 30 mm	1.0000	5	_	u. persion	124	2	
Dielectric				BP	0.9500	5		sh Point C	124.		
A 20 to B 138 °C		4402	5	te tc	0.9407 0.256	5		e Point			
c 1138 7	218.2		5	∆Hc kcal/m		<u> </u>		Spec.		ŧ	
A* 20 to		9145	5	ΔHf ΔFf			X-I	ra V. Ray Dif.			
B*[135 °C	1248.8	,	5	Viscosity			<b> </b>	ared		<u> </u>	
c	-			centistokes 7°C				ubility in <sup>†</sup> eton <b>e</b>	}		
t <sub>k</sub> to				ŋ °c				rbon tet.			
A'  to B'  °C							Et	her	{		
c, 3	-			B <sup>V</sup> to A <sup>V</sup> °C				Heptane hanol			
A!* to				v			W:	ter iter in	[		
B'* °(		540	-	(B <sup>V</sup> )  to			- W	ret III		+	
Acl 138 to	1639.	549	5	(A <sup>V</sup> )  °C	<del> </del>						
Cc -	258.		5	c <sub>p</sub> liq. °K							
Cryos. A' consts. B'				c <sub>p</sub> vap. °K					-		
t <sub>e</sub> °C	119.9	6	5	c <sub>v</sub> vap.							
$T_{\mathbf{R}} = 0.7$	'5 T <sub>c</sub>						+ g1	ams/100 gra	ms solve	nt	
	CES: 1-	Dow			Calc, from de	t. da	ta 5	-Calc. by for	mula		
SOURCE:	TION			.PI							
PURIFICA LITERATI		FFRF		.PI							
	AL RE	. DRE.	MOE:	<b>,</b> .							
L											

							No. 132
NAME	2, 4-Dim	ethyl-	-trans-3-hexene	•		STRUCTURAL 1	ORMULA
						Сн3Сн Сн=С	CH CH
						cn <sub>3</sub> cn cn-с	
Mole % Pur.		lecul		Molecular Weight 112.2	:08	03	3
		Ref.	Ī		Ref.		Ref.
F.P. ℃			dt/dP			f   to	
F.P. 1009	•		*C/mm 25*C	0,6109	5	g <u>°K</u>	
B. P. *C 760 mm	107.6	2	BP	0.0459	4	h	
100	49.50	4	t <sub>e</sub>	0.0363	5	f' to	
30 10	23.86 4.40	4 5	30 mm	0.6419	5	h' '	
1	-27.70	5	ΔHm cal/g		_	m   to	
Pressure mm 25°C	21 02	5	ΔHv cal/g 25°C	81.00	5	n ºK	
t <sub>e</sub>	31.82 1025.	5	30 mm BP	81.16	5	0	
Density		<b>†</b>	t_	69.57 68.13	5	m'   to	
g/ml 20°0	0.7145 0.7101	2 2	te (4, 6)	68.08	5	n	
d <sub>4</sub> 25 30	0.7056	4	ΔHv/T <sub>e</sub>	19.52	5	Surface tension	
a b	0.7323	4 4	d 23 to e 130 °C		5	dynes/cm. 20°C	20.74 5
Ref. Index	-0.0388	+	d' to			30 40	19.71 5 18.71 5
n <sub>D</sub> 20°0	1.4126	2		0,240	5	Parachor [P]	
25 30	1.4101 1.4076	2	d g/ml vc ml/g	4.161	5	20°C	
"C"	0,7686	4	, C	272.	5	40	
MR (Obs.	39.13	2	P <sub>c</sub> mm	18641.	5		335.2 5
MR (Calc. (nD-d/2)	) 38.677 1.0554	5 2	25°C	1.0000	5	Exp. L.1.%/wt. u.	
Dielectric	<del></del>	+-	30 mm BP	1.0000 0.9500	5	Dispersion	
A 23 t		5	te	0.9409	5	Flash Point °C Fire Point	l i
B (135 °C		5	t <sub>c</sub>	0.256	5	M Spec.	
A*  23 to	<del></del>	5	ΔHc kcal/m ΔHf			Ultra V.	
B≠ 130 °0		5	ΔFf			X-Ray Dif. Infrared	
K C			Viscosity centistokes			Solubility in +	
t <sub>k</sub>		ĺ	η °c			Acetone Carbon tet.	
t	_1	<u> </u>		]		Benzene	
В' •						Ether n-Heptane	
C'	-	-	B <sup>V</sup> to A <sup>V</sup> C			Ethanol Water	
A'* to B'* *			(B <sup>V</sup> ) to		1	Water in	
Ac   135 to	7.3551	5	(A <sup>V</sup> ) °C				
Bc tc_°	257.	5	c <sub>p</sub> liq. °K			1	
Cryos. A		Ť	c <sub>p</sub> vap. °K				
consts. B			ll -				
t <sub>e</sub> °C	118.37	5	c <sub>v</sub> vap.	<u></u>			
$T_R = 0$ .						+ grams/100 gran	
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by for	nula
SOURCE:	TION:		PI				
PURIFICA	RE REFERE		.PI				
- IERAIL	NE KEFEKE	NUES	);				

No 133

·									No. 1	33
NAME	2,5	Dime	thyl	-cis-3-hexene			ST	RUCTURAL	FORMUL	A
							(	сн сн=с	нсн сн.	
Mole	Ref.	Mole	cul		Molecular			ċн <sub>3</sub>	ċн,	5
% Pur.		Form	mula	C <sub>8</sub> H <sub>16</sub>	Weight 112.20	08			, 	
			Ref.			Ref.				Ref.
F.P. °C F.P. 100				dt/dP	ļ		f	to		
B. P. °C	-			°C/mm 25°C	0.4958	5	g	° <u>K</u> _		
760 mm	102.		2	BP	0.0453 0.0363	4 5	h f'	4-		+-
100 30	19.31		4	t <sub>e</sub> 30 mm	0.6335	5	g'	to		
10	0.11	20	5	ΔHm cal/g	0.0333	<del> </del>	h'			
Pressure	-31.56	<del>`</del>	5	ΔHv cal/g	<u> </u>		m	to		
mm 25°C	40.18	,	5	25°C 30 mm	78, 96 79, 73	5	n o	<u>*K</u>		
t <sub>e</sub>	1009.		5	BP	68.37	5	m'	to		+
Density g/ml 20°0	0,71	ا ه	2	te te (d, e)	67.03 66.99	5	n'	<u>•</u> K_		
dt 25 4 30	0.70	6	2	ΔHv/T	19.52	5	ە'			1_
a 30	0.70		4	d 15 to	82, 38	5		face tension	20.22	_
ь	-0.03		4	_d'120 °C to	0.1374	5	gyn	es/cm. 20°C 30	20.22 19.31	5
Ref. Index		$\Box$		e' °C			<u> </u>	40	18.42	5
<sup>n</sup> D 20°0	1.40		2	d <sub>c</sub> g/ml	0.244	5	Par	achor [P] 20°C		
30	1.40	2	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4, 103 264.	5		30 40		
"C"	0.76	18	4	P <sub>c</sub> mm	18625.	5			335.2	5
MR (Obs. MR (Calc.		7	<b>2</b> 5	PV/RT			Exp	. L.1.%/wt.		
(nD-d/2)	1.05		2	25°C 30 mm	0.9989	5	Dis	u. persion	119.	2
Dielectric				BP	0.9500 0.9414	5	Fla	sh Point C		
A 15 to B 130 °C		698	5	te tc	0. 256	5		e Point		_
c	219.62		5	ΔHc kcal/m ΔHf				Spec. raV.		
A*  15 to		205	5	ΔFf				lay Dif. ared		
к — — -	_			Viscosity			l <del></del>	ability in +		+-
t <sub>k</sub> tē	-			centistokes り °C			Ac	etone		
x				•				rbon tet.		
A'  to								her Heptane		
c' '				B <sup>V</sup> to C			Et	hanol		
A'* to B'* °C								iter iter in		
Ac  130 to		83	5	(B') to						T
Bc tc °C	1608.		5	c <sub>p</sub> liq. °K						
Cryos. A	258.		2	1						
consts. B				P						
t <sub>e</sub> °C	112.07		5	c <sub>v</sub> vap.						<u></u>
$T_{\mathbf{R}} = 0.$								ams/100 gra		nt
	CES: 1-I	Dow 2		PI 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:	TION			PI PI						
PURIFICA LITERATU		ED EN							<del></del>	
LILERAIC	NE REF	er en	CES	<b>)</b> ;						
1										

· · · · · · · · · · · · · · · · · · ·	·						No. 134	4
NAME	2, 5-Dime	thyl	-trans-3-hexen	e	_	STRUCTURAL I	FORMULA	١.
						СН3СН СН=СН	сн сн.	
	2 ( )			)/-1			Ċн <sub>3</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	208	3	3	
		Ref.	F		Ref			Ref.
F, P. *C			dt/dP	1		f to		
F.P. 100%			*C/mm			gK		1
B. P. °C	100		25°C BP	0.4958 0.0453	5 4	h		}
7 <b>60 mm</b> 100	102. 44.62	2 4	t <sub>e</sub>	0.0363	5	f' to		
30	19.31	4	30 mm	0.6335	5	g'		l
10 1	0.1120 -31.56	5	∆Hm cal/g			h'		ļ
Pressure	<del> </del>	<del> </del>	AHv cal/g			m to		1
mm 25°C	40.18	5	25°C 30 mm	78.96 79.73	5 5	"		1
t <sub>e</sub>	1009.	5	BP	68.37	5	m1 1 to		-
Density g/ml 20°C	0.710	2	te (d.e)	67.03	5	m'   to		
<sub>a</sub> t 25	0.706	2	'e 'a', c'	66.99	5	0'		
<sup>4</sup> 30	0.702	4	ΔHv/T <sub>e</sub>	19.52	5	Surface tension		
	0.726	4	d   15 to		5	dynes/cm. 20°C	20.22	5
B-C Index	-0.038	4	-a to			30 40	19.31 18.42	5
Ref. Index n <sub>D</sub> 20°C	1.406	2	e' i °C		$\vdash$	Parachor [P]		<u> </u>
45	1.404	2	d g/ml	0.244 4.103	5	20°C		1
"C"	1.402	4	t <sub>c</sub> *C	264.	5	30 40		1
	0.7618	4	P <sub>c</sub> mm	18625.	5		335.2	5
MR (Obs.) MR (Calc.)	38.8 38.677	2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.051	2	25°C 30 mm	0.9989	5	u. Dispersion	119.	2
Dielectric			BP	0.9500	5	Flash Point °C	117.	<u> </u>
A 15 to		5	t <sub>e</sub>	0.9414 0.256	5	Fire Point		1
B 1130_°C	1304.5 219.62	5	t <sub>c</sub>	+	ابًا	M Spec.		
A* 15 to	1.39205	5	ΔHf	ſ		Ultra V. X-Ray Dif.		
B* 120 °C		5	ΔFf	<u> </u>	$\perp$	Infrared		
K — — —			Viscosity centistokes			Solubility in +		
t to	-		7 0	1	1 1	Acetone Carbon tet.		1
t <sub>x</sub> i						Benzene		İ
A' to B' C						Ether		i
č, – – <u>–</u>	•]		B <sup>V</sup>   to			n-Heptane Ethanol		
A¹* to			AV C	_		Water		
B'* °C			(B <sup>V</sup> ) to	1		Water in		<del> </del>
Ac   130 to		5 5	(A <sup>V</sup> ) °C					
Bc _tc_°C	258.	5	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> *C	112.07	5	c <sub>v</sub> vap.					<u> </u>
$T_{\mathbf{R}} = 0.75$						+ grams/100 gran	ns solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc, from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:		A	PI					
PURIFICAT			PI					
LITERATU	RE REFERE	NCES	5 <b>:</b>					

									No. 1	35	
NAME	3,4	Dime	thyl-	cis-3-hexene			STRUCTURAL FORMULA				
				т-		_	С	$H_3CH_2C = C$			
Mole % Pur.	Rei	Mo Fo	lecula rmula	ar C <sub>8</sub> H <sub>16</sub>	Molecular Veight 112.20	08		Ċн <sub>3</sub> Ċı	H <sub>3</sub>		
			Ref.		T T	Ref.	Π			Ref.	
F.P. °C				dt/dP			f	1			
F.P. 1009			$\vdash$	°C/mm			g	to K		1	
B. P. °C			$\vdash$	25°C	1.0610	5	h				
760 mm	122.		2	BP	0.0473	4				+-	
100	62.0		4	t <sub>e</sub>	0.0363	5	f' g'	to _°K			
30 10	35.5		4	30 mm	0.6637	_5_		<u> </u>		1	
10	15.2		5	ΔHm cal/g			h'			-	
Pressure				ΔHv cal/g			m	to oK		1	
mm 25°C	17.3	3	5	25°C	85.63	5				1	
te	1066.		5	30 mm BP	84.76 72.64	5				↓	
Density				t <sub>e</sub>	70.94	5	m'   n'	to		1	
g/m1 20°C			2	te (d, e)	70.87	5	0'	•K			
d <sub>4</sub> 25	0.7		2	ΔHv/T <sub>e</sub>	19.52	5				<u> </u>	
	0.7		4	d 36 to	89.74	5		ace tension		1_	
a b	-0.0		4 4	_e_ _135 °C	0.1401	5	dyne	s/cm. 20°C 30	24.79 23.73	5	
Ref. Index	-	<u> </u>	+	d'   20 to	87.70	5	•	40	22.71	5	
n <sub>D</sub> 20°C		130	2	e'   36 °C	0.0826	5	Par	achor [P]	<b></b>	$\vdash$	
45	1.4		2	d g/ml	0.252	5		20°C		1	
30	1.4	25	4	vc ml/g tc °C	3.970 299.	5		30		1	
"C"	0.7	644	4	P <sub>c</sub> mm	20507.	5		40 Suad	335.2	5	
MR (Obs.)		3	2	PV/RT	20301.	-			333.2	+	
MR (Calc.			5	25°C	1.0000	5	Exp	. L.1.%/wt. u.			
(nD-d/2)	1.0	)57	2	30 mm	1.0000	5	Disp	persion	127.	2	
Dielectric				BP	0.9500	5		h Point C		+-	
A 36 to		5905	5	te t c	0.9396 0.256	5	Fire	Point			
B (_156 °C C	1377.7 215.8		5 5	ΔHc kcal/m	0.230	ļ <u> </u>	M.	Spec.			
A* 36 to		9265	5	ΔHf				aV.			
B*  145 °C			5	ΔFf				ay Dif. ared			
к ——	-			Viscosity		1		bility in +	ļ	+	
t <sub>k</sub> – to	-1			centistokes 7 °C				etone			
t <sub>x</sub>   °C				ŋ °C				rbon tet.		1	
A'   20 to	7 :	5168	5					nzene ner			
B' 36 ℃	1582.1		5			₩-		Heptane			
C'	233.8	32	5	B <sup>V</sup> to A <sup>V</sup> °C			Etl	nanol			
A'* 20 to		4680	5					ter ter in			
B'* 36 ℃			5	(B <sup>v</sup> )  to			<b>- "</b>		<del> </del>	+	
Acl 156 to	7.3	700	5	(A <sup>V</sup> )  °C					ĺ		
Bc tc °C	257.		5	c <sub>p</sub> liq. °K							
Cryos, A°			$\vdash$	1							
consts. B°				р -	[				1		
t <sub>e</sub> °C	134.6	3	5	c <sub>v</sub> vap.					İ	1	
$T_{\mathbf{R}} = 0.7$	5 T <sub>C</sub>						+ gr	ams/100 gra	ms solve	nt	
REFEREN		Dow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da		Calc. by for			
SOURCE:			A	PI					<del>-</del>		
PURIFICA	TION:			PI							
LITERATU		FERE						· · · · · · · · · · · · · · · · · · ·			
										-	

Mole % Pur.  F.P. °C F.P. 1009 B.P. °C			-trans-3-hexen	<u> </u>		STRUCTURAL			
% Pur. F.P. °C F.P. 100	Ref.	M-11							
F.P. *C F.P. 100		Formul		Molecular Weight 112.2	08	сн <sub>3</sub> сн <sub>2</sub> с = с сн <sub>3</sub> сн	сн <sub>2</sub> сн <sub>3</sub>	ţ	
F.P. 100		Ref			Ref			Ref	
F.P. 100		1	dt/dP	1	1	f   to	Γ	+	
B. P. *C	•	$\neg$	*C/mm		1 1	f to	ĺ		
	<del>                                     </del>		25°C	1.0610	5	h .	ĺ	}	
760 mm	122.	2	BP t	0.0473 0.0363	5	f' to		+-	
100 30	62.00 35.50	4 4	t <sub>e</sub> 30 mm	0,6637	5	g'    K_			
10	15.27	5	ΔHm cal/g	1,005.	╁┷┨	h'	ì	1	
1	-18.61	5	ΔHv cal/g		+-1	m   to		T	
Pressure mm 25°C	17.33	5	25°C	85.63	5	n °K_		1	
t <sub>e</sub>	1066.	5	30 mm	84.76	5	<u> </u>		1	
Density			BP	72.64 70.94	5 5	m'   to	l		
g/ml 20°		2	t <sub>e</sub> (d, e)	70.87	5	n'   °K_	i		
d <sub>4</sub> 25	0.743 0.739	2 4	ΔHv/T <sub>e</sub>	19.52	5		ļ	4_	
a .	0.763	4	d   36 to	89.74	5	Surface tension	24 70	5	
b	-0.038	4	135 °C		5	dynes/cm. 20°C	24.79 23.73	5	
Ref. Inde:			d'   20 to		5 5	40	22.71	- 5	
<sup>n</sup> D 20°0		2	d g/ml	0, 252	5	Parachor [P]			
30	1.428 1.425	2 4	d g/ml v ml/g	3.970	5	20°C 30	ļ	1	
"C"	0.7644		ic C	299.	5	40	ļ		
MR (Obs.		2	P <sub>c</sub> mm	20507.	5		335.2	5	
MR (Calc.	38.677	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.	İ		
(nD-d/2)	1.057	2	30 mm	1.0000	5	Dispersion	127.	2	
Dielectric			BP	0.9500	5 5	Flash Point °C		+	
A   36 t B   156 °		05   5	te t <sub>c</sub>	0.9396	5	Fire Point			
c —	215.82	5	ΔHc kcal/m	<u> </u>	$\dagger$	M Spec.		1	
A*   36 t	1.3920	55 5	ΔHf ΔFf		1 1	Ultra V. X-Ray Dif.		1	
B* 145 °	2 1294.3	5		ļ	+	Infrared			
c	_[		Viscosity centistokes	ì	1 1	Solubility in +		Ţ	
t <sub>k</sub>   t			η •c		1 1	Acetone Carbon tet.		1	
'x I		<u>,                                    </u>		ļ	i i	Benzene		į	
A'   20 t		58   5		İ	L	Ether n-Heptane	1	1	
C'	233.82	5	B <sup>V</sup> to			Ethanol		}	
A1# 20 t			A <sup>V</sup> C			Water Water in			
B'* 36 °		5	(B <sup>V</sup> ) to		1 1		<del> </del>	+	
Ac   156 t	7.3700 1700.	5   5	(A <sup>V</sup> )  °C		$\sqcup$		1		
Ccc_	257.	5	c <sub>p</sub> liq. °K	j	j ]		}	1	
Cryos, A			c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	134.63	5	c <sub>w</sub> vap.						
$T_R = 0.$						grams/100 gran	ns solver	at	
REFEREN	CES: 1-Dov	v 2-AF	PI 3-Lit. 4-0	Calc. from de	t. dat				
SOURCE:			PI						
PURIFICA	TION:	A:	PI						
LITERATU	RE REFER	ENCES	:	<del></del>					

								37
NAME	2-n-Prop	yl-1-	pentene			STRUCTURAL	FORMUL	Α
						CH: (CH ) C	СН	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	08	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> с с	3 <sup>H</sup> 7	
		Ref.			Ref.			Ref
F.P. °C	_	<u> </u>	dt/dP			f to		
F.P. 100	*	-	°C/mm 25°C	0,8975	5	g <u>°K</u>	-	1
B. P. °C 760 mm	117.7	2	BP	0.0469	4	h	<u> </u>	├
100	58.25	4	t <sub>e</sub>	0.0363	5	f' to g' <u>*K</u>	l	
30 10	32.00 -11.97	4 5	30 mm	0.6573	5	h'	·	1
1	-21.58	5	ΔHm cal/g		-	m to		+-
Pressure		_	ΔHv cal/g 25°C	84, 22	5	n K	.	
mm 25°C	20.83	5	30 mm	83.66	5	<u> </u>	l	
Density			BP te	71.69 7 <b>0</b> .07	5	m¹ to		
g/ml 20°		2	te (a, e)	70.01	5	n'   <u>*</u> K	1	l
dt 25 4 30	0.7198 0.7156	2	ΔHv/T <sub>e</sub>	19.51	5		<b></b>	<b>├</b> ─
a	0.7408	4	d 32 to	88.13	5	Surface tension dynes/cm, 20°C	21.87	5
Ъ	-0.03834	4	130 °C d 120 to	0.1397 86.20	5	<b>8</b> 30	20.86	5
Ref. Inde:		١, ١	e' 32 °C	0.0794	5	40	19.89	5
<sup>n</sup> D 20°	1.4111	2 2	d <sub>c</sub> g/ml	0. 243	5	Parachor [P] 20°C		
30	1.4085	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4.123 288.	5	30		
"C"	0.7603	4	P <sub>c</sub> mm	19365.	5	40 Sugd	335.2	5
MR (Obs. MR (Calc		<b>2</b> 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)	1.0516	2	25°C 30 mm	1.0000	5 5	u. Dispersion	123.	2
Dielectric	:		BP	1.000 <b>0</b> 0.9500	5	Flash Point C	123.	+-
A 32 to		5	t e t c	0.9400 0.256	5	Fire Point		
B (148 °C	216.64	5	ΔHc kcal/m	0.230	-	M. Spec.		
A* 32 to		5	ΔHf			Ultra V. X-Ray Dif.		
B*[138 °C		- 5	ΔFf			Infrared		
c			Viscosity centistokes			Solubility in +		
t <sub>k</sub> [ -te			η· •c			Acetone Carbon tet.		
A'  20 to		5				Benzene		
B' j_32 °		5		<del> </del>	ļļ	Ether n-Heptane		
C'	234.64	5	B <sup>V</sup> to A <sup>V</sup> °C			Ethanol	ł	
A'* 20 to B'* 32 °C		5	⊩. <u>-v</u>			Water Water in		
Ac 148 to		5	(B')  to					
Bc tc *	C 1677.4	5		<del>                                     </del>	$\vdash$		1	
Cc	256.71	5	Р.					
Cryos, A consts, B			c <sub>p</sub> vap. °K					
te °C	129.78	5	c <sub>w</sub> vap.					
$T_{R} = 0.$	75 T <sub>c</sub>					grams/100 gra	ms solver	nt
REFEREN	CES: 1-Dow	Z-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	rmula	
SOURCE:		А	PI					
PURIFICA			PI					
LITERATI	URE REFERE	NCE	5:					

F.P. °C									No. 13	38
Molecular   C <sub>8</sub> H <sub>16</sub>   Molecular   C <sub>8</sub> H <sub>16</sub>   Molecular   Weight   112.208   CH(CH <sub>3</sub> ) <sub>2</sub>	NAME	2-Is	ргору	71-1-	-pentene			STRUCTURAL	FORMUL	A
Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.		Ref.	Mole	cula	C <sub>8</sub> H <sub>16</sub>		208	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>2</sub> с= сн	сн <sub>2</sub> (Сн <sub>3</sub> ) <sub>2</sub>	
F.P. *C   F.P. 100%				_			_	<del></del>		Ref
F. P. 100%   C	F.P.°C	T	T		de/dD	T		6 1 1 45		1
B.P. "C   7.60 mm   113.			$\neg \uparrow$		°C/mm		1.		l	
100								h ¦	l	1
30			.					f' to		
1	30					0.6501	5	g'	ļ	
Pressure mm 25°C					ΔHm cal/g			h'		
The color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the color of the		-24.70	<del>`</del>	-		<del> </del>				
Density g/ml 20°C		25. 38		5					-	
Density g/ml 20°C   0,725   2   4 25   0,721   2   4 30   0,717   2   4 4 5   0,721   2   4 4 30   0,741   4 5   0,038   4   4	t <sub>e</sub>							<del></del>		┼
A   25 to   1, 44 to   25 to   1, 39279   B*   134 * C   C   C   C   C   C   C   C   C   C	Density		_		t	69.18	5			
A	g/ml 20°C				'e (u, e)	1	1 1		1	
Ref. Index	d4 30							Surface tensis	<del>                                     </del>	+
Ref. Index		+			::				21.99	5
No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.   No.	ъ	-0.03	8	4				30		5
1.412   2			.			:			20.08	13
"G"					d <sub>c</sub> g/ml					
MR (Obs.) 38.7	30	1.40	9	4	v ml/g			30	ļ	1
MR (Calc.) 38.677 25	"C"	0.75	99	4		1	1 1		335 2	5
No.   1.052   2   25°C   1.0000   5   Dispersion   122.			_			17,131.	╁┷┤		333.2	+-
Dielectric   Same   1,0000   5   5   5   6,94966   5   6   1344.9   5   6   1344.9   5   6   1.39279   5   6   1263.1   5   6   1   1   1   1   1   1   1   1   1					25°C	1.0000	5		ļ	
A   25 to   6,94966   5   1344.9   5   217.53   5   AHc   kcal/m   AHf   AFf	<del></del>	+	-+						122.	2
The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliation   The Foliatio		6 94	966	<u> </u>						}
A*   25 to   1.39279   5	B 1144 °C	1344.9	ļ	5		0.256	5			
A		+								į.
Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Viscosity   Centistokes   Viscosity   Centistokes   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosity   Viscosit			279							ļ
C	K LIST C	- 1203.1		1	Viscosity		$\Box$			+
Carbon tet.   Benzene   Ether   n-Heptane   Ethanol   Water   Water in				- 1		. 1				1
A'   to   B'   - °C   B'   to   A'     °C   B'   to   A'     °C   B'					יי י			Carbon tet.		
B	A' to	-	-+				1 1			1
A   + to     A   - C     Water   Water in		<u>2</u>		- 1		<del> </del>	┼┈┤	n-Heptane		
Bi		+	-+		AV C		1 1		1	İ
Ac   144 to   7.3606   5   (A <sup>V</sup> )   °C						-				
Bc   t_c °C   1658.   5   c_p   liq. °K	Ac   144 to	7.36	06	5	v. '	1				
Cryos. A°	Bcit °C	1658.		5		+	$\vdash$			
consts. B°         p         p         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c         c		257.		5						
T <sub>R</sub> = 0.75 T <sub>C</sub> + grams/100 grams solvent  REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det, data 5-Calc, by formula  SOURCE: API  PURIFICATION: API	consts. B°				р .					
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API				5	c <sub>w</sub> vap.	<u> </u>		L		
SOURCE: API PURIFICATION: API										ıt
PURIFICATION: API		<u> D</u>	ow 2			Calc. from de	t. dat	ta 5-Calc. by for	mula	
ALIERATURE REFERENCES:										
	LIERATO	RE REF	EKEN	CES	:					

							No. 13	39	
NAME _	2-Ethyl-	3-me	thyl-l-pentene			STRUCTURAL FORMULA			
						сн,сн,сн с	= CH,		
Mole	Ref. Mo	lecul		Molecular		с́н <sub>3</sub> с			
% Pur.	For	mula		Weight 112.20	08				
		Ref.			Ref.		,	Ref.	
F. P. °C F. P. 100%	-	_	dt/dP °C/mm			f to			
B. P. °C		-	25°C	0.7348	5	g '° <u>K</u> h			
760 mm	112.5	2	BP t <sub>e</sub>	0.0464	4 5	f' to		$\vdash$	
100 30	53.73 27.79	4	30 mm	0.6495	5	g'   '° <u>K</u>			
10 1	7.99 -25.16	5	ΔHm cal/g			h'		<u> </u>	
Pressure	23.10		ΔHv cal/g			m to			
mm 25°C	25.96	5 5	25°C 30 mm	82.73 82.35	5				
Density	1039.	)	BP	70.59 69.06	5	m¹ to			
g/ml 20°C	0.729	2	t <sub>e</sub> (d, e)	69.00	5	n'   <u>•K</u>	ļ		
d <sub>4</sub> 30	0.725 0.721	2 4	AHv/T <sub>e</sub>	19.51	5		ļ	<del> </del>	
a	0.745	4	d 25 to e 124 °C	86, 20	5 5	Surface tension dynes/cm. 20°C	22.48	5	
ь	-0.038	4	-d1 10	0.1388	"	30 40	21.50 20.54	5	
Ref. Index	1.4142	2	e'   °C		_	Parachor [P]	20.51	+-	
25 30	1.4118	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	0.248 4.038	5	20°C			
"C"	1.4094 0.7561	4	v <sub>c</sub> m1/g t <sub>c</sub> °C	282.	5	30 40			
MR (Obs.)	38.5	2	P <sub>c</sub> mm	19563.	5	Sugd.	335, 2	5	
MR (Calc.) (nD-d/2)	38.677	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt. u.			
Dielectric	1.050	2	30 mm BP	1.0000	5	Dispersion	122.	2	
A 25 to	6.94746	5	t.	0.9500 0.9404	5	Flash Point C Fire Point			
B (143 ℃ C	1342.5 217.62	5 5	tc ΔHc kcal/m	0.256	5	M. Spec.		╁	
A*  25 to	1.39112	5	∆Hf			Ultra V. X-Ray Dif.			
B* 134 °C	1260.8	5	ΔFf		<u> </u>	Infrared			
K — — —			Viscosity centistokes			Solubility in +			
t <sub>k</sub> to			η ℃			Acetone Carbon tet.	ŀ		
A'I to		_				Benzene Ether			
B'°C			BV I		-	n-Heptane			
A¹* to		-	B <sup>V</sup>			Ethanol Water			
B'≠ °C			(B <sup>V</sup> )  to			Water in		<del> </del>	
Ac 143 to Bc to °C	7.3586	5	(A <sup>V</sup> )  °C						
Bc t <sub>c</sub> °C	1656. 258.	5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K						
te °C	123.91	5	c <sub>w</sub> vap.						
$T_{\mathbf{R}} = 0.75$						grams/100 gra	ms solver	nt	
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for			
SOURCE:			PI						
PURIFICAT			PI						
LITERATUR	RE REFERE	NCES	<b>5:</b>						

							No. 140
NAME	2-E	thyl-4-m	ethyl-l-pentenc	<u> </u>		STRUCTURAL FO	ORMULA
						CH CH CH C	- CU
						CH <sub>3</sub> CH CH <sub>2</sub> C	
Mole % Pur.	Ref.	Molecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular Weight 112.2	08	CH <sub>3</sub> C <sub>2</sub>	<b>~</b> 5
<u> </u>		Ref.		weight 112.2	Ref		Ref
F.P. °C	T	1.61.	dt/dP	T	1		
F.P. 100%	<del>                                     </del>		*C/mm			f to g *K	-
B. P. *C			25°C	0.6758	5	h	l
760 mm 100	110.3	2 4	BP t <sub>e</sub>	0.0462 0.0363	5	f' to	
30	26.02		30 mm	0.6461	5	g'   'K_	ļ
10 1	6.32		ΔHm cal/g			h'	
	-26.66		ΔHv cal/g	<del>                                     </del>		m to	
Pressure mm 25°C	28,46	. 5	25°C	81.95	5	n <u>*K</u>	
. t <sub>e</sub>	1033.	5	30 mm BP	81.80 70.12	5		
Density			11 +	68,63	5	m' to	
g/ml 20°C	0.71		t_ (a,e)	68.58	5		ŀ
d <sub>4</sub> 25 30	0.71		ΔHv/T <sub>e</sub>	19.51	5	Surface tension	
	0.73		d 20 to		5	dynes/cm. 20°C	21.33 5
Ъ	-0.03	85 4		7		30	20.31   5
Ref. Index	1.41	05 2	e' 'c		$\sqcup$	Parachor [P]	19.33 5
25	1.40		d <sub>c</sub> g/ml	0.242 4.128	5	20°C	l
30	1.40	53 4	d g/ml vc ml/g tc °C	277.	5	30	
"C"	0.75		P <sub>c</sub> mm	18963.	5	40 Sugd. 3	35.2 5
MR (Obs.) MR (Calc.)	38.68 38.67		PV/RT	<del> </del>	╁	Exp. L.1.%/wt.	
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	.
Dielectric			BP BP	1.0000 0.9500	5		22. 2
A 20 to			t <sub>e</sub>	0.9406	5	Flash Point °C Fire Point	i
B (139 °C	1334.6	5 5	t <sub>c</sub>	0, 256	-	M Spec.	
A*  20 to	<del></del>	<del></del>	ΔHc kcal/m ΔHf			Ultra V.	1
B* 130 °C	1.39 1253.3	159   5	ΔFf	L		X-Ray Dif. Infrared	
K – – –	1	- 1	Viscosity			Solubility in +	
1k   10	1		centistokes 7 °C	1		Acetone	ì
t⊊i •c	İ		•	1		Carbon tet. Bensene	
A' to B' C				1		Ether	
B' •C	-1		B <sup>V</sup>   to			n-Heptane Ethanol	
A'* to			A <sup>V</sup> C	_[		Water	
B'* *C			(BV) to	1		Water in	
Ac   139 to	7.35	63 5	(A <sup>V</sup> )  °C	1			
Bc tc C	257.	5	c <sub>p</sub> liq. •K				1
Cryos. A			c <sub>p</sub> vap. °K	1			
te °C	121,42	5	c <sub>w</sub> vap.				
$T_{\mathbf{R}} = 0.7$			1	<u> </u>	<u> </u>	+ grams/100 gram	
REFERENC		ow 2-AI	PI 3-Lit. 4-0	Calc from de	t da	ta 5-Calc. by form	
SOURCE:		AF		40			
PURIFICAT	ION:	AF	<del></del>				
LITERATU							

No. 141 3-Ethyl-2-methyl-1-pentene NAME STRUCTURAL FORMULA  $CH_3CH_2CH$   $C = CH_2$ Ċ<sub>2</sub>H<sub>5</sub>ĊH<sub>3</sub> Mole Ref. Molecular Molecular C8H16 % Pur. Formula Weight 112, 208 Ref. F.P. °C F.P. 100% dt/dP f to °C/mm <u>°K</u> g 25°C 0.6682 5 B. P. °C h BP 0.0461 760 mm 110. <sup>t</sup>e 0.0363 5 ſ١ 51.56 to 100 4 g' <u>°к</u> 4 30 25.78 30 mm 0.6456 5 10 6.10 5 h' ∆Hm cal/g -26,86 m to AHv cal/g Pressure n ۰ĸ 25°C 81.84 mm 25°C 28.82 0 30 mm 81.73 te 1032. 5 ΒP 70.07 5 m' to Density te (d, e) 68.59 5 n' °K g/ml 20°C 0.730 0.726 2 68.54 5 ٥' 25 ž  $\mathbf{d_{4}^{t}}$ ΔHv/Te 5 19.52 30 0.722 4 Surface tension 20 to 85, 30 5 0.746 4 dynes/cm. 20°C 22,60 ٠Ć 0.1385 121 Ъ -0.038 4 30 5 21.61 ďΠ to 40 20.65 Ref. Index 5 e' °C 20°C 1.415 2 Parachor [P]  $^{n}D$ d<sub>c</sub> g/ml 0.249 25 1.413 2 20°C vc ml/g t °C 4.020 30 1,410 4 30 t<sub>c</sub> 279. 5 "C" 40 0.7564 4 P<sub>c</sub> mm 19545. 5 Sugd. 335.2 5 MR (Obs.) 38.5 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 25°C 1.0000 5 (nD-d/2) 1.050 2 30 mm 1.0000 2 Dispersion 122. Dielectric ВP 5 0.9500 Flash Point C 0.9407 5 A 20 to 6.94544 5 Fire Point 0.256 В \_141 °C 1333.6 M. Spec. С 218. 5 ΔHc kcal/m Ultra V. ΔHf A\* 20 to 1.39179 5 X-Ray Dif. ΔFf B\*[131°C 1252.4 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. t<sub>x</sub> °C Benzene A'I to Ether В' °C n-Heptane B<sup>V</sup> | C to Ethanol A'\* °C Water to (B<sup>V</sup>) Water in B'\* °C to Ac | 141 to 7.3568 5 (AV) °C Bc tc °C 1645. c<sub>p</sub> liq. °K 258. 5 Cryos, A° consts, B° cp vap. °K c<sub>w</sub> vap. te °C 121.08  $T_{\mathbf{R}} = 0.75 \, \mathbf{T_{\mathbf{c}}}$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: **PURIFICATION:** LITERATURE REFERENCES:

NAME	3-E	thy1-3-	-me	thyl-1-pentene			STRUCTURAL	No. 14 FORMUL	
							CH <sub>3</sub>		
	7.6				Malala	$\neg$	CH <sub>3</sub> CH <sub>2</sub> C	CH=CH <sub>2</sub>	
Mole % Pur.	Ref.		mul		Molecular Weight 112.2	208	с <sub>2</sub> н	5	
			Ref.			Ref	T		Ref
F.P. °C	T			1. (15	T	1		1	-
F.P. 100%	+		$\dashv$	dt/dP *C/mm			f to		
B. P. °C	<del>                                     </del>	+		25°C	0.7214	5	h .		İ
76 <b>0 mm</b>	112.	- 1	2	BP	0. 0463 0. 0363	5	f' to		$\vdash$
100 30	53.31		4 4	t <sub>e</sub> 30 mm	1	5	g'   K		
10	7.64		5		0.6486	-	h'	1	1
1	-25.47		5	ΔHm cal/g		$\vdash$	m   to		+
Pressure				ΔHv cal/g 25°C	82.45	5	n '°K		1
mm 25°C	26.48 1038.	3	5	30 mm	82.25	5	•	]	
Te te	1038.		-	BP	70.50	5	m' to		$t^-$
Density g/ml 20°C	0.73	305	2	te te (d, e).	68.99	5 5	n'  K_	1	
at 25	0.72	264	2	ΔHv/T <sub>e</sub>	19.52	5	o' '		1
<sup>4</sup> 4 30	0.72	223	4		<del></del>		Surface tension	<b></b>	T
	0.74		4	d 20 to		5 5	dynes/cm. 20°C	22.66	5
ь	-0.0	381	4	a'   - to	1	`	30 40	21.65 20.67	5
Ref. Index		.	2	e' i °C	<u> </u>	L_	Parachor [P]	20.01	+-
<sup>n</sup> D 20°C	1.4		2	d <sub>c</sub> g/ml	0.248	5	20°C		
30	1.41	13	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	4. 039 282.	5	30		1
"C"	0.76	511	4	P <sub>c</sub> mm	19557.	5	40 Sugd	335.2	5
MR (Obs.)			2	PV/RT	17,331.	<u> </u>	Exp. L. 1, %/wt.	333.2	1
MR (Calc. (nD-d/2)			5	25°C	1.0000	5	u.		
	1.05	23	2	30 mm	1.0000	5	Dispersion	117.	2
Dielectric	1			BP t <sub>e</sub>	0.9500	5	Flash Point °C		
A 20 to		1826	5	t c	0. 256	5	Fire Point		<u> </u>
č	217.72	2	5	ΔHc kcal/m			M Spec.	1	
A*   20 to	1.39	246	5	ΔHf		1 1	Ultra V. X-Ray Dif.		
B* 133 °C		- 1	5	ΔFf		<b>├</b> ─┤	Infrared		
K	1	1		Viscosity centistokes			Solubility in +		
ել		İ		η °c	}		Acetone Carbon tet.		
t <sub>x</sub> l	<u> </u>				İ		Benzene		l
A' to B' c							Ether		
c, L S	4	-		B <sup>v</sup> to			n-Heptane Ethanol		ļ
A'* to			$\neg$	AY C			Water		1
B'* °C		ľ		(B <sup>V</sup> ) to	1		Water in		<del> </del>
Ac   143 to		94	5	(A <sup>V</sup> ) °C	1				1
Bc tc_°C	1 - 0	1	5	c <sub>p</sub> liq. °K	+			1	1
	258.							1	
Cryos. A° consts. B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	123.34	1	5	c <sub>v</sub> vap.	L		<u> </u>		<u> </u>
$T_R = 0.7$							grams/100 gra		ıt
REFEREN	.೯೦: 1-L	70W 2			Jaic, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			AI						
PURIFICAT			AI						
LITERATU	RE REF	EREN	CES	:					

							No. 14	3
NAME _	3-Ethyl-4	-met	hyl-l-pentene			STRUCTURAL	FORMUL	A
						сн <sub>3</sub> сн сн	сн=сн,	
Mole % Pur.		ecula mula		Molecular Veight 112.2	208	ch3c2H		
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%			dt/dP °C/mm			f to		
B. P. °C 760 mm 100 30 10 1	107.5 49.40 23.77 4.31 -27.79	2 4 4 5 5	25°C BP te 30 mm  AHm cal/g  AHv cal/g	0.6083 0.0459 0.0363 0.6418	5 4 5 5	h   to g'   c 'K h'   to m   to n   o'K		
mm 25°C	31.97 1025.	5 5	25°C 30 mm BP	80.95 81.12 69.54	5 5 5	m' to		
Density g/ml 20°C d <sup>t</sup> 25 4 30	0.7200 0.7158 0.7116	2 2 4	t <sub>e</sub> (d, e)  ΔHv/T <sub>e</sub> d 20 to	68.10 68.06 19.52	5 5 5	n' o' o'K		
a b	0.7367 -0.0 <sub>3</sub> 83	4	_e_ _130 °C d'  to	0.1382	5	dynes/cm. 20°C 30 40	21.38 20.40 19.43	5 5 5
Ref. Index <sup>n</sup> D 20°C 25 30	1.4097 1.4072 1.4047	2 2 4	e'   °C  d g/ml vc ml/g tc °C	0.245 4.088 273.	5 5 5	Parachor [P] 20°C 30		
"C"	0.7577	4	P <sub>c</sub> mm	19008.	5	40 Sugd.	335.2	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	38.59 38.677 1.0497	2 5 2	PV/RT 25°C 30 mm BP	1.0000 1.0000 0.9500	5 5 5	Exp. L.1.%/wt. u. Dispersion	117.	2
A 20 to B   137°C C	6.94340 1324.7 218.57	5 5 5	te tc AHc kcal/m	0,9409 0,256	5	Flash Point C Fire Point M. Spec.		
A* 20 to B* 137°C K	1.39246	5 5	ΔHf ΔFf Viscosity	!		Ultra V. X-Ray Dif. Infrared Solubility in +		
c t <sub>k</sub> to C A'   to			centistokes γ °C			Acetone Carbon tet. Benzene		
B' i°C_ C' to			B <sup>V</sup> to A <sup>V</sup> C			Ether n-Heptane Ethanol Water Water in		
B'* °C  Ac 137 to Bc tc °C Cc	7.3545 1633. 258.	5 5 5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C c <sub>p</sub> liq. °K			water in		
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	118.26	5	c <sub>w</sub> wap.	L	L		l	
$T_R = 0.79$	<del>X</del>					grams/100 gra		t
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:	'ION ·		PI Dī					
PURIFICAT	TON: RE REFERE		PI					
			••					

No. 144 NAME 2, 3, 3-Trimethyl-1-pentene STRUCTURAL FORMULA CH<sub>3</sub> сн<sub>3</sub>сн<sub>2</sub>с C = CH, Mole Ref. Molecular Molecular CaH16 ĊH<sub>2</sub>ĊH<sub>2</sub> % Pur Formula Weight 112, 208 Ref Ref Ref F.P. °C F.P. 100% -69 2 dt/dP to \*C/mm 25\*C g 0.6276 ٠5 B. P. \*C h BP 0.0459 4 760 mm 108.31 2 0.0363 5 f' to 100 50.12 4 g\* ۰ĸ 30 5 30 mm 0.6429 24.45 4 10 4.96 5 h' ∆Hm cal/g 1 5 -27, 20 to AHv cal/g m Pressure °K 25°C n 81.27 mm 25°C 30.87 5 0 30 mm 81.35 5 t<sub>e</sub> 1027. 5 ВP 69.75 5 Density g/ml 20°C m to te (d, e) 68.30 5 •ĸ 0.7352 2 68.25 o١ 0.7308 2 dt AHv/Te 5 19.53 30 0,7263 4 Surface tension 1 20 84.73 5 0.7529 4 dynes/cm. 20°C 23.25 125 •c 0.1383 5 Ъ -0.0388 4 30 22.14 5 ď١ to 40 21.06 5 Ref. Index e' •c 20°C 1.4174 [P]2 n<sub>D</sub> Parachor d<sub>c</sub> g/ml 0.25 5 25 20°C 1.4151 2 ml/g 4.0 5 30 1.4128 4 c 30 •c 5 ŧč 276. 40 "C" 0.7552 4 Pc 19451. 5 5 mm Sugd. 335.2 MR (Obs.) 38.41 2 PV/RT Exp. L.1. %/wt. MR (Calc.) 38.677 25°C 1.0000 5 (nD-d/2)1.0498 30 mm 1.0000 Dispersion 122. 2 Dielectric RP 0.9500 Flash Point °C 0.9408 A 20 to 6.94586 5 Fire Point tç 0.256 1132 °C 1328.2 M Spec. c 218.42 AHc kcal/m Ultra V ΔHf A\* | 20 to 1.39405 5 X-Ray Dif. ΔFf B\* 130 °C 1247.2 Infrared ĸ Viscosity Viscom, centistokes °C Solubility in Acetone to °C Carbon tet. Benzene A' to Ether В' •c n-Heptane Ċ Ethanol to °C A'\* Water to •C B'\* (BV) Water in to Ac | 139 to 7.3573 (AV) °C Bc \_tc\_ °C 1638. cp liq. °K Cc 5 258. Cryos. Aº ۰ĸ c<sub>p</sub> vap. consts. B° t<sub>e</sub> °C vap. 119.17 5  $T_R = 0.75 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

<u> </u>		<del></del>	<del></del>					No. 14	±0
NAME	2, 3,	4-Trime	thyl-1-pentene			ST	RUCTURAL		A
			—				CH <sub>3</sub> CH CH	C = CH <sub>2</sub>	
Mole	Ref.	Molecu	lar C tr	Molecular			ċн <sub>3</sub> ċн <sub>3</sub>	ĊН <sub>3</sub>	
% Pur.		Formul	a C <sub>8</sub> H <sub>16</sub>	Weight 112.2	.08				
		Ref			Ref.				Ref.
F.P. °C		L_	dt/dP		I	ſ	to		
F.P. 100%	•		*C/mm 25*C	0 (104	١,	g	• <u>K</u>		1
B. P. °C 760 mm	108.	2	BP	0.6194 0.0459	5 4	h			ـــــ
100	49.82		t <sub>e</sub>	0.0363	5	f'	to		1
30	24. 15		30 mm	0.6427	5	g'	' <u>°K</u>		1
10	4.67		∆Hm cal/g			h'	ļ		-
Pressure		<del> </del> _	ΔHv cal/g			m n	to eK		1
mm 25°C	31.35		25°C 30 mm	81.10 81.22	5	0	<u> </u>		1
t <sub>e</sub>	1026.	5	BP	69.63	5	m'			<del> </del>
Density g/ml 20°C		.   .	te (d. s)	68.19	5	n'	to to		1
g/mi 20 C	0.72		t <sub>e</sub> (d, e)	68, 14	5	ە'			
d <sub>4</sub> 25 30	0.72		ΔHv/T <sub>e</sub>	19.52	5	Sur	face tension		+
a	0.74		d 20 to	84.55 0.1381	5		es/cm. 20°C	22.47	5
Ь	-0.03	8 4	-∥d'∣ to	0.1301	_	8	30 40	21.50 20.54	5
Ref. Index		.5 2	e' °C	<b></b>	ļ	B	achor [P]	20.54	+ 5
25	1.41	.3 2	d <sub>c</sub> g/ml	0.249	5	Fai	20°C		İ
30	1.41		vc m1/g tc °C	4.015 276.	5		30		
"C"	0, 75	75 4	P <sub>c</sub> mm	19460.	5	Ĭ	40 Sugd.	335.2	5
MR (Obs.)		. 2	PV/RT		+-	Éxt	L.1.%/wt.		+
MR (Calc. (nD-d/2)	38.67		25°C	1.0000	5	l l	u.		
Dielectric			30 mm BP	1.0000 0.9500	5		persion	122.	2
A 20 to	1	26 5	- t	0.9408	5		sh Point C e Point		
B 139 °C	1326.1	5	_ t <sub>c</sub>	0.256	5		Spec.		1-
c	218.48		ΔHc kcal/m ΔHf				ra V.		
A* 20 to B* 129 °C		111 5	ΔFf				Ray Dif. ared		
K	-		Viscosity			<b>!</b>			+
t,tō	-	l	centistokes		1		ability in T		
t <sub>k</sub> to		i	7 °C				rbon tet.		
A'   to	<b>—</b>				1		nzene her		
B'°	<u>-</u>	ŀ	B <sub>v</sub> to			n-	Heptane		
	+		B to		į.		hanol ater		
A'* to B'* °C			$(B^{V})$ = $\frac{1}{to}$	1	1		ter in		
Ac   139 to		41 5	(A <sup>V</sup> )  °C						
Bc tc °C	1636. - 258.	5	c <sub>p</sub> liq. °K			1			
Cc		-   3	-{I			1			1
Cryos, A° consts, B°			c <sub>p</sub> vap. *K						
te °C	118.83	5	c <sub>w</sub> vap.	ļ					
$T_R = 0.7$	5 T <sub>c</sub>					+ g1	ams/100 gra	ms solve	nt
REFEREN	CES: 1-I	Dow 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ata 5	-Calc. by for	mula	
SOURCE:			API						
PURIFICA'	TION:	1	API						
LITERATU	RE REF	ERENCE	S:						
ļ									

								No. 14	6
NAME	2,4	, 4-T	rimet	hyl-l-pentene			STRUCTURAL	FORMUL	Α.
							CH <sub>3</sub>		
		T -					сн <sub>3</sub> с сн <sub>2</sub> с	- CH	
Mole	Ref	. Mo	lecul	ar C <sub>8</sub> H <sub>16</sub>	Molecular		- 1		
% Pur.		Fo	rmul	a 08116	Weight 112.20	08	CH <sub>3</sub> CH	·3	<del></del>
			Ref.			Ref.			Ref.
F.P. °C F.P. 1007	-93.4	80	2	dt/dP			f to		
	<u>'</u>		-	*C/mm 25°C	0,4858	· <sub>5</sub>	g <u>*K</u> _	i	
B. P. *C 760 mm	101.4	4	2	BP	0.0452	4	h + -		<del> </del>
100	44.1	4	4	t <sub>e</sub>	0.0363	5	f' to		
30 10	18.8		4 5	30 mm	0.6326	5	h' '		
1	-31.9		5	ΔHm cal/g			<del> </del>		+-
Pressure				ΔHv cal/g 25°C	70 75	5	m to		-
mm 25°C	41.1 1007.	0	5	30 mm	78.75 79.60	5		1	1
Denelte	1007.		1 3	BP	68.26	5	m' to		1
Density g/ml 20°C	0.7	150	2	te te (d, e)	66. 93 66. 89	5	n'  °K	1	
<sub>a</sub> t 25	0.7	108	2	ΔHv/T <sub>e</sub>	19.52	5	o'		
		065	4	d   18 to		5	Surface tension		
a b	-0.0	319	4	e   120 °C		5	dynes/cm. 20°C	20.79 19.81	5
Ref. Index		301	+ -	d'   to			40	18.85	5
n <sub>D</sub> 20°C		086	2			5	Parachor [P]		$\top$
25 30		060	2	d g/ml v ml/g	0.244 4.10	5	20°C		
"C"		034	4	tc °C	264.	5	30 40		
		611		P <sub>c</sub> mm	18658.	5		335.2	5
MR (Obs.) MR (Calc.			2 5	PV/RT			Exp. L.1.%/wt.		
(nD-d/2)		511	2	25°C 30 mm	0.9984 1.0000	5	u. Dispersion	122.	2
Dielectric				BP	0.9500	5	Flash Point °C	122.	+-
A 15 to		3714	5	te .	0.9414	5	Fire Point		1
B 1129 °C	219.7		5	t <sub>c</sub>	0.230	-	M Spec.		1
A*   15 to		9284	5	ΔHc Kca1/m		1	Ultra V.		
B* 120 °C			5	ΔFf		<u></u>	X-Ray Dif. Infrared		
к ——	-			Viscosity			Solubility in +	-	†
t <sub>k</sub>	-			centistokes 7 °C		'	Acetone		
t <sub>x</sub> i •c	7			'		1	Carbon tet. Benzene		
A'   to							Ether		
B', L _ º	4			B <sup>V</sup>   to		<u> </u>	n-Heptane Ethanol		
A'* to	;			A <sup>V</sup> C			Water		
B'* °(				(B <sup>V</sup> ) to	7)		Water in	<u> </u>	—
Ac   129 to		486	5	(A <sup>V</sup> ) •C	1			1	
Bc tc_°C	258.		5	c <sub>p</sub> liq. °K					
Cryos, A°	<del></del>		+-						
consts. B			L	c <sub>p</sub> vap. °K	1	:			
te °C	111.4	3	5	c <sub>w</sub> vap.					
$T_R = 0.7$	5 T <sub>c</sub>		1		.1		grams/100 gra	ms solver	nt.
REFEREN		Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:				PI					
PURIFICA'	TION:			PI					
LITERATU	RE REI	ERE							

No. 147 3, 3, 4-Trimethyl-1-pentene NAME STRUCTURAL FORMULA CH<sub>3</sub> сизси с CH=CH, Molecular Mole Ref. Molecular င်မ₃င်မ₃ C8H16 Weight 112.208 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C °K g 0.5543 5 B. P. °C h ВP 0.0456 760 mm 105. 2 t<sub>e</sub> 0.0363 5 ſ١ 47, 24 to 100 4 g' °К 30 21.76 4 30 mm 0.6380 5 10 2.42 5 h! ∆Hm cal/g -29.49 5 to m AHv cal/g Pressure ۰ĸ n 25°C 80.06 5 mm 25°C 35.46 5 o 30 mm 80,50 5 5 1018. te BP 5 69.03 m' to Density te te (d, e) 67.64 5 n' ۰ĸ g/ml 20°C 0.729 2 67.59 5 o' d<sub>4</sub> 25 0.725 2 ΔHv/T 5 19.53 30 0.721 4 Surface tension 15 83,50 5 to 0.745 a 4 dynes/cm. 20°C 22.47 ٠<u>c</u> 0.1378 5 ᇷᅱ 125 ь -0.038 4 30 21.50 to 40 20.55 5 Ref. Index e¹ [P] <sup>n</sup>D 20°C 1.4144 Parachor d<sub>c</sub> g/ml 0.25 5 25 2 1.4120 20°C vc ml/g 3.99 30 4 1.4096 30 t<sub>c</sub> 271. 5 40 "C" 0.7564 4 P<sub>c</sub> mm 19425. 5 Sugd 335.2 5 MR (Obs.) 38.5 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 38.677 5 25°C 1,0000 (nD-d/2) u. 2 1.0499 30 mm 1.0000 Dispersion 5 117. 2 Dielectric ВP 0.9500 5 Flash Point C A 15 to 0.9411 5 6.94133 5 Fire Point 0.256 1315.8 В 135 °C 5 M. Spec. С 5 219. AHc kcal/m Ultra V. AHf A\* 15 to 1.39312 5 X-Ray Dif. ΔFf B\* 120 °C 1235.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. ۰c Benzene A' | to Ether В' ۰c n-Heptane B<sup>v</sup> | C' to Ethanol °C Water A1# to Water in B'\* (B<sup>V</sup>)| °C to Acl 135 to 7.3531 5 (A<sup>V</sup>)| °C Bc tc °C 1624. c<sub>p</sub> liq. ۰ĸ Cc 258 5 Cryos. A° c vap. °K consts. B° c<sub>v</sub> vap. te °C 11,5.44 5  $T_R = 0.75 T_C$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: LITERATURE REFERENCES:

						1	lo. 148
NAME	3, 4, 4	-Trime	thyl-1-pentene			STRUCTURAL FO	RMULA
Γ						CH <sub>3</sub>	
					$\neg$	сн <sub>3</sub> с сн сн=с	CH <sub>2</sub>
Mole % Pur.	Ref. N	Aoleculi Formuli		Molecular Weight 112.2	08	<b>്</b> ċн₃ċн₃	_
70 Pur.		Ref		weight 110.2	Ref		Ref
F, P. *C	Τ	- Kei	14/17	Τ	I Ver	· · · · · · · · · · · · · · · · · · ·	
F.P. 100%	<del> </del>	+	dt/dP *C/mm			f to g *K	
B. P. °C	<b>†</b>		25°C BP	0.534 0.0455	5 4	h	ĺ
760 mm 100	104. 46.36	2 4	t.	0.0363	5	f <sup>1</sup> to	
30	20.94	4	30 mm	0.6365	5	g' <u>•</u> K	ļ
10 1	1.65 -30.18	5	ΔHm cal/g			h'	
Pressure	1	+	ΔHv cal/g			m to	
mm 25°C	36.97	5	25°C 30 mm	79.69 80.24	5 5	<del> </del>	
t <sub>e</sub>	1015.	5	BP	68.81	5	m'   to	
Density g/ml 20°C	0,719	2	te te (d, e)	67.43 67.39	5	n'  •K_	
at 25	0.715	2	ΔHv/T	19.53	5	0'	
	0,711	4	d   15 to		5	Surface tension	
a b	0.735 -0.038	4 4	<u>e   120 °C</u>	0.1377	5		1.26   5 0.31   5
Ref. Index			d' to		1 1		9.38 5
<sup>n</sup> D 20°C	1.412 1.410	2 2	d_g/ml	0,247	5	Parachor [P]	
30	1.408	4	d g/ml v ml/g t °C	4.055	5	20°C	
"C"	0.7628	4	F	268.	5	40	
MR (Obs.)	38.9	2	P <sub>c</sub> mm PV/RT	18987.	-	Sugd. 33 Exp. L.1.%/wt.	5.2 5
MR (Calc.) (nD-d/2)	38.677 1.053	5 2	25°C	1.1161	5	u.	
Dielectric	1	+-	30 mm BP	1.0000 0.9500	5 5	Dispersion 11	7. 2
A   15 to	6,9398	9 5	te	0.9412	5	Flash Point °C Fire Point	
B 1_133 ℃		5	tc	0.256	5	M Spec.	
C A*   15 to	219.24	5 7 5	ΔHc kcal/m ΔHf			Ultra V.	
B* 120 °C	1.3927	'   5	ΔFf			X-Ray Dif. Infrared	İ
K ———			Viscosity			Solubility in +	
t <sub>e</sub> to	1		centistokes 7°C		1 1	Acetone	ł
<u>'x  </u>			•			Carbon tet. Benzene	
A'   to B' °C						Ether n-Heptane	- 1
c, =	1		B <sup>v</sup> to			Ethanol	
A¹* to			A <sup>V</sup> C			Water Water in	
B'* °C	7 2514	+-	(B <sup>V</sup> ) to	1			
Ac   133 to	7.3514 1618.	5	(A <sup>V</sup> )  °C	<b> </b>	$\vdash$	1	
Cc	258.	5	c <sub>p</sub> liq. °K			1	
Cryos, A° consts, B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	114.32	5	c <sub>w</sub> vap.				
$T_{R} = 0.75$		لتـــــــــــــــــــــــــــــــــــــ	<u> </u>	L	L	+ grams/100 grams	anlunt
REFERENC		2-AP	I 3-Lit, 4-0	Calc, from de	t. dat		
SOURCE:		AP					
PURIFICAT	ION:	AP	rI				
LITERATU	RE REFER	ENCES	:				

Mole			19
Mole	TURAL	FORMUL	A
Ref.	сн <sub>2</sub> с = с <sub>2</sub> н <sub>5</sub>		
F. P. °C			Ref
F.P. 100%  B.P. °C 760 mm 117.0 2100 57.63 4 30 31.43 4 10 11.43 5 1 -22.07 5  Pressure mm 25°C 21.47 5 te 1052. 5  Density g/ml 20°C 4 30 0.735 2 d4 30 0.731 4  B.P. °C/mm 25°C 83.98 5 h    AHV cal/g 25°C 83.98 5 BP 71.55 5 te (d,e) 69.94 5 o' AHV/Te 19.51 5  AHV/Te 19.51 5  Surface dynes/c:  a 0.755 4 b -0.038 4  Ref. Index nD 20°C 1.4247 2 25 1.4222 2 30 1.4198 4  "C" 0.7637 4  MR (Obs.) 39.0 MR (Calc.) 38.677 5 (nD-d/2) 1.0552 2  Dielectric  A 31 to 6.95222 5 B 149.°C 1358 9  "C/mm 25°C 0.8733 5 BP 0.0468 4 f'   15   16   17   18   18   19   11   17   18   19   10   10   10   11   15   16   16   17   18   18   19   11   11   12   12   12   12   12   12		T	101
B.P. °C 760 mm 117.0 2 100 57.63 4 30 31.43 4 10 11.43 5 1 -22.07 5  Pressure mm 25°C 21.47 5 1052.  Density g/ml 20°C 0.735 2 d. 25°C 30 mm 83.48 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 5 BP 71.55 6 br>K		1	
To   To   To   To   To   To   To   To	<del>-</del>		
100			├
10	to		
1 -22.07 5 AHm cal/g  Pressure mm 25°C 21.47 5 30 mm 83.48 5 0 0 1 20°C 1052. 5 4 6 69.88 5 1 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	<u>*K</u>		
Pressure mm 25°C   21. 47   5   1052.   5   5   BP   71. 55   5   5   The strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the strength of the s			-
mm 25°C	to		
Density g/ml 20°C   0.739   2 dt 25   0.735   2 dt 30   0.731   4 d	<u>•</u> K	1	ŀ
Density g/ml 20°C   0.739   2 dt 25   0.735   2 d4 30   0.731   4		L	<b>↓</b> —
g/ml 20°C 0.739 2 dt 25 0.735 2 d4 30 0.731 4	to °K		
a     0.731     4       a     0.755     4       b     -0.038     4       Ref. Index     0.1394     5       nD     20°C     1.4247     2       25     1.4222     2     0.0789     5       30     1.4198     4     0.250     5       "C"     0.7637     4     0.290     5       MR (Obs.)     39.0     2     Pv/RT       (nD-d/2)     1.0552     2     25°C     1.0000     5       Oilelectric     BP     0.9500     5       B   149 °C     1358 9     5     Flash Pt       Fire Po	<sub>v</sub> -	1	
a 0.755 4 6 0.1394 5 dynes/ci 8 1.4247 2 1.4227 2 30 1.4198 4 1 2 1.5 1 0.250 5 290. 5 290. 5 290. 5 25 1.0000 5 25°C 1.0552 2 2 30 mm 1.0000 5 25°C 1.0552 2 30 mm 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000 5 25°C 1.0000			-
b -0.038 4 d e 129 C 0.1334 5 d' 15 to 85.95 5 d' 15 to 85.95 5 d' 15 to 85.95 5 2 0.0789 5 Paracho  **Ref. Index nD 20°C 1.4247 2 1.4222 2 2 30 1.4198 4 to ml/s 20052.  **With the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the company of the compa		23.74	5
Ref. Index  nD 20°C 1. 4247 2 25 1. 4222 2 30 1. 4198 4  "C" 0.7637 4  MR (Obs.) 39.0 2 MR (Calc.) 38. 677 5 (nD-d/2) 1.0552 2  Dielectric B 1.149 °C 1.358 9  MR 1	m. 20°C	22.72	5
n <sub>D</sub>   20°C   1.4247   2   1.4222   2   30   1.4198   4   v <sub>c</sub> ml/g   3.996   5   5   290.   5   5	40	21.72	5
30   1.4198   4   vc ml/g   3.996   5   290.   5			
MR (Obs.)   39.0   2   MR (Calc.)   38.677   5     5     C   1.0000   5     C   1.0552   2     C   30 mm     1.0000   5       C   1.0552   2     C   1.0000   5     C   1.0000   5     C   C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5   C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5     C   1.0000   5	20°C <b>30</b>		
MR (Obs.) 39.0 2 PV/RT 25°C 1.0000 5 Dispers:  Dielectric BP 0.9500 5 Flash Pt C 1.358 9 5 tc 0.256 5 Fire Poi	40	1	
MR (Calc.) 38.677 5 25°C 1.0000 5 Dispers  Dielectric BP 0.9500 5 Flash Pt  te 0.9401 5 Fire Poi  B 144 °C 1358 9 5 tc 0.256 5	Sugd.	335.2	5
(nD-d/2) 1.0552 2 25°C 1.0000 5 Dispers:  Dielectric BP 0.9500 5 C 1.0000 5 Dispers:  A 31 to 6.95222 5 te 0.9401 5 Flash Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Potential Pote			
Dielectric         BP         0.9500 5         5         Flash Post           A 31 to 6.95222 5         te 0.9401 5         Flash Post Post Post Post Post Post Post Post	u. ion	127.	2
A 31 to 6.95222 5 te 0.9401 5 Fire Poi		121.	+-
B   149 °C   1358 9   5   °C   0.256   5			
C 216 77 5 AHC heal/m M. Spec		-	$\vdash$
AHC REAL/III Ultra V.			
A*  31 to			1
K Viscosity		ļ	+
c centistokes Solubilit			
tk Carbon	tet.		
t <sub>x</sub>   Benzen   A'   15 to   7.34929   5   Ether	e		1
D1 21 00 115/2 2   5	ane		
C' 234.77 5 B to Ethano			1
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	in		
D 31 0 1403.5 5 (B // 10			1
Ac  149 to 7.3633 5 (A <sup>V</sup> )  °C   Be  t <sub>c</sub> °C   1677. 5 c lig °K			1
$\begin{bmatrix} \frac{BC}{Cc} & \frac{C}{c} & \frac{C}{257} & \frac{1617}{257} & \frac{5}{5} & \frac{C}{p} & \text{liq.} & \text{°K} \end{bmatrix}$			
Cryos. A c c vap. °K			
t <sub>e</sub> °C 128.99 5 c <sub>w</sub> vap.		1	
	/100 are	ms solver	
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc			
SOURCE: API	,		
PURIFICATION: API			
LITERATURE REFERENCES:			

						<u></u>	No. 15	0
NAME	3-Ethyl-4	-me	thyl-cis-2-pente	ene		STRUCTURAL I	FORMULA	4
						CH <sub>3</sub>		
			Ι.			CH3CH C = C	CHCH <sub>3</sub>	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	08	C <sub>2</sub> H <sub>5</sub>		
		Ref.			Ref			Ref.
F.P. °C			dt/dP			f   to		
F.P. 1007	·		*C/mm 25*C	0.8401	5	g <u>°K</u>		ł
B. P. °C 760 mm	116.	2	BP	0.0467	4	h		
100	56.76	4	t.	0.0363	5	f' to		l
30 10	30.61 10.66	4 5	30 mm	0.6548	5	h'		
1	-22,77	5	ΔHm cal/g		$\vdash$	m   to		<del> </del>
Pressure mm 25°C	22.40	5	ΔHv cal/g 25°C	83.66	5	n 'K_		
t <sub>e</sub>	1049.	5	30 mm BP	83.22 71.33	5 5	° ;		L
Density			t <sub>e</sub>	69.74	5	m'   to		
g/ml 20°C	0.739 0.735	2 2	'e (a, e)	69.68	5	0,		
d <sub>4</sub> 25	0.731	4	ΔHv/T <sub>e</sub>	19.51	5	Surface tension		<del>                                     </del>
a b	0.755	4	d 31 to	87.48 0.1392	5 5	dynes/cm. 20°C	23.74	5
Ref. Index	-0.038	-	d' 15 to 31 °C	85.60	5	30 40	22.72 21.72	5
n <sub>D</sub> 20°C	1.424	2		0.0779	5	Parachor [P]		
30	1.422 1.419	2	d g/ml v ml/g	3.995	5	2 <b>0°C</b> 30		
"C"	0.7625	4	*c C	289.	5	40		_
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	20021.	5		335.2	5
MR (Calc. (nD-d/2)	) 38.677 1.054	5	25°C	1.0000	5	Exp. L.1.%/wt. u.		
Dielectric	1.051	-	30 mm BP	1.0000 0.9500	5	Dispersion	124.	2
A   31 to	6,95084	5	t <sub>e</sub>	0.9401	5	Flash Point °C Fire Point		
B 1148°9	217.	5	t <sub>c</sub> ΔHc kcal/m	0.256	5	M Spec.		$t^{-}$
A* 31 to		5	ΔHf			Ultra V.		
B* 138°C	1272.8	5	ΔFf			X-Ray Dif. Infrared		ĺ
K C			Viscosity centistokes			Solubility in +		
t <sub>k</sub>   to		١.	η •c			Acetone Carbon tet.		
A' 15 to		5				Benzene Ether		
B'   31 °C	1559.4	5	B <sup>V</sup>   to			n-Heptane		
A'* 15 to	235.	5	B <sup>V</sup>   to A <sup>V</sup>   °C			Ethanol Water		
A'* 15 to B'* 31 °C		5 5	(B <sup>V</sup> ) to			Water in		
Ac   148 to	7.3620	5	(A <sup>V</sup> ) °C					
Bc tc_°C	257.	5	c <sub>p</sub> liq. °K					
Cryos, A		Ť	c <sub>p</sub> vap. °K					
consts. B°			<u> </u>					
t <sub>e</sub> °C	127.86	5	c <sub>v</sub> vap.					
T <sub>R</sub> = 0.7		2 :-				f grams/100 gran		t
	CES: 1-Dow	2-AI		alc. from de	t. da	ta 5-Calc. by for	nula	
SOURCE:	TION:		PI					
	RE REFERE							
		.010	••					
1								
1								
L								

							No. 151
NAME	3-Ethyl-	4-me	thyl-trans-2-pe	entene		STRUCTURAL	FORMULA
Mole % Pur.	Ref. Mol	ecul		Molecular Veight 112,20	08	сн <sub>3</sub> сн <sub>3</sub> сн с = с <sub>2</sub> н	снсн <sub>3</sub>
	1	Ref.			Ref.		Ref
F. P. °C F. P. 100%			dt/dP °C/mm 25°C	0.7881	5	f to	
B. P. °C 760 mm 100 30 10	114.3 55.31 29.27 9.40	2 4 4 5	BP t <sub>e</sub> 30 mm ΔHm cal/g	0.0466 0.0363 0.6520	4 5 5	f' to g' °K	
Pressure mm 25°C t <sub>e</sub>	24.03 1044.	5 5 5	ΔHv cal/g 25°C 30 mm BP	83.16 82.83 71.00	5 5 5	m to	
Density g/ml 20°C dt 25 d <sub>4</sub> 30	0.7350 0.7308 0.7266	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	69. 44 69. 38 19. 52	5 5	m' to	
a b	0.7518 -0.0 <sub>3</sub> 83	4	d 29 to e 126 °C d' 15 to	86.90 0.1391 85.05	5 5 5	Surface tension dynes/cm. 20°C 30 40	23.23 5 22.17 5 21.15 5
Ref. Index n <sub>D</sub> 20°C 25 30	1.4210 1.4183 1.4158	2 2 4	e'   29 °C d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.0760 0.248 4.028 285.	5 5 5 5	Parachor [P] 20°C 30	
"C"	0.7615	4	tc°C Pcmm	19717.	5	40 <b>Sug</b> d.	335.2 5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	38.71 38.677 1.0535	2 5 2	PV/RT 25°C 30 mm	1.0000 1.0000	5	Exp. L.1.%/wt. u. Dispersion	124. 2
A 29 to B 1146 °C	6.95117 1349.7 217.28	5 5 5	BP t e t c AHc kcal/m	0.9500 0.9403 0.256	5 5 5	Flash Point C Fire Point M. Spec.	
A* 29 to B* 136 °C K	1.39291 1267.6	5	ΔHf ΔFf Viscosity			Ultra V. X-Ray Dif. Infrared	
t <sub>k</sub> to			centistokes 7°C			Solubility in Acetone Carbon tet. Benzene	
A'   15 to B'   29 °C C'	7, 35081 1553, 9 235, 28	5 5 5	B <sup>V</sup>			Ether n-Heptane Ethanol Water	
A'* 15 to B'* 29 °C Ac 146 to	1.74885 1454.3	5	(B <sup>V</sup> )  to			Water in	
Bc t <sub>c</sub> °C	7.3622 1665. 257.	5 5 5	c <sub>p</sub> liq. °K				
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	125.93 T	5	c vap.			+ « • • • • • • • • • • • • • • • • • •	
T <sub>R</sub> = 0.75	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	grams/100 gra ta 5-Calc. by for	
SOURCE:			PI		ud	3-021c. by 101	
PURIFICAT	ION:		.PI				
LITERATUR	RE REFERE	NCES	5:				

							No. 152	2
NAME	2, 3, 4-Tri	meth	yl-2-pentene			STRUCTURAL I	FORMULA	L
Ī						сн <sub>3</sub> сн с = 0	C + CH -	
			T		$\neg$	Ċн <sub>3</sub> ċн <sub>3</sub> ċн <sub>3</sub>	он.	
Mole % Pur.		lecul rmul		Molecular Weight 112.2	08	303	3	
		Ref.			Ref	T		Ref.
F.P. *C	-113.3	2	dt/dP	1		f to		
F.P. 100%			°C/mm	0.0400	ا ۔ ا	g <u>*K</u>		
B. P. °C 760 mm	116.26	2	25°C BP	0.8499 0.0467	5 4	h		
100	57.02	4	t.	0.0363	5	f' to		
30 10	30.86	5	30 mm	0.6550	5	g'   'K_ h'		
1	-22.54	5	ΔHm cal/g			<del>                                     </del>		
Pressure			ΔHv cal/g 25°C	83.79	5	m to		
mm 25°C	22.11 1050.	5	30 mm	83.33	5	•		ĺ
Density		i -	BP	71.43 69.83	5	m¹   tổ		
g/ml 20°C		2	te (d, e)	69.77	5	n'  •K_		
dt 25 4 30	0.7391 0.7348	2 4	AHv/T	19.52	5	0'		
	0.7606	4	d   31 to		5	Surface tension dynes/cm. 20°C	24, 31	5
ъ	-0.0385	4	a, - 135 °C		5	30	23.19	5
Ref. Index			e' 31 °C		5	40	22.11	5
<sup>n</sup> D 20°C	1.4275	2	d g/ml	0.250	5	Parachor [P] 20°C		
30	1.4223	4	tc *C	3.997	5	30		İ
"C"	0.7639	4	P <sub>c</sub> mm	20011.	5	40 Sugd	335.2	5
MR (Obs.) MR (Calc.		2	PV/RT			Exp. L.1.%/wt.	333.2	Ť
(nD-d/2)	38.677 1.0558	5 2	25°C 30 mm	1.0000	5	u.		
Dielectric			BP BP	1.0000 0.9500	5	Dispersion	127.	2
A 31 to		5	t <sub>e</sub>	0.9401 0.256	5	Flash Point °C Fire Point		•
B 1.149.℃	1356.9 217.	5	t <sub>c</sub>	0.230	<del>                                     </del>	M Spec.		
A*   31 to	+	5	ΔHf		1 1	Ultra V. X-Ray Dif.		Ì
B* 138 °C		5	ΔFf	ļ	1	Infrared		1
K c			Viscosity centistokes			Solubility in +		
tk to			η •c	}		Acetone Carbon tet.		]
t <sub>x  </sub> *C		_				Benzene		1
B' _ 31 °C		5			$oxed{oxed}$	Ether n-Heptane		l
C'	235.	5	B <sup>V</sup>   to			Ethanol		
A'* 15 to B'* 31 °C		5 5		-		Water Water in		
Ac   149 to	7.3646	5	1	1				
Bc t °C	1674.	5	c <sub>p</sub> liq. °K	<del></del>	-			
Cc	257.	5	1 -					
Cryos. A° consts. B°	1		c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	128.15	5	c <sub>v</sub> vap.					
$T_R = 0.7$			4		·	+ grams/100 gran	ns solven	 t
	CES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for		
SOURCE:		AI	PI					
PURIFICAT		AF						
LITERATU	RE REFERE	VC ES	<b>3:</b>					

Гт							No. 153
NAME	2, 4, 4-Tz	imet	thyl-2-pentene			STRUCTURAL 1	<b>FORMULA</b>
] [						CH <sub>3</sub>	
						сн3с сн=с	CH.
Mole	Ref. Mo	lecul	ar c. 1	Molecular			
% Pur.		mul		Weight 112.20	8	CH <sub>3</sub> C	Н <sub>3</sub>
		Ref.			Ref.		Ref
F.P. °C	-106.330	2	dt/dP			f to	
F.P. 1009			°C/mm	]		f to to	j
B. P. °C			25°C	0.5523	5	h	1
760 mm	104.91	2	BP	0.0456	5	<del></del>	
100	47.15	4	t <sub>e</sub>	0.0363		g' to to	
30 10	21.67	5	30 mm	0.6379	5_	h'	1
i	-29.57	5	ΔHm cal/g			<del>}</del>	
Pressure			ΔHv cal/g		1	m to	1
mm 25°C	35.62	5	25°C 30 mm	80.01	5	<del>-</del>	j
te	1017.	5	BP	80.46 68.99	5 5		
Density	_		t_	67.60	5	m' to	}
g/ml 20°0		2	t <sub>e</sub> (d, e)	67.56	5		
dt 25	0.7176 0.7133	2 4	ΔHv/T <sub>e</sub>	19.52	5		
a		4	d 15 to	83.45	5	Surface tension	21 60 5
b	0.7387 -0.0 <sub>3</sub> 84	4	_e120 °C	0.1378	5	dynes/cm. 20°C	21.60   5   5
Ref. Index	<del></del>	Ė	d' to	]		40	19.60 5
n <sub>D</sub> 20°C		2		0.515	+-	Parachor [P]	
25	1.4135	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	0.245 4.073	5	20°C	
30	1.4110	4	tc *C *	270.	5	30	
"C"	0.7668	4	P <sub>c</sub> mm	18976.	5	40 Sugd. 3	335.2 5
MR (Obs.)		2	PV/RT	-	<del>  </del>	Exp. L.1.%/wt.	
MR (Calc. (nD-d/2)	) 38.677 1.0551	5 2	25°C	1.0000	5	u.	
		۲	30 mm	1.0000	5		125. 2
Dielectric		ļ	BP	0.9500 0.9411	5	Flash Point C	
A 15 to B   134 °C		5	te t <sub>C</sub>	0.256	5	Fire Point	
č 🗀 🚾 🗅	219.07	5	ΔHc kcal/m	-	_	M. Spec.	
A* 15 to		5	ΔHf	i		Ultra V. X-Ray Dif.	
B* 120 °C		5	ΔFf		<u> </u>	Infrared	
к — — -			Viscosity			Solubility in +	
t <sub>k</sub> – to	-		centistokes 7 °C	l	İ	Acetone	-
k x			7		1	Carbon tet.	
A' to		-				Benzene• Ether	
B' °C			<del></del>		├	n-Heptane	
C'		L	B <sup>V</sup> to C			Ethanol	1
A'* to					1	Water Water in	[
B'* °C	<del></del>	<b>—</b>	(B <sup>V</sup> )  to	1			
Ac 134 to		5	(A <sup>V</sup> )  °C		<u></u>	1	i
Bc tc C	1622.	5	c <sub>p</sub> liq. °K	1			
Cryos, A		<u> </u>	f.	1			1
consts. B			c <sub>p</sub> vap. °K	1	1		1
t <sub>e</sub> °C	115.34	5	c <sub>v</sub> vap.				
$T_R = 0.7$	5 T <sub>c</sub>			•		grams/100 gran	ns solvent
	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by form	
SOURCE:			PI			<del></del>	
PURIFICA	TION:		PI				
	JRE REFERE	NCES	5:				
			· · · · · · · · · · · · · · · · · · ·				

						No.	154
NAME	3, 4, 4	-Trimeth	yl-cis-2-pente	ne		STRUCTURAL FORM	
			,			CH <sub>3</sub>	
1_						сн, с с = снсн	•
Mole	Ref.	Molecul		Molecular		ĊH3ĊH3	3
% Pur.	لييلي		a 98-16	Weight 112.	$\overline{}$	3 3	5.
	T	Ref.			Ref		Ref.
F.P. °C F.P. 100%	<del> </del>		dt/dP *C/mm			f to to	1
B. P. °C	<del>                                     </del>		25°C	0.7214	5	g   <u>*K</u>	ı
76 <b>0 mm</b>	112.	2	BP	0.0463	4 5		
100 30	53.31 27.41		t <sub>e</sub> 30 mm	0.6486	5	f' to g' 'K	- 1
10	7.64	5	ΔHm cal/g	0.0400	ť	h'	
1	-25.47	5	ΔHin cal/g		-	m to	
Pressure mm 25°C	26,48	5	25°C	82.58	5	n •K	
t <sub>e</sub>	1038.	5	30 mm	82.25	5	0	
Density			BP te	70.51	5	m'   to	- 1
g/ml 20°C	0.73		te (a, e)	68.94	5	n'   °K	- 1
dt 25 4 30	0.73	- 1 - 1	AHv/T <sub>e</sub>	19.52	5		
<b>a</b>	0.75	5 4	d   20 to	86.05	5	Surface tension dynes/cm. 20°C 23.7	4 5
ь	-0.03	8 4	_e,		5	30 22.7	
Ref. Index	1.42	3 2	e'			40 21.7	2 5
<sup>n</sup> D 20°C	1.42		d g/ml	0.252	5	Parachor [P] 20°C	ì
30	1.41	8 4	v <sub>c</sub> ml/g t <sub>c</sub> °C	3.969	5	30	
"C"	0.76		P <sub>c</sub> mm	19940.	5	40 Sugd. 335.2	. 5
MR (Obs.) MR (Calc.)	38.7 38.67	7   2	PV/RT		<b>-</b>	Exp. L.1.%/wt.	
(nD-d/2)	1.05		25°C 30 mm	1.0000	5	u.	2
Dielectric			BP	0.9500	5	Dispersion 124. Flash Point °C	
A 20 to			t <sub>e</sub>	0.9405 0.256	5	Fire Point	İ
B <u>  144 °C</u> C	1341.1	5 5	t <sub>c</sub> AHc kcal/m	0,230	-	M Spec.	
A*  20 to	1.39	+-	ΔHf	j		Ultra V. X-Ray Dif.	
B* 133 ℃		5	ΔFf		_	Infrared	ľ
K — — —			Viscosity centistokes			Solubility in +	
t <sub>k</sub>   to	1		η °c			Acetone Carbon tet.	
t c						Benzene	
A'   to B'   *C						Ether n-Heptane	- 1
c,			B <sup>V</sup>   to	Ţ.		Ethanol	
A'* to B'* °C			$\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})} - \frac{{}^{\mathbf{c}}C}{to}$	-[		Water Water in	
Ac   144 to		98 5					
Bc t °C	1655.	5			$\vdash$		
Cc	258.	5	c <sub>p</sub> liq. °K				1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	123.34	5	c <sub>v</sub> vap.				
$T_R = 0.7$						+ grams/100 grams sol	vent
REFERENC	ES: 1-De	ow 2-AF	PI 3-Lit, 4-0	Calc. from de	t. da	ta 5-Calc, by formula	
SOURCE:	···-	AF					
PURIFICAT		AF					
LITERATU	RE REFE	ERENCES	:				

							No. 1	
NAME	3,4,	4-Trime	thyl-trans-2-pe	ntene		STRUCTURAL	FORMUL	٠A
1					1	CH <sub>3</sub>		
						CH <sub>3</sub> C C = C	CH CH <sub>3</sub>	
Mole	Ref.			Molecular	.	сн,сн,	•	
% Pur.		Formu		Weight 112.2	-	3 3 T		D - 6
	<del></del>	Re		<del> </del>	Ref.	<del>-                                    </del>		Ref
F.P. °C F.P. 100%			dt/dP °C/mm			f to		
B. P. °C	<b>,</b>		25°C	0.7214	5	g <u>*K</u>		
760 mm	112.	2	BP	0.0463	4	h		-
100	53.31	4	t <sub>e</sub>	0.0363	5	f' to c'K		
30 10	27.41 7.64		30 mm	0.6486	5_			1
1 1	-25.47		ΔHm cal/g			h'		+-
Pressure			AHv cal/g		!	m to		
mm 25°C	26.48		25°C 30 mm	82.58 82.25	5	<del></del>		
t <sub>e</sub>	1038.	5	BP	70.51	5	- <u></u>		+
Density	,   ,	,   -	t <sub>e</sub> ,	68.99	5	m' to		1
g/ml 20°C	0.73		t <sub>e</sub> (d, e)	68.94	5	ö'		
d <sub>4</sub> 30	0.73		ΔHv/T <sub>e</sub>	19.52	5	Sunfa an dan sida		+-
a	0.75	5 4	d 20 to	86.05	5	Surface tension dynes/cm. 20°C	23,74	5
ь	-0.03	8 4	e123 °C to	0.1387	5	<b>8</b> 30	22.71	5
Ref. Index		_   _	e' °C			40	21.72	5
<sup>n</sup> D 20°C	1		d <sub>c</sub> g/ml	0, 252	5	Parachor [P] 20°C		
30	1.42		V mi/g	3, 969	5	30		
"C"	0.76		11 _	283.	5	40		1
MR (Obs.)		2	P <sub>c</sub> mm	19940.	5	Sugd.	335.2	5
MR (Calc.			PV/RT 25°C	1 0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.05	4 2	30 mm	1.0000	5	u. Dispersion	124.	2
Dielectric			BP	0.9500	5	Flash Point °C		+
A 20 to			t <sub>e</sub>	0.9405 0.256	5	Fire Point		
B (144 °C	- 1341.1 217.72	5 5	t <sub>c</sub> ΔHc kcal/m	+	-	M. Spec.		
A* 20 to		+	ΔHf	1		Ultra V. X-Ray Dif.		İ
B*[133 °C		5	ΔFf		1	Infrared		1
к			Viscosity			Solubility in +		T
t <sub>k</sub> — tō	-		7 °C			Acetone		
t <sub>x</sub> °C	:		'			Carbon tet. Benzene		
A'   to			1			Ether		
B'  °	-		B <sub>v</sub> to		T	n-Heptane		
	+		A C			Ethanol Water		
A'* to			$\frac{ \mathbf{B}^{\mathbf{v}} }{ \mathbf{B}^{\mathbf{v}} } = \frac{1}{to}$	-		Water in		
Ac  144 to	7.35	98 5	(A <sup>V</sup> )  °C					
Bc tc °C	1655.	5		<del> </del>				
Cc	258.	5	_  P -					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
		<del>-   -</del>	c, vap.	1				
t <sub>e</sub> °C	123.34	5		1		+ /		
$T_R = 0.7$				<u> </u>		grams/100 gram		nt
REFEREN	UES: 1-I			Calc. from de	et. da	ita 5-Calc. by for	mula	
SOURCE:			API					
PURIFICA			API					
LITERATU	RE REF	ERENCE	cs:					

-							No. 156	5
NAME	2-Isopro	oyl - 3	-methyl-1-bute	ne		STRUCTURAL 1	ORMULA	
					$\neg$	CH3		
			T			сн3сн с	CH,	
Mole		lecul		Molecular Weight 112.2	ا يمر	сн(сі	H <sub>3</sub> ),	
% Pur.	1 1 1 1	rmul Ref.		weight 112.2	Ref			Ref.
F.P. *C	T	Rei.		1	Kei.			Kei.
F.P. 1009	<u></u>	<del>                                     </del>	dt/dP *C/mm			f to		
B. P. *C		$\vdash$	25°C	0.5340	5	h .		
760 mm 100	104.	2	BP t <sub>e</sub>	0.0455 0.0363	<b>4</b> 5	f' to		
30	46.36 20.94	4	90 mm	0.6365	5	g'   'K_		
10 1	1.65	5 5	ΔHm cal/g			h'		
Pressure	-30.18	13	ΔHv cal/g			m to		
mm 25°C		5	25°C 30 mm	79.69 80.24	5	n •K		
t <sub>e</sub>	1015.	5	BP	68.81	5			-
Density g/ml 20°0	0,722	2	te (d.e)	67.44 67.39	5	m'   to		
at 25	0.718	2	'e ('', '')	1	5	o'		
<sup>d</sup> 4 30	0.714	4	ΔHv/T <sub>e</sub>	19.53 83,13	5	Surface tension		
a b	0.738	4	d   15 to		5	dynes/cm. 20°C	21.62	5
Ref. Index	-0.038	4	d' to			30 40	20.66 19.72	5 5
n <sub>D</sub> 20°0	1.4085	2	1	0,247	5	Parachor [P]		_
25 30	1.4061 1.4037	2	d g/ml vc ml/g	4.04	5	20°C		
"C"	0,7535	4	tc °C	269.	5	30 40		
MR (Obs.		2	P <sub>c</sub> mm	19096.	5	Sugd.	335.2	5
MR (Calc.		5	PV/RT 25°C	1 0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0475	2	30 mm	1.0000 1.0000	5	u. Dispersion	122.	2
Dielectric			BP	0.9500	5	Flash Point °C		$\vdash$
A 15 t B <u>1133</u> •		5	t <sub>e</sub> t <sub>c</sub>	0.9412 0.256	5	Fire Point		
c L	219.24	5	ΔHc kcal/m			M Spec. Ultra V.		
A*  15 to		5	ΔHf ΔFf			X-Ray Dif.	•	1
B* 125 °C	2 1231.9	5	Viscosity		<del> </del>	Infrared		<u> </u>
° .—	_		centistokes	1	İ	Solubility in +		
t <sub>k</sub>   to			η ∘c		1	Carbon tet.		
A' to		$\vdash$		1		Benzene Ether		İ
B'	2		B <sup>V</sup>   to		<b>├</b> ─	n-Heptane		İ
C'			B to			Ethanol Water		
A'* to B'* *			(B <sup>V</sup> ) - to	1		Water in		
Ac   133 to		5	(A <sup>V</sup> ) °C	1				
Bc tc_		5	c <sub>p</sub> liq. °K		<b>†</b>	1		
Cryos. A	258.	+ -	11					
consts. B			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	114, 32	5	c <sub>v</sub> vap.					
$T_R = 0.7$	75 T <sub>c</sub>	<b>-</b>	и	·	L	grams/100 gran	ns solveni	
	CES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from de	t. da	ta 5-Calc. by for		
SOURCE:		AF						
PURIFICA	TION:	AF	PI					
LITERATU	RE REFERE	NCES	S:	·				
L								

NAME	2-Ethyl-3	, 3-d	imethyl-l-buten	e		STRUCTURAL	FORMUL	A
	2-tert-Bu	tyl-l	-butene			СH <sub>3</sub>		
Mole % Pur.	Ref. Mo	lecul rmula		Molecular Veight 112,20	08	сн <sub>3</sub> с с = сн <sub>3</sub> с <sub>2</sub> н		
		Ref.	· · · · · · · · · · · · · · · · · · ·	l I	Ref.			Ref.
F. P. °C			dt/dP			f to		
F.P. 1009	6		°C/mm			g  K		
B. P. ℃		$\vdash$	25°C	0.6682	5	h		
760 mm	110.	2	BP	0.0461 0.0363	5	f' to		$\vdash$
100 30	51.56 25.78	4 4	t <sub>e</sub>		5	g'°K		
10	6.10	5	30 mm	0.6456	⊢°⊣	h'		
1	-26.86	5	ΔHm cal/g		-	m   to		t-
Pressure			ΔHv cal/g 25°C	81.84	5	n eK		
mm 25°C		5	30 mm	81.73	5	•		
t <sub>e</sub>	1032.	5	BP	70.07	5	m' to		t
Density g/ml 20°0	0,728	2	t <sub>e</sub> (d, e)	68.59 68.53	5	n'   °K		
dt 25 4 30	0.724	2	te (4, 6)	į.	1 1	o'		
<sup>4</sup> 30	0.72 <b>0</b>	4	ΔHv/T <sub>e</sub>	19.52	5	Surface tension		
a	0.744	4	d 20 to	85.30 0.1385	5	dynes/cm. 20°C	22.35	5
ь	-0.038	4	1-a-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1	0.1303		<b>8</b> 30	21.37	5
Ref. Index	-		e'   °C			40	20.42	5
<sup>n</sup> D 20°0	1.4159 1.4135	2 2	d <sub>c</sub> g/ml	0.248	5	Parachor [P] 20°C		
30	1.4111	4	V_ml/g	4.035 279.	5	30		
"C"	0.7601	4	1. ~	19472.	5	40 Suad	335.2	5
MR (Obs.	38.7	2	P <sub>c</sub> mm	19472.	•		335. 4	+-
MR (Calc.		5	25°C	1,0000	5	Exp. L.1.%/wt. u.		İ
(nD-d/2)	1.0519	2	30 mm	1.0000	5	Dispersion	122.	2
Dielectric			BP	0.9500	5	Flash Point °C		
A 20 to B 141 °C		5	te tc	0.9407 0.256	5	Fire Point		
č '	218.	5	ΔHc kcal/m			M. Spec.		
A* 20 to	1,39179	5	ΔHf		1	Ultra V. X-Ray Dif.		
B*[131°C	1252.4	5	ΔFf		-	Infrared		
K			Viscosity centistokes		'	Solubility in +		
t <sub>k</sub>  to			η °c			Acetone Carbon tet.		
t <sub>x</sub>   °C	_1					Benzene		
A' to						Ether		
B'°	-	1 1	B <sup>V</sup> to A <sup>V</sup> C			n-Heptane Ethanol		
A'* to		$\vdash$	Ā <sup>V</sup> I °C	1		Water		1
B'* °C			(B <sup>V</sup> )  to			Water in		<u> </u>
Ac 141 to		5	(A <sup>V</sup> )  °C					
Bc tc C	1645.	5	c liq. °K					1
	258.	3						
Cryos, Acconsts, B			c <sub>p</sub> vap. *K	1				
te °C	121.08	5	c <sub>v</sub> vap.					
$T_R = 0.7$	75. <b>T</b> c	•		•	<b>-</b>	grams/100 gra	ms solver	nt
	CES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			PI					
PURIFICA	TION:	A	PI					
	JRE REFERE	NCES	5:					
1								

			_					No.158	
NAME	l-Nor	nene					STRUCTURAL	FORMULA	١.
L							сн <sub>3</sub> (сн <sub>2</sub> ) <sub>6</sub> с⊦	H=CH,	
Mole	Ref.	Mole	ecul	C9H18	Molecular Weight 126.	224	3 20	-	
% Pur.		-	mul Ref	7 10 1	Weight 126.	Ref			Ref.
F.P. °C	-81.37		2	dt/dP	1	Kei	f to	1	101
F.P. 100%				°C/mm		_	f to g *K		
B. P. *C	146 06	. 1		25°C BP	3.018 0.04944	5 4	h i		
760 mm 100	146.86 84.21		2	te	0.0362	5	f		
30 10	56.55 35.44	1	2 2	30 mm	0.6928	5	g' °K		
i	1.1		5	ΔHm cal/g			h'	0.0045	<del> </del>
Pressure				ΔHv cal/g 25°C	86.83	5	m   300 to n   600 °K	0.0247 0.0013	4
mm 25°C	5.33 1132.	9	5	30 mm	82.32	5	0	-0.0 <sub>6</sub> 49	4
Density	<del>                                     </del>			BP te	69.58 67.48	5 5	m'   700 to	0.1080	
g/ml 20°C	0.72		2 2	t <sub>e</sub> (d, e)	67.36	5	n'   1000 °K	0.0011 -0.0 <sub>6</sub> 38	
d <sub>4</sub> 30	0.72		4	ΔHv/T <sub>e</sub>	19.55	5	Surface Accessor	0.0650	<u>ٻ</u>
a	0.74		4	d   25 to		5 5	Surface tension dynes/cm. 20°C	21.83	5
ь	-0.03	780	4		'		30 40	20.91	5
Ref. Index n <sub>D</sub> 20°C	1.41	572	2	e' ; °C	<del></del>	-	Parachor [P]		H
25 30	1.41		2 4	d g/ml vc ml/g	0,25 4.0	5 5	20°C	'	Ì
"C"	0.75	+	4	T <sub>C</sub> T	321.	5	30 40		
MR (Obs.)	43.41		2	P <sub>c</sub> mm	18000.	5		374.2	5
MR (Calc.)	43.29	5	5	PV/RT 25°C	1.0000	5	Exp. L.1.%/wt.	'	
(nD-d/2) Dielectric	1.05	111	2	30 mm	1.0000	5	Dispersion	114.4	2
A 25 to	6,95	387	-2	BP t <sub>e</sub>	0.9463 0.9338	5	Flash Point °C		
B <u>[173</u> °C	1435.4	- 1	2	t <sub>c</sub>	0.245	5	Fire Point M Spec.		┢─
C 25 4	205.53		2	ΔHc kcal/m ΔHf	1341.90	2	Ultra V.		
A*   25 to B* <sub>1</sub> 165 °C		103	5	ΔFf	<u></u> _		X-Ray Dif. Infrared		
K — — —	1			Viscosity centistokes			Solubility in +		m
t <sub>k</sub>  to	1			7 40 °C	0.688	2	Acetone Carbon tet.		
t <sub>x</sub> i °C	ļ <u> </u>			60 80	0.568 0.488	2 2	Benzene		
A' to B' °C	İ			100	0.426	2	Ether n-Heptane		ŀ
c'				B <sup>V</sup>   30 to A <sup>V</sup>   70 °C	434.21 2.4510	4	Ethanol		ļ
A'* to B'* °C				<u> </u>	-1	4	Water Water in	1	į
Ac   173 to	7,17	08	5	(B <sup>V</sup> ) 70 to	388.81 2.58744	4			
	1518. 211.		5	c <sub>p</sub> liq. °K		H			
Cryos. A°	<del> </del>	$\dashv$		c <sub>p</sub> vap.300°K	0.38231	2		1	
consts. B°				- 400	0.48204	2			
t <sub>e</sub> °C	162.52		5	c <sub>v</sub> vap.					
T <sub>R</sub> = 0.75							f grams/100 gran		<u>t</u>
REFERENC	£S: 1-D	ow 2	-AF		Calc. from det	da	ta 5-Calc. by for	mula	
SOURCE: PURIFICAT	ION:		AP						
LITERATU		EREN							
				•					

No. 159 1-Decene NAME STRUCTURAL FORMULA CH3(CH2)7CH=CH2 Molecular C<sub>10</sub>H<sub>20</sub> Mole Ref Molecular Weight 140,260 % Pur Ref. Ref. Ref. F. P. °C -66:310 2 dt/dP f to F.P. 100% °C/mm ۰ĸ g 25°C 8.778 5 B.P. °C h BP 0.05157 2 760 mm 170,570 2 5 f† ž 0.03615 to 105.198 100 ۰ĸ g' 30 76.33 2 30 mm 0.7231 5 10 54.29 2 h' ∆Hm cal/g 18.2 5 1 0.0246 300 m AHv cal/g Pressure 600 0.0013 n ۰ĸ 25°C 86.70 5 mm 25°C 1.632 5 ٥ -0.0649 4 30 mm 79.78 5 1197. te 5 BP 66.84 5 m 700 0.1096 Density to 4 te te (d, e) 64.46 5 n' 1000 °K 0.0011 g/ml 20°C 0.74081 2 5 64.27 ۰' -0.0,38 4 dt 4 25 0.73693 AHv/Te 19.55 5 0.73305 Surface tension 25 to 90.27 5 0.75632 22.68 5 dynes/cm. 20°C 189 °C 0.1374 5 ь -0.03775 21.75 30 to 40 20.84 5 Ref. Index e' °C 20°C 1.42146 2 nD Parachor [P] d<sub>c</sub> g/ml 0.243 5 1.41913 25 2 20°C vc ml/g t °C 5 4.11 30 1.41675 4 30  $t_c$ 343. 5 "C" 40 0.7563 4 P<sub>c</sub> mm 5 16209. 5 413.2 Sugd. MR (Obs.) 48.059 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 47.913 1.0000 25°C 5 u. (nD-d/2)1.05106 2 30 mm 112.9 2 1.0000 5 Dispersion Dielectric 0.9450 BP Flash Point °C 0.9303 5 A 25 to 6.96034 Fire Point 5 B |233 °C 1501.872 0.243 2 M. Spec. ċ 197.58 2 AHc kcal/m 1488,82 2 Ultra V. A\* 76 to B\* 199 °C ΔHf 1.45987 1417.2 5 X-Ray Dif. ΔFf Infrared ĸ Viscosity Solubility in centistokes Acetone to 40°C 0.855 2 ٠c Carbon tet. 60 0.692 2 Benzene 80 0.583 2 A۱ to Ether 100 0.501 2 °C B١ n-Heptane B<sub>v</sub> C' 30 to 479.3 4 Ethanol 2.40181 70°C A1\* Water to Water in (B<sup>V</sup>)| R!# ۰c 70 to 433.9 4 Acl233 to (AV)| 110°C 7,8050 Ž.53743 4 Bc tc 5 2317. c<sub>p</sub> liq. 5 Cc 302. c<sub>p</sub> vap.300K Cryos. Aº 0.38329 2 consts. Bº ž 0.48332 400 c, vap. te °C 189.28 5  $T_R = 0.82 T_c$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No.160	
NAME	l-Und	ecen	•			STRUCTURAL	FORMULA	
l l								
Mole % Pur.	Ref. Mo	lecul	ar C <sub>11</sub> H <sub>22</sub>	Molecular Weight 154.2	86	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>8</sub> C	CH=CH <sub>2</sub>	
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		Ref.			Ref			Ref.
F.P. ℃	-49.185	2	dt/dP	1		f   to		
F.P. 1009			°C/mm	26.16	_	g °K		ļ
B. P. *C	192.671	2	25°C BP	26.16 0.05348	5 2	h		
760 mm 100	124.851	2	t.	0.03615	5	f' to		
30 10	94.89 72.02	4 5	30 mm	0.7506	5	g' K		
1	34.54	5	ΔHm cal/g	L		h'		<u> </u>
Pressure			ΔHv cal/g 25°C	86.85	5	m   300 to n   600 °K	0.0233 0.0014	
mm 25°C	0.490 1255.	5	30 mm	77.50	5	•	-0.0 <sub>6</sub> 50	
Density	11233.	-	BP	64.33 61.70	5	m'   700 to	0.1143	4
g/ml 20°C	0.75032	2	te te (d, e)	61.44	5	n' 1000°K	0.0011	4
dt 25 4 30	0.74655	2	AHV/T	19.53	5	0'	-0.0 <sub>6</sub> 38	4
a 30	0,74278	4	d   25 to	90.27	5	Surface tension dynes/cm, 20°C	22 20	١,
ь	0.7654 -0.03754	4			5	dynes/cm. 20°C	23.39 22.46	5
Ref. Index			d'   to			40	21.56	5
<sup>n</sup> D 20°C	1.42609	2 2	d <sub>c</sub> g/ml			Parachor [P]		
30	1.42167	4	A <sup>C</sup> mr/R	344	_	20°C		
"C"	0.7545	4	,	364. 14948.	5	40	453.3	_
MR (Obs.)		2	P <sub>c</sub> mm	14740.	3	Exp. L. 1, %/wt.	452.2	5
MR (Calc. (nD-d/2)	) 52.531 1.05093	5 2	25°C	1.0000	5	u.		
Dielectric	1.03073	-	30 mm BP	1.00 <b>00</b> 0.9420	5 <b>5</b>	Dispersion	112.1	2
A   25 to	6.96662	2	te	0.9253	5	Flash Point °C Fire Point	1	
B 257 °C	1562.47	2	tc			M Spec.		-
C	189.74	2	ΔHc kcal/m ΔHf	1635.75	2	Ultra V.		
A* 94 to	1.49542	5	ΔFf			X-Ray Dif. Infrared		Į
к — — —	-		Viscosity			Solubility in +		-
t <sub>k</sub>			centistokes 7 40 °C	1.054	2	Acetone	1	1
t <sub>x</sub> •c	;		60	0.835	2	Carbon tet. Benzene		
A'   to			80 100	0.691 0.586	2 2	Ether		
B' °	<del>-</del>		B <sup>V</sup>   30 to	527.6	4	n-Heptane Ethanol		
A'* to	,		A <sup>V</sup>   70 °C	2.33787	4	Water		
B'* °(	<b></b>		(B <sup>V</sup> ) 70 to	471.6	4	Water in		-
Ac   257 to	7.9272	5	(A <sup>V</sup> )  110 °C	<b>Ž.</b> 50 <b>3</b> 97	4			
Cc	312.	5	c <sub>p</sub> liq. •K					
Cryos. A°			cp vap.300°K	0.38403	2			
consts. B°	<del></del>		400	0.48443	2			
t <sub>e</sub> °C	214.20	5	c <sub>v</sub> vap.	l		L	L	
$T_{\mathbf{R}} = 0.83$						grams/100 gran		t
	CES: 1-Dow			Calc. from det	t. da	ta 5-Calc, by for	mula	
SOURCE:		AF	**					
PURIFICA		AF						
LITERATU	RE REFERE	VCES	<b>5:</b>					

								No. 161	l
NAME	l -Dode	cene				ST	RUCTURAL	FORMUL	A.
					$\neg$		сн <sub>3</sub> (сн <sub>2)9</sub> с	H=CH <sub>2</sub>	
Mole % Pur.	Ref. Mo	lecul		Molecular Veight 168.3	,,				
W 1 u1.	F0	Ref.	12 24 1	vergnt 100.5	Ref.	T			Ref.
F. P. °C	-35,230	2	dt/dP		Ker.	<b>-</b>			
F.P. 100%		H	°C/mm			f g	to °K		
B. P. °C			25°C BP	71.96 0.05522	5	h	!		
760 mm 100	213.357 143.315	2 2	te	0.03609		f'	to		
30	112.35	4	30 mm	0.7757	5	g'	°K		
10	88.72 49.20	5	ΔHm cal/g			h'			
Pressure	17.20	╁	ΔHv cal/g			m n	300 to	0.0231 0.0014	4
mm 25°C	0.1446	5	25°C 30 mm	86.74 75.41	5	0	000 ·K	-0.0650	4
t <sub>e</sub>	1309.	5	BP	62.15	5	m'	700 to	0,1145	4
Density g/ml 20°C	0.75836	2	te te (d, e)	59.30 58.97	5	n'	1000 °K	0.0011	4
dt 25	0.75474	2	e (a, e)	19.54	5	0'		-0.0 <sub>6</sub> 38	4
	0.75112	4	ΔHv/T <sub>e</sub>	90.14	ا ج ا	Sur	face tension		
a b	0.77284 -0.03724	4 4	_e_ _238 °C	90.16	5		es/cm. 20°C	23.99	5 ∽5
Ref. Index	-0.03124	-	d' to °C			"	30 40	23.09 22.21	5
n <sub>D</sub> 20°C	1.43002	2	d <sub>c</sub> g/ml		$\vdash$	Par	achor [P]		
25	1.42782	2 4	v <sub>c</sub> ml/g t <sub>c</sub> °C				20°C 30		
"C"	0.7530	4	tc °C	384.	5		40		
MR (Obs.)		2	P <sub>c</sub> mm	13911.	5			491.2	5
MR (Calc.)	57.149	5	PV/RT 25°C	1,0000	5	Exp	u. L.1.%/wt.		
(nD-d/2)	1.05084	2	30 mm	1.0000	5	Dis	persion	110.7	2
Dielectric A 112 to			BP te	0.9400 0.9214	5		sh Point °C		
B 280 °C	6.97522 1619.86	2 2	tc	.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			e Point		_
С	182.27	2	ΔHc kcal/m	1782.68	2		Spec. ra V.		
A* 112 to B* 248 °C	1.52979	5	ΔHf ΔFf			X-I	Ray Dif.		
K 240 C	1554.5		Viscosity				ared		
t, -to			centistokes				ability in <sup>T</sup> etone		
t <sub>k</sub> to ℃			7 40°C	1.286 0.995	2 2	Ca	rbon tet.		
A'   to			80	0.811	2		nzene her		
B'°C	.		B <sup>V</sup> to	0.676	2	n-	Heptane		
A'* to	<del> </del>	$\vdash$	A C	581.2 2.25329	4		hanol ater		
B'* °C			(B <sup>V</sup> )	52 <u>1</u> .0	4	W	ter in		
Acl 280 to	8.0711	5	(A <sup>V</sup> )	Ž. 43370	4				
Bc tc C	2792. 327.	5	c <sub>p</sub> liq. °K						
Cryos, A°	† · · · · · · · · · · · · · · · · · · ·	┯┥	c <sub>p</sub> vap.300°K	0.38464	2	Ì			
consts. B°			1 400	0.48529					
t <sub>e</sub> °C	237.59	5	c <sub>v</sub> vap.						
$T_{\mathbf{R}} = 0.84$	T <sub>c</sub>					+ g:	rams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da				
SOURCE:		AF	PI						
PURIFICAT	TION:	AF	PI						
LITERATU	RE REFERE	NCES	<b>5:</b>						

									No. 162	
NAME	1-7	ride	cene				STRU	CTURAL 1	FORMULA	
Mole % Pur.	Ref.	Mo	lecul		Molecular Weight 182.3	38	C	H <sub>3</sub> (CH <sub>2</sub> ) <sub>10</sub> C	CH=CH <sub>2</sub>	
			Ref.			Ref	r			Ref
F.P. °C	-23,07	0	2	dt/dP			f			
F.P. 100%				*C/mm	1		g	to °K		
B. P. °C	1			25°C BP	207.11 0.05680	5 2	h			
760 mm 100	232.78	3	2 2	t <sub>e</sub>	0.0361	5	f +	to		
30	128.84	ŀ	4	30 mm	0.7980	5	g'	°K		ŀ
10 1	104.5		5	ΔHm cal/g			h'			
Pressure	1		<u> </u>	ΔHv cal/g			m	300 to	0.0233	
mm 25°C	0.05	477	5	25°C	86.76 73.58	5 5	n	600°K	0.0014 -0.0 <sub>6</sub> 50	4
t <sub>e</sub>	1358.		5	30 mm BP	60.05	5				-
Density	0.7/	<b>-</b>		t <sub>e</sub> ,	56.96	5	m'	700 to 1000 K	0.1116 0.0011	
g/ml 20°C	0.76		2 2	te (a, e)	56.57	5	01	1000 20	-0.0639	
d <sup>t</sup> 25 4 30	0.75		4	ΔHv/T <sub>e</sub>	19.50	5	Surfa	ce tension	-	_
	0.77	97	4	d 25 to	90.36 0.1302	5		/cm. 20°C	24.52	5
<u>ь</u>	-0.03	12	4	_d' to			,	30 40	23.61 22.73	5
Ref. Index n <sub>D</sub> 20°C	1,43	36	2	e' i °C		$\vdash$	Parac	hor [P]	22.13	-
D 25	1.43		2	d g/ml v ml/g			1 ****	20°C		
30	1.42		4	tc °C	401.	5		30 40		
"C"	0.75		4	P <sub>c</sub> mm	12776.	5	ł		530.2	5
MR (Obs.) MR (Calc.)	61.96		2	PV/RT		$\vdash$	Exp.	L.1.%/wt.		<u> </u>
(nD-d/2)	61.76		5 2	25°C 30 mm	1.0000 1.0006	5		u.	110.	2
Dielectric				BP BP	0.9370	5	Dispe	Point °C	110.	-
A   128 to		92	2	te	0.9165	5	Fire			
B	1662.68 173.90		2 2	t <sub>c</sub>	1020 (0	2	M Spe	c.		
A*   128 to	1.55		5	ΔHc kcal/m ΔHf	1929.60	-	Ultra	v.		1
B* 270°C	1577.9	000	5	ΔFf			X-Ra			ļ
к ———	Ï			Viscosity				llity in +		┢─╴
ξ <sub>k</sub>	-		l i	centistokes 7 40 °C	1.56	2	Acet	one		ļ
tx c	l			60	1.184	2	Cart Benz	on tet.		1
A'   to				80 100	0.945 0.780	2 2	Ethe	r		ļ
B', ∟ _ °C	·			B <sup>V</sup> to			n-He Etha	ptane		l
A¹* to	f			A <sup>V</sup>   °C			Wate	r		
B'* °C				(BV)			Wate	rin		├
Ac   300 to	8.20	20	5	(A <sup>V</sup> )						l
Bc tc_°C	3037. 341.		5	c <sub>p</sub> liq. °K						ļ
Cryos, A°			۲	c <sub>p</sub> vap.300°K	0.38522	2				l
consts. B°	Ĺ.,			}	0.48602					1
t <sub>e</sub> °C	259.52		5	c <sub>v</sub> vap.						ļ
$T_{\mathbf{R}} = 0.85$	Тc						+ grar	ns/100 gran	ns solven	t
REFERENC	ES: 1-D	ow	2-AF	PI 3-Lit. 4-C	alc, from det	dat				
SOURCE:			API							
PURIFICAT	ION:		API							
LITERATUI	RE REFI	EREN	CES	:						

									No.163	
NAME	1-	Tetr	adec	ene			ST	RUCTURAL	FORMUL	A.
								СН <sub>3</sub> (СН <sub>2</sub> ) <sub>11</sub> С	H=CH <sub>2</sub>	
Mole % Pur.	Ref.	Mo!	ecul		Molecular Weight 196.30	64				
70 1 411		10.	Ref.	1 17 20 1	Weight 170.3	Ref.	Γ	<del> </del>		Ref.
F.P. °C	-12,85		2	dt/dP			f	4-		
F.P. 100%				°C/mm			g	to °K		
B.P. °C				25°C BP	497.07 0.05820	5 2	h			
760 mm	251.10 177.1	0	2 2	te	0.0361	5	f'	to		
30	144.40		4	30 mm	0.8181	5	g'	°K		
10	119.0 74.5		5	ΔHm cal/g			h'			_
Pressure				ΔHv cal/g			m n	300 to	0.0231 0.0014	
mm 25°C	0.01	815	5	25°C 30 mm	86.67 71.91	5	0	••	-0.0650	4
Density	1406.		5	BP	58.18	5	m'	700 to	0.1155	-
g/ml 20°C	0.77	13	2	t <sub>e</sub> (d, e)	54.89 54.42	5	n'	1000 °K	0.0011	4
dt 25 4 30	0.76		2	AHv/Te	19.48	5	ο'		-0.0 <sub>6</sub> 38	4
a 30	0.76		4	d 25 to	90,50	5		face tension	24.00	_
b	-0.03		4	_e_ 290°C	0.1288	5	dyn	es/cm. 20°C 30	24.99 24.07	5
Ref. Index				d' to				40	23.17	5
<sup>n</sup> D 20°C	1.43		2	d <sub>c</sub> g/ml			Par	achor [P]		ĺ
30	1.43		4	v ml/g	416.	5		30		ĺ
"C"	0.75	07	4	tc °C P <sub>c</sub> mm	11740.	5		40 Suad	569.2	5
MR (Obs.)			2	PV/RT	11110	-	Evr	. L.1.%/wt.	307.2	<del>-</del> -
MR (Calc. (nD-d/2)	) 66.38 1.05		5 2	25°C	1.0000	5	_	u.		ĺ
Dielectric	<b>†</b>			30 mm BP	1.0000 0.9350	5		persion	109.	2
A 144 to	6.96	15	2	t <sub>e</sub>	0.9129	5		sh Point °C e Point		
B   319 °C	$- \begin{vmatrix} 1699.76 \\ 165.53 \end{vmatrix}$		2	tc ΔHc kcal/m	2067,52	2	М.	Spec.		
A* 144 to			5	ΔHf	2007.52	-		ra V. Ray Dif.		
B* 290 °C			5	ΔFf				ared		
K — —				Viscosity centistokes				bility in +		
t <sub>k</sub> to				η 40°C	1.85	2		etone rbon tet.		
t <sub>x</sub> °C		0.41	-	60 80	1.40	2 2	Be	nzene		
B' 144 °C		041	5	100	0.891	2		her Heptane		ĺ
	203.3		5	B <sup>V</sup>   to A <sup>V</sup>   °C			Et	hanol		
A'* 20 to B'* 144 °C		444	5	$\frac{1}{ \mathbf{B}^{\mathbf{v}} } - \frac{1}{ \mathbf{B}^{\mathbf{v}} }$	ŀ			iter iter in		
Acl 319 to		16	5	(A <sup>V</sup> )						
Bc tc °C	3298.	-	5	c <sub>p</sub> liq. °K		$\vdash$				
Cryos. A°	356.		5	P	0.305//					1
consts. B°				c <sub>p</sub> vap.300°K	0.38566 0.48665	2				
te °C	280.17		5	c <sub>w</sub> vap.						
$T_R = 0.86$	T <sub>c</sub>						+ g1	ams/100 gra	ms solven	t
REFEREN		)ow	2-A	PI 3-Lit. 4-0	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:			AP	I						
PURIFICAT			AP							
LITERATU	RE REF	EREI	NCES	<b>5</b> :						

·							<u>-</u>		No. 164	
NAME		l-Per	ntad	ecene			STR	UCTURAL 1	FORMULA	
Mole % Pur.	Ref.	Mole For	ecul:	arC <sub>15</sub> H <sub>30</sub>	Molecular Weight 210.3	90		СН <sub>3</sub> (СН <sub>2</sub> )12	сн≖сн₂	
			Ref.			Ref				Ref.
F.P. °C	-3.73	0	2	dt/dP			f	to		
F.P. 100%				*C/mm	1629.5	5	g	°K		
B. P. °C	3/0 17			25°C BP	0.0596	4	h			
760 mm 100	268.17 192.5	1	2 2	t.	0.0361	5		to		
30	159.1	- 1	4	30 mm	0,8362	5	g'	°K		
10 1	133.7		5	ΔHm cal/g			h'			
Pressure	1 /1.2		<u>-</u>	ΔHv cal/g	1		m	300 to	0.0232	
mm 25°C	0.00	609	5	25°C	86.60 70.37	5 5	n o	600 °K	0.0014 -0.0 <sub>6</sub> 50	
t <sub>e</sub>	1448.0		5	30 mm BP	56.41	5	ļ			-
Density	0.77			t	52.92	5	m' n'	700 to	0.1155 0.0011	4
g/ml 20°C	0.77		2 2	t <sub>e</sub> (d, e)	52.39	5	0'	1000 K	-0.0638	
dt 25 4 30	0.76		4	ΔHv/T <sub>e</sub>	19.44	5	Sund.			
a	0.79	04	4	d   159to		5		s/cm. 20°C	25.39	5
Ъ	-0.03	70	4			5 5	3	30	24.49	5
Ref. Index			١ ,	e' 159°C		5	<u> </u>	40	23.61	,
n <sub>D</sub> 20°C	1.43		2 2	d g/ml vc ml/g			Para	chor [P] 20°C		!
30	1.43		4	tc °C	431.	5		30		
"C"	0.74	97	4	1 -	10918.	5	}	40 50 ad	608.2	5
MR (Obs.)			2	P <sub>c</sub> mm PV/RT	10718.	1	F		608,2	-
MR (Calc. (nD-d/2)	71.00 1.05		5 2	25°C	1.0000	5	Exp.	L. 1. %/wt. u.		
Dielectric	1.03	-	-	30 mm	1.0000 0.9325	5	Disp	ersion	108.	2
A 159 to	6, 95	02	2	BP t <sub>e</sub>	0.9086	5		h Point °C		
B 337°C			2	tc				Point		ļ
c ——	157.		2	ΔHc kcal/m	2223.44	2	M S <sub>I</sub>			
A* 159 to		900	5	ΔHf ΔFf				y Dif.		
B* ∟305 °C	1647.3	i	5	Viscosity	<del> </del>	-	Infra			
c	_	ŀ		centistokes				bility in +		
tk to				7 40 °C	2.21	2 2		tone bon tet.		
t <sub>x</sub> i *C	<u> </u>			60 80	1.64 1.265	2	Ber	zene		
B' °C				10 <b>0</b>	1.016	2	Eth	er Ieptane		
<u>c'</u>		i		B <sub>v</sub> l to			Eth	anol		
A¹* to				AV I _ °C			Wat	ter ter in		
B'* °C	<del></del>			(B <sup>V</sup> )			- " a	er m		1
Ac 337 to Bc t <sub>c</sub> °C		72	5	(A <sup>V</sup> )	L					ĺ
Cc Cc	376.	- 1	5	c <sub>p</sub> liq. ∘K						1
Cryos. A°				cp vap.300 °K	0.38609	2				
consts. B°	<b>_</b>			400	0.48719	2				
t <sub>e</sub> °C	299.58		5	c <sub>v</sub> vap.						
$T_{R} = 0.87$	T <sub>c</sub>						+ gra	ms/100 grai	ns solven	t
REFERENC	ES: 1-D	ow 2	-AF		Calc. from det	t. dat	a 5-	Calc. by for	mula	
SOURCE:			AF	PI						
PURIFICAT	ION:		AF	PI						
LITERATU	RE REF	EREN	CES	:						

No. 165 l-Hexadecene NAME STRUCTURAL FORMULA CH3(CH2)13CH=CH2 Molecular C16H32 Mole Ref. Molecula r 224.416 % Pur. Weight Ref. Ref. Ref. 4.120 2 F. P. ℃ dt/dP f to F.P. 100% °C/mm g ۰c 25°C 4630. 5 B. P. °C h BP 0.0610 4 284.4 2 760 mm t<sub>e</sub> 0.0362 5 ſ١ 207.2 to 100 2 °C g' 173.13 4 30 30 mm 0.8527 5 10 147.2 5 h! ∆Hm cal/g 103.9 5 0.0231 300 to m AHv cal/g Pressure ۰ĸ n 600 0.0014 25°C 86.45 mm 25°C 0.0352 68.96 30 mm 5 1487. 5 o -0.0650 t. BP 54.75 5 m' | to 700 0.1158 4 Density 5 51.07 te (d, e) g/ml 20°C 0.78112 n' 1000 °K 0.0011 50.46 5  $d_4^t$ 25 0.77759 ۰, -0.0638 4 AHv/T 5 19.39 30 0.77406 Surface tension d 173 91.08 5 0.79524 to dynes/cm. 20°C 25.75 5 1310 ᇷᅴ •c 0.1277 5 Ъ -**0.0**3706 30 24.83 5 to 40 23.94 5 Ref. Index e! °C 20°C [P] 1.44120  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 25 1.43907 2 20°C vc ml/g t °C 30 1.43694 4 30  $^{\mathbf{t}}_{\mathbf{c}}$ 5 444. 40 "C" 0.7489 4  $P_c$  mm 10008. 5 Sugd. 647. 2 5 MR (Obs.) 75.898 PV/RT Exp. L.1.%/wt. MR (Calc.) 75.621 25°C 1.0000 (nD-d/2) 1.05064 2 30 mm 1.0000 Dispersion 107.1 2 0.9300 Dielectric BP 5 Flash Point °C 0.9043 5 A 173 to 6.936 2 Fire Point B \_352°C 1755.2 2 M. Spec. Ultra V. С 148. 2 2370.37 AHc kcal/m 2 ΔHf A# 173 to 1.59011 5 X-Ray Dif. ΔFf B\*[325 °C 1674.1 Infrared ĸ Viscosity Solubility in centistokes Acetone to 40 °C t<sub>k</sub> [ t<sub>x [</sub> 2.62 2 Carbon tet. °C 60 1.89 2 Benzene 80 1.46 2 A' I to Ether 100 1.149 B' °C n-Heptane Т C 35 Ethanol to 740.1 5 **v** | 70 •c **Z.** 05527 Water AI\* to Water in B'\* °C (BV) 70 to 685.6 5 Ac 352 to (AV) 105 .c 8.6611 5 5 ₹. 22324 3905. Bc \_tc\_ 5 °C c<sub>p</sub> liq. Cc 5 394. c<sub>p</sub> vap.300K Cryos. A 0.38643 2 consts. B° 0.48767400 c vap. t, °C 5 317.97 = 0.87T grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

							<b>No.</b> 166	
NAME	l-Heptadeo	ene			_	STRUCTURAL 1	FORMULA	
Mole % Pur.		lecul rmul	ar C <sub>17</sub> H <sub>34</sub>	Molecular Weight 238.4	142	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>14</sub> С	H=CH <sub>2</sub>	
	<del></del>	Ref.			Ref			Ref
F.P. °C	11.2	2	dt/dP			f to		
F.P. 100%			°C/mm	İ		g °K		
B. P. °C			25°C BP	12449. 0.0622	5 4	h .		
760 mm 100	299.7 221.0	2	t <sub>e</sub>	0.0362	5	$\frac{1}{\mathbf{f}} + \frac{1}{\mathbf{to}}$		
30	186.34	4	30 mm	0.8672	5	g' K		
10 1	160.	2	ΔHm cal/g		1	h'		
	116.	)	ΔHv cal/g			m   300 to	0.0229	4
Pressure mm 25°C	0.0,708	5	25°C	86.40	5	n 600 °K	0.0014 -0.0650	
te	1522.	5	30 mm BP	67.65 53.06	5		-0.0630	
Density			te (d.e)	49.31	5	m'   700 to	0.1160	4
g/ml 20°C	0.7852 0.7817	2 2	e (-, -,	48.51	5	n' 1000 *K	0.0011 -0.0 <sub>6</sub> 38	4
d <sup>t</sup> 25 4 30	0.7782	4	ΔHv/T <sub>e</sub>	19.33	5	1		<u> </u>
a	0,7992	4	d 186 to	91.63	5	Surface tension dynes/cm. 20°C	26.07	5
ь	-0.0370	4	_e _ 325 °C to	0.1287	5	y 30	25.16	5
Ref. Index	, ,,,,,,	١.,	e' *C		5	40	24.26	5
<sup>n</sup> D 20°C	1.4432	2 2	d <sub>c</sub> g/ml			Parachor [P] 20°C		
30	1.4390	4	V <sub>c</sub> mi/g	455.	5	30		
"C"	0.7482	4	l C	9252.	5	40	686.2	5
MR (Obs.)	80.54	2	P <sub>c</sub> mm	7232.	-	Sugd.	000.2	<del>-</del> -
MR (Calc.) (nD-d/2)		5 2	25°C	1.0000	5	Exp. L.l.%/wt. u.		
Dielectric	1.0506	-	30 mm	1.0000	5	Dispersion	107.	2
	/ 030	_	BP t <sub>e</sub>	0.9250 0.8995	5	Flash Point °C		
A 186 to B 1366 °C	6.920 1774.6	2	tc			Fire Point		ļ
с	139.7	2	ΔHc kcal/m	2517.30	2	M Spec. Ultra V.		l
A*   186 to	1.59829	5	ΔHf ΔFf			X-Ray Dif.		ĺ
B*  340 °C K	1695.7	5	Viscosity		$\vdash$	Infrared		ļ
¢			centistokes			Solubility in + Acetone		l
t <sub>k</sub> to			7 40 °C	3.08 2.18	2	Carbon tet.		ĺ
A' to	<del>                                     </del>	$\vdash$	80	1.66	2	Benzene Ether		ĺ
B'i °C			100	1.30	2	n-Heptane		ĺ
c'	ļ		B <sup>V</sup>   35 to A <sup>V</sup>   70 °C	783.1 3.98823	5	Ethanol Water		ĺ
A'* to B'* °C				699.7	5	Water in		ĺ
Ac   366 to	8.8342	5	(B <sup>V</sup> ) 70 to (A <sup>V</sup> ) 105 °C	Z. 23908	5			
Bc t °C	4239.	5		2.23908	9			1
CE	416.	5	c <sub>p</sub> liq. ∘K	1				ĺ
Cryos. A° consts. B°	1		cp vap.300°K	0.38672	2			ĺ
t <sub>e</sub> °C	225 11	5	c <sub>v</sub> vap.	0.48809	2			ĺ
$T_{R} = 0.88$	335.11		L *	l	<u> </u>	·		L
		2 4 7	OT 2 124 4 4			grams/100 gram		<u> </u>
REFERENC	20: 1-DOM	2-AI		alc. from de	. da	ta 5-Calc. by for	mula	
SOURCE:	ION.	AF						
PURIFICAT	RE REFEREN	AF						

								No. 167	7
NAME	1-Octadecen	е				ST	RUCTURAL	FORMUL	A.
					1			c c	
Mole % Pur.		ecul		Molecular Veight 252.4	68		СН <sub>3</sub> (СН <sub>2</sub> ) <sub>15</sub>	CH=CH <sub>2</sub>	
	1	Ref.			Ref.				Ref.
F.P. °C	17.6	2	dt/dP			f	to		
F. P. 100% B. P. °C 760 mm	314.2	2	°C/mm BP	0.0633	4	g h	' ° <u>K</u>		
100	234.2	2	t <sub>e</sub>	0.03596	5	f'	to		
<b>3</b> 0 10	198.91 172.	2	30 mm	0.8805	5	g'	'° <u>K</u>		
i	128.	5	ΔHm cal/g			_h'	300 to	0.0230	4
Pressure mm 93.77 °C		5	ΔHv cal/g 25°C 30 mm	66.44	5	m n o	600 •K	0.0014 -0.0 <sub>6</sub> 50	4
t <sub>e</sub>	1575.	5	BP	52,04	5	m'	700 to	0.1164	4
Density g/ml 20°C	0.7888	2	te te (d, e)	48.18 47.31	5	n'	1000 °K	0.0011	4
dt 25	0.7853	2	ΔHv/T <sub>e</sub>	19.45	5	0'		-0.0 <sub>6</sub> 38	4
	0.7818	4	d 199 to	91. 20	5	Sur	face tension		
a b Ref. Index	0.8028 -0.0 <sub>3</sub> 70	4	e 360 °C to	0.1246	5	g dyn	es/cm. 20°C 30 40	26.36 25.43 24.53	5 5 5
n <sub>D</sub> 20°C		2	e'   °C		$\vdash$	Par	achor [P]		<u> </u>
25 30	1.4428 1.4408	2	d g/ml v ml/g				20°C		
"C"	0.7476	4	vc m1/g tc °C	466.	5		30 40		
	<del></del>		P <sub>c</sub> mm	8609.	5			725.2	5
MR (Obs.) MR (Calc. (nD-d/2)		2 5 2	PV/RT 25°C	1.0000	5		L.1.%/wt. u.		
Dielectric		-	30 mm BP	1.0000 0.9303	5		persion	107.	2
A 199 to	6,901	2	te tc	0.9053	5		sh Point C e Point		
B 1380 °C C	1789.4 131.	2 2		2//4 22	2		Spec.		
A* 199 to B* 361 °C	1.58563 1707.4	5	ΔHc kcal/m ΔHf ΔFf	2664.22	2	Ult:	ra V. Ray Dif. ared		
к — — —			Viscosity centistokes			Sol	ability in +		
t <sub>k</sub> to			7 40 °C	3.59 2.51	2 2		rbon tet.		
t <sub>x</sub> °C	<del>                                     </del>	-	80	1.87	2		nzene her		
B' °C	_		100	1.46	2		Heptane		
C'	ļ		B <sup>V</sup>   35 to A <sup>V</sup>   70 °C	810.4 3.96761	5		hanol ater		
A'* to B'* °C			(B <sup>V</sup> )  70 to	560.8	5		iter iter in		
Acl 380 to Bc t <sub>c</sub> °C	9.04000 4643.	5	(A <sup>V</sup> )  105 °C	₹. 69007	5				
Cc C	443.1	5	c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap.300K 400	0.38702 0.48846	2				
t <sub>e</sub> °C	352.14	5	c <sub>v</sub> vap.						
$T_{R} = 0.8$							ams/100 gra		t
	CES: 1-Dow			Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		AP							
PURIFICAT		AP							
LITERATU	RE REFERE	NCES	<b>5:</b>						

								No. 168	
NAME	l-Nona	lecen	e				STRUCTURAL		
I						$\neg$			
ļ						$\neg$	СH <sub>3</sub> (СH <sub>2</sub> ) <sub>16</sub>	CH=CH <sub>2</sub>	
Mole % Pur.	Ref.		ecul mul		Molecular Weight 266.4	.94			
<del></del>			Ref.	1 1/ 30 1		Ref		<del></del>	Ref.
F. P. °C	23.4		2	dt/dP	1		f   to	T	
F.P. 100%				*C/mm	l		f to g °K	ł	ļ
B. P. °C	1			106.43 °C BP	125.89	5 2	h	ł	l
760 mm 100	328.0 247.		2 2	t	0.03604	5	f' to		i
30	211.15	.	4	30 mm	0.8925	5	g' K	į	
10 1	184.		2 5	ΔHm cal/g			h'	ļ <u>.</u>	<u> </u>
Pressure	<b>-</b>	$\dashv$	$\neg$	ΔHv cal/g			m   300 to n   600 °K	0.0230	
mm 106.43°			5	25°C 30 mm	86.30 65.33	5. 5	0 000 1	-0.0650	
t <sub>e</sub>	1606.06	<u>'</u>	5	BP	50.79	5	m'   700 to	<del>                                     </del>	<del> </del>
Density g/ml 20°C	0.79	20#	2	te te (d, e)	46.63 45.83	5	n' 1000 *K	0.1187	
at 25	0.78	86	2	ΔHv/T <sub>e</sub>	19.38	5	o' ¦	-0.0638	
	0.78		4	d   211 to		5	Surface tension		
a b	0.80		4 4	<u> </u>		5	dynes/cm. 20°C	26.60 25.70	5
Ref. Index	1 3,53	-	-	d'   to			40	24.83	5
n <sub>D</sub> 20°C	1.44		2	<del></del>	<del></del>	$\vdash$	Parachor [P]		
25 30	1.44		2 4	d g/ml v ml/g	ł		20°C 30	1	l
"C"	0.74		4	16 °C	477.	5	40	1	
MR (Obs.)	89.83		2	P <sub>c</sub> mm	8079.4	5		764.2	5
MR (Calc.)	89 47	5	5	PV/RT 106.43 °C	1.0000	5	Exp. L.1.%/wt. u.	1	İ
(nD-d/2)	1.05	06"	2	30 mm	1.0000	5	Dispersion	106.	2
Dielectric	ļ			BP	0.9300	5	Flash Point °C		
A 211 to B 394 °C		1	2 2	te t <sub>c</sub>	0.7005		Fire Point		
c	122.		2	ΔHc kcal/m	2811.14	2	M Spec. Ultra V.	İ	
A* 211 to	1.58		5	ΔHf ΔFf			X-Ray Dif.	İ	
B* <sub>L</sub> 37 <u>7</u> °C	1720.93	1	5	Viscosity	<del> </del>	$\vdash$	Infrared	ļ	<u> </u>
·	4	l		centistokes			Solubility in + Acetone		ŀ
t <sub>k</sub> to		1		7 40 ℃	4.17 2.87	2 2	Carbon tet.	İ	
A' to	<del> </del>			80	2.12	2	Benzene Ether		
B' ∟ _ °C	.]	l		B <sup>V</sup>   35 to	1.63	2	n-Heptane		1
C' to	<u> </u>			B'   35 to A'   70 °C	846.6 3.91708	4	Ethanol Water	1	
B'* °C				(BV) 70 to	752.3	4	Water in		
Ac   394 to	9. 29		5	(AV) 105 °C	2.19638	4			
Bc tc_°C	5158.13		5	c <sub>p</sub> liq. °K	<del> </del>				
Cryos. A°	479.41			l <sup>-</sup>	0 20725			1	
consts. B°		i		c <sub>p</sub> vap.300°K	0.38725 0.48879	2 2		1	
t <sub>e</sub> °C	367.84	Ť	5	c <sub>w</sub> vap.					
$T_{\mathbf{R}} = 0.8$			unde	rcooled liquid	<u> </u>		grams/100 gra	ms solven	<u></u>
					Calc. from det	da:	ta 5-Calc. by for	mula	
SOURCE:			AP					<del></del>	
PURIFICAT	ION:		AP						
LITERATUI	RE REFI	EREN							
1									

No. 169 l-Eicosene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_{17}CH=CH_2$ Mole Molecular Molecular C20H40 Weight 280.520 % Pur. Formula Ref. Ref. F.P. °C F.P. 100% 28.6 2 dt/dP f to °C/mm g °<u>K</u> 117.0°C 127.12 5 B. P. \*C h BP 0.06527 4 760 mm 341,2 2 te 0.03634 5 ſ١ 100 to 259. 2 g' <u>°К</u> 30 222.92 4 0.9036 5 30 mm 10 195. 2 h' ∆Hm cal/g 1 149. 300 to 0.0230 m AHv cal/g Pressure n 600 °K 0.0014 25°C 86.27 5 mm 117.0°C 0.1000 o -0.0650 4 30 mm 64.32 5 te 5 1622. ΒP 49.34 5 700 to m 0.1166 4 Density 5 44.87 te (d, e) n' 1000 °K 0.0011 g/ml 20°C 0.7950 2 44.12 0.7916 0.7882 ۰' -0.0638 4 25  $\mathbf{d_{4}^{t}}$ 2 AHv/T 19.19 5 30 4 Surface tension 223 to 92.55 5 0.8086 a dynes/cm, 20°C 26.85 <u> 1 383</u> 5 •C 0.1266 h -0.0368 30 25.94 ď٦ to 40 25.06 Ref. Index 5 e' •c <sup>n</sup>D 20°C 1.4480<sup>‡</sup> 1.4459<sup>‡</sup> [P] Parachor dc g/ml 25 2 20°C vc ml/g t °C 1.4439<sup>‡</sup> 30 4 30 486. 5 t<sub>c</sub> 40 "C" 0.7466 4 5 7425. P<sub>c</sub> mm Sugd. 803.2 5 94.47<sup>‡</sup> MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 94.093 25°C 1,0000 5 1.0506 (nD-d/2)2 106.<sup>≠</sup> 30 mm 1.0000 ž Dispersion Dielectric BP 0.9250 5 Flash Point C 5 A 223 to 6.859 te t 0.8894 2 Fire Point B 405 °C 1807.9 2 M. Spec. Ċ 113. 2 AHc kcal/m 2958.06 Ultra V. ΔHf A\* 223 to 1.59840 5 X-Ray Dif. ΔFf B\*[393 °C 1734.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to 40 °C 4.81 2 °C Carbon tet. 60 3.27 2 Benzene 80 2.38 2 Α' to Ether 100 1.81 2 B <u>•с</u> n-Heptane B<sup>V</sup> | 35 A<sup>V</sup> | 70 C' 5 874.5 to Ethanol 3.89001 5 A'\* °C Water to (BV) 70 Water in B'\* °C 5 783.6 to Ac 405 to 9.50720 (A<sup>V</sup>)| 105 Z. 15801 5 5 °C 5591. Bc tc C c<sub>p</sub> liq. Cc 5 505. c<sub>p</sub> vap 300°K 0.38749 Cryos, Aº 2 consts. B 400 0.48909 c<sub>v</sub> vap. te °C 5 382.48  $T_{R} = 0.89T$ for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: PURIFICATION: LITERATURE REFERENCES:

										No. 170	)
NAME		1 - F	lene	icose	ne			STRUCTU	RAL I	FORMUL	A.
								сн <sub>3</sub> (сі	1. ) CI	H=CH_	
Mole % Pur.	R	lef.	Mo Fo	lecul	ar C <sub>21</sub> H <sub>42</sub>	Molecular Weight 294,5	46	03(0.	-2/18	2	
				Ref.			Ref				Ref
F. P. *C	33	3.3		2	dt/dP	T		f	to		
F.P. 1007					*C/mm			g	°K		1
B.P. °C		_			0.1 mm BP	146.20	5 4	ъ ;			1
760 mm 100	355	5. 1.27		5	te	0.0354	5	<u>-</u>	to		
30	233	3.06		5	30 mm	0.9666	5	g'	°K		
10 1		3.53 3.7 <b>5</b>		5	ΔHm cal/g			h'			
0.1		5.62		5	ΔHv cal/g			m (	to		1
Press. mm				$\vdash$	0.1 mm	69.75	5	n	°К		
t <sub>e</sub>	1602	2.		5	30 mm BP	59.60 47.98	5 5	<u> </u>			
Density	. 🗆	. 70	#		l t	44.61	5	m'   n'	to °K		1
g/ml 20°C	, ,	0.79 0.79	43 <sup>‡</sup>	2 2	t <sub>e</sub> (d, e)	44.22	5	6 !			1
d <sub>4</sub> 25	6	7.79	09 <sup>‡</sup>	4	ΔHv/T <sub>e</sub>	19.39	5	Sunface to			┼
a		81	13	4	d   233 to		5	Surface te dynes/cm.		27.06	5
ь	$\overline{}$	0.03	68	4	d'   393 - °C		5	) y	30	26.15	5
Ref. Index	. 1 1	1.44	0.4 <del>‡</del>	١, ١	e' 233 °C		5	<del></del>	40	25.27	5
<sup>n</sup> D 25	1 1	44	7 2 T	2 2	d <sub>c</sub> g/ml			Parachor	[P] 20°C		l
30	1	. 44	5 <b>2</b> ₹	4	vc ml/g tc °C	1	1 1		30		1
"C"		. 74		4	P <sub>c</sub> mm		1 1		40 Sugd.	842.2	5
MR (Obs.)		.09		4	PV/RT	<del> </del>	$\vdash$	Exp. L.1.		012.2	13
MR (Calc. $(nD-d/2)$	) 98	3.71 .05	1 05≠	5 2	0.1 mm	1.0000	5	u.	,0, w.c.	,	1
Dielectric		2.10		5	30 mm BP	1.0000	5	Dispersion		105.≠	2
A   233 to		. 54		4	t <sub>e</sub>	0.8948 0.8627	5	Flash Point			
B 443°C	2457	. 3		4	t <sub>c</sub>	<u> </u>					╁
С	172	<u> </u>		5	ΔHc kcal/m			M Spec. Ultra V.			
A* 2 33 to B* 403 °C		31	717	5	ΔFf			X-Ray Dif	•		1
K Lion		. 0			Viscosity			Infrared	in +	l	-
t	_				centistokes	1		Solubility Acetone	ın T		Į
t <sub>x</sub>   t <sub>0</sub>					η •ο			Carbon to	et.		
A' to				1	1			Benzene Ether			
B' •	<u>:</u>			]	B <sup>V</sup> to	<del> </del>	1	n-Heptan	e		]
C'	+			1	B' to	i		Ethanol Water			i
A'* 115 to B'* 233 *(		. 23	287	5	(BV)	-		Water in			
Ac  to	<del></del>	<u> </u>		H	(A <sup>V</sup> )						
Bcit 0				1	c <sub>p</sub> liq. °K	+	+				
	1-			$\vdash$	_	1					
Cryos, A° consts, B°				i	c <sub>p</sub> vap. °K	1					Ì
t <sub>e</sub> °C	394	. 33		5	c <sub>v</sub> vap.						
≠ for und								f grams/l	00 grai	ns solver	ıt
REFEREN	CES:	1 - Do	w	2-AF	PI 3-Lit. 4-	Calc, from de	t. da	ta 5-Calc.	by for	mula	
SOURCE:				API							
PURIFICA				API							
LITERATU	RE R	EFE	CREI	NCES	<b>::</b>						

								<b>N</b> o. 17	1
NAME	1-Docose	ne				STRU	CTURAL	FORMUI	-A
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 308.5	72	СН3	(СН <sub>2</sub> ) <sub>19</sub> СІ	H=CH <sub>2</sub>	
		Ref		T	Ref.				Ref
F. P. °C	37.8	2	dt/dP			f	to		
F.P. 100%			°C/mm			gl	°K		1
B. P. °C	2/2	١.	0,1 mm BP	148.82 0.0656	5 4	h			
760 mm 100	367. 281.87	2 5	t <sub>e</sub>	0.0354	5	f'	to		
30	243.02	5	30 mm	0.9831	5	g'	°K		
10 1	212.98 162.33	5	ΔHm cal/g			h'			
0.1	123.52	5	ΔHv cal/g			m	to		1
Press. mm			0.1 mm 30 mm	68.09	5	n	°K		ļ
t <sub>e</sub>	1632.3	5	BP	58.15 47.39	5	m'			+-
Density g/ml 20°C	0.8002	2	te (d.e)	43.34	5	n' i	to °K		ŀ
dt 25		2	'e (-, c,	43.0	5	0'			1
4 30	0.7934	4	ΔHv/T <sub>e</sub>	19.63	1	Surfac	e tension		+-
a L	0.8138	4	d 243 to e 408 °C	80.49 0.0919	5 5		cm. 20°C	27.27	5
b Def 7.1	-0.0368	*	d' 122 to	78.37	5	8	30 40	26.35 25.46	5
Ref. Index	) , ,=o=F	2	e'   243 °C	0.0832	5	Paracl		23.40	13
25		2	d g/ml			Faraci	20°C		
30	1, 1104	4	vc ml/g tc°C			ļ	30 40		
"C"	0.7457	4	P <sub>c</sub> mm					881.2	5
MR (Obs.) MR (Calc.)	103 329	<b>2</b> 5	PV/RT			Exp. I	J. 1. %/wt.		1
(nD-d/2)	1.0505	2	0.1 mm 30 mm	1.0000 1.0000	5			105.≠	2
Dielectric	2.10	5	BP	0.8943	5	Disper		105.	<u>  -</u>
A 243 to	7.55128	4	t <sub>e</sub>	0.8612	5	Flash Fire P	Point °C		
B   458 °C	2505.7 169.5	4 5	t <sub>c</sub> ΔHc kcal/m		1	M. Spe			+
A*  243 to	2.33832	5	ΔHC RCa1/H	1		Ultra '	٧.		
B*[418 °C		5	ΔFf			X-Ray Infrare			
K — — —		1	Viscosity	1		Solubil			+
t <sub>k</sub> – tō	·	l	rentistokes 7°C			Aceto	ne		1
tx C		1	'			Benze	on tet.	Ì	1
A'   to B'   °C					1	Ether	•		1
C'	-	ĺ	B <sup>V</sup> to			n-Hej Ethan		,	
A** 122 to	2,25251	5	AV I ℃			Wate	r		
B'* 243 °C	2390, 03	5	(B <sup>V</sup> )	-[		Water	r in		
Acl to			(A <sup>V</sup> )[					İ	
Bc tc °C	-	1	c <sub>p</sub> liq. °K						
Cryos. A°	<del> </del>	<del>                                     </del>	-11	İ		1		1	1
consts. B			P -						
t <sub>e</sub> °C	408.05	5	c <sub>v</sub> vap.					<u> </u>	$\perp$
# for unde	rcooled liquid					† gran	s/100 gra	ms solve	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5-Ca	lc. by for	mula	
SOURCE:		API							
PURIFICAT	TION:	API							
LITERATU	RE REFERE	NCE	S:						

							No. 172	
NAME	1-Tricos	ene				STRUCTURAL 1	FORMULA	
14.1.	2 / 1/					CH <sub>3</sub> (CH <sub>2</sub> ) <sub>20</sub> C	H=CH,	
Mole % Pur.	Ref. Mo	rmu]	ar C <sub>23</sub> H <sub>46</sub>	Molecular Weight 322.	598	3 2 20	•	
		Ref.		<del>T</del>	Ref.			Ref.
F.P. *C	41.6	2	dt/dP			f to		
F. P. 1009	6		°C/mm		ا ۔ ا	g °K		
B.P. °C	270	,	0.1 mm BP	151.55 0.0667	5 4	h		
760 mm 100	379. 292.47	5	t.	0.0355	5	f' to		ŀ
30	252.95	5	30 mm	1.0001	5	g' °K		
10	222.39 170.83	5	ΔHm cal/g			h'		
0. 1	131.32	5	ΔHv cal/g			m to		ŀ
Press. mn		_	0.1 mm 30 mm	66.50 56.80	5	ö		1
t <sub>e</sub>	1662.0	5	BP	45.63	5	m'   to		
Density g/ml 20°0	0.8023	2	te te (d, e)	42.20	5	n' 'K		
dt 25	0.79905	2	ΔHv/Te	41.85 19.59	5	0'		
	0.7958	4	d   254 to	<del></del>	5	Surface tension		
a b	0.8156 -0.0 <sub>3</sub> 66	4	e   419 °C		5	dynes/cm. 20°C	27.44 26.55	
Ref. Index		┝∸	d'   132 to		5	40	25.68	
n <sub>D</sub> 20°0	~   1 4 5 1 7 7	2		0.0797	1	Parachor [P]		
25	1.4516, 1.4496, 1.4475	2	d g/ml vc ml/g		i	20°C		
"C"	0,7453	4	I'c C			40		l
MR (Obs.		2	P <sub>c</sub> mm			Sugd.	920.2	5
MR (Calc.	1 107 047	5	PV/RT	1 0000	_	Exp. L.1.%/wt.		
(nD-d/2)	1.0505*	2	0.1 mm 30 mm	1.0000	5	u. Dispersion	105.≠	2
Dielectric		5	BP t <sub>e</sub>	0.8936 0.8596	5	Flash Point °C		
A 254 t		4	tc	0.0370		Fire Point		
c	167.5	5	ΔHc kcal/m			M Spec. Ultra V.	1	
A*   254 to			ΔHf ΔFf			X-Ray Dif.		
B* 429 °C	2477.9	5	Viscosity	<b>+</b>	<del> </del>	Infrared		
\$ .— —.	_		centistokes			Solubility in + Acetone		ĺ
t <sub>k</sub>   t			<b>η °</b> ℃	•		Carbon tet.		
A' i to	<del></del>	-				Benzene Ether		
B' º	의	1	B <sup>V</sup>   to	<u> </u>	-	n-Heptane		
C'	2 2252	-	B' to			Ethanol Water		1
A'* 132 to B'* 254 °C		5	(BV)	-		Water in		<u> </u>
Ac  te		<u> </u>	(A <sup>V</sup> )					
Bc tc_'	의	ĺ	c <sub>p</sub> liq. °K					
Cc	+		11 -					
Cryos. A'			c <sub>p</sub> vap. °K					İ
t <sub>e</sub> °C	421.77	5	c <sub>v</sub> vap.					
# for und	ercooled liquid			•		+ grams/100 gran	ns solven	t
	CES: 1-Dow	2-A1	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		AP	I					
PURIFICA		AP						
LITERATU	RE REFERE	1CES	S:					
l								
1								
i .								

No 173

								No. 17	<del></del>		
NAME	l-Tetracosene						STRUCTURAL FORMULA				
				CII (CII ) CII CII							
Mole % Pur.	Ref. Mo	lecula mula	C24H48	Molecular Weight 336.	624	C	:H <sub>3</sub> (СН <sub>2</sub> ) <sub>21</sub> СН	I=CH <sub>2</sub>			
		Ref		T	Ref.	Г			Ref.		
F.P. °C	45.3	2	dt/dP			f	to				
F.P. 1009	6		°C/mm			g	°K				
B.P. °C			0.1 mm BP	153.90 0. <b>0</b> 677	5 4	_h_	l				
760 mm 100	390. 302.20	2 5	t <sub>e</sub>	0.0355	5	f¹	to				
30	262.09	5	30 mm	1.0151	5	g'	°K		1		
10 1	231.06 178.73	5 5	ΔHm cal/g	<u> </u>		h'			+		
0.1	<b>13</b> 8, 61	5	ΔHv cal/g	65.04	5	n n	to •K				
Press. mn	n   169 <b>0.</b>	5	0.1 mm 30 mm	55.51	5	٥					
e Density	10,0.	<del>                                     </del>	BP	44.52 41.05	5	m'	to				
g/ml 20°0	0.8045	2	t <sub>e</sub> t <sub>e</sub> (d, e)	40.70	5	n' o'	*K				
dt 25 4 30	0.8011 0.7977	2 4	AHv/Te	19.69	5				<b> </b>		
a 30	0.1711	4	d 263 to	78.01	5		face tension es/cm. 20°C	27.62	5		
ъ	-0.0 <sub>3</sub> 68	4	d 431 °C	0.085 <b>9</b> 75.73	5	8 20	30	26.69	5		
Ref. Index	~ 1 4527F	2	e' 263 °C	0.0772	5		4.0	25.80	5		
<sup>n</sup> D 20°0	1.4506 1.4485	2	d <sub>c</sub> g/ml			Par	achor [P] 20°C				
30	1.4485	4	t <sub>c</sub> *C				30				
"C"	0.7449	4	P <sub>c</sub> mm				40 Sugd.	959.2	5		
MR (Obs. MR (Calc.	1 112 565	2 5	PV/RT			Exp	. L.1.%/wt.				
(nD-d/2)	1.0505	2	0.1 mm 30 mm	1.0000	5		u. persion	104. ≠	2		
Dielectric	2.11	5	BP	0.8933	5		sh Point °C	104.	+		
A 263 to		4	t <sub>e</sub>	0.8585	5		e Point				
B (481 °C	2601.2 165.	4 5	ΔHc kcal/m	<del> </del>	-		Spec.				
A*  263 to		5	ΔHf ΔFf				ra V. Ray Dif.				
B*[441_°C	2519.5	5	Viscosity				ared				
c	_		centistokes				ubility in <sup>†</sup> etone				
t <sub>k</sub> to			η °C			Ca	rbon tet.				
A'   to		$\vdash$					nzene her				
B'°	2		B. to		1	n-	Heptane				
A'* 140 to	2.29129	5	A l to				hanol ater				
B'* 263 °C		5	(B <sup>V</sup> )	1		W	ater in	ļ	4_		
Ac to			(A <sup>V</sup> )								
Bc tc °C	<u>-</u>		c <sub>p</sub> liq. *K								
Cryos. A		$\Box$	c <sub>p</sub> vap. °K								
consts, B		اــا	c <sub>v</sub> vap.								
t <sub>e</sub> °C	434.4 ercooled liquid	5	.vr.	<u> </u>	L	L		L	<u></u>		
			PI 3-Lit. 4-	Calc from de	+ 1-		Calc by for		nt		
SOURCE:	OEG: 1-DOW	API		Calc. Irom de	., Q8	- D	-Carc. by for				
PURIFICA	TION:	API		<del></del>							
	JRE REFERE										

							No. 174	
NAME	l-Penta	osen	e		_	STRUCTURAL :	FORMULA	4
			· · · · · · · · · · · · · · · · · · ·			СН <sub>3</sub> (СН <sub>2</sub> ) <sub>22</sub> СН	I=CH.	
Mole	Ref. Mo	lecul	ar C <sub>25</sub> H <sub>50</sub>	Molecular		3,011,2,2201	2	
% Pur.		Ref.		Weight 350.6	Ref			Ref.
F.P. *C	48.7	2	i	T	Kei		T	Ker.
F.P. 100%		ᆣ	dt/dP *C/mm	]		f to g 'K		1
B.P. °C	<u> </u>	<del> </del>	0.1 mm	156.49	5	h .	İ	1
760 mm	401.	2	BP	0.0686 0.0355	4 5	$\left  \frac{\ddot{\mathbf{f}}}{\mathbf{f}'} + \frac{\ddot{\mathbf{b}}}{\mathbf{t}_0} \right $		
100 30	311.91 271.18	5	t <sub>e</sub> 30 mm	1.0309	5	g'   °K	Ì	
10	239.66	5	ΔHm cal/g	1	+-	h'	ĺ	
1 0, 1	186.48 145.70	5	ΔHv cal/g	<del>                                     </del>	$\vdash$	m   to		
Press. mm		<del>  '</del> -	0.1 mm	63.53	5	n °K		1
t <sub>e</sub>	1717.3	5	30 mm	54.27	5 5	<u> </u>		
Density	+	1	BP t_	43.51 40.02	5	m'   to		
g/ml 20°C	0.8063 0.8030	2 2	te te (d, e)	39.69	5	n' K	1	
dt 25 4 30	0.7998	4	AHv/T <sub>e</sub>	19.48	5	<u> </u>		<u> </u>
	0,8196	4	d   272 to		5	Surface tension dynes/cm, 20°C	27,77	5
b	-0.0366	4	d'   443 to		5	30	26.87	5
Ref. Index	4		e' 272 °C		5	40	26.00	5
n <sub>D</sub> 20°C	1.4536 1.4515	2 2	d g/ml			Parachor [P]	}	1
30	1.4495	4	d g/ml vc ml/g tc °C			20°C 30	ļ	
"C"	0.7447	4	11 -			40		
MR (Obs.)	117.68	2	P <sub>c</sub> mm			Sugd.	998.2	5
MR (Calc.)	117.183	5	PV/RT 0.1 mm	1,0000	5	Exp. L.1.%/wt.		ł
(nD-d/2)	1.0505		30 mm	1.0000	5	Dispersion	104.	2
Dielectric	2.11	5	BP	0.8927	5	Flash Point °C		$\vdash$
A  272 to B  493 °C		4	te t <sub>c</sub>	0.8571	5	Fire Point		
č	163.5	5	ΔHc kcal/m		$\vdash$	M Spec.		
A*  272 to	2.40531	5	ΔHf			Ultra V. X-Ray Dif.		1
B* 453 °C	2570.2	5	ΔFf	· <del> </del> · · · · · · · · · · · · · · · · · · ·		Infrared		1
c V			Viscosity centistokes			Solubility in +		1
tk to	1		7 °C	: [		Acetone Carbon tet.		1
*x 1		_				Benzene		
A'   to B'   °C	1	1				Ether n-Heptane		
_c,	1	1	B <sup>v</sup> to			Ethanol		ļ
A'* 147 to			A <sup>V</sup> °C			Water Water in		
B'* 272 °C	2531.2	5	(B <sup>V</sup> )			Water III		
Ac to		1	(A <sup>V</sup> )	<u> </u>			1	
Cc L-c-	1	1	c <sub>p</sub> liq. ∘K	. [				1
Cryos, A° consts, B°			с <sub>р</sub> vaр. °К					
t <sub>e</sub> °C	446.99	5	c <sub>v</sub> vap.					
<pre># for unde:</pre>	rcooled liquid	Ĺ				grams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da			
SOURCE:		AP					_	
PURIFICAT	ION:	AP	I					
LITERATU	RE REFERE	NCES	5:					

							No. 17	5		
NAME	l-Hexaco	sene				STRUCTURAL	FORMUL	A		
					CH <sub>3</sub> (CH <sub>2</sub> ) <sub>23</sub> CH=CH <sub>2</sub>					
Mole % Pur.	Ref. Mol	ecul		Molecular Veight 364.6	76	3,01,2,2	3011-0112			
		Ref.	1		Ref.			Ref.		
F.P. °C F.P. 100%	51.8	2	dt/dP °C/mm			f to				
B.P. °C 760 mm 100	411. 320.74	2 5	0.1 mm BP t <sub>e</sub> 30 mm	158.69 0.0695 0.0356 1.0448	5 4 5	h   t				
30 10 1	279.47 247.53 193.62	5 5 5	ΔHm cal/g			h'				
0.1 Press.mm	152.26	5	ΔHv cal/g 0.1 mm	62.15	5	m t				
t e Density	1742.7	5	30 mm BP	53.07 42.50 38.98	5 5	m' t	0			
g/ml 20°C dt 25 4 30	0.8082 0.8048 0.8014	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	38.66 19.43	5	n' o' Surface tension	K			
a b	0.8218 -0.0 <sub>3</sub> 68	4	d 280 to e 454 °C d' 154 to	75.52 0.0804 73.02	5 5 5	dynes/cm. 20°0	27.92 26.99	5 <b>5</b> 5		
Ref. Index n <sub>D</sub> 20°C 25 30		2 2 4	e'   280 °C d g/ml v ml/g t °C	0.0714	5	Parachor [P] 20°( 30	26.09	3		
"C"	0.7443	4	P <sub>c</sub> mm			40 Sug	d. 1037.2	5		
MR (Obs.) MR (Calc. (nD-d/2) Dielectric		2 5 <b>2</b>	PV/RT 0.1 mm 30 mm BP	1.0000 1.0000 0.8924	5 <b>5</b> 5	Exp. L.1.%/wt u. Dispersion		2		
A   280 to B   504 °C	7.58699 2694.3 161.5	4 4 5	te tc	0.8561	5	Flash Point °C Fire Point M. Spec.				
A*  280 to B*  464 °C	2.42316 2611.2	5	ΔHc kcal/m ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared				
c t <sub>k</sub> to t <sub>x</sub> °C	-		Viscosity centistokes			Solubility in Acetone Carbon tet. Benzene				
A'   to B'   _ °C C'			B <sup>V</sup> to A <sup>V</sup> °C			Ether n-Heptane Ethanol Water				
A'* 154 to B'* 280 °C	2.33132 2570.7	5 <b>5</b>	(B <sup>V</sup> )			Water in		ļ		
Ac to Bc t <sub>c</sub> °C Cc	_		(A <sup>V</sup> )  c <sub>p</sub> liq. °K							
Cryos. A° consts. B°			c <sub>p</sub> vap. °K							
te °C	458.48 rcooled liquid	5	c <sub>v</sub> vap.			+ grows /100 =		<u></u>		
		2-4	PI 3_1.i+ 4 /	Calc from de	+ 4-	f grams/100 g ta 5-Calc. by f		ıt		
SOURCE:	225. 1-DOW	AP		oute, moni de	ua	J-Caic, by I	~u.a			
	TION:									
	PURIFICATION: API LITERATURE REFERENCES:									

							<b>No.</b> 176	
NAME	l-Heptaco	sene				STRUCTURAL :	FORMULA	
						<b></b>		
Mole	Ref. Mo		1	Molecular		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>24</sub> (	CH=CH <sub>2</sub>	
% Pur.	Fo Fo	rmul		Weight 378.7	02			
	_	Ref.			Ref.			Ref.
F.P. C F.P. 100%	54.7	2	dt/dP			f to		
B. P. °C	<u> </u>	-	*C/mm 0.1 mm	160.89	5	g °K		
760 mm	421.	2	BP t	0.0704	4			
100 30	329.58 287.77	5	t <sub>e</sub> 30 mm	0.0356 1.0587	5	g' to		1
10	255.40	5	ΔHm cal/g		1	h'		
1 0.1	200.76 158.83	5 5	ΔHv cal/g			m   to		
Press. mm			0.1 mm 30 mm	60.86	5	n °K		
t <sub>e</sub>	1767.5	5	BP	51.96 41.57	5 <b>5</b>	m'   to		$\vdash$
Density g/ml 20°C	0.8097	2	te te (d, e)	38.04 37.74	5	n' 'K		
dt 25 4 30	0.8064 0.8032	2	ΔHv/T	19.11	5	o' ¦		
a 30	0.8032	4	d   289 to	74.39	5	Surface tension dynes/cm. 20°C	39.05	_
ь	-0.0366	4	e   465 °C d'   161 to		5 <b>5</b>	30 gynes/cm. 20-C	28.05 27.14	5
Ref. Index	1 45537	2	e'   289 °C		5	40	26.26	5
D 25	1 1 4532	,	d <sub>c</sub> g/ml v <sub>c</sub> ml/g			Parachor [P] 20°C		
30	1,4513		tc °C			30 40		
"C"	0.7441	4	P <sub>c</sub> mm	ļ		<b>!</b> ]	1076.2	5
MR (Obs.) MR (Calc.	1 126 210	<b>2</b> 5	PV/RT		T .	Exp. L.1.%/wt.		
(nD-d/2)	1.0504	2	0.1 mm 30 mm	1.000 <b>0</b> 1.00 <b>00</b>	5 <b>5</b>	u. Dispersion	104.≠	2
Dielectric	2.12	5	BP t <sub>e</sub>	0.8919 0.8549	5	Flash Point °C		
A 289 to B <u> 515 °C</u>		4	tc	0.031)		Fire Point		
С	159.5	5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		
A* 289 to B* 475 °C		5	ΔFf			X-Ray Dif. Infrared		
к — — –	1	-	Viscosity			Solubility in +	<del> </del>	├
t <sub>k</sub>			rentistokes °C			Acetone		1
'x			∥'			Carbon tet. Benzene		ļ
A'   to B'   °C						Ether n-Heptane		
C'	<u> </u>		B <sup>V</sup> to A <sup>V</sup> C			Ethanol		
A'*161 to B'*289 °C		5 5	$\frac{A^{\mathbf{v}}}{(B^{\mathbf{v}})} - \frac{{}^{\mathbf{c}}}{-}$	-[		Water Water in		
Ac  to	+	<del>-</del>	(A <sup>V</sup> ) <sub>1</sub>					
Bc tc_C	<u>:</u>		c <sub>p</sub> liq. °K	<del> </del>	<del>                                     </del>			1
Cryos, A*	<del> </del>		11					İ
consts. B°			P					
t <sub>e</sub> °C	469.86	5	c <sub>v</sub> vap.	I				
	rcooled liquid					grams/100 gra		<u>t</u>
SOURCE:	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
PURIFICAT	rion.	AP AP						
	RE REFEREN							

NAME	1-Octaco	sene			- 1					
						ST	RUCTURAL	FORMUL	A.	
L i							CH <sub>3</sub> (CH <sub>2</sub> ) <sub>25</sub> CH=CH <sub>2</sub>			
Mole % Pur.		lecul mul		Molecular Veight 392.7	28		3 2 25			
		Ref.			Ref.				Ref.	
F.P. °C F.P. 1009	57.5	2	dt/dP °C/mm	1/2 0/		f g	to °K			
B. P. °C 760 mm 100 30	430. 337.52 295.19	2 5 5	0.1 mm BP t <sub>e</sub> 30 mm	163.06 0.0712 0.0356 1.0718	5 2 5 5	_h_ f' g'	to *K			
10 1 0.1	262.42 207.07 164.59	5 <b>5</b>	ΔHm cal/g ΔHv cal/g			h' m n	to °K			
Press. mm	1789.2	5	0.1 mm 30 mm BP	59.46 50.81 40.61	5 5	o m'				
Density g/ml 20°C dt 25 d4 30	0.8114 <sup>‡</sup> 0.8080 <sup>‡</sup> 0.8046 <sup>‡</sup>	2 2 4	te te (d,e) AHv/Te	37.06 36.82 19.32	<b>5</b> 5 5	n' o'	to °K			
a b Ref. Index	0.8250 -0.0 <sub>3</sub> 68	4	d 297 to e 475 °C d 168 to e' 297 °C	73.13 0.0756 70.37	5 5 <b>5</b>		face tension es/cm. 20°C 30 40	28.18 27.25 26.34	5 5 5	
n <sub>D</sub> 20°C 25 30	1.4560 1.4540 1.4519	2 2 4	e'   297 °C  d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.0663	5	Par	20°C 30 40			
"C"	0.7438	4	P <sub>c</sub> mm				Sugd.	1115.2	5	
MR (Obs.) MR (Calc. (nD-d/2)	131.337 1.0504 <sup>‡</sup>	2 5 2	PV/RT 0.1 mm 30 mm	1.0000 1.000 <b>0</b>	5 5	-	u. u. persion	103.≠	ż	
Dielectric  A 297 to  B 525 °C	7.60499 2780.18	5 4 4	BP t e t c	0.8912 0.85 <b>3</b> 5	5		sh Point °C e Point			
A*  297 to B*  485 °C	158.5 2.46441	5	ΔHc kcal/m ΔHf ΔFf			Ult:	Spec. ra V. Ray Dif. rared			
K to to			Viscosity centistokes 7°C			Soli Ac Ca	ubility in + etone arbon tet.			
A'   to B'   _ °C	<u> </u>		B <sup>V</sup>			Et n- Et	her Heptane hanol			
A'* 168 to B'* 297 °C Ac to		5	(B <sup>V</sup> )				ter in		-	
Bc tc C			c <sub>p</sub> liq. °K							
Cryos, A° consts, B°			c <sub>p</sub> vap. °K							
te °C	480.28	5	c vap.	l	L	L	4	<u> </u>	<u> </u>	
	rcooled liquid		DI 2 144 4 4	Cala 6 1			Cala by for		it	
SOURCE:	CES: 1-Dow			Calc. from de	t. Qa	ua 5	-Caic. by for	IIIUIA		
PURIFICA	TION:	API								
	RE REFERE							· · · · · · · · · · · · · · · · · · ·		

								No. 178	
NAME	1 -No	naco	sene				STRUCTURAL :	FORMULA	A.
							СН <sub>3</sub> (СН <sub>2</sub> ) <sub>26</sub> С	H=CH,	
Mole % Pur.	Ref.	Mo:	lecul rmul	ar C <sub>29</sub> H <sub>58</sub>	Molecular Weight 406.7	54	3 2 20	2	
			Ref.			Ref			Ref.
F. P. *C	60.0		2	dt/dP	T		f   to		
F.P. 100%	1			*C/mm			g °K		
B. P. °C				0.1 mm BP	165.14 0.0721	5 4	h ¦		1
760 mm 100	440. 346.3	7	5	t <sub>e</sub>	0.0357	5	f' + to		
30	303.5	1	5	30 mm	1.0853	5	g' K		Í
10 1	270.3		5	ΔHm cal/g			h'		
0. 1	171.2		5	AHv cal/g	1		m   to		
Press. mm				0.1 mm	58.43	5	n °K		
t <sub>e</sub>	1814.7		5	30 mm BP	49.87 39.82	5	<u> </u>		L
Density		, , , , #		t_	36.26	5	m'   to	ļ	ļ
g/ml 20°C	1 08	127 ‡ 094 ‡	2 2	e (=, =,	36.0	5	o' l		1
dt 25 4 30	0.8	062	4	ΔHv/T <sub>e</sub>	19.28	5			-
a	0.8		4	d   305 to		5	Surface tension dynes/cm. 20°C	28,30	5
Ъ	-0.0	366	4	d'   485 °C		5	<b>y</b> 30	27.39	5
Ref. Index	1	-/-#		e' 305 °C		5	40	26.50	5
n <sub>D</sub> 20°C	1.4	567 547 ±	2 2	d g/ml			Parachor [P] 20°C	1	İ
30	1.4	526 <sup>‡</sup>	4	∥ v <sub>c</sub> mı/g			30		
"C"	0.7	436	4	H C			40	1154 2	5
MR (Obs.)	136.2	3	2	P <sub>c</sub> mm	<u> </u>			1154.2	-
MR (Calc.) (nD-d/2)	135.9	55 504 <sup>≠</sup>	5	0.1 mm	1.0000	5	Exp. L.1.%/wt. u.	١.	
Dielectric			2	-30 mm	1.0000	5	Dispersion	103.	2
	2.17		5	BP t <sub>e</sub>	0.8910 0.8526	5	Flash Point °C		
A 305 to		0654	4	tc	0.0520		Fire Point		<u> </u>
c	156.		5	AHc kcal/m		m	M Spec.		
A*   305 to		7604	5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* 1495 °C	2732.0		5		<del>                                     </del>	├	Infrared		
·	1			Viscosity centistokes	1		Solubility in +		
tk to				7 °C	· ]		Acetone Carbon tet.		
t <sub>x</sub> c							Benzene		1
B' C	}				L		Ether n-Heptane		
C'	<u> </u>			B <sup>v</sup>   to			Ethanol	Ī	
A!#174 to		3008	5	AV   - °C	_		Water Water in		
B'*305 °C	2687.9		5	(B <sup>V</sup> )	1	Í	water in	<u> </u>	+
Ac to			ŀ	(A <sup>V</sup> )	<u> </u>	<u> </u>			
Cc L-c-	1			c <sub>p</sub> liq. °K		1			
Cryos, A° consts, B°				c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	491.82	:	5	c <sub>w</sub> vap.					
# for under	cooled lie	quid	<b></b>	<u> </u>	<u> </u>	1	grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow	2-AI	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for		
SOURCE:			AP						
PURIFICAT	ION:		AP	I .					
LITERATUI	RE REF	ERE							

								No. 17	9	
NAME	l-Triacontene						STRUCTURAL FORMULA			
						CH <sub>3</sub> (CH <sub>2</sub> ) <sub>27</sub> CH=CH <sub>2</sub>				
Mole % Pur.	Ref. Mo	ecul:		Molecular Weight 420.78	30		3. 2.27	2		
		Ref.			Ref.				Ref.	
F.P. °C F.P. 100	62.4	2	dt/dP °C/mm	.,,,	_	f g	to °K			
B.P. °C 760 mm 100 30 10	448. 353.43 310.14 276.61	2 5 5 5	0.1 mm BP t <sub>e</sub> 30 mm  AHm cal/g	166, 92 0, 0728 0, 0357 1, 0964	5 4 5 5	_h_ f' g' h'	to °K			
0.1 Press. mn	219.97 176.49	5 5	ΔHv cal/g 0,1 mm	57.20	5	m n o	to °K			
t <sub>e</sub> Density g/ml 20°0	1834.4	2	30 mm BP t <sub>e</sub> t <sub>e</sub> (d, e)	48.82 38.95 35.37 35.17	5 5 5 5	m' n'	to •K			
d <sub>4</sub> 25	0.8141 <sub>4</sub> 0.8107 <sub>4</sub> 0.8075 <sup>#</sup>	2 4	ΔHv/T <sub>e</sub>	19.22 71.04	5	Sur	face tension		-	
a b Ref. Index	0.8273 -0.0 <sub>3</sub> 66	4	e  495 °C to	0.0716 68.27	5 5	dyn 8	es/cm. 20°C 30 40	28.40 27.49 26.60	5 5	
n <sub>D</sub> 20°0 25 30		2 2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.0627	5	Par	20°C			
"C"	0.7433	4	P <sub>c</sub> mm				40 Sugd.	1193.2	5	
MR (Obs. MR (Calc. (nD-d/2) Dielectric	) 140.373 1.0503≠	2 5 2	PV/RT 0.1 mm 30 mm BP	1.00 <b>0</b> 0 1.0000 <b>0.</b> 8906	5 5 5	Dis	L.l.%/wt. u. persion	103.≠	2	
A 313 to B 545 °C	7.61184 2850.4	4 4	te tc	0.8517	5	Fir	sh Point °C e Point Spec.		-	
A* 313 to B* 505 °C		5 5 · 5	ΔHc kcal/m ΔHf ΔFf			Ult:	ra V. Ray Dif.			
t <sub>k</sub> t <sub>o</sub>	;		Viscosity centistokes 7°C			Ac Ca Be	ubility in + cetone rbon tet.			
B' °C	<u>:</u>	5	B <sup>V</sup> to A <sup>V</sup> °C			n- Et Wa	her Heptane hanol ater			
B'*313 °C	2720.4	5	(B <sup>V</sup> )  (A <sup>V</sup> )			Wa	ater in		+-	
Cryos. A			c <sub>p</sub> liq. °K c <sub>p</sub> vap. °K							
consts. B	501.03	5	c vap.							
	ercooled liquid		_ <del>v</del> -	<u> </u>	L	<del>ل _</del> _	rams/100 gra	me solve-	<u></u>	
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da		-Calc. by for			
SOURCE:		API								
PURIFICA	TION:	API								
LITERATU	JRE REFERE	NCES	5:							

<del></del>							······································	No. 18	0
NAME	1 - F	lentr	iacor	ntene			STRUCTURAL	FORMUL	<b>A</b>
Mole % Pur,	Ref.	Mo	lecul	ar C <sub>31</sub> H <sub>62</sub>	Molecular Weight 434.	806	СH <sub>3</sub> (СH <sub>2</sub> ) <sub>28</sub> С	H=CH <sub>2</sub>	
70 Fur.		FO	Ref	a 31 62	weight 454.	Ref		<del></del>	Ref
F.P. °C	64.6		2	44/45	1	I.C.	1 .	1	-
F.P. 100%			۲	dt/dP *C/mm			f to		l
B. P. °C	1			0.1 mm	168.97	5	h	1	
760 mm	457.		2 5	BP t <sub>e</sub>	0.0736 0.0357	4 5	<u>f'</u> + <u>to</u>		
100 30	361.30		5	30 mm	1.1092	5	g' K		
10	283.6		5	ΔHm cal/g			h'		ì
1 0. 1	182.3		5	ΔHv cal/g	<del> </del>	<del>                                     </del>	m to		
Press. mm	<del></del>		-	0.1 mm	56.12	5	n °K		1
t <sub>e</sub>	1857.6		5	30 mm BP	47.91 38.20	5	<u> </u>	L	<u> </u>
Density		#		t_	34.61	5	m'   to		ĺ
g/ml 20°C	0.81	153 <sup>‡</sup> , 120 ‡	2 2	e (a, e,	34.43	5	n' K		
d <sup>t</sup> 25 4 30	0.80	153° <sub>≠</sub> 120 <sup>‡</sup> 088 <sup>‡</sup>	4	ΔHv/T <sub>e</sub>	19.17	5			├
a	0.8		4	d 320 to		5	Surface tension dynes/cm. 20°C	28.51	5
<u>b</u>	-0.0	366	4	d'   504 °C		5 5	30	27,60	5
Ref. Index	1.4	580 <sup>#</sup>	2	e'   320 °C		5	40	26.71	5
n <sub>D</sub> 20°C			2	d <sub>c</sub> g/ml			Parachor [P] 20°C		
30	1.45	540 <sup>±</sup>	4	vc ml/g tc °C	1	l	30		l
"C"	0.74		4	P <sub>c</sub> mm		l	40 Sund	1232,2	5
MR (Obs.)			2	PV/RT	<del> </del>	₩	Exp. L.1.%/wt.	1232.2	13
MR (Calc.) (nD-d/2)	144.99	91 503	5 2	0.1 mm	1.0000	5	u.	١,	
Dielectric	2.12		5	30 mm BP	1.0 <b>000</b> 0.8905	5	Dispersion	103. <sup>‡</sup>	2
A 320 to			4	te	0.8510	5	Flash Point °C	1	1
B [554 °C	2890.4		4	t <sub>c</sub>			Fire Point		╁
C	153.		5	ΔHc kcal/m ΔHf			M Spec. Ultra V.		1
A*   320 to B*   514 °C		0893	5	ΔFf		l	X-Ray Dif.		1
к 🗀 =	-			Viscosity		1	Infrared Solubility in +		<u> </u>
£ to	-			centistokes		ŀ	Solubility in +		
t <sub>k</sub> to	1			η •ο	1		Carbon tet.		
A' to	<b>†</b>		$\vdash$				Benzene Ether		
B' °	.[			B <sup>V</sup>   to	<del> </del>	-	n-Heptane		
	<del> </del>		<del> </del>	B to		1	Ethanol Water		1
A'* 187 to B'* 320 °C		1119	5	(BV)	-		Water in		
Acl to	<u> </u>		$\vdash$	(A <sup>V</sup> )					
Bcit C				c <sub>p</sub> liq. °K	-	+	1		
Ce	<del> </del> -			_		1			
Cryos. A° consts. B°	ļ			c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	511.43		5	c <sub>v</sub> vap.		<u> </u>	I .		
# for unde							f grams/100 gra	ms solven	t
REFERENC	ES: 1-D	OW	2-AF		Calc, from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT			AP						
LITERATU	RE REFI	EREN	ICES	<b>:</b>					

								No.181	
NAME	l-Dotri	acon	tene			ST	RUCTURAL	FORMUL	A
							СН <sub>3</sub> (СН <sub>2</sub> ) <sub>29</sub> С	I=CH_	
Mole % Pur.	Ref. Mo	lecul mula	ar C <sub>32</sub> H <sub>64</sub>	Molecular Weight 448,83	32		3, 5 <u>2</u> , 29	2	
		Ref.			Ref.				Ref.
F. P. °C	66.7	2	dt/dP	ł		f	to		
F.P. 100%			*C/mm 0.1 mm	170.75	5	g	•к		i
B.P. °C 760 mm	465.	2	BP	0.0743	4	h_	<del>!</del>		l
100	368.45	5	t <sub>e</sub>	0.0357 1.1203	5	f' g'	to °K		
30 10	324.22 289.96	5 <b>5</b>	30 mm	1.1203		h'			
1	232.06	5	ΔHm cal/g		$\vdash$	m	l to		<del>                                     </del>
0.1	187.58	5	ΔHv cal/g 0,1 mm	55.04	5	n	°K		
Press. mm	1879.8	5	30 mm	46.98	5	°	i		
Density			BP t <sub>e</sub>	37.47 33.87	5	m'	to		
g/ml 20°C	0.8165 0.8132	2	t <sub>e</sub> (d, e)	33.70	5	n' o'	•K		
dt 25	0.8132 <sub>7</sub>	2 4	ΔHv/T <sub>e</sub>	19.14	5		<u> </u>		
a	0,8298	4	d 327 to		5		face tension es/cm. 20°C	28.61	5
ь	<b>-0.</b> 0366	4	d 192 to		5	8,	30	27.70	5
Ref. Index	1 4505#	١. ا	e' 327 °C		5	<u> </u>	40	26, 81	5
<sup>n</sup> D 20°C	1.4585	2 2	d <sub>c</sub> g/ml			Par	rachor [P] 20°C		
30	1.4546	4	vc ml/g tc °C				30		
"C"	0.7429	4	P <sub>c</sub> mm				40 Sugd.	1271.2	5
MR (Obs.) MR (Calc.)	150.13 149.609	2 5	PV/RT	·	$\vdash$	Ext	. L.1.%/wt.		-
(nD-d/2)	1.0503	2	0.1 mm	1.0000	5		u.	10 <b>3</b> .≠	
Dielectric	2.13	5	30 mm BP	1.0000 0.8910	5		persion	103.	2
A 327 to	7.62424	4	t e t c	0.8511	5		sh Point °C e Point		
B 1563 ℃ C	2924.3 151.5	4 5	ΔHc kcal/m		$\vdash$	M.	Spec.		
A*  327 to	2,52222	5	ΔHc Kcal/Hi			Ult	ra V.		
B*(523 °C	2837.3	5	ΔFf				Ray Dif. rared		l
K c			Viscosity centistokes		1	Sol	ubility in +		T
t <sub>k</sub> Tto			η °C				etone rbon tet.		
1x			•				enzene		
A'   to B'   °C	1	١.,					her		į
č, ' =			B <sup>V</sup>   to A <sup>V</sup>   °C				Heptane hanol		ļ
A!* 192 to	2.42526	5		_			ater ater in		
B'* 327 °C	2791.2	5	(B <sup>V</sup> )			<del></del>	ater in		-
Ac to Bc tc °C	1		(A <sup>V</sup> )	<b>_</b>	$\sqcup$				
Co			c <sub>p</sub> liq. °K						
Cryos, A° consts, B°			c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	520.74	5	c <sub>w</sub> vap.						
	rcooled liquid	<u> </u>	ш		لــــــا	+ g:	rams/100 gra	ms solver	ıt
REFERENC		2-A	PI 3-Lit. 4	-Calc. from de	t. da				
SOURCE:		AP							
PURIFICAT	ION:	AP							
	RE REFERE								

F.P. °C 68.7 2 F.P. 100%  B.P. °C 760 mm 473. 2 100 375.52 5 30 330.85 5 10 296.24 5 1 237.75 5 0.1 192.82 5  Press. mm t t e Density g/ml 20°C 0.8176 2 dt 25 0.8143 2 dt 25 0.8181 4  a 0.8309 4 b -0.0366 4 b -0.0366 4  Ref. Index nD 20°C 1.4591 2 25 1.4571 2 30 1.4550 4  MR (Obs.) 154.27 3 MR (Calc.) 154.227 3 MR (Calc.) 154.227 3 MR (Calc.) 154.227 3 MR (Calc.) 154.227 3 MR (Calc.) 154.227 3 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Dos.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Calc.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 154.227 4 MR (Dos.) 15								No. 182	
Molecular   Formula   C 3 H 66   Molecular   Fig.   Molecular   Fig.   Molecular   Fig.   Molecular   Fig.   Molecular   Fig.   Molecular   Fig.   Molecular   Fig.   Fig.   Molecular   Fig.   Fig.   Molecular   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.   Fig.	NAME	l - Tritria	c ont	ene		ı	STRUCTURAL	FORMULA	
Mole									
Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.				T		一	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>30</sub> CF	I=CH <sub>2</sub>	
Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.   Ref.		Ref. Mo	lecul	ar C33H66	Molecular	858			
F. P. *C	- A 1 U.				weight 102.		<del></del>		Ref.
F. P.   100%	FP •C	68.7			T	1		г	-
B. P. 'C   760 mm   473.   2   375.52   5   10   375.52   5   10   276.24   5   10   277.75   10   10   10   10   10   10   10   1							1 1 111		
The control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the control of the	B.P. °C			0.1 mm					İ
100	760 mm			11 .					
10					1		1 1		İ
192.82   5	10	296.24	5				h'	1	
Press, mm t t					<del> </del>	-			
The content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the	Press mm		-	0.1 mm	54.03		l i		ĺ
Density g/ml 20°C   0.8176   2   2   4   25   0.8143   2   2   4   30   0.8117   4   4   25   0.8117   4   4   4   4   25   0.8117   4   4   4   4   25   0.8117   4   4   4   4   20.49   5	t		5				•		<u> </u>
Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution   Solution	Density			l t					İ
A				e (4, 6)		5			
A	d4 30			ΔHv/T <sub>e</sub>	20.49	5			ļ
Ref. Index   nD   20°C   1.4591   2   2.5   1.4591   2   2.5   30   1.4596   4   4   4   4   50°   4   4   4   50°   4   4   4   50°   4   4   4   50°   4   4   4   50°   4   4   50°   4   4   50°   4   4   50°   4   4   50°   4   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   4   50°   5   5   50°   5   5   5   5   5   5   5   5   5		0.8309	4					28.69	5
Ref. Index nD 20°C   1.4591	Ъ	-0.0366	4		-1		<b>y</b> 30	27.77	5
1.4571								26.88	5
"C"   0.7428   4   V <sub>c</sub>   C   C   V <sub>c</sub>   MR (Cobs.)   154.79   2   1.0503   2   2   1.0503   2   2   2   1.0503   2   2   2   3.0   3.0   4.0   3.0   3.0   4.0   3.0   3.0   4.0   3.0   3.0   4.0   3.0   3.0   4.0   3.0   3.0   4.0   3.0   3.0   3.0   3.0   3.0   4.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0   3.0	n <sub>D</sub> 20°C			d <sub>c</sub> g/ml				1 1	
MR (Obs.)   154.79   2   PV/RT   1.0000   5				V <sub>C</sub> m1/g	1				
MR (Calc.) (nD-d/2)         134.27 / 1.0503 f 2 / 2.7 / 1.0503 f 2 / 2.050 mm         PV/RT	"C"	0.7428	4	C -					۱ ـ
Non-d/2								1310.2	5
Dielectric   2.13   5   30 mm   BP   0.8930   5   5   5   5   5   5   5   5   5		) 154.227 1.0503	5	A	1.0000	5		1 1	i
A   334 to   7.62921   4   2958.3   150.   5   A+   2958.3   150.   5   A+   334 to   B+   532 °C   K   C   C   C   C   C   C   C   C				30 mm	1.0000	5		103. ₹	2
B   572°C   2958.3   4   t <sub>c</sub>		+		11					
A*   334 to   2.53183   5   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f   A+f				t.		ľ			
A*   334 to			5	∆Hc kcal/m					
Nitrared   Solubility in   Acetone   Carbon tet.   Benzene   Ether   n-Heptane   Ethanol   Water   Water in	A*  334 to	2,53183					X-Ray Dif.		
C	K L322	2868.4	5	<b> </b>	<del>                                     </del>	$\vdash$			
Carbon tet.   Benzene   Ether   n-Heptane   Ethanol   Water in				centistokes	1		li corability in		
A'   to B'   - C' C'	1k   10			<b>7</b> °C					
B'   C'   B'   to   A'   198 to   2.43886   5   B'   to   A'   0°C     (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )   C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )   C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )   C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )   C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )     C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A' )   C   (A'	A' I to								
A'* 198 to	B' •C		İ	L	<b>_</b>	<b>-</b>			ì
B1* 334 °C   2823.7   5     BV				B to					
Ac   to   Co   Co   Co   Co   Co   Co   Co				<u> </u>	-[				
C <sub>C</sub>   C <sub>C</sub>   C <sub>C</sub>   C <sub>C</sub>   C <sub>D</sub>   C <sub>D</sub>   liq. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   C <sub>D</sub>   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap. °K   vap.		<del></del>	,	11	1				Г
Cryos. A° consts. B° cp vap. °K custs. B° cp vap. °K cv vap. cp vap. °K cryos. Consts. B° cp vap. °K cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. cp vap. c	Bc t °C				<b>+</b>	$\vdash$			
consts. B°         P         P           te °C         530.21         5         cv vap.           # for undercooled liquid         grams/100 grams solvent           REFERENCES: 1-Dow         2-API         3-Lit.         4-Calc. from det. data         5-Calc. by formula           SOURCE:         API		<del> </del>		(I -	1				
te °C 530.21 5 Cv vap. grams/100 grams solvent  # for undercooled liquid  REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det, data 5-Calc, by formula  SOURCE: API  PURIFICATION: API		1		c <sub>p</sub> vap. °K	1				
# for undercooled liquid  REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det, data 5-Calc, by formula  SOURCE: API  PURIFICATION: API		520 21	-	c. vap.					1
REFERENCES: 1-Dow 2-API 3-Lit, 4-Calc, from det. data 5-Calc, by formula  SOURCE: API  PURIFICATION: API			١,	L • ·	L	L	II.		<u> </u>
SOURCE: API PURIFICATION: API			2 4 7	OT 2 T24 4	Colo 6 1		grams/100 gran	ns solvent	<u> </u>
PURIFICATION: API		) - NOW			Caic. Irom de	L. da	ua 5-Caic, by for	mula	
		TON.							
MILARIURE REFERENCES:									
	MIEKATU	RE REFERE	NCES	<b>5:</b>					

								No. 183	3
NAME	1-Teti	atri	acontene		_	ST	RUCTURAL	FORMUL.	A.
				<del></del>			רוו (רוו ) רו	H-CH	
Mole % Pur.		lecul mul		Molecular Weight 476.88	84	·	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>31</sub> С	1-0112	
		Ref.			Ref.				Ref
F.P. ℃	70.5	2	dt/dP			f	to		
F.P. 100%	<del> </del>		*C/mm 0,1 mm	174.32	5	g	°K		
B. P. °C 760 mm	481.	2	ВР	0.0757	4	_ <u>_</u> ,_			
100	382.59	5	t <sub>e</sub>	0.0358	5 5	f' g'	to °K		İ
30 10	337.48 302.53	5	30 mm	1.1427	"	h'			
1	243.45	5 5	ΔHm cal/g ΔHv cal/g	<del></del>	+	m	to		
0.1 Press.mm	198.06	) 5	0.1 mm	53.08	5	n o	°K		
t <sub>e</sub>	1917.3	5	30 mm BP	45.30 36.03	5				
Density	0 010/#		t_	32.40	5	m' n'	to °K		
g/ml 20°C	0.8186 0.8153 0.8121	2 2	'e (u, e)	32.29	5	ō'	- 1		
d <sup>t</sup> 25 4 30	0.8121	4	ΔHv/T <sub>e</sub>	19.02	5	Sur	face tension		
<u>a</u>	0.8319 -0.0366	4 4	d 341 to e 531 °C	67.11 0.0646	5 5	dyn	es/cm. 20°C	28.78	5
Ref. Index	-0.0300	-	d 204 to	64.13	5	, a	30 40	27.87 26.97	5
n <sub>D</sub> 20°C	1.4596	2	e'   341 °C	0.0558	5	Par	achor [P]		
D 25 30	1.4596 1.4576 1.4556	2	dcg/ml vcml/g				20°C		ĺ
"C"		4	v <sub>c</sub> ml/g t <sub>c</sub> °C				30 <b>40</b>		
MR (Obs.)	0.7425	2	P <sub>c</sub> mm					1349.2	5
MR (Calc.)	150 045	5	PV/RT	1 2222		Exp	. L.1.%/wt.		
(nD-d/2)	1.0503	2	0.1 mm 30 mm	1.0000	5	Dis	u. persion	103. <sup>‡</sup>	2
Dielectric	2.13	5	BP	0.8895	5	Fla	sh Point °C		
A 341 to B  581 °C	7.63409 2992.2	4	te t	0.8484	"		e Point		
с	148.5	5	∆Hc kcal/m				Spec. ra V.		
A* 341 to B* 541 °C	2552.8 2905.2	5	ΔHf ΔFf			X-I	Ray Dif.		
K STI	2705.2	] "	Viscosity				ared		
t, to			centistokes				ibility in <sup>†</sup> eton <b>e</b>		
t <sub>k</sub> to t <sub>x</sub> °C	ļ		η °C				rbon tet.		
A'   to							her		
B'  °C			B, to	<del> </del>	$\dagger$		Heptane hanol		
A!* 204 to	2.45201	5.	A I °C			w.	ater		
B'* 341 °C	2856.2	5	(B <sup>V</sup> )	-		W	ter in		_
Acl to Bc te °C			(A <sup>V</sup> )	1	$\perp$				
Cc re-re	-		c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	539.14	5	c <sub>v</sub> vap.						
	rcooled liquid					+ g:	ams/100 gra	ms solven	t
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:		API							
PURIFICAT		API							
LITERATU	RE REFERE	NCE	S:						

							No. 184			
NAME	1-Pentat	riac	ontene			STRUCTURAL I	FORMULA			
Mole	Ref. Mo	laani		Molecular		СН <sub>3</sub> (СН <sub>2</sub> ) <sub>32</sub>	CH=CH <sub>2</sub>			
% Pur.	Fo	rmul	ar C <sub>35</sub> H <sub>70</sub>	Weight 490.	10					
		Ref.			Ref.			Ref.		
F.P. *C	72.3	2	dt/dP			f to				
F.P. 100%	<b></b>	<b>├</b>	*C/mm 0.1 mm	176.10	5	g °K				
B.P. °C 760 mm	489.	2	BP	0.0764	4					
100	389.65	5	t <sub>e</sub>	0.0358 1.1538	5	f' to g' °K				
30 10	344.11 308.82	5	30 mm	1.1536	1-	h'				
1	249.15 203.29	5	ΔHm cal/g		<del>  -  </del>	m to		_		
0.1 Press.mm		1	ΔHv cal/g 0.1 mm	52, 18	5	n 'K				
t <sub>e</sub>	1937.2	5	30 mm BP	44.53	5	<u>° i</u>				
Density	4		t_	35.38 31.74	5 5	m'   to				
g/ml 20°C	0 81637	2	i e (u, e,	31.65	5	o'				
dt 25 4 30	0.8131≠	4	ΔHv/T <sub>e</sub>	18.89	5	Surface tension		├		
a	0.8329	4	d   347 to e   539 °C		5	dynes/cm. 20°C	28.87	5		
ь	-0.0366	4	a 709 to	63,22	5	30 40	27.95 27.05	5 5		
Ref. Index		2	e'   347 °C	0.0543	5	Parachor [P]	21.03	-		
25	1.45817	2	d g/ml vc ml/g			20°C		ŀ		
30	1,4500	4	tc °C	ļ	İ	30 40				
"C"	0.7425	4	P <sub>c</sub> mm				1388.2	5		
MR (Obs.) MR (Calc.	1 163 463	2 5	PV/RT	T		Exp. L.1.%/wt.				
(nD-d/2)	1.0503	2	0,1 mm 30 mm	1.0000	5	u. Dispersion	102.≠	2		
Dielectric	2.13	5	BP	0.8892	5	Flash Point °C	102.	<del>-</del>		
A 347 to		4	t <sub>e</sub> t <sub>c</sub>	0.8475	5	Fire Point				
c	147.	5	ΔHc kcal/m		T	M Spec.				
A*   347 to		5	ΔHf ΔFf			Ultra V. X-Ray Dif.				
B* _549°C	2938.7	5	Viscosity		┼─	Infrared				
c	_]	ŀ	centistokes	İ		Solubility in + Acetone				
t <sub>k</sub> to			η ∘c			Carbon tet.				
A' to				1		Benzene Ether	i			
B' _ °C	<u> </u>		F-V 1	<del> </del>	<b>├</b> ─	n-Heptane				
C'	2 4/455	<del>  </del>	B <sup>V</sup>   to			Ethanol Water				
A'* 209 to B'* 347 °C		5	(BV)	-	1	Water in		<u> </u>		
Ac to		i	(A <sup>V</sup> )		1					
Bc tc_C	<u>-</u>		c <sub>p</sub> liq. °K		<b>†</b>	1				
Cryos. A	<b>†</b>	$\vdash$	ii -				1			
consts. B			р -							
t <sub>e</sub> °C	548.39	5	c <sub>v</sub> vap.							
# for undercooled liquid # grams/100 grams solvent										
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula										
SOURCE:	PION.	API								
	RE REFERE	API NCE								
			<del></del>							

No.185 l-Hexatriacontene NAME STRUCTURAL FORMULA  $CH_3(CH_2)_{33}CH=CH_2$ Molecular C36H72 Mole Ref Molecular Weight 504.936 % Pur Ref. Ref Ref. F.P. \*C F.P. 100% 73.9 2 dt/dP f to °C/mm °K g 177.73 0.1 mm 5 B. P. °C h BP 0.077 2 496. 760 mm 2 t<sub>e</sub> 0.0358 5 f 100 395.83 5 ۰ĸ 1.1638 5 g' 349.90 30 mm 5 30 10 314.30 5 h' ∆Hm cal/g 227, 27 5 1 m to AHv cal/g 0.1 207.81 5 n ۰ĸ 51.22 43.73 0.1 mm 5 5 Press. mm 0 30 mm 1957.8 5 34.77 BP 5 m' Density to 31.12 5 0.8205<sup>†</sup> n' g/ml 20°C te (d, e) 2 31.06 5 ٥'  $\mathbf{d_{4}^{t}}$ 25 0.8172 2 ΔHv/T<sub>e</sub> 20.46 5 0.8140 30 Surface tension d 353 65.18 5 a 0.8337 4 28.93 1 546 dynes/cm, 20°C °C 0.0613 b -0.0366 5 28.01 ă 214 30 62.18 5 40 27.11 5 Ref. Index e' | 353 0.0527 5 1.4605  $\mathbf{n}_{\mathbf{D}}$ 20°C [P] Parachor d<sub>c</sub> g/ml 25 1.4585 2 20°C 1.4566 vc ml/g t\_°C 30 4 30 <sup>t</sup>c "C" 40 0.7423 4  $P_c$  mm Sugd. 1427.2 5 MR (Obs.) 168.71 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 168.081 5 0.1 mm 1,0000 1.0503<sup>#</sup> (nD-d/2)2 30 mm 1,0000 5 102. Dispersion 2 Dielectric 2.13 5 BP 0.8900 5 Flash Point °C A 353 to 0.8480 7.64540 4 Fire Point 3058.9 B 596 °C M. Spec. C 146. 5 AHc kcal/m Ultra V. ΔHf A\* 353 to 2.57957 5 X-Ray Dif. ΔFf B\*| 556 °C 2969.8 Infrared K Viscosity Solubility in c centistokes Acetone to Carbon tet. t<sub>x</sub> °C Benzene A' to Ether B' °C n-Heptane B<sub>v</sub> | C to Ethanol ۰c Water A'\* 214 to 2.47928 5 Water in B'\* 353 °C (B<sup>V</sup>) 2920.1 Acl to (A<sup>V</sup>)| Bc °C c<sub>p</sub> liq. °К Cc Cryos. A. cp vap. ٩K consts. B c, vap. te °C 556.60 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							No. 186	
NAME	l -Hepta	<b>tria</b> c	ontene			STRUCTURAL	FORMULA	<b>L</b>
			-			: ·		
Mole	Back Ma	11		Molecular		сн <sub>3</sub> (сн <sub>2</sub> ) <sub>34</sub> (	CH=CH <sub>2</sub>	
% Pur.	Ref. Mo	rmul	ar C 37H74	Weight 518.	962			
		Ref.			Ref			Ref.
F.P. °C	75.5	2	dt/dP			f to		
F. P. 1009			*C/mm 0.1 mm	179.12	5	g °K		
B.P. °C 760 mm	503.	2	BP	0.0776	4	<u>+ </u>	1	
100	402.03	5	t <sub>e</sub>	0.0359	5	f' to		
30 10	355.74 319.85	5	30 mm	1,1731	5	h'	1	
1	259.17	5	ΔHm cal/g	<b></b>	<u> </u>	m to	<del> </del>	ļ
0.1 Press. mn	212.53	5	ΔHv cal/g 0.1 mm	50,43	5	n °K	1	
t <sub>e</sub>	1972.4	5	30 mm	43.01	5.	°	Ĺ	<u> </u>
Density		┢	BP t <sub>e</sub>	34.11 30.45	5	m'   to		
g/ml 20°0	0.8214	2	t <sub>e</sub> (d, e)	30.4	5	n' °K	1	1
dt 25 4 30	0.8181 0.8189 0.8149	2	AHv/T <sub>e</sub>	20.46	5		<u> </u>	├
8	0.8347	4	d   359 to		5	Surface tension dynes/cm, 20°C	29.02	5
<u>ь</u>	-0.0366	4	a   554 °C		5	30	28.10	5
Ref. Index	1 4610	2	e' 359 °C		5	40	27.20	5
<sup>n</sup> D 20°C	1 1 45007	2	d <sub>c</sub> g/ml			Parachor [P] 20°C		1
30	1.4570	4	tc °C			30	İ	
"C"	0.7421	4_	P <sub>c</sub> mm			40 Sugd	1466.2	5
MR (Obs.) MR (Calc.	1 172 400	2 5	PV/RT		1	Exp. L.1.%/wt.	1	
(nD-d/2)	1.0502	2	0.1 mm 30 mm	1.0000	5	u.	102. #	
Dielectric		5	BP	1.000 <b>0</b> 0.8888	5	Dispersion Flash Point °C	102.	2
A 359 to B   604 °C	7.64418 3081.9	4	t <sub>e</sub> t <sub>c</sub>	0.8462	5	Fire Point		
C POT	144.	<b>4</b> 5	ΔHc kcal/m	<del> </del>	$\vdash$	M Spec.		
A* 359 to		5	ΔHf			Ultra V. X-Ray Dif.	]	
B* 564 °C	2993.7	5	ΔFf	<del> </del>	-	Infrared		
c			Viscosity centistokes	1		Solubility in +		
tk to			η °c			Acetone Carbon tet.		İ
t'x i °C		-		}		Benzene Ether		ļ
B' _ °C			ļ	ļ	-	n-Heptane	İ	l
C'		<u> </u>	B <sup>V</sup>   to A <sup>V</sup>   °C			Ethanol Water		
A'* 219 to B'* 359 *(		5	(BV)	-		Water in		
Acl to		Ť	(A <sup>V</sup> )	1				
Bc tc *C			c <sub>p</sub> liq. °K	<del> </del>				
Cryos. A	<del></del>	$\vdash$	1) -					
consts. B		<u>_</u>	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	564.62	5	c <sub>v</sub> vap.					L
	ercooled liquid					f grams/100 gra	ms solven	
	CES: 1-Dow	2-AI		Calc, from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP						
PURIFICA		AP						
LITERATU	RE REFERE	VCES	5:					

No.187 l-Octatriacontene NAME STRUCTURAL FORMULA CH3(CH2)35CH=CH2 Molecular C38H76 Ref. Mole Molecular Weight 532,988 % Pur. Ref. Ref. Ref. F.P. °C 77.0 2 dt/dP f to F.P. 100% °C/mm g °K 0.1 mm 180.76 B. P. °C h 0.0782 BP 4 510. 760 mm 2 t<sub>e</sub> 0.0359 5 ſ١ to °K 100 408.21 5 5 5 g' 1.1831 30 mm 30 361.52 h١ 10 325.33 5 ∆Hm cal/g 1 264.11 5 m to ∆Hv cal/g 0.1 217.05 5 n ۰ĸ 0.1 mm 49.56 5 Press. mm o 30 mm 42.29 <sup>t</sup>e 1990.9 5 ВP 33.54 5 m' to Density 5 te (d, e) 29.86 n' g/ml 20°C 0.8223 5 29.81 0.8190 o'  $\mathbf{d_{4}^{t}}$ 25 2 AHV/T 20.36 5 0.8158 30 Surface tension 365 to 63.59 5 0.8356 4 dynes/cm. 20°C 29.10 5 ٠c e 562 d' 224 0.0589 60.50 -0.0366 ь 4 30 28.17 to 40 27.27 5 Ref. Index 365 0.0504 5 e' °C 1.4614 20°C [P] nD 1.4594 Parachor d<sub>c</sub> g/ml 25 2 20°C 1.4574 vc ml/g 30 4 30  $t_c$ ٠c 0.7419 40 "C" 4 1505.2 5 P<sub>c</sub> mm Sugd MR (Obs.) 177.99 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 177.317 5 1.0502<sup>#</sup> 0.1 mm 1.0000 5 (nD-d/2)2 102. 30 mm 1.0000 2 Dispersion Dielectric 2.13 5 ΒP 0.8889 Flash Point °C A 365 to B 612 °C 0.8459 7,65055 4 ţe. Fire Point 3114.6 4 M. Spec. C 143. 5 AHc kcal/m Ultra V. ΔHf A\*| 365 to 2.60324 5 X-Ray Dif. ΔFf B\* 572 °C 3025.6 Infrared Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C Benzene A۱ to Ether B١ °C n-Heptane Bv | Av | C' to Ethanol °C Water 2,50025 A! # 224 to 5 Water in (B<sup>V</sup>)I B'\* 365 °C 2973.6 Acl (A<sup>V</sup>)| to Bc °C c<sub>p</sub> liq. °K Cc Cryos. A° °K c<sub>p</sub> vap. consts. B° c<sub>v</sub> vap. te °C 572.76 5 # for undercooled liquid grams/100 grams solvent 5-Calc. by formula REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data SOURCE: API **PURIFICATION:** API LITERATURE REFERENCES:

							No. 188	
NAME	l-Nonatr	iacor	ntene			STRUCTURAL 1	FORMULA	<b>.</b>
						CH (CH )	CU-CU	
Mole % Pur.	Ref. Mo	lecul rmul	ar C <sub>39</sub> H <sub>78</sub>	Molecular Weight 547.	014	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>36</sub>	CH=CH <sub>2</sub>	
		Ref.			Ref	<u> </u>		Ref.
F. P. *C	78.4	2	dt/dP	Γ		f   to		
F.P. 100%			*C/mm			f to g °K		
B. P. °C			0.1 mm BP	182.40	5 2	h		
760 mm 100	517. 414.39	2 5	t	0.0359	5	f' + to		
30	367.31	5	30 mm	1.1931	5	g' °K		
10	330.81	5	ΔHm cal/g			h'		
1 0. 1	269.06 221.57	5	ΔHv cal/g		$\vdash$	m   to		
Press. mm		-	0.1 mm	48.75	5	n °K		1
t <sub>e</sub>	2006.7	5	30 mm BP	41.60 32.96	5			
Density	1		t	29.27	5	m' l to		İ
g/ml 20°C	0.82317	2	e (-, -,	29. 25	5	n' 'K		1
d <sup>t</sup> 25	0.8198 0.8198 0.8166	2	AHv/T <sub>e</sub>	20.35	5			<u> </u>
8	0.8364	4	d   370 to	62.82	5	Surface tension dynes/cm, 20°C	29.16	5
ъ	-0.0366	4	d 1 569 to		5	30	28, 24	5
Ref. Index	#		e' 370 °C		5	40	27.34	5
n <sub>D</sub> 20°C	1.4618 <sup>‡</sup> 1.4598 <sup>‡</sup>	2	d_g/ml	1		Parachor [P]		
30	1.4618	4	d g/ml vc ml/g			20°C 30		1
"C"	0,7418	4	1c			40		
MR (Obs.)	182,63	2	P <sub>c</sub> mm			<del></del>	1544.2	5
MR (Calc.)	181 035	5	PV/RT 0.1 mm	1,0000	5	Exp. L.1.%/wt.		
(nD-d/2)	1.0502	2	30 mm	1,0000	5	u. Dispersion	102. #	2
Dielectric	2.14	5	BP	0.8881	5	Flash Point °C	<u> </u>	<del>                                     </del>
A 370 to		4 5	te t <sub>c</sub>	0.8446	5	Fire Point M Spec.		_
A* 370 to	2,61874	5	ΔHc kcal/m ΔHf			Ultra V.	ļ	
B* 579 °C		5	ΔFf	<u> </u>		X-Ray Dif. Infrared		
С	1		Viscosity centistokes	ļ		Solubility in +		
t <sub>x</sub> to			η ∘c			Acetone Carbon tet.		
A' to		$\vdash$			1	Benzene Ether		
B', ∟ _ °⊆			B <sup>V</sup>   to		+-	n-Heptane Ethanol	1	
A'+228 to	2,51369	5	AV I C	1		Water	i	l
B'*370 °C		5	(BV)			Water in	<b></b>	<u> </u>
Ac to			(A <sup>V</sup> )	<u> </u>	_			ŀ
Co C -	<b> </b>		c <sub>p</sub> liq. ∘K					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	580.82	5	c <sub>v</sub> vap.	1	<u> </u>	L	<u> </u>	<u> </u>
# for unde	rcooled liquid					grams/100 gram		t
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:		API						
PURIFICAT		API						
MILKAIU	RE REFERE	NCES	<b>:</b>					

No. 189

							No. 18	9
NAME	1-Tet	raconter	ne			STRUCTURAL	FORMUL	.A.
	1T	<del></del>			$\neg$	СН <sub>3</sub> (СН <sub>2</sub> ) <sub>37</sub>	CH=CH <sub>2</sub>	
Mole % Pur.	Ref.	Molecul		Molecular Weight 561.	040			1
<del></del>		Ref.		Weight 501.	Ref.	<del></del>		Ref.
F.P. °C	79.8	2	dt/dP		Xe.			1
F.P. 1009		-+-	*C/mm		1 1	f to g °K		
B. P. °C			0.1 mm	183.76	5	h i	1	
760 mm	523.	2	BP t	0.0794 0.0359	4 5	f' to		
100 30	419.69 372.28	5	t <sub>e</sub> 30 mm	1,2016	5	g' K		
10	335.51	5	ΔHm cal/g		$\vdash$	h'	Ì	
1 1	273.31	5	ΔHv cal/g		$\vdash$	m to		
0.1 Press. mn	225,48	5	0.1 mm	47.92	5	n oK		
t <sub>e</sub>	2022.3	5	30 mm BP	40.90 32.39	5 5			
Density		4	t.	28.70	5	m' to		
g/ml 20°0	0 820	ハドアーラー	t <sub>e</sub> (d, e)	28.73	5	0'	`	
dt 25	0.817	73 <sup>‡</sup> 4	ΔHv/T <sub>e</sub>	20.33	5		<del> </del>	+
a	0,837	71 4	d 376 to	61.94	5	Surface tension dynes/cm. 20°C	29.22	5
ь	-0.036	66 4	6   575 °C   232 to	0.0565 58.70	5 5	الا 30 30	28.29	5
Ref. Index	- 1 4/-	22 2 2	e' 376 °C	0.0478	5	40	27.39	5
n <sub>D</sub> 20°0	1 460	17' 1 7 1	d <sub>c</sub> g/ml			Parachor [P] 20°C		
30	1.458	32 <sup>≠</sup> 4	vc ml/g tc °C	:		30		
"C"	0.741	7 4	P <sub>c</sub> mm			40 Suad	. 1583.2	5
MR (Obs.)		2	PV/RT		$\vdash$	Exp. L.1.%/wt.	. 1303.2	+
MR (Calc. (nD-d/2)	186,553 1. <b>0</b> 50	5 2 <sup>#</sup> 2	0.1 mm	1.0000	5	u.	1 4	
Dielectric		5	30 mm BP	1.0000 0.8881	5	Dispersion	102.	2
A 376 to		$\rightarrow$	t	0.8443	5	Flash Point °C Fire Point		
B   625 °C	3174.1	4					<del> </del>	+
C	141.	5	ΔHc kcal/m ΔHf	ĺ		M. Spec. Ultra V.		
A*  376 to B*  585 °C		81 5	ΔFf			X-Ray Dif. Infrared		
K 55	- 3001.3		Viscosity			Solubility in +		+-
t <sub>1.</sub>	-		centistokes			Acetone		
t <sub>k</sub> to			<i>7</i> °C			Carbon tet. Benzene		
A'   to						Ether		
B', L _ °	-		B <sub>v</sub> to		$\vdash$	n-Heptane Ethanol		
A'* 232 to	2 525	53 5	A I C			Water		
B'* 376 °C		52 5 5	(B <sup>V</sup> )	1		Water in	ļ	↓
Ac to			(A <sup>V</sup> )					
Bc tc C	<u>-</u>		c <sub>p</sub> liq. °K				1	
Cryos, A	+	-	1					
consts. B			р -					
t <sub>e</sub> °C	587.80	5	c <sub>v</sub> vap.					
	ercooled lie	quid				grams/100 grams/	ams solve	nt
REFEREN	CES: 1-Do	w 2-A	PI 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE:		API						
PURIFICA	TION:	API						
LITERATU	RE REFE	RENCES	S:					

							No. 1	
NAME	Vinyl bromi	.de				STRUCTURAL	FORMULA	
	Bromoethen							
	1					CHBr=CH <sub>2</sub>		
Mole % Pur. 100	Ref. Mo	lecul rmul		Molecular Weight 106.9	60			
		Ref.			Ref			Ref.
F.P. °C	-139.54	1	dt/dP			f to		
F.P. 1009	-139.54	1	*C/mm 25*C	0.0300	5	g .		
B. P. °C 760 mm	+15.80	2	BP	0.038	2	<u>+</u> +		
100	-31.7	5	t <sub>e</sub>	0.03736	5	f' to		
30 10	-52.3	5	30 mm	0.5224	5	h'		
1	<u> </u>		ΔHm cal/g	11.44	4	m l to		
Pressure mm 25°C	1022	_ ا	ΔHv cal/g 25°C	50.5	5	n °K		
t <sub>e</sub>	1033. 777.0	5 5	30 mm	80.8	5	o i		
Density			BP te (d.e)	52.4 51.6	5	m'   to		
g/ml 20°C	1 1 4720F	2 2	e (4,0)	51.2	5	n' K		
dt 25 4 30	1.47384	5	ΔHv/T <sub>e</sub>	19.12	5	<u> </u>		
8	1.5730	4	d   -55 to e   20 °C	58.0 0.417	5	Surface tension dynes/cm. 20°C	20.04	4
ь	-0.00372	4				3° 25 30	20.04', 18.97 <sup>‡</sup> 17.92 <sup>‡</sup>	4
Ref. Index	- 1 441#	2	e' i °C			Parachor [P]	11.92	4
45	1.435	2	d g/ml vc ml/g			20°C		
30 "C"		_	tc °C			30 40		
MR (Obs.)	0.5847	5	P <sub>c</sub> mm	1			152.1	5
MR (Calc.		5	PV/RT		_	Exp. L.1.%/wt.		
(nD-d/2)	0.6944	5	25°C 30 mm	0.9512 1.0000	5	u. Dispersion		
Dielectric		1	BP t <sub>e</sub>	0.9650 0.9642	5	Flash Point °C		_
A -60 to B60 °0	6.66715 C 953.4	5	tc	0.7042	-	Fire Point		
c	236.0	5	ΔHc kcal/m			M Spec. Ultra V.		
A*  -60 to		<b>5</b>	ΔHf ΔFf			X-Ray Dif.		
B* ⊢ 20 °C	899.8	] 3	Viscosity		T-	Infrared Solubility in +		├—
c	_	l	centistokes 7 -20 °C	0. 2759	1	Solubility in +		
t <sub>k</sub>   t <sub>0</sub>   t <sub>0</sub>			-10	0.2528	i	Carbon tet. Benzene		
A' to			0	0.2393	1	Ether		
B' <u>*</u>	익	1	B <sup>V</sup>   -30 to	213.8	4	n-Heptane Ethanol		
A'* to	,		A   10 °C	Z. 59636	4	Water		
B'* °(		L	(B <sup>V</sup> )			Water in		<u> </u>
Ac to			(A <sup>V</sup> )			]		
Cc CC			c <sub>p</sub> liq. °					
Cryos. Acconsts. B		1	c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	16.44	5	c <sub>v</sub> vap.	1				
for the l	iquid at satura			5°C		grams/100 gran	ns solven	t
	CES: 1-Dow	2-A1		Calc. from det	da.	ta 5-Calc, by for	mula	
SOURCE:			, 2					
PURIFICA			, 2					
LITERATU	RE REFERE	NCES	5:					
1								
ĺ								
1								

								No. 2	
NAME	Vinyl chl	oride				ST	RUCTURAL	FORMUL	Α
	Chloroetl	nene					CH -CH	Cl	
Mole % Pur. 100	Ref. Mo	ecul		Molecular Veight 62.50	ı		CH <sub>2</sub> =CH	C1	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-153.79	1	dt/dP °C/mm			f g	to		
B.P. °C 760 mm 100 30 10	-13.37 -55.8 -73.95 -87.5	1 4 4 4	25°C BP t <sub>e</sub> 30 mm	0.01355 0.03423 0.03675 0.4500	4 4 5 4	f' g'	to * <u>K</u>		
1	-109.42	4	ΔHm cal/g	18.14	4	m	to		<del>                                     </del>
Pressure mm 25°C t <sub>e</sub>	2660.0 692.3	<b>4</b> 5	AHv cal/g 25°C 30 mm BP	71.26 93.50 79.53	5 5 5	n o	<u>•K</u>		
Density g/ml -30° d <sup>t</sup> -25 4 -20	0.99986 0.99176 0.98 <b>34</b> 3	1 1 1	te te (d, e) AHv/Te	79.84 80.08 19.38	5 5	m' n' o'	to •K		
a b Ref. Index	0.95421 -0.00158	4	d -80 to e 20 °C d to	76.45 0.2306	5		face tension es/cm-30°C -20 -10	23.87 22.27 20.88	5 5 5
<sup>n</sup> D -10°C 25 30	1.4046	5	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.370 2.70 156.5	5 5 5	Par	achor [P] 20°C 30	20.00	
"C"	0,54	5	P <sub>c</sub> mm	42000.	5		40 Sugd.	138.4	5
	15.836 0°C 0.9213	5 5	PV/RT 25°C 30 mm	0.9178 1.0000	5		. L.1.%/wt.	2.5 69.9	1 1
Dielectric	6.26#	1	BP t	0.9640 0.9652	5	Fla	sh Point C	-78.0	1
A -100 to B _ 50 °C	6.49712 783.4	4	t e t c	0. 264	5		Point		-
A*  -80 to B*  0 °C	230.0 0.84290 727.85	5 5	ΔHc kcal/m ΔHf ΔFf	-144.0	5	Ultz X-F	Spec. a V. ay Dif. ared		
K c t <sub>k</sub> c to to to to to to to to to to to to to			Viscosity centistokes 7 -40 °C -30 -20 -10	0.3339 0.3026 0.2780 0.2563	1 1 1 5	Solu Ac Ca Be Et	ability in + etone rbon tet. nzene her		
C'	-		B <sup>V</sup>   -50 to A <sup>V</sup>   -10 °C	234.945 2.51616	4	Et. Wa	Heptane hanol iter		
B¹* °C	<u> </u>		(B <sup>v</sup> )  to			Wa	ter in	0.11	1
Ac 50 to Bc t <sub>c</sub> °C Cc		5 5 5	(A <sup>V</sup> )  °C c <sub>p</sub> liq. 293°K	0.38	1				
Cryos. A° consts. B°	0.04002	1	c <sub>p</sub> vap.298°K	0.205	1				
t <sub>e</sub> °C	-15.76	5	c <sub>v</sub> vap.						
$T_R = 0.7$			≠ 17.2°C				ams/100 gra		nt
	ES: 1-Dow			Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:	TON.	Do							
PURIFICAT	TION: RE REFEREI		stillation						
Z. Z.K. I U	NEFERE	TOES							

							No. 3	
NAME	cis-1-C	hloro-l	-propene			STRUCTURAL	FORMUL	A.
						CUCI-CU	CU	
Mole	Ref.	Molecu	10.00	Molecular		CHC1=CH-	CH <sub>3</sub>	
% Pur.	2	Form	lar C <sub>3</sub> H <sub>5</sub> Cl	Weight 76.	527			
		Re	ſ.		Ref.			Ref.
F.P. °C F.P. 100%	-134.8	2	dt/dP *C/mm	1		f to		
B. P. °C	+	-	- 25°C	0.0472	5	g *2	1	1
760 mm 100	+32.8	2 5	BP t <sub>e</sub>	0.0379 0.03578	5 5	$\frac{1}{5} + \frac{1}{5}$		
30	-36.2	5	30 mm	0.5266	5	g' 'K	:	İ
10 1	-52.1	5	ΔHm cal/g			h'	ļ	
Pressure			ΔHv cal/g 25°C	81.72	5	m to		
mm 25°C	575.0 817.9	5	30 mm	92.30	5			1
Density	+	+-	BP t	80.42 80.38	5 <b>5</b>	m¹   tó		
g/ml 20°C			te (d, e)	80.06	5	n' 'K		1
d4 25	0.92		ΔHv/T <sub>e</sub>	19.96	5	Surface tension	<del> </del>	+-
a b	0.96		d   -40 te		5	dynes/cm. 20°C	21.84	5
Ref. Index	-0.00	144 4	-a'	5		3 25 30	21.10	5
n <sub>D</sub> 20°C	1.40		<u> </u>	0.337	5	Parachor [P]		
25 30	1.40	00 2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g	2.97	5	20°C 30		
"C"	0.57	780 4	վ <sup>†</sup> Շ <sup>+</sup> Շ	217.8 35700.	5 <b>5</b>	40		_
MR (Obs.)			P <sub>c</sub> mm	3700.	3	Exp. L.1.%/wt.	.177.4	5
MR (Calc. (nD-d/2)	20.45		25°C	0.9631	5	u.		
Dielectric			30 mm BP	1.0000 0.9560	5	Dispersion	<del> </del>	—
A -40 to			<b>∄ :•</b>	0.9540 0.265	5	Flash Point °C Fire Point		
B ∟9 <u>5</u> °C	1065.3	5 5	t <sub>c</sub>		<u> </u>	M Spec.		1
A* -40 to			ΔHf ΔFf			Ultra V. X-Ray Dif.		1
B* _ 45 °C K	994.8	5	Viscosity	<del> </del>	<del>                                     </del>	Infrared		4
c	_		centistokes	.		Solubility in + Acetone		
t <sub>x</sub>   to			7	1	1	Carbon tet. Bensene		
A' to			1			Ether		
B' °	4		B <sup>V</sup>   to			n-Heptane Ethanol		
A'* to				<u>:  </u>	١ .	Water Water in		
B'* °C		270 5	(B <sup>V</sup> )	j			<del> </del>	†
Bc t °C	1322.0	5	c <sub>p</sub> liq. °	<del></del>	-	1		
Cc	270.	5	⊣l ⁻					
Cryos. A° consts. B°			c <sub>p</sub> vap. *K	`				
t <sub>e</sub> °C	34.93	5	c <sub>v</sub> vap.					
						+ grams/100 gra	ms solver	ıt
REFEREN	CES: 1-D			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:	rion.		PI					
LITERATU			PI 'S•	<del></del>				
	KEF	ENTOE						

								No. 4	
NAME _	trans-1-Chlo	pro-	l-propene		-	STRUCTUI	RAL	FORMUL	A
Mole % Pur.		ecul	ar C <sub>3</sub> H <sub>5</sub> C1	Molecular Veight 76.5	27	CHC1=	CH-C	СН <sub>3</sub>	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-99.0	2	dt/dP °C/mm 25°C	0.05405	5	f g	to °C		
B. P. °C 760 mm 100 30 10	37.4 -11.2 -32.6 -48.7	2 5 5	BP t <sub>e</sub> 30 mm	0.0340 0.0385 0.03594 0.5341	5 5	h	to *C		
l Pressure			ΔHm cal/g ΔHv cal/g 25°C	83.51	5	m   n	to •K		
mm 25°C t <sub>e</sub> Density	487.2 830.8	5	30 mm BP	93.81 81.57 81.03	5 5	o m'	to		-
g/ml 20°C dt 25 4 30	0.935 0.927 0.9185	2 2 4	t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	81. 11 19. 82	5 5	n'   o'	•ĸ		
a b Ref. Index	0.9696 -0.00162	<b>5</b> 5	d -40 to e 50 °C d' to	88.11 0.1748	5			21.77 21.13 20.33	5 5 5
n <sub>D</sub> 20°C 25 30	1.4054 1.400	2 2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	0.333 3.000 225.0	5 5 5	Parachor [	P] 20°C		
"C"	0.578	5	P <sub>c</sub> mm	35900.	5		10 Sugd.	177.4	5
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	20.828 20.454 0.9379	4 5 4	PV/RT 25°C 30 mm BP	0.9654 1.0000 0.9560	5 . 5	Exp. L.1.% u. Dispersion	/wt.		
A -40 to B 1100 °C	6.88340 1078.3 232.	5 5 5	te tc ΔHc kcal/m	0.9525 0.265	5	Flash Point Fire Point M. Spec.	•c		
A* -40 to B* 50 °C K	1.24083 10 <b>07.</b> 1	5 <b>5</b>	ΔHf ΔFf Viscosity			Ultra V. X-Ray Dif. Infrared	+		
t <sub>k</sub> to to A' to			centistokes η °C			Solubility in Acetone Carbon tet Benzene Ether	•		
B'°C C' to			B <sup>V</sup> to A <sup>V</sup> C			n-Heptane Ethanol Water			
B'* °C  Ac 100 to  Bc t <sub>c</sub> °C	7. 28034 1349.	5 5	(B <sup>V</sup> )  (A <sup>V</sup> )  c liq. °C			Water in			
Cryos, A° consts, B°	270.	5	c <sub>p</sub> liq. °C						
t <sub>e</sub> °C	40.03	5	c <sub>w</sub> vap.			† grams/10	0 gra	ms solver	nt
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da				
SOURCE:		AP							
PURIFICAT	ION:	AP	r						
LITERATUE	RE REFEREI	NCES	5:						

									No. 5	
NAME	1,1,	2-Tr	ichlo	roethylene			STRUCTU	RAL 1	FORMUL	A
Mole % Pur.	Ref.	Mo	lecul:	C2HC13	Molecular Weight 131.3	99	cc	1 <sub>2</sub> =CH	Cl	
7		1 10	Ref.		028	Ref				Ref
F, P. °C				dt/dP	Γ		f	to		
F.P. 100	•			°C/mm			g   _	_ <u>*K</u>		
B. P. °C	7	_		25°C BP	0.2890 0.04369	5 5	h			
760 mm 100	87.0 31.5		1 4	t.	0.0359	5	f'	to		
30	6.9	0	5	30 mm	0.6178	5	g'   '_	_ <u>_</u> K_		
10 1	-11.8 -42.8		5	ΔHm cal/g			h'			1
Pressure	+		H	ΔHv cal/g			m	to °K		
mm 25°C	74.3		5	25°C 30 mm	62.33 64.02	5 5	-			
t <sub>e</sub>	975.5	Z 	5	BP	56.43	5	m'	to		+-
Density g/ml 20°	1.4	6422	1	te te (d, e)	55.61 55.62	5 5	n _	_ °K_		1
dt 25	1.4	5541	1	ΔHv/T <sub>e</sub>	19.81	5	0'			
<b>4</b> 60		9501	1	d 7 to	64,67	5	Surface te	nsion		
a b	-0.0	9948	4 4			5	dynes/cm.	20°C 30	29.28 27.94	1 1
Ref. Index		01/4	H	d'   to				40	26.76	li
n <sub>D</sub> 20°	1.4	7734	1	d <sub>c</sub> g/ml	0,513	5	Parachor	[P]		1
25 50		7457 6056	1 1	( A - m1/8	1.950	5		20°C 30	208.7	4
"C"	0.4		4	tc *C	298.	5		40	208.8	4
MR (Obs.	- <del></del>		4	P <sub>c</sub> mm	36876.	5		Sugd.	212.8	5
MR (Calc.	) 25.5	7	5	PV/RT 25°C	0.0057	5	Exp. L.1.	%/wt.		1
(nD-d/2)	0.7	4523	4	30 mm	0.9957 1.0000	5	u. Dispersion	1		
Dielectric			$oldsymbol{\sqcup}$	BP	0.9578	5	Flash Poir			+-
A 7 t		2808	4 4	te t <sub>c</sub>	0.9507 0.265	5	Fire Point			
c Liss	230.		4	ΔHc kcal/m		1	M Spec.			
A* 7 to	1.54	4642	5	ΔHf			Ultra V. X-Ray Dif			
B* 106 %	1230.2		5	ΔFf	<u> </u>	-	Infrared			
c	_			Viscosity centistokes			Solubility	in +		-
tk				7 20 °C	0.3844	1	Acetone Carbon te	t.		
A' t			$\vdash$	40 60	0.3291 0.2867	1 1	Benzene			
B' •				80	0.2553	1	Ether n-Heptan			1
C'				B <sup>V</sup>   10 to A <sup>V</sup>   70 °C	311.03	4	Ethanol			1
A'* to				75V. — — —	2,52395	4	Water Water in			
Ac   155 to	<del></del>	675	5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C	1					T
Bcit . º	C 1675.	013	5		ļ	-				
Cc — —	280.		5	c <sub>p</sub> liq. ∘K	1					
Cryos, A consts. B				c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	95.60	0	5	c <sub>v</sub> vap.	L		L <u>.                                    </u>			
T <sub>R</sub> = 0.75		<b></b>	<del></del>				grams/l			nt
REFEREN	CES: 1-D	юw.	Z-AF		calc, from de	t. da	ta 5-Calc.	by for	mula	
SOURCE:	TION:									
PURIFICA LITERATU		ERF		stillation.			<del></del>			

No. 6 Perchloroethylene STRUCTURAL FORMULA NAME CC12=CC12 Molecular C2Cl4 Mole Molecular Weight 165.848 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm g \_°K 25°C 1.0102 5 B. P. ℃ h ВP 0.04721 5 760 mm 120.97 1 t<sub>e</sub> 0.0357 5 ſ١ to 100 60.97 4 g' <u>°К</u> 0.6670 5 30 34.37 4 30 mm 10 14.02 5 h! ∆Hm cal/g 5 -20.12 m to AHv cal/g Pressure °K n 25°C 57.09 5 mm 25°C 18.47 5 o 30 mm 56.65 5 5 1086. te ВP 50.0 5 m' to 48.96 Density 5 5 t<sub>e</sub> (d, e) n' °K g/ml 20°C 1,62272 1 48.98 ۰,  $d_4^t$ 25 1.61446 1 19.93 5 AHV/Te 30 1.60590 4 Surface tension d 34 to 59.29 5 a 1.65599 4 dynes/cm. 20°C 31.33 1 0.0768 5 134 °C 25 to ь -0.00166 4 30.81 30 1 d٦ 58.27 5 40 30.20 1 e ¹ Ref. Index 34 °C 0.0472 5 20°C 1.50534 Parachor [P] n<sub>D</sub> 0.573 d<sub>c</sub> g/ml 4 25 1.50284 1 20°C 241.8 4 vc ml/g t °C 1.745 4 1.50004 30 4 30 243.3 4 <sup>t</sup>c 340. 5 4 40 244.6 "C" 0.4095 4 P<sub>c</sub> mm 33687. 5 Sugd. 250.0 5 MR (Obs.) 30.34 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30.437 25°C 1,0000 (nD-d/2) 0.69417 4 30 mm 1.0000 5 Dispersion Dielectric BP 0.9672 Flash Point C 0.9580 5 A 34 to 7.02003 Fire Point 0.255 В 1415.5 187°C 4 M. Spec. Ultra V. С 221. 4 AHc kcal/m ΔHf A\* 34 to 1.58865 X-Ray Dif. ΔFf B\* 144°C 1321.5 Infrared ĸ Viscosity Solubility in centistokes Acetone to 20 0.54976 1 t<sub>x</sub> Carbon tet. 40 0.45702 1 Benzene 60 0.39347 1 25 to 7.41073 Ether 0.34061 80 B' 1622.1 34 °C 5 n-Heptane  $\mathbf{B}^{\mathbf{v} \top}$ C' 239. 5 10  $\frac{353.2}{2.53223}$ Ethanol A | 90 °C A1\* Water 25 to 1.96901 5 34 °C (BV) Water in 1518.6 5 to Ac | 187 to 7.4489 5 (A<sup>V</sup>)| °C 1787. Bc tc °C c<sub>p</sub> liq. ۰ĸ Cc 271. 5 c<sub>p</sub> vap. Cryos. A °K consts. B° c vap. t<sub>e</sub> °C 134.28 5  $T_{R} = 0.75 T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula Dow SOURCE: Distillation PURIFICATION: LITERATURE REFERENCES:

· · · · · · · · · · · · · · · · · · ·								No. 7		
NAME	Неха	chlor	opro	pene			STRUCTURAL FORMULA			
<u> </u>	Hexa	chlor	opro	pylene			C C1 <sub>3</sub> C C1=	C Cl.		
Mole % Pur.	Ref.	Mo Fo	lecul rmul	ar C <sub>3</sub> Cl <sub>6</sub>	Molecular Weight 248.7	72	3			
			Ref.			Ref			Ref	
F.P. °C	-57.1	2	1	dt/dP			f to			
F.P. 1009	•			*C/mm 25*C	48.562	5	g <u>*K</u> _			
B. P. °C 760 mm	214.1	4	1	BP	0.05767	4	h			
100	141.1	3	i	t.	0.03718	5	f' to		1	
30 10	108.9°	7	4 5	30 mm	0.805 <b>0</b>	4	g'  K_			
1	43.5		5	∆Hm cal/g	<u> </u>		h'		<u> </u>	
Pressure	1			ΔHv cal/g	52.40	_	m to		1	
mm 25°C	0.2	717	5	25°C 30 mm	53.40 47.91	5 <b>5</b>			ŀ	
t <sub>e</sub>	1336.0		3	BP	40.7	5	m'   to		$\vdash$	
Density g/ml 20°(	1.7	6376	1	te (d, e)	39.04 38.89	5	n' K	-		
t 25	1.7	5 <b>6</b> 56	1	ΔHv/T	19.06	5	٥'		ļ	
	1.74		4	d   100 to	<del>                                     </del>	5	Surface tension			
a b	-0.00	9 <b>2</b> 56	4	<u>e   245 °C</u>	<b>0</b> . 0687	5	dynes/cm. 20°C	38.12 36.95	1 1	
Ref. Index			<u> </u>	d'   20 to		5	40	35.87	ì	
n <sub>D</sub> 20°0		4956	1	d <sub>c</sub> g/ml	0.58	5	Parachor [P]			
25 30		4720 3570	1 1	ll v mi/g	1.72	5	20°C 30	350.5 349.2	4	
"C"	<del></del>	0748	4	1c 3C	447.	5	40	350.9	4	
MR (Obs.			4	P <sub>c</sub> mm	23966.	5	Sugd.	363.4	5	
MR (Calc.	) 44.7	87	5	PV/RT 25°C	1,0000	5	Exp. L.1.%/wt.		1	
(nD-d/2)	<del></del>	6768	4	30 mm	1.0000	5	u. Dispersion			
Dielectric			1	BP	0.9522 0.9348	<b>5</b>	Flash Point °C		<del>                                     </del>	
A 109 t B 267		2329	1 1	t <sub>e</sub> t <sub>c</sub>	0.229	5	Fire Point		<u> </u>	
c —	193.8		ī	ΔHc kcal/m			M Spec. Ultra V.			
A*   110 to		1793	5	ΔHf ΔFf			X-Ray Dif.			
B* _250°	1550.5	ı	5	Viscosity	<del> </del>	$\vdash$	Infrared		L_	
c	_			centistokes			Solubility in + Acetone		ŀ	
tk			1	7 20 °C	1.9698	1 1 1	Carbon tet.		İ	
A' 20 to		664	5	60	1.0652	i	Benzene			
B' <u>[109</u> •	1863.7		5	80	0.8515	1	Ether n-Heptane			
C'	212.9		5	B <sup>V</sup>   35 to A <sup>V</sup>   90 °C	601.4 2.22746	4 4	Ethanol	0.015	1	
A'* 20 to B'* 109 °C		332	5 5		- 2.22140	7	Water Water in	0.013	li	
Ac   269 to		256	5		1					
Bc t	2041.		5		<del> </del>	$\vdash$				
Cc	245.7		5	c <sub>p</sub> liq. °K	1					
Cryos. A'consts. B'				c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	240, 4	5	5	c <sub>w</sub> vap.						
$T_R = 0.7$							f grams/100 gran		t	
REFEREN	CES: 1-D	ow	2-AI		Calc. from det	t. da	ta 5-Calc, by for	mula		
SOURCE:	PION			ow .						
PURIFICA		ERE	_	ist. chromat.						

									No. 1
NAME	Propadi	ene					ST	RUCTURAL	FORMULA
	Allene								
Mole % Pur	Ref.		ecul: muls		Molecular Weight 40.	062		CH <sub>2</sub> =C=C	CH <sub>2</sub>
			Ref.		i cigai io.	Ref.	ī	<del></del>	Re
F. P. °C	-136.		2	dt/dP		1	ſ		
F.P. 100%				°C/mm	İ		g	to *C	
B. P. °C				25°C BP	0.0110	5 4	h	ļ l	
760 mm 100	-34.5 -73.0	ł	2 4	t	0.0372	5	_r,	to	
30	-88. 2	l	4	30 mm	0.3674	5	g'	*c	
10 1	-99.		4	ΔHm cal/g			h'		
Pressure	<del> </del>			ΔHv cal/g			m	to *K	
mm 25°C	4286.	- 1	5	25°C 30 mm	153.0	_	n	, A	
t <sub>e</sub>	637.1		5	BP	153.8 109.0	5 5	<u> </u>		
Density				t <sub>e .</sub> .	112.0	5	m'	to to	
g/ml 20°C	1			te (d, e)	112.5	5	۰,		
dt 25 4 30		l		AHv/T <sub>e</sub>	19. 23	5		face tension	
a·				d -90 to		5		es/cm. 20°C	
ь				d' to			8	30 40	
Ref. Index		ł		e'   °C		$\perp$	Bar	achor [P]	
25				d <sub>c</sub> g/ml			Fai	20°C	
30				vc ml/g tc °C	120.0	2		30	
"C"				P <sub>c</sub> mm				40 Sugd.	
MR (Obs.) MR (Calc.		1	- 1	PV/RT		+	Ext	L.1.%/wt.	
(nD-d/2)	1			25°C 30 mm	0.8750	5 <b>5</b>		u.	ļ
Dielectric				BP	1.0000 0.9700	5		persion sh Point °C	
A -95 to	5.64	457	2	t <sub>e</sub>	0.9769	5		e Point	Ì
B ∟25 °C C	441.0 194.0		2 2	tc ΔHc kcal/m	4	4-4	M.	Spec.	
A*  -90 to		775	5	ΔHC KCal/m			U1t:	ra V.	ŀ
B*  15 °C		,,,,	5	ΔFf				Ray Dif.	
к — — —	-			Viscosity				ability in +	
t <sub>k</sub> – tō	-	ľ		centistokes り °C			Ac	etone	
t <sub>x</sub> °C			- 1	'				rbon tet.	
A'   to							Et	her	
B'°	-			B <sup>v</sup>   to oc				Heptane hanol	İ
A¹* to				A <sup>V</sup> I °C			w.	ater	
B1* °C				(B <sup>V</sup> )			W.	ater in	
Acl to		Ī		(A <sup>V</sup> )					
Bc tc °C	-	- 1		c <sub>p</sub> liq. °C					
Cryos. A*		$\dashv$	-	c <sub>p</sub> vap. *K					
consts. B°				р.					
t <sub>e</sub> °C	-38.8		5	c <sub>v</sub> vap.					
							† g:	ams/100 gra	ms solvent
REFEREN	CES: 1-E	ow	2-A	PI 3-Lit. 4	-Calc. from d	et. da	ta 5	-Calc. by for	mula
SOURCE:			AP	I					
PURIFICA'	rion:		ΑP	I					
LITERATU	RE REF	EREN	CES	<b>5:</b>					

							No. 2	
NAME	1, 2-Buta	diene				STRUCTURAL	FORMULA	4
Mole % Pur.	Ref. M	olecul ormul	ar C <sub>4</sub> H <sub>6</sub>	Molecular Weight 54.08	8	CH <sub>2</sub> =C=CH	CH <sub>3</sub>	
	<del></del>	Ref.			Ref			Ref
F.P. ℃	-136, 190	2	dt/dP	T		f   to		
F.P. 100%			°C/mm		1 1	g L°K		
B. P. *C			25°C BP	0.0234 0.03495	5 4	h ;		
760 mm 100	10.85	2 2	t	0.03500	5	f' to		Г
30	-54.	2	30 mm	0,5022	5	g'   ' <u>*</u> K_		
10 1	-69.07 -94.48	2 5	ΔHm cal/g			h'		
Pressure	+-/:	+	AHv cal/g			m   300 to	0.0654	
mm 25°C	1259.8	5	25°C	104.13	5	n 600 °K	0.0011 -0.0 <sub>6</sub> 42	
t <sub>e</sub>	758.60	5	30 mm BP	117.39 106.74	5	<u> </u>		
Density	0.652	Τ,	t	106.76	5	m'   700 to n' ,  1000 °K	0.1346	4
g/ml 20°C	0.646	2 2	te (d, e)	106.75	5	o'   '= = = :	0.0 <sub>3</sub> 90 -0.0 <sub>6</sub> 30	4
dt 25 4 30	0.640	4	ΔHv/T <sub>e</sub>	20.33	5	Surface tension	<del>                                     </del>	<del> </del>
a	0.6770	4	d   -54 to		5	dynes/cm, 20°C	16.56	5
ь	-0.0011	4	-å,-¦¹ 1- °€		,	30 40	15.24	5
Ref. Index			e' i °C	ļ		Parachor [P]	13.95	13
D 25			d g/ml	0.2468 4.051	3	20°C		
30	<u> </u>		tc °C	170.6	3	30		
"C"			P <sub>c</sub> mm	33770.	3	40 Sugd	168.2	5
MR (Obs.) MR (Calc.		5	PV/RT	<del> </del>	-	Exp. L.1.%/wt.	1	<u> </u>
(nD-d/2)	17.130	'	25°C	0.9461	5	u.		1
Dielectric			30 mm BP	1.0000 0.9600	5	Dispersion		
A -54 to	7, 1619	2	t <sub>e</sub>	0.9601	5	Flash Point °C Fire Point		
B ∟68 °C		2	t <sub>c</sub>		_	M Spec.		
C	251.0	2	ΔHc kcal/m ΔHf gas	588.37 38.77	2 2	Ultra V.		1
A* -54 to B* 21 °C	1.37634 1045.57	5	ΔFf gas	47.43	2	X-Ray Dif. Infrared		
K '	-		Viscosity			Solubility in +		╁
t.		1	centistokes 7°C			Acetone		Ì
t <sub>k</sub> to			7 °⊂	1		Carbon tet. Benzene		1
A' to		<del>                                     </del>				Ether		Ì
B', ∟ _ °	2		B <sup>V</sup> to		<del> </del>	n-Heptane		
A'* to		+	Av i °C			Ethanol Water		l
B'* °C			(BV) to	-		Water in	ļ	<u> </u>
Ac  68 to	7.62883		(A <sup>V</sup> )  °C	ı			1	
Bc tc °C	1438.26 293.28	5	c <sub>p</sub> liq. •K	<del></del>				
Cryos, A°	273,20	+	() <del>-</del>	ì				
consts. B			c <sub>p</sub> vap.300°K 400	0.35553 0.43522	2 2			
t <sub>e</sub> °C	10,80	5	c <sub>v</sub> vap.		-			
$T_{R} = 0.7$	5 T <sub>C</sub>		≠ at saturation	pressure	L	grams/100 gra	me solven	<del></del> -
	CES: 1-Dow				t. da	ta 5-Calc, by for		*
SOURCE:		AF						
PURIFICA?	TION:	AF	PI					
LITERATU	RE REFER	ENCES	: 3 ASTM Sp.	Tech. Pub.	109			
			-					

No. 3 1.3-Butadiene NAME STRUCTURAL FORMULA CH,=CHCH=CH, Molecular C4H6 Mole Ref. Molecular Weight 54.088 % Pur Formula Ref. Ref. Ref. F.P. °C F.P. 100% -108.915 2 dt/dP f to °C/mm °K g 25°C 0.0154 5 B.P. °C h BP 0.03377 4 760 mm -4.413 2 0.03542 5 f١ -47.035 100 2 ŧ, to g' °K -65.74 2 30 0.4675 5 30 mm -79.89 10 2 h! ∆Hm cal/g -103.17 5 300 to -0.0514 m ∆Hv cal/g Pressure 0.0016 600 °K n 25°C 92,25 mm 25°C 2104.5 5 -0.0698 o 4 30 mm 112,75 5 714.96 5 te 98.89 5 ВP 700 to 0.1766 m' 4 Density 99.26 5 0.0<sub>3</sub>87 -0.0<sub>6</sub>30 n' 1000 °K g/ml 20°C te (d, e) 99.25 0.6211 2 5 0.6149 ٥' 4 2 d⁴₄ 25 AHv/Te 5 20.09 30 0.6086 4 Surface tension 97.90 d -66 to 5 0,6480 4 а dynes/cm. 20°C 13.41 5 <u>-6 °C</u> 0.2259 5 12.20 Ъ -0.0011 4 5 ă۳ 30 to 40 11.01 5 Ref. Index e' <sup>n</sup>D 20°C [P] Parachor d<sub>c</sub> g/ml 0.245 2 25 20°C v<sub>c</sub> ... ml/g 4.082 30 30 t<sub>c</sub> 152. 2 40 "C" P<sub>c</sub> mm 32452. 2 168.2 5 Sugd. MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 21,008 5 25°C 0.9269 (nD-d/2)30 mm 1.0000 5 Dispersion Dielectric BP 0.9600 5 Flash Point C -66 to 0.9616 5 6.85941 2 Fire Point 0.271 2 ž В 46 °C 935.53 M. Spec. Ultra V. 239.55 С 2 ∆Hc kcal/m 575.93 2 ΔHf 21.21 A\* -66 to 1.11738 5 X-Ray Dif. ΔFf B\* 5 °C 872.39 Infrared ĸ Viscosity Solubility in c centistokes Acetone to  $t_k$ Carbon tet.  $\mathbf{t_{\underline{x}}}$ °C Benzene A۱ to Ether В' °C n-Heptane B<sub>v</sub> C' to Ethanol °C Water A1\* to Water in B'\* (B<sup>V</sup>) °C to 7.29710 1202.54 Ac 46 to (A<sup>V</sup>) 5 °C Bc °C °K cp liq. 277.80 5 Cc c<sub>p</sub> vap300°K Cryos. A° 0.35331 2 consts. B° 0.44908 2 400 t, °C c, vap. -5.97 5  $T_R = 0.75 T_c$ at saturation pressure grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

								No. 4	
NAME	1,2-F	Pentadier	ıe .			STR	UCTURAL 1	FORMULA	
					$\dashv$		сн -с <del>-</del> снс	н сн	
Mole % Pur.	Ref.	Molecu Formu	lar C <sub>5</sub> H <sub>8</sub>	Molecular Weight 68.11	4		сн <sub>2</sub> =с <b>-</b> снс	203	
		Ref			Ref				Ref.
F.P. *C	-137, 20		dt/dP	1		f	to		
F.P. 100%			°C/mm			g	L		ł
B. P. °C			25°C BP	0.0691 0.03867	5 4	h	1		
760 mm 100	44.85		t.	0.03528	5	f'	to		
30	-26.04	4 2	30 mm	0.5457	5	g'	K_		
10 1	-42.6 -69.99	9 5	ΔHm cal/g			h'			
Pressure	-07.7	<del>'   '</del>	ΔHv cal/g	†		m	300 to	0.0148	
mm 25°C	367.20	6 5	25°C	99.47	5	n o	_600 °K	0.0014 -0.0 <sub>6</sub> 73	
t <sub>e</sub>	853.68	3 5	30 mm BP	108.86 95.76	5		1		
Density	2	),,,,T	l t	95.12	5	m' n'	700 to	0.1189 0.0010	
g/ml 20°C		9257   2 3760   2	t <sub>e</sub> (d, e)	95.12	5	ö'	·	-0.0 <sub>6</sub> 37	4
d <sub>4</sub> 25 30		3259 4	AHv/T <sub>e</sub>	20.15	5	Surf	ace tension	<del>-</del>	$\vdash$
	0.71		d   -26 to		5		s/cm. 20°C	19.57	5
ь	-0.0	947 4	╢╬┼╌╬			3.	30	18.41	5
Ref. Index	1,42	2091 2	e' i •C	·		-	40	17.27	5
<sup>n</sup> D 25		773 2	d g/ml v ml/g	1		Para	chor [P] 20°C		1
30	1.41	446 4	v <sup>c</sup> ml/g t <sub>c</sub> *C	221.	5		30		
"C"	0,80	080 4	P <sub>c</sub> mm	32089.	5	ļ	40 Sugd	207.2	5
MR (Obs.)	24.93		PV/RT	-	-	Evn	L.1.%/wt.	201.2	<u> </u>
MR (Calc. (nD-d/2)	24.35		25°C	0.9744	5	DAP.	u.		ŀ
Dielectric	2, 02		30 mm BP	1.0000 0.9575	5	Disp	ersion	164.6	2
A   -26 to	+	100 2	t.	0.9543	5		h Point °C		
B97 °C			tc		<u> </u>		Point		├—
С	234.65		AHc kcal/m	736, 25	2	M Sp Ultra			l
A*   -26 to B*   58 °C	1.29	9889 5	ΔHf gas ΔFf gas	34.80 50.29	2	X-R	y Dif.		
B* ∟ 58 °C	1079.85	'   '	Viscosity	30.27	-	Infra	<del></del>		ļ
·	4		centistokes				oility in +	<b>∞</b>	
t <sub>k</sub> to		ı	7 .c	1			bon tet.	œ	Ì
t <sub>x</sub>   *C			1	1		Ber Eth	zene	<b>00</b>	1
B' •C		- 1	-	<b></b>	<u> </u>		leptane	<b>&amp;</b>	
C'	<u> </u>		B <sup>V</sup> to	1		Eth Wat	anol	oc	
A'* to B'* *C		l		-{			er in		ĺ
Ac  97 to	+	719 5	1	1					
Bc t °C	1462.26	5   5		<del></del>	$\vdash$				
Cc — —	276.31	5	c <sub>p</sub> liq. ∘K						İ
Cryos, A° consts, B°			c <sub>p</sub> vap.300°K 400	0.37144 0.46099	2 2				
t <sub>e</sub> °C	48,31	5	c <sub>v</sub> vap.	<u> </u>		<u> </u>			
$T_R = 0.7$							ms/100 gran	ns solven	<u> </u>
REFERENC	ES: 1-D			Calc. from det	da:	ta 5-	Calc. by for	mula	
SOURCE:			PI						
PURIFICAT			PI						
LITERATU	RE REFI	ERENCE	S:						

No. 5 1, cis-3-Pentadiene STRUCTURAL FORMULA NAME CH2=CHCH=CHCH3 Mole Ref. Molecular Molecular C5H8 Weight 68.114 % Pur Formula Ref. Ref. Ref. F, P. °C -140,820 2 dt/dP f to F.P. 100% °C/mm g <u>°K</u> 25°C 0.0673 5 B. P. °C h BP 0.03875 4 760 mm 44.068 2 0.03544 5 f١ te to 100 -5.02 2 g <u>°K</u> -26.67 5 30 2 30 mm 0.5421 10 -43.1 2 h' ∆Hm cal/g 5 -70, 22 -0.0544 m 300 to AHv cal/g Pressure 0.0015 600 °K n 25°C 98.84 5 mm 25°C 379.07 o -0.0<sub>6</sub>73 4 30 mm 109.02 5 5 851.42 te ВP 95.09 5 m 700 to 0.1498 4 5 Density 94.44 te (d, e) 0.0<sub>3</sub>91 -0.0<sub>6</sub>29 n' 1000 °K g/ml 20°C 4 0.69102 2 94.42 o'  $\mathbf{d_4^t}$ 25 0.68592 2 ΔHv/Te 20.06 5 30 0.68078 4 Surface tension ď -27 to 103.76 5 0.7116 a 4 dynes/cm. 20°C 19.39 5 0.1969 å, 47 °C -0.0397 ь 30 18.21 5 5 40 17.05 Ref. Index e' 20°C 1.43634 Parachor [P]  $^{n}D$ d<sub>c</sub> g/ml 25 1.43291 2 20°C vc ml/g 30 1.42943 4 30 <sup>t</sup>c 219. 5 40 "C" 0.8377 4 P<sub>c</sub> mm Sugd. 207. 2 30901. 5 5 MR (Obs.) 25.790 2 PV/RT Exp. L. l. %/wt. MR (Calc.) 25,626 25°C 0.9739 5 (nD'-d/2)1.09083 2 30 mm 1.0000 2 5 Dispersion 243.8 Dielectric 2.06 5 BP 5 0.9575 Flash Point C 0.9543 5 -27 to 6.94178 1118.37 Fire Point 2 \_\_ <u>96 °C</u> M. Spec. Ultra V. С 231.33 2 AHc kcal/m 720.15 2 2 ∆Hf gas 18.70 A\* -27 to 1,23515 5 X-Ray Dif. ΔFf 34.88 2 gas B\*[ 57 °C 1046.11 Infrared ĸ Viscosity Solubility in centistokes Acetone to tk tx Carbon tet. °C Benzene ٨' to Ether B' °C n-Heptane B<sub>v</sub> C to Ethanol °C Water A'\* to Water in °C B'\* (B<sup>V</sup>) to Acl 96 to 7.37139 1417.55 5 (A<sup>V</sup>)| °C °C Bc tc c<sub>p</sub> liq. ۰ĸ Cc' 272.55 5 c<sub>p</sub> vap300°K Cryos, A 0.33326 2 consts. B° 400 0.43310 c<sub>v</sub> vap. te °C 47.45  $T_{R} = 0.75 T$ grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: API PURIFICATION: LITERATURE REFERENCES:

							No. 6			
NAME	1, trans-3	-Pen	tadiene			STRUCTURAL FORMULA				
				·····						
Mole % Pur.		lecul rmul	ar C <sub>5</sub> H <sub>8</sub>	Molecular Weight 68.114	4	CH <sub>2</sub> =CHCH=	ECH CH <sub>3</sub>			
		Ref.			Ref.			Ref		
F.P. °C	-87.470	2	dt/dP			f   to				
F.P. 100%			°C/mm	1	_	g K				
B. P. *C			25°C BP	0.0632	5 4	h				
760 mm 100	42.032 -7.06	2	t	0.03567	5	f' to				
30	-28.7	2	30 mm	0.5414	5	g'	-			
10 1	-45.1 -72.15	2 5	ΔHm cal/g			h'				
Pressure	+	<del>                                     </del>	ΔHv cal/g			m   300 to	0.0087 0.0014	4		
mm 25°C	410.5	5	25°C	97.06 107.39	5	n 600 °K	-0.0 <sub>6</sub> 66	4		
t <sub>e</sub>	845.6	5	30 mm BP	93.76	5	1 700				
Density	0 (7/02		te (d. a)	93.17	5	m'   700 to n'   1000 K	0.1303	4		
g/ml 20°C	0.67603 0.67102	2 2	'e (u, c)	93.15	5	0'	-0.0637	4		
dt 25 4 30	0.66597	4	ΔHv/T <sub>e</sub>	19.93	5	Surface tension	<del>                                     </del>	-		
a	0.6963	4	d   -29 to		5	dynes/cm. 20°C	17.75	5		
ь	-0.0395	4		1	ا آ	30 40	16.66 15.58	5		
Ref. Index	1,43008	2	e' i °C			Parachor [P]	13.30	<u> </u>		
D 25	1.42669	2	d g/ml			20°C				
30	1.42302	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	214.	5	30				
"C"	0.8448	4	P <sub>c</sub> mm	29456.	5	40 Sugd	207.2	5		
MR (Obs.)	26. 032	2	PV/RT			Exp. L.1.%/wt.		_		
MR (Calc.) (nD-d/2)	1.09206	5 2	25°C	0.9724	5-	u.	[ ]	_		
Dielectric	2,04	5	30 mm BP	1.0000 0.9575	5	Dispersion	245.7	2		
A   -29 to		2	te	0.9545	5	Flash Point °C Fire Point				
B92 °C	1108.94	2	t <sub>c</sub>	<u> </u>	L		<del> </del>			
С	232, 34	2	ΔHc kcal/m ΔHf gas	720.05 18.60	2 2	M Spec. Ultra V.				
A*   -29 to B*55 °C	1.21772	5	ΔFf gas	35.07	2	X-Ray Dif.				
K L==	-		Viscosity			Infrared Solubility in +				
t.	-{	į	centistokes 7 °C	1		Solubility in + Acetone	oc			
t <sub>k</sub> to t <sub>x</sub> to			<b>7</b> °⊂			Carbon tet.	00			
A'   to	<del> </del>	_		1		Benzene Ether	80	Ì		
B' °	4		B <sup>V</sup>   to		-	n-Heptane	œ			
	<del> </del>	-	B' to	İ		Ethanol Water	<b>∞</b>			
A'* to B'* °C		1	(BV) to	-		Water in				
Ac   92 to	7.35341	5	(A <sup>V</sup> ) °C							
Bc t C		5	c <sub>p</sub> liq. °K	<u> </u>						
Cc	273.32	5	41 -	1						
Cryos. A° consts. B°		_	c <sub>p</sub> vap.300°K	0.36556 0.45806	2	E				
t <sub>e</sub> °C	45.22	5	c <sub>v</sub> vap.	l		L,	<u> </u>	<u> </u>		
$T_{R} = 0.7$						f grams/100 gra		<u> </u>		
	ES: 1-Dow			Calc. from det	. da	ta 5-Calc, by for	mula			
SOURCE:		AF								
PURIFICAT		AI								
LITERATU	RE REFERE	NCES	5:							

								No. 7		
NAME	1,4-P	entadiene	e			STRUCTURAL FORMULA				
Mole % Pur.	Ref.	Molecul Formula		Molecular Weight 68.114	1		CH <sub>2</sub> =CHCH <sub>2</sub>	сн=сн2		
		Ref.			Ref.				Ref.	
F.P. °C F.P. 100	-148, 27! %	5 2	dt/dP °C/mm			f g	to*K			
B. P. ℃			25°C BP	0.0382	5 4	h	l			
760 mm	25.96 -20.96		t <sub>e</sub>	0.03573	5	f'	to			
30	-41.5	2	30 mm	0,5142	5	g'	'° <u>K</u>			
10 1	-57.1 -82.69	5	ΔHm cal/g			h'				
Pressure	-02.07		ΔHv cal/g			m	300 to	0.0221 0.0013	4	
mm 25°C	734.3	5	25°C	88.28	5	n o	_600 <u>•</u> K	-0.0666	4	
t <sub>e</sub>	800.1	5	30 mm BP	101.49 88.09	5	<u> </u>			<u> </u>	
Density			te	87.82	5	m' n'	700 to	0.1175 0.0010	4	
g/ml 20°0	0.66 0.65		te (d, e)	87.80	5	0'	======	-0.0637	4	
d <sub>4</sub> 25 30	0.65		ΔHv/T <sub>e</sub>	19.90	5	C	face tension	-	-	
a	0.68		d -41 to e 27 °C		5		es/cm. 20°C	16.09	5	
Ъ	-0.03	92 4	-å-  - <sup>27</sup>			8	30	15.03	5	
Ref. Index	-	876 2	e'   °C			<u> </u>	40	13.98	5	
<sup>n</sup> D 20°0	1.38 1.38		d <sub>c</sub> g/ml			Par	achor [P]			
30	1.38	247 4	v <sub>c</sub> m1/g t <sub>c</sub> °C	187.	5		30			
"C"	0.78	57 4	P <sub>c</sub> mm	27238.	5	ł	40 Snad	207.2	5	
MR (Obs.			PV/RT	1.230.	-	E	L. 1. %/wt.	201.2	۲	
MR (Calc. (nD-d/2)	.) 24.35 1.05		25°C	0.9589	5	Exp	u.			
Dielectric		5	30 mm	1.0000	5	Dis	persion	153.1	2	
A -41 to			BP t_	0.9580 0.9565	5		sh Point C			
B _ 72 °C			t t				e Point		-	
С	232.35	4 2	AHc kcal/m	726.65	2		Spec. ra V.			
A* -41 to			ΔHf gas ΔFf gas	25.20 40.69	2 2	X-I	Ray Dif.			
B* _37°0	957.53	5	Viscosity				ared		<u> </u>	
° ~	_		centistokes				ubility in <sup>†</sup> etone			
t <sub>k</sub> to			η °C				rbon tet.			
A'I to							nzene her			
B'		ļ	ļ				Heptane			
C'			B <sup>V</sup> to A <sup>V</sup> °C				hanol			
A'* to B'* °(			III. IV	_			ater ater in			
Ac 72 to		510 5	1 ' 2''							
Bc tc °C		5   5		+	$\vdash$					
Cc	270.91	5	c <sub>p</sub> liq. °K							
Cryos. A consts. B			c <sub>p</sub> vap.300°K 400	0.36997 0.45952	2 2					
t <sub>e</sub> °C	27,43	5	c vap.	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	-					
$T_R = 0.$		t	П			+ 91	rams/100 gra	ms solven	ı <b>t</b>	
REFEREN		ow 2-A	PI 3-Lit. 4-	Calc, from de	t. da					
SOURCE:		AF								
PURIFICA	TION:	AF								
LITERATI										

								No. 8	
NAME	2,3-P	entad	iene				STRUCTURAL	FORMULA	<b>L</b>
Γ									
Mole % Pur.	Ref.	Mole For	ecul mul	ar C <sub>5</sub> H <sub>8</sub>	Molecular Weight 68, 11	4	CH <sub>3</sub> CH=C=C	нсн <sub>3</sub>	
			Ref.			Ref			Ref.
F, P. *C	-125.65	0	2	dt/dP	T		f to		
F.P. 100%				*C/mm		_	g L _ °K		
B. P. °C	10.00	_		25°C BP	0.0768 0.03870	5 4	h		
760 mm 100	48.26		2	t.	0.03505		f' to		
30	-22.1		2	30 mm	0.5377	5	g'  K_		
10 1	-38.4 -65.24	. 1	2	ΔHm cal/g			h'		
Pressure	+	-+	_	AHv cal/g			m   300 to	0.0660	
mm 25°C	320.0	- 1	5	25°C	102.91	5	n 600 °K	0.0011 -0.0 <sub>6</sub> 37	4
t <sub>e</sub>	862.8		5	30 mm BP	114.00 97.70	5	<u> </u>	<del></del>	—
Density	0.40	502	,	t <sub>e</sub> , .	96.86	5	m'   700 to n'   1000 K	0.0790 0.0010	
g/ml 20°C	0.69		2 2	t <sub>e</sub> (d, e)	96.82	5	0' 1000 1	-0.0637	4
d 25 4 30	0.68		4	ΔHv/T <sub>e</sub>	20.28	5	Surface tension	- 0	<del> </del>
	0.71		4	d   -22 to		5 5	dynes/cm. 20°C	19.87	5
ь	-0.03	96	4	52 to		"	30	18.69	5
Ref. Index		842	2	e' j •			40	17.53	-
<sup>n</sup> D 20°C	1.42		2	d g/ml vc ml/g tc °C	l	l	Parachor [P] 20°C		Ì
30	1.42	170	4	tc *C	226.	5	30		
"C"	0.81	87	4	P <sub>c</sub> mm	31482.	5	40 Sugd	2 <b>0</b> 7.2	5
MR (Obs.)	25.23		2	PV/RT	137.102.	<del>  </del>	Exp. L.1.%/wt.	201.2	<del>-</del>
MR (Calc. (nD-d/2)	24.35		5 2	25°C	0.9770	5	u.		l
Dielectric	2.04	$\rightarrow$	5	30 mm BP	1. <b>0</b> 000 0.9570	5	Dispersion	174.5	2
A   -22 to		-	2	t <sub>e</sub>	0.9534	5	Flash Point °C Fire Point		l
В <u>[101 °С</u>	1086.64	.	2	tc	_L				├
C	223.04	+	2	ΔHc kcal/m ΔHf gas	734.55 33.10	2 2	M Spec. Ultra V.		1
A*   -22 to B*   62 °C	1.18		5	ΔHf gas ΔFf gas	49.22	2	X-Ray Dif.		
K - = =	-			Viscosity			Infrared Solubility in +		
c tto	-	j		centistokes	.		Solubility in + Acetone		1
tk to				η •	'		Carbon tet.	00	ŀ
A' to	<del> </del>	+	-				Benzene Ether	90 90	1
B' •	<u>:</u>			B <sup>V</sup>   to	<del> </del>	$\vdash$	n-Heptane	- ×	1
	<del> </del>			B' to			Ethanol Water	<b>∞</b>	l
A'* to B'* *C		-		(BV) to	-		Water in		
Ac   101 to	7, 30	054	5	(A <sup>V</sup> )	1				
Bcit C	1372.39		5	c <sub>p</sub> liq. °K	<del></del>	<del> </del>			1
Cc	263.23		5	l -	ł	1 1	}		1
Cryos. A° consts. B°		l		c <sub>p</sub> vap.300°K 400	0.35675 0.43897				
t <sub>e</sub> °C	52.05		5	c <sub>v</sub> vap.					
$T_R = 0.7$	5 T <sub>c</sub>			<u> </u>		<u> </u>	+ grams/100 gra	ms solveni	<u></u>
REFEREN	ES: 1-D	ow 2	-AI	PI 3-Lit. 4-	Calc. from det	t. dat	ta 5-Calc. by for		
SOURCE:				PI					
PURIFICAT	ION:		A	PI					
LITERATU	RE REFI	EREN	CES	):					

								No. 9	
NAME	3-Methyl-	1,2-1	outadiene			ST	RUCTURAL	FORMUL	A
							CH =C=C (	CH C	
Mole	Ref. Mo	lecul		Molecular			CH <sub>2</sub> =C=C C	3	
% Pur.		mul	C <sub>5</sub> H <sub>8</sub>	Weight 68.11	4				
		Ref.		ļ	Ref.				Ref.
F.P. *C F.P. 1009			dt/dP °C/mm	1		f	to		
B. P. °C			25°C	0.0588	5	g h	'° <u>K</u>		
760 mm	40.	2	BP t <sub>e</sub>	0.03796 0.035 <b>0</b> 7	4 5	f!	to		$\vdash$
100 30	-8. -30.	2 2	30 mm	0.5353	5	g'	<u>*K</u>		
10	-46. -72.69	2 5	AHm cal/g			h'			
Pressure	-12.07	Ť	ΔHv cal/g			m n	300 to 600 °K	0.0588 0.0012	4
mm 25°C	438.6	5	25°C 30 mm	97.50 107.81	5 5	0	1 2000 201	-0.0 <sub>6</sub> 51	4
t <sub>e</sub> Density	840.4	3	BP	94.63 94.08	5	m'	700 to	0.1525	4
g/m1 20°C		2	t <sub>e</sub> (d, e)	94.08	5	n' o'	l   17000 -k	0.0 <sub>3</sub> 91 -0.0 <sub>6</sub> 29	4
dt 25 4 30	0.675 0.67 <b>0</b>	2 4	AHv/Te	20.27	5		L	-0.0627	<u> </u>
a	0.7002	4	d -30 to	102.21	5		face tension es/cm. 20°C	18.16	5
Ъ	-0.0394	4	-di-)- 10 to	0.1894		8	30 40	17.05 15.95	5
Ref. Index		2	e'   °C	ļ		Par	achor [P]	13.73	-
D 25 30	1.407 1.404	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g				20°C		
"C"	0.8028	4	tc °C	212.	5		30 <b>4</b> 0		
MR (Obs.)		4	P <sub>c</sub> mm	31519.	5	<u> </u>		207.2	5
MR (Calc. (nD-d/2)		5 2	PV/RT 25°C	0.9713	5	Exp	u. L.1.%/wt.		
Dielectric	1.99	5	30 mm BP	1.0000 0.9580	5		persion		
A -30 to	7.005	2	t.	0.9555	5		sh Point C e Point		
B ∟90 °C	_ 1130. 234.	2 2	tc ΔHc kcal/m	732.45	2	<u> </u>	Spec.		-
A* -30 to	1,29994	5	ΔHf gas	31.00	2		ra V. Ray Dif.		
B*[_53 °C		5	ΔFf gas	47.47	2		ared		
c	_		Viscosity centistokes		1		ubility in +	_	
t <sub>k</sub> to	:		η •c			Ca	rbon tet.	ος ος	
A'   to	<del> </del>						enzene her	<b>8</b> 0	
B', L°			B <sub>v</sub> to				Heptane hanol	oc oc	
A'* to			B to €C	_		W	ater	•	1
B'* °C	+		(B <sup>V</sup> )  to				ater in		
Ac 90 to		5 5	(A <sup>V</sup> )  °C						
Cc	274.34	5	c <sub>p</sub> liq. °K	ļ					
Cryos, A° consts, B°			c <sub>p</sub> vap.300°K	0.37144	2				
t <sub>e</sub> °C	42.93	5	c <sub>v</sub> vap.	0.45512	2				
$T_R = 0.7$			ı · ·	<del></del>	l	+ 91	rams/100 gra	ms solven	ıt.
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		A	PI						
PURIFICA			PI						
LITERATU	RE REFERE	NCE	S:						

								No. 10	
NAME	2-Methyl	-1,3-	butadiene			STRUC	TURAL 1	FORMULA	
	Isoprene					c	H,=C CH	=CH,	
Mole % Pur.		lecul		Molecular Weight 68.11	4		ch <sub>2</sub> =C CH ch <sub>3</sub>	L	
		Ref.			Ref.				Ref.
F. P. *C	-145.950	2	dt/dP	[		f	to		
F.P. 1009	6		*C/mm 25*C	0.0403	5	g	_ <u>•K</u>		
B. P. *C 760 mm	34, 067	2	BP	0.0492 0.03817	4	h			
100	-14, 21	2	t <sub>e</sub>	0.03582	5	f'	to		
30	-35.5	2	30 mm	0,5315	5	g'	•K_		
10 1	-51.6 -78.08	5	ΔHm cal/g			h'			
Pressure		1	ΔHv cal/g		_	m (	300 to	-0.0351 0.0016	4
mm 25°C		5	25°C 30 mm	92.34 103.42	5	0 !	_000 _1,_	-0.0 <sub>6</sub> 95	4
t <sub>e</sub>	824.2	5	BP	90.67	5	<u> </u>	700 to	0.1880	4
Density g/ml 20°0	0,68095	2	t <sub>e</sub>	90.23	5	m'   n'	11000 eK	0.0,88	4
dt 25	0.67587	2	'e (", c)	90.24	J	0'		0.0 <sub>3</sub> 88 -0.0 <sub>6</sub> 29	4
<sup>4</sup> 30	0,67074	4	ΔHv/T <sub>e</sub>	19.85	5	Surface	tension		
a.	0.7017	4	d   -35 to		5		m. 20°C	18.22	5
b	-0.0395	4		1	1	<b>)</b> *	30 40	17.08 15.95	5
Ref. Index		2	e' i °C	; <u> </u>	<u> </u>	Parache		13.73	$\dashv$
25	1.41852	2	d g/ml v ml/g	ł	ł	Faracin	20°C		
30	1.41524	4	t <sub>c</sub> °C	202.	5		30		
"C"	0.8237	4	P <sub>c</sub> mm	28860.	5	1	40 Sugd.	207.2	5
MR (Obs.)		2	PV/RT		<u> </u>	Exp. L.	1.%/wt.		$\vdash$
MR (Calc. (nD-d/2)	25.626	5 2	25°C	0.9671	5	1	u.		
Dielectric	<del></del>	5	30 mm BP	1.0000	5	Dispers		224.8	2
A   -35 t		2	t <sub>e</sub>	0.9568	5	Flash P			
B84 °	C 1080.996	2	t <sub>c</sub>			M Spec.			
С	234.67	2	ΔHc kcal/m ΔHf	719.55 11.80	2 2	Ultra V			
A*   -35 to B*46 °C		5	ΔFf	11.00	-	X-Ray			
K	- 100/.21		Viscosity			Infrared			
c	_		centistokes	}	l	Solubili Aceton	·y		
tk   to		1	η ∘c			Carbon	tet.		
A' to	<del></del>	+-+			1	Benzer Ether	ne		
B! L '	<u>c </u>	1	_ V	<u> </u>	-	n-Hept			
C'	_	1	B <sup>V</sup>   to A <sup>V</sup>   °C			Ethano Water	ol .		
A'* to B'* *	- 1			-	ĺ	Water	in		
Acl 84 to	7 33735	5	. v. '						
Bcit c	C 1374.92	5				1			
	213.34	5	c <sub>p</sub> liq. °K		1	ll			
Cryos. Acconsts. B			c <sub>p</sub> vap.300°K 400	0.36997 0.46686	2 2				
t <sub>e</sub> °C	36.44	5	c <sub>v</sub> vap.		L				
$T_{\mathbf{R}} = 0.7$						grams	/100 gran	ns solvent	
REFEREN	CES: 1-Dow	2-AF		Calc. from det	. da	ta 5-Cal	c. by for	mula	
SOURCE:		AF	PI						
PURIFICA	TION:	AF	PI						
LITERATU	RE REFERE	NCES	<b>:</b>						

						No. 11				
NAME	1,2-Hexa	liene				STRUCTURAL FORMULA				
Mole % Pur.		lecul mula		Molecular Weight 82.14	.0	CH <sub>2</sub> =C=CH(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>				
		Ref.			Ref.	Re				
F.P. °C F.P. 100%			dt/dP °C/mm			f to g oK				
B. P. °C 760 mm 100	76. 23.40	2	25°C BP t <sub>e</sub>	0.2012 0.04136 0.03497	5 4 5	h   ft   to				
30	0.064	5	30 mm	0,5853	5_	g'   '° <u>K</u>				
10	-17.70 -47.11	5	ΔHm cal/g			h'				
Pressure mm 25°C t <sub>e</sub>	107.7 940.1	5	ΔHv cal/g 25°C 30 mm	98.35 102.88	5	m   to   o				
Density g/ml 20°C	0.7149 0.7102	2 2	BP t <sub>e</sub> t <sub>e</sub> (d, e)	89.22 88.03 87.99	5 5 5	m' to n' K				
dt 25 4 30	0.7055	4	ΔHv/T <sub>e</sub>	20.31	5	S				
a b	0.7337 -0.0 <sub>3</sub> 92	4 4	d 0 to e 83 °C d to	0.1799	5	Surface tension   dynes/cm. 20°C   21.04   5   5   5   5   5   5   5   5   5				
Ref. Index n <sub>D</sub> 20°C 25 30	1.4282 1.4252	2	d g/ml v ml/g			Parachor [P] 20°C				
"C"	1.4222 0.7955	4	vc ml/g tc °C Pc mm			30 40 Sugd. 246. 2 5				
MR (Obs.) MR (Calc. (nD-d/2)		5 2	PV/RT 25°C 30 mm	0.9915 1.0000	5	Exp. L.1.%/wt.				
Dielectric	2.04	5	BP	0.9550	5	Dispersion Flash Point <sup>6</sup> C				
A 0 to B 1 123 °C C	7.03380 1248.39 224.6	4 4 5	te tc ΔHc kcal/m	0.9490	5	Fire Point M. Spec.				
A* 0 to B* 93 °C	1.37445	5	ΔHf ΔFf			Ultra V. X-Ray Dif. Infrared				
K c t <sub>k</sub> to			Viscosity centistokes 7°C			Solubility in + Acetone Carbon tet.				
A'   to B'   _ °C			B <sub>v</sub> to			Benzene Ether n-Heptane				
A¹* to B¹* °C			$ \begin{vmatrix} \mathbf{B} \\ \mathbf{A}^{\mathbf{V}} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   \mathbf{B}^{\mathbf{V}} \mathbf{O} \\   $	-		Ethanol Water Water in				
Ac to Bc t <sub>c</sub> °C			(A <sup>V</sup> )  °C c <sub>p</sub> liq. °K		-					
Cryos, A° consts, B°			c <sub>p</sub> vap. °K							
t <sub>e</sub> °C	82.84	5	c <sub>w</sub> vap.	<u> </u>						
B D D D D D D D D D D D D D D D D D D D		-	<b>.</b>			† grams/100 grams solvent				
	CES: 1-Dow	2-A		Calc. from de	t. da	ita 5-Calc. by formula				
SOURCE:	TION:	AP								
	RE REFERE									
			•							

							No. 12
NAME	l, cis-3-H	exadi	ene			STRUCTURAL FO	
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 82.14	10	CH2=CHCH=CHC	H <sub>2</sub> CH <sub>3</sub>
		Ref.			Ref		Ref.
F, P. *C			dt/dP			f to	
F.P. 100%			*C/mm		_	g	
B. P. *C			25°C BP	0.1807 0.04106	5 4	h	
760 mm 100	73. 20. 78	5	t	0.03498	5	f¹ to	
30	-2.37 -20.00	5	30 mm	0.5808	5	g'	
10 1	-49.18	5	ΔHm cal/g	}		h'	
Pressure	<del> </del>	$\vdash$	ΔHv cal/g			m to to	1
mm 25°C	121.55	5	25°C 30 mm	96.88 101.84	5 5	"	
t <sub>e</sub>	931.49	5	BP	88.32	5		
Density g/ml 20°C	0.705	١,	to (d. a)	87.17	5	m'   to	ļ
dt 25 4 30	0.705	2 2	e (4, 6)	87.16	5	0,	1
<sup>a</sup> 4 30	0.6950	4	ΔHv/T <sub>e</sub>	20.30	5	Surface tension	
	0.7250	4	d   -2 to e   79 °C	101.41 0.1793	5	dynes/cm. 20°C	19.89 5
b	-0.0398	4	a'	0.1173	ا ً ا		18.76 5 17.65 5
Ref. Index n <sub>D</sub> 20°C	1,438	2	e' i •c			Parachor [P]	11.03
D 25	1.435	2	d g/ml v ml/g			20°C	
30	1.432	4	tc C			30	
"C"	0.8241	4	P <sub>c</sub> mm			40 Sugd. 24	46.2 5
MR (Obs.)	30.585	4	PV/RT	<u> </u>	-	Exp. L. 1, %/wt.	
MR (Calc.) (nD-d/2)	30.244 1.086	5 2	25°C	0.9901	5	u.	
Dielectric	2.07	5	30 mm BP	1.0000 0.9550	5		25. 2
A -2 to	<b>↓</b>	4	te	0.9493	5	Flash Point °C Fire Point	ł
B   118 °C	1236.47	4	t <sub>c</sub>				
C	225.1	5	ΔHc kcal/m ΔHf	l I		M Spec. Ultra V.	
A*  -2 to B* 89 °C	1.37294 1160.15	5	ΔFf			X-Ray Dif.	
K - 57 5	1100.13	"	Viscosity			Infrared Solubility in +	
t.  to			centistokes			Solubility in +	
t <sub>k</sub> to	1		η ℃			Carbon tet.	
A' to	<del> </del>	<del>                                     </del>		}	li	Benzene Ether	
B' <u>*C</u>			B <sup>V</sup>   to	<u> </u>		n-Heptane	
C' to		-	B <sup>V</sup> to			Ethanol Water	Í
A'* to B'* °C		1	<u> </u>	{	li	Water in	
Acl to	<b>†</b>	1	(A <sup>V</sup> ) to				
Bc t °C	]		c <sub>p</sub> liq. °K		╁┈┤		1
Cc		-	ll -	1	}		
Cryos. A° consts. B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	79.49	5	c <sub>w</sub> vap.		Ш	+ grams/100 grams	solvent
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-C	alc, from de	t. da	ta 5-Calc. by formu	
SOURCE:			PI				
PURIFICAT	ION:	Al	PI				
	RE REFERE						
			•				

No. 13 1, trans-3-Hexadiene NAME STRUCTURAL FORMULA CH,=CHCH=CHCH,CH, Mole Ref. Molecular Molecular C6H10 % Pur Formula Weight 82.140 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm ٩ĸ g 25°C 0.1807 5 B. P. °C h BP 0.04106 4 760 mm 73. 2 t<sub>e</sub> 0.03498 5 f 100 20.78 5 to g¹ 5 <u>°К</u> 30 -2.370.5808 5 30 mm 10 -20.00 5 h! ∆Hm cal/g 1 -49.18 5 to AHv cal/g Pressure n ۰ĸ 25°C 96.88 5 mm 25°C 121.55 5 o 5 101.84 30 mm te 931.49 5 BP 88.32 5 m to Density 5 te (d, e) 87.17 n' °K g/ml 20°C 0.705 87.16 2 5 o' 25 0.700 2  $d_4^t$ AHv/T 20.30 5 30 0.6950 4 Surface tension -2 d to 101.41 0.1793 5 0.7250 4 19.89 5 dynes/cm. 20°C <u>79 ℃</u> 5 Ъ 18.76 -0.0398 4 30 ď٦ to 5 17.65 40 Ref. Index e¹ <sup>n</sup>D 20°C 1.438 [P] Parachor d<sub>c</sub> g/ml 25 1.435 2 20°C ,с .С ml/g 30 1.432 4 30  $\mathbf{t_c}$ 40 "C" 4 0.82415 P<sub>c</sub> mm 246.2 Sugd. MR (Obs.) 30.585 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30, 244 5 25°C 0.9901 (nD-d/2)1.086 2 30 mm 1.0000 5 225. 2 Dispersion 5 Dielectric 2.07 BP 0.9550 5 Flash Point °C 0.9493 A -2 to B 118 °C 7.02864 1236.47 te tc 44 Fire Point M. Spec. 225.1 5 C AHc kcal/m Ultra V ΔHf 1.37294 -2 to A\*| 5 X-Ray Dif. B\*| 89 °C ΔFf 1160.15 Infrared ĸ Viscosity Solubility in centistokes Acetone t<sub>k</sub> to °C Carbon tet. t<sub>x</sub> °C Benzene A to Ether В °C n-Heptane B<sub>V</sub> C to Ethanol °C A'\* Water to Water in B'\*  $(\mathbf{B}^{\mathbf{V}})$ °C to Acl to (AV) °C Bc °C c<sub>p</sub> liq. °K Cc Cryos, Aº °K cp vap. consts. B° c, vap. te °C 79.49 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 4-Calc, from det. data 3-Lit. 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 14	
NAME	1, cis-4-H	lexad	iene		STRUCTURAL FORMULA			
Mole % Pur.	Ref. Mo	0	CH <sub>2</sub> =CHCH <sub>2</sub> CH=CHCH <sub>3</sub>					
		Ref.			Ref			Ref
F, P. *C			dt/dP		П	f   to		
F.P. 100%			°C/mm		1 _ 1	g   <u>*K</u>		
B. P. *C	4.5		25°C BP	0.1363 0.04028	5 4	h ;		
760 mm 100	65. 13.80	5	te	0.03499	5	f¹ to		
30	-8.88	5	30 mm	0.5689	5	g'		
10 1	-26.15 -54.71	5	ΔHm cal/g			h¹		
Pressure	-51.11	-	ΔHv cal/g			m   to		1
mm 25°C	167.20	5	25°C	92.97	5 5	n ' •K_		
t <sub>e</sub>	908.62	5	30 mm BP	99.03 85.91	5	<del>!</del>		├
Density	0.700		te (d.e)	84.95	5	m'   to		
g/ml 20°C	0.700 0.695	2 2	e (4, 6)	84.93	5	0,   =		
d 25 4 30	0.690	4	ΔHv/T <sub>e</sub>	20.30	5	Surface tension		-
a	0.7200	4	d   -9 to		5	dynes/cm. 20°C	19.31	5
ь	-0.0397	4				30	18.19	5
Ref. Index	1 415	١,	e' j *(	<u> </u>		40	17.10	5
n <sub>D</sub> 20°C	1.415	2 2	d g/ml vc ml/g			Parachor [P] 20°C		
30	1.409	4	v <sub>c</sub> ml/g t <sub>c</sub> °C			30		1
"C"	0.7888	4	P mm			40 Sugal	246.2	5
MR (Obs.)	29.385	4	P <sub>c</sub> mm PV/RT	<b>+</b>		Exp. L.1.%/wt.	240.2	+-
MR (Calc.) (nD-d/2)	28.974 1.065	5 2	25°C	0.9860	5	u.		
Dielectric	2,00	5	30 mm BP	1.0000	5 5	Dispersion		
A -9 to	7.01701	4	t <sub>e</sub>	0.9550 0.9500	5	Flash Point °C		
B 1 108 °C		4	tc	l		Fire Point		├
<u>c                                    </u>	226.6	5	ΔHc kcal/m			M Spec. Ultra V.		
A*  -9 to	1.37095	5	ΔHf ΔFf			X-Ray Dif.		
B*	1131.31	5	Viscosity		$\vdash$	Infrared		<u> </u>
c			centistokes			Solubility in + Acetone		
tk to	İ		η ∘c			Carbon tet.		
A' to	<del> </del>	-				Benzene Ether		
B'		1	<u> </u>	<del></del>		n-Heptane		1
C'	ļ	ļ	B <sup>V</sup>   to			Ethanol Water		
A'* to B'* °C		1		-1		Water in		1
Ac to	<del> </del>		, v.'	1				
Bc t C	j	1	<del></del>	+	$\vdash$			
Cc	·!	<b>L</b>	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> *C	70.57	5	c <sub>v</sub> vap.			+ ~~~~ (100		<u></u>
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-	Calc, from de	t det	grams/100 grants  'grams/100 grants  'grams/100 grants		<u> </u>
SOURCE:		Al		ue	., ual	- J-Carc. by for		
PURIFICAT	ION:	AJ	***					
	RE REFERE							
			•					

NAME	l, trans-4							
1	1, 114113-1	-Hex	adiene		STRUCTURAL FORMULA			
			CH -CHCH CH CHCH					
Mole % Pur.	Ref. Mo	lecul: rmula		Molecular Veight 82.140	,	CH <sub>2</sub> =CHCH <sub>2</sub> CF	н=СнСн <sub>3</sub>	
		Ref.			Ref.			Ref.
F.P. °C F.P. 100%			dt/dP °C/mm			f to		
B. P. °C 760 mm	65.	,	25°C <b>B</b> P	0.1363 0.04028	5 4	h		
100	13.80	5	t <sub>e</sub>	0.03496	5.	f' to		
30 10	-8.88 -26.15	5	30 mm	0.5689	5	h' <u></u>		
11	-54.71	5	ΔHm cal/g ΔHv cal/g		+	m to		
Pressure mm 25°C	167.20	5	25°C	93.02	5	n   <u>*K</u>		
t <sub>e</sub>	909.69	5	30 mm BP	99.03 86.00	5 5	m¹ to		
Density g/ml 20°C	0,700	2	te te (d, e)	85.03 85.02	5	n' K		
dt 25	0.695	2	ΔHv/T <sub>e</sub>	20.31	5	0'		
a 30	0.6900	4	d -9 to	97.46	5	Surface tension dynes/cm. 20°C	19.31	5
ь	-0.0397	4	_e	0.1762	5	8 30	18.19	5
Ref. Index	1.415	2	e'   °C			Parachor [P]	17.10	5
25	1.412	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g			20°C	·	
30 "C"	0.7888	4	tc °C			30 40		
MR (Obs.)	29.385	4	P <sub>c</sub> mm			Sugd.	246.2	5
MR (Calc.)	28.974	5	PV/RT 25°C	0.9866	5	Exp. L.1.%/wt. u.		
(nD-d/2) Dielectric	1.065 2.00		30 mm BP	1.0000	5	Dispersion		
A -9 to	7.01701	4	t	0.9560 0.9510	5	Flash Point C Fire Point		
B   108 °C_	1206.11 226.6	<b>4</b> 5	-c			M. Spec.		
A* -9 to	1.36916	5	ΔHc kcal/m ΔHf			Ultra V. X-Ray Dif.		
B* _81 °C	1130.92	5	ΔFf		$ \parallel$	Infrared		
K			Viscosity centistokes			Solubility in +		
t <sub>k</sub> to			η °c			Acetone Carbon tet.		ı
A'  to		$\vdash$				Benzene Ether		i
B'°C			Bv to			n-Heptane Ethanol		
A'* to						Water		i
Bi* °C	ļ		(B <sup>V</sup> )  to			Water in	<del>  </del>	
Ac to Bc tc °C			(A <sup>V</sup> )  °C		1			1
Cc — —		$\sqcup$	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	70, 61	5	c <sub>v</sub> vap.			L	L	
REFERENC	ES: 1-Dow	2. 4	PI 3-Lit. 4-0	Calc from do	+ 4-	grams/100 gra ta 5-Calc, by for		ŧ
SOURCE:			PI 3-LR. 4-0	Jake. If Offi de	s. ua	a J-Caic, by 101		
PURIFICAT	ION:		PI	<del></del>				
LITERATUR	RE REFERE	NCES	<b>3</b> :					

						<b>No.</b> 16		
NAME	1,5-Hexad	li <b>e</b> ne			STRUCTURAL FORMULA			
Mole	Ref. Mo	lecul	ar	Molecular		CH <sub>2</sub> =CH(CH <sub>2</sub> ) <sub>2</sub> CH=CH <sub>2</sub>	2	
% Pur.	Fo	rmul	ar C <sub>6</sub> H <sub>10</sub>	Weight 82.14	10			
	<b></b>	Ref.		<b>_</b>	Ref.		Ref	
F.P. °C	<del>,  </del>	-	dt/dP *C/mm	1		f to oK	ļ	
B. P. *C	-	$\vdash$	25°C	0.1125	5	g   <u>*K</u>		
760 mm 100	59.46 8.95	2 5	BP t <sub>e</sub>	0.03976 0.03499	<b>4</b> 5	f' to	$\vdash$	
30	-13.43	5	30 mm	0.5609	5	g!K		
10 1	-30.44 -58.5 <b>9</b>	5	ΔHm cal/g			h'	L.	
Pressure			ΔHv cal/g	90,27	5	m to K		
mm 25°C	208.01 893.84	5	25°C 30 mm	97.02	5	0		
Density	+ 075.01	<u> </u>	BP	84.29 83.44	5 <b>5</b>	m¹   to		
g/ml 20°C		2	t <sub>e</sub> (d, e)	83.42	5	n'   <u>*K</u>		
dt 25 4 30	0.6878 0.6833	2 4	ΔHv/T <sub>e</sub>	20.30	5		├-	
	0.7104	4	d   -13 to		5	Surface tension dynes/cm. 20°C 18.46	5	
b D. C. T. I.	-0.0386	4		5		30 17.47 40 16.50	5	
Ref. Index		2	e'   •c		-	Parachor [P]	Ť	
25 30	1.4010 1.3978	2	d g/ml vc ml/g			20°C		
"C"	0.7780	4	ic o			40		
MR (Obs.)		4	P <sub>c</sub> mm			Sugd. 246, 2	5	
MR (Calc. (nD-d/2)	28.974 1.0580	5 2	PV/RT 25°C	0.9834	5	Exp. L.1.%/wt.		
Dielectric	1.97	5	30 mm BP	1.0000 0.9560	<b>5</b>	Dispersion	_	
A   -13 to	7.00740	4	te	0.9515	5	Flash Point °C Fire Point	ŀ	
B 1 102 °C	2 1184.99 227.7	4 5	t <sub>c</sub>	<del> </del>	-	M Spec.		
A*   -13 to		5	ΔHf	1		Ultra V. X-Ray Dif.		
B*	1110.81	5	ΔFf Viscosity	<del></del>		Infrared		
¢ .— —.:	_	l	centistokes	1		Solubility in + Acetone		
t <sub>k</sub> to			η °C			Carbon tet. Benzene	1	
A'   to						Ether		
B' °	4		B <sup>V</sup>   to			n-Heptane Ethanol		
A'* to			A <sup>v</sup> C	_]		Water Water in	l	
B'* °C	<del></del>	-	(BV) to	1		4042 711	<del>                                     </del>	
Bc t °C			(A <sup>V</sup> ) <sub> </sub> °C	<del></del>				
Cryos, A	<del>}</del>	-	P -					
consts. B°			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	64.45	5	c <sub>v</sub> vap.				L	
D. D. D. D. D. D. D. D. D. D. D. D. D. D						grams/100 grams solven	t	
SOURCE:	CES: 1-Dow	Z-AI	PI 3-Lit, 4-0 PI	Calc. from det	t. da	ta 5-Calc, by formula		
PURIFICA?	TION:		PI					
	RE REFERE							

No. 17 2, 3-Hexadiene STRUCTURAL FORMULA NAME CH2CH=C=CHCH2CH3 Molecular C6H10 Mole Ref. **Molecular** % Pur Weight 82.140 Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C °K g 0.1514 5 B. P. °C h ВP 0.04057 4 760 mm 68.0 2 t<sub>e</sub> 0.03495 5 f١ to 100 16.42 5 g <u>«к</u> 30 -6.45 5 30 mm 5 0.5734 10 -23.85 -52.65 5 h! AHm cal/g 5 1 to m ∆Hv cal/g Pressure ۰ĸ n 25°C 94.47 mm 25°C 148.45 5 0 30 mm 100.07 5 918, 28 5 t<sub>e</sub> BP 86.90 5 m' to Density te te (d, e) 5 85.86 n' °K g/ml 20°C 0.680 2 85.85 5 01  $\mathbf{d_4^t}$ 25 0.675 2 ΔHv/T<sub>e</sub> 5 20,32 30 0.670 4 Surface tension -6 98.93 5 d to 0,7000 dynes/cm. 20°C 17.20 5 74 •C 0.1769 5 ь -0.03974 30 16.18 5 a 5 40 15.19 Ref. Index e' [P] 20°C 1.395 2 Parachor <sup>n</sup>D d<sub>c</sub> g/ml 25 2 20°C 1.392 vc ml/g t °C 30 1.389 4 30 t<sub>c</sub> 40 "C" 0,7750 4 P<sub>c</sub> mm Sugd. 246.2 5 28.95 28.974 MR (Obs.) 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 5 25°C 0.9881 5 5 5 (nD-d/2) 1.055 2 30 mm 1.0000 Dispersion Dielectric 1.95 5 ВP 0.9560 Flash Point C 0.9507 5 A -6 to 7.02235 4 Fire Point 110 °C В 1218.02 M. Spec. С 226.1 5 AHc kcal/m Ultra V. A\* -6 to B\* 84 °C ΔHf 1.37077 5 X-Ray Dif. ΔFf 1142.22 Infrared ĸ Viscosity Solubility in centistokes Acetone Carbon tet. °C  $\mathbf{t_x}$ Benzene ΑΊ to Ether B' °C n-Heptane B<sub>v</sub> C' to Ethanol °C Water A'\* to Water in (BV) B'\* °C to Acl to (A<sup>V</sup>) °C Βc °C c<sub>p</sub> liq. ۰ĸ Cc' Cryos. A° c<sub>p</sub> vap. •ĸ consts. B° c<sub>v</sub> vap. te °C 5 73.95 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula API SOURCE: API **PURIFICATION:** LITERATURE REFERENCES:

NAME cis-2, cis-4-Hexadiene						$\neg$	No. 18				
NAME	C 18 - 2	nexadiene	$\dashv$	STRUCTURAL FORMULA							
Mole % Pur.							сн <sub>3</sub> сн=снсн=снсн <sub>3</sub>				
		Re				Ref				Ref	
F.P. °C			dt/dP				f	to		T	
F.P. 100%			*C/mm		0 2225	ا ۔ ا	g	<u>•ĸ</u>			
B. P. *C 760 mm	١.,	١,	25°C BP		0.2325 0.04175	5 4	h				
100	80. 26.89	5			0.03492	5	£	to			
30 10	3, 32				0.5912	5	g'   '	_ <u>_</u> K_	-		
10	-14.62 -44.34			g			h'		ļ		
Pressure	†		AHv cal/g	3	100 38		m n	to °K	ĺ		
mm 25°C	91.55				100.38 104.30	5 5			1		
t <sub>e</sub>	953.22	5	BP		90.57	5	m' i	to	<u> </u>	+	
Density g/ml 20°C	0.72	0 2	te (d.s)		89.28 89.25	5 5	n'	_ °K			
<sub>a</sub> t 25	0.71	5 2	'e \-', -'		20.34	5	o'		]		
<sup>4</sup> 30	0,71		4 1 2	<b>A</b>	104.90	5	Surface tens	ion		T	
a b	0.74		- 1 07	to ℃		5	dynes/cm.	20°C	21.65	5	
Ref. Index	-0.03	70 7	-   d'	to				3 <b>0</b> 40	20.45 19.28	5	
n <sub>D</sub> 20°C			e¹	°C		$\vdash$	Parachor [	P]		+-	
- 25	1.44				i	1 1	1	20°C		1	
30 "C"	1.44		-11 to 10				1	30 40			
	0,82		∥P_mm		ł				246.2	5	
MR (Obs.) MR (Calc.)			PV/RT				Exp. L.1.%	/wt.		$\top$	
(nD-d/2)	1.09				0.9937 1.0000	5 5	u.		225.	2	
Dielectric	2.10	5			0.9565	5	Dispersion Flash Point	•	225.	+-	
A 3 to					0.9502	5	Fire Point	C			
B [_128 •C	1263.15 223.8	4 5				$\vdash$	M Spec.			1	
A*  3 to	+		-	m		li	Ultra V.				
B*  97 °C						Ш	X-Ray Dif. Infrared			İ	
к — — —	1	-	Viscosity		1		Solubility in	+	<u> </u>	+	
\$ - to	-	!	centistoke	<b>"</b> ℃	l		Acetone			1	
t <sub>X</sub> i °C		İ	'			1 1	Carbon tet. Benzene	•	ŀ		
A' to							Ether				
B', L _ •	-	1	B <sup>V</sup>	to		$\vdash$	n-Heptane Ethanol				
A'* to	<b>†</b>		AV	•c			Water				
B'* °C	:		(B <sup>V</sup> )	to	1		Water in			+	
Ac  to			(A <sup>V</sup> )	°C							
Bc tc C	-		c <sub>p</sub> liq.	°К							
Cryos. A°			c <sub>p</sub> vap.	۰ĸ							
consts. B°	<u> </u>		<b>-</b> ∥								
t <sub>e</sub> °C	87.36	5	vep.		<u> </u>		+ (1.00		<u></u>		
REFERENC	ES: 1-D	ow 2-1	API 3-Lit.	4-0	alc. from det	t. dat	grams/100 a 5-Calc.b			10	
SOURCE:			API				- 32-0. 0				
PURIFICAT	ION:		API								
LITERATU		ERENCI	CS:		· · · · · · · · · · · · · · · · · · ·						

No. 19 cis-2, trans-4-Hexadiene NAME STRUCTURAL FORMULA CH3CH=CHCH=CHCH3 Molecular C6H10 Mole Ref. Molecular % Pur. Weight 82.140 Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C °<u>K</u> g 0.2325 0.04175 5 4 B. P. °C h ВP 760 mm 80. 2 0.03498 5 ſ١ 100 26.89 5 to <u>°К</u> 3.32 5 5 g' 30 0.5912 30 mm 10 -14.62 5 h' ∆Hm cal/g 5 1 -44.34 to AHv cal/g Pressure n ۰ĸ 100.31 25°C 5 91.55 mm 25°C o 30 mm 104.30 5 5 950.95 5 te ΒP 90.38 m to Density 89.10 5 te te (d, e) n' °K g/ml 20°C 89.06 5 0.720 2 o'  $\mathbf{d_{4}^{t}}$ 25 0.715 2 AHv/Te 20,30 5 30 0.710 4 Surface tension d 3 104.90 5 0.7400 a b dynes/cm. 20°C 21.65 5 °C 0.1816 5 ᇷᅥ 87 -0.0398 4 30 20.45 5 to 19.28 5 40 Ref. Index e' °C 20°C 1.450 2 Parachor [P] nD d<sub>c</sub> g/ml 25 1.447 2 20°C vc ml/g t °C 30 1.444 4 30  $\mathbf{t_c}$ "C" 40 0.8277 4 P<sub>c</sub> mm 5 246.2 Sugd. MR (Obs.) 30,658 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30,244 5 25°C 0.9930 (nD-d/2) 1.090 2 1.0000 30 mm 5 Dispersion 225. 2 Dielectric 2.10 5 BP 0.9545 5 Flash Point C 0.9481 A 3 to 7,03864 t<sub>e</sub> Fire Point B | 127 °C 1263.15 M. Spec. c 223.8 5 AHc kcal/m Ultra V ΔHf A\*Ī 3 to 1.37555 X-Ray Dif. ΔFf B\*| 97 °C 1185.70 Infrared K Viscosity Solubility in centistokes to Acetone Carbon tet. ٠c  $\mathbf{t_x}$ Benzene A۱ to Ether B١ °C n-Heptane B<sup>v</sup> | C' to Ethanol °C A'\* Water to ٠c Water in B'\* (B<sup>V</sup>)| to Acl (A<sup>V</sup>) to °C Bc °C c<sub>p</sub> liq. ۰ĸ Cc Cryos. Aº c<sub>p</sub> vap. °K consts. B° te °C c, vap. 87.28 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API **PURIFICATION:** LITERATURE REFERENCES:

							No. 20	
NAME	trans	-2, tran	ns-4-Hexadiene			STRUCTURAL		
Mole	Ref.	Moles		Molecular		сн <sub>3</sub> сн=снсн	=CHCH <sub>3</sub>	
% Pur.	Kei.	Form	ular C <sub>6</sub> H <sub>10</sub>	Weight 82.14	10			
		Re	f.		Ref.			Ref.
F.P. °C F.P. 1007			dt/dP			f to		1
B. P. °C	<del>'</del>		*C/mm 25*C	0, 2325	5	g <u>°K</u>	ł	1
760 mm	80. 26.89	2 5		0.04175 0.03498	4 5	f' to		+
100 30	3.32	5	30 mm	0.5912	5	g'   'K_		1
10 1	-14.62 -44.34					h'		ļ
Pressure	+		ΔHv cal/g	100 31	5	m to	1	1
mm 25°C	91.55 950.95		30 mm	100.31 104.30	5	0  =	1	1
Density	730.73		—∥ вь	90.38 89.10	5	m'   to		
g/m1 20°C			11 6	89.06	5	n'   ' <u>*</u> K	-	
dt 25 4 30	0.71			20.30	5	Surface tension		┼
a	0.74			o 104.90 C 0.1816	5	dynes/cm. 20°C	21.65	5
b Ref. Index	-0.03	98 4	- a'	5		30 40	20.45 19.28	5
n <sub>D</sub> 20°0	1.45		4 - /1	-		Parachor [P]		T
25 30	1.44		V mi/g			20°C 30		
"C"	0, 82	77 4	11 ~	1		40	246.2	5
MR (Obs.)				+		Exp. L.1.%/wt.	246,2	+3
MR (Calc. (nD-d/2)	30.24		25°C	0.9930	5	u.		
Dielectric	2.10	5	30 mm BP	1.0000 0.9545	5	Dispersion Flash Point °C	225.	2
A 3 to				0.9481	5	Fire Point		<u> </u>
B (_127 °C	223.8	5	AHc kcal/m	+		M Spec.		
A*  3 to						Ultra V. X-Ray Dif.		
B* ⊢ 97 °C	1185.70	5	Viscosity			Infrared Solubility in +		+
t <sub>k</sub>	-	į	centistokes	-		Solubility in + Acetone		-
t <sub>X</sub> °C			] '			Carbon tet. Benzene		
A' to						Ether n-Heptane		1
C'	<u>-</u>		B <sup>V</sup> to			Ethanol		}
A!# to B!# *0			$\frac{1}{ B^{\vee} } = \frac{1}{t}$	-		Water Water in		1
Acl to	<del></del>		$- \begin{vmatrix} (B^{\vee})   & to \\ (A^{\vee})   & \circ C \end{vmatrix}$	1				
Bc tc_*(			c <sub>p</sub> liq. •k	<del></del>	$\vdash$			
Cryos. A	+			1				
consts. B			_ P • • •				1	
t <sub>e</sub> °C	87.28	5	c <sub>v</sub> vap.			L <u>,</u>		<u> </u>
DEFEDEN	CEC. 1 D		D. 2			grams/100 gran		ıt
SOURCE:	. E3: 1-D		API 3-Lit. 4- API	Calc. from de	t. da	ta 5-Calc. by for	mula	
PURIFICA'	TION:		API	· <del></del>				
LITERATU								
· · · · · · · · · · · · · · · · · · ·								

No. 21 3-Methyl-1, 2-pentadiene NAME STRUCTURAL FORMULA CH2=C=CCH2CH3 Molecular C6H10 ĊH Mole Ref. Molecular Weight 82,140 % Pur Ref. Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm °K g 25°C 0.1625 5 B.P. °C h ВP 0.04077 4 760 mm 70. 2 0.0350 5 ſ١ 100 18.17 5 °<u>K</u> g' 30 -4.82 30 mm 0.5764 5 10 -22.31 5 h' ∆Hm cal/g 1 -51.26 5 m to ∆Hv cal/g Pressure n °K 25°C 95.38 mm 25°C 137.1 5 o 30 mm 100.78 5 t. 922.4 5 BP 5 87.38 m¹ Density to 86.30 5 n' g/ml 20°C °K\_ te (d, e) 0.715 2 5 86,28 ۰'  $d_4^t$ 25 0.710 2 ΔHv/T 20.29 5 30 • 0.705 4 Surface tension d -5 99.91 0.7350 4 4 dynes/cm. 20°C 21.03 °C 0.1791 76 -0.0398 ь 30 19.85 5 to 40 18.69 5 Ref. Index e¹ °C 20°C 1.425 2  $^{n}D$ [P] Parachor d<sub>c</sub> g/ml 25 1.422 2 ۍ. °C 20°C ml/g 30 4 1.419 30  $\mathbf{t_c}$ 40 "C" 0.7898 4  $P_c$  mm 5 Sugd. 246.2 MR (Obs.) 29.375 4 PV/RT Exp. L. 1. %/wt. MR (Calc.) 5 28,974 25°C 0.9884 1.0000 5 (nD-d/2)1.068 2 30 mm Dispersion Dielectric 2.03 5 BP 0.9545 Flash Point C 0.9490 A -5 to 7.02488 4 Fire Point B [116 °C] 1225.40 4 M. Spec. C 225.7 5 ∆Hc kcal/m Ultra V ΔHf A\* -5 to 1.37362 5 X-Ray Dif. ΔFf B\* 86 °C 1149.83 Infrared ĸ Viscosity Solubility in centistokes Acetone tō tk tx Carbon tet. °C Benzene A١ to Ether В' °C n-Heptane  $\overline{B_{\mathbf{v}}^{\mathbf{v}}}$ C to Ethanol °C Water A'\* to Water in B'\* °C (B<sup>V</sup>)| to Acl (AV) to °C Βc °C cp liq. ۰ĸ Cc Cryos. A° c<sub>p</sub> vap. °K consts. B° c vap. te °C 76.12 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API API PURIFICATION: LITERATURE REFERENCES:

						N	o. 22
NAME	4-Methyl-	1,2-	pentadiene			STRUCTURAL FOR	MULA
						CH2=C=CHCH C	н.
Mole	Ref. Mo	lecul		Molecular		Ċн <sub>3</sub>	3
% Pur.		rmul		Weight 82.14	0	J	
		Ref.			Ref.		Ref
F.P. °C F.P. 1007			dt/dP			f to	ŀ
B. P. *C		-	*C/mm 25°C	0.1625	5	g   <u>*K</u>	l
760 mm	70.	2	BP	0.04077 0.03498	4 5	f' to	
100 30	18.17 -4.82	5	t <sub>e</sub> 30 mm	0.5764	5	g'   'K_	
10 1	-22.31 -51.26	5	ΔHm cal/g		$\Box$	h'	
Pressure	-51.20	-	ΔHv cal/g	1		m to	
mm 25°C	137.08	5	25°C 30 mm	95.40 100.78	5 5	ö	
Deneity	922.91	5	BP	87.42	5	m'   to	
Density g/ml 20°0	0.708	2	te (d, e)	86.35 86.32	5	n'	
dt 25	0.703 0.698	2 4	AHV/T	20.30	5	0'	
	0.7280	4	d   -5 to		5	Surface tension dynes/cm. 20°C 20	0.22 5
ь	-0.0398	4	-a,  -76 %	5	5	30   19	9.07 5 7.95 5
Ref. Index		2	e' i °0	<u> </u>	$\vdash$	Parachor [P]	1.73
25	1.421	2	d <sub>c</sub> g/ml v <sub>c</sub> ml/g			20°C	
"C"	0.7959	4	tc °C	1		30 40	l
MR (Obs.)	+	4	P <sub>c</sub> mm	<u> </u>		Sugd. 246	5, 2 5
MR (Calc.	) 28.974	5 2	PV/RT 25°C	0.9886	5	Exp. L.1.%/wt.	
(nD-d/2) Dielectric	2,03	5	30 mm BP	1.0000	5	Dispersion	
A -5 to		4	t <sub>e</sub>	0.9550 0.9495	5	Flash Point °C Fire Point	1
B L115*9	1225.40	4 5	t <sub>c</sub>	<u> </u>		M Spec.	
A*  -5 to	<del>+</del>	5	ΔHc kcal/m ΔHf			Ultra V.	1
B* 86°0	1149.63	5	ΔFf	<b></b>		X-Ray Dif. Infrared	
K	1		Viscosity centistokes			Solubility in +	
t <sub>k</sub>  tc			η •	;		Acetone Carbon tet.	
t <sub>x</sub> ' °C						Benzene Ether	
B' °	2		B <sup>V</sup>   to	<del> </del>		n-Heptane	Ì
A'* to			B' to			Ethanol Water	
B'* °(			(B <sup>V</sup> ) to	7		Water in	
Ac   to			(A <sup>V</sup> )  •c	:			
Bc t <sub>c</sub> °C			c <sub>p</sub> liq. °K	·			
Cryos, A <sup>c</sup> consts, B <sup>c</sup>			с <sub>р</sub> <b>va</b> p. °К				
t <sub>e</sub> °C	76.14	5	c <sub>v</sub> vap.				
REFEREN	CES: 1-Dow	2 . 41	PI 3-Lit. 4-	Cala (==== 1 :		grams/100 grams	solvent
SOURCE:			PI 3-Lit. 4-	Carc. from de	. aai	ta 5-Calc. by formula	1
PURIFICA	TION:		PI				<del></del>
	RE REFERE						

No. 23 2-Methyl-1, cis-3-pentadiene NAME STRUCTURAL FORMULA CH2=CCH=CHCH3 Molecular C6H10 ċн<sub>3</sub> Ref. Mole Molecular % Pur. Weight 82,140 Formula Ref Ref. Ref. F.P. °C dt/dP f to F.P. 100% °C/mm 25°C g <u>.°K</u> 0.2012 5 B. P. °C h ВP 0.04136 4 760 mm 76. 2 <sup>t</sup>e 0.03497 5 ſ١ 100 to 23.40 5 g' °<u>K</u> 30 5 0.5853 5 0.64 30 mm 10 -17.70 5 h' AHm cal/g 5 -47.11 m to AHv cal/g Pressure °K n 25°C 98.35 5 mm 25°C 107.71 5 o 30 mm 102.88 5 940.07 5 te ΒP 89.22 5 m' to Density te (d, e) 5 5 88 03 °K g/ml 20°C 0.719 2 87.99 o' 25 0.714 2  $\mathbf{d_{4}^{t}}$ AHv/T 20.31 5 30 0.709 4 Surface tension ī to 102.89 5 0,7390 dynes/cm. 20°C a 21.52 5 °C 0.1799 -0.0398 ь 4 20.32 5 30 ď٠ to 5 40 19.16 Ref. Index e' <sup>n</sup>D 20°C 1.446 2 [P] Parachor dc g/mi 25 1.443 2 20°C vc ml/g t °C 30 1.440 4 30 tc "C" 40 4 0.8219 P<sub>c</sub> mm Sugd 246.2 5 30, 465 MR (Obs.) 4 PV/RT Exp. L. 1. %/wt. MR (Calc.) 30.244 25°C 0.9915 (nD-d/2)1.087 2 30 mm 1.0000 5 2 Dispersion 225. Dielectric 2.09 5 BP 0.9550 5 Flash Point °C 0.9490 te tc 7.03380 l to 4 Fire Point B (123 °C) 1248.39 M. Spec. 224.6 AHc kcal/m Ultra V. ΔHf A\* 1 to 1.37445 5 X-Ray Dif. ΔFf 9<u>3</u> ℃ 1171.46 B\* Infrared ĸ Viscosity Solubility in c centistokes Acetone to °C ٠c Carbon tet. Benzene A' to Ether B °C n-Heptane B<sub>V</sub> | C' to Ethanol °C Water A'\* to Water in (BV) B!\* °C to Acl to (AV) °C  $Bc_1$ °C c<sub>p</sub> liq. ۰ĸ Cc Cryos, A° ٩K cp vap. consts. B° c<sub>v</sub> vap. te °C 82.84 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

No. 24 NAME 2-Methyl-1, trans-3-pentadiene STRUCTURAL FORMULA CH2=CCH=CHCH3 Molecular C6H10 ĊH3 Mole Ref. Molecular Weight 82.140 % Pur. Formula Ref. Ref. Ref F.P. \*C F.P. 100% dt/dP to °C/mm <u>°</u>K g 25°C 0.2012 5 B. P. °C h BP 0.04136 4 760 mm 76. 2 0.03499 5 ſ١ to ŧ. 100 23, 40 5 °K g' 30 0.06 5 30 mm 5 0.5853 10 -17.70 5 h' ΔHm cal/g 1 -47.11 5 to m ΔHv cal/g Pressure °K n 25°C 98.33 5 mm 25°C 107.71 5 o 30 mm 102.88 939.51 5 t<sub>e</sub> BP 5 89.17 m١ to Density 87.98 5 te (d, e) ٩ĸ g/ml 20°C 0.719 2 5 87.94 01 25 0.714 2 ď4 ΔHv/Te 20.30 5 30 0.709 4 Surface tension 102.89 0 to 5 0.7390 4 dynes/cm. 20°C 21.52 83 •C 0.1805 ь 4 -0.0398 5 30 20.32 ď to 40 19.16 5 Ref. Index •C 20°C 1.446 [P] 2  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 25 2 4 1.443 20°C ml/g t<sub>C</sub> 1.440 30 30 •C 40 "C" 0.8219 4 Sugd. 5 mm 246.2 MR (Obs.) 30.465 4 PV/RT Exp. L.1. %/wt. MR (Calc.) 30,244 5 25°C 0.9913 5 ź (nD-d/2)1.087 30 mm 1.0000 Dispersion 2 5 225. Dielectric 2.09 5 BP 0.9545 5 Flash Point °C 0.9485 7.03380 0 to 4 Fire Point tc В 123°C 1248.39 4 M Spec. C 224.6 5 AHc kcal/m Ultra V ΔHf A\*| 0 to 1.37535 5 X-Ray Dif. ΔFf B\* 93 °C 1171.66 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in Acetone t<sub>x</sub> to Carbon tet. •c Benzene A' to Ether B١ •c n-Heptane Ċ١ В Ethanol ·c A'\* Water to Water in (BV) B'\* •c to Acl to (AV) °C Bc ٠c cp liq. °K Cc Cryos. A° consts. B° c<sub>p</sub> vap. °K t<sub>e</sub> °C c, vap. 82.82 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 25

ГТ-							No. 2	
NAME	3-Methyl-	l, ci	s-3-pentadiene			STRUCTURAL	FORMUL	-A
	<del></del>					CH <sub>2</sub> =CHC=C	нсн,	
Mole	Ref. Mo	ecul	ar	Molecular		с́н <sub>з</sub>	3	
% Pur.	For	mula		Weight 82.140	)	3		
		Ref.			Ref.			Ref.
F. P. °C			dt/dP			f to		
F.P. 100%	ļ	$\vdash$	°C/mm 25°C	0.2086	5	g ' <u>°K</u>		
B. P. °C 760 mm	77.	2	BP	0.04145	4	h		—
100	24.27	5	t <sub>e</sub>	0.03499	5	f' to		1
30 10	0.88	5 5	30 mm	0.5868	5	h'		
1	-46.42	5	ΔHm cal/g	<b>_</b>		m l to		+
Pressure mm 25°C	102 42	5	ΔHv cal/g 25°C	98.83	5	n ºK		
mm 25°C	103.43 942.37	5	30 mm	103.24	5	°		
Density		$\vdash$	BP	89.50 88.26	5	m! to		
g/m1 20°C	0.735 0.730	2 2	le (4, 6)	88.25	5	n' LºK		
dt 25 4 30	0.730	4	ΔHv/T <sub>e</sub>	20.30	5			$\vdash$
a	0.7550	4	d   1 to e   84 °C	103.39 0.1804	5	Surface tension dynes/cm. 20°C	23.51	5
b	-0.0398	4	-ă	0.1004		30 40	22.23 20.98	5
Ref. Index n <sub>D</sub> 20°C	1.452	2	e' _ °C	ļ	<u> </u>	Parachor [P]	20.70	+
D 25	1.449	2	d g/ml			20°C		
30	1.446	4	vc ml/g tc °C			30 40		1
MR (Obs.)	0.8142 30.148	4	P <sub>c</sub> mm			Sugd.	246.2	5
MR (Calc.)	30.244	5	PV/RT 25°C	0.0017	5	Exp. L.1.%/wt.		
(nD-d/2)	1.085	2	30 mm	0.9917 1.0000	5	u. Dispersion	225.	2
Dielectric A 1 to	7, 03502	5	BP t <sub>e</sub>	0.9545 0.9484	5	Flash Point °C Fire Point		
B   126 ℃ C	1252.08	4 5	tc AHc kcal/m		-	M. Spec.		+
A* 1 to	1.37541	5	ΔHf ΔFf			Ultra V. X-Ray Dif.		
B* _ 94 °C K	1175.17	5	Viscosity		<del> </del>	Infrared		ļ
c			centistokes		ļ	Solubility in * Acetone		
t <sub>k</sub> to °C			η °C	}		Carbon tet. Benzene		
A'   to				İ		Ether		
B'°C_	•		B <sub>v</sub> to			n-Heptane Ethanol		
A¹* to			AV I C		}	Water		
B¹* °C			(B <sup>V</sup> )  to			Water in		+
Ac to Bc t <sub>c</sub> °C			(A <sup>V</sup> )  °C	<b></b>	<u> </u>			
ac c			c <sub>p</sub> liq. *K					
Cryos. A° consts. B°			c <sub>p</sub> vap. *K					
t <sub>e</sub> °C	83.93	5	c <sub>w</sub> vap.					l
						grams/100 gra	ms solve	nt
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:			.PI					
PURIFICAT			PI					
LITERATU	RE REFERE	NCES	<b>5:</b>					

No. 26 NAME 3-Methyl-1, trans-3-pentadiene STRUCTURAL FORMULA CH2=CHC=CHCH3 Molecular C6H10 ĊH, Ref. Mole Molecular % Pur. Weight 82,140 Formula Ref Ref. Ref F.P. °C F.P. 100% dt/dP to °C/mm °K g 25°C 0.2086 5 B. P. °C h 4 BP 0.04145 77. 760 mm 2 t<sub>e</sub> 0.03499 5 24.27 ſ١ to 5 100 g' °K 30 0.88 5 30 mm 0.5868 5 10 -16.93 h' ∆Hm cal/g -46.42 5 1 to m AHv cal/g Pressure °K n 25°C 98.83 mm 25°C 103.43 5 0 30 mm 103.24 5 te 942.37 5 ВP 89.50 5 Density g/ml 20°C m to 88.26 5 te (d, e) ۰ĸ 0.735 2 88, 25 5 ۰, 0.730 2  $\mathbf{d_{4}^{t}}$ ΔHv/T 20.30 5 30 4 0.725 Surface tension 103, 39 ı to 0.7550 4 23.51 dynes/cm. 20°C \*C 0.1804 -0.0398 84 ь 4 5 30 22.23 40 20.98 5 Ref. Index °C  $\mathbf{n}_{\mathbf{D}}$ 20°C 1.452 [P] d vc tc Parachor g/ml 25 1.449 2 ml/g 30 1.446 4 30 °C 40 "C" 0.8142 4 5 Sugd. 246. 2 mm MR (Obs.) 30.148 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30, 244 25°C 0.9917 5 2 1.085 (nD-d/2) 1.0000 30 mm 5 2 Dispersion 225. 2.11 5 Dielectric 0.9545 BP 5 Flash Point °C 0.9484 1 to 7.03502 4 Fire Point t<sub>c</sub> <u> 126 °C</u> 1252.08 С 224.4 5 M Spec. AHc kcal/m Ultra V ΔHf 1.37541 A\* l to 5 X-Ray Dif. ΔFf B\* <u>94</u> ℃ 1175.17 Infrared ĸ Viscosity Viscos.., centistokes °C Solubility in Acetone to t<sub>x</sub> Carbon tet. •c Benzene A' to Ether B١ •c n-Heptane Ċ١ В to Ethanol Á °C A'\* Water to Water in B'\* ٠c (BV) to Acl to (AV) °C Bc °C cp liq. °K Cc Cryos. A° consts. B° c<sub>p</sub> vap. °K t<sub>e</sub> °C c, vap. 83.93 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

No. 27 4-Methyl-1, 3-pentadiene NAME STRUCTURAL FORMULA CH2=CHCH=CCH3 Molecular C6H10 Ċнз Molecular Weight 82.140 Mole Ref. % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g <u>°K</u> 0.2034 5 B. P. °C h ВP 0.04139 760 mm 76.3 2 5 t<sub>e</sub> 0.03499 5 f 100 23.66 to 5 g' <u>°К</u> 30 0.31 0.5857 5 30 mm 10 -17.47 5 h' ∆Hm cal/g 1 -46.91 5 m to ΔHv cal/g 25°C Pressure n ۰ĸ 98.47 mm 25°C 106.42 o 30 mm 102.99 940.37 5 t<sub>e</sub> ВP 5 89, 26 m' to Density te te (d, e) 5 88.06 'n °K g/ml 20°C 0.719 2 88.02 5 ٥' 25 0.714 2  $\mathbf{d_{4}^{t}}$ ΔHv/Te 20.30 5 30 0.709 4 Surface tension 0 to 103.04 5 0.7390 a dynes/cm. 20°C 21,52 ٠Ç <u>83</u> 0.1806 -0.0398 ь 4 20.32 5 30 ď٠ to 40 5 Ref. Index 19.16 e ¹ 20°C 1.451 2 [P]  $^{n}D$ d g/ml vc ml/g tc °C Parachor 25 1.448 2 Z0°C 30 1.445 4 30 t<sub>c</sub> 40 "C" 0.8306 4 P<sub>c</sub> mm Sugd 246.2 5 MR (Obs.) 30.760 4 PV/RT Exp. L. l. %/wt. MR (Calc.) 30, 244 25°C 0.9914 (nD-d/2)1.092 2 30 mm 1.0000 Dispersion 2 225 Dielectric 5 2.10 BP 0.9545 Flash Point C 0.9485 0 to te t 7.03324 4 Fire Point 1249.05 B | 123 °C 4 M. Spec. С 224.5 5 AHc kcal/m Ultra V ΔHf A\*I 0 to 1.37449 5 X-Ray Dif. ΔFf B\* 9<u>3 °C</u> 1172.30 Infrared ĸ Viscosity Solubility in centistokes  $\mathbf{t_k}$ Acetone to °C Carbon tet. °C Benzene A to Ether B ٠c n-Heptane B<sup>V</sup> | C' to Ethanol °C Water to B'\* Water in ۰c (BV) to Acl to (A<sup>V</sup>) °C Bc ۰c c<sub>p</sub> liq. ۰ĸ Cc Cryos. A° ۰ĸ c<sub>p</sub> vap. consts. B° c vap. t<sub>e</sub> °C 5 83.15 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

							No. 28	
NAME	2-Meth	nyl-1,4-1	pentadiene			STRUCTURAL	FORMUL	A
						CH <sub>2</sub> =C CH <sub>2</sub> C	н=Сн	
Mole	Ref.	N - 11		Molecular		с́н <sub>3</sub>	2	
Mole % Pur.	Kei.	Formula		Meight 82.14	10	3		
		Ref			Ref			Ref
F, P. °C			dt/dP			f   to		Т
F.P. 100%			°C/mm		_	gK		
B. P. °C			25°C BP	0.1000 0.03940	5 4	ъ ;	l	
760 mm 100	56. 5.95	5	t	0.03496	5	f' to		$\Gamma$
30	-16.22	5	30 mm	0.5557	5	g'	-	
10 1	-33.07 -60.96	5	ΔHm cal/g			h'	ļ	_
Pressure	-00.70	-+-	ΔHv cal/g			m to		
mm 25°C	237.83	5	25°C 30 mm	88.68	5 5	n   ' °K	1	
t <sub>e</sub>	884.46	5	BP BP	95.84 83.34	5	<u> </u>	<b></b>	+-
Density	0.40	. ] , ]	te (d. a)	82.56	5	m'   to	l	
g/ml 20°C t 25	0.694 0.689		e (a, e,	82.54	5	0' =-	1	
dt 25 4 30	0.684		ΔHv/T <sub>e</sub>	20, 31	5	Surface tension	† ·	T
<b>a</b>	0.714		d   -16 to _e   _61 °C	93. <b>04</b> 0.1731	5	dynes/cm, 20°C	18.63	5
b	-0.039	96 4	_d'to			30 40	17.53 16.45	5
Ref. Index	1.405	5 2	e' i •c		ļ	Parachor [P]	10.15	Ť
D 25	1.402	2   2	d g/ml vc ml/g			20°C	ļ	
30	1.399		tc C			30 40		
"C"	0.777	-	P <sub>c</sub> mm	1			246.2	5
MR (Obs.) MR (Calc.)	29.009		PV/RT	<del> </del>	Н	Exp. L.1, %/wt.	<del> </del>	+
(nD-d/2)	28.974 1.058		25°C	0.9816	5	u.		1
Dielectric	1.97	5	30 mm BP	1.0000 0.9565	5	Dispersion	ļ	∔_
A   -16 to	7, 005	07 4	t <sub>e</sub>	0.9523	5	Flash Point °C Fire Point	1	1
B	1172.94	4	tc	<u> </u>		M Spec.	<b></b>	+
C 1/1	228.4	5	ΔHc kcal/m ΔHf	1		Ultra V.		
A*   -16 to B*71 °C	1.367   1099.20	730   5	ΔFf			X-Ray Dif. Infrared		
к ———			Viscosity			Solubility in +	<del> </del>	+
t <sub>k</sub>			centistokes 7°C			Acetone		1
tx C			'			Carbon tet. Benzene		
A' to				]		Ether		l
B' ∟ _ °C			B <sup>V</sup>   to			n-Heptane Ethanol		
A¹* to			AV   °C			Water	1	
B'∗ °C			(BV) to	1		Water in		╄
Ac to			(AV) °C	1				
Bc t <sub>c</sub> C			c <sub>p</sub> liq. °K					
Cryos. A*			3	1				
consts. B°			Р -	[				
t <sub>e</sub> °C	60.61	5	c <sub>w</sub> vap.	L		L	l	
REFERENC	ES: 1-Do	w 2-AF	PI 3-Lit. 4-0	Calc. from de	· de	grams/100 gra ta 5-Calc. by for		nt
SOURCE:		API		de	ua	- J-Carc. by for		
PURIFICAT	ON:	API						
LITERATUR								

No. 29 3-Methyl-1, 4-pentadiene NAME STRUCTURAL FORMULA CH2=CHCHCH=CH2 Molecular C6H10 сн3 Mole Ref. Molecular Weight 82,140 % Pur. Formula Ref. Ref. Ref. F.P. °C F.P. 100% dt/dP f to °C/mm 25°C g °K 0.0966 5 B. P. ℃ h ВP 0.03930 4 760 mm 55. 2 0.03498 5 ſ١ te to 100 5.08 5 g' <u>°К</u> 30 -17.02 5 30 mm 0.5541 5 10 -33,83 5 h' ∆Hm cal/g 5 -61.64 to m ΔHv cal/g 25°C Pressure °K n 88.17 5 5 mm 25°C 247.21 o 30 mm 95.51 881.08 t<sub>e</sub> BP 83.00 5 m' to Density te (d, e) 82.24 5 n' °K g/ml 20°C 0.695 2 82.23 ٥'  $d_4^t$ 25 0.690 2 AHv/Te 20.30 5 30 0.685 4 Surface tension d -17 92.55 5 to 0.7151 4 dynes/cm. 20°C 18.73 å-\_ 59 0.1736 -0.0<sub>3</sub>96 ь 4 30 17.63 to 16.54 5 40 Ref. Index e' <sup>n</sup>D 20°C 1.405 2 [P] Parachor d<sub>c</sub> g/ml 25 1.402 2 20°C vc ml/g t °C 30 1.399 4 30 t<sub>c</sub> 40 "C" 0.7764 4 P<sub>c</sub> mm Sugd 246.2 5 MR (Obs.) 28.967 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 28.974 5 0.9806 25°C 5 (nD-d/2)1.058 2 30 mm 1.0000 Dispersion Dielectric 1.97 5 ВP 0.9560 Flash Point C 0.9519 A -17 to 7.00219 4 Fire Point B \_ 97 °C 1168.41 4 M. Spec. C 228.5 5 AHc kcal/m Ultra V ΔHf A\* -17 to 1.36668 5 X-Ray Dif. ΔFf B\* 69 °C 1095.11 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. °C Benzene A' to Ether В' °C n-Heptane B<sub>v</sub> | C١ to Ethanol A'\* °C Water to Water in B'\* °C (B<sup>V</sup>) to (A<sup>V</sup>)| Acl to °C Βc ۰c c<sub>p</sub> liq. ۰ĸ Cc Cryos. Aº •ĸ cp vap. consts. B° c, vap. te °C 59.49 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula API SOURCE: PURIFICATION: API LITERATURE REFERENCES:

												No. 30	)
NAME	2-M	ethyl-	-2,3-	pentad	iene				ST	RUCTURA	L I	FORMU L	A
										CH <sub>3</sub> C = C	=CH	ICH.	
Mole	Ref	Mo	lecul	9.7		Molecula				ċн,	-	3	
% Pur.	Ker.		rmul		5 <sup>H</sup> 10	Weight		0					
			Ref.					Ref					Ref
F.P. °C F.P. 1007				dt/d					f	1	to		
B. P. *C	-		-	25*			744	5	g	'	<u>•K</u>		
760 mm	72.	_	2	BP			4097 3498	4 5	f'	<del>                                     </del>	to		+
100 30	19.9		5	t <sub>e</sub>	mm	0.5		5	g'	l '	_K_		1
10 1	-20.7 -49.8	7	5		cal/g				h'	<u> </u>			
Pressure	-47.8	<u>'</u>	3		cal/g				m	1	to °K		İ
mm 25°C	126.5		5	25°	C mm	96.3		5 5	n o	! '	<u>_</u> _		
t <sub>e</sub>	928.6	3	5	BP		88.0	1	5	m'	<del></del>	to		+-
Density g/ml 20°C	0.7	11	2	te (	(d, e)	86.9		5	n'	i '	<u>"K</u>		
dt 25 4 30	0.7 0.7	06	2 4	ΔH	v/T <sub>e</sub>	20.3		5	0'	<u>i</u>			ļ
a 30	0.7		4	d !	-3 to	100.9	1	5		ace tensies/cm. 20		20.57	5
ь	-0.0		4	-읍,	_78_ <b>°</b> C		792	5	3,11	30	)	19.41	5
Ref. Index		25	2	6'	•(				-	40 achor [P]		18,28	5
D 25	1.4		2	d <sub>c</sub> g	g/ml				Par		)°C		1
30	1.4		4	t <sub>c</sub>	C C	1				30 40			
"C"	0.7		4	P <sub>c</sub> r	nm						igd.	246.2	5
MR (Obs.) MR (Calc.			5	PV/1		0.9	904	5	Exp	. L.1.%/v	vt.		
(nD-d/2)	1.0		2	30 1	mm	1.0	000	5	Dis	u. persion			
Dielectric		3 2740	5 4	BP t <sub>e</sub>		0.9		5		h Point °	С		$\top$
A   -3 to	1232.7	8	4	tc					<del></del>	Point			<del>-</del> }
C	225.3		5	ΔHc ΔHf	kcal/m				M S Ultr	pec. a V.			
A* -3 to B* 88 °C		7287 4	5	ΔFf						ay Dif. ared			Î.
к ——-	-				osity					bility in	+		╁
k				η	istokes °C	:			Ac	etoné rbon tet.			1
x '				'					Be	nzene			
A'   to									Etl n-l	ner Heptane			
C'			1	B <sup>V</sup>	to °C			1 1	Etl	anol			1
A'* to				(B <sup>V</sup> )		-				ter ter in			1
Ac  to	+			(A <sup>V</sup> )		į.							Γ
Bc Ltc_°	<u> </u>			c <sub>p</sub> li		<del></del>		1					
Cryos. A	<del>-  </del>		$\vdash$	c <sub>p</sub> v		1							
consts. B	<b>_</b>		_	c <sub>v</sub> v									
t <sub>e</sub> °C	78.3	1	5		·	<u> </u>			+	/: 00		<u> </u>	<u> </u>
REFEREN	CES: 1-I	Dow	2-AI	PI 3-1	Lit. 4-	Calc. fro	m dei	t. dat		Calc. by			ıt
SOURCE:				PI									
PURIFICA	TION:		A1	PI									
LITERATU	RE REF	ERE	NCES	i:									

No. 31 2-Ethyl-1, 3-butadiene NAME STRUCTURAL FORMULA CH,=C CH=CH,  $\dot{C}_2H_5$ Mole Ref. Molecular Molecular  $C_{6}H_{10}$ % Pur. Weight 82.140 Formula Ref. F.P. °C F.P. 100% dt/dP to °C/mm ٩Ķ g 25°C 0.1941 5 B. P. ℃ h BP 0.04126 4 760 mm 75. 2 <sup>t</sup>e 0.03499 5 ſ١ 100 22.53 5 to 5 g' <u>°к</u> 30 -0.74 30 mm 5 0.5837 10 -18.46 5 h' AHm cal/g 1 -47.79 5 m to ∆Hv cal/g Pressure n °K 25°C 97.84 5 mm 25°C 112.14 5 o 30 mm 102.55 5 te 936.65 5 ВP 88.88 5 m to Density 87.70 5 te te (d, e) n' °K g/ml 20°C 5 0.717 2 87.67 ٥' 25 0.712  $\mathbf{d_{4}^{t}}$ AHv/T 20.30 5 0.707 30 4 Surface tension d 0 to 102,41 5 0.7370 dynes/cm. 20°C 21.28 5 <u>82</u> <u>•c</u> 0.1805 ь -0.0398 4 30 20.09 5 d to 5 e¹ 40 18.93 Ref. Index 20°C <sup>n</sup>D 1.445 1.442 2 [P] Parachor d<sub>c</sub> g/ml 25 20°C ^c .C ml/g 30 1.439 4 30  $t_c$ "C" 40 0.8225 4 P<sub>c</sub> mm Sugd 246.2 5 MR (Obs.) 30.490 4 PV/RT Exp. L.1.%/wt. MR (Calc.) 30, 244 5 25°C 0.9908 5 (nD-d/2)1.086 2 30 mm 1.0000 5 Dispersion 225. 2 Dielectric 2.09 5 BP 0.9545 5 Flash Point C A 0 to 7.03112 te 0.9486 4 Fire Point B (121 °C 1243.85 M. Spec. C 224.7 5 AHc kcal/m Ultra V ΔHf A\*| 0 to 1.37396 5 X-Ray Dif. ΔFf B\*I 92 °C 1167.37 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. t<sub>x</sub> °C Benzene to Ether B' °C n-Heptane B<sup>V</sup> | C to Ethanol °C A'\* Water to Water in B'\* °C (B<sup>V</sup>) to Acl (AV) to °C Bc °С cp liq. °K Cc c<sub>p</sub> vap. Cryos. A° °K consts. B° te °C c, vap. 81,70 5 grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 5-Calc. by formula 4-Calc, from det. data API SOURCE: **PURIFICATION:** API LITERATURE REFERENCES:

							No. 32
NAME	2, 3-Dim	thyl-	1,3-butadiene			STRUCTURAL FO	
						CH <sub>3</sub>	_
			T			CH <sub>2</sub> =C C=CI	·12
Mole % Pur.	Ref. Me	olecul ormul		Molecular Weight 82.14	10	CH <sub>3</sub>	
		Ref.			Ref		Ref.
F.P. *C	-76,005	2	dt/dP			f to	
F.P. 100%	<u> </u>		*C/mm 25*C	0,1557	5	g <u>*K</u>	
B. P. *C 760 mm	68.78	2	BP	0.04064	4	h	
100 30	17,11	5	t <sub>e</sub>	0.03494	5	f' to g'K	
10	-5.79 -23.23	5	30 mm	0.5745	5	h'	
1	-52.08	5	ΔHm cal/g ΔHv cal/g	<del> </del>	$\vdash$	m   to	
Pressure mm 25°C	143.84	5	25°C	94.89	5	n   •K_	
t <sub>e</sub>	920.51	5	30 mm BP	100.38 87.18	5 <b>5</b>	°	
Density	0.73/-	1.	t,	86.14	5	m' to to	1
g/ml 20°C	0.7267	2 2	t (d, e)	86.11	5		
dt 25 4 30	0.7177	4	ΔHv/T <sub>e</sub>	20,33	5	Surface tension	-+
a b	0.7447	4	d   -6 to		5	dynes/cm. 20°C	22.44 5
Ref. Index	-0.0 <sub>3</sub> 87	+-	d' i to	1			21.32   5 20.21   5
n <sub>D</sub> 20°C	1.4394	2				Parachor [P]	
25 30	1.4362	2 4	d g/ml v ml/g			20°C	
"C"	0.8019	4	*c ~			40	
MR (Obs.)	<del></del>	4	P <sub>c</sub> mm			Sugd. 2	46.2 5
MR (Calc. (nD-d/2)	30.244	5	PV/RT 25°C	0.9885	5	Exp. L.1.%/wt. u.	Ì
Dielectric	2.07	5	30 mm BP	1.0000 0.9560	5	Dispersion 2	25. 2
A -6 to	+	-	t <sub>e</sub>	0.9507	5	Flash Point °C Fire Point	
В ∟,16°С	1220.88	4	t <sub>c</sub>			M Spec.	
C	225.9	5	ΔHc kcal/m ΔHf			Ultra V.	
A*  -6to B* 85°C		5	ΔFf			X-Ray Dif. Infrared	ļ
к — — —	-		Viscosity			Solubility in +	<del></del>
t <sub>k</sub>		1	centistokes 7 °C			Acetone	
x '	1		'		1	Carbon tet. Benzene	
A' to						Ether n-Heptane	
c,	1		B <sup>V</sup> to			Ethanol	}
A'* to			A <sup>V</sup> C			Water Water in	
B'* °C	<del></del>	+	(B <sup>V</sup> ) to				
Bc t *C			(A <sup>V</sup> )  °C	<del>                                     </del>	$\vdash$		
Cc	<del></del>	1	c <sub>p</sub> liq. °K				
Cryos, A° consts, B°			c <sub>p</sub> vap. °K				
t <sub>e</sub> °C	74.82	5	c <sub>v</sub> vap.				
					_	grams/100 grams	solvent
REFEREN	CES: 1-Dow			Calc, from det	t. da	ta 5-Calc, by formu	
SOURCE:		Al	PI				
PURIFICAT		Al					
LITERATU	RE REFERE	NCES	<b>5</b> :				

No. 1 Ethyne NAME STRUCTURAL FORMULA Acetylene CH=CH Molecular 26.036 Molecular C2H2 Ref. Mole % Pur Formula Ref. Ref. Ref. -80.8<sup>₹</sup> <u>F.P.</u> ℃ 2 dt/dP f to F.P. 100% °C/mm °K g 25°C B. P. °C h ВP 0.018 2 -84.0<sup>#</sup> 760 mm 2 0.02582 5 f t<sub>e</sub> 100 -108.3 2 to g' ۰ĸ 0.3036 5 -120.1 2 30 mm 30 10 -129.5 2 h! ∆Hm cal/g 1 -153.35 0, 1355 300 to m AHv cal/g(a) 0,0012 Pressure ņ 600 25°C mm 25°C ò -0.0<sub>6</sub>86 4 30 mm 196.37 5 494.3 5 BP 193.46 5 5 700 0. 3444 m' 4 to Density te te (d, e) 195.04 0. 0<sub>3</sub>37 -0. 0<sub>7</sub>98 n' 1000 4 g/ml 20°C 4 dt4 25 AHv/T 30 Surface tension d to a dynes/cm. 20°C <u>с</u> Ъ 30 ai to 40 Ref. Index °C <sup>n</sup>D 20°C Parachor [P] d<sub>c</sub> g/ml 0.231 25 20°C 4.329 vc ml/g t °C 2 30 30 t<sub>c</sub> 2 36.3 40 "C" 46816. 2 5 P<sub>c</sub> mm Sugd. 90.4 MR (Obs.) PV/RT Exp. L.1. %/wt. MR (Calc.) 9.434 25°C u. (nD-d/2) 1.0000 30 mm 5 Dispersion Dielectric 0.9700 5 BP Flash Point °C 0.9778 5 A -81 to 7,0949 2 Fire Point 5 0.273-60 °C 709.1 M. Spec. 253.2 C 2 AHc kcal/m 2 300,10 Ultra V. ΔHf 54.194 2 A\* -81 to 1.60062 5 X-Ray Dif. ΔFf 50,000 2 -60 °C B\* 676.6 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet. t<sub>x</sub> •c Benzene ۸ to Ether В' °C n-Heptane C'  $\mathbf{B}^{\mathbf{v}}$ Ethanol B' I °C Water A1\* to Water in B'\* (B<sup>V</sup>) °C A<sup>S</sup>
B<sup>S</sup>
C<sup>S</sup>
A\*S 9.1363 2 Acl (A<sup>V</sup>)| to 1230,3 Βc °C cp liq. 300°K 280.7 10.532 2 11.973 3, 1388 5 400 2 A\*s cp vap 300°K 1160.71 Cryos. Aº 0.40452 2 consts. B° 400 0.45986 2 c<sub>w</sub> vap. te °C -89.6 Saturation press (triple pt.) #Sublimation point grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: PURIFICATION: LITERATURE REFERENCES: (a) Heat of sublimation

							No. 2	
NAME	Propyne				Ī	STRUCTURAL	FORMULA	<u> </u>
	Methylace	tylene				СН₃С≡СН		
Mole	Ref. M	olecul	27	Molecular		030 - 0		
% Pur.		ormul		Weight 40.	062			,
	т.	Ref.		T	Ref			Ref
F.P. °C F.P. 100%	-102.7	2	dt/dP *C/mm			f to	]	ĺ
B. P. °C	<del></del>	+	25°C	0.0201	١, ١	g   °K		
760 mm 100	-23.22 -61.1	2 2	BP t <sub>e</sub>	0.0301 0.0334	5	$\left  \frac{1}{\mathbf{f}'} + \frac{1}{\mathbf{to}} \right $		
30	-78.2	2	30 mm	0.413	5	g' 'K		İ
10 1	-90.2 -110.6	2	ΔHm cal/g			h'		_
Pressure	<u> </u>	+	ΔHv cal/g			m   300 to n 600 °K	0.0996	
mm 25°C	666.5	5	25°C 30 mm	152.2	5	0	-0.0646	
t <sub>e</sub> Density		+-	BP	131.0 131.9	5 5	m'   700 to	0.1830	4
g/ml 20°C	;		t <sub>e</sub> (d, e)	132.15	5	n' 1000 °K	0.0 <sub>3</sub> 74 -0.0 <sub>6</sub> 24	4
d <sup>t</sup> 25 4 30			ΔHv/T <sub>e</sub>	21.39	5		-0.0624	-
a	†	†	d   -78 to		5	Surface tension dynes/cm. 20°C		
b		+				30 40		
Ref. Index			e' ; °C	0,229	5	Parachor [P]		
25 30			d g/ml vc ml/g tc °C	4.367	5	20°C 30		
"C"	<del> </del>	+	tc °C	121.6	2	40		_
MR (Obs.)			P <sub>c</sub> mm PV/RT	38000.	5		129.4	5
MR (Calc. (nD-d/2)	14.052	5	25°C			Exp. L.1.%/wt. u.		İ
Dielectric	<del></del>	+-	30 mm BP	1.0000 0.9700	5	Dispersion		
A   -78 to		2	te	0.9699 0.270	5	Flash Point °C Fire Point		
B  _24 °C	801.91 228.93	2 2	t <sub>c</sub> ΔHc kcal/m	442.07	2	M Spec.		
A*  -78 to	0,89250	5	ΔHf	44.319	2	Ultra V. X-Ray Dif.	1	
B* L-16 °C	738.2	5	ΔFf Viscosity	46.313	2	Infrared		
С	_		centistokes			Solubility in + Acetone		
t <sub>k</sub> to			7 ℃			Carbon tet.		
A' to				1		Benzene Ether		1
B' ∟ _ º	2		B <sup>V</sup>   to		$\vdash$	n-Heptane Ethanol		
A'* to			AV I _ °C			Water		
B'* °C	+	+	(B <sup>V</sup> )			Water in	<del>                                     </del>	<del>                                     </del>
Ac to			(A <sup>V</sup> )	<u> </u>				
	<b>-</b>	+	c <sub>p</sub> liq. °K				}	
Cryos, A° consts, B°			c <sub>p</sub> vap.300°K 400	0.36319 0.43258				1
t <sub>e</sub> °C	-26.2	5	c <sub>v</sub> vap.					
						f grams/100 gran		t
	CES: 1-Dow			Calc, from det	t. da	ta 5-Calc. by for	mula	
SOURCE:	rion.	AP	<del></del>					
	RE REFERE							
			•					

No. 3 NAME 1-Butyne STRUCTURAL FORMULA Ethylacetylene C2H5C≣CH Mole Molecular Molecular C4H6 % Pur Formula Weight 54,088 Ref. Ref. Ref. F.P. °C F.P. 100% -125.720 2 dt/dP f to °C/mm °K g 25°C 0.0208 5 B. P. °C h BP 0.0336 5 5 760 mm 8.09 2 0.0337 f† t<sub>e</sub> to -34.56 100 2 °K g' -53, 42 30 30 mm 0.4724 5 10 -67.75 2 h' ∆Hm cal/g -91.42 5 1 m AHv cal/g Pressure 25°C 105.61 5 mm 25°C 0 30 mm 5 125.13 te 758.4 5 BP 109.82 m Density te (d, e) 109,82 5 0.65 0.65 n' °K g/ml 20°C 2 ٥' 25 2  $\mathbf{d_{4}^{t}}$ AHv/Te 21.12 5 30 0.65 4 -60 Surface tension 0.6511 5 111.83 0.2489 8 dynes/cm. 20°C 16.40 30 °C Ъ 0.03129 5 30 16.25 5 5 to 40 16.03 Ref. Index °C [P] n<sub>D</sub> 20°C Parachor d<sub>c</sub> g/ml 25 20°C vc ml/g tc °C 30 30 40 "C"  $P_c$  mm Sugd MR (Obs.) PV/RT Exp. L.1.%/wt. MR (Calc.) 25°C (nD-d/2)30 mm Dispersion Dielectric BP Flash Point °C A -67 to 6.97497 986.46 2 Fire Point В <u>63 °C</u> M. Spec. С 232.85 2 ∆Hc kcal/m 589.08 Ultra V. ΔHf 39.48 2 A≠l -60 to 1.20441 5 X-Ray Dif. ΔFf 48.30 B\* 18 °C 920.64 5 Infrared ĸ Viscosity Solubility in c centistokes Acetone to Carbon tet. **t**x\_\_ °C Benzene A۱ to Ether B' °C n-Heptane B<sup>V</sup> | C' to Ethanol °C Water A'\* to Water in B!# (B<sup>V</sup>) °С Acl (AV) Bc ۰c c<sub>p</sub> liq. °K Cc Cryos. Aº ۰ĸ cp vap. consts. B° c, vap. te °C  $T_R = 0.75 T_C$ # at saturation pressure grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: **PURIFICATION:** LITERATURE REFERENCES:

					Т	CTRUCTURAL	No. 4	
NAME	2-Butyne			<del></del>		STRUCTURAL	r ORMULA	•
	Dimethyla	cetyl	ene		_	CH3CEC CH3		
Mole % Pur.		lecul rmul		Molecular Weight 54.0	88	, , ,		
		Ref.	[		Ref			Ref
F.P. °C	-32.260	2	dt/dP			f to		
F.P. 100%			°C/mm			g °C		
B. P. °C	-/		25°C BP	0.0379 0.0358	5 5	h ,		
760 mm 100	26.97 -18.7	2	te	0.0341	5	f' to		ŀ
30	-38.98	2	30 mm	0.5097	5	g' °C	1	
10 1	-54.46 -80.0	2	∆Hm cal/g			h'		
Pressure	00.0	۲	ΔHv cal/g			m   300 to n   600 °K	0.1082	
mm 25°C	706.6	5	25°C 30 mm	117.43	5	<b>1</b>	0.0 <sub>3</sub> 84 -0.0 <sub>6</sub> 17	4
t <sub>e</sub>	808.6	5	BP	117.03	5 5	0		├-
Density	0 (0)0	,	t <sub>e</sub> ,	116.5	5	m'   700 to n'   1000 °K	0.1020	4
g/ml 20°C	0.6910 0.6856	2	[ [a,e)	116.65	5	0'	0.0 <sub>3</sub> 93 -0.0 <sub>6</sub> 30	4
d <sub>4</sub> 25 30	0.6801	4	ΔHv/T <sub>e</sub>	20, 87	5	Surface tension		$\vdash$
a	0.7130	4	d   -40 to e   35 °C		5	dynes/cm. 20°C	21.19	5
b	-0.00101	4	d' - to			30 40	19.80	5
Ref. Index n <sub>D</sub> 20°C	1.3921	,	e'		$\sqcup$	Parachor [P]	18.42	5
<sup>n</sup> D 20°C	1.3921	2	d g/ml v ml/g			20°C		
30	1.3865	4	vc ml/g tc °C	212.	5	30		
"C"	0.7574	4	P <sub>c</sub> mm	38478.	5	40 Sugd	168.4	5
MR (Obs.)	18.64	2	PV/RT	-	-	Exp. L.1.%/wt.	100.1	Ť
MR (Calc.) (nD-d/2)	18.670 1.0466	5	25°C	0.9653	5	u.		
Dielectric	1.0100	ļ <u> </u>	30 mm BP	1.0000 0.9650	5	Dispersion		L
A -32 to	7, 07871	2	t <sub>e</sub>	0.9628	5	Flash Point °C Fire Point		
B91°C	1104.72	2	t <sub>c</sub>	L		M Spec.		$\vdash$
<u>c</u>	236.19	2	ΔHc kcal/m ΔHf	584.57 34.97	2 2	Ultra V.		
A*  -32to B*  39°C	1.28120 1032.5	5	ΔFf	44. 32	2	X-Ray Dif. Infrared		1
K - = =	1032.3		Viscosity			<del>-</del>		$\vdash$
t			centistokes 7°C			Solubility in + Acetone		1
tk C			<b>7</b> °℃		1 1	Carbon tet.		
A' to		╁──			1	Benzene Ether		
B' ∟_ °C			B <sup>V</sup> to	<del> </del>	<del> </del>	n-Heptane		
		<del> </del>	B' to			Ethanol Water		
A'* to B'* °C			(BV)	-		Water in		L
Ac  91 to	7, 34794	5	(A <sup>V</sup> )					
Bc t °C	1354.8	5	c <sub>p</sub> liq. °C	<del> </del>	$\vdash$			
Cc — —	277.8	5	11 -					
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.34573 0.41821	2			
t <sub>e</sub> °C	28.65	5	c <sub>v</sub> vap.			L,	<u> </u>	L
$T_{\mathbf{R}} = 0.75$		olid				grams/100 gra		<u>t</u>
	ES: 1-Dow	2-AI		Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		API						
PURIFICAT		API						
LITERATUE	RE REFERE	NCES	<b>5:</b>					

								<b>No.</b> 5	
NAME	l-Pent	yne				ST	RUCTURAL	FORMUL	A
Mole % Pur.	Ref.	Molecula Formula		Molecular Weight 68.	114		CH≡C(CH <sub>2</sub> ) <sub>2</sub>	сн <sub>3</sub>	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100°	-105.7	2	dt/dP			f	to		
B. P. °C 760 mm 100 30	40.18 -7. -28.	2 2 2	°C/mm 25°C BP t <sub>e</sub> 30 mm	0.0586 0.0374 0.0342 0.5251	5 5 5	g h f' g'	*K to *K		
10 1	-44. -70.	2 5	ΔHm cal/g			h'			
Pressure mm 25°C		4 5	ΔHv cal/g 25°C 30 mm BP	100.63 111.13 97.28	5 5 5	m n o	300 to	0.0650 0.0012 -0.0 <sub>6</sub> 48	4
Density g/ml 20°0 dt 25 d4 30	0.690 0.684		t <sub>e</sub> t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	96. 62 96. 63 20. 78	5 5	m' n' o'	700 to 1000 °K	0.1410 0.0 <sub>3</sub> 94 -0.0 <sub>6</sub> 32	4 4
a b Ref. Inde:	K		d -50 to e   50 °C d'   to	1	5		face tension es/cm. 20°C 30 40	19.34 18.13 16.94	5 5 5
<sup>n</sup> D 20°0 25 30	1. 385 1. 382		d g/ml vc ml/g tc °C			Par	achor [P] 20°C 30 40		
MR (Obs.	) 23.14	2	P <sub>c</sub> mm		$\sqcup$			207.4	5
MR (Calc (nD-d/2) Dielectric	1.040		PV/RT 25°C 30 mm BP	0.9810 1.0000 0.9693	5 5 5	Dis	u. L.1.%/wt. u. persion sh Point °C	119.	2
A _50 to		63 2 2 2 2	te t <sub>c</sub> ΔHc kcal/m	0.9666	5	Fir M.	Spec.		
A*  -50 to B*  50 °C	1,256		ΔHf ΔFf Viscosity	34.50 50.16	2 2	X-I	ra V. Ray Dif. Pared		
c t <sub>k</sub> — to t <sub>x</sub>   °C			centistokes 7 °C			A c Ca Be	ability in + etone rbon tet.		
B'  °(	5		B <sup>V</sup> to A <sup>V</sup> C	_		n- Et Wa	her Heptane hanol ater ater in		
Ac to	,		(B <sup>V</sup> )  (A <sup>V</sup> )  c <sub>p</sub> liq. °K			-			
Cryos. A consts. B	•		c <sub>p</sub> vap. 300K	0.37114 0.45409	<b>2</b> 2				
t <sub>e</sub> °C	43.45	5	c <sub>v</sub> vap.			L <u>,</u> _			
<b>D.D.D.</b>							ams/100 gra		ıt
	CES: 1-Do	w 2-A	Pi 3-Lit. 4-	Calc. from de	t. da	ta 5	-Calc. by for	mula	
SOURCE: PURIFICA	TION								
	JRE REFE	RENCES	S:						
		·····							

NAME											
ľ		Z-F	enty	ne				STR	UCTURAL	FORMULA	
					<del></del>				C2H5C≣CCI	H <sub>3</sub>	
Mole % Pur.		Ref.	Mo Fo	lecul: rmul:	ar C <sub>5</sub> H <sub>8</sub>	Molecular Weight 68.	114				
				Ref			Ref				Ref
F.P. °C	-	109.3		2	dt/dP	T	$\Box$	f	to		Г
F.P. 1009					°C/mm		1 .	g	*K		1
B.P. °C					25°C BP	0, 1003 0, 0392	5	h			
760 mm 100	ı	56.07 6.20		2 2	te	0.0343	5				
30		-16.0		2	30 mm	0.556	5	g'	*K		l
10 1		-32.8 -61.		2 5	ΔHm cal/g			h'	<u> </u>		L
Pressure	╁			$\vdash$	ΔHv cal/g			m	300 to	0.0691 0.0010	
mm 25°C		235.8		5	25°C 30 mm	108.5 115.58	5	n o	000 -K	-0.0 <sub>6</sub> 30	
t <sub>e</sub>	18	898.4		5	BP BP	102.6	5	<del>  </del>	700		-
Density		0.71			te (d.s)	101.66	5	m' n'	700 to	0.1628	4
g/ml 20°C	1	0.710 0.705		2 2	e (a, e)	101.7	5	٠,		0. 0 <sub>3</sub> 81 -0. 0 <sub>6</sub> 22	4
d <sub>4</sub> 25 30		0.700		4	AHv/T <sub>e</sub>	20.71	5	Sund	ace tension	ļ <del>-</del>	├
a .	1	0.731		4	d   -19 to		5 5		s/cm. 20°C	21.84	5
ь	_	-0.00	101	4				γ.	30	20.54	5
Ref. Index		1.403	20	2	e' j •(	; <u> </u>	$oxed{oxed}$		40	19.27	5
<sup>n</sup> D 20°C	1	1.400		2	d g/ml vc ml/g			Para	chor [P] 20°C		1
30	$\perp$	1.397	79	4	t <sub>c</sub> °C	241.	5		30		1
"C"		0.757	73	4	P <sub>c</sub> mm	28675.	5	j	40 Suad	207.4	5
MR (Obs.)		23.43		2	PV/RT	20075.	╁┷┤	Evn	L.1.%/wt.		<del>  -</del>
MR (Calc. (nD-d/2)	1	23, 288 1, 048		5 2	25°C	0.9901	5		u.		1
Dielectric	+	1.97	_	5	30 mm BP	1.0000	5		ersion		L
A   -19 to		7. 051	89	2	te	0.9663	5		h Point °C Point		1
B 1113°C	2 11	193.05	,	2	t <sub>c</sub>		$\perp$				┼
С	+	229. 96		2	ΔHc kcal/m ΔHf	732.25	2	M S		}	1
A*   -19to		1.30 <i>6</i>	669	5	ΔFf	30.80 46.41	2 2		ay Dif.	1	1
K	-				Viscosity		╅	Infra			₩
t.	_				centistokes	1			bility in + etone	l	1
t <sub>k</sub>   to					γ ∘⊂			Ca	rbon tet.		
A' to	,							Ber Eth	nzene	1	
B' °	2				B <sup>v</sup>   to	<del> </del>	$\vdash$		leptane		1
A'* to	+				B to			Eth Wa	anol	1	
A'* to					(B <sup>V</sup> )	-			ter in		
Ac   113 to	,	7, 229	935	5	(A <sup>V</sup> ) <sub>1</sub>	1					T
Bc t °C	C   14	24.6		5	c <sub>p</sub> liq. •K		+-+			1	
Cc — —	_	272.6		5	_	1				į	1
Cryos. A° consts. B°					c <sub>p</sub> vap.300°K 400	0.34780 0.42869	2 2				
t <sub>e</sub> °C	$\perp$	61.14		5	c <sub>v</sub> vap.					ſ	
$T_{\mathbf{R}} = 0.75$								+ gra	ms/100 grai	ns solven	t
	CES	: 1-Do	w	2-AF	I 3-Lit. 4-	Calc, from de	t. dat	a 5-	Calc. by for	mula	
SOURCE:				AP							
PURIFICA'			· D	AP							
LITERATU	KE	REFE	CREN	ICES	:						

Density									No. 7	
Mole	AME _	3-Methy	l-1-b	utyne			ST	RUCTURAL	FORMUL.	A
F. P. °C   -89.7   2   F. P. 100%						14		СН <sub>3</sub> (СН <sub>2</sub> ) <sub>2</sub> С	СН	
F. P. 100%  B. P. °C 760 mm 26. 35 100			Ref.			Ref.				Ref.
The content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the content of the	.P. 100%	-89.7	2	°C/mm	0.0376	5				
Pressure   mm 25°C   723, 4   4   799, 8   5   723, 4   4   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 8   5   799, 63   799, 26   5   799, 26   5   799, 26   5   799, 26   5   799, 26   5   799, 26   5   799, 26   5   799, 29   5   799, 29   5   799, 29   5   799, 29   5   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29   799, 29	760 mm 100 30 10	-19.4 -39.5	4	BP t <sub>e</sub> 30 mm	0.0362 0.0348	5 5	f' g'	°K		
Density g/ml 20°C	ressure nm 25°C			25°C 30 mm BP	106.8	5	n o	600 °K	0. 0293 0. 0013 -0. 0 <sub>6</sub> 64	4
Ref. Index   1, 3723   2   1, 3695   2   2, 5   30   1, 3667   4   28 c	/ml 20°C	0.660	2	le (a, e)	90.26 90.29	5	n' o'	1000 •K	0.1145 0.0010 -0.0 <sub>6</sub> 37	4 4
No.   20°C   1.3723   2   1.3695   2   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.3667   4   1.		0,691		_e 28 °C to				es/cm. 20°C 30	16. 67 15. <b>41</b> 14. 17	5 5 5
MR (Obs.)   23.261   2	D 20°C 25 30	1.3695 1.3667	2 4	d g/ml v ml/g	188.	5	Par	20°C 30		
Dielectric 1.88 5  A   -40 to   6.88971 5 te   0.9549 5   5	R (Obs.) R (Calc.)	23.261 23.288	2 5	P <sub>c</sub> mm PV/RT			Exp	Sugd.	207.4	5
B   -73 °C   1017.50   5   5	ielectric	1.88	5	30 mm BP t	1.0000 0.9566	5 5	Fla	persion sh Point °C		
Solubility in + Actone   Carbon tet.   Benzene   Ether   N   C   Carbon tet.   Benzene   Ether   N   C   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Benzene   Ether   N   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.   Carbon tet.	+ -40 to	1017.50 227.46 1.21594	5 5	ΔHc kcal/m ΔHf	32.60	2	M. Ult:	Spec.		
C!	to °C	952.8	5	Viscosity centistokes	17,12	-	Solu Ac Ca Be Et	ubility in + retone rbon tet.		
Bc to °C       1329.4       5       cp liq. °K         Ccvos. A° consts. B°       cp vap.300°K 400       0.36894 2 0.45659       2         te °C       27.76       5       cv vap.       5       cr vap.         TR = 0.75 Tc       *grams/100 grams         REFERENCES: 1-Dow       2-API       3-Lit.       4-Calc. from det. data       5-Calc. by forms         SOURCE:       API	'* to						Et Wa	hanol iter		
consts. B°         p         400 cv vap.         0.45659 2         2           te °C         27.76 5         5         cv vap.         2         grams/100 grams           TR = 0.75 Tc         *grams/100 grams         *grams/100 grams         *grams/100 grams         *grams/100 grams           SOURCE:         API         *grams/100 grams         *grams/100 grams         *grams/100 grams	c_tc_°C	1329.4	5	c <sub>p</sub> liq. °K						
T <sub>R</sub> = 0.75 T <sub>c</sub>	onsts. B°	27.54		1 -100						
REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by forms SOURCE: API		1	5	, Ab.			L+	ame/100 ~	m a a a l	<u> </u>
SOURCE: API			2-A	PI 3-Lit 4-0	Calc. from de	t. da				Ľ
					ue	J. ua		Jul		
		ION:						***************************************		
LITERATURE REFERENCES:	TERATUI	RE REFERE	NCES	5:						

NAME	1-H	exyne	:				STRUCTURAL	FORMUL	A
							C H CECH		
Mole % Pur.	Ref.	Mol	ecul		Molecular Weight 82.1	40	C <sub>4</sub> H <sub>9</sub> C≣CH		
<u> </u>			Ref.			Ref	<del></del>		Re
F.P. °C	-131.9		2	dt/dP	1		f   to	1	+
F.P. 100%				*C/mm			f to g °K	1	
B.P. °C				25°C BP	0.1670 0.042	5 2	h	ł	1
760 mm 100	71.33	'	2 4	t <sub>e</sub>	0.0360	5	f' + to		
30	-5.2		4	30 mm	0.5854	5	g' °K		
10 1	-22.9 -52.2		5	ΔHm cal/g			h'		
Pressure	-32.2	$\dashv$		ΔHv cal/g			m   300 to	0.0576	
mm 25°C	136.05	.	5	25°C	93.6	5	n 600 °K	0.0012 -0.0 <sub>6</sub> 48	
t <sub>e</sub>	925.6		5	30 mm BP	98.94 85.40	5	ļ. <u></u>	<del> </del>	+-
Density g/ml 20°C	0.71		,	te , , ,	84.3	5	m'   700 to n'   1000 °K	0.1451	4
	0.71		2 2	te (d, e)	84. 27		0'	0.0 <sub>3</sub> 96 -0.0 <sub>6</sub> 32	4
dt 25 4 30	0.70	57	4	ΔHv/T <sub>e</sub>	19.73	5	Surface tension	<del>                                     </del>	t
a L	0.73		4	d   -8 to e   78 °C		5	dynes/cm. 20°C	21.16	5
b Pot Indon	-0.03	96	4	d' - to	1		30 40	19.99 18.85	5
Ref. Index		89	2	e' i *C		$\vdash$	Parachor [P]	10.03	۲
- 25	1.39	60	2	d g/ml			20°C	1	ł
30	1.39		4	v <sub>c</sub> ml/g t <sub>c</sub> °C	248.	5	30 40		
	0.74	-	4	P <sub>c</sub> mm	22734.	5		246.4	5
MR (Obs.) MR (Calc.			2 5	PV/RT		_	Exp. L.1.%/wt.		Т
(nD-d/2)	1.04		2	25°C 30 mm	0.9913	5 5	u. Dispersion	115.	2
Dielectric	1.96		5	BP	0.9550	5	Flash Point °C		屵
A   -8 to	6.91	212	5	t <sub>e</sub> t <sub>c</sub>	0.9479	5	Fire Point		
B   118 °C	225.		5	ΔHc kcal/m	882.85	2	M Spec.		T
A*  -8 to	1.26	142	5	∆Hf	29.55	2	Ultra V. X-Ray Dif.		1
B* ∟ 88 °C	1119.4	1	5	ΔFf	52.17	2	Infrared		
c				Viscosity centistokes		1 1	Solubility in +		
tk   to		Ì		η •c			Acetone Carbon tet.		
A' to							Benzene	1	
B'		İ				$\sqcup$	Ether n-Heptane	}	
C'	<b></b>			B <sup>V</sup>			Ethanol		ŀ
A'* to B'* °C		Į		$\frac{ \mathbf{A}_{\mathbf{v}} }{ \mathbf{B}_{\mathbf{v}} } - \frac{\mathbf{c}}{ \mathbf{c} }$	-		Water Water in		
Ac   118 to		827	5	(A <sup>V</sup> )					
Bc t °C	1453.8	۱ ،	5		<del> </del>	$\vdash\vdash$		1	
Cc	268.5		5	c <sub>p</sub> liq. °K				1	
Cryos. A° consts. B°				c <sub>p</sub> vap.300°K 400	0.37460 0.46104				
t <sub>e</sub> °C	77.76		5	c <sub>v</sub> vap.	0.40104			1	
$T_{R} = 0.75$	1	1	ات	<u> </u>	I	1	+ grams/100 gra	me eclas-	<u> </u>
REFERENC		ow i	2 - AF	PI 3-Lit. 4-0	Calc from det	dat	ta 5-Calc. by for		ıt
SOURCE:			AP				5 CLIC. Dy 101		
PURIFICAT	TION:	-	AP	•					
LITERATU		EREN							
				•					

								No. 9	
NAME	l -Heptyne					ST	RUCTURAL	FORMUL	A.
							a a-		
Mole % Pur.	Ref. Mo	ecul	ar C <sub>7</sub> H <sub>12</sub>	Molecular Weight 96.16	6		C <sub>5</sub> H <sub>11</sub> C≡	СН	
		Ref.			Ref.				Ref
F.P. °C F.P. 100%	-80.9	2	dt/dP °C/mm 25°C	0.4081	5	f g	to *K		
B. P. °C 760 mm 100 30	99.74 39.63 13.56	2 4 4	BP t <sub>e</sub> 30 mm	0.0450 0.03836	5 5	f' g'	to •K		
10	-6.0	5	ΔHm cal/g	0.6491	- "	h'	<del>-</del>		l
Pressure mm 25°C	-38. 0 52. 51	5	ΔHv cal/g 25°C 30 mm	85.69 87.26	5 5	m n o	1 600 °K	0.0526 0.0012 -0.0 <sub>6</sub> 48	4 4 4
Density g/ml 20°C		2	BP t <sub>e</sub> (d, e)	75.40 73.85 73.88	5 5 5	m' n'	700 to	0.1346 0.0010 -0.0 <sub>6</sub> 34	4 4
d <sub>4</sub> 25	0.7283 0.7238	2 4	AHv/T <sub>e</sub>	18.50	5			1, 16 1	<u> </u>
a b	0.7508 -0.0 <sub>3</sub> 89	4	d 14 to e 111 °C d' to	89.13 0.1377	5 5		face tension es/cm. 20°C 30 40	22.34 21.25 20.19	5 5 5
Ref. Index n <sub>D</sub> 20°C 25 30		2 2 4	e'   °C  d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	280.	5	Par	20°C		
"C"	0.7427	4	t <sub>c</sub> °C P <sub>c</sub> mm	19163.	5		40 Sugd	285.4	5
MR (Obs.) MR (Calc. (nD-d/2)		2 5 2	PV/RT 25°C 30 mm	0.9996 1.0000	5 5	_	L.l.%/wt. u. persion	113.	2
Dielectric  A 14 to B 142 °C	1.98 6.68593 1216.6	5 5 5	BP t t c	0.9622 0.9529	5	Fla	sh Point C e Point		
A* 14 to B* 121 °C	1.05646	5 5 5	ΔHc kcal/m ΔHf ΔFf	1029.77 24.62 54.18	2 2 2	Ult: X-F	Spec. ra V. Ray Dif.		
c t <sub>k</sub> to			Viscosity centistokes 7°C			Solu Ac Ca Be	ared  ubility in + etone rbon tet.		
A'   to B'   _ °C C' _ to	-		B <sup>V</sup> to A <sup>V</sup> °C			n- Et	her Heptane hanol ster		
B'* °C	7,10459	5	(B <sup>V</sup> )  to (A <sup>V</sup> )  °C			W	iter in		
Bc tc °C	265.0	5	c <sub>p</sub> liq. °K c <sub>p</sub> vap.300°K	0.37716	2				
consts. B°		5	p 400 c vap.	0.46597	2				
$T_R = 0.7$		<b>-</b>	<del>  </del>	<u> </u>		+ g1	ams/100 gra	ms solven	t
	CES: 1-Dow	2-A	PI 3-Lit. 4-	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		A	\PI						
PURIFICA			\PI						
LITERATU	RE REFERE	NCE	S:						

							No. 10	
NAME	1-Octyne					STRUCTURAL	FORMULA	¥.
						C.H. CECH		
Mole % Pur.	Ref. Mo	lecul	ar C <sub>8</sub> H <sub>14</sub>	Molecular Weight 110.1	92	C <sub>6</sub> H <sub>13</sub> C≡CH		
		Ref.			Ref.			Ref
F.P. °C	-79.3	2	dt/dP			f to		
F.P. 1009	•		°C/mm 25°C	1.3009	5	g °K		
B. P. °C 760 mm	126.2	2	BP	0.0471	2	<u> </u>		
100	66.39	5	te	0.0358	5	f' to		
30 10	39.86 19.68	5	30 mm	0.6651	5	h'		
1	-13.73	5	ΔHm cal/g	<u> </u>	-	m   300 to	0.0487	4
Pressure	12 (0	_	ΔHv cal/g 25°C	90.74	5	n 600 °K	0.0012	4
mm 25°C	13.60 1077.7	5	30 mm	88.54	5	<u> </u>	-0.0 <sub>6</sub> 48	4
Density	<del> </del>	<del>                                     </del>	BP te	75.99 74.18	5	m'   700 to	0.1326	4
g/ml 20°0		2	e (a, e)	74, 11	5	n' 1000 °K	0.0010	4
dt 25	0.7419 0.7377	2	ΔHv/T <sub>e</sub>	19.82	5		-0.0 <sub>6</sub> 34	
a	0.7629	4	d   25 to		5	Surface tension dynes/cm. 20°C	23, 27	5
ь	0.000836	4	_a,_  150_ °C		5	<b>3</b> 0	22.23	5
Ref. Index		2	e' ' °(			40	21.23	5
<sup>n</sup> D 20°C	1.4134	2	d <sub>c</sub> g/ml			Parachor [P] 20°C		
30	1.4109	4	vc ml/g tc °C			30		
"C"	0.7146	5	P <sub>c</sub> mm			40 Sugd.	324.4	5
MR (Obs.) MR (Calc.		<b>2</b> 5	PV/RT		-	Exp. L.1.%/wt.		
(nD-d/2)	37.142 1.0428	2	25°C 30 mm	1.0016 1.0000	5 <b>5</b>	u. Dispersion	111.	2
Dielectric			BP	0.9500	5	Flash Point °C	1111	<u> </u>
A 25 to		4	t <sub>e</sub>	0.9394	5	Fire Point		ł
B 1 170 °C	2 1413.8 215.0	4 5	t <sub>c</sub>	1176, 70	2	M Spec.		
A*  33 to	1.44610	5	∆Hf	19.70	2	Ultra V. X-Ray Dif.		
B* 160 °C	<u>C</u> 1329. 7	5	ΔFf	56. 19	2	Infrared		
C			Viscosity centistokes			Solubility in +		
t <sub>k</sub>   to			η •c			Acetone Carbon tet.		ł
t'x i °C						Benzene		
B' °0					<u> </u>	Ether n-Heptane		
C'	<u>_</u>		B <sup>V</sup> to A <sup>V</sup> °C			Ethanol Water		
A'* to B'* °0			$\frac{\mathbf{A}}{(\mathbf{B}^{\mathbf{V}})_1} \frac{\mathbf{C}}{\mathbf{C}}$	_		Water in		ļ
Ac to	+		(A <sup>V</sup> )					
Bc t °C			c <sub>p</sub> liq. •K	<del> </del>	+-		j	1
Cc — -			{{	1				l
Cryos. Acconsts. B			c <sub>p</sub> vap.300°K 400	0.3790 0.4696	2 2			
t <sub>e</sub> °C	139.16	5	c <sub>v</sub> vap.					
B C C C C C C C C C C C C C C C C C C C	cne i n					f grams/100 gra		<u>t</u>
	CES: 1-Dow	2 - Al	-1 3-Lit. 4-	Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	TION.							
	RE REFERE	IC EC						
: KAIU	NE REFERE	10ES	<b>,</b>					

								No.11	
NAME	l-Nonyne					ST	RUCTURAL	FORMUL	A.
							C H C=C	u	
Mole % Pur.	Ref. Mo	ecul mula		Molecular Weight 124.2	218		C <sub>7</sub> H <sub>15</sub> C≡C		
	1	Ref.			Ref.	Ĺ		T	Ref
F.P. °C F.P. 100%	-50.	2	dt/dP °C/mm 25°C	2.727	5	f g	to °K		
B.P. °C 760 mm 100 30 10	150.8 84.23 55.19 33.28	2 5 5 5	BP t <sub>e</sub> 30 mm	0.0530 0.0385 0.7247	5 5	_h _f' g' h'	to *K		
1	2.64	5	ΔHm cal/g			m	300 to	0.0459	4
Pressure mm 25°C t <sub>e</sub>	6.26 1148.0	5	ΔHv cal/g 25°C 30 mm BP	83.15 79.30 67.35	5 5 5	n o	600 •K	0.0013 -0.0 <sub>6</sub> 49	
Density g/ml 20°C dt 25 d4 30	0.7568 0.7527 0.7486	2 2 4	te te (d, e) AHv/Te	65.28 64.52 18.40	5 5	n' o'	700 to 1000 °K	0.1312 0.0010 -0.0 <sub>6</sub> 35	4 4
a b	0.7732 -0.0 <sub>3</sub> 82	5	d 50 to e 160 °C d' to	86.20 0.1250	5 5		face tension es/cm. 20°C 30 40	24.02 23.00 22.00	5 5 5
Ref. Index <sup>n</sup> D 20°C 25 30	1.4217 1.4193 1.4169	2 2 4	e'   °C  d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	334.	5	Par	20°C 30 40		
MR (Obs.)	41.64	2	P <sub>c</sub> mm	14300.	5		<u>_</u>	363.4	5
MR (Calc.) (nD-d/2) Dielectric A 50 to	41.760 1.0430 6.77410	5 2 5	PV/RT 25°C 30 mm BP te	1.0000 1.0000 0.9482 0.9346	5 5 5 5	Dis Fla	b. L.1.%/wt. u. persion sh Point °C e Point	110.	2
B   223 °C C	1404.7 210.	5 5	t <sub>c</sub> ΔHc kcal/m ΔHf ΔFf			M. Ult:	Spec. ra V. Ray Dif.		
B* 180 °C K  c  t <sub>k</sub> to  t <sub>x</sub> °C  A' to	1316.75	5	Viscosity centistokes 7 °C			Solu Ac Ca Be	ared  ubility in + etone urbon tet. enzene		
A'* to B'* °C			B <sup>V</sup>			n- Et Wa	her Heptane hanol ater ater in		
Ac to Bc t <sub>c</sub> °C			(A <sup>V</sup> )  c <sub>p</sub> liq. °K						
Cryos, A° consts, B°			c <sub>p</sub> vap.300°K 400 c <sub>v</sub> vap.	0.38046 0.47248	2 2				
t <sub>e</sub> °C T <sub>R</sub> = 0.82 7	168.2	5	vb.	L	لــــا	L	/1.00	L	<u> </u>
REFERENC		2-A	PI 3-Lit. 4-0	Calc, from de	+ 4-		rams/100 gra -Calc. by for		.t
SOURCE:	EG. 1-DOW	API		Carc. Irom de	t. ua	3	-Caic. by 10:		
PURIFICAT	ION:	API							
	RE REFERE								-

							No. 12	
NAME	1-Decyne					STRUCTURAL	FORMULA	1
Mole	Ref. Mo	lecul	27	Molecular		C <sub>8</sub> H <sub>17</sub> C≡CH		
% Pur.	Kei. Mo	rmul	ar C <sub>10</sub> H <sub>18</sub>	Weight 138.	244			
		Ref.			Ref.			Ref.
F.P. °C	-44.	2	dt/dP			f to		
F.P. 100%			°C/mm	10.1246	5	g °K		
B. P. °C 760 mm	174.	2	25°C BP	0.0514	5	<u> </u>		
100	108.48	5	t <sub>e</sub>	0.0358	5	f' to		ŀ
30	79.28 57.00	5	30 mm	0.7333	5	g' K		
10 1	20.00	5	∆Hm cal/g			h'		
Pressure	<del> </del>	1-	ΔHv cal/g			m   300 to n   600 °K	0.0424	
mm 25°C	1.42	5	25°C 30 mm	88.69 81.14	5	0 1 000 K	0.0013 -0.0 <sub>6</sub> 49	4
t <sub>e</sub>	1198.8	5	BP	68.78	5	}	<del></del>	
Density g/ml 20°C	0 7/55	,	t <sub>e</sub> ,	66.49	5	m'   700 to n'   1000 °K	0.1345 0.0010	
	0.7655	2 2	t <sub>e</sub> (d, e)	66.38	5	0'	-0.0635	
d <sub>4</sub> 25 30	0.7577	4	ΔHv/T <sub>e</sub>	19.63	5	Surface tension		-
a	0.7811	4	d   78 to		5	dynes/cm. 20°C	24.65	5
ь	-0.0378	4				30 40	23.66	5
Ref. Index		,	e'				22.70	-
<sup>n</sup> D 20°C	1.4242	2 2	d g/ml vc ml/g		1	Parachor [P] 20°C		
30	1.4219	4	t <sub>c</sub> *C	354.	5	30		1
"C"	0.7402	4	P <sub>c</sub> mm	17454.	5	40 Sugd	402.4	5
MR (Obs.)		2	PV/RT	1	-	Exp. L. 1, %/wt.	102.1	<del>-</del>
MR (Calc. (nD-d/2)	1.0437	5 2	25°C	1.0000	5	u.		ŀ
Dielectric	1	+-	30 mm BP	1.0000 0.9400	5	Dispersion	108.	2
A   78 to	7,10870	5	t <sub>e</sub>	0.9250	5	Flash Point °C Fire Point		
B   246 °C	160 6	5	t <sub>c</sub>	L		<del></del>	<b></b>	
С	206.0	5	ΔHc kcal/m ΔHf	1470.54	2	M Spec. Ultra V.		İ
A* 78 to B* 204 °C	1.60046	5	ΔFf	ļ		X-Ray Dif.		
K L	2 131 7. 03		Viscosity	<u> </u>		Infrared Solubility in +		-
·	_		centistokes	1		Solubility in + Acetone		1
t <sub>k</sub>   to t <sub>x</sub>   °C			η •c			Carbon tet.		
A' to	<del> </del>	1-				Benzene Ether		1
B' ∟ _ °	<u>:</u>	1 1	B <sup>v</sup> to	<del> </del>	-	n-Heptane		1
C'		-	B' to A' ℃	}		Ethanol Water		
A'* to B'* °C			(BV)	-		Water in		l
Acl to	<del></del>	$\vdash$	(A <sup>V</sup> )	ļ				
Bcit c°C			<del></del>					l
	<del>-</del>	$\vdash$	<b>-</b>			1		
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.38157 0.47481	2			
t <sub>e</sub> °C	192.7	5	c <sub>v</sub> vap.			1		L
$T_{\mathbf{R}} = 0.83$						fgrams/100 gra	ms solven	t
REFERENC	CES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from det	t. da	ta 5-Calc. by for	mula	
SOURCE:		AP	PI .					
PURIFICAT	rion:	AP	1					
LITERATU	RE REFERE	NCES	):					

No. 13 1-Undecyne NAME STRUCTURAL FORMULA C9H19C≡CH Molecular C11H20 Mole Ref. Molecular % Pur. Weight 157, 270 Ref. Ref. Ref. F. P. ℃ -25 2 dt/dP f to F.P. 100% °C/mm ۰ĸ g 110.84 0.1 mm 5 B.P. °C h BP 0.0534 5 760 mm 195. 2 t<sub>e</sub> 0.0361 5 ſ١ to 100 126.84 4 4 ۰ĸ g' 0.7642 5 30 mm 30 96.43 h' 73.20 10 5 ∆Hm cal/g 34.61 5 m 300 to 0.0405 0.1 5.51 5 AHv cal/g 0,0013 ۰ĸ n 600 88.55 0.1 mm Press. mm -0.0<sub>6</sub>49 o 4 30 mm 75.22 5 1248. t<sub>e</sub> 5 BP 63, 31 5 0.1329 0.0010 m' 700 to Density te (d, e) 60.92 5 1000 n' 4 g/ml 20°C 0.7728 2 60.74 5 ٥' -0.0<sub>6</sub>35 4 0.769**0**  $\mathbf{d_{4}^{t}}$ 25 2 ΔHv/T<sub>e</sub> 5 19.57 30 0.7652 4 Surface tension 7 97 86.88 5 a 0.7880 4 dynes/cm. 20°C 22.13 5 217 °C 0.1209 ь -0.0376 4 21.27 20.44 30 to 5 40 Ref. Index e¹ 1.4306 2 <sup>n</sup>D 20°C [P] Parachor dc g/ml 25 1,4284 2 Z0°C vc ml/g t °C 1.4262 4 30 30 t<sub>c</sub> 373 5 40 0.7398 4 "C" P<sub>c</sub> mm 16891. 5 5 Sugd. 441.4 MR (Obs.) 50.97 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 50.996 5 0.1 mm 1,0000 5 1.0442 2 11. (nD-d/2)1.0000 30 mm Dispersion 2 107. 0.9339 Dielectric 2.05 5 BP Flash Point °C 0.9169 A 80 to 7.13064 4 Fire Point B 1\_267 °C 1687.2 4 M. Spec. C 202. 5 AHc kcal/m 1617.48 Ultra V. ΔHf 4.92 ž 1.67084 A\* 80 to 5 X-Ray Dif. ΔFf 62.21 2 B\* 2 67°C 1599.5 5 Infrared K Viscosity Solubility in centistokes Acetone to Carbon tet. t<sub>x</sub> °C Benzene A'ı to Ether В' ۰c n-Heptane B<sup>v</sup> C' to Ethanol B' ' °C Water A1\* to B'\* (B<sup>V</sup>) Water in °C Acl 267 to 8.22204 (A<sup>V</sup>)| Bc \_t<sub>c</sub> °C 2858.81 5 c<sub>p</sub> liq. 342.89 5 c<sub>p</sub> vap.300°K Cryos. Aº 0.38248 2 consts. B° 400 0.47665 c, vap. te C 216.2 5  $T_R = 0.84 T_C$ grams/100 grams solvent 5-Calc. by formula REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No.14	
NAME	l-Do	decyne				STRUCTURAL	FORMULA	1
						C <sub>10</sub> H <sub>21</sub> C≣C	н	
Mole % Pur.	Ref.	Molecul Formul	ar C <sub>12</sub> H <sub>22</sub>	Molecular Weight 166.2	296	10 21		
		Ref.			Ref.	i		Ref
F.P. *C	-19.	2	dt/dP	T				Г
F. P. 100%			*C/mm	1		g to		ł
B. P. 'C			0.1 mm	115.26	5	h		l
760 mm	215.	2	BP t <sub>e</sub>	0. 0550 0. 0357	5	<u>f</u> ' + <del>to</del>		l
100 30	144.63		30 mm	0.7913	5	g' K		1
10	89.09		ΔHm cal/g	<del>                                     </del>	$\vdash$	h'	1	1
1	49.05 18.8	5		<del> </del>		m   300 to	0.0391	4
O. 1	<del></del>	-   -	ΔHv cal/g 0.1 mm	81.75	5	n 600 °K	0.0013	
Press. mm	1313.0	5	30 mm	75.06	5	•	-0.0650	4
e Density			BP	63.49	5	m'   700to	0.1278	
g/ml 20°C	0.77	88 2	te te (d, e)	60.89 60.74	5 5	n' 1000•K	0.0011	
dt 25 d4 30	0.77		ΔHv/T	19.76	5	o' ¦	-0.0 <sub>6</sub> 36	4
	0,77		d   100 to		5	Surface tension		
a b	0.79		e 1250 °C		5	dynes/cm. 20°C	25.62	5
Ref. Index			d'  to	1		30 40	24.66	5
n <sub>D</sub> 20°C		40 2	e' i •c		<b> </b>	Parachor [P]	-	Ť
25	1.43	18 2	d g/ml vc ml/g			20°C		1
30	1.42	96 4	tc °C	391.	5	30		l
"C"	0.73	96 4	P <sub>c</sub> mm	15719.	5	40 Sugd	480.4	5
MR (Obs.)		2	PV/RT		1	Exp. L.1.%/wt.	100.1	ř
MR (Calc. (nD-d/2)	55.61- 1.04-		0.1 mm	0.9251	5	u.		
Dielectric	2.06		30 mm	1.0000 0.9391	5	Dispersion	107.	2
A 113 to	+		BP t <sub>e</sub>	0.9391	5	Flash Point °C		
B 1288_•C		4 5	t <sub>c</sub> AHc kcal/m	1764.40	2	Fire Point M Spec.		
A* 100 to			ΔHf	-0.01	2	Ultra V.		
B* 288 °C		5	ΔFf	64. 22	2	X-Ray Dif. Infrared	ŀ	1
к — — —			Viscosity			Solubility in +		$\vdash$
t <sub>k</sub>	-		centistokes 7 °C			Acetone		
tx c	; [		l '			Carbon tet. Benzene		1
A'   to					1	Ether		
B' L • C	4		B <sup>V</sup> to	<del> </del>		n-Heptane		
A'* to	<del>                                     </del>		B to			Ethanol Water	Ì	
B'* *C			(BV)	-		Water in		
Ac  288 to	8.42	471 5	(A <sup>V</sup> )					
Bcj t C	3185.77	5	c <sub>p</sub> liq. °K		+	1		
Cc	362.54	5	it -		1			
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.38329 0.47818	2 2			
te °C	239,2	5	c <sub>v</sub> vap.	L			<u> </u>	L_
$T_{R} = 0.84$						grams/100 gra	ms solven	t
REFEREN	ES: 1-Do	w 2-AF	PI 3-Lit. 4-0	Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		API	[					
PURIFICAT		API						
LITERATU	RE REFE	RENCES	):					

								<b>No</b> . 15	
NAME	l-Tridecy	ne				ST	RUCTURAL	FORMUL	A.
							C <sub>11</sub> H <sub>23</sub> C≣CI	н	
Mole % Pur.	Ref. Mo	ecul mul		Molecular Weight 180.3	22		11230-0.	•	
		Ref.			Ref.				Ref.
F.P. °C F.P. 100%	-5.	2	dt/dP °C/mm 0.1 mm	119.74	5	f g	to °K		
B. P. °C 760 mm 100 30 10	234. 160.72 127.97 102.95	2 4 4 5	BP t <sub>e</sub> 30 mm ΔHm cal/g	0.0573 0.0364 0.8232	5 5	h f' g' h'	to °K		
0.1	61.31 29.89	5 5	ΔHv cal/g 0,1 mm	84, 53	5	m n	900 °K	0.0376 0.0013	4
Press. mm	1350.7	5	30 mm BP	71.78 60.04	5 5	o m'	700 to	-0.0 <sub>6</sub> 50	4
Density g/ml 20°C dt 25 d4 30	0.7842 0.7806 0.7770	2 2 4	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	57.33 57.10 21.5	5 5 5	n' o'	1000 •K	0.0011 -0.0 <sub>6</sub> 36	4
a b Ref. Index	0.7986 -0.0 <sub>3</sub> 72	4 4	d 120 to e 260 °C d' to e' °C	85.87 0.1104	5		face tension es/cm. 20°C 30 40	26.03 25.09 24.17	5 5 5
n <sub>D</sub> 20°C 25 30	1.4371 1.4349 1.4327 0.7394	2 2 4	d <sub>c</sub> g/ml v <sub>c</sub> ml/g t <sub>c</sub> °C	408. 14663.	5	Par	20°C 30 40	510.4	
MR (Obs.) MR (Calc.) (nD-d/2) Dielectric	60.25 60.232 1.0449 2.06	2 5 2	P <sub>c</sub> mm PV/RT 0.1 mm 30 mm BP	1.0000 1.0000 0.9299	5 5 5	_	L.1.%/wt. u.	519. 4 106.	2
A 120 to B 308 °C	7.15674 1834.4	4	te t <sub>C</sub>	0.9095	5	Fir	sh Point °C e Point Spec.		
A* 120 to B* 270 °C K	195. 1.73019 1742.5	5 5 5	ΔHc kcal/m ΔHf ΔFf Viscosity	1911.32 -4.93 66.23	2 2 2	Ult:	ra V. Ray Dif.		
c t <sub>k</sub>			centistokes 7°C			Ac Ca Be Et	ubility in † etone rbon tet. nzene her		
B'   _ °C C' A'* to B'* °C			B <sup>v</sup>   to A <sup>v</sup>   °C (B <sup>v</sup> )	-		Et Wa	Heptane hanol ater ater in		
Acl 308 to Bc t <sub>c</sub> °C Cc	8.66116 3575.90 387.79	5 5 5	(A <sup>V</sup> )  c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c vap 300°K	0.38392 0.47948	2				
t <sub>e</sub> °C T <sub>R</sub> = 0.85	260.6	5	c <sub>v</sub> vap.	<u> </u>					
REFERENC		2. 4	DI 3-1:4 4	Cala from 3-	- نہ ہ		Cala by for		t
SOURCE:	1-DOM	2-A AP		Calc. from de	ι, α <b>a</b>	. La 5	-Calc. by for	muia	
PURIFICAT	ION:	AP							
LITERATUE	RE REFERE				-		4, 10		

							No. 16	
NAME	l-Tetra	dec yn				STRUCTURAL	FORMULA	`
Mole	Ref. M	lolecul	ar C <sub>14</sub> H <sub>26</sub>	Molecular	.,,	C <sub>12</sub> H <sub>25</sub> C≡C	Н	
% Pur.	<del>_</del> <del>_</del>	Ref.		Weight 194.3	Ref			Ref.
F, P. *C		2		<del></del>	Kei,	· · · · · · · · · · · · · · · · · · ·	T	Ker.
F.P. 100%	0.		dt/dP *C/mm			f to		l
B. P. 'C	<b>†</b>	+-	0.1 mm	123.53	5	h .	1	
$760~\mathrm{mm}$	252.	2	BP t <sub>e</sub>	0.0582 0.0358	5	$\frac{1}{f'} + \frac{1}{to}$	1	]
100 30	177.42 143.95	4	30 mm	0.8425	5	g'   °K		1
10	118.31	5	ΔHm cal/g	<u> </u>		h'	1	1
1 0, 1	75.54 43.16	5	ΔHv cal/g	<del>                                     </del>	-	m   300 to		
Press, mm		+-	0,1 mm	82.84	5	n 600 °K		
t <sub>e</sub>	1401.3	5	30 mm BP	70.33 58.89	5	<u> </u>	-0.0 <sub>6</sub> 50	1
Density			l +	56.05	5	m'   700 to		
g/ml 20°C	0.7888	2 2	te (d, e)	55, 83	5	n' 1000 °K	0.0011 -0.0 <sub>6</sub> 36	
d <sup>t</sup> 25 4 30	0.7816		ΔHv/T <sub>e</sub>	19.66	5	<u></u>	11.6	<u> </u>
a .	0, 8032	4	d   145 to	85.52	5 <b>5</b>	Surface tension dynes/cm. 20°C	26.38	5
ъ	-0.0372	4	-å,-¦ <sup>281</sup> - to	0.1059	"	30	25.43	5
Ref. Index	1.4396	2	e'			40	24.51	5
n <sub>D</sub> 20°C	1.4375	2	d g/ml v ml/g			Parachor [P]	.	İ
30	1.4354	4	t <sub>c</sub> °C	422.	5	30		1
"C"	0.7391	4	P <sub>c</sub> mm	13515.	5	40 Suga	. 558. 4	5
MR (Obs.)	64.89	2	PV/RT		<u> </u>	Exp. L.1.%/wt.		<del>  _</del>
MR (Calc.) (nD-d/2)	64.850 1.0452	5 <b>2</b>	0.1 mm	1.0000	5	u.		į .
Dielectric	2,07	5	30 mm BP	1.00 <b>0</b> 0 <b>0</b> .9 <b>30</b> 6	5	Dispersion	105.	2
A 1145 to	+		te	0.9092	5	Flash Point °C Fire Point		
B (326_°C		4	t <sub>c</sub>		<u> </u>	M Spec.	+	┼─
C	191.	5	ΔHc kcal/m ΔHf	2058.24 -9.86	2 2	Ultra V.		·
A* 145 to B* 291 °C	1.8236 1833.2	7   5	ΔFf	68.24	2	X-Ray Dif.		1
к — — —	1		Viscosity		1	Infrared Solubility in +	+	<b>├</b> ──
t <sub>k</sub>	-{		centistokes 7°C		ł	Solubility in +		l
tx C	1		<b>7</b> °⊂			Carbon tet.		}
A' to	†	+				Benzene Ether		1
B', ∟ °⊆	.[		B <sup>V</sup>   to	<del> </del>	├	n-Heptane		
A'* to	<del> </del>	+	B' to		i	Ethanol Water		ł
B'* *C	ł		(B <sup>V</sup> )	1	ŀ	Water in		<u> </u>
Ac   326 to		7 5	(A <sup>V</sup> )		l			
Bc tc_C		5	c <sub>p</sub> liq. °K	<u> </u>	-	1	İ	
Cryos, A*	413.26	<del>  }</del> -	li .					1
consts. B°			c <sub>p</sub> vap.300°K	0.38452 0.48058	2			
t <sub>e</sub> °C	280.8	5	c <sub>w</sub> vap.	1	<u> </u>	L		<u> </u>
$T_R = 0.86$						grams/100 gra	ıms solven	t
REFERENC	ES: 1-Dow		PI 3-Lit. 4-(	calc, from de	t. da	ta 5-Calc. by fo	rmula	
SOURCE:		API						
PURIFICAT		API						
LITERATU	RE REFER	ENCES	<b>3:</b>					

								No. 17	
NAME	1-Pentade	cyne				ST	RUCTURAL	FORMUL	A
					_		C <sub>13</sub> H <sub>27</sub> C≡CH	-I	
Mole % Pur.		lecul:		Molecular Weight 208.3	74		13 27		
		Ref.			Ref.				Ref
F.P. °C F.P. 100%	10.	2	dt/dP °C/mm			f g	to °K		
B.P. °C 760 mm 100	268. 190.84	2	0.1 mm BP t <sub>e</sub>	127.20 0.06 <b>03</b> 0.0366	5 5	h f' g'	to *K		
30 10	156.26 129.79	4 5	30 mm	0.8700	5	h'	K		ĺ
1 0, 1	85.69 52.33	5	ΔHm cal/g ΔHv cal/g			m	300 to 600 °K	0.0357	4 4
Press. mm	1427.3	5	0,1 mm 30 mm BP	79.33 67.33	5 5	٥		0.0013 -0.0 <sub>6</sub> 50	
g/ml 20°C	0.7928 0.7893	2 2	te te (d, e) AHv/Te	55.69 52.77 52.48	5 5 5	m' n' o'	700 to 1000 °K	0.1294 0.0011 -0.0 <sub>6</sub> 36	4 4 4
a b	0.7858 0.8068 -0.0370	4 4 4	d 150 to e 305 °C	19. 22 83. 61 0. 1042	5 <b>5</b>		face tension es/cm. 20°C 30	26.69 25.76	5
Ref. Index n <sub>D</sub> 20°C 25 30	1.4419 1.4398 1.4377	2 2 4	e'   °C  d <sub>c</sub> g/ml v <sub>c</sub> ml/g			Par	40 achor [P] 20°C	24.85	5
"C"	0.7390	4	v <sub>c</sub> ml/g t <sub>c</sub> °C P <sub>c</sub> mm				30 40 Sugd	597.4	5
MR (Obs.) MR (Çalc.) (nD-d/2)	69.53 69.468 1.0455	2 5 2	PV/RT 0.1 mm 30 mm	1.0000	5	_	L.1.%/wt. u. persion	105.	2
Dielectric A 140 to	2.08 7.20553	5 4	BP t_	0.9208 0.8969	5	Fla	sh Point °C e Point	103.	-
B   349°C C A* 150 to	1972.07 188.	<b>4</b> 5	tc  AHc kcal/m  AHf	2205.17 -14.78	2 2 2	M. Ult:	Spec. ra V. Ray Dif.		
B* 310°C  K  c t <sub>k</sub> c  t <sub>x</sub> °C  A' to  B' °C	1881.34	5	ΔFf Viscosity centistokes γ °C	70.25	2	Solu Ac Ca Be Et	ared ubility in + etone rbon tet. nzene her		
A <sup>1</sup> * to B <sup>1</sup> * °C			B <sup>V</sup> to A <sup>V</sup> C (B <sup>V</sup> )			Et Wa	Heptane hanol ater ater in		
Ac to Bc t <sub>c</sub> °C Cc			(A <sup>V</sup> )  c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c vap.300°K 400	0.38498 0.48154					
t <sub>e</sub> °C	298.8	5	c <sub>v</sub> vap.			L.	ame/100	me solus-	<u></u>
REFERENC	ES: 1-Dow	2-A	PI 3-Lit. 4-0	Calc. from de	t. da		ams/100 gra -Calc. by for		•
SOURCE:		AF							
PURIFICAT	ION:	AF							
LITERATUI	RE REFERE	NCES	<b>5:</b>						

<del></del>					—	·	No. 18	
NAME	1-Hexa	lecyn	e		_	STRUCTURAL	FORMULA	
Mole	Ref. Mo	lecul	ar C <sub>16</sub> H <sub>30</sub>	Molecular	400	C <sub>14</sub> H <sub>29</sub> C≡CH	I	
% Pur.	F	_		Weight 222.		r	·*···	-
	T	Ref.		1	Ref			Ref
F.P. *C F.P. 100%	15.	2	dt/dP	1	1 1	f to		
	<u> </u>	-	*C/mm 0.1 mm	130.28	5	g °K		1
B.P. °C 760 mm	284.	2	BP	0.0583	5	h +		
100	208.55	4	t <sub>e</sub>	0.0344	5	f' to		
30 10	174.24 147.8	4 5	30 mm	0.8670	5	g' °K		ĺ
10	103.3	5	ΔHm cal/g			h'		
0.1	69.2	5	AHv cal/g			m   300 to n   600 °K	0.0349	
Press. mm			0.1 mm 30 mm	(0.72	ا ہ ا	0 000 K	-0.0650	
t <sub>e</sub>	1465.5	5	BP BP	68.73 57.04	5 <b>5</b>	ļ. <u> </u>	<del>                                     </del>	
Density	0.70/5		ll t	53.99	5	m'   700 to n'   1000 °K	0.1290	
g/ml 20°C ,t 25	0.7965	2	t <sub>e</sub> (d, e)	53.7	5	o' 1000 K	-0.0636	
d <sub>4</sub> 25 30	0.7895	4	ΔHv/T <sub>e</sub>	21.7	5	S-6	6	-
a	0.8105	4	d 172 to		5	Surface tension dynes/cm. 20°C	26.99	5
Ъ	-0.0370	4	$\begin{bmatrix} -\mathbf{e} \\ -\mathbf{d} \end{bmatrix} = \begin{bmatrix} \frac{31}{1}^7 \\ \mathbf{to} \end{bmatrix}$		5	<b>3</b> 0	26.05	5
Ref. Index			e'   •c			40	25.14	5
n <sub>D</sub> 20°C	1.4440 1.4419	2 2	d_g/ml			Parachor [P]		ł
30	1.4398	4	d g/ml vc ml/g			20°C 30		
"C"	0,7388	4	ic C	1		40	ł	l
MR (Obs.)	74.17	2	Pcmm	<u> </u>			636.4	5
MR (Calc.)	74.086	5	PV/RT	1 0000	_	Exp. L.1.%/wt.		į
(nD-d/2)	1.0457	2	0.1 mm 30 mm	1.0000	5	u. Dispersion	104.	2
Dielectric	2.08	5	BP	0.9197	5	Flash Point °C	10	<del> -</del>
A 172 to	7, 47511	4	t <sub>e</sub>	0.8955	5	Fire Point		
B 1_367°C C	2154.7 185.	4 5	t <sub>c</sub>	1252.00		M Spec.	ļ	
A* 172 to	2,12483	5	∆Hf	2352.09	2 2	Ultra V.	į	
B* _ 327 °C		5	ΔFf	72.26	2	X-Ray Dif. Infrared	]	1
к — — —			Viscosity			Solubility in +	<del> </del>	├
t <sub>k</sub>			centistokes 7°C			Acetone		1
t <sub>x</sub> °C		i .	η ∘c			Carbon tet.	İ	-
A' to	<b> </b>	+		1	1 1	Benzene Ether		
B' <u>*C</u>			B <sup>V</sup>   to	<del> </del>	$\vdash \vdash \vdash$	n-Heptane		
		┼	B to A °C			Ethanol Water		ł
A'* to B'* *C			(BV)	-		Water in		İ
Ac  to	<b></b>	+-	(A <sup>V</sup> ) <sub>1</sub>					T
Bc t C					$\vdash$			
Cc		₩	c <sub>p</sub> liq. ∘K					
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.38539 0.48237	2			
t <sub>e</sub> °C	316.2	5	c <sub>w</sub> vap.					
						fgrams/100 gra	ms solven	t
	ES: 1-Dow			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:		Al						
PURIFICAT		Al				····		
LITERATUE	RE REFERE	NCES	5 <b>:</b>					

Mole % Pur.  F.P. °C F.P. 100%  B.P. °C 760 mm 100 30 10 1 0,1	22.		C <sub>17</sub> H <sub>32</sub>	Molecular Weight 236,42		ST	RUCTURAL C <sub>15</sub> H <sub>31</sub> C≡		A
% Pur. F.P. °C F.P. 100% B.P. °C 760 mm 100 30 10	22.	Ref.					C, EH, , C≡	СН	
% Pur. F.P. °C F.P. 100% B.P. °C 760 mm 100 30 10	22.	Ref.							
B. P. °C 760 mm 100 30 10	22.	Ref.			6		15 31		
B. P. °C 760 mm 100 30 10	299.	2			Ref.				Ref.
B. P. °C 760 mm 100 30 10	299.		dt/dP			f	to		
760 mm 100 30 10		-	°C/mm 0.1 mm	133,58	5	g	°K		
100 30 10 1		2	BP	0.0597	5	_h_			
10 1	221.80 186.68	4	t <sub>e</sub> 30 mm	0.0345 0.8879	5	f' g'	to °K		
	159.6	5	ΔHm cal/g	0.007	-	h'		·	
	114.0 79.1	5	ΔHv cal/g	<del> </del>	$\vdash$	m	300 to	0.0342	
Press. mm		+-	0.1 mm			n o	600 <b>∘K</b>	0.0013 -0.0 <sub>6</sub> 50	
t <sub>e</sub>	1502.1	5	30 mm BP	66.70 55.18	5 5		700 .		ļ
Density g/ml 20°C	0.7996	, <b>#</b> 2	te te (d, e)	52.06	5	m' n'	700 to	0.1278 0.0011	
dt 25	0.7961	2	ΔHv/T <sub>e</sub>	51.5 21.5	5	٥'		-0.0 <sub>6</sub> 37	4
	0.7926		d 184 to	85.85	5		face tension		
a b	0.8136		_e <u>  334 °C</u>	0.1026	5	dyn	es/cm. 20°C 30	27.22 26.28	5
Ref. Index	1		d'  to				40	25.37	5
<sup>n</sup> D 20°C	1.4457 1.4437	*   2   2	d <sub>c</sub> g/ml	1		Par	achor [P] 20°C		
30	1.4417		V mi/g				30		
"C"	0.7386		tc°C P <sub>c</sub> mm				40 Sugd	675 <b>.4</b>	5
MR (Obs.) MR (Calc.		, 5	PV/RT		$\vdash$	Ext	L.1.%/wt.		-
(nD-d/2)	1.0459	≠   3   2	0.1 mm 30 mm	1.0000	5	-	u.	104	
Dielectric	2.09	5	RP	1.0000 0.9176	5 5		persion sh Point °C	104.	2
A 184 to B   384 °C	7,4882		te t <sub>c</sub>	0.8921	5		e Point		
c 1204 C	182.	5	ΔHc kcal/m	2499.01	2		Spec.		
A*  184 to			ΔHf ΔFf	-24.64 74.27	2 2		ra V. Ray Dif.		
B* 344 °C	- 2125.3	5	Viscosity	14.21	-		ared		<u> </u>
°	-		centistokes				ubility in Tetone		
t <sub>k</sub> to			ŋ °C			Ca	rbon tet.		
A'   to							nzene her	į	
B'°	-	1 1	B <sub>v</sub> to		1-1		Heptane hanol		
A'* to	+	_	AV I °C			Wa	ater		
B'* °C	+		(B <sup>V</sup> )			W <sub>4</sub>	ter in		₩
Ac to			(A <sup>V</sup> )		1				
Cc			c <sub>p</sub> liq. °K						
Cryos. A° consts. B°			c <sub>p</sub> vap300°K	0.38579 0.48311	2 2				
t <sub>e</sub> °C	333.7	5	c <sub>v</sub> vap.						
	rcooled liqu	id				+ g1	ams/100 gra	ms solven	ıt
REFEREN	CES: 1-Dov	w 2-A	PI 3-Lit. 4-	Calc. from de	t. da				
SOURCE:		AP	I						
PURIFICA		<b>A</b> P							
LITERATU	RE REFER	(ENCES	<b>5</b> :						

							No. 20	
NAME	1-Octade	c <b>yne</b>			_	STRUCTURAL 1	FORMULA	
					_	C <sub>16</sub> H <sub>33</sub> C≡	СН	
Mole		olecul		Molecular		16 33		
% Pur.	F	ormul		Weight 250.45	-			Ref
E D 4C	27.	Ref.			Ref		1	Kei
F.P. C F.P. 100%	<del> </del>	+-	dt/dP *C/mm		1 1	f to		1
B.P. °C	<b>†</b>	+	0.1 mm	136.86	5	h .		
760 mm	313.	2	BP t <sub>e</sub>	0.0609 0.0345	5 5	$\frac{1}{f'} + \frac{1}{to}$		
100 30	234.16 198.25	4	30 mm	0.9080	5	g'   °K		
10	170.5	5	AHm cal/g			h'		
1 0. 1	123.8 88.1	5	ΔHv cal/g	<del> </del>		m   300 to	0.0334	
Press. mm	<del> </del>	+-	0.1 mm		ا ا	n 600 °K	0.0013 -0.0 <sub>6</sub> 50	
t <sub>e</sub>	1540.6	5	30 mm BP	64.71 53.55	5	ļ.—————	-0.0630	Ľ
Density	4		t	50.39	5	m'   700 to n'   1000 °K	0.1301	
g/ml 20°C	0.8025 0.7990	2 2	te (d, e)	50.2	5	n'   1000 °K	0.0011 -0.0 <sub>6</sub> 36	
d <sub>4</sub> 25	0.7955 <sup>#</sup>	4	ΔHv/T <sub>e</sub>	21.4	5	<u> </u>		H
a .	0.8165	4	d   196 to		5 <b>5</b>	Surface tension dynes/cm, 20°C	27.46	5
<u>b</u>	-0.0370	4	<u>-a,</u> - 350 €			30	26.51	5
Ref. Index	1 4474#		e' •C			40	25.59	5
n <sub>D</sub> 20°C	1.4474 1.4453	2 2	d g/ml vc ml/g			Parachor [P] 20°C	}	
30	1.4432 <sup>#</sup>	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	1		30	ł	
"C"	0.7386	4	1 -		1	40 Sund	714 4	5
MR (Obs.)	83.451 <sup>#</sup>	2	P <sub>c</sub> mm	ļ		<u> </u>	714.4	-
MR (Calc.) (nD-d/2)	83.322 1.0461	5 2	0.1 mm	1.0000	5	Exp. L.1.%/wt.	,	
Dielectric	2.09	5	30 mm	1.0000	5	Dispersion	10 <b>4</b> . <sup>≠</sup>	2
A   196 to	<del></del>	+	BP t <sub>e</sub>	0.9178 0.8914	5 5	Flash Point °C		
B 1400 °C	7.50787 2281.1	4	tc	0.0711		Fire Point	ļ	-
c	180.	5	∆Hc kcal/m	2645.94	2	M Spec. Ultra V.		
A*   196 to	2.19183		ΔHf ΔFf	-29.56 76.28	2 2	X-Ray Dif.		
B* ∟360 °C	2187.9	5	Viscosity	10.20	Ī	Infrared		<u> </u>
·	İ		centistokes	1		Solubility in + Acetone		1
tk   to	İ		η •c	1		Carbon tet.		
tx C		+		1		Benzene Ether		
B' ∟ °C	1		<del> </del>	<b> </b>	_	n-Heptane		
C'			B <sup>V</sup>   to	1		Ethanol	1	
A'* to B'* °C			(BV)	-		Water Water in		l
Ac  to	<del> </del>	+-	(A <sup>V</sup> ))	1				Π
Bc t C		1		<del> </del>	-		1	1
Cc			c <sub>p</sub> liq. ∘K	1				ł
Cryos. A° consts. B°			c <sub>p</sub> vap.300°K 400	0.38610 0.48377				
t <sub>e</sub> °C	348.0	5	c <sub>v</sub> vap.					
for under	cooled liqui	d				grams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	Calc, from det	da	ta 5-Calc. by for	mula	
SOURCE:		AF						
PURIFICAT		AP						
LITERATU	RE REFERI	NCES	3:					

								No. 21	
NAME	l-Nonade	cyne				ST	RUCTURAL	FORMUL	A
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 264.4	78		C <sub>17</sub> H <sub>35</sub> C≡C	Н	
		Ref.		T	Ref.		<del></del>		Ref.
F, P, *C	33.	2	dt/dP			f	to		
F.P. 100%		1.	°C/mm			g	°K		
B. P. °C			0.1 mm	139.89	5	h			
760 mm	327.	2	BP t <sub>e</sub>	0.0621 0.0344	5	_f'	to		
100 30	246.54 209.87	4 4	90 mm	0.9272	5	g'	°ĸ		
10	181.6	4	ΔHm cal/g	11,72.12	+-	h'	,		
1	133.9	5	ΔHv cal/g	<b>+</b>	<del>                                     </del>	m	300 to	0.0329	4
0.1 Press.mm	97.4	5	0.1 mm		1 1	n	600 °K	0.0013	
t <sub>e</sub>	1580.0	5	30 mm	63.00	5	l °	İ	-0.0 <sub>6</sub> 50	4
Density		<b>†</b>	BP t <sub>e</sub>	52.12 48.92	5	m'	700 to	0.1267	
g/m1 20°C	0.8050‡	2	te (d, e)	48.68	5	n' o'	1000 <b>°K</b>	0.0011	
dt 25 4 30	0.8016 0.7982	2	ΔHv/T <sub>e</sub>	21.2	5		<u> </u>	-0.0 <sub>6</sub> 37	*
. 50		4	d 208 to	82,47	5		face tension		_
ь	0,8186 -0,0368	4	_e_ _366_ °C	0.0928	5	dyn	es/cm. 20°C 30	27.65 26.73	5
Ref. Index			d' to				40	25.83	5
n <sub>D</sub> 20°C	1.4488	2	d <sub>c</sub> g/ml	<del></del>		Par	achor [P]		
25 30	1.4467 1.4446 <sup>‡</sup>	2 4	v ml/g		1 1	İ	20°C		
"C"		4	tc °C				<b>30</b> <b>4</b> 0		
MR (Obs.)	0.7384 88.088 <sup>≠</sup>	2	P <sub>c</sub> mm				Sugd.	753.4	5
MR (Calc.)	87 940	5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.0463	2	0.1 mm 30 mm	1.0000	5	Die	u. persion	103.≠	١,
Dielectric	2.10		BP	0.9184	5		sh Point °C	103.	2
A 208 to	7.51711	4	t <sub>e</sub>	0.8912	5		e Point		
B  _416 °C_ C	2336.7 177.	4 5	t <sub>c</sub> ΔHc kcal/m	2792.86	2	M.	Spec.		
A*  208 to	2,21395	5	ΔHf	-34.49	2		ra V.		
B*[_376 °C	2241.3	5	ΔFf	78.29	2		Ray Dif. ared		
K ———			Viscosity			Sol	ubility in +		<del>                                     </del>
t <sub>k</sub>   to			centistokes り °C			Ac	etone		
t⊊   °C			'				rbon tet. enzen <b>e</b>		
A'  to					1		her		
B'°C			B <sub>v</sub> to	†	$\vdash$		Heptane hanol		
A¹* to		1	A I °C				ater		
B'* °C			(B <sup>V</sup> )	-		W	ater in		L
Acl to			(A <sup>V</sup> )						
Bc tc °C			c <sub>p</sub> liq. °K		T-				
				0.004:-					
Cryos. A* consts. B*			c <sub>p</sub> vap.300°K 400	0.38642 0.48435	2				
t <sub>e</sub> °C		5	c <sub>w</sub> vap.	1		L			<u>L</u>
	cooled liquid						rams/100 gra		t
	ES: 1-Dow	2-A	PI 3-Lit. 4	Calc, from de	t. da	ta 5	-Calc. by for	mula	
SOURCE:		API							
PURIFICAT		API							
LITERATUR	E REFERE	NCES	5:						

								No. 22	
NAME	l-Eicosyne						STRUCTURAL FORMULA		
Mole Ref. Molec % Pur. Form				ular C <sub>20</sub> H <sub>38</sub> Molecular ula C <sub>20</sub> H <sub>38</sub> Weight 278.504			C <sub>18</sub> H <sub>37</sub> C≡C	н	
			Ref.		***************************************	Ref			Ref.
F. P. *C	36.		2	dt/dP			f to		
F. P. 100%	T			°C/mm			g °K		1
B. P. °C				0.1 mm BP	142.65	5	h .		
760 mm	340.	.	2	t	0.0633 0.0347	5 5	$\frac{1}{\mathbf{f}} + \frac{1}{\mathbf{to}}$		
100 30	258.03		4	30 mm	0.9540	5	g' K		l
10	191.8		5	ΔHm cal/g	····		h'		Ì
1 0. 1	143.2	- 1	5	ΔHv cal/g	<del> </del>	$\vdash$	m   300 to	0.0323	4
Press. mm	100.0	-		0, 1 mm			n 600 °K	0.0013	
t <sub>e</sub> .	1601.5		5	30 mm	60.40	5	° ;	-0.0650	4
Density	<del></del>			BP t	50.36 47.09	5 5	m'   700 to	0.1264	
g/ml 20°C	0.80	73 7	2	te te (d, e)	46.82	5	n' 1000 °K	0.0011	
d <sub>4</sub> 25	0.80 0.80	397 05	2 4	ΔHv/T <sub>e</sub>	21.4	5	l	-0.0637	4
a 30				d   219 to	81.68	5	Surface tension	27.04	1_
b	0.82		4	<u>e   381 °C</u>	0.0921	5	dynes/cm. 20°C	27.84	5
Ref. Index	1	-	$\dashv$	d'   to	1		40	26.01	5
n <sub>D</sub> 20°C	1.45 1.44	01 🗯	2			$\vdash \dashv$	Parachor [P]		
25 30	1.44	81. <del>7</del>	2	d g/ml v ml/g			20°C		1
"C"	1.44		4	tc °C			30 40		1
	0.73		4	P <sub>c</sub> mm			li .	792.4	5
MR (Obs.) MR (Calc.)	92.72 92.55		2 5	PV/RT	<del> </del>	1	Exp. L.1.%/wt.		
(nD-d/2)	1.04	65#	2	0.1 mm	1.0000	5	u.	#	1_
Dielectric	2.10		5	30 mm BP	1.0000 0.9122	5 5	Dispersion	103.≠	2
A   219 to	7,52	336	4	te	0.8833	5	Flash Point °C Fire Point		İ
B   431 °C	2386.3		4	tc				<del> </del>	<del> </del>
	174.		5	ΔHc kcal/m ΔHf	2939. 78	2	M Spec. Ultra V.	1	İ
A*   219 to	2.24	673	5	ΔFf	-39.41 81.00	2 2	X-Ray Dif.	l	
B* ∟391 °C	2274.3		"	Viscosity		$\vdash$	Infrared	<b></b>	├
°			- 1	centistokes			Solubility in + Acetone		1
tk C	ľ			ነ የ • ℃			Carbon tet.	ļ	
A' to	<del> </del>						Benzene		1
B' C						$\perp$	Ether n-Heptane		1
C'				B <sup>V</sup> to			Ethanol		
A!* to			1	A <sup>V</sup> C			Water Water in	1	1
B'* °C	<del> </del>		-	(B <sup>V</sup> )	1			†	<b>†</b>
Ac to				(A <sup>V</sup> )		1		1	
Cc				c <sub>p</sub> liq. •K	1				
Cryos, A° consts, B°				c <sub>p</sub> vap.300°K	0.38677 0.48488	2			
	379.8		5						
<pre># for under</pre>	cooled li	quid					+ grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow 2	2 - AF	PI 3-Lit. 4-0	alc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICATION: API									
LITERATU	RE REF	EREN	CES	:					

No. 23 l-Heneicosyne NAME STRUCTURAL FORMULA C<sub>19</sub>H<sub>39</sub>C≡CH Molecular C21H40 Mole Ref Molecular Weight 292.530 % Pur. Ref. Ref. Ref. F.P. °C F.P. 100% 2 41. dt/dP f to °C/mm g °K 0.1 mm 145.39 B. P. °C h BP 0.0643 5 760 mm t<sub>e</sub> 352. 2 0.0354 5 f١ to 100 268.63 4 g' °K **3**0 mm 0.9700 5 30 230.60 4 10 201.2 h١ 5 ∆Hm cal/g 151.7 5 to m 0.1 AHv cal/g 113.8 ٩K n 0.1 mm Press, mm 0 30 mm 58,90 <sup>t</sup>e 1594.2 5 BP 48.06 m' Density to te (d, e) 44.72 5  $0.8094_{\pm}^{\ddagger}$   $0.8060_{\pm}^{\ddagger}$   $0.8026^{\ddagger}$ n' °K g/ml 20°C 2 44.32 5 01  $d_4^t$ 25 ΔHv/T<sub>e</sub> 20.9 5 30 4 Surface tension 229 81.85 d to 5 0.8230 4 28.00 5 dynes/cm, 20°C <u> 394</u> °C 0.0960 ь -0.0368 27.07 5 4 30 ď 40 26.17 5 Ref. Index e' °C 20°C 1.45137 2 [P] n D Parachor 1.4493 1.4473 dc g/ml 25 2 20°C vc ml/g 30 4 30 t<sub>c</sub> 40 "C" 0.7383 4 P<sub>c</sub> mm 831.4 Sugd. 5 MR (Obs.) 97.33 \* 2 Exp. L.1. %/wt. PV/RT MR (Calc.) 97.176 0.1 mm 1.0466<sup>‡</sup> 1.0000 u. (nD-d/2)2 30 mm 103.<sup>≠</sup> 1.0000 5 Dispersion 2 Dielectric 2.11 5 BP 0.8948 5 Flash Point °C A 229 to 0.8628 7.53602 4 Fire Point B 1444 °C 2439.3 M. Spec. C 172. 5 ∆Hc kcal/m Ultra V ΔHf A\* 229 to B\* 404 °C 2.30870 5 X-Ray Dif. ΔFf 2360.3 Infrared ĸ Viscosity Solubility in c centistokes Acetone t<sub>k</sub> to Carbon tet. °C Benzene A' | to Ether ٠<u>c</u> B١ n-Heptane Bv | Av | C١ Ethanol to °C A!\* Water to Water in (BV)I B'\* °C Acl (A<sup>V</sup>)|  $Bc_1$ °C c<sub>p</sub> liq. Cc Cryos, Aº cp vap. °K consts. B° c<sub>v</sub> vap. te °C 393.3 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

<del></del>							No. 24	
NAME	1-Doc	osyne	<u> </u>			STRUCTURAL :	FORMUL	A
						C <sub>20</sub> H <sub>41</sub> C≣C	н	
Mole	Ref. Mo	lecul	ar a	Molecular		20-41		
% Pur.	Fo	rmul	ar C <sub>22</sub> H <sub>42</sub>	Weight 306.5	56			
	<del></del>	Ref.			Ref			Ref
F.P. °C F.P. 1007	45.	2	dt/dP *C/mm			f to		1
B. P. °C	+	-	0.1 mm	147.86	5	g °K	ł	1
760 mm	363.	2	BP	0.0653 0.0354	5 5	$\left  \frac{\mathbf{h}}{\mathbf{f'}} + \frac{\mathbf{h}}{\mathbf{f}} \right $		1
100 30	278.35 239.72	4	t <sub>e</sub> 30 mm	0.9774	5	g' to	l	
10	209.9	5	AHm cal/g	<b>†</b>		h'		
1 0. 1	159.5 121.0	5 <b>5</b>	ΔHv cal/g	†	$\vdash$	m to to		
Press. mn			0.1 mm 30 mm	58,13	5	n ! °K	1	1
t <sub>e</sub>	1621.8	5	BP	46.74	5	m' l to	<del> </del>	+-
Density g/ml 20°0	0.8114	2	te te (d, e)	43.38 43.0	5 5	n' K		1
dt 25 4 30	0.8114 <sup>‡</sup> 0.8080 <sup>‡</sup> 0.8046 <sup>‡</sup>	2 4	ΔHv/T	20.9	5	o'		
a 30	0.8250	4	d   238 to	80.28	5	Surface tension	20.17	T.
<u> </u>	-0.0368	4	d' 1 407 to		5	dynes/cm. 20°C	28.17 27.24	5
Ref. Index	1 4524	,	e'			40	26, 33	5
n <sub>D</sub> 20°0	1.4524 <sup>‡</sup> 1.4504 <sup>‡</sup> 1.4484	2 2	d g/ml v ml/g			Parachor [P] 20°C	}	1
30		4	vc ml/g tc °C			30		
"C"	0.7381	4	P <sub>c</sub> mm	1		40 Sugd.	870.4	5
MR (Obs.) MR (Calc.	1 101 704	2 5	PV/RT	<b>†</b>	$\vdash$	Exp. L.1.%/wt.	<del>                                     </del>	+
(nD-d/2)	1.0468	2	0.1 mm 30 mm	1.0000	5	u. Dispersion	103. ≠	2
Dielectric		5	BP	0.8943	5	Flash Point °C	-	+-
A 238 to B 457 °C		4	te t <sub>c</sub>	0.8615	5	Fire Point		
c	170.	5	∆Hc kcal/m			M Spec. Ultra V.	1	
A*   238 to		5	ΔHf ΔFf			X-Ray Dif.	1	
B* ∟417 °C	2 2400.0	5	Viscosity	<u> </u>		Infrared Solubility in +		ــــ
t <sub>k</sub>	<u>-</u>		centistokes 7°C	1		Acetone		
t <sub>x</sub> (		1	7 ℃			Carbon tet. Benzene		1
A'   to				l		Ether		
c, – – -	-		B <sup>V</sup> l to	1		n-Heptane Ethanol		
A'* to			AV I °C	_[		Water		
B'* *(	<del>-</del>	1	(B <sup>V</sup> )			Water in	ļ	+
Ac  to		1 1	(A <sup>V</sup> )	ļ	1_1		1	
Cc —			c <sub>p</sub> liq. °K					
Cryos, A <sup>c</sup> consts, B <sup>c</sup>			c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	405.7	5	c <sub>v</sub> vap.					
	rcooled liquid					grams/100 grai	ms solver	ıt_
	CES: 1-Dow	2-AF	PI 3-Lit. 4-0	Calc. from de	t. da			
SOURCE:		AP						
PURIFICA		AP						
- IEKATU	RE REFERE	NCES	<b>:</b>					

No. 25 NAME 1-Tricosyne STRUCTURAL FORMULA C<sub>21</sub>H<sub>43</sub>C≡CH Molecular C23H44 Mole Ref Molecular Weight 320.582 % Pur Formula Ref. Ref. Ref. F. P. °C 49. dt/dP f to F.P. 100% °C/mm g °K 0.1 mm 150.33 5 B.P. °C h 5 BP 0.0663 374. 760 mm 2 <sup>t</sup>e 0.0355 ſ١ 288.06 to 100 4 ٩ĸ g' 30 mm 0.9928 5 248.83 4 30 218.5 5 h' 10 ∆Hm cal/g 167.3 5 1 m to ∆Hv cal/g 0.1 128.1 5 ۰ĸ n 0.1 mm Press. mm o 30 mm 56.68 5 <sup>t</sup>e 1648.9 5 ВP 45.54 5 m' | to Density te (d, e) 42.18 5 0.8131 \( \frac{\pm}{4} \)
0.8100 \( \frac{\pm}{4} \)
0.8069 \( \frac{\pm}{4} \) n' °K g/ml 20°C 2 41.82 5 ٥' 25 30 2  $\mathbf{d_{4}^{t}}$ ΔHv/T<sub>e</sub> 20.9 5 4 Surface tension 248 78,84 to 0.8255 4 a 28.30 dynes/cm. 20°C 419 0.0890 ᇷᅴ °C ъ -0.0362 4 27.45 26.62 5 30 to 5 Ref. Index 40 e' 1.4534 2 nD 20°C [P] Parachor d<sub>c</sub> g/ml 25 1.4514 2 20°C 1.4494 vc ml/g tc °C 30 4 30 40 "C" 0.7381 4 P<sub>c</sub> mm 909.4 5 Sugd 106.62 # MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 106.412 1.0469 0.1 mm 1.0000 (nD-d/2) 11. 2 102. 1.0000 30 mm 2 5 Dispersion Dielectric 2.11 5 BP 0.8936 5 Flash Point °C 0.8600 A 248 to 7.55521 4 Fire Point B \_469°C 2533.5 M. Spec. C AHc kcal/m Ultra V. ΔHf A\* 248 to 2.35591 X-Ray Dif. ΔFf B\* 429°C 2453.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to Carbon tet.  $\mathbf{t}_{\mathbf{x}}$ °C Benzene to Ether B' °C n-Heptane B<sup>v</sup> | c' Ethanol ۰c A'\* Water to B'\* °C (BV) Water in (A<sup>V</sup>)| Aci to Вс ۰c cp liq. ۰ĸ Cc Cryos. A c<sub>p</sub> vap. °K consts. B° te °C c, vap. 418.1 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det, data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

						Т			No. 26	
NAME	1-1	Tetrac	osy	ne			STRUCTU	RAL 1	FORMUL	A
							с <sub>22</sub> н <sub>4</sub>	5C≣CH	ī	
Mole % Pur.	Ref.	Mole	cula nula	C24H46	Molecular Weight 334.6	.08				
	*****		lef.			Ref				Ref
F.P. °C	52.		2	dt/dP			f	to		
F.P. 100%				°C/mm			g	°K	1	1
B. P. °C				0.1 mm BP	152.80 0.0672	5	h			
760 mm 100	385. 297.78		2 4	te	0.0355	5	<u> </u>	to	İ	
30	257.94		4	30 mm	1.0082	5	g'	°K		
10 1	227.1	ł	5	ΔHm cal/g			h'			
0. 1	135.3	- 1	5	ΔHv cal/g	† · · · · · · · · · · · · · · · · · · ·		m (	to		
Press mm	<u> </u>			0.1 mm	55.00	_	n	°K	1	
t <sub>e</sub>	1677.	i	5	30 mm BP	55.36 44.42	5				
Density		40#		t	41.01	5	m'   n' '	to °K		
g/ml 20°C	0.81 0.81	48 14	2 2	t <sub>e</sub> (d, e)	40.66	5	0, 1	K	ļ	
d <sub>4</sub> 25	0.80	80 <sup>#</sup>	4	ΔHv/T <sub>e</sub>	20.8	5	<u> </u>			┼
	0.82		4	d   257 to		5	Surface te dynes/cm.		28,45	5
Ъ	-0.03		4	-6,-  432 °C		5	3,	30	27.51	5
Ref. Index		≠		e'   *C				40	26,59	5
n <sub>D</sub> 20°C	1.45 1.45	44 / 23 <sup>‡</sup>	2 2	d g/ml v ml/g	1		Parachor	[P] 20°C		
30	1.45	02#	4	C				30		
"C"	0.73		4		Ì			40	040 4	_
MR (Obs.)	111.26	7	2	P <sub>c</sub> mm					948.4	5
MR (Calc.)	111.03	0 #	5	PV/RT 0.1 mm	1,0000	5	Exp. L.1.	%/wt.		1
(nD-d/2)	1.04	70.	2	30 mm	1.0000	5	Dispersion	a	102. #	2
Dielectric	2.11		5	BP	0.8933	5	Flash Poir	nt °C		+
A 257 to B 482 °C	7.56	438	4 4	ر و و	0.8589	5	Fire Point	:		
c — <u>-</u>	166.	- 1	5	ΔHc kcal/m	<u> </u>		M Spec.			
A*   257 to	2.37	736	5	ΔHf ΔFf			Ultra V. X-Ray Dif			ļ
B* L442 °C	2499.5	1	5		<del> </del>	-	Infrared	-		
С	İ	1		Viscosity centistokes			Solubility	in +		T
tk   to	1	- 1		η °c	1		Acetone Carbon to	••		
<u>x</u>				•			Benzene			1
A'   to B'   °C	İ						Ether			
č, – – <u>–</u>		- 1	1	B <sup>v</sup>   to			n-Heptan Ethanol	е		
A'* to			$\neg$	A <sup>v</sup> I °C	1	}	Water			1
B'* °C				(B <sup>V</sup> )	1		Water in			<del> </del>
Ac  to	ļ	- 1	1	(A <sup>V</sup> )	1		Ì			
Bc tc_C		1		c <sub>p</sub> liq. °K			i			-
Cryos, A*	<del></del>	t	$\neg$							
consts. B°			- 1	р.						
t <sub>e</sub> °C	430.4		5	c <sub>v</sub> vap.						Ì
<pre># for under</pre>	cooled li						f grams/l			nt
REFERENC	ES: 1-D	ow 2-	-AP	I 3-Lit. 4-0	Calc, from de	t. da	ta 5-Calc.	by for	mula	
SOURCE:										
PURIFICAT	ION:									
LITERATUR	E REF	ERENC	ES	:						

No. 27 NAME 1-Pentacosyne STRUCTURAL FORMULA C<sub>23</sub>H<sub>47</sub>C≡CH Molecular C<sub>25</sub>H<sub>48</sub> Mole Ref Molecular Weight 348.634 % Pur Formula Ref Ref. Ref. F.P. °C F.P. 100% 55. 2 dt/dP f to °C/mm ۰ĸ g 0.1 mm 155.00 5 B. P. °C h ВP 0.0681 5 395. 306.62 760 mm 2 t<sub>e</sub> 0.0355 5 f١ 100 4 ۰ĸ g' 1.0221 5 266.24 30 mm 30 4 10 235.0 5 h' ∆Hm cal/g 182.3 5 1 to 0.1 141.9 5 ∆Hv cal/g n ٩K 0.1 mm Press. mm 0 30 mm 54.06 1701.6 5 <sup>t</sup>e 43.32 5 BP m' Density to 39.89 5 te te (d, e) n' g/ml 20°C 0.8163 0.8129 4 °К 2 39.56 5 ۰,  $\mathbf{d_4^t}$ 25 2 AHv/Te 0.8095 5 20.8 30 4 Surface tension 265 76.26 d 5 0.8299 4 28.57 5 dynes/cm. 20°C 1 443 0.0834 5 °C ь -0.0368 4 27.63 30 to 40 26.71 5 Ref. Index e' °C 1.4552 20°C n<sub>D</sub> [P] Parachor dc g/ml 25 1.4532 2 Z0°C 1.4512 vc ml/g 30 4 30 ŧċ 40 "C" 0.7380 4  $P_c$  mm 5 987.4 Sugd. MR (Obs.) 115.89 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 115.648 5 1.0000 0.1 mm 1.0471 (nD-d/2)5 2 30 mm 102. 1.0000 5 Dispersion 2 Dielectric 2.12 5 BP 0.8927 5 Flash Point °C 265 to 0.8575 7.57091 4 Fire Point 2621.8 В 1\_493 °C 4 M. Spec. C 164. 5 AHc kcal/m Ultra V. ΔHf A\* 265 to 2.39684 5 X-Ray Dif. ΔFf B\*|\_453 °C 2540.2 Infrared ĸ Viscosity Solubility in c centistokes Acetone t<sub>k</sub> | to  $\mathbf{t_x}$ Carbon tet. °C Benzene A' | to Ether B <u>°с</u> n-Heptane B<sup>V</sup> A<sup>V</sup> C' to Ethanol °C A'\* Water to Water in B'\* (BV) °C Acl to (AV) Bc °C cp liq. °К Cryos. Aº ٩K c<sub>p</sub> vap. consts. B° t<sub>e</sub> °C c<sub>v</sub> vap. 441.7 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc, by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

										No. 28	
NAME	1-F	lexa c os	yne					STRUCTU	RAL :	FORMUL.	A.
							4	$C_{24}H_{4}$	<sub>9</sub> C≣CH		
Mole % Pur.	Ref.	Molec Form	cular nula	С <sub>26</sub> н <sub>50</sub>	Molecular Weight 362	. 660		24 4	7		
			ef.				ef				Rei
F.P. °C	57.			dt/dP			ヿ	f l	to	T	1-
F.P. 100%			$\neg$	*C/mm	1			g	°K		1
B. P. °C			7	0.1 mm BP	157.19 <b>0</b> .069		5	h			1
760 mm	405.	.	2	t <sub>e</sub>	0.035		5	<u></u>			
100 30	315.45 274.53		4	e 30 mm	1.035	، ا و	5	g'	°ĸ		1
10	242.9		5	ΔHm cal/g		_	٦	h'			}
1	189.4		" ⊩		+	+	-	m (	to		†
0.1 Press. mm	148.5		-1	∆Hv cal/g 0.1 mm	1		1	n i	°К		1
t <sub>e</sub>	1727.	l	5	30 mm	52.86		5	°			İ
e Density	<del> </del>	<del>,+</del> -	$\dashv$	BP	42.32 38.87		5	m'	to		
g/ml 20°C	0.81	77 #	2	te te (d, e)	38.55		5	n'	°K	ļ	1
dt 25 4 30	0.81	43' <sub>4</sub>	2	ΔHv/T <sub>e</sub>	20.8		5	0'			1
					0 75.04	+;	5	Surface te			_
a b	-0.03		4	e   454 °	C 0.080		5	dynes/cm		28.68 27.74	5
Ref. Index	-	<del>,</del>	—- ii ·		Ĉ			1 -	30 40	26.82	5
n <sub>D</sub> 20°C	1.45	60 <sup>‡</sup>	- 11	. ,	4	-+		Parachor	[P]		1-
- 25	1.45	407	2	d g/ml vc ml/g					20°C	ì	1
30	1.45	520		tc °C	1	- 1	ł		30 40		
"C"	0.73		!!	P <sub>c</sub> mm	}	-				1026.4	5
MR (Obs.) MR (Calc.)	120.53 120.26	. A i		PV/RT		+	$\dashv$	Exp. L.1.	%/wt.		1
(nD-d/2)	1.04	72#	2	0.1 mm	1.000		5	u.			
Dielectric	2,12		5	30 mm BP	1.000 0.892		5	Dispersion		102.≠	2
A   274 to	7,57	727	4	t <sub>e</sub>	0.856		5	Flash Poin			İ
B 1_504°C		İ	4	t <sub>c</sub>			_	M Spec.			
С	162.		—-	ΔHc kcal/m ΔHf		l		Ultra V.			
A* 274to B* 464°C	2.41	490		ΔFí		-	- 1	X-Ray Dif			l
K 1-101 0	2300.0		- II	Viscosity			$\neg$	Infrared		<b></b>	
·		- 1	1 9	centistokes	1	J		Solubility Acetone	in +		1
t <sub>x</sub> to °C		1	_ ∥ '	7 °	7	- }		Carbon to	et.		
A' to	ļ					İ	- 1	Benzene Ether			1
B' ∟ °C		- 1	-				_	n-Heptan	e		1
<u>c'</u>			1	B <sup>V</sup>   to			ı	Ethanol			
A'* to		- 1	- 1	- <u></u> '	4	ł	- 1	Water Water in			
B'* °C		-+	<del></del>	(B <sup>V</sup> )		]				1	+
Ac to			- 11	(A <sup>V</sup> )	ļ		_			1	1
Cc			_   '	c <sub>p</sub> liq. °p	۲	į	-				
Cryos. A° consts. B°			7	c <sub>p</sub> vap. °F	c						
t <sub>e</sub> °C	452.9		5	vap.	1					<u> </u>	
# for unde								f grams/1	<b>0</b> 0 gran	ms solven	t
	ES: 1-D	ow 2-			Calc, from	det.	dat	ta 5-Calc.	by for	mula	
SOURCE:			API								
PURIFICAT			API								
LITERATUI	RE REF	ERENC	ES:								

No. 29 NAME 1-Heptacosyne STRUCTURAL FORMULA C25H51CECH Molecular C27H52 Mole Molecular Weight 376.686 % Pur. Ref Ref. Ref. F.P. °C F.P. 100% 60. 2 dt/dP f to °C/mm °K g 159.39 0.1 mm 5 B. P. °C h BP 0.0699 5 760 mm 415. 2 t<sub>e</sub> 0.0356 5 ſ١ to 100 324.29 4 g¹ ۰ĸ 1.0498 5 30 mm 30 282.82 4 250.7 10 5 h! ∆Hm cal/g 196.6 5 1 m to ∆Hv cal/g 0.1 155.0 5 ٥ĸ n 0.1 mm Press. mm 0 30 mm 51.75 5 t<sub>e</sub> 1751.8 5 BP 5 41.38 m' | Density to 37.91 t<sub>e</sub> (d, e) n' g/m1 20°C ۰ĸ 0.81907 2 37.64 5 0.8156 0.8122 ٥' dt 4 25 2 AHv/Te 20.6 5 30 4 Surface tension 282 73.94 5 to a 0.8326 28.79 5 dynes/cm. 20°C <u>| 465</u> ٠C 0.0784 -0.0368 ь 27.85 5 4 ăĦ 30 to 26.93 5 40 Ref. Index e¹ ۰c 20°C 1.4568 2 [P]  $\mathbf{n}_{\mathbf{D}}$ d<sub>c</sub> g/ml v<sub>c</sub> ml/g t<sub>c</sub> °C Parachor 25 1.4548 2 20°C 1.4528 30 4 30  $\mathbf{t_c}$ 40 "C" 0.7380 4  $P_c$  mm Sugd. 1065.4 5 MR (Obs.) 125.17 ≠ 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 124.884 0.1 mm 30 mm 1,0000 5 1.0473 u. (nD-d/2) 2 1.0000 102. 5 Dispersion 2 Dielectric 2.12 5 BP 0.8919 Flash Point °C 0.8553 A 282 to 7.58347 4 Fire Point 2704.0 B |\_515°C 4 M. Spec. C 160. 5 ΔHc kcal/m Ultra V. ΔHf A\* 282 to 2.43266 5 X-Ray Dif. ΔFf B\*|\_475°C 2621.3 Infrared K Viscosity Solubility in c centistokes Acetone tk tx to Carbon tet. °C Benzene A to Ether B °C n-Heptane B<sup>V</sup> | C to Ethanol °C Water A1\* to Water in (B<sup>V</sup>) B'\* °C Acl (AV) to  $Bc_1$ °C cp liq. Cc Cryos. Aº cp vap. ۰ĸ consts, B° c vap. te °C 464.2 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

NAME	1-C	ctacosyne	•			STRUCTURAL	FORMUL.	
						C <sub>26</sub> H <sub>53</sub> C≣C	н	
Mole % Pur.	Ref.	Molecul Formul		Molecular Weight 390.	712	261153		
		Ref	<del></del>		Ref			Re
F.P. °C	62.	2	dt/dP	1	1	6 1	<u> </u>	+
F.P. 100%			*C/mm		1 1	f to g °K		
B.P. °C	<del>                                     </del>		0.1 mm	161.32	5	h		ļ
760 mm	424.	2	BP	0.0707 0.0356	5			1
100 30	332.25	4 4	t <sub>e</sub> 30 mm	1,0621	5	g' to		
10	257.8	5		1	+	h'	}	
1	203.0	5 5	ΔHm cal/g	<del> </del>	-	m to	<del>                                     </del>	十
0.1 Press. mm	161.0	-   3	ΔHv cal/g 0.1 mm			n °K		
t <sub>e</sub>	1773.5	5	30 mm	50.65	5	°		1
e Density	<del></del>		BP	40.44 36.94	5	m'   to		T
g/ml 20°C	0.82	02 2	t <sub>e</sub> (d, e)	36.64	5	n' °K	l	Ì
dt 25 4 30	0.81 0.81	687   2   34	ΔHv/T	20.7	5	0'		
			d   290 to	72.82	5	Surface tension		
a b	0.83		_e <u>  4</u> 76 °C		5	dynes/cm. 20°C	28.89 27.94	5
Ref. Index	<del></del>		d'   to		i i	40	27.02	5
n <sub>D</sub> 20°C	1 45	75 2		1	1-	Parachor [P]		$\top$
25 30	1.45	55 <sup>‡</sup> 2	d g/ml vc ml/g			20°C		
"C"	1.45	35 4	tc° °C°		1 1	30 40		
	0.73		P <sub>c</sub> mm				1104.4	5
MR (Obs.) MR (Calc.	1 120 50	2 1 5 1	PV/RT			Exp. L.1.%/wt.		$\top$
(nD-d/2)	1.04	74 <sup>‡</sup>   2	0.1 mm 30 mm	1.0000	5	u.	102.≠	2
Dielectric	2.12		BP	1.0000 0.8912	5	Dispersion Flash Point °C	102.	1-
A   290 to		724 4	t <sub>e</sub>	0.8539	5	Fire Point		İ
B 1 526 °C		4 5	t <sub>c</sub>	<del> </del>	$\vdash$	M Spec.		†
A*  290 to	158. 2.44		ΔHc kcal/m ΔHf			Ultra V.	İ	1
B* 486 °C		5	ΔFf			X-Ray Dif. Infrared		1
к	1		Viscosity			Solubility in +	<del> </del>	+
t to	-		centistokes			Acetone		İ
t <sub>x</sub> to			<b>7</b> °⊂		1	Carbon tet.		1
A' to	<del> </del>	-				Benzene Ether		
B' *	<u>:  </u>		B <sup>V</sup> to	<del> </del>	+	n-Heptane		
	ļ		B' to			Ethanol Water		
A'* to B'* *C			(BV)	-		Water in		
Acl to	+	-+-	(A <sup>V</sup> )					T
Bci t C			<del></del>	<del> </del>	+			
Cc	<b>1</b>		c <sub>p</sub> liq. °K			ş-		1
Cryos. A° consts. B°			c <sub>p</sub> vap. °K					
te °C	474.3	5	c <sub>v</sub> vap.	<u> </u>		<u> </u>		L
for unde						grams/100 gra		nt
REFEREN	JES: 1-D			Calc. from de	t. da	ta 5-Calc. by for	mula	
SOURCE:	NOV:	AP						
PURIFICAT		AP						
LITERATU	RE REFI	ERENCES	<b>:</b>					

No. 31 l-Nonacosyne NAME STRUCTURAL FORMULA C<sub>27</sub>H<sub>55</sub>CH≣CH Molecular C29H56 Mole Ref Molecular Weight 404.738 % Pur. Ref. Ref. Ref. F. P. °C 65. 2 dt/dP f to F.P. 100% °C/mm ۰ĸ g 0.1 mm 163.23 5 B.P. °C h BP 0.0714 5 760 mm 432. 2 ſ١ 0.0357 5 100 to 339, 31 4 ۰ĸ g1 30 mm 1.0737 5 30 296.90 4 10 264.1 5 h' ∆Hm cal/g 5 208.7 m to AHv cal/g 0.1 166.1 5 ۰ĸ n 0.1 mm 30 mm Press. mm o 49.51 1793.8 <sup>t</sup>e 5 BP 5 39.52 m Density te te (d, e) 36,05 5 g/ml 20°C n† ۰ĸ 0.8213 2 5 0.8180 35.78 ۰,  $d_4^t$ 25 2 ΔHv/Te 0.8147<sup>#</sup> 20.5 5 30 4 7 297 Surface tension 71.45 5 0.8345 4 dynes/cm. 20°C 28.98 5 l <u>485</u> °C 0.0739 ь -0.0366 4 30 28.06 5 to 27.16 5 40 Ref. Index e' °C 20°C 1.4581 n<sub>D</sub> Parachor [P] d<sub>c</sub> g/ml 1.4561<sup>‡</sup> 1.4541<sup>‡</sup> 25 2 20°C vc ml/g 30 4 30 "C" 40 0.7379 4  $P_c$  mm 5 1143.4 Sugd. MR (Obs.) 134.45 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 134.120 5 1.0475 1.0000 0.1 mm 30 mm (nD-d/2) 2 102. 1.0000 5 Dispersion 2 Dielectric 5 2.13 BP 0.8910 Flash Point °C <sup>t</sup>e 0.8531 A 297 to 7.59699 Fire Point B 535°C 2777.8 4 M. Spec. 157. C 5 AHc kcal/m Ultra V. ΔHf A\* 297 to 2.46916 5 X-Ray Dif. ΔFf B\*|\_495°C 2694.2 Infrared ĸ Viscosity Solubility in centistokes Acetone to °C Carbon tet. °C Benzene A۱ to Ether B١ °C n-Heptane B<sup>V</sup> | C' to Ethanol °C Water A1\* tο (B<sup>V</sup>)| Water in B'\* °C Acl (AV) to Bc ۰c c<sub>p</sub> liq. °К Cc Cryos. Aº cp vap. ۰ĸ consts. B° c<sub>v</sub> vap. te °C 483.3 for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc. from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

MAME							No. 32	
NAME	l-Tria	conty	ne			STRUCTURAL	FORMULA	A.
						С <sub>28</sub> н <sub>57</sub> С≢СН		
Mole	Ref. M	olecul		Molecular		28 57		
% Pur.	F	ormul	ar C <sub>30</sub> H <sub>58</sub>	Weight 418.	764			
		Ref.			Ref			Ref.
F.P. C	67.	2	dt/dP			f to		
F.P. 100%	<u>'</u>	-	*C/mm 0.1 mm	165.2	5	g °K		1
B.P. °C 760 mm	441.	2	BP	0.0722	5	h +		1
100 30	347.27	4	t <sub>e</sub> 30 mm	0.0357 1.086	5	f' to g' 'K		ĺ
10	304.38 271.2	4 5	ΔHm cal/g	1.000	<del>                                     </del>	h'		ĺ
1 0. 1	215.1 172.1	5	ΔHv cal/g	ļ	├	m to		
Press. mm	<del></del>	+ -	0.1 mm		1	n °K	!	
t <sub>e</sub>	1816.2	5	30 mm BP	48.55 38.71	5	<u> </u>		
Density g/ml 20°C	0.8224	,	t	35.19	5	m' to		
dt 25	0.8224, 0.8190, 0.8156,	2	le (d, e)	34.99 20.6	5	0'		
			ΔHv/T <sub>e</sub>	<del></del>	5	Surface tension	•	$\vdash$
a b	0.8360 -0.0368	4 4	<u>e   495 °C</u>	0.0721	5	dynes/cm. 20°C	29.07 28.12	5
Ref. Index		+ -	d'   to			40	27.20	5
n <sub>D</sub> 20°C	1.4587 ±	2 2	d <sub>c</sub> g/ml	1	$\vdash$	Parachor [P]		
30	1.4587; 1.4567; 1.4547	4	II V mi/o		1	20°C 30		
"C"	0.7378	4	-		1	40	1102 4	5
MR (Obs.)		2	P <sub>c</sub> mm PV/RT	ļ		Exp. L.1.%/wt.	1182.4	-
MR (Calc. (nD-d/2)	138.738 1.0475	5 2	0.1 mm	1.0000	5	u.	4	İ
Dielectric	2.13	5	30 mm BP	1.0000 0.8906	5	Dispersion	101.≠	2
A 304 to		4	t <sub>e</sub>	0.8521	5	Flash Point °C Fire Point		
3 <u>  545 °C</u>	2813.0 155.	4 5	t <sub>c</sub> AHc kcal/m	ļ	_	M Spec.		<del>                                     </del>
A*   304 to	<del></del>	5	ΔHf		1	Ultra V.		
B* 505 °C	2729.0	5	ΔFf	<u> </u>		X-Ray Dif. Infrared		
c c	İ		Viscosity centistokes	Į	Į.	Solubility in +		
t			η •c	1	1	Acetone Carbon tet.		1
t c		$\perp$			ł	Benzene		
B'				<u> </u>	<u> </u>	Ether n-Heptane		1
C'	<b>-</b>	$\perp$	B <sup>V</sup> to A <sup>V</sup> C			Ethanol Water		
A'* to B'* *C			(BV)	-		Water Water in		L
Aci to	+	$\dagger$	(A <sup>V</sup> )		İ			
Bc tc_C			c <sub>p</sub> liq. °K	<del> </del>	-	1		1
Gryos. A*	<del> </del>	+	-					
consts. B.			р .	1	l			
t <sub>e</sub> °C	493.4	5	c <sub>v</sub> vap.					
	rcooled liquid	1				+ grams/100 gran	ns sol <b>ve</b> n	t
	CES: 1-Dow	2-AI		Calc, from de	t. da			
SOURCE:		AP						
PURIFICAT		AP		·- ·-		····		
MITERATU	RE REFERE	NCES	<b>5:</b>					

No. 33 NAME l-Hentriacontyne STRUCTURAL FORMULA C29H59C≡CH Molecular Mole Ref Molecular C31H60 % Pur Weight 432,790 Formula Ref. Ref Ref. 69. F.P. C F.P. 100% 2 dt/dP f to °C/mm °K g 167. OG 0.1 mm 5 B.P. °C h BP 0.0729 5 449. 354. 32 760 mm 0.0357 f to °K t, 100 4 g' 1.0976 5 30 mm 310.98 30 4 h' 10 277.4 5 ∆Hm cal/g 220.7 5 m to AHv cal/g 0.1 177.2 5 ٩K n 0.1 mm Press. mm ٥ 30 mm 47.55 t<sub>e</sub> 1836.8 5 BP 37.90 5 m' to Density 34.39 5 n' ۰ĸ g/ml 20°C 0.8234 0.8200 0.8166 te (d, e) 2 34.20 5 ۰,  $\mathbf{d_4^t}$ 25 2 ΔHv/T<sub>e</sub> 5 20.6 30 4 Surface tension 311 ā 69.30 5 29.16 0.8370 4 dynes/cm. 20°C <u>| 504</u> °C 0.0699 ь -0.0368 4 30 28.21 5 40 27,28 5 Ref. Index e' °C 1.4593 20°C [P]  $\mathbf{n}_{\mathbf{D}}$ Parachor d<sub>c</sub> g/ml 1.4573 25 2 20°C 1.4553<sup>‡</sup> vc ml/g 30 4 30 <sup>t</sup>c 40 "C" 0.7378 4 P<sub>c</sub> mm 5 Sugd. 1221.4 143,71 \* MR (Obs.) **2** 5 PV/RT Exp. L. 1. %/wt. MR (Calc.) 1**43.3**56 1.0476 0.1 mm 1.0000 5 (nD-d/2)2 30 mm 101. <sup>‡</sup> 1.0000 5 Dispersion 2 Dielectric 5 2.13 BP 0.8905 5 Flash Point °C A 311 to 0.8515 7.60994 Fire Point 2851.7 В \_554°C 4 M. Spec. 154. C 5 ∆Hc kcal/m Ultra V. ΔHf A\* 311to 2.50239 5 X-Ray Dif. ΔFf B\*[\_514°C 2766.9 Infrared K Viscosity Solubility in centistokes c Acetone to °C ٠c Carbon tet. Benzene A' to Ether B' i °C n-Heptane B<sup>V</sup> | C' to Ethanol °C A'\* Water Water in B'\* °C (B<sup>V</sup>)I Acl (A<sup>V</sup>)|  $Bc_1$ •c c liq. ۰ĸ Cc Cryos, Aº ٠ĸ c<sub>p</sub> vap. consts. B. c vap. te °C 502.4 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

									No. 34	
NAME	l - Doi	triaco	ontyr	ie .			STRUCTU	RAL :		۸.
Mole	Ref.	Mol	ecul	ar C <sub>32</sub> H <sub>62</sub>	Molecular		с <sub>30</sub> н <sub>6</sub>	l C≣CI	H	
% Pur.		For	mul	C <sub>32</sub> H <sub>62</sub>	Weight 446.8	16				
			Ref.			Ref.				Ref.
F.P. *C	71.		2	dt/dP	1		f	to		1
F.P. 100%	<b>}</b>			*C/mm 0,1 mm	168,72	5	g	°K		
B.P. °C 760 mm	457.		2	BP	0.0736	5	<u> </u>			ĺ
100	361.40		4	t <sub>e</sub>	0.0357 1.1084	5	f' g'	to °K	ļ	l
30 10	317.64 283.7		4 5	30 mm	1.1004	1,	h'	**	1	į .
1	226.5	- 1	5	ΔHm cal/g		<b>.</b>	m	to	<b>-</b>	
0.1	182.5		- 5	ΔHv cal/g 0.1 mm			n }	°K		ì
Press.mm	1858.9		5	30 mm	46.66	5	°			ĺ
e Density	1000.7		-	BP	37.18 33.67	5	m'	to		
a/m1 20°C	0.82	43 <sup>‡</sup>	2	te te (d, e)	33.47	5	n'	•K		
dt 25	0, 82 0, 82 0, 81	107	2	AHV/T	20.6	5	0'			<u>L</u>
.4 JU	0.81	75	4	d   318 to	68.25	5	Surface ter		20.35	_
b	-0.03		4	<u> </u>	0.0680	5	dynes/cm.	30°C	29.23 28.31	5
Ref. Index		4		d'   to				40	27.40	5
n <sub>D</sub> 20°C	1.45	98‡	2 2	d <sub>c</sub> g/ml			Parachor			
30	1.45 1.45 1.45	58 <sup>#</sup>	4	V_ mi/g				20°C 30		i i
"C"	0,73		4	Te "C				40		
MR (Obs.)	148, 37	#	-2	P <sub>c</sub> mm					1260.4	5
MR (Calc.)	147 97	4 . 1	5	PV/RT 0.1 mm	1.0000	5	Exp. L.1.9 u.	/wt.		1
(nD-d/2)	1.04	77'	2	30 mm	1.0000	5	Dispersion		101.≠	2
Dielectric	2.13		_5	BP t <sub>e</sub>	0.8910	5	Flash Poin	t °C		T
A   318 to B   563 °C	7.61 2880.8	'''	4	t <sub>c</sub>	0.8515	] ]	Fire Point			<u> </u>
c ———	152.		5	ΔHc kcal/m			M Spec. Ultra V.			1
A*   318 to B*   523 °C	2.51 2794.9	198	5	ΔHf ΔFf			X-Ray Dif.			
K L = S	- , , . ,	J	١	Viscosity			Infrared	n +	<u> </u>	┼
°		1	1	centistokes			Solubility i Acetone	n T		1
t <sub>k</sub> to		l		η •c			Carbon te	t.		
A' to							Benzene Ether			
B' <u>*</u> C			İ	B <sup>V</sup>   to	1	+-	n-Heptane			
A'* to				B to C	İ		Ethanol Water			}
B'* °C			- 1	(BV)	-		Water in			<u> </u>
Ac  to		$\neg \uparrow$		(A <sup>V</sup> )					1	
Bc t <sub>c</sub> C				cp liq. °K					ļ	
Cryos. A*				c <sub>p</sub> vap. °K						
t <sub>e</sub> °C	511.4		5	c <sub>v</sub> vap.						
<pre>for under</pre>		quid					+ grams/10	0 gran	ms solven	t
REFERENC	ES: 1-D	ow 2	2-AF	PI 3-Lit. 4-0	Calc. from de	t. da				
SOURCE:			ΑP							
PURIFICAT	ION:		AP	PI						
LITERATUR	E REFI	EREN	CES	:						

								No. 35	
NAME	1-Tr	itriacon	tyne		_	ST	RUCTURAL	FORMUL	A
Mole % Pur.	Ref.	Molecula Formula	ar C 33H64	Molecular Weight 460.8	42		C <sub>31</sub> H <sub>63</sub> C≡C	н	
		Ref.			Ref.				Ref
F. P. °C	73.	2	dt/dP			f	to		
F.P. 100%			°C/mm	170.00	_	g	°K		
B. P. °C			0.1 mm BP	170.36 0.0742	5	h			
760 mm	464.	2	t	0.0358	5	fı	to	•	l
100 30	367.58 323.42		30 mm	1.1184	5	g'	°K		
10	289.2	5	ΔHm cal/g		m	h'			
1 0.1	231.4 187.1	5	ΔHv cal/g	T	lП	m	to		
Press. mm	107.1		0.1 mm			n o	°K		
t <sub>e</sub>	1873.0	5	30 mm BP	45.71 36.35	5				
Density		4	1 t.	32.82	5	m' n'	to		
g/ml 20°C	0.82 0.82	52 2 2 18 2	'e (", ",	32.66	5	, n	*K		İ
d <sup>t</sup> 25 4 30	0.81	84 4	ΔHv/T <sub>e</sub>	20.5	5		L		
	0.83		d 324 to		5		face tension es/cm, 20°C	29.31	5
Ъ	-0.03		d' 521 °C		5	8 2	30	28, 35	5
Ref. Index			e' ' 'C				40	27,42	5
<sup>n</sup> D 20°C	1.46	03 7 2	d <sub>c</sub> g/ml	1		Par	achor [P]		
30	1.45 1.45	837 2 63 4	v_ml/g				20°C 30		
"C"	0.73		1 -				40		
MR (Obs.)	153.01		P <sub>c</sub> mm				Sugd.	1299.4	5
MR (Calc.)	152 50	2   5	PV/RT 0,1 mm	1 0000	ا ۔ ا	Exp	L.1.%/wt.		
(nD-d/2)	1.04	77 <sup>‡</sup> 2	30 mm	1.000 <b>0</b> 1.0000	5	Dis	u. persion	101.≠	2
Dielectric	2,13		BP	0.8895	5	Fla	sh Point °C		
A 324 to B 571 °C	7.61 2913.4	812 4 4	t <sub>e</sub> t <sub>c</sub>	0.8494	5	Fir	e Point		
c (-1.1.5)	151.	5	ΔHc kcal/m	<del></del>			Spec.		Π
A* 324 to	2.53	156 5	ΔHf				ra V. Ray Dif.		
B*  531 °C	2828.3	5	ΔFf				ared		
K — — —			Viscosity centistokes				ubility in +		
t <sub>k</sub> Tto			η •c	:			etone rbon tet.		
*x   *C							enzene		
A'   to B'   °C	l						her		
č, '			B <sup>v</sup> to				Heptane hanol	1	
A'* to			AV   °C	:_		w.	ater		
B'* °C			(B <sup>V</sup> )			w	ater in	ļ	-
Acl to			(A <sup>V</sup> )						
Bc tc °C	ļ		c <sub>p</sub> liq. °K	:					
Cryos. A*			3	.					
consts. B	1		P	.				]	
t <sub>e</sub> °C	519.3	5	c <sub>v</sub> vap.					,	
≠ for under					•	+ g:	rams/100 gra	ms solver	ıt
REFERENC			PI 3-Lit. 4	-Calc. from de	t. da				
SOURCE:		API							
PURIFICAT	ION:	API							
LITERATUE									

							No. 36	
NAME	1-Tetrat	riacon	yne			STRUCTURAL I	FORMULA	4
						C 11 C-C11		
Mole	Ref. N	(alasul		Malagulan		C <sub>32</sub> H <sub>65</sub> C≣CH		
% Pur.	Rei. N	Formul	ar C <sub>34</sub> H <sub>66</sub>	Molecular Weight 474,	868			
		Ref.			Ref.			Ref.
F, P. °C	74.	2	dt/dP			f to		
F. P. 1007			*C/mm	172 02	5	g °K		1
B.P. °C 760 mm	472.	2	0.1 mm BP	172.02 0.0749	5	<u> </u>		
100 11111	374.66	4	t <sub>e</sub>	0,0358	5	f' to		l
30 10	330.08 295.6	4 5	30 mm	1.1292	5	g' K h'		1
1	237.2	5	ΔHm cal/g			ļ <u>-</u>		₩
0, 1	192.4	5	ΔHv cal/g			m to	!	1
Press. mn	1893.8	5	0.1 mm 30 mm	44.93	5	•		
e Density		-	BP t	35.70 32.17	5	m'   to		Г
g/ml 20°0	0.8260 0.8227	ž 2	te (d, e)	32.02	5	n' *K		
dt 25 4 30	0.8227 0.8194	¥ 2 4	ΔHv/T <sub>e</sub>	18.81	5	0'		<u> </u>
a	0.8392		d   330 to	66.38	5	Surface tension	29.37	5
b	-0.0366				5	dynes/cm. 20°C	28.45	5
Ref. Index		4	d'   to			40	27.54	5
n <sub>D</sub> 20°0	1 4500		d g/ml v ml/g			Parachor [P] 20°C		
30	1.4568	≠   4	vc ml/g	ł		30		
"C"	0.7377					40	1220 4	5
MR (Obs.)			P <sub>c</sub> mm	<del> </del>		Exp. L.1.%/wt.	1338.4	13
MR (Calc. (nD-d/2)	) 157.210 1.0478	≠ 5 2	0.1 mm	1.0000	5	u.		ļ
Dielectric		5	30 mm BP	1.0000 0.8895	5 <b>5</b>	Dispersion	101. <sup>≠</sup>	2
A   330 to		9 4	te	0.8489	5	Flash Point °C Fire Point		i
B _ 580 °C		4	t <sub>c</sub>			M Spec.		+-
C	149.	7 5	ΔHc kcal/m ΔHf			Ultra V.		1
A*  330 to B*  540 °C		'   5	ΔFf			X-Ray Dif. Infrared		ĺ
к ——-	-	1	Viscosity			Solubility in +		$\vdash$
* to	<del>,  </del>		centistokes 7 °C			Acetone		1
t <sub>x</sub> ; •(	7		'	į		Carbon tet. Benzene		
A'   to						Ether		ł
B', L _ •	-	1 1	B <sup>V</sup>   to	<u> </u>	$\vdash$	n-Heptane Ethanol		1
A1* , to	,		A <sup>V</sup>   °C			Water		
B'* '	3		(B <sup>V</sup> )			Water in		+-
Ac  to			(A <sup>V</sup> )					
Bc tc_*C			c <sub>p</sub> liq. °K			Ĭ		1
Cryos, A			c <sub>p</sub> vap. *K					
consts. B	+	_	c, vap.					
t <sub>e</sub> °C	528.3	5	-vp.	<u> </u>		L <u>.</u>	L	<u> </u>
	rcooled liqu		7 2 7 1			grams/100 gran		t
	CES: 1-Dow			Calc. from de	t. da	ta 5-Calc, by for	mula	
SOURCE:	TION:	AI						
	RE REFER	AI						
- LEKAIU	ne refek	ENCES	) <del> </del>					

No. 37 l-Pentatria contyne NAME STRUCTURAL FORMULA C<sub>33</sub>H<sub>67</sub>C≡CH Molecular C35H68 Mole Ref. Molecular % Pur Weight 488.894 Ref. Ref Ref. F.P. °C F.P. 100% 76. 2 dt/dP f to °C/mm 1 ۰ĸ g 0.1 mm 177,23 5 B. P. 'C h 0.0755 BP 5 479. 760 mm 0.0358 2 5 f to 100 380.84 ۰ĸ g' 30 mm 1.1392 5 30 335,87 4 10 301.0 h' 5 ∆Hm cal/g 242.1 1 5 m AHv cal/g 0. 1 196.9 5 n °К 0.1 mm 30 mm Press. mm 0 44.09 <sup>t</sup>e 1911.2 5 35.02 5 ΒP 31.48 m to Density 5 te (d, e) 0.8268 0.8235 1 n' ۰ĸ g/ml 20°C 2 31.34 5 ۰, 25 dt4 ΔHv/Te 18.77 5 0.8202 30 4 Surface tension d 336 65.37 5 0.8400 4 29.44 5 dynes/cm. 20°C <u>| 537</u> ٠Ç 0.0634 -å--0.0366 Ъ 28.51 5 4 30 to 5 27.61 40 Ref. Index e¹ °C 1.4612 20°C  $\mathbf{n}_{\mathbf{D}}$ Parachor [P] d<sub>c</sub> g/ml 1.4593 2 25 20°C 1.4574 vc ml/g 30 4 t<sub>c</sub> 30 40 "C" 0.7376 4 5 P<sub>c</sub> mm 1377.4 Sugd. 162. Z8 # MR (Obs.) 2 PV/RT Exp. L.1.%/wt. MR (Calc.) 161.828 1.0479 1.0000 0.1 mm 5 (nD-d/2) 2 101. ¥ 30 mm 1.0000 5 2 Dispersion Dielectric 2,13 5 BP 0.8892 5 Flash Point °C 0.8481 A 336 to 7.62600 4 Fire Point 2975.2 B 1587 °C M. Spec. C 148. 5 AHc kcal/m Ultra V. A\*|336 to B\*|547 °C AHf 2.55750 5 X-Ray Dif. ΔFf 2889.0 Infrared Viscosity Solubility in centistokes Acetone tk tx to Carbon tet. °C Benzene A' to Ether B <u>°с</u> B<sup>V</sup> | n-Heptane C to Ethanol °C Water to Water in B'\* °C (B<sup>V</sup>) Acl to (A<sup>V</sup>) Bc °C c<sub>p</sub> liq. ۰ĸ Cryos. A cp vap. •ĸ consts. B° c vap. te .C 536.2 5 # for undercooled liquid grams/100 grams solvent REFERENCES: 1-Dow 2-API 3-Lit. 4-Calc, from det. data 5-Calc. by formula SOURCE: API PURIFICATION: API LITERATURE REFERENCES:

							No. 38	
NAME	l-Hexatriac	ontyr	ie			STRUCTURAL		4
Mole % Pur.	Ref. Mo	lecul		Molecular Weight 502.	920	С <sub>34</sub> Н <sub>69</sub> С <u>≡</u> €	СН	
		Ref.			Ref			Ref.
F.P. °C	77.	2	dt/dP			f   to		
F.P. 100%			°C/mm		_	g °K		
B. P. *C	1		200.97°C BP	175.29	5	h i		
760 mm 100	486. 387.53	2	te	0.03519	5	f' - to		
30	342.34	4	30 mm	1.1453	5	g' °K		
10	307.2 247.4	5	ΔHm cal/g			h'		
Pressure	241.4	1 3	ΔHv cal/g			m   to		
mm 200.97°	0.1000	5	200.97°C	50.77	5	n °K		
te	1954.3	5	30 mm BP	43. 29 34. 37	5	<u> </u>		<del></del>
Density	ŧ		t_	30.84	5	m'   to		
g/ml 20°C	0.8275 <sup>‡</sup> 0.8242 <sup>‡</sup>	2	t <sub>e</sub> (d, e)	30.79	5	0'		
dt 25 4 30	0.8209	4	ΔHv/T <sub>e</sub>	18.98	5	Surface tension		<del> </del>
a	0.8407	4	d   342 to		5	dynes/cm. 20°C	29.50	5
ь	-0.0366	4	e   545 °C		5	30	28.57	5
Ref. Index		2	e'			40	27.66	5
n <sub>D</sub> 20°C	1 4507	1 2	d g/ml v ml/g			Parachor [P] 20°C		
30	1.4577	4	d g/ml vc ml/g tc °C			30		
"C"	0.7378	4	P <sub>c</sub> mm			40 Sugd	1416.4	5
MR (Obs.)		2	PV/RT	<b></b>	_	Exp. L.1.%/wt.	1410.4	<del>  </del>
MR (Calc. (nD-d/2)	166.446 1.0479 <sup>‡</sup>	5 2	200.97°C	1.0000	5	u.	1 4	1
Dielectric	2,14	5	30 mm BP	1.0000	5	Dispersion	101. ≠	2
A   342 to	+	-	t <sub>e</sub>	0.8475	5	Flash Point °C		
B 595 °C		4	tc			Fire Point		<del>├</del>
С	147.	5	AHc kcal/m			M Spec. Ultra V.	1	1
A* 342 to			ΔHf ΔFf		1	X-Ray Dif.		1
B* ∟ <sup>555</sup> °C	2931.19	5	Viscosity		<del> </del>	Infrared		ـــ
° .—	_		centistokes			Solubility in + Acetone		
t <sub>k</sub>  to			η ∘c			Carbon tet.		
A' to	1	$\vdash$		ļ		Benzene Ether		
Bii °C			<u> </u>		<u> </u>	n-Heptane		
C'		<del> </del>	B <sup>V</sup>   to	į.		Ethanol		
A'* to B'* °C			(BV)	-		Water Water in		1
Ac to	<del></del>	+	(B')    (A')					
Bc t °C								
Cc	<b>1</b>	<u> </u>	c <sub>p</sub> liq. °K					
Cryos. A° consts. B°			c <sub>p</sub> vap. *K	1				
t <sub>e</sub> °C	545.40		c <sub>v</sub> vap.	L		L		
for under	cooled liquid					grams/100 gran	ms solven	t
REFERENC	ES: 1-Dow	2-AI	PI 3-Lit. 4-0	alc. from det	t. da	ta 5-Calc, by for	mula	
SOURCE:	API							
	TION: API							
LITERATU	RE REFERE	NCES	<b>:</b>					
1								

								No. 39	
NAME	1-Hept	atria	contyne			ST	RUCTURAL	FORMUL	A
							C35H71 C≣CH	ī	
Mole % Pur.	Ref. Mo	leculi mula		Molecular Weight 516.9	946				
		Ref.			Ref.				Ref
F.P. °C	79.	2	dt/dP			f	to		
F.P. 100%	<del></del>	$\vdash \dashv$	*C/mm 0.1 mm	176.68	5	g	°K		
B. P. °C 760 mm	493.	2	BP	0.0767	5	_h _f'			
100 30	393.21	4	t <sub>e</sub> 30 mm	0.0359	"	g'	to •K		
10	347.49 312.1	4 5	ΔHm cal/g		+	h'			i
1 0, 1	252.2 206.2	5	ΔHv cal/g		+	m	to		
Press. mm	200.2	اثا	0.1 mm	42.50	_	n o	°К		
t <sub>e</sub>	1946.4	5	30 mm BP	42.58 33.76	5	m'			<u> </u>
Density g/ml 20°C	0.8282	2	t <sub>e</sub> (d, e)	30, 21	5 5	n'	to   °K		
dt 25	0 02407	2	ΔHv/T <sub>e</sub>	30.13 18.90	5	0'			
	0.8216	4	d 348 to	63, 64	5		face tension	20	
a b	0.8414	4	_e   553 °C	0.0606	5	dyn	es/cm. 20°C 30	29.56 28.63	5
Ref. Index			d'  to				40	27.72	5
<sup>n</sup> D 20°C	1.4621 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4601 ± 1.4	2	d <sub>c</sub> g/ml	<u> </u>	1	Par	achor [P]		
30	1.4581 ≠	2 4	v <sub>c</sub> ml/g		1 1		20°C 30		
"C"	0.7377	4	t <sub>c</sub> °C P <sub>c</sub> mm				40 Sugd.	1455.4	5
MR (Obs.)	171.56 ≠	2	PV/RT		$\vdash$	Evr	L.1.%/wt.	1133.1	<u> </u>
MR (Calc.) (nD-d/2)	171.034 1.0480≠	5 2	0.1 mm 30 mm	1.0000	5	-	u.	≢	_
Dielectric	2.14	5	BP	1.0000	5		persion sh Point °C	101.	2
A 348 to	7.63157	4	t <sub>e</sub>	0.8468	5		e Point		Ì
B 1603 °C	3031.0 145.	4 5	tc ΔHc kcal/m	-	+		Spec.		
A* 348 to	2,58066	5	∆Hf				ra V. Ray Dif.		
B*[563 °C K	2944.0	5	ΔFf Vicesites		$\vdash$		ared		
c			Viscosity centistokes				ubility in +		
t <sub>k</sub> to t <sub>c</sub> °C	1		ŋ °C				etone rbon tet.		
t <sub>x  </sub> °C							enzene her		
B'°C			B <sub>v</sub> to		+	n-	Heptane		
A¹* to		-	B to C	1	1 1		hanol ater		
B'* °C			(B <sup>V</sup> )				ater in		<u> </u>
Acl to			(A <sup>V</sup> )						
Bc tc C	-		c <sub>p</sub> liq. °K						
Cryos, A° consts, B°			c <sub>p</sub> vap. *K						
t <sub>e</sub> °C	553.0	5	c <sub>w</sub> vap.						
	rcooled liquid						rams/100 gra		ıt
	ES: 1-Dow			Calc. from de	et. da	ta 5	-Calc. by for	mula	
SOURCE:	701	API					-		
PURIFICAT	ION: RE REFEREI	API							

								No. 40	
NAME	1 -Oc	tatriac	ontyne			STRUCT	URAL I	FORMULA	
						C. H.	73С≣СН		
Mole	Ref. M	(alaan)		Molecular		36	73		
% Pur.	Kei. M	ormul	ar C <sub>38</sub> H <sub>74</sub>	Weight 530.	972				
		Ref.			Ref.				Ref.
F.P. *C	80.	2	dt/dP			f	to		
F.P. 1007	<u>'</u>	-	*C/mm 0.1 mm	178.05	5	g	•K		
B.P. *C 760 mm	499.	2	BP	0.0773	5	<del>-</del> + -		İ	
100 30	398.51 352.45	4	t <sub>e</sub> 30 mm	0.0359	5 <b>5</b>	f' g'	to •K		l
10	316.8	5	ΔHm cal/g	1.1007	+-	h' i			ł
1 0, 1	256.4 210.1	5	ΔHr cal/g	+		m l	to	l	$\vdash$
Press. mm		+-	0.1 mm			n i	•K		l
t <sub>e</sub>	1962.3	5	30 mm BP	41.81 33.14	5				<u> </u>
Density g/ml 20°C	0.8289	<i>į</i>	l t	29.59	5	m'	to •K		l
dt 25	1 00054		e (4, 5,	29.52 18.86	5	0'		ĺ	l
	0.0223	4	ΔHv/T <sub>e</sub>	<del> </del>	5	Surface t	ension		
a b	0.8421 -0.0366	4	e   560 °C		5	dynes/cm	a. 20°C	29.62 28.69	5
Ref. Index			d'   to			1	40	27.78	5
<sup>n</sup> D 20°C	1		d <sub>c</sub> g/ml		$\vdash$	Parachor			
30	1.4605	4 4	vc ml/g tc °C				20°C 30		l
"C"	0.7377	4					40		_
MR (Obs.)			P <sub>c</sub> mm	<del> </del>	-	Exp. L.1		1494.4	5
MR (Calc. (nD-d/2)	) 175.652 1.0480 <sup>7</sup>	4 5 2	0.1 mm	1.0000	5	u		1	
Dielectric		5	30 mm BP	1.0000	5	Dispersion		101.#	2
A 353 to			t <sub>e</sub>	0.8466	5	Flash Poir			
B <u>[610</u> •0	3057.7 144.	4 5	t <sub>c</sub> ΔHc kcal/m	<del> </del>	-	M Spec.			
A*   353 to			ΔHf			Ultra V. X-Ray Di	e .		
B* 570 °C	2970.0	5	ΔFf	<del> </del>	<del> </del>	Infrared	<b></b>		
С			Viscosity centistokes			Solubility			
tk   to			η •c	1		Acetone Carbon			
t c		+	·		1	Bensene Ether	t	l	Ì
B' •	2		B <sup>V</sup>   to		-	n-Hepta:	ne	•	İ
A¹* to	<del></del>	+	B' to			Ethanol Water			
B'* *C			(BV)	-[		Water in	·		<u> </u>
Ac  to			(A <sup>V</sup> )	1					
Bc tc_*C	4		c <sub>p</sub> liq. •K					1	
Cryos, A			c <sub>p</sub> vap. *K					1	
consts. B	<del> </del>	-	l -					1	
t <sub>e</sub> °C	560.0	5	c <sub>v</sub> vap.		L	Ļ		L	<u>L</u>
	cooled liquid		OT 3 744 4	Cala (==== 1				ms solven	<u>t </u>
SOURCE:		2-AF		Calc, from de	t. Q8	ua 5-Calc.	. by ior	mula	
PURIFICA"	TION:	AF							
	RE REFER								

Mole % Pur.	1 - N	onatr	iaco	ntuna		- 1				
				ityne			ST	RUCTURAL	FORMUL	.A
								C <sub>37</sub> H <sub>75</sub> C≡CH	I	
	Ref.	Mol For	ecula mula		Molecular Weight 544.9	98		37 75		
			Ref.	r	T	Ref.		<del></del>		Ref
F.P. °C	82.		2	dt/dP			f	to		T
F.P. 100	%			°C/mm			g	l °K		1
B.P. °C				0.1 mm BP	179.41	5 5	h	l		1
-760 mm 100	505. 403.8	,	2	t <sub>e</sub>	0.0359	5	f'	to		
30	357.4		4 4	30 mm	1.1754	5	g'	•к		
10 1	321.5 260.7		5	∆Hm cal/g			h'			↓_
0.1	214.0		5	ΔHv cal/g			m n	to or to		
Press. mr				0.1 mm 30 mm	41.00	_	0			Ī
<sup>t</sup> e	1975.5		5	BP	41.09 32.53	5	m'			╁
Density g/ml 20°	റിരം	295#	2	te (d. a)	28.97	5	n'	to   °K		
at 25	0.8	2627	2	t <sub>e</sub> (d, e) ΔHv/T <sub>e</sub>	28.95	5	٥'			
4 30	0.8	229	4		<del></del>	1	Sur	face tension		$\top$
<b>a</b> b	-0.0		4	d 358 to e 567 °C	61.83	5		es/cm. 20°C	29.67	5
Ref. Inde		300	-	d'			•	30 40	28.74 27.83	5
<sup>n</sup> D 20°	ā 1 1 1.	628‡	2	e'	<del></del>	+	Par	achor [P]		+-
25 30		608 588	2 4	dc g/ml vc ml/g				20°C		
"C"	0.7		4	v <sub>c</sub> ml/g t <sub>c</sub> °C				30 40		
MR (Obs.			2	P <sub>c</sub> mm					1533.4	5
MR (Calc	1 180 2	70	5	PV/RT			Exp	L.1.%/wt.		
(nD-d/2)	1.04	481 <sup>≠</sup>	2	0.1 mm 30 mm	1.0000	5	Dis	u. persion	101.≠	2
Dielectric		4	5	BP	0.8881	5		sh Point °C		┿
A 358 to		1056	4	te t <sub>C</sub>	0.8453	5		e Point		
C 1010 7	143.		4 5	ΔHc kcal/m	<del> </del>	+-1		Spec.		
A*1358 to	2.60	759	5	ΔHf ΔFf				ra V. Ray Dif.		
B* 576 °C	2996.9		5	·	<del> </del>	+		ared		
c				Viscosity centistokes				ubility in +		
t <sub>k</sub>				<b>" °</b> C				etone rbon tet.		
'x							Be	nzene		
A'   to				ļ		1		he <i>r</i> Heptane		
C'				BV to			Et	hanol		
A'* to				<del></del>	_			ater ater in		
B'* °C	<del></del>			(B <sup>V</sup> )			— <u> </u>			+
Acl to				(A <sup>V</sup> )	<del> </del>	+				
Cc L				c <sub>p</sub> liq. °K						
Cryos. A consts. B				c <sub>p</sub> vap. °K						İ
t <sub>e</sub> °C	566.9		5	c <sub>w</sub> vap.						
≠ for und	ercooled l	i <b>q</b> uid					+ g:	rams/100 gra	ms solve	nt
REFEREN	CES: 1-I	Dow .	2-A	PI 3-Lit. 4-	Calc. from de	et. da				
SOURCE:		A	API							
PURIFICA			API							
LITERAT	URE REF	EREI	NCES	<b>5</b> :						

	<del></del>							No. 42	
NAME	l-Tetracontyne				_	STRUCTURAL FORMULA			
Mole % Pur.	Ref.	Mole	ecula mula	C <sub>40</sub> H <sub>78</sub>	Molecular Weight 559.	024	C <sub>38</sub> H <sub>77</sub> C≡C	Н	
			Ref.			Ref.			Ref
F. P. °C	83.		2	dt/dP			f to		
F.P. 100%				°C/mm			g °K		1
B. P. 'C	513			0.1 mm BP	181.05 0.0784	5	h '		
760 mm 100	512. 409.99	,	2	t <sub>e</sub>	0.0359	5	f' to		
30	363, 21		4	30 mm	1.1854	5	g' c'K	İ	1
10 1	327.0 265.6	- 1	5	ΔHm cal/g		П	h!		
0. 1	218.5	- 1	5	ΔHv cal/g		$\Box$	m   to		ļ
Press. mm	1993.7		5	0.1 mm 30 mm BP	40.45 32.01	5	n o K		
Density		4		t_	28.45	5	m'   to		l
g/ml 20°C dt 25 d4 30	0.83 0.82 0.82	:687	2 4	te (d, e)  AHv/Te	28.45 18.75	5 5	n' o'		<u> </u>
	0.84		4	d   364 to		5	Surface tension dynes/cm, 20°C	29.72	5
Ъ	-0.03		4	-a,, 575_ <b>*</b> €		5	30	28.79	5
Ref. Index		#	٦	e' •C		$\perp$	40	27.87	5
<sup>n</sup> D 25	1.46	12#	2	d g/ml vc ml/g			Parachor [P] 20°C		
30	1.45	92*	4	v <sub>c</sub> ml/g t <sub>c</sub> °C	1		30		
"C"	0.73		4	P <sub>c</sub> mm			40 Suad	1572.4	5
MR (Obs.)		#	2	PV/RT	<del> </del>	╁─┤	Exp. L.1.%/wt.	1572.4	13
MR (Calc. (nD-d/2)	184.88	81 <sup>#</sup>	5 2	0.1 mm	1.0000	5	u.	4	1
Dielectric	2.14		5	30 mm BP	1.0000 0.8881	5	Dispersion	101. ≠	2
A   364 to		695	4	t <sub>e</sub> t <sub>c</sub>	0.8449	5	Flash Point °C Fire Point		
B [ 625 °C	142.		<b>4</b> 5	∆Hc kcal/m	<del> </del>		M Spec. Ultra V.		
A*   364 to B*   585 °C		122	5	ΔHf ΔFf			X-Ray Dif. Infrared		
K — — —	]		}	Viscosity centistokes			Solubility in +		†
t <sub>k</sub>   to		ļ		7 00			Acetone Carbon tet.		
A' to							Benzene		
B' ∟ °C			ŀ	ļ.,	<b></b>	4	Ether n-Heptane		
C'	ļ			B <sup>V</sup> to A <sup>V</sup> °C			Ethanol Water		1
A'* to B'* °C		1	l	(BV)	-		Water in		
Ac  to Bc  tc °C	1			(A <sup>V</sup> )					
Cc				c <sub>p</sub> liq. °K	1				1
Cryos, A° consts, B°				c <sub>p</sub> vap. °K					
t <sub>e</sub> °C	575.0		5	c <sub>v</sub> vap.	<u> </u>				
# for unde							grams/100 gra	ms solven	t
REFERENC	ES: 1-D	ow 2	-AF	PI 3-Lit. 4-0	Calc. from de	t. dat	ta 5-Calc. by for	mula	
SOURCE:			AP						
PURIFICAT LITERATU		FD FX	AP						
				•					

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